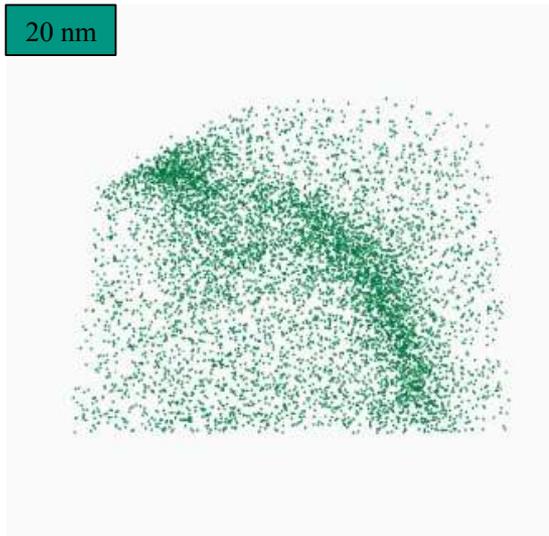


Outward diffusion through protective Alumina on NiAl-alloys

Torben Boll¹, Olof Bäcké², Krystyna Stiller²

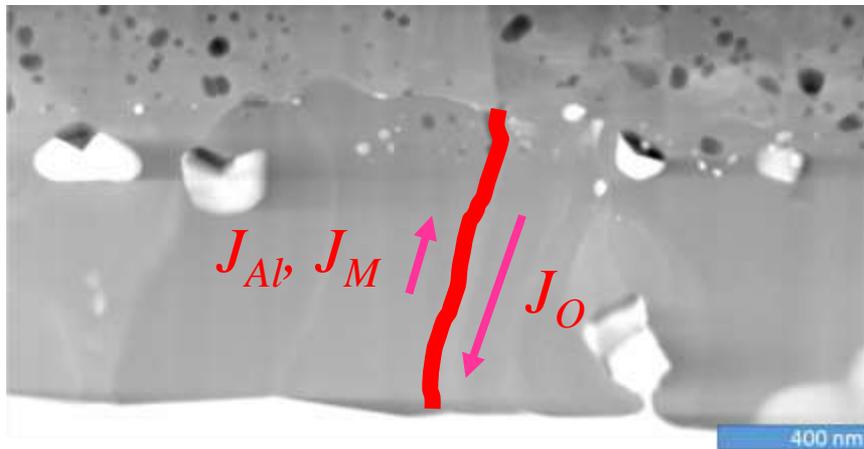
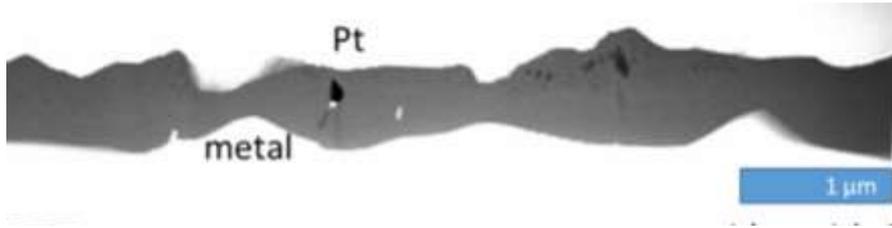
1 Institut für Angewandte Materialien - Werkstoffkunde (IAM-WK), Karlsruher Institut für Technologie (KIT), 76344 Eggenstein-Leopoldshafen, Germany

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Atom Probe Tomography:
Zr at a grain boundary in Al_2O_3 ,
Each dot represents one atom,
Al and O atoms are not displayed

TEM of typical oxide on NiAl



- Protective Al_2O_3 coating on NiAl-alloy
- O (and all other elements) in α -alumina diffuse mostly via grain boundaries (GBs)
- Minor outward diffusion of metal
- Decoration of GBs will influence the diffusion and thus oxidation
- Apparently grows inwards

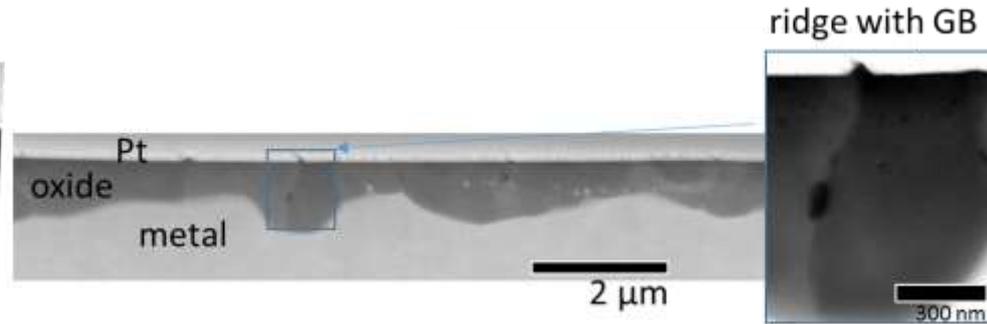
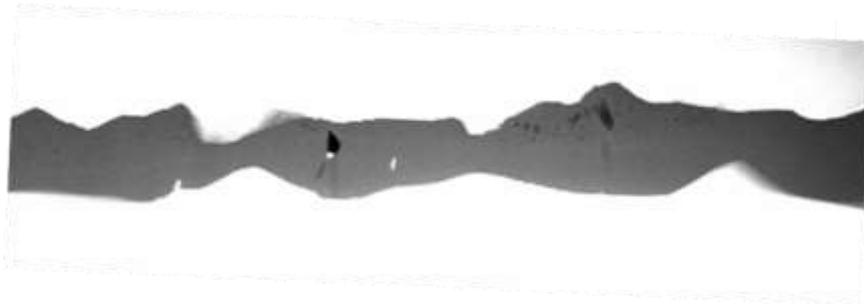
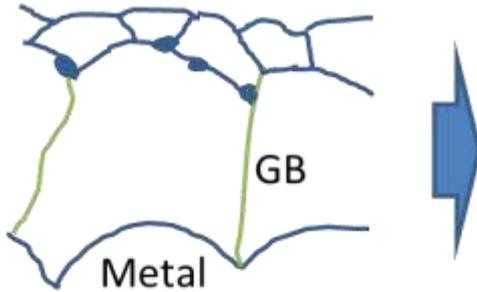
Material	Ni at.%	Al at.%	Zr ppma	Hf ppma	C ppma	S ppma	Cr ppma
Zr-doped	49.95	49.99	520	0	0	3	0
Hf-doped	49.83	50.07	0	480	36	<3	100
undoped	49.9	50.1		<1	40	<4	<100

Outward diffusion: Exp. idea

a) After 1st exposure

small grains
+ voids

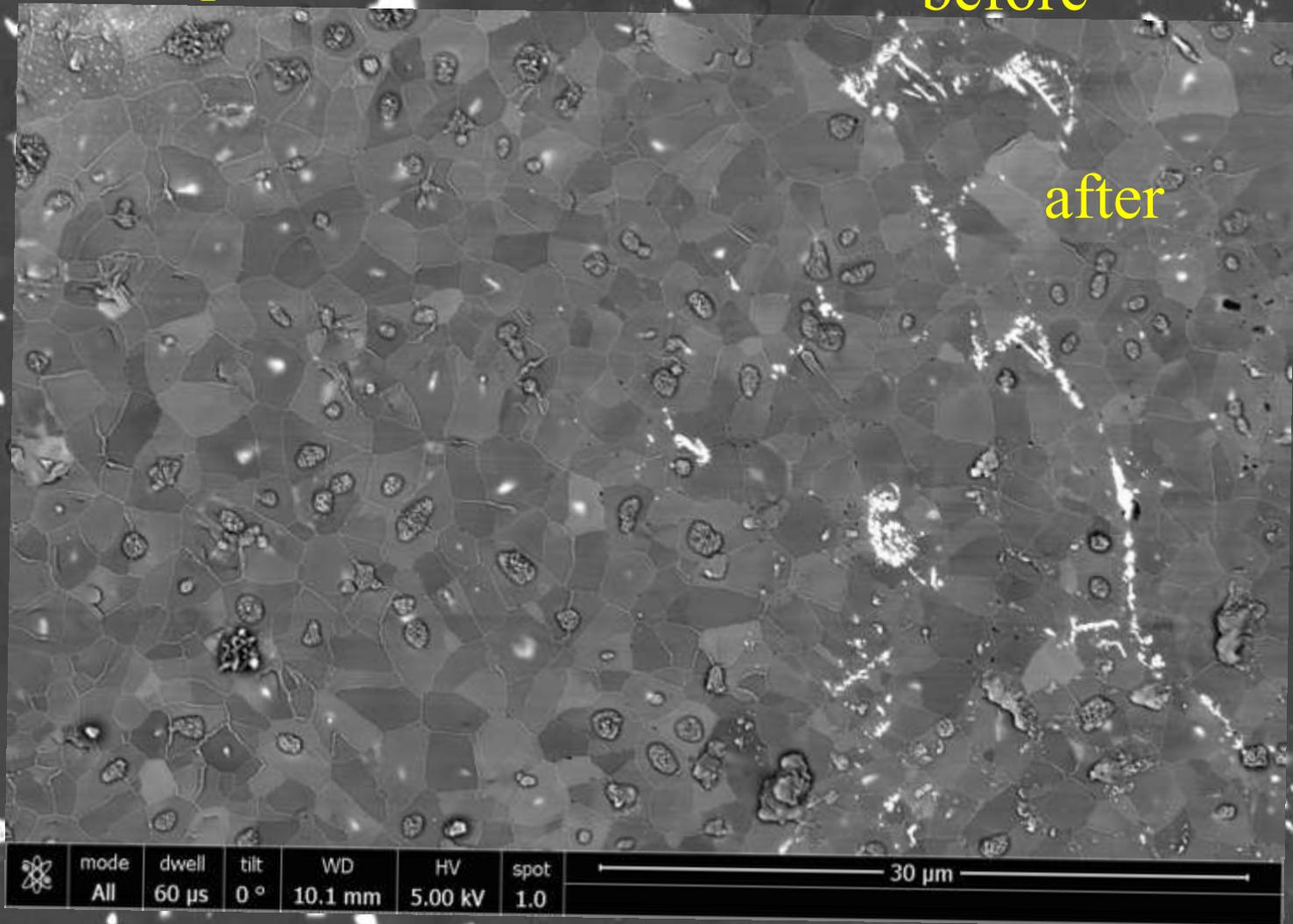
large Al₂O₃
grains



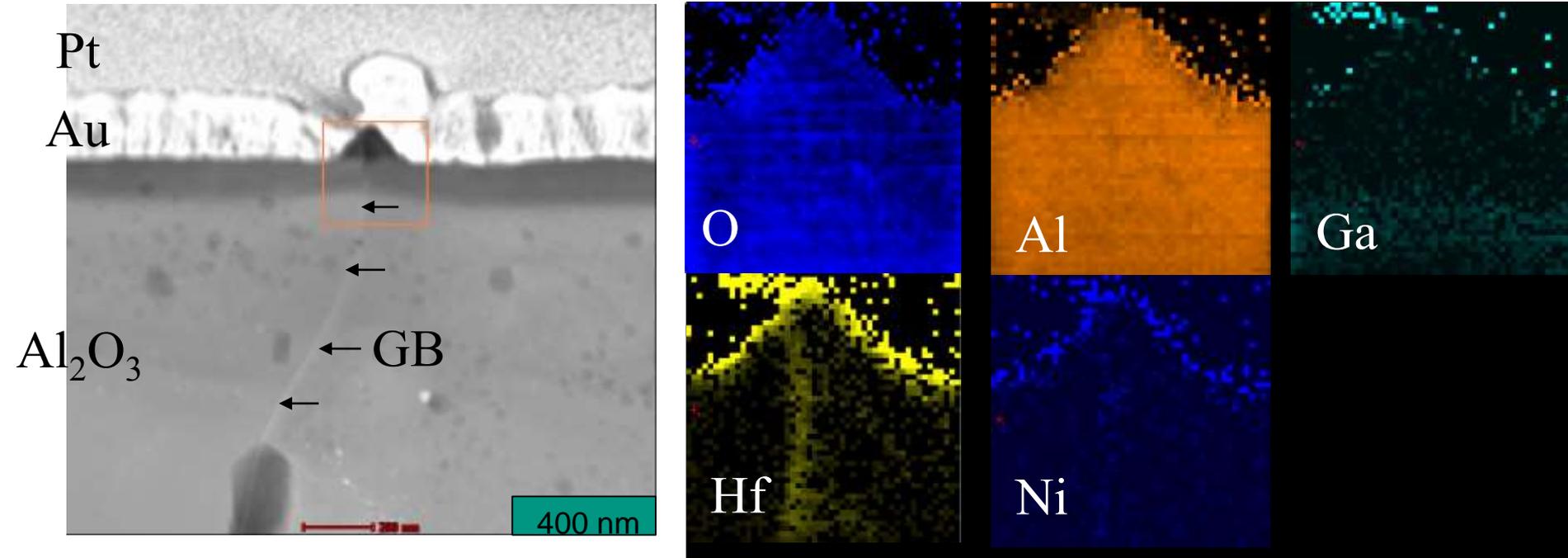
Hf 10h exposure

before

after

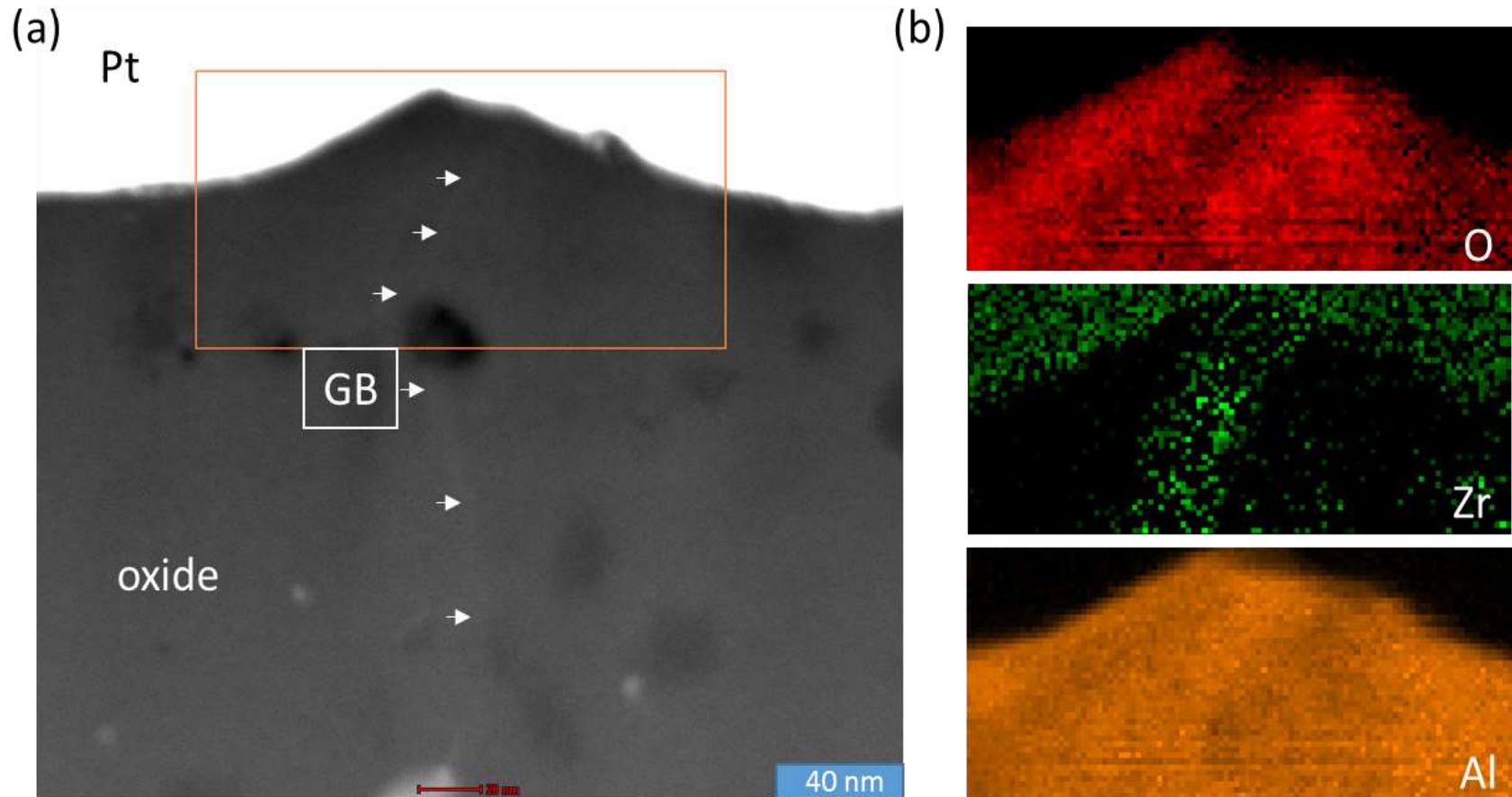


TEM of mech. pol. Hf sample



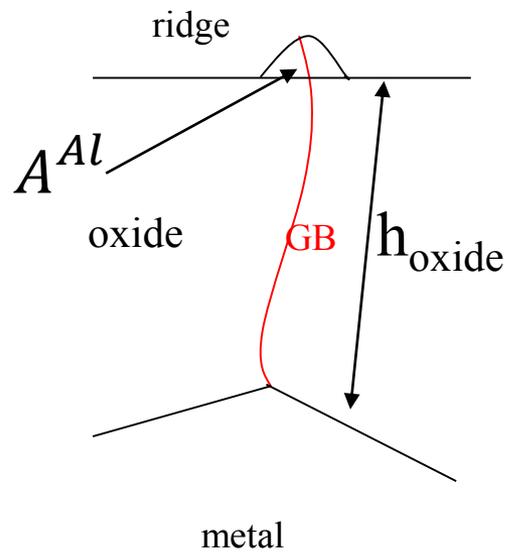
- No Ga contamination
- GB enriched with Hf and some Ni

TEM of Zr sample



- Zr enriched at the GB

How to calculate the flux



- Calculate the flux
 - Number of diffused Al-atoms N_{GB}^{Al}
 - Exposure time Δt (10h)

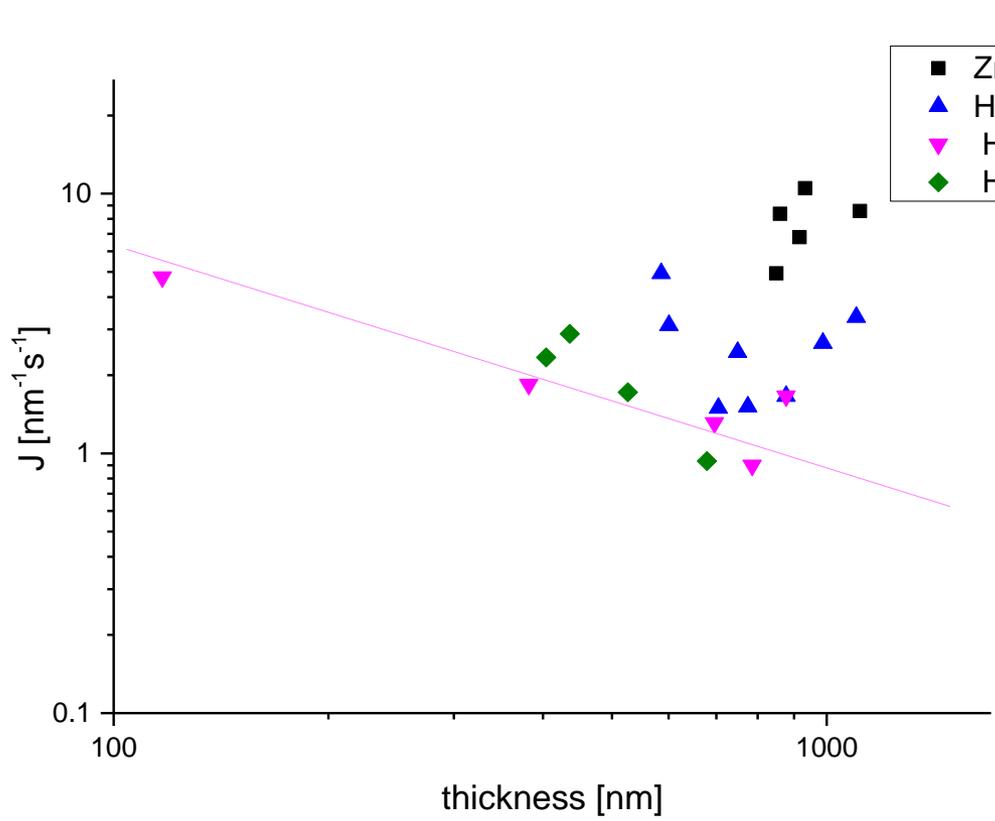
- Calculate number of atoms
 - Volume of ridge $V^{Al} = A^{Al} L_{GB}$
 - Length of GB L_{GB} (not height!)
 - Cross section area of ridge A^{Al}
 - Volume of Al_2O_3 unit cell: $V_u = 2.54 \cdot 10^{-22} \text{ cm}^3$
 - Number of Al atoms per unit cell: 12

$$J_{Al} = \frac{N_{GB}^{Al}}{L_{GB} \Delta t}$$

$$N_{GB}^{Al} = \frac{12 V^{Al}}{V_u}$$

$$J_{Al} = \frac{12 A^{Al}}{V_u}$$

Flux of Al through GBs at 1100°



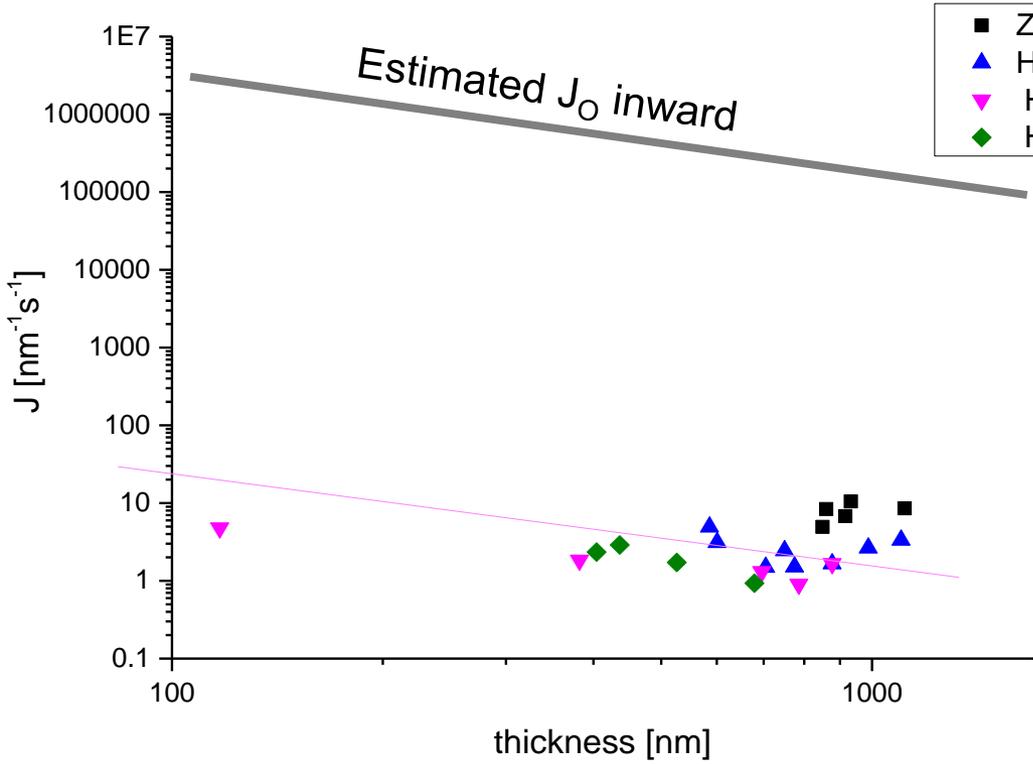
■ Mech. polishing enhances ridge growth

■ Zr allows higher outward flux than Hf

Should follow Fick's 1. law (assuming h_{oxide} is constant)

$$J_{\text{GB}}^{\text{Al}} = -\frac{A}{h_{\text{oxide}}}$$

Flux of Al through GBs at 1100°



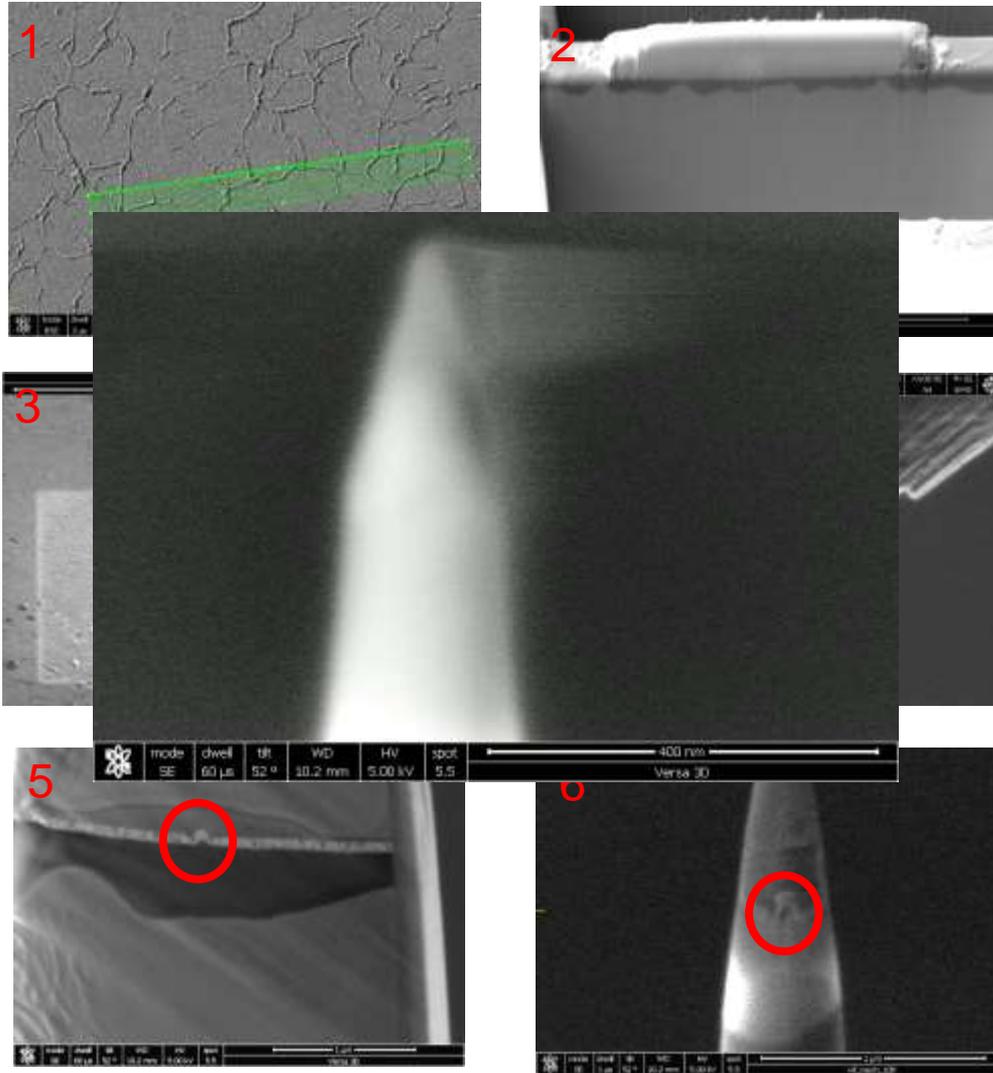
- Mech. polishing enhances ridge growth
- Zr allows higher outward flux than Hf
- Inward flux six orders of magnitude larger

Should follow Fick's 1. law (assuming h_{oxide} is constant)

$$J_{\text{GB}}^{\text{Al}} = -\frac{A}{h_{\text{oxide}}}$$

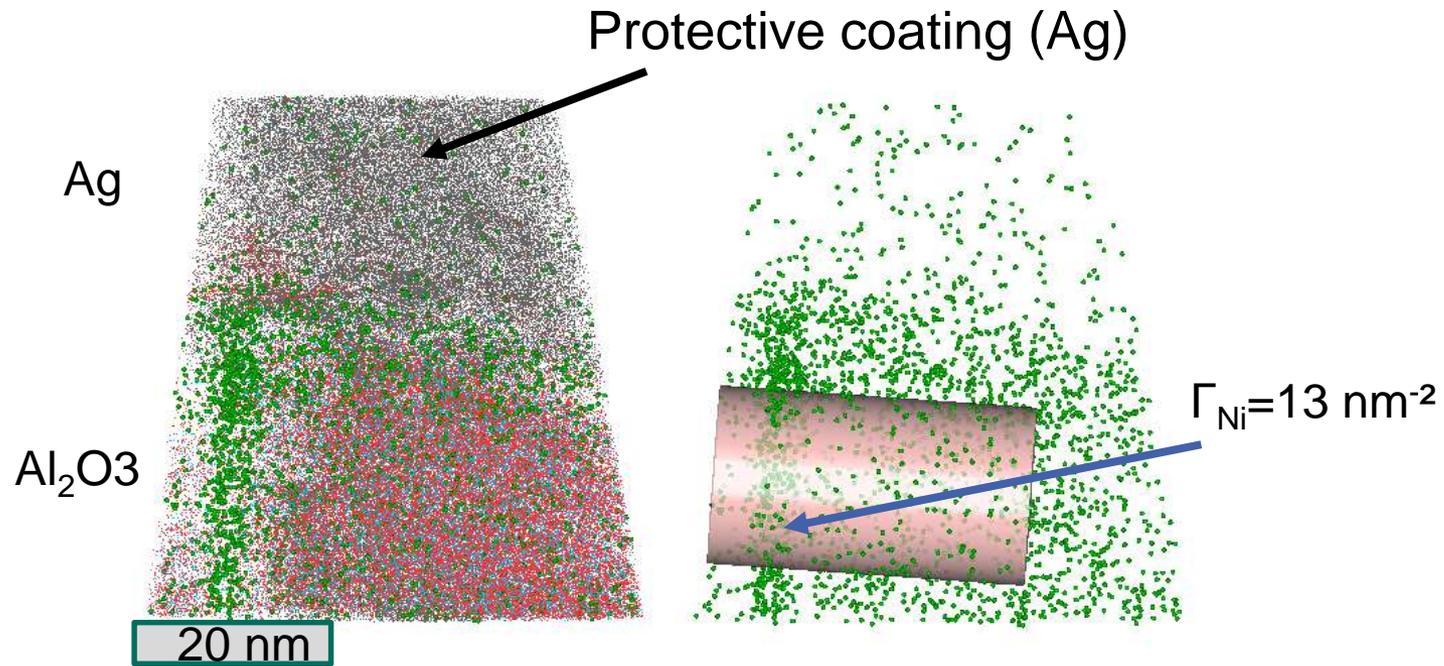
[1] T. Boll, K. A. Unocic, B. A. Pint, A. Mårtensson, and K. Stiller, "Grain Boundary Chemistry and Transport Through Alumina Scales on NiAl Alloys," *Oxid. Met.*, pp. 1–11, 2017.

Sample preparation of surface features with FIB

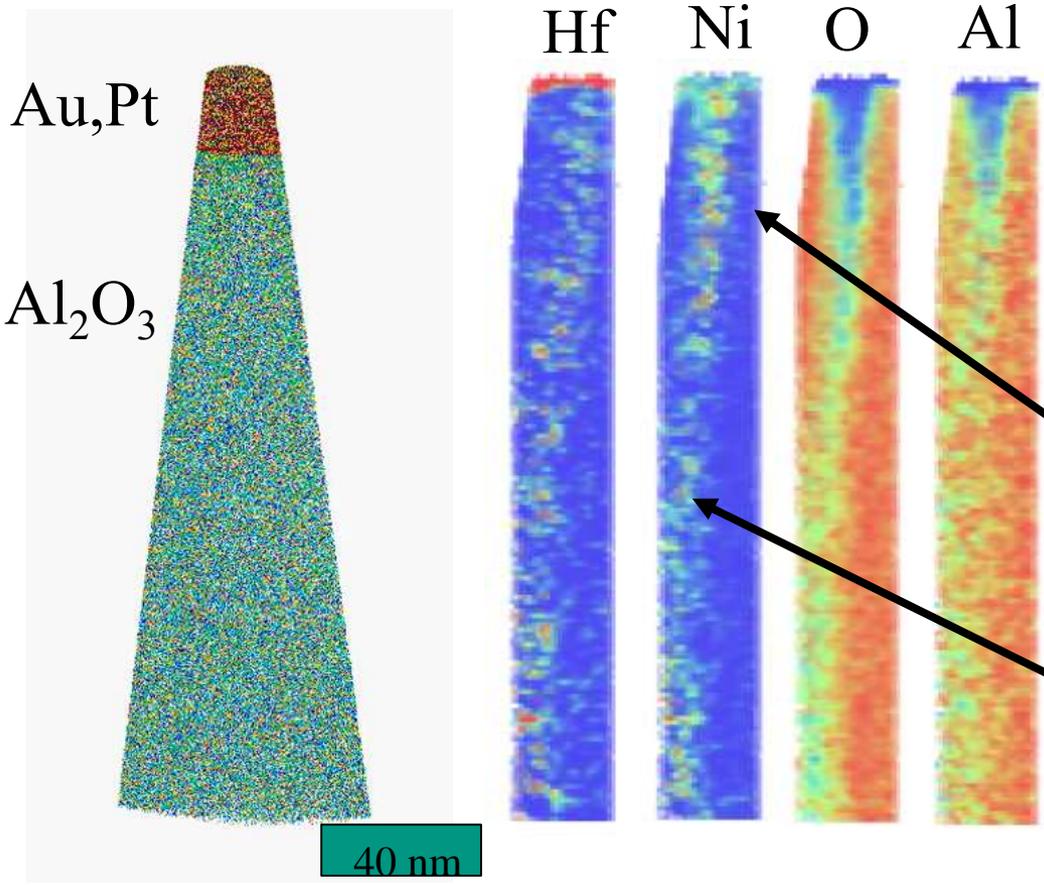


- Protective sample coating (Au, Ag)
- Mark area with Pt, deposit 2 μ m Pt on 2x10 μ m
- Cut lamella
- Lift out
 - Attach lamella to OP
 - Cut lamella loose
 - Attach to micro tip
 - Cut loose
- Find feature (ridge)
- Annular milling at feature (ridge) while controlling position

Undoped NiAl



APT of Hf sample

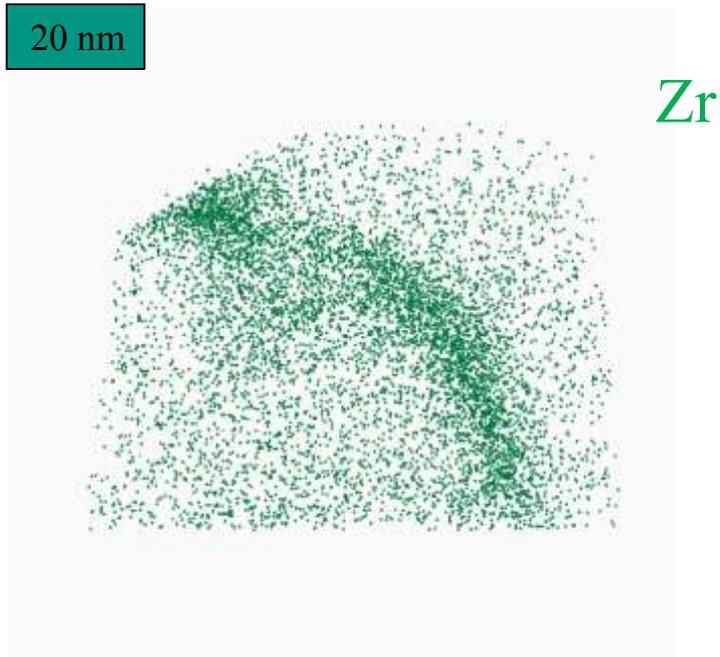


- GB with Hf, Ni
- Ni enriched at surface
- Gibbsian excess Γ
(Number of additional atoms per area in GB):

Hf: 0.5 nm^{-2}
Ni: 2.6 nm^{-2}

Hf: 0.35 nm^{-2}
Ni: 0.59 nm^{-2}

APT of Zr sample



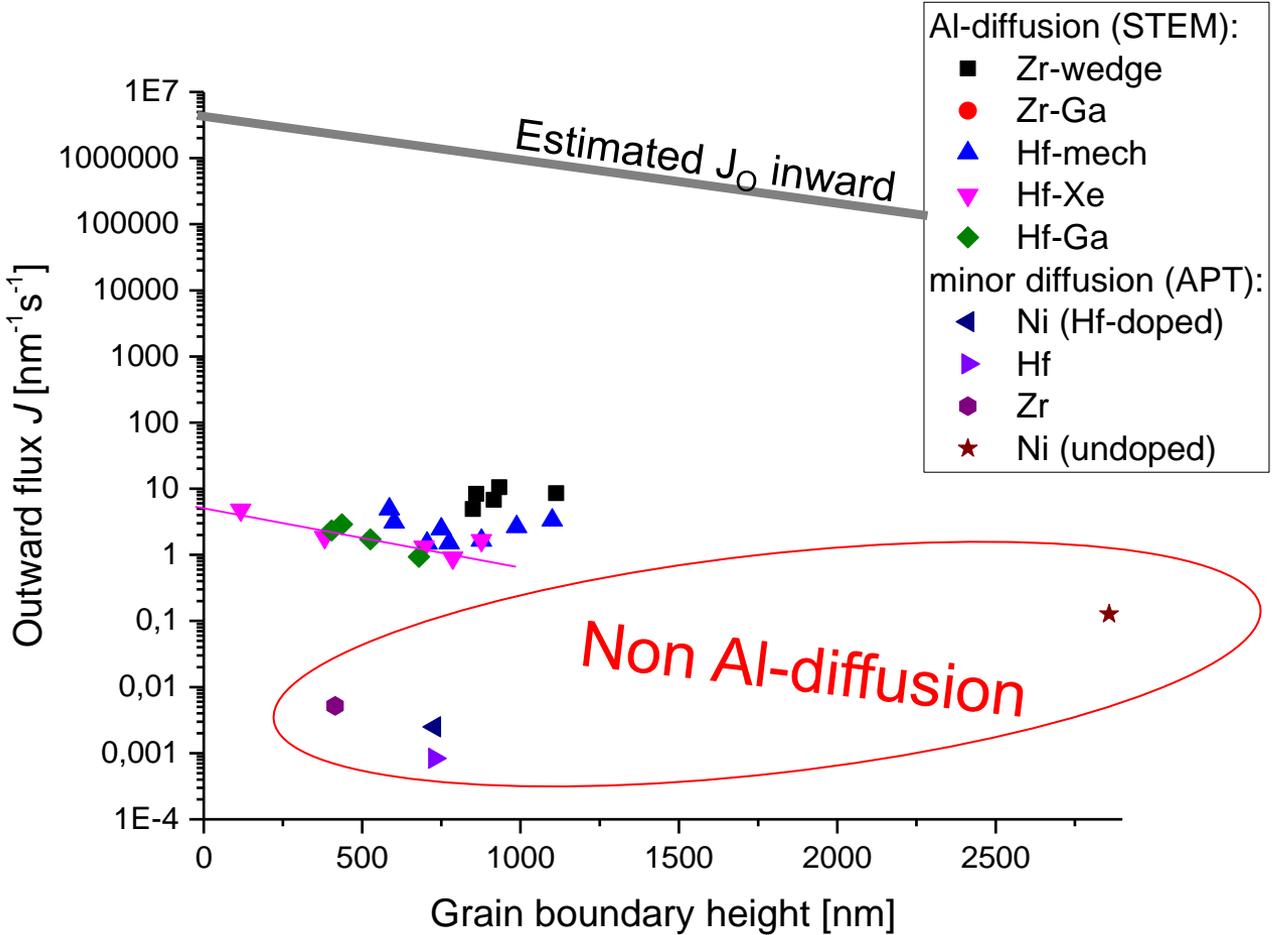
- Protective Ag on top of ridge-GB
- No Ni found
- $\Gamma_{\text{Zr}}: 2.5 \text{ nm}^{-2}$

Outward flux of Ni, Hf, Cr

- Outward flux into ridge GB
- Gibbsian excess
- Additional GB length (ridge height fro
- $J_X = \Gamma_X h_{\text{ridge}}$

→ amount of diffused material

$$J_O \gg J_{Al} \gg J_{Hf, Ni, Zr}$$



Conclusions

- **Outward Diffusion of Al** along Al_2O_3 GBs is observed by STEM Mechanical polishing introduces defects that promote diffusion
 - Hf reduces Al-outward diffusion stronger than Zr
- Zr is enriched at GBs → **Outward diffusion of Zr, Hf**
- Hf is enriched at GBs
- Ni at the GB and at the top of the ridge → **Outward diffusion of Ni**
- $J_{\text{O}} \sim 10^6 \text{ nm}^{-1}\text{s}^{-1} \gg J_{\text{Al}} \sim 1 \text{ nm}^{-1}\text{s}^{-1} \gg J_{\text{Hf,Ni,Zr}} \sim 10^{-3} \text{ nm}^{-1}\text{s}^{-1}$
- *Absence of reactive elements* → $J_{\text{Ni}} \sim 10^{-2} \text{ nm}^{-1}\text{s}^{-1}$

Thank you for your attention

You also want APT results: knmf.kit.edu, or contact me
KNMF grants APT time to suitable projects

Acknowledgements

Kinga A. Unocic, Bruce A. Pint (ORNL): NiAl+Hf and NiAl+Zr samples

Patrik Alnegren: Help with 2nd exposure and sputter coating: