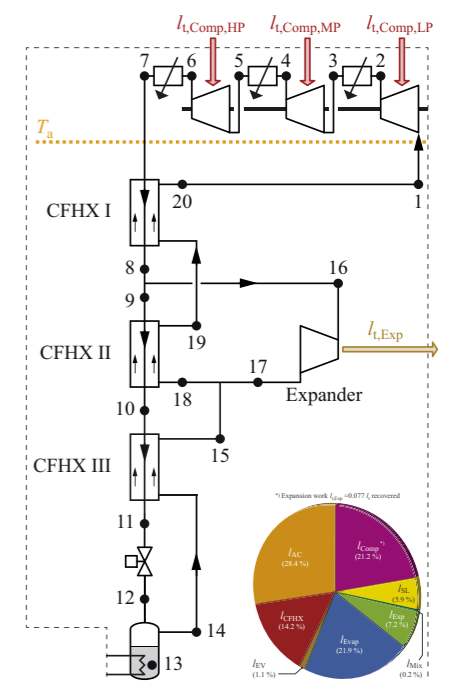
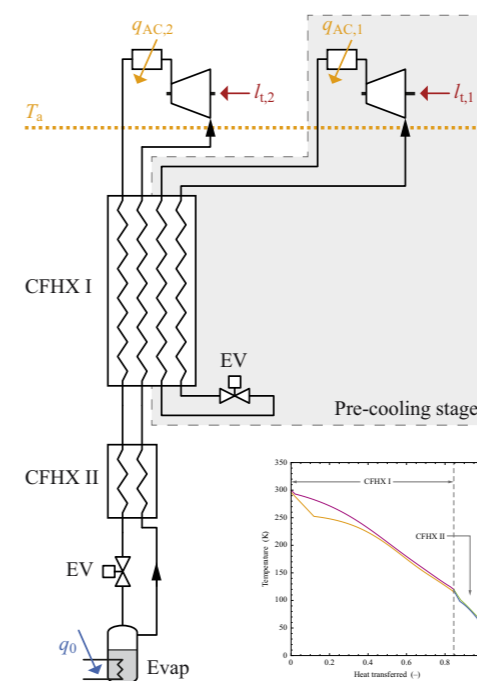
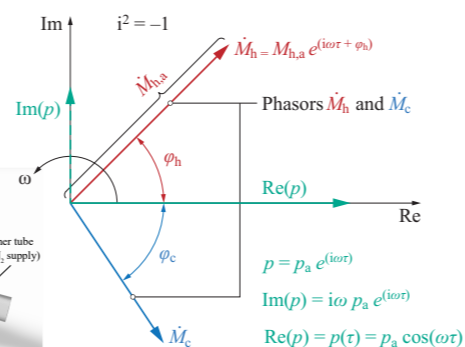
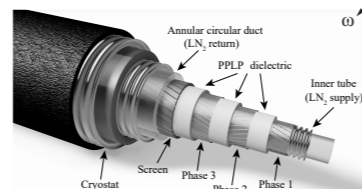
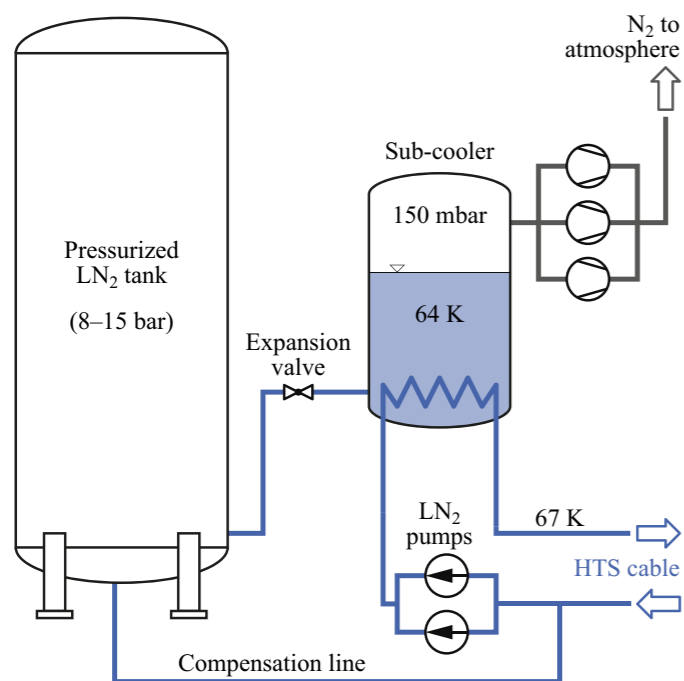


# Cooling technology for HTS power applications

Prof. Dr.-Ing. Steffen Grohmann  
ESAS Summer School, Bologna, June 8-14, 2016

INSTITUTE OF TECHNICAL THERMODYNAMICS AND REFRIGERATION (ITTK)  
INSTITUTE OF TECHNICAL PHYSICS (ITEP)



# Outline

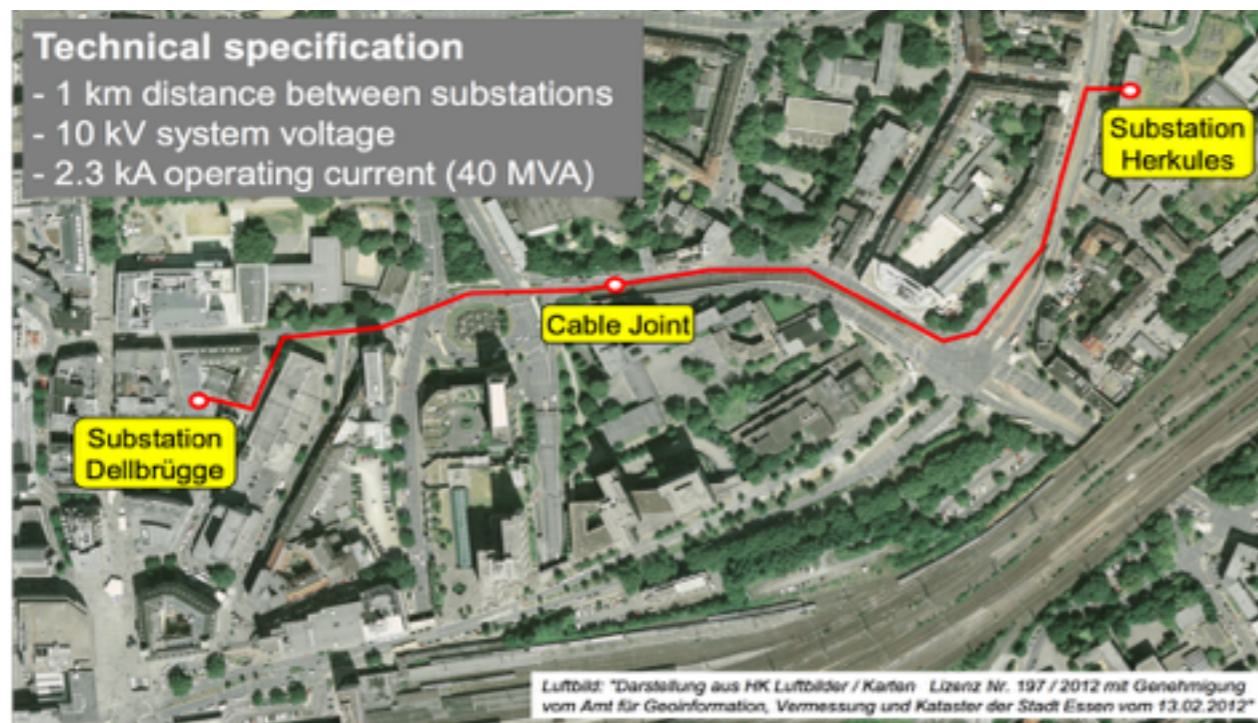
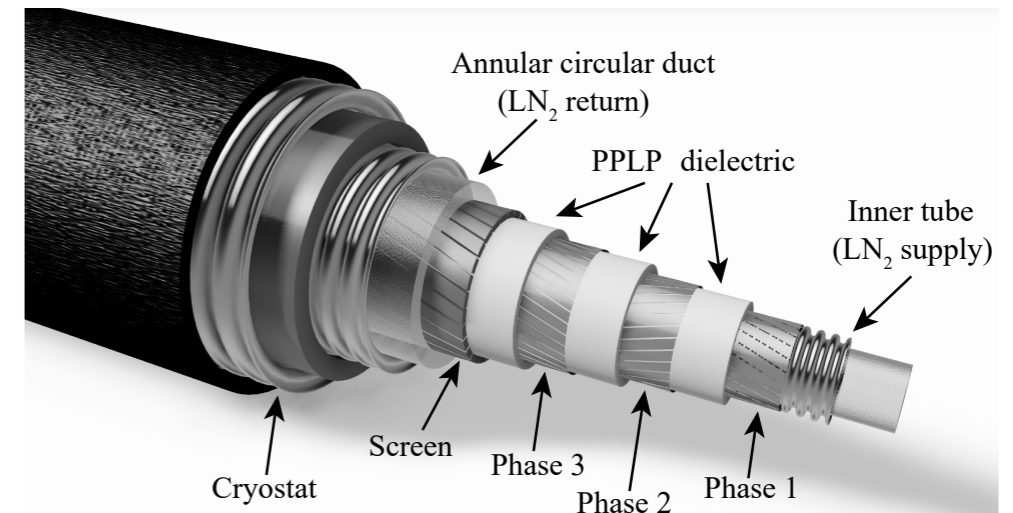
- Taking into account different backgrounds of the course participants, the *aim* of this course is
  - to *provide an overview* of existing HTS cooling technologies
  - to *explain* their *technological differences*
  - to *discuss* future development *potentials*
  
- Therefore, the course is structured in the following sections
  - ▶ Technology overview
  - ▶ Cryocoolers
  - ▶ Cryoplants (refrigerators)
  - ▶ Development potentials
  - ▶ Summary

Cooling technology for HTS power applications  
**TECHNOLOGY OVERVIEW**

# Three-phase HTS power cables

## Example

- *AmpaCity* project
- Partners RWE, Nexans, KIT
- Cable length  $L = 1000 \text{ m}$
- Voltage  $U = 10\,000 \text{ V}$
- Nominal current  $I_N = 2\,300 \text{ A}$
- Nominal capacity  $P_N = \sqrt{3} \cdot U \cdot I_N = 40 \text{ MW}$



Cable installation route in downtown Essen, Germany

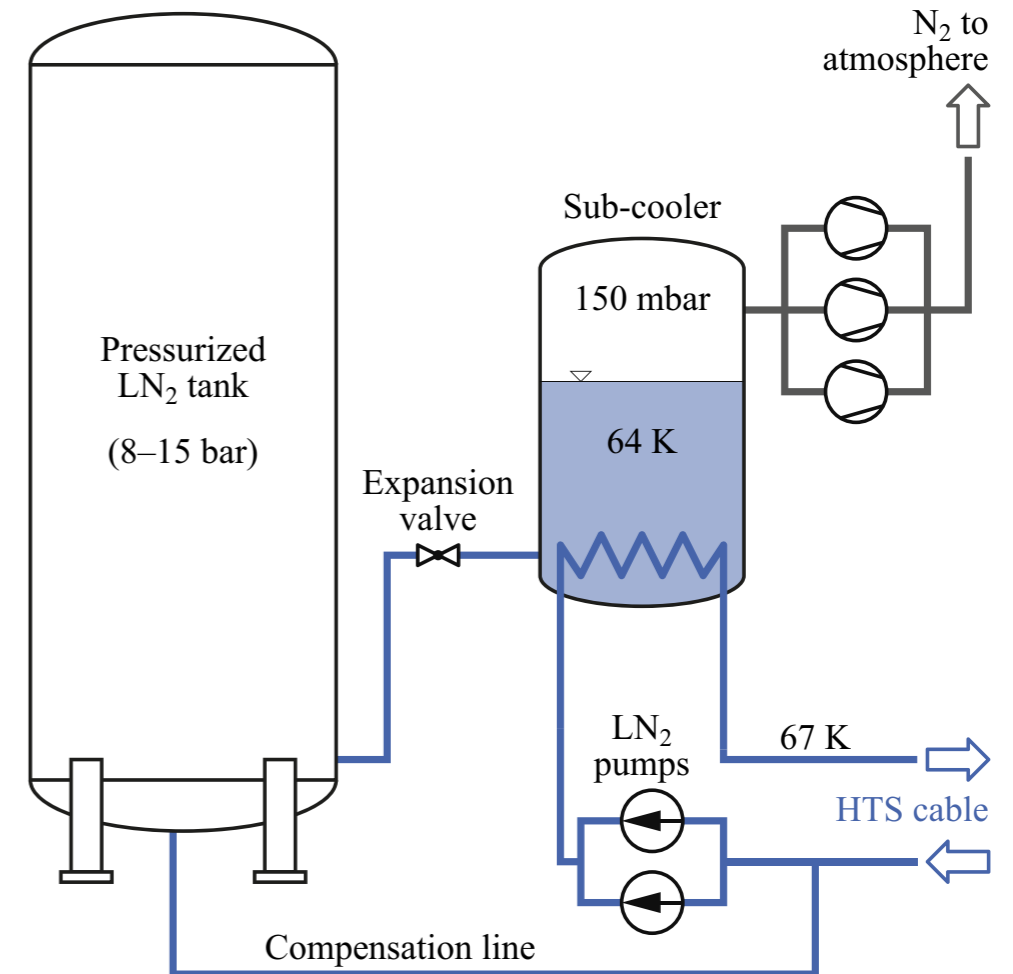


Official start of field test on April 30, 2014

# Cooling unit in the AmpaCity project

## ■ Specification<sup>1)</sup>

- Sub-cooled LN<sub>2</sub> cooling of the HTS cable
- Open LN<sub>2</sub> cooling cycle
- Each one redundant LN<sub>2</sub> and vacuum pump
- Nominal cooling capacity  $\dot{Q}_{0,N} = 4 \text{ kW @ } 67 \text{ K}$
- Nominal LN<sub>2</sub> consumption  $\dot{M}_{\text{LN}_2,N} = 110 \text{ kg/h}$



1) Herzog, F.; Kutz, T.; Stemmler, M.; Kugel, T.: Cooling unit for the AmpaCity Project – one year successful operation. IWC 2015, Matsue - Japan

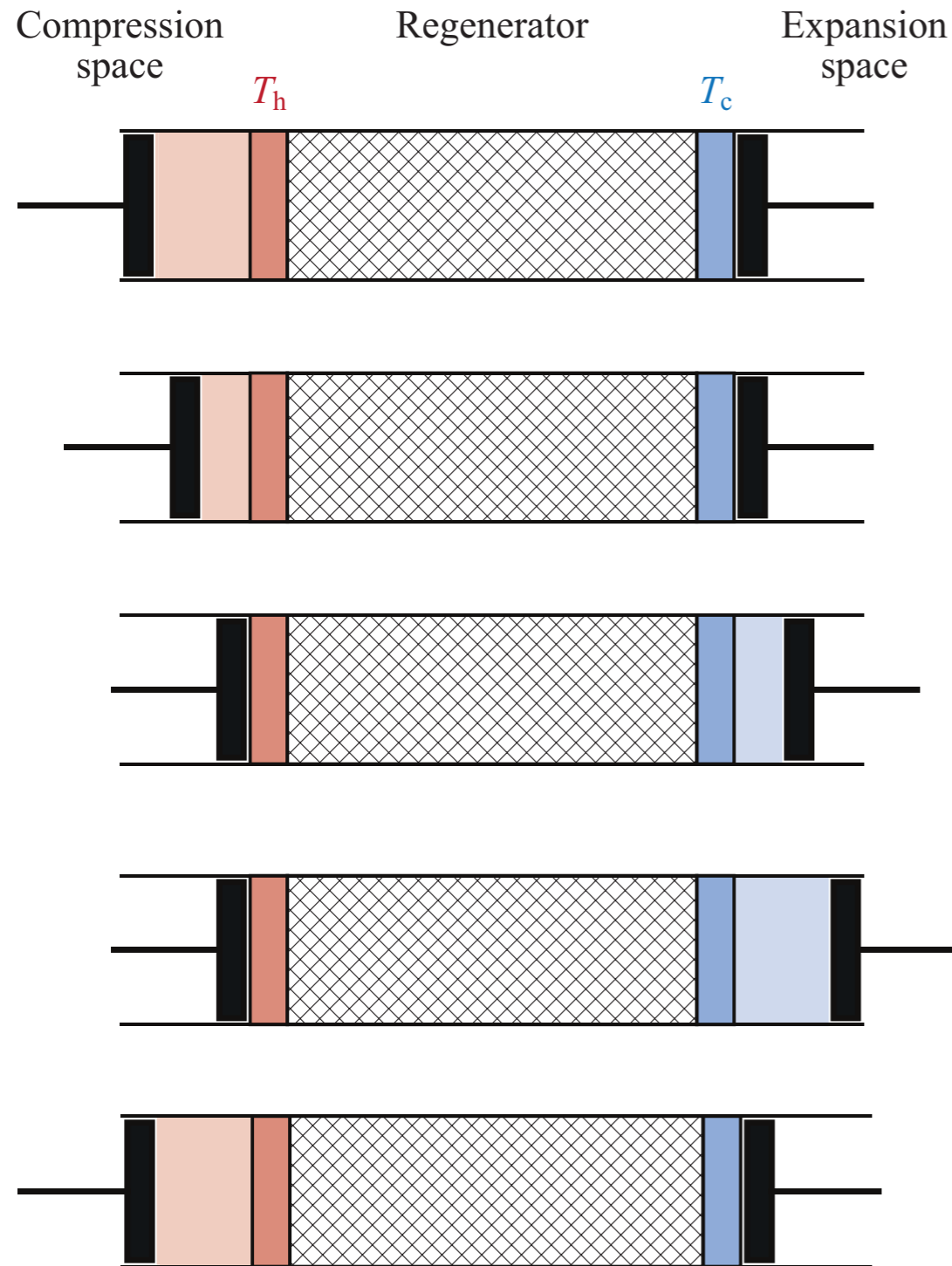
# Cooling methods

## General overview

	Open cycle	Closed cycle	
		Cryocooler	Cryoplant (refrigerator)
Principle	Evaporation of LN <sub>2</sub> obtained from air separation units (ASU) <u>Air composition:</u> <ul style="list-style-type: none"> <li>● 78.09 Vol% N<sub>2</sub></li> <li>● 20.95 Vol% O<sub>2</sub></li> <li>● 0.934 Vol% Ar</li> </ul> + CO <sub>2</sub> , Ne, He, Kr, H <sub>2</sub> , Xe, O <sub>3</sub> , Rn + Water, HCs, CFCs, dust, ...	Oscillating regenerative process	Continuous recuperative cycle
Capacities	Any range	Small	Large
Temperatures	$T_{\min} > T_{\text{tr},\text{N}_2} = 63 \text{ K}$	Any down to ~2 K	Any down to ~1 K
Operation supplies	Electricity LN <sub>2</sub>	Electricity (Cooling water)	Electricity (Cooling water)
Other criteria	Investment cost, operating cost, reliability, space requirements, noise, maintenance etc. are <i>project-dependent</i>		

Cooling technology for HTS power applications  
**CRYOCOOLERS**

# Principle of Stirling cryocoolers



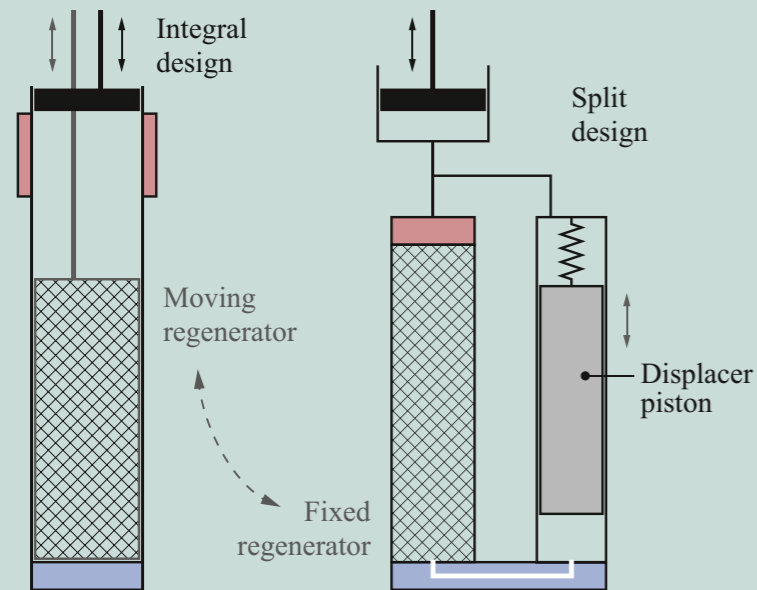
## Steps in a Stirling cycle

- I→II: The gas is in the compression space. During *isothermal* compression, heat is removed at  $T_h$
- II→III: Isochoric displaced ( $V = \text{const.}$ ) of the gas through the regenerator, cooling the gas down to  $T_c$
- III→IV: The gas is *isothermally* expanded in the expansion space, absorbing the heat  $Q_c$  at  $T_c$
- IV→I: Isochoric displacement back in the compression space, warming the gas up to  $T_h$  while cooling the regenerator



# Cryocooler types

## Stirling



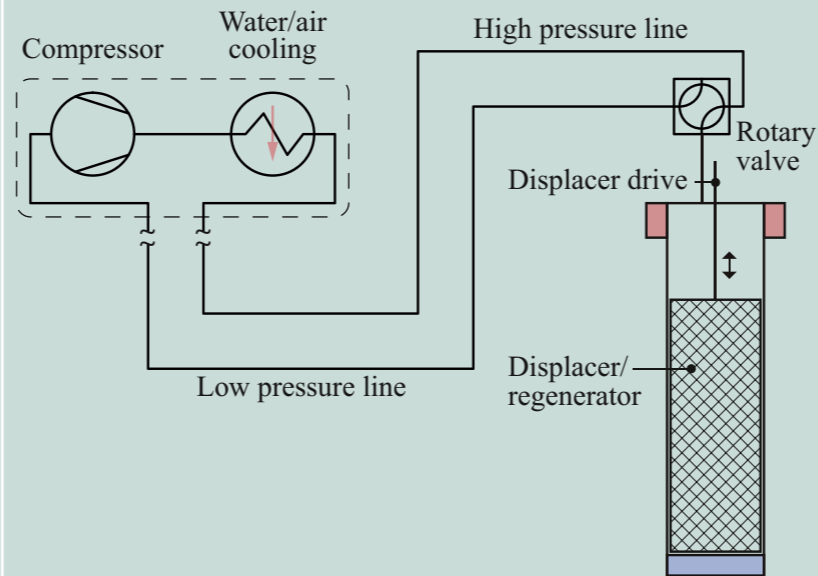
### Integral Stirling

- Build-in 90° phase shift
- + High efficiency, low weight
- Vibration, noise, lifetime

### Split Stirling

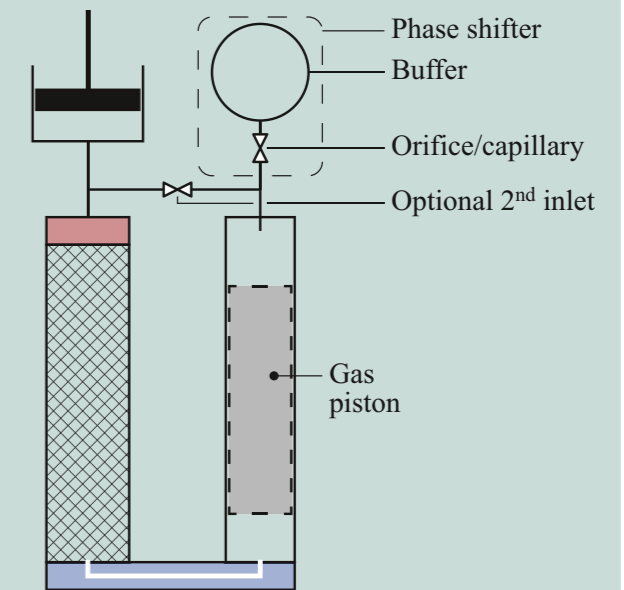
- Resonance frequency depending on load and ambient temperature
- + Low vibration and noise
- Less efficient

## Gifford-McMahon



- Continuously working compressor unit (air/water cooled)
- Compression and expansion by switching to high-pressure and low-pressure lines (amount of fluid contained changes)

## Pulse tube



- Replacement of the displacer piston by a gas piston (wear, reliability)
- Control of the phase shift between mass flow and pressure waves by orifice or capillary combined with a buffer vessel
- Thermodynamically equivalent to Stirling cycle

# Cryocooler types (examples)

## Stirling



0.5-4 W  
@ 77 K

Source: <http://www.aim-ir.de/>

Source: <http://www.stirlingcryogenics.com/>

1000 W  
@ 77 K



## Gifford-McMahon



Source:  
<http://www.shicryogenics.com>



200 W  
@ 77 K



140 W  
@ 77 K



Source: <https://leyboldproducts.oerlikon.com/>

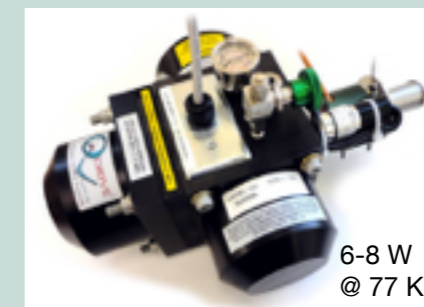


## Pulse tube



90 W  
@ 77 K

Source: <http://www.cryomech.com/>



6-8 W  
@ 77 K

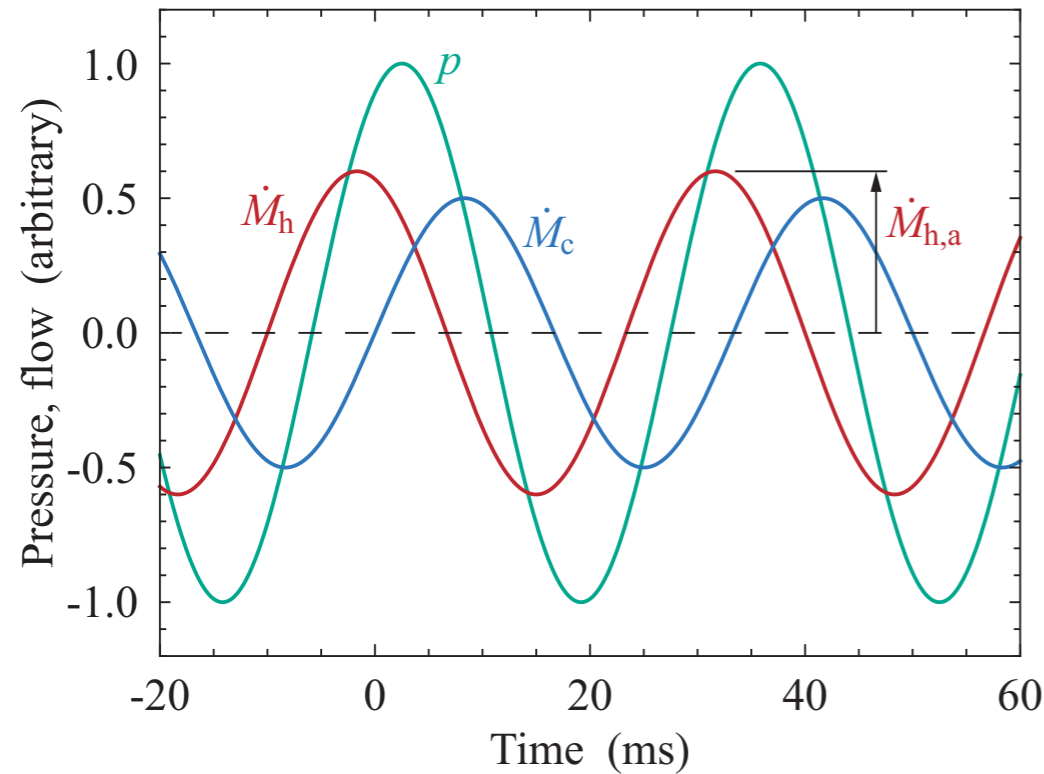
Source: <http://www.chartindustries.com>



150 W @ 77 K

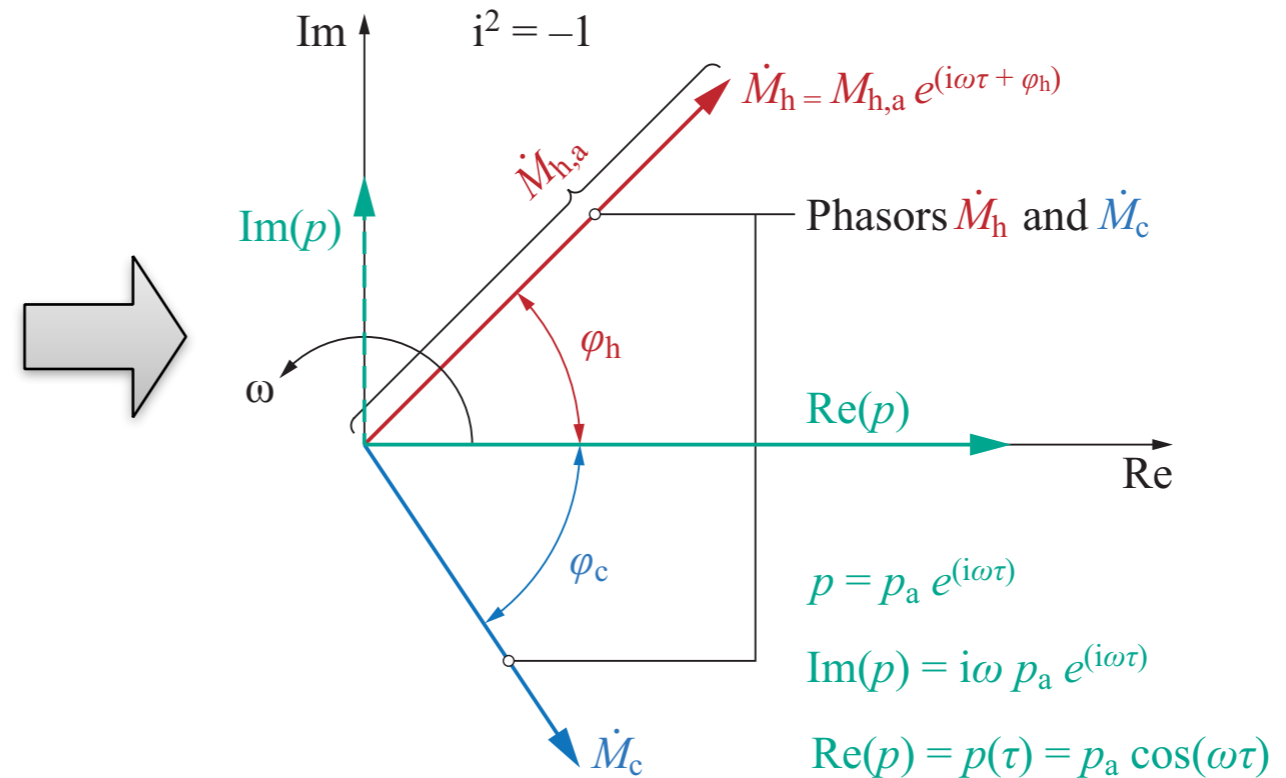
# Thermodynamic modelling

## Time domain



- Sinusoids
- Math: Differential equations

## Frequency domain



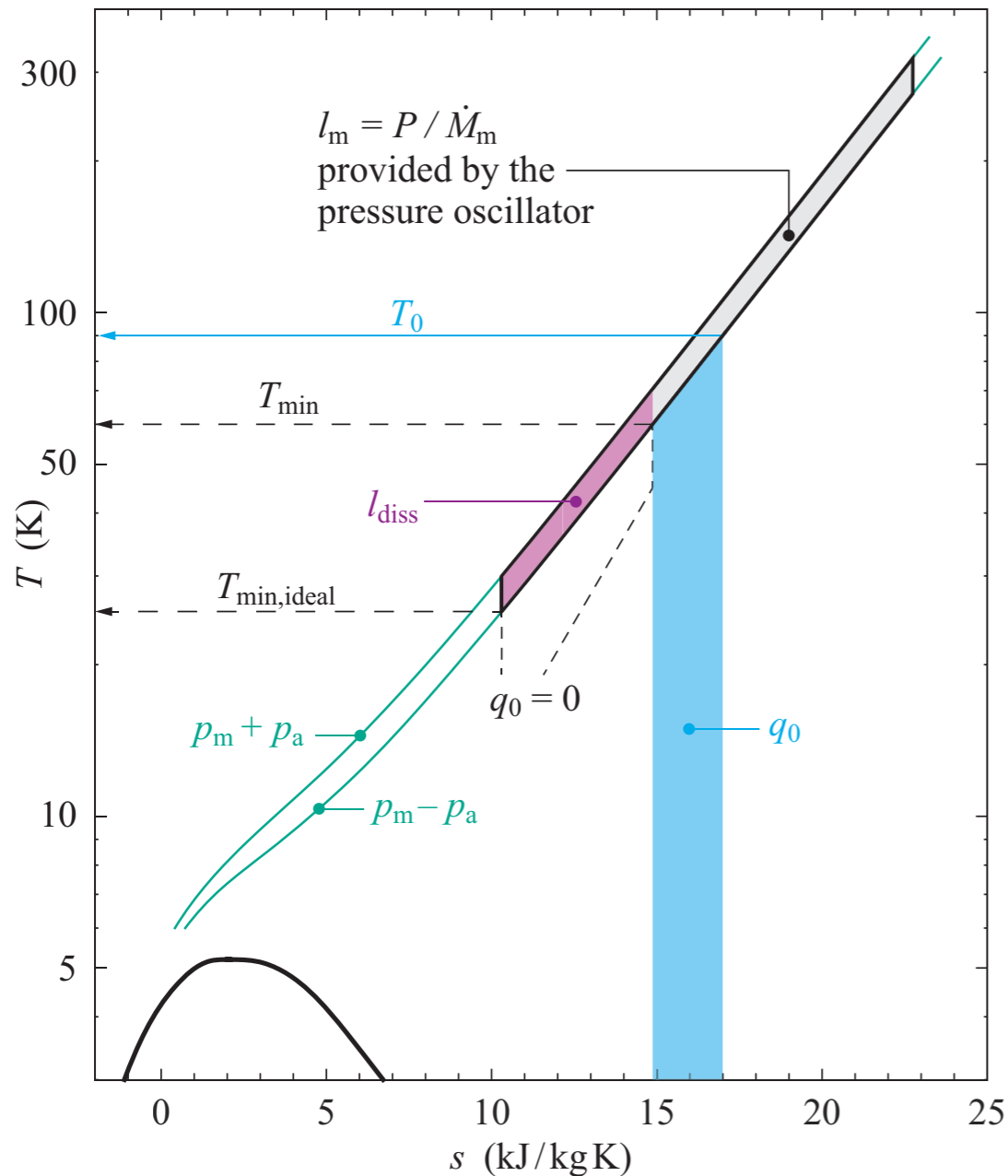
- Phasors ( $\tau = 0$ , complex space)
- Math: Algebra, real and imaginary parts

## Acoustic power of the pressure oscillator for the real gas EOS $p v = Z R T$

$$\triangleright P = \frac{1}{2} Z R T_m \dot{M}_a \frac{p_a}{p_m} \cos(\varphi_{\dot{M}p})$$

# Illustration of cryocooler operation

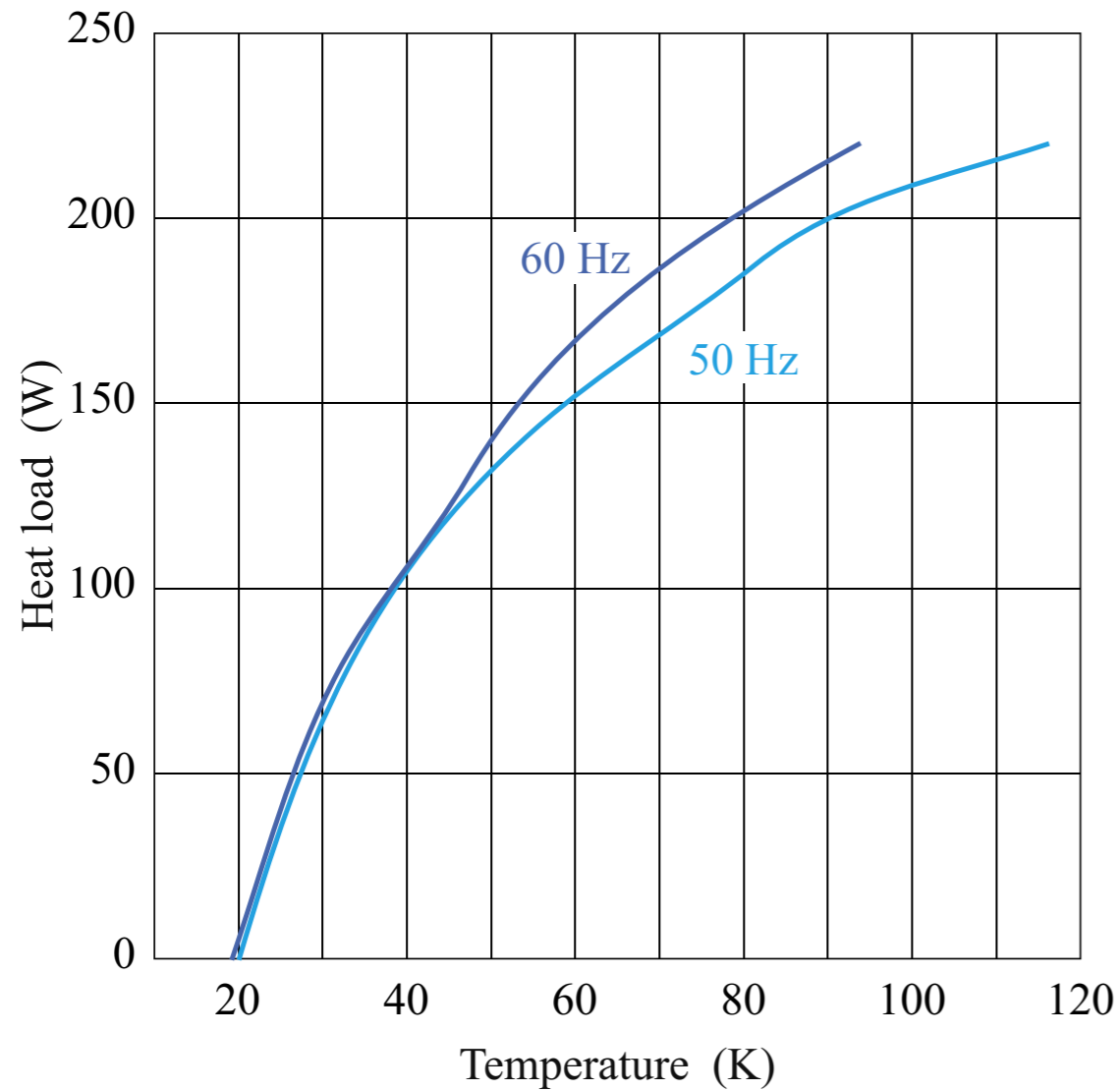
## ■ $T, s$ – diagram (simplified)



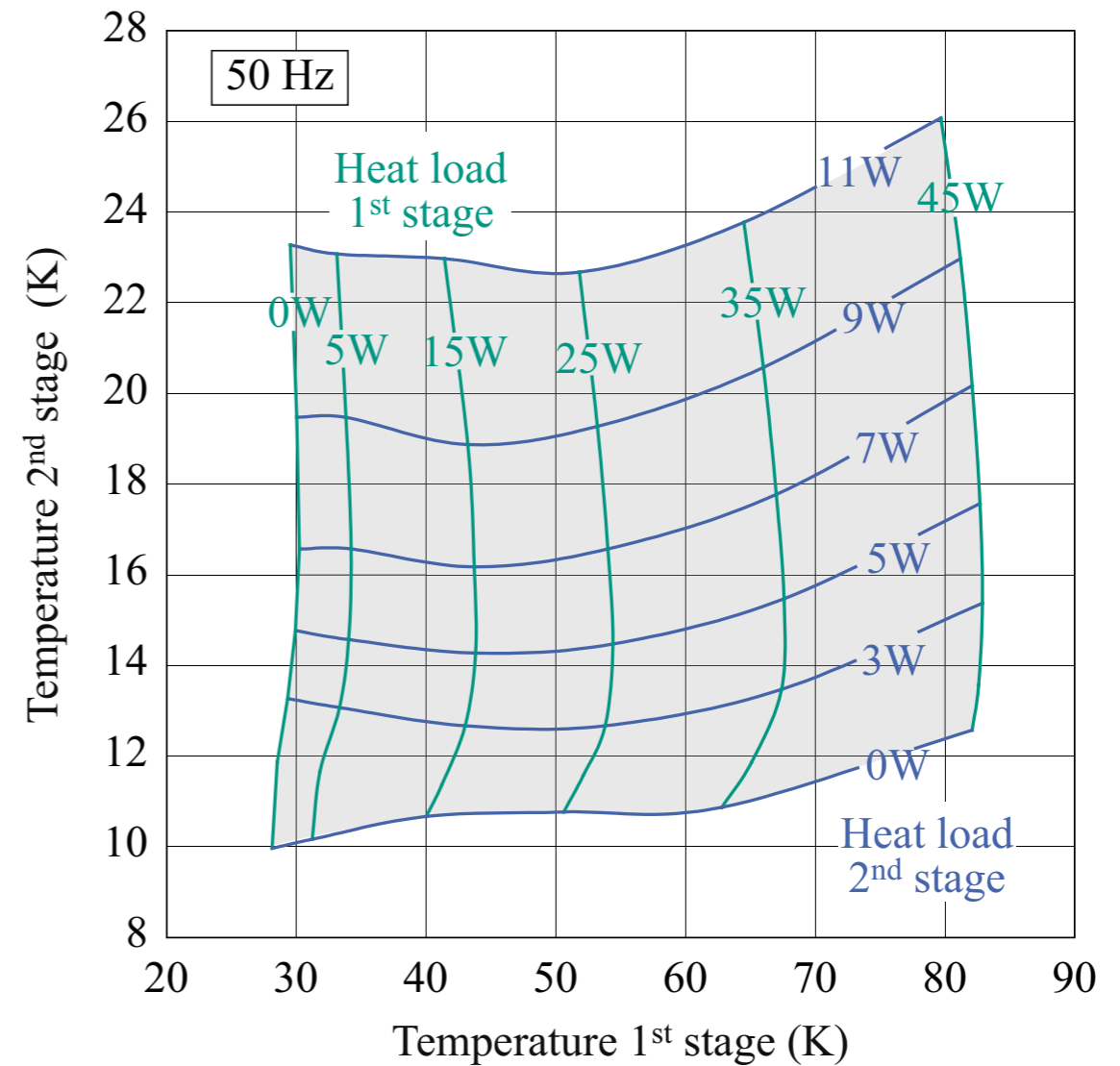
- Consider  $P$  as the average *acoustic power* provided by the pressure oscillator, causing pressure oscillations  $p_m \pm p_a$ 
  - Minimum temperature  $T_{\min,ideal}$  of the ideal cryocooler cycle at zero load  $q_0 = 0$
- Additional entropy production due to gradients in  $\{T, p\}$  in a *real* cryocooler
  - Minimum temperature  $T_{\min}$  of the real cryocooler at zero load  $q_0 = 0$
- Cooling power  $q_0$  available at  $T_0 > T_{\min}$ 
  - Work available to compensate the entropy difference between **absorbing** / **dissipating** the heat  $q_0$  between  $T_0 / T_h$
  - The higher the temperature  $T_0$ , the larger the cooling capacity  $q_0$
  - ▶ **Strong dependence**  $q_0 = f(T_0)$

# Exemplary performance plots

## Single-stage cryocoolers



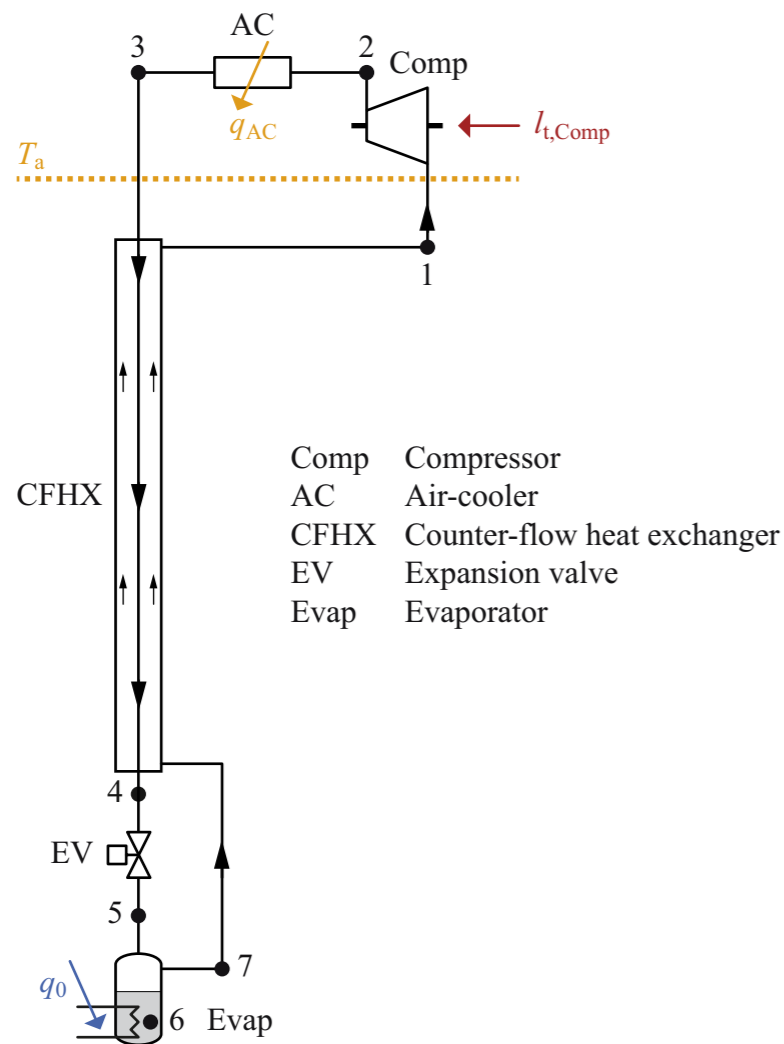
## Two-stage cryocoolers



Cooling technology for HTS power applications  
**CRYOPLANTS (REFRIGERATORS)**

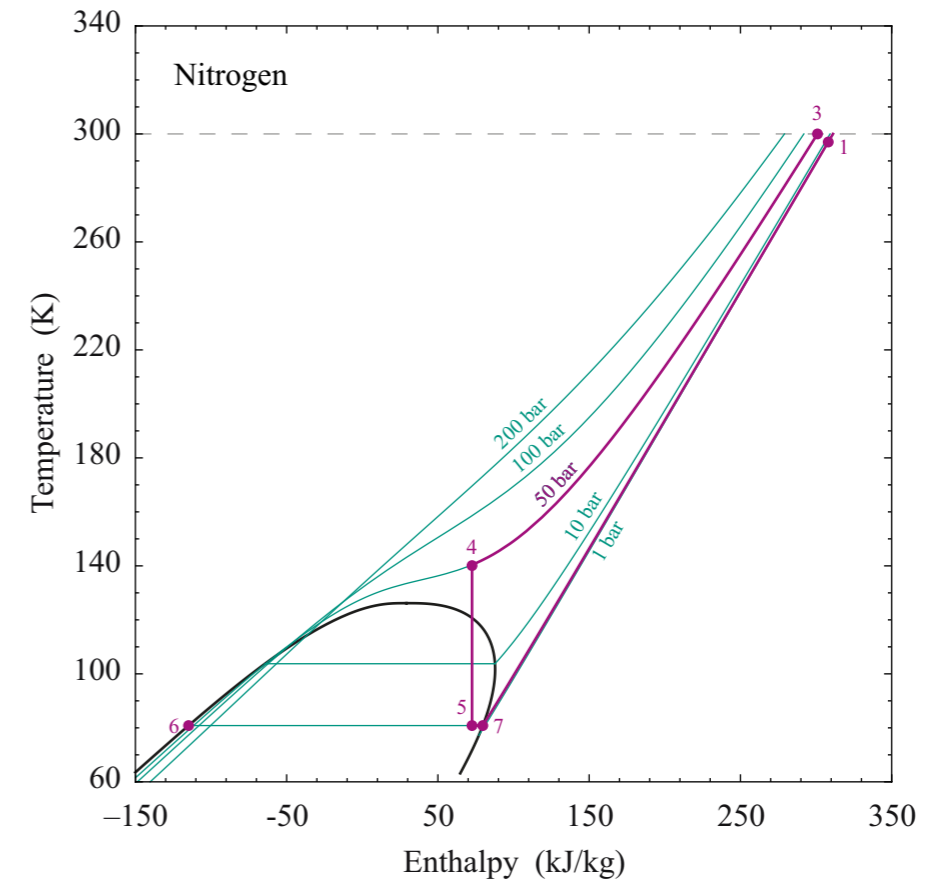
# Linde-Hampson cycle

## Layout



- ▶ **Simplest** configuration
- ▶ Positive Joule-Thomson coefficient  $\mu_{JT} = (\partial T / \partial p)_h$  required

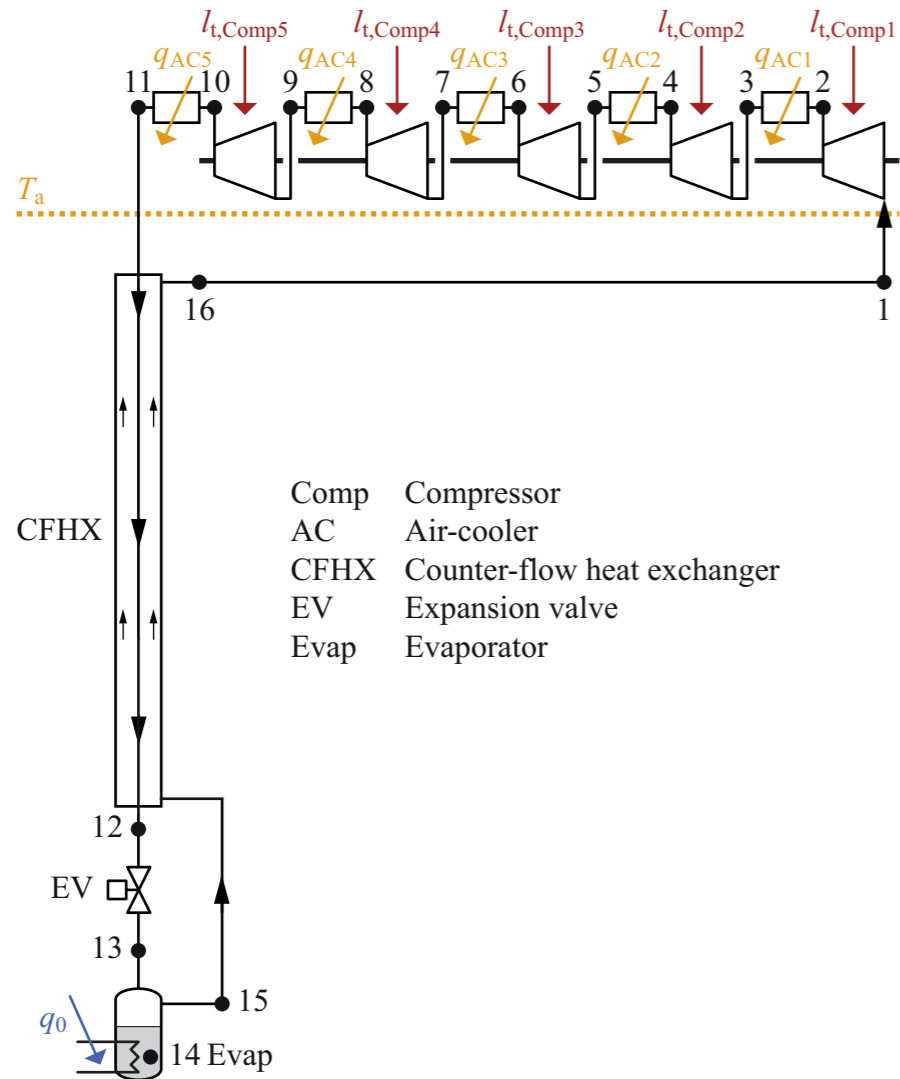
## Process in $T, h$ -diagram



- ▶ Very low cooling capacity  $q_0 = h_7 - h_5$
- ▶ Very low efficiency  $\eta = q_0 / l_t$
- ▶ Warm process part at  $T > T_a$  not shown

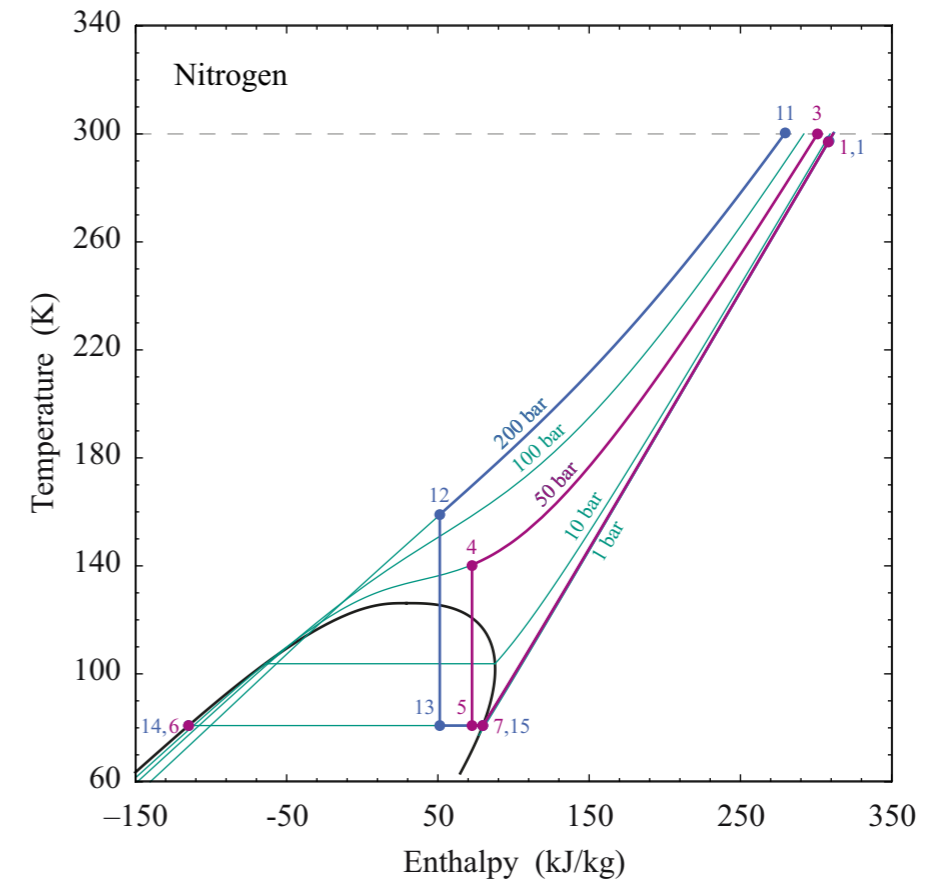
# Linde-Hampson cycle

## Layout



- ▶ Multi-stage compression to e.g.  $p_H = 200$  bar
- ▶ Different number of stages depending on compressor technology (flow rates)

## Process in $T, h$ -diagram

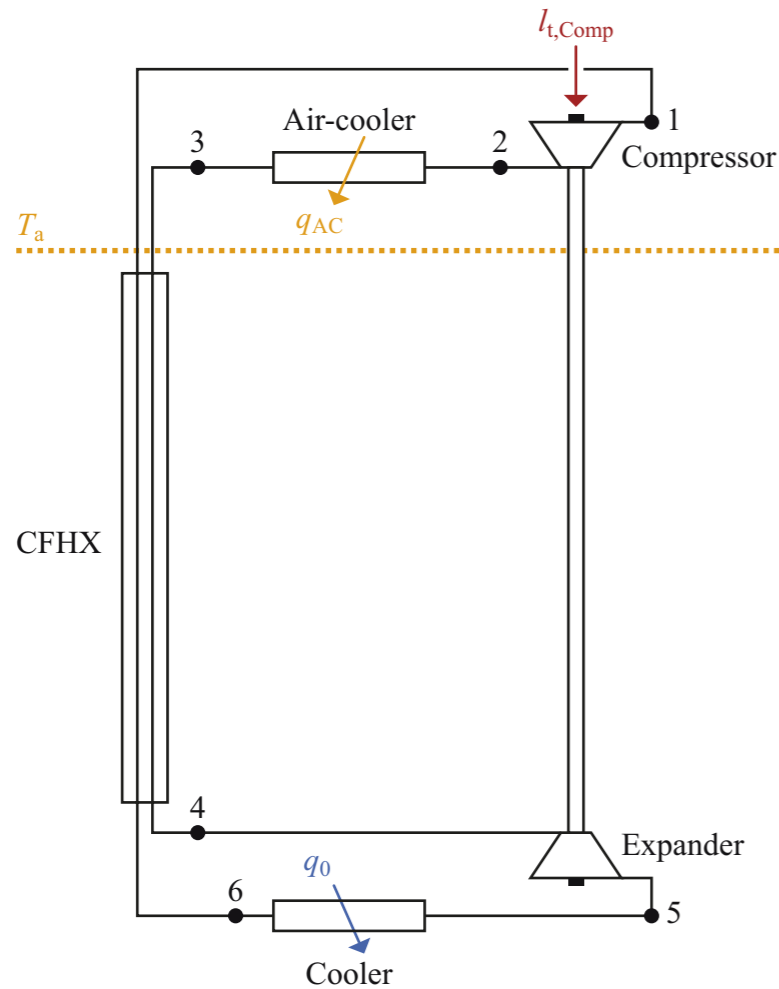


- ▶ Higher specific cooling capacity  $q_0 = h_7 - h_5$
- ▶ Still low efficiency  $\eta = q_0 / l_t$
- ▶ Warm process part at  $T > T_a$  not shown

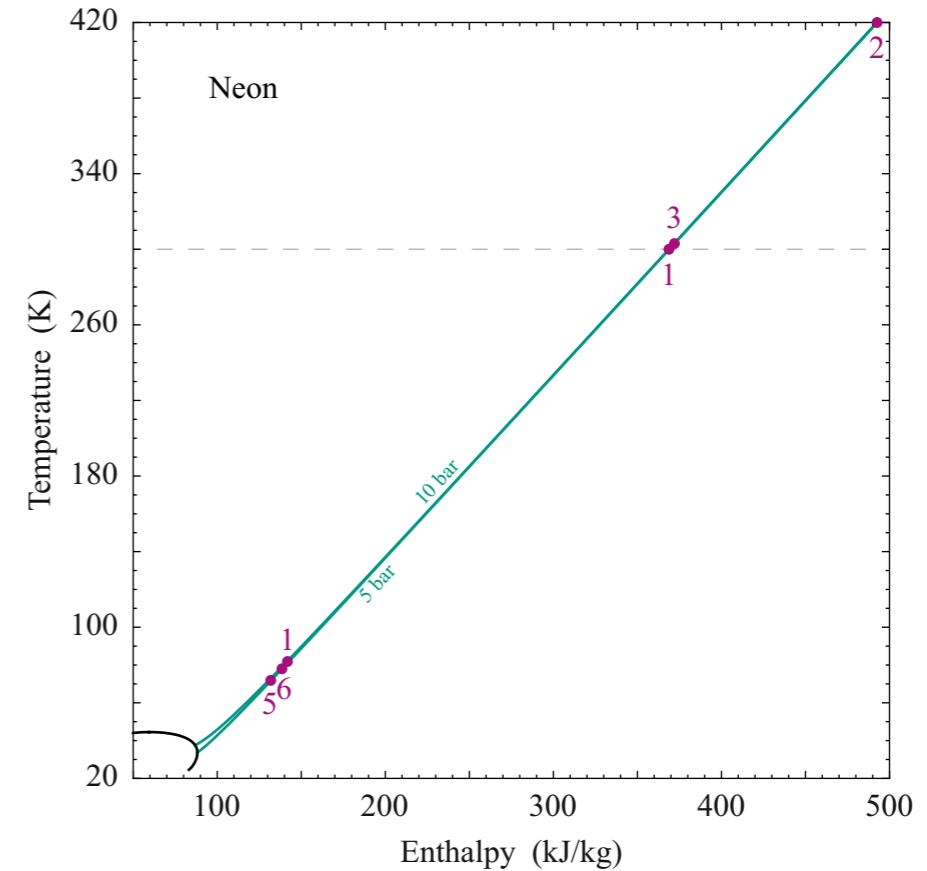


# Brayton cycle

## Layout



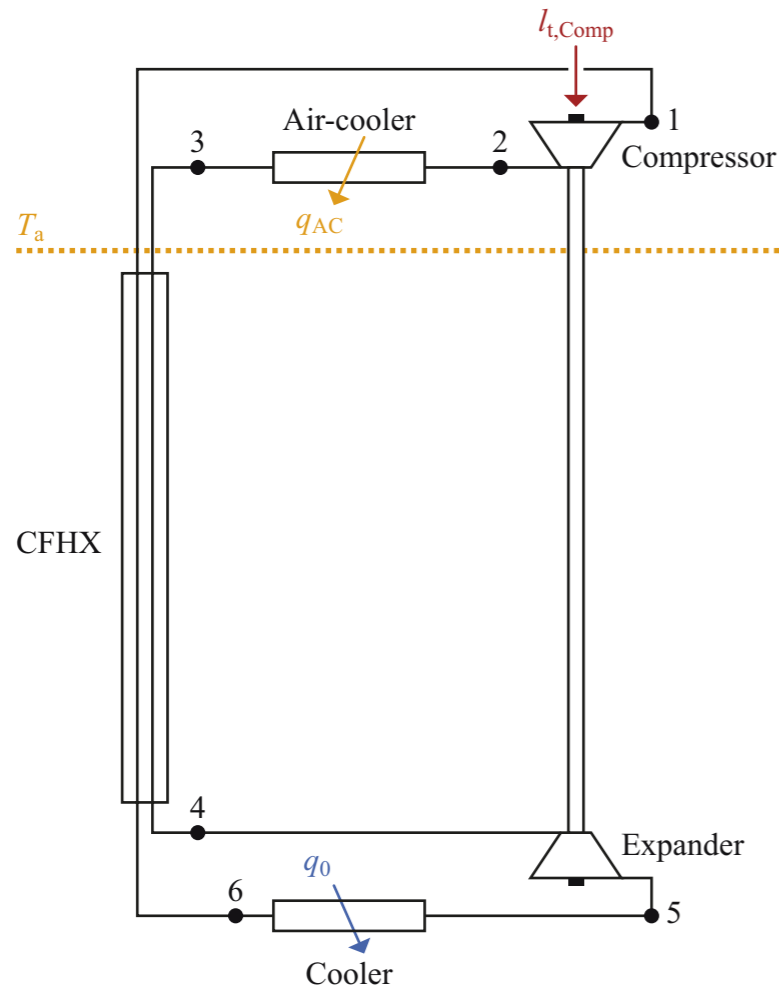
## Process in $T, h$ -diagram



- ▶ Expansion of *single-phase* working fluid in super-heated region (e.g. Ne)
- ▶ Recovery of expansion work

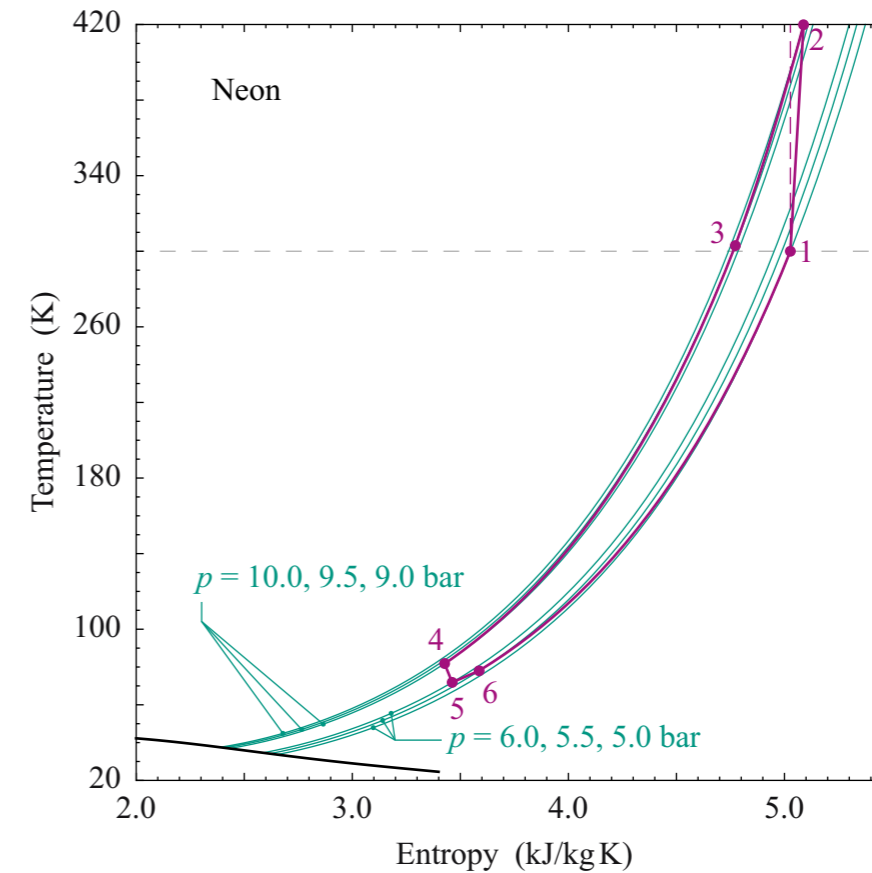
# Brayton cycle

## Layout



- ▶ Expansion of *single-phase* working fluid in super-heated region (e.g. Ne)
- ▶ Recovery of expansion work

## Process in $T, s$ -diagram



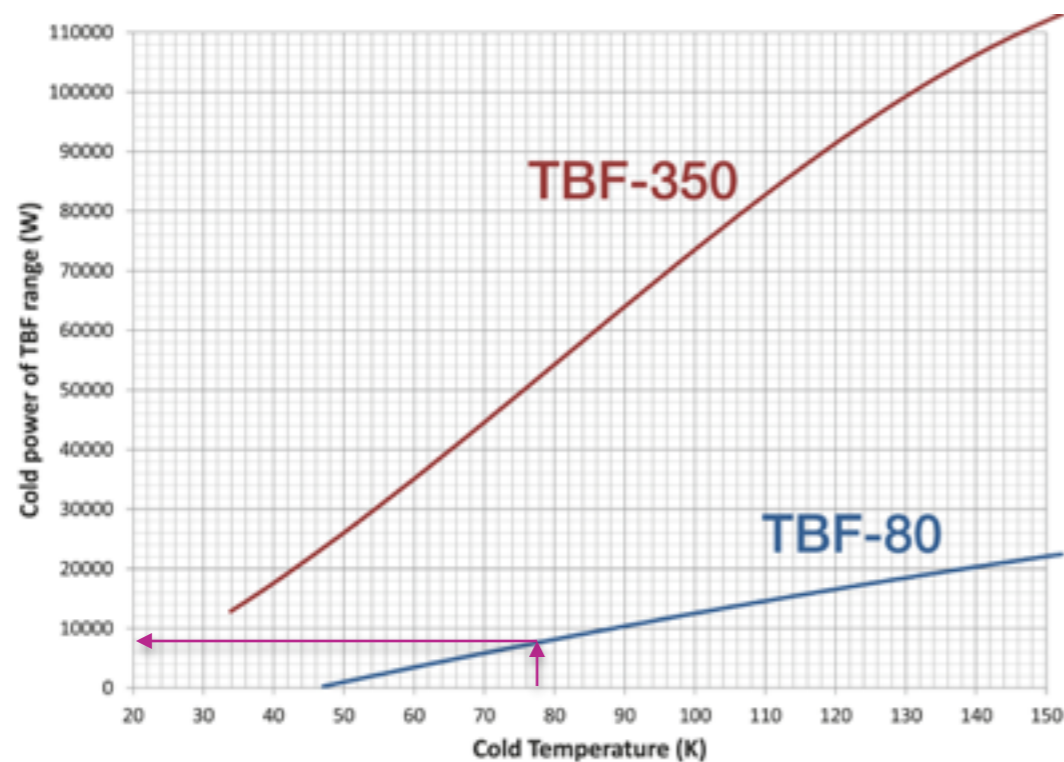
- ▶ Efficient due to turbo-expander, but cost factor of cold turbo-machinery
- ▶ Costly and rare working fluid (0.0018 Vol% in air)

# Turbo-Brayton refrigerator

- Commercial system available (Air Liquide)
- Applications
  - Gas (re)liquefaction (HCs, air gases)
  - Cryogenic gas purification/separation
  - ▶ Smallest unit suitable for HTS power applications

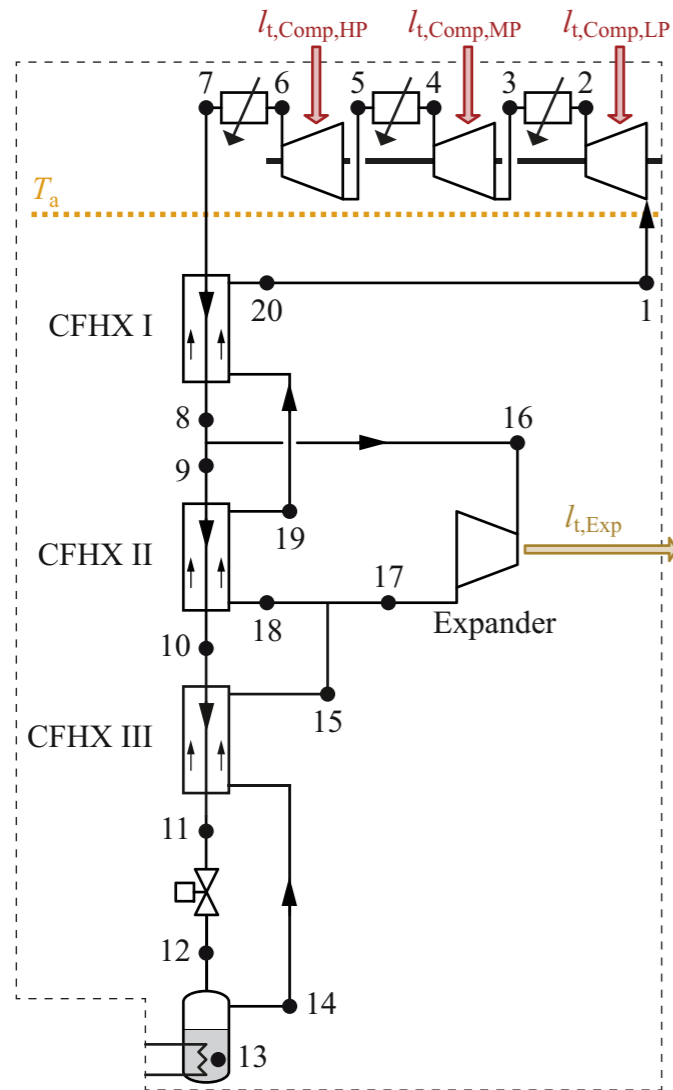


Source: <http://www.airliquideadvancedtechnologies.com/>



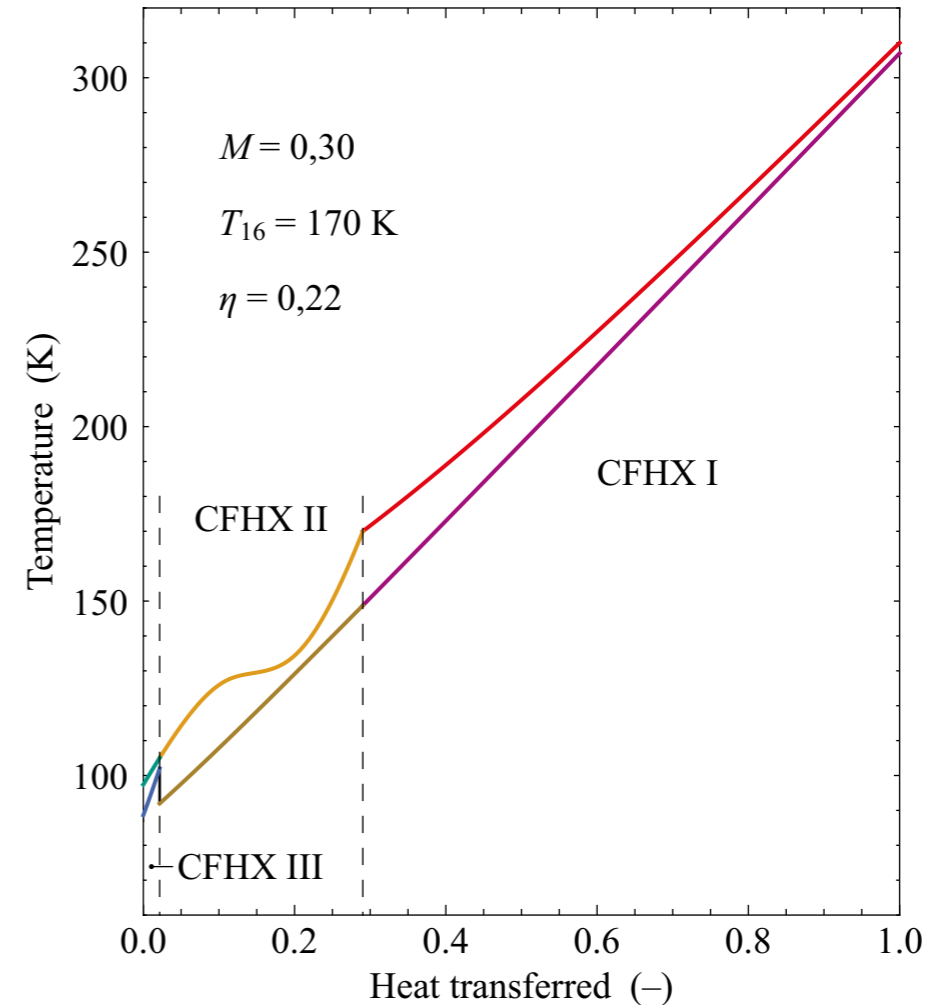
# Claude cycle

## Layout



- ▶ Single-phase expansion of *partial flow* in super-heated region
- ▶ Pre-cooling of remaining flow

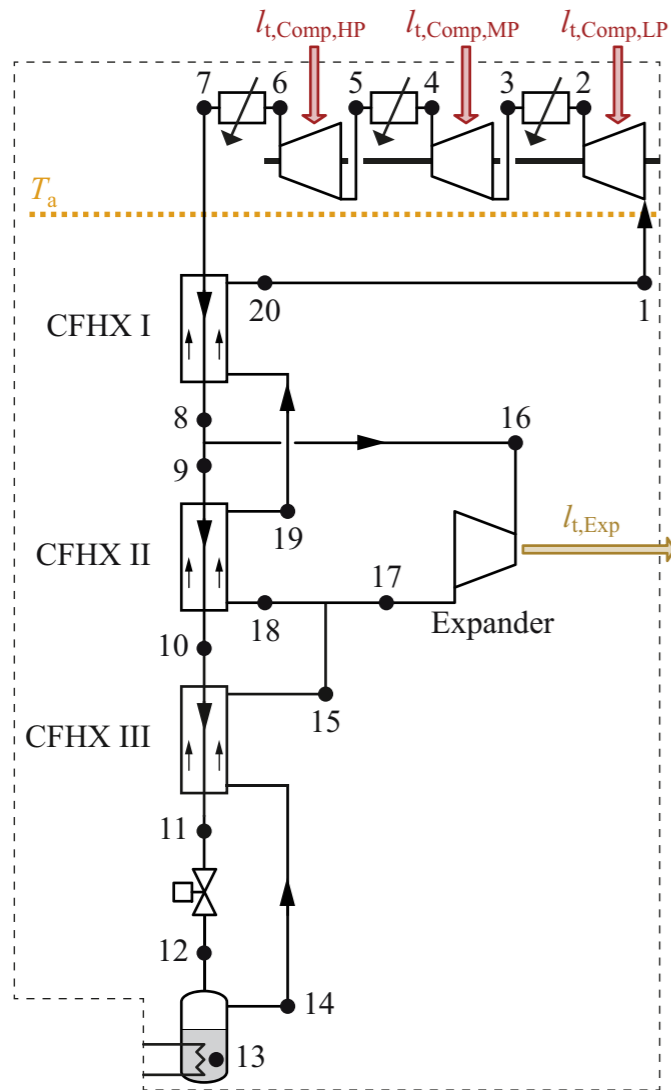
## Temperature profiles in CFHX I–III



- ▶ Efficient due to turbo-expander, but cost factor of cold turbo-machinery
- ▶ Pinch-points to be considered

# Claude cycle

## Layout

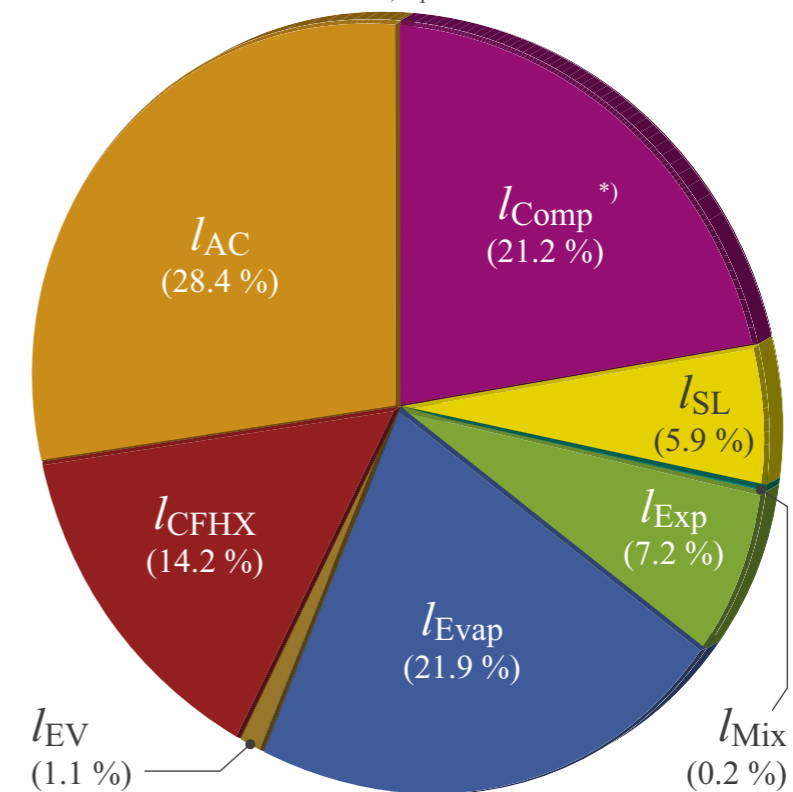


- ▶ Single-phase expansion of partial flow in superheated region
- ▶ Pre-cooling of remaining flow

## Sources of energy demand

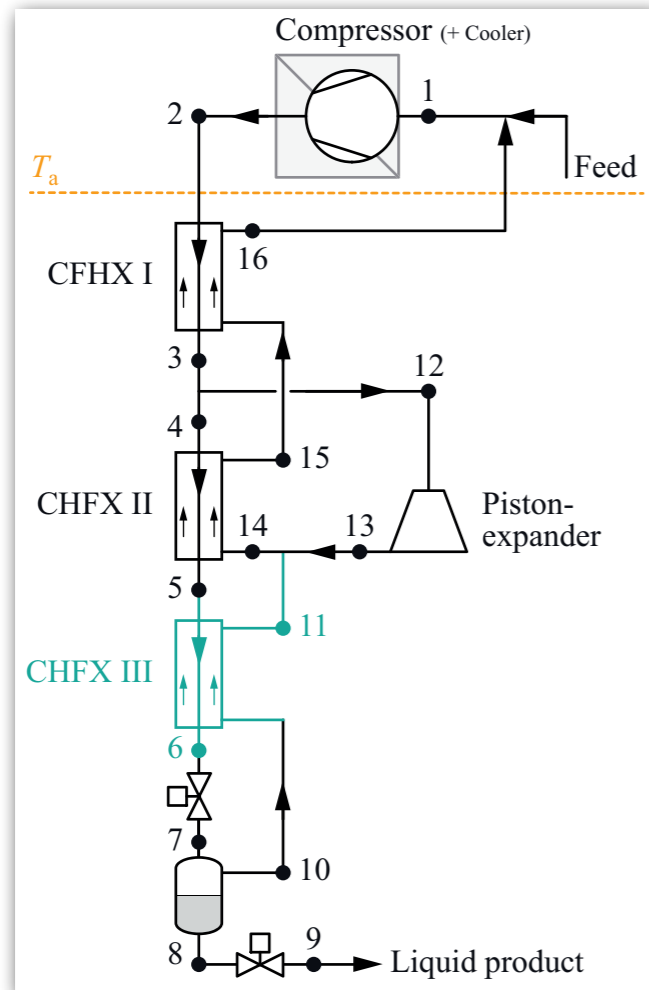
- Example process with  $N_2$ ,  $T_a = 300$  K,  $p_1 = 1$  bar,  $p_7 = 40$  bar,  $\Delta p_{ij} = 0.5$  bar,  $\eta_{Comp} = 0.7$ ,  $\eta_{Exp} = 0.8$ ,  $\Delta T_{min} = 3$  K in all heat exchangers
- Most efficient operating point at a cold flow  $M = 0.3$  und  $T_{16} = 170$  K (expander flow  $1-M$ )

\*) Expansion work  $l_{t,Exp} = 0.077 l_t$  recovered

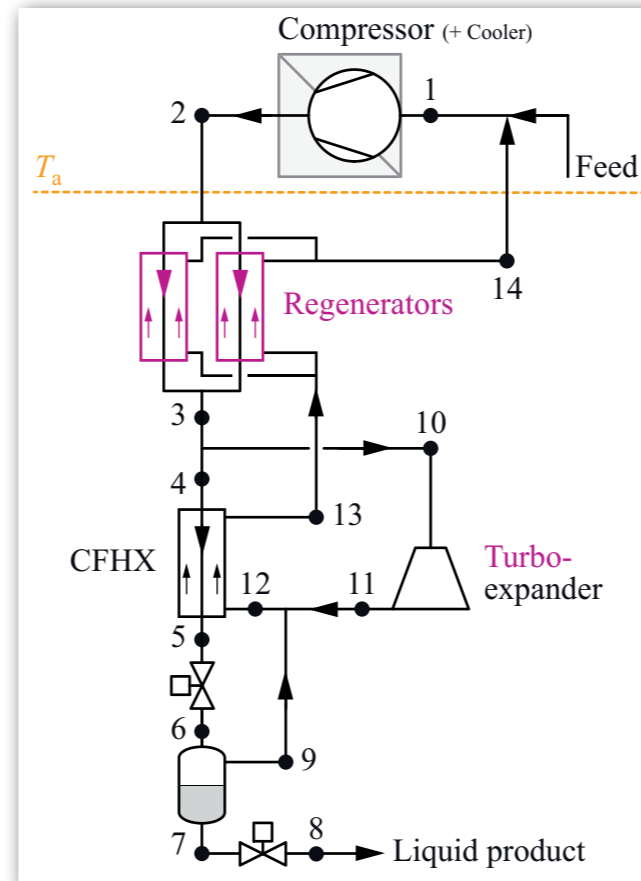


# Variants of the Claude cycle

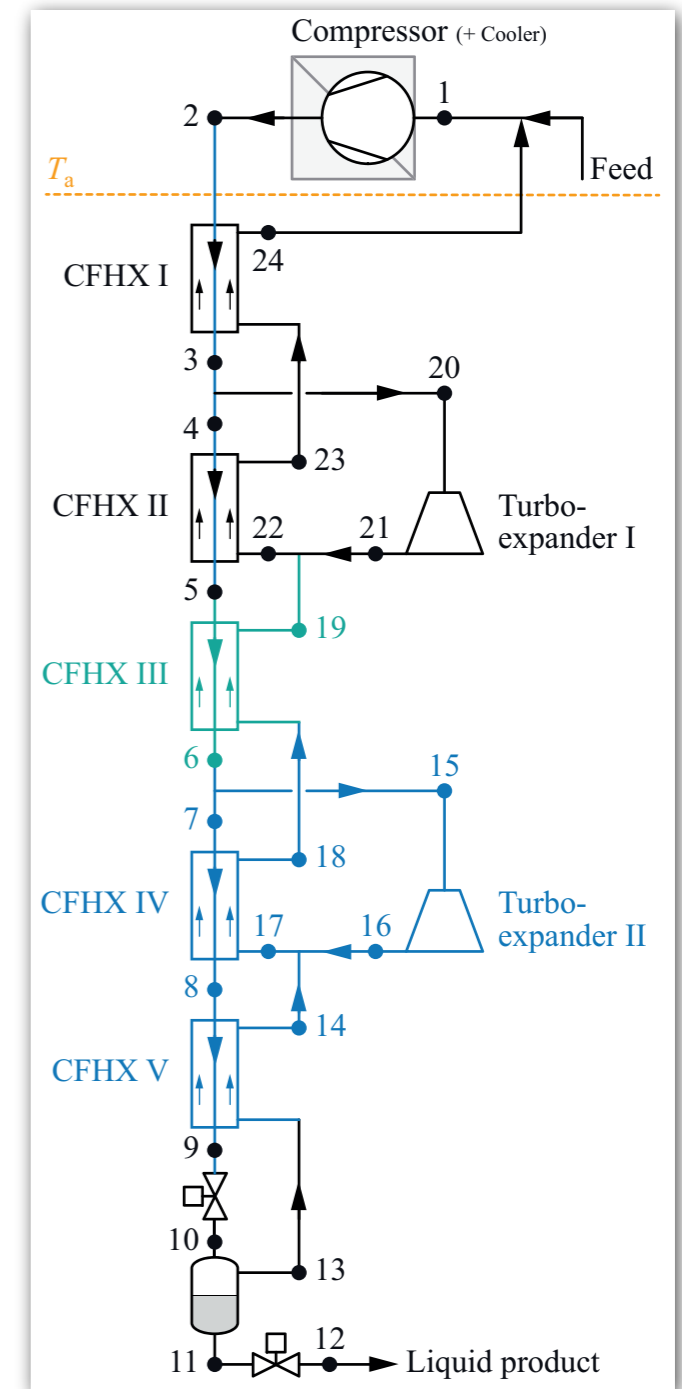
## ■ Claude cycle



## ■ Kapitza cycle



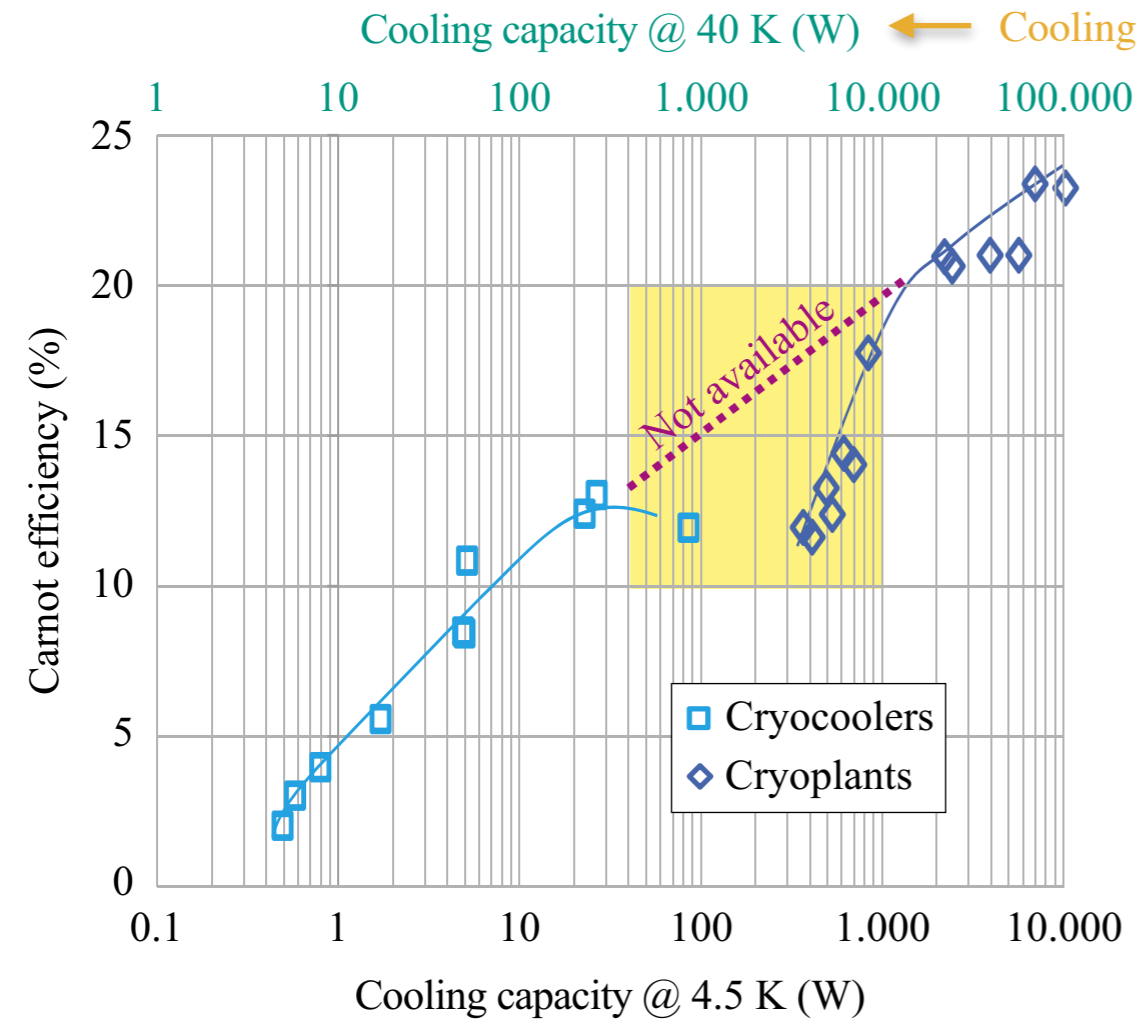
## ■ Collins cycle



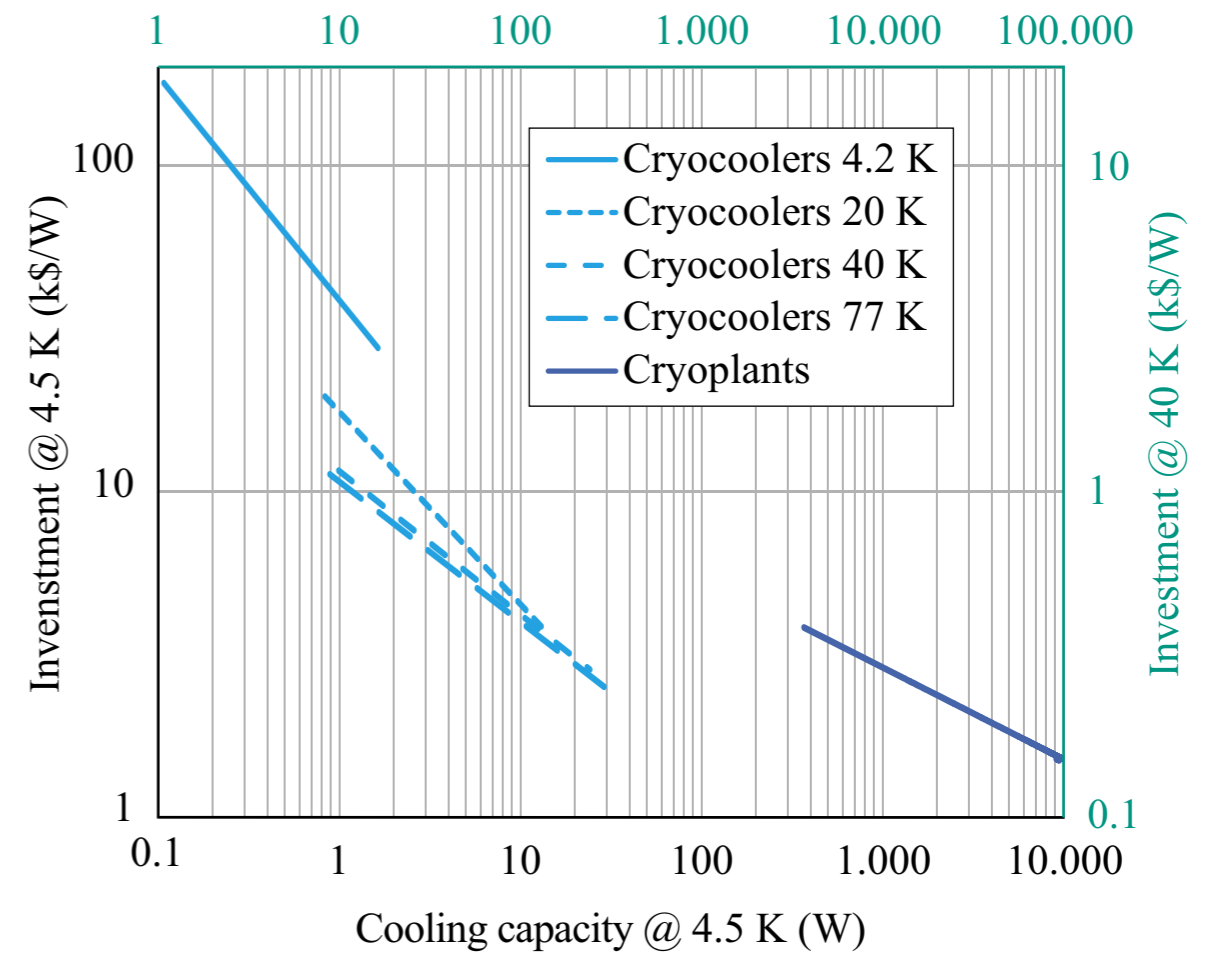
Cooling technology for HTS power applications  
**DEVELOPMENT POTENTIALS**

# Technology comparison

## Efficiency



## Specific investment cost



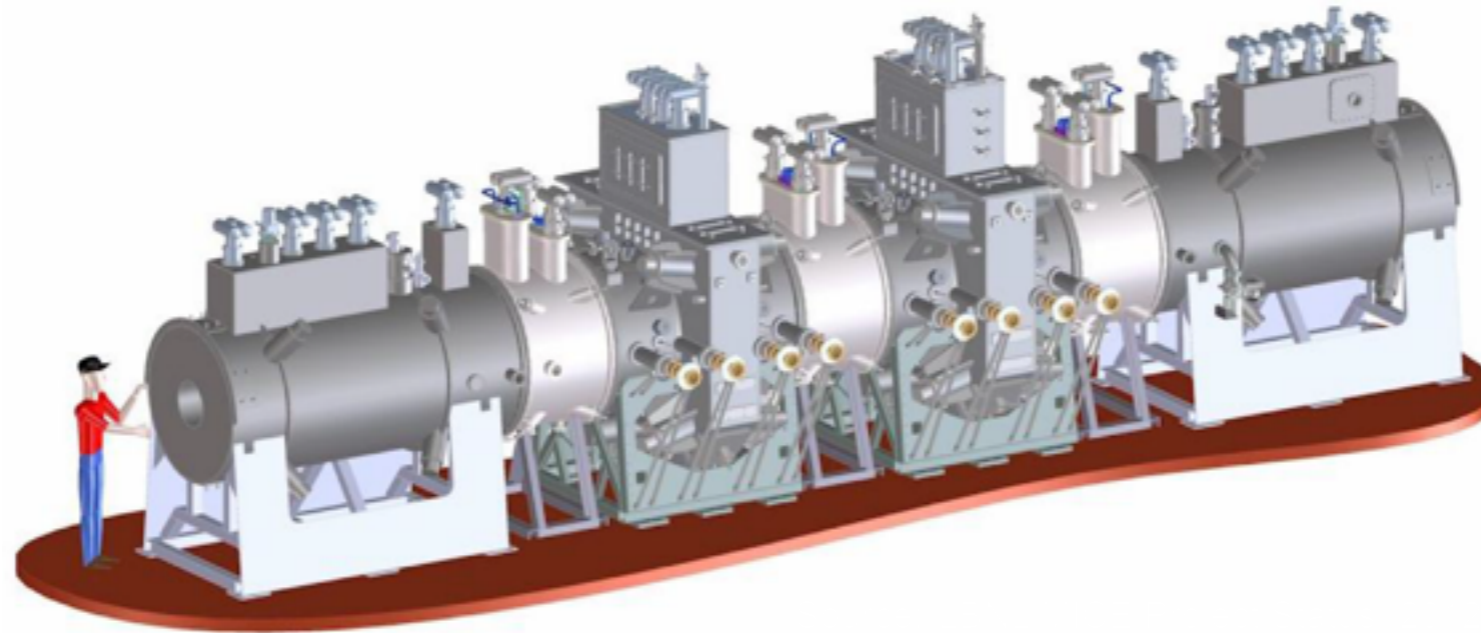
Source: Decker, L.: Overview on cryogenic refrigeration cycles for large scale HTS applications. International Workshop on Cooling System for HTS Applications (IWC-HTS), October 14-16, 2015, Matsue, Japan

► We can conclude that *either technology* has *limitations* in the HTS power application range



# Makeshift solution at 4 K

- Excessive cryocooler use on a superconducting magnet and RF cryostat



- **25 GM cryocooler on one cryostat**

- ▶ Cooling power:

$$\dot{Q}_0 = 37.5 \text{ W @ } \underline{4.2 \text{ K}} \rightarrow (25 \times 1.5 \text{ W @ } 4.2 \text{ K})$$

- ▶ Power consumption:

$$P = 180 \text{ kW } (25 \times 7.2 \text{ kW})$$

- ▶ Specific power consumption:

$$P^* = 4800 \text{ W/W}_{\text{Cooling power}}$$

- ▶ Other issues:

Space, vibrations, noise, reliability, maintenance, ...

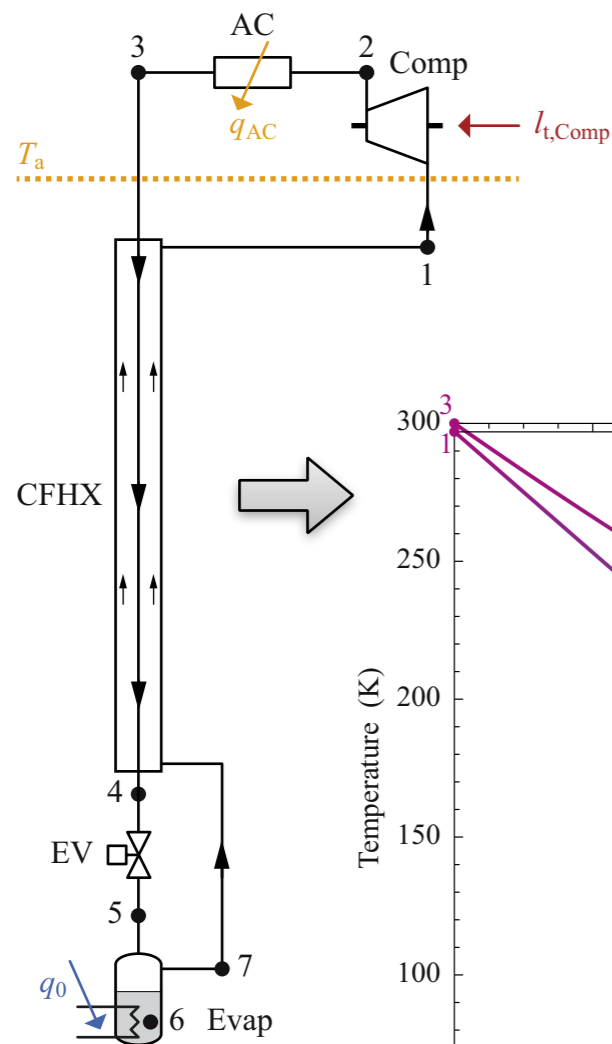
- Comparison to recuperative cycle (Collins process)

- ▶ Specific power consumption:

$$P^* = 250 \dots 400 \text{ W/W}_{\text{Cooling power}}$$

# Back to the Linde-Hampson cycle

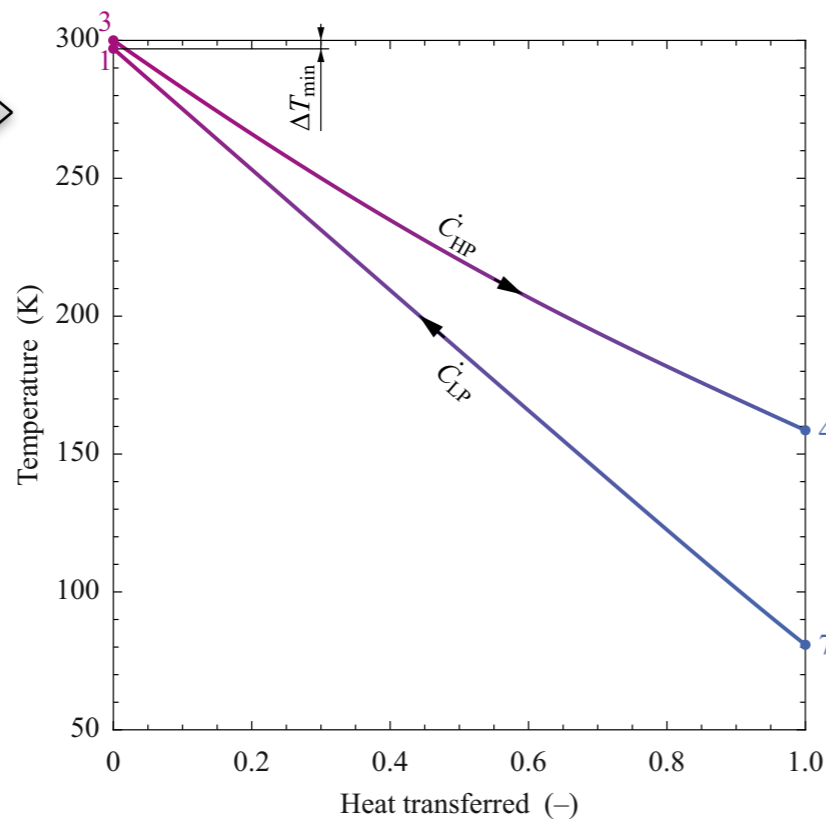
## Reason for poor efficiency



► Different capacity flows  $\dot{C} = \dot{M} \cdot c_p(T, p)$  yield increasing  $\Delta T$

► Large entropy production  $\Delta s_{\text{irr}} = \int \frac{T_h - T_c}{T_h \cdot T_c} dq$

↑ 1<sup>st</sup> Law  
└─┘ 2<sup>nd</sup> Law



## Technical options

► Aim:  $\dot{M}_{\text{HP}} \cdot c_{p,\text{HP}} = \dot{M}_{\text{LP}} \cdot c_{p,\text{LP}}$  in order to keep  $\Delta T_{\text{min}}$  along the CFHX

I) Adaptation of flow rates  $\dot{M}$

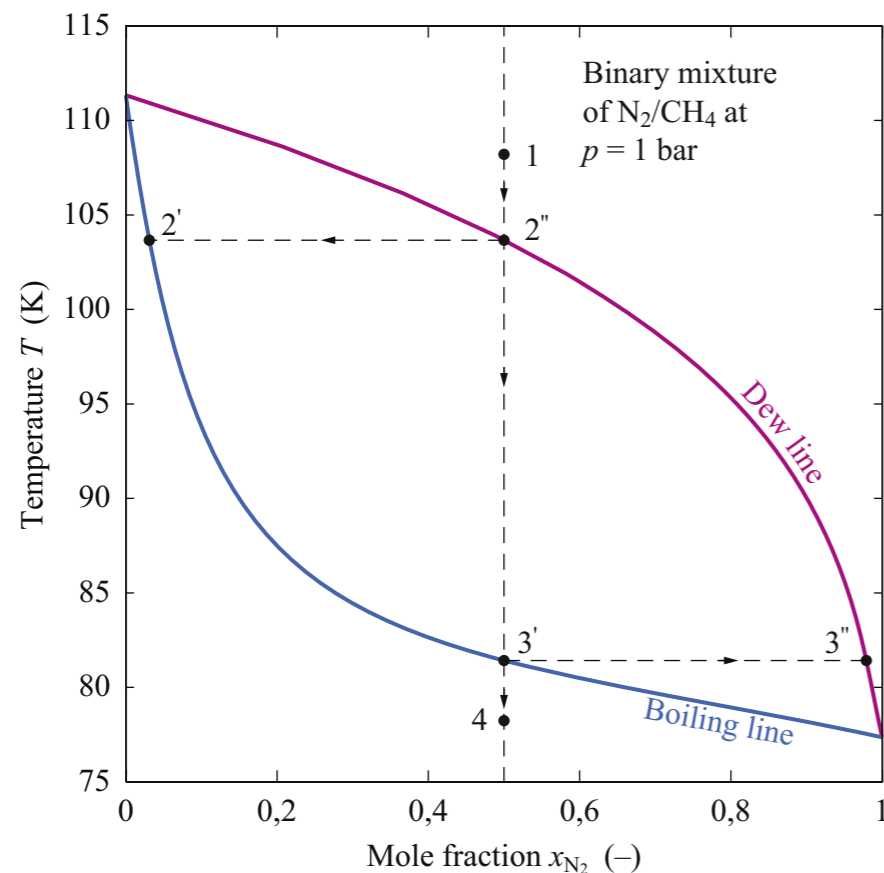
► The Claude cycle includes this solution in CFHX II

II) Manipulation of spec. heat capacities by using *wide-boiling refrigerant mixtures*

► Mixed refrigerant cycle (MRC)

# Phase behavior of zeotrope mixtures

## Phase diagram (example)



► Different concentrations of saturated liquid and vapor phases

## Closed condensation 1 → 4

(1) Single-phase super-heated vapor

(2) Saturated vapor 2''; first liquid drop 2'

► High-boiler CH<sub>4</sub> has lower vapor pressure and condenses first

(3) Saturated liquid 3'; last vapor bubble 3''

(4) Single-phase sub-cooled liquid

## Basics (ideal mixtures)

► Dalton's law  $p_i = y_i \cdot p$

► Raul's law  $p_i'' = x_i \cdot p_{\text{sat},i}$

► Equilibrium  $p_i = p_i''$

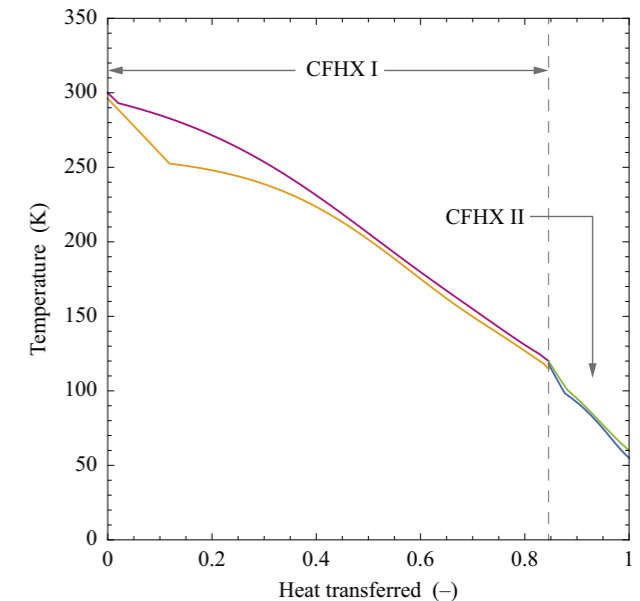
● Equivalent for heating/boiling from 4 → 1

► Low-boiler N<sub>2</sub> has higher vapor pressure and evaporates first

# Effects of using refrigerant mixtures

## ■ Partial condensation and evaporation (HP and LP sides), respectively, of mixture components along the CFHX

- High-boiling components condense/evaporate at the warm end
- Low-boiling components condense/evaporate at the cold end



## ■ Reduced entropy production $\Delta s_{\text{irr}}$ during heat transfer

- Manipulation of capacity flows  $\dot{C} = \dot{M} \cdot c_p(T, p)$  through the specific heat capacities of the condensing/evaporating components  $i$  (phase change implies  $c_{p,i} \rightarrow \infty$ )
- ▶ Minimization of temperature gradients in the CFHX

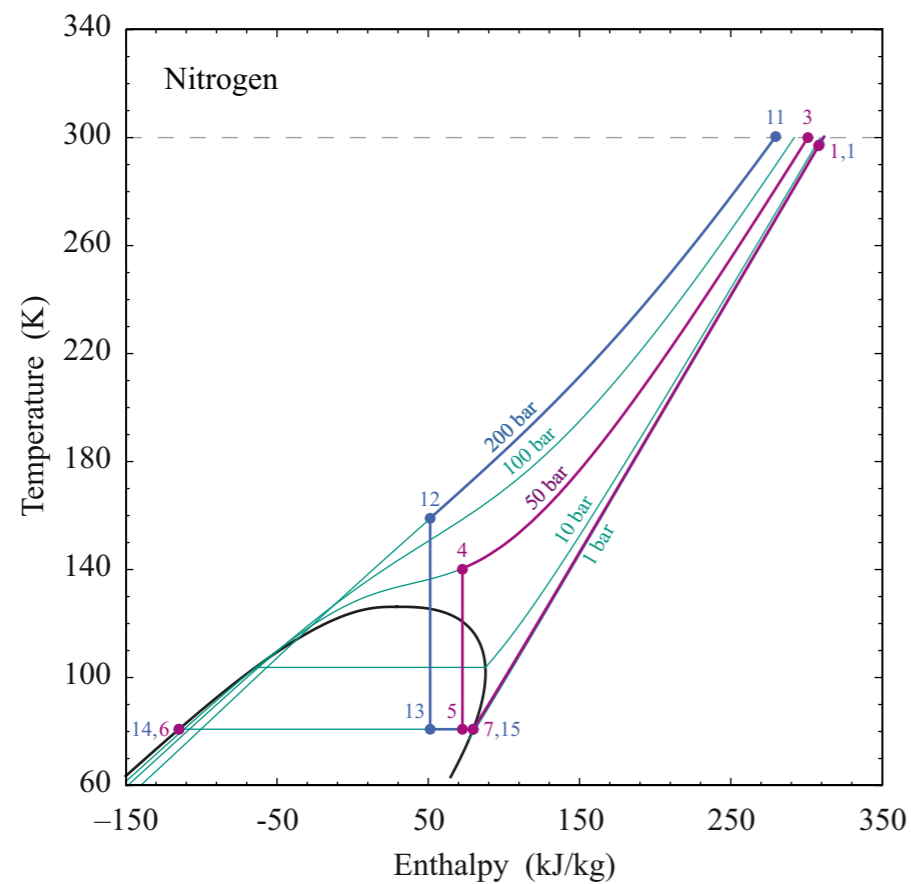
## ■ Reduced entropy production $\Delta s_{\text{irr}}$ during throttling

- Higher Joule-Thomson coefficients  $\mu_{\text{JT}} = \left(\frac{\partial T}{\partial p}\right)_h$  at lower temperature
- ▶ Lower  $\Delta p$  required for the same  $\Delta T$

# Throttling in the Linde-Hampson cycle

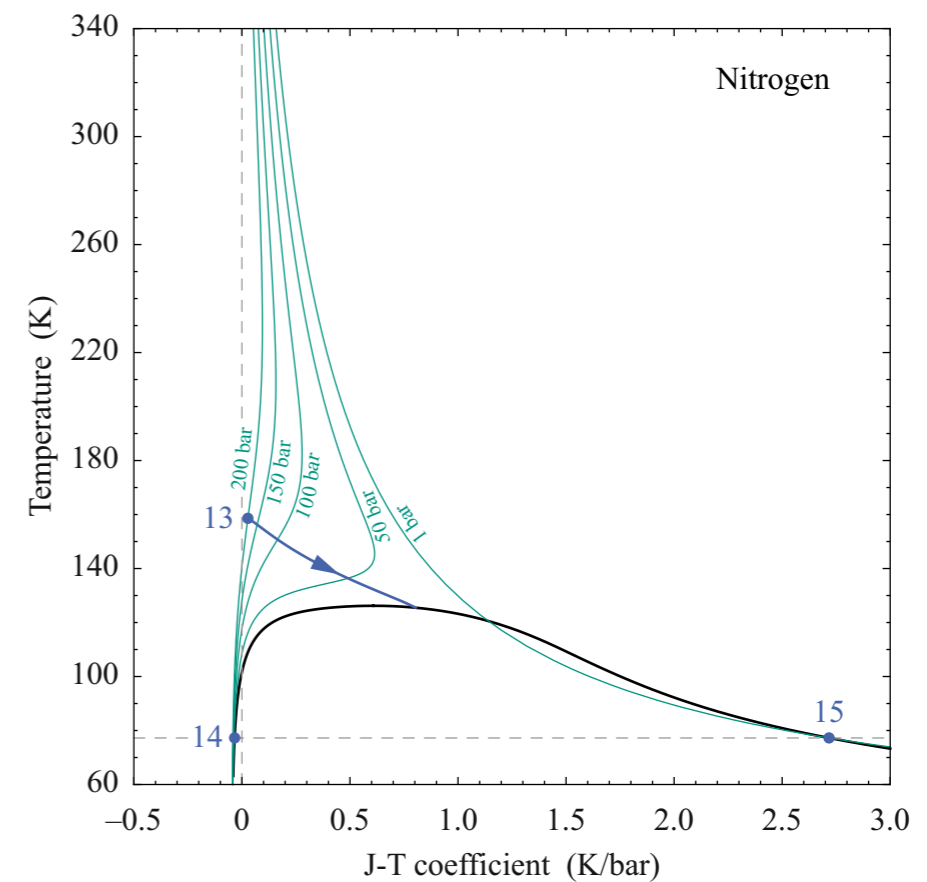
## Example process

- Higher specific cooling capacity at higher pressure
- ▶ Necessity of c. 200 bar in case of N<sub>2</sub>



## Joule-Thomson coefficient of N<sub>2</sub>

- $\mu_{JT} \uparrow$  bei  $T \downarrow$ 
  - ▶ Better pre-cooling ( $T_{12}$ ) with mixtures
- $\mu_{JT} \uparrow$  bei  $p \downarrow$ 
  - ▶ Lower pressure needed as well



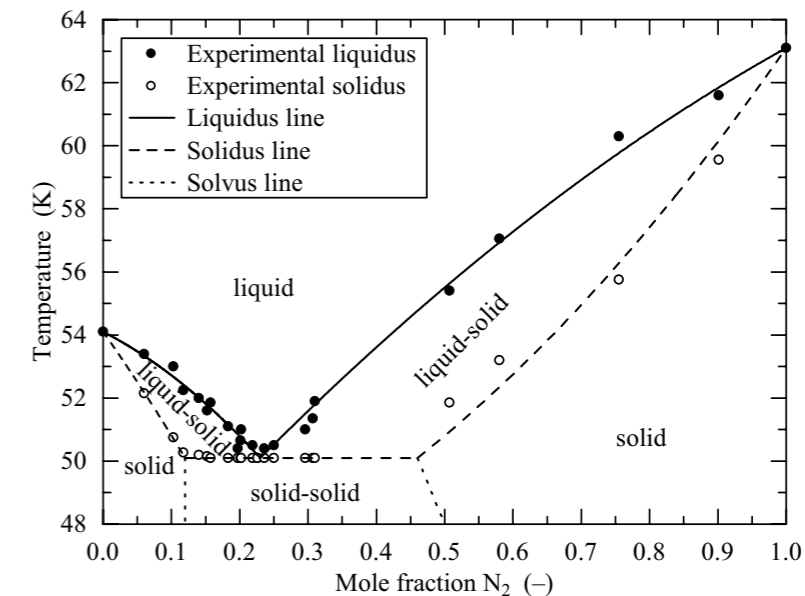
# Low-temperature limits of MRCs

## State-of-the-art

- MRCs are *widely used* in natural gas liquefaction since the 1970s
  - ▶ C3MR process (C3 pre-cooling)
  - ▶ DMR process (double MR process)
  - ▶ MFC process (mixed fluid cascade)
- Typical use of hydrocarbon mixtures (C<sub>1</sub>...C<sub>5</sub>) at  $T > 100$  K
  - ▶  $T_{nb,CH_4} = 112$  K
- Low-temperature limits
  - Achievable temperature influenced by lowest boiling component
  - Some reduction of boiling temperatures  $T_b(p)$  by addition of inert gases (N<sub>2</sub>, ...)
  - ▶  $\underbrace{y_i \cdot p}_{p_i} = x_i \cdot p_{sat,i}$ , i.e.  $T_b = f(p_i)$
  - Low-temperature limit determined by *freezing* of high-boilers and oil

## Theoretical limit for HTS application

- Classical limit of vapor compression cycles given by the *triple point* of N<sub>2</sub>
  - ▶  $T_{tr,N_2} = 63$  K
- N<sub>2</sub>/O<sub>2</sub> binary mixtures, however, have freezing points as low as ~50 K
  - ▶ Eutectic point at 50.1 K and 77 mole-% O<sub>2</sub>

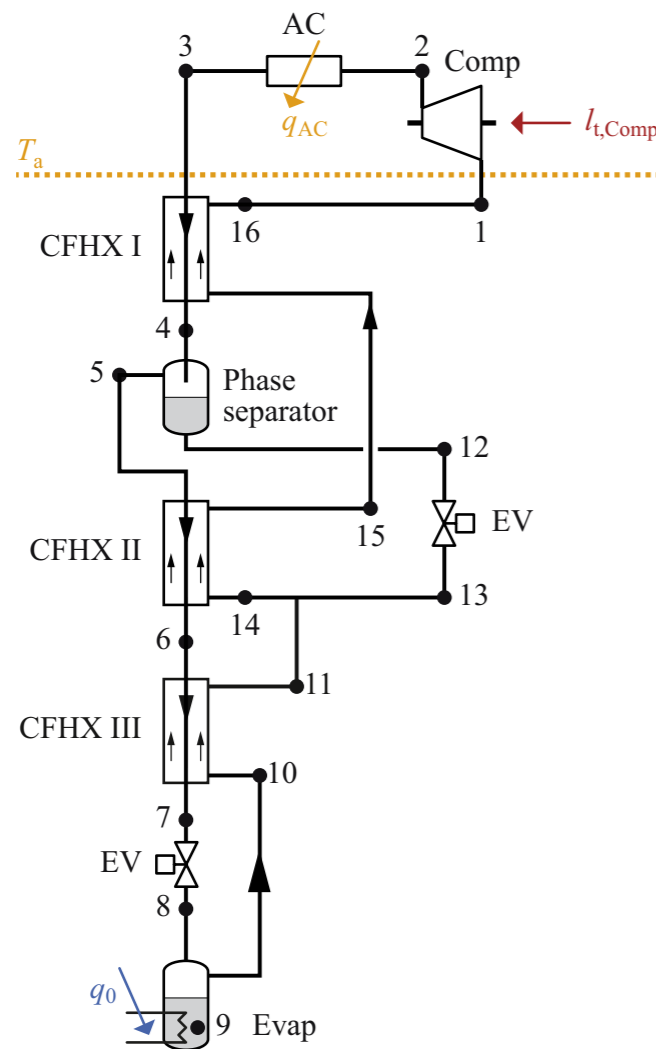


Data source: Ruhemann, M. et al.: Zustandsdiagramme niedrig schmelzender Gemische; II. Das Schmelzdiagramm Sauerstoff-Stickstoff und das Zustandsdiagramm Stickstoff-Kohlenoxyd. Phys. Z. Sowjetunion 1935, 8, 326.

- Data for ternary Ne/N<sub>2</sub>/O<sub>2</sub> mixtures not yet available (→ [studies at KIT](#))

# Prevention of freeze-out

## Kleemenko cycle



## Principle

- Similarity with the Claude cycle, but *phase separator* and *expansion valve* instead of the expander
- Separation of high-boilers (e.g. C<sub>5</sub>, C<sub>4</sub>) as well as oil at ~250...270 K

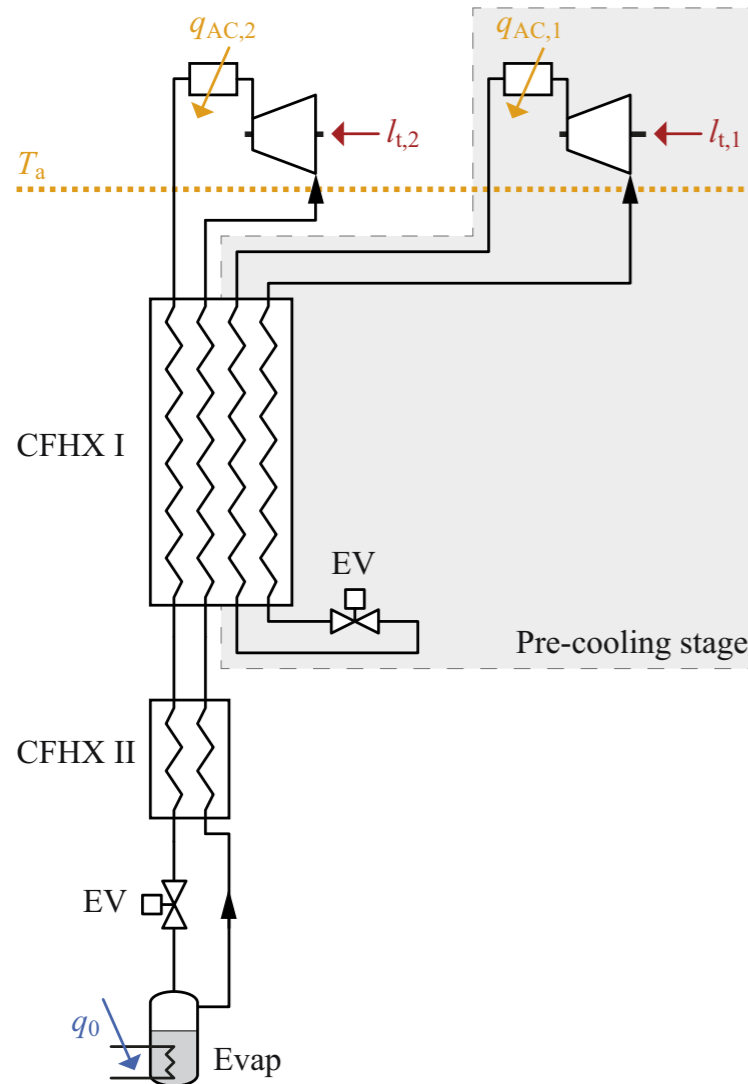
- + Concentration-dependent freezing point
- No complete separation

## Various process configurations

- Several phase separators, especially for low-temperature operation
- Complexity, ...

# Cryogenic mixed refrigerant cascade (CMRC)

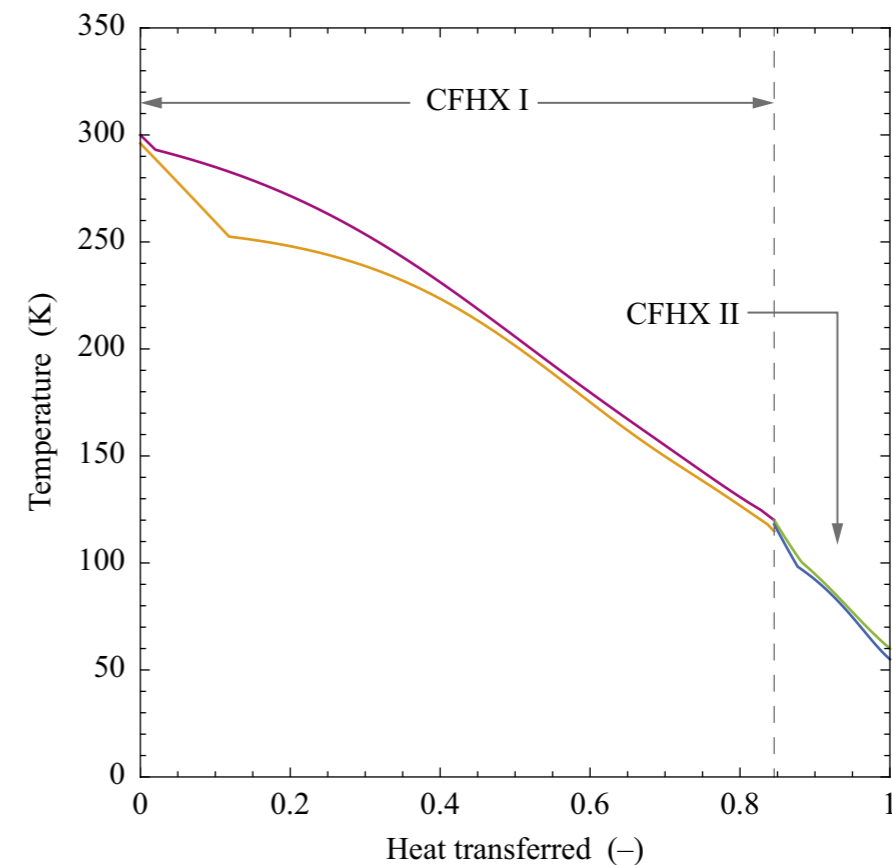
## Layout



## Principle

- Separate pre-cooling and second stage
- + No high-boilers and no risk of freeze-out in the second stage

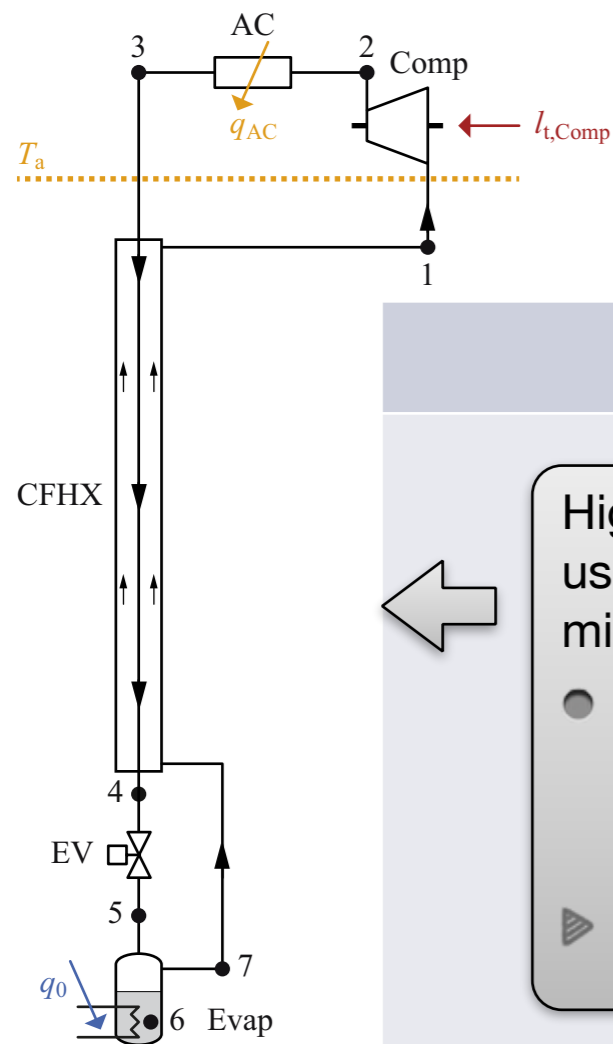
## Example temperature profiles





# Approaches for cooling system improvements

## Linde-Hampson process



### Options for efficiency increase

Higher efficiency by using refrigerant mixtures (MRC)

● „Molecular Engineering“

▶ Negligible cost

Higher efficiency by increased system complexity

● Cold expander

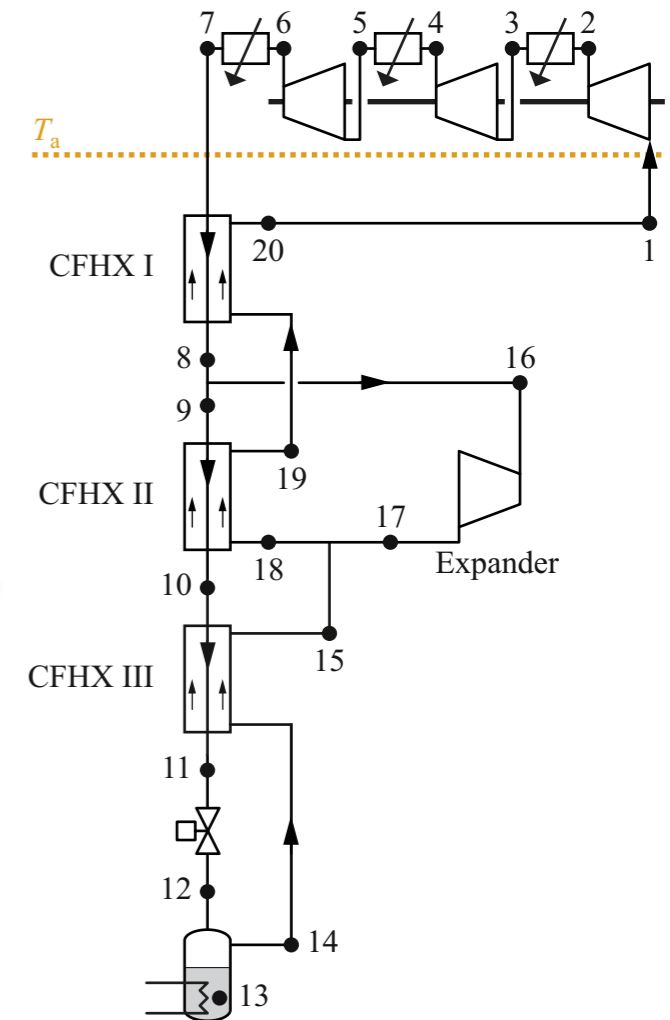
● Number of HX ↑

● Control ↑

■ Investment cost ↑

Use of standard refrigeration technology

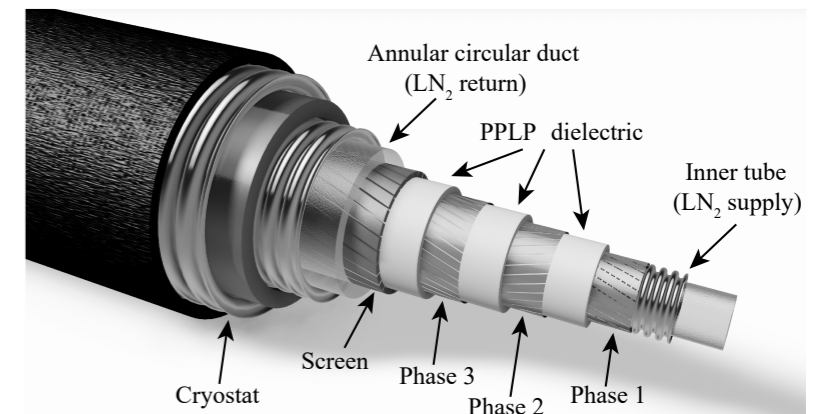
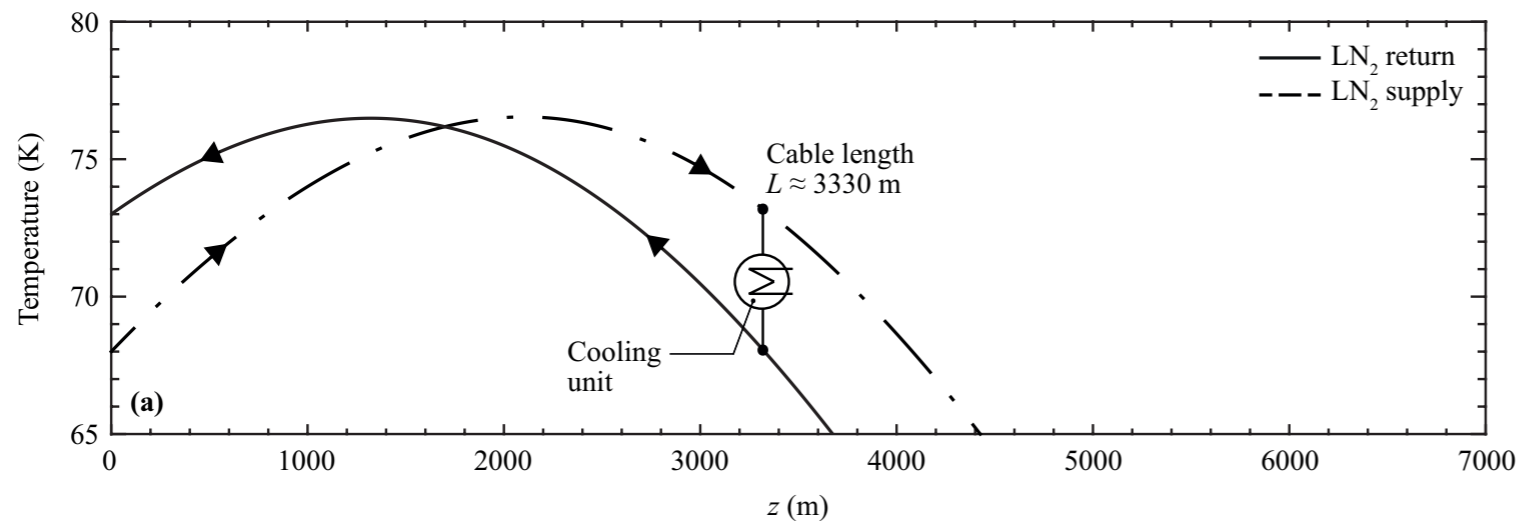
## Claude process



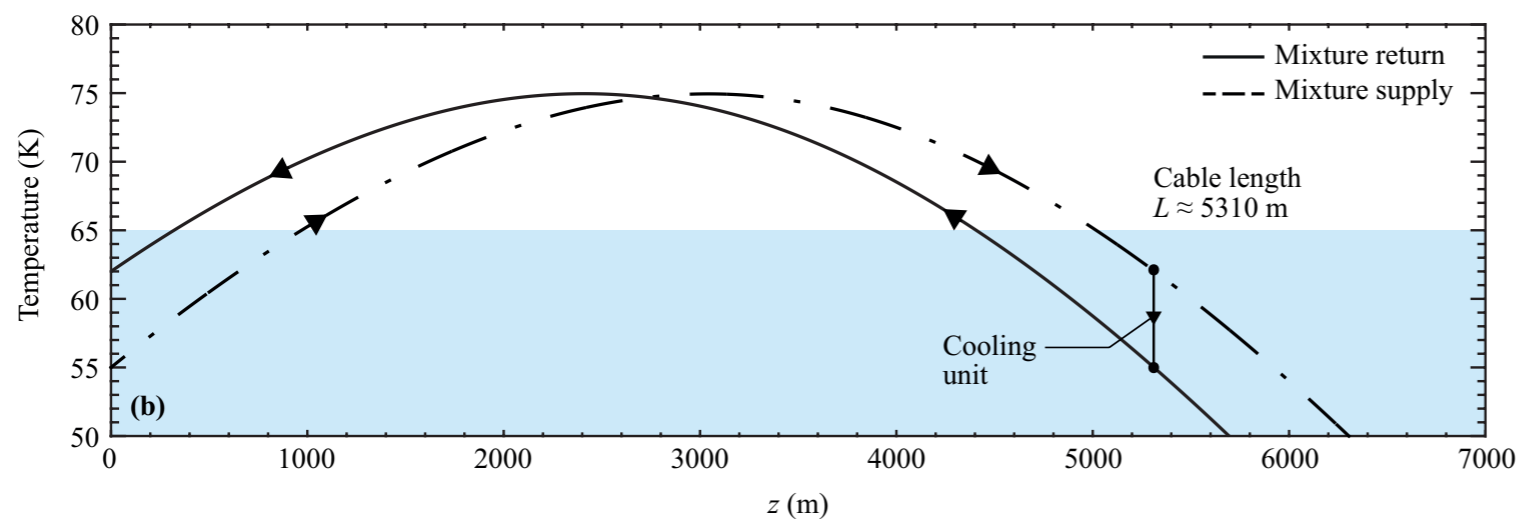
# Potential HTS mixture applications

## ■ Use of wide-boiling refrigerant mixtures in closed cycle cooling units

- Longer power cable lengths by re-cooling units at the far cable end



## ■ Potential use of O<sub>2</sub>/N<sub>2</sub> mixtures as cooling liquid in HTS power cables



Source: Shabagin, E.; Heidt, C.; Strauß, S.; Grohmann, S.: Three-dimensional modelling of temperature and pressure profiles in concentric three-phase HTS power cables. *Cryogenics* 2016 (to be published).

- **Open issues:** Material compatibility, safety, ...

Cooling technology for HTS power applications  
**CONCLUSIONS**

# Conclusions

