

Karlsruhe Institute of Technology



Mechanical Characterization of Electrochemically based W – Cu Joints for low Temperature **Heat Sink Application**

Wolfgang Krauss, Julia Lorenz, Jürgen Konys, Ermile Gaganidze KIT, Hermann-von-Helmholtz-Platz 1, 76344 Eggenstein-Leopoldshafen, Germany

Motivation

Joining of the armor material tungsten to copper components, which will act as low temperature heat sinks in divertor application, is a challenging process and showed lacks due to their restricted miscibility and alloyability.

Electroplating has the feature to generate layers acting as active interlayers which can overcome such lacks. The interlayers must have the ability to react with both alloys W or Cu.

Application of electroplating technology for joining low temperature heat sinks

View of alloy sequence



Divertor

Mechanical behavior and characterization of joints

Behavior of joints bonded at 700°C vs bonding time





Pressing load: 7 Mpa

Crack in Pd-Cu

This work was performed to demonstrate that joints can be processed applying the electrochemical plating tool and bonding.

The joints were qualified by thermomechanical testing for the demonstration of future applicability in fusion development.



Joints processed by electroplating

Tools and plating development for adapted joints

Layout of sample processing for brazing of components

Pressing load

Cu part

W part

Interlayer

F

Anvil

The development of joining technology by electroplating consists of the two main parts :

- Electroplating **
- Joining by diffusion bonding *

Pd was selected as interlayer metal to improve reactions. Electroplating from aqueous electrolytes was performed similarly to Cu or Ni plating with a current density of 8 mA/cm² and pH value of 7 to 8.

Surface conditioning and electroplating of interlayers









Bonded at 700°C, 350 N



[MPa]

Shea



reaction zone

Shear testing of W - Cu joints

120 80 40 0 120 240 30 60 Bonding time t [min]

load [kN]

Shear

Bonded at 700°C

4 h, 350 N D 17

Bonded at 600°C

| Sample | Time [h] | Load [N] | Strength [MPa] |
|--------|----------|----------|----------------|
| D 19 | 4 | 350 | 40 |
| D 22 | 8 | 350 | 51 |
| D 23 | 4 | 500 | 47 |
| D 20 | 8 | 500 | 63 |

| Sample | Time t [h] | Shear strength [MPa] | Bonding force F [N] | Temp. T [°C] |
|--------|---------------|----------------------------|---------------------------|-----------------|
| D 10 | 1 | 1 | 500 N | 600°C |
| D 15 | 4 | 40 | 350 N | 600°C |
| D 02 | 0.5 | 1 | 100 N | 700°C |
| D 12 | 0.5 | 50 | 350 N | 700°C |

General behavior

• Strength depends on T, t, F

- No spalling of Pd from W
- Shear strength comparable with conventional brazing values near 200 MPa



4 h, 350 N, 700°C

300

www.kit.edu

200

Position X [µm

Behavior of W - CuCrZr joints

Mechanical characterization of joints

Shear testing of diffusion bonded samples





Bonded at 700°C, 4 h



Conclusions

-100

The investigations showed that electroplating is an effective technology in the field of the joining of fusion relevant materials and components.

Reactive Pd layers were successfully deposited on W, Cu and CuCrZr by electroplating to obtain improved bonding

200

-200

Pd reacted with both parts W and Cu / CuCrZr to be joined as expected



Testing conditions:

Sample size: Diameter 8 mm

Temperature range: RT

Atmosphere: Air

During heating: Preloading of 0.03 kN

0.01 mm/s **Displacement rate:**

- Under industrial view Pd plating on Cu alloy instead on W is recommended
- Mechanical testing was performed at RT with shear strength of around 100 MPa
- Shear strength increased with bonding time
- W CuCrZr joints showed lower strength compared to W –Cu joints
- The strength of CuCrZr may initiate microcracks due to expansion mismatch
- An additional layer of pure Cu between Pd and CuCrZr may be recommended

Acknowledgment



This work has been partly carried out within the framework of the EUROfusion Consortium and has received funding from the Euratom research and training programme 2014-2018 under grant agreement No 633053. The views and opinions expressed herein do not necessarily reflect those of the European Commission.

KIT – The Research University in the Helmholtz Association