

Mass transfer and catalytic reaction in Taylor flow: parametric numerical study for frozen hydrodynamics

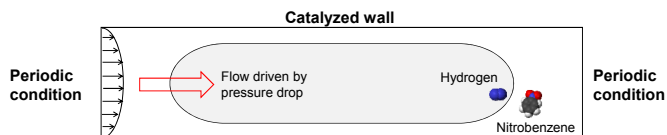
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Objectives

- Goal: Simulation of catalyzed hydrogenation of nitrobenzene to aniline within a Taylor flow in monolith reactor
- Methodology
 - Coupling two computer codes: TURBIT-VOF^[1] for the gas-liquid flows and DETCHEM^[2] for the reaction kinetics
 - Moving reference frame approach for mass transfer
 - Langmuir-Hinshelwood type global reaction kinetics^[3]

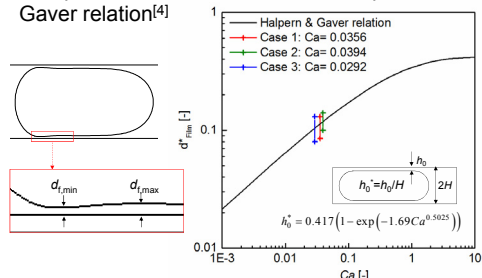


- Test conditions
 - Temperature: 298K, pressure: 7bar, channel height: 100 μ m
 - Reynolds number: 80~110, capillary number: 0.03~0.04

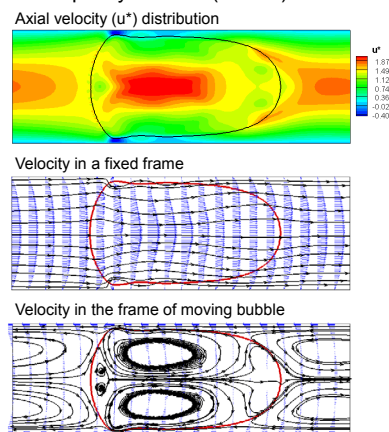
Numerical approaches

Hydrodynamics

- Two-dimensional incompressible Navier-Stokes equation with Volume-of-fluid (VOF) method
- Isothermal simulation
- Constant physical properties during calculation
- Fine mesh resolution near the boundary wall
- Comparison of film thickness with Halpern & Gaver relation^[4]

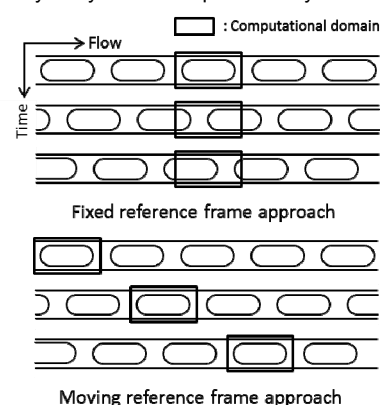


Velocity fields for different Reynolds and capillary number (case 1)



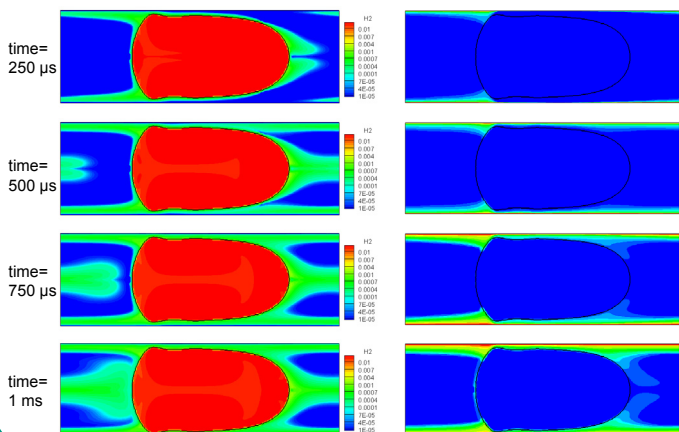
Mass transfer in the moving frame

- Solve concentration equations with frozen hydrodynamics at quasi steady state

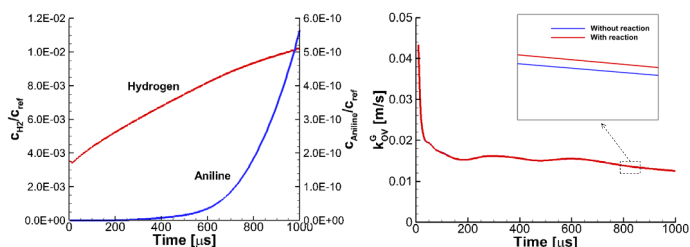


Results of reactive mass transfer in the moving frame

Temporal distributions of hydrogen and aniline



Mean concentrations and mass transfer coefficient



- Hydrogen mass transfer is largest at the rear part of the bubble where the liquid film thickness is thinnest
- The aniline is mainly produced in rear part as well
- The mean concentration of hydrogen in liquid phase is increasing by mass transfer from the gas phase and that of aniline is also increasing due to the reaction
- The enhancement of mass transfer coefficient by reaction is small

Conclusions

- The gas-liquid Taylor flow of nitrobenzene and hydrogen has been successfully computed with real viscosities and diffusivities
- Moving frame approach allows to save the computational time without loss of physical information
- The estimation of film region is most important to investigate the catalytic reaction within the Taylor flow

References

[1] Onea et al., Chem. Eng. Sci., 64 (2009) 1416-1435
[4] Halpern et al., J Comput Phys, 115(2) (1994) 366-375

[2] Deutschmann et al., DETCHEM™ User Manual, 2012, <http://www.detchem.com>

[3] Höller et al., Chem. Eng. Technol. 23 (2000) 3, 251-255