

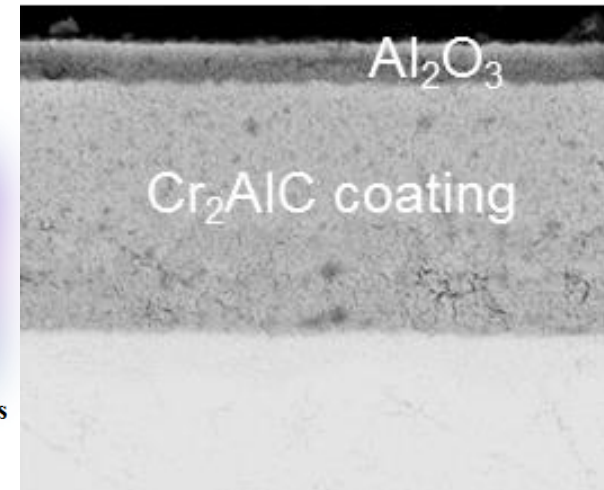
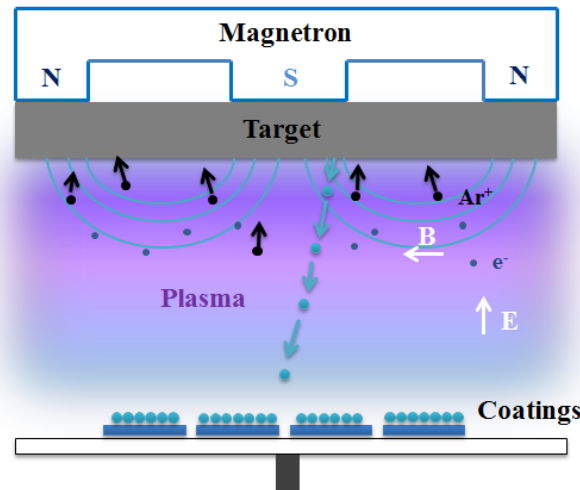
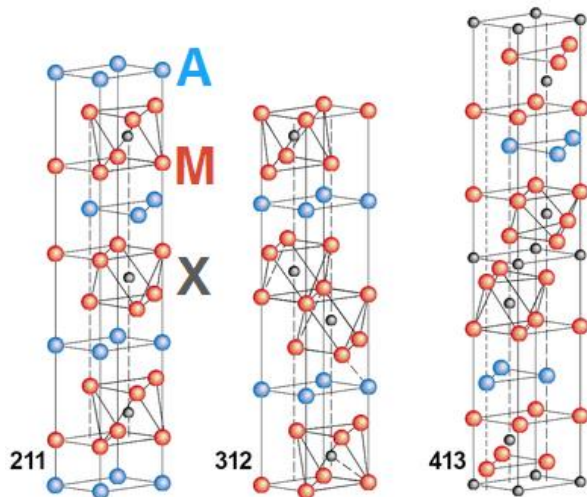
# Magnetron-sputtered Al-containing MAX phase carbide thin films and their application as oxidation-resistant coatings

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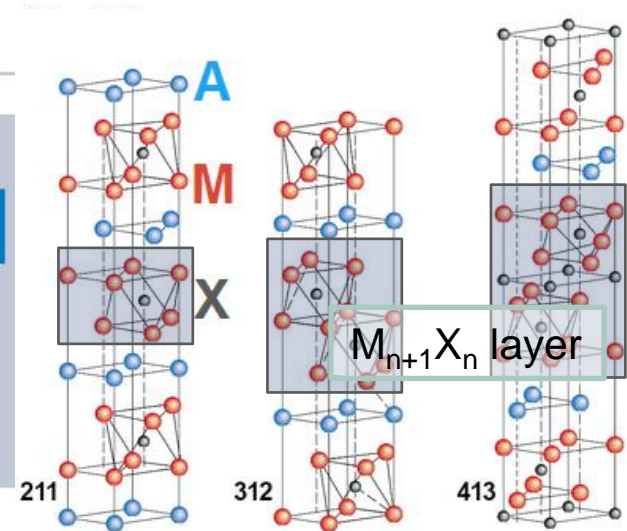
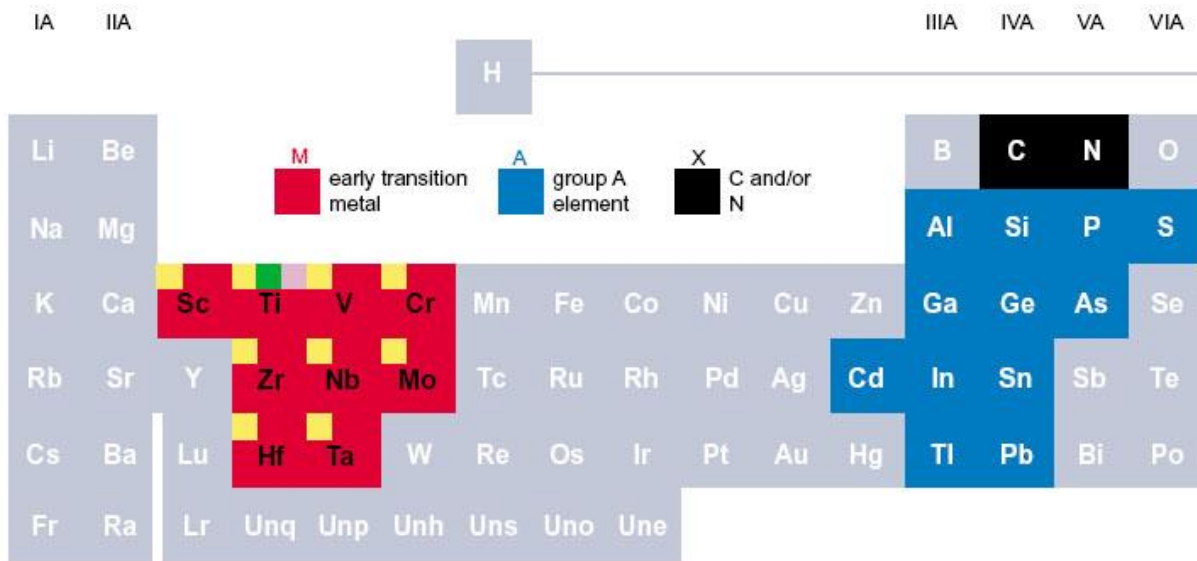
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# Introduction

- MAX phases: ternary carbide and nitride compounds described by the general formula  $M_{n+1}AX_n$ , where n typical is 1, 2, 3
- Nanolaminated crystal structure:  $M_6X$  octahedral interleaved with A layers

## $M_{n+1}AX_n$ (MAX) phase



**Nanolaminated crystal structure**

***M-X: strong covalent-ionic bonds***

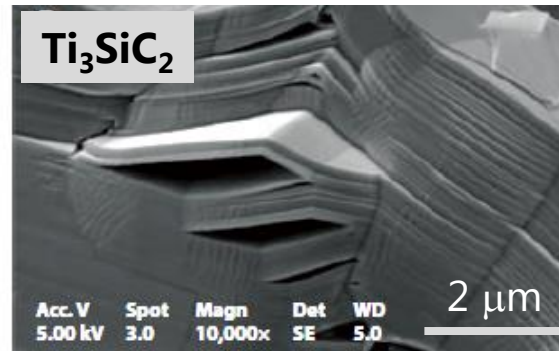
***M-A: weak metallic bonds***

# Introduction

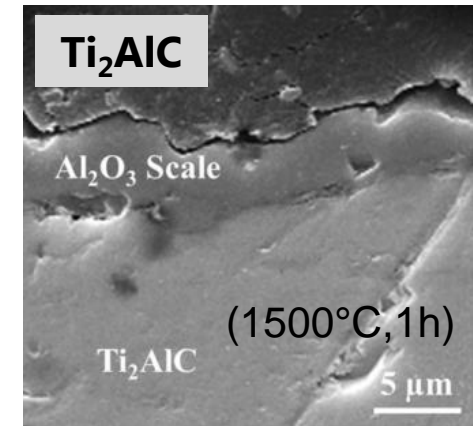
- Unique combination of physical, chemical, mechanical properties of both metals and ceramics



***Machinability***



***Damage tolerance***



***Oxidation resistance***

***Low density***

***High melting/decomposition temperature***

***Thermal conductivity***

- Promising application prospects as structural materials and protective coatings in harsh environments

# Motivation

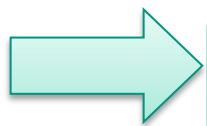
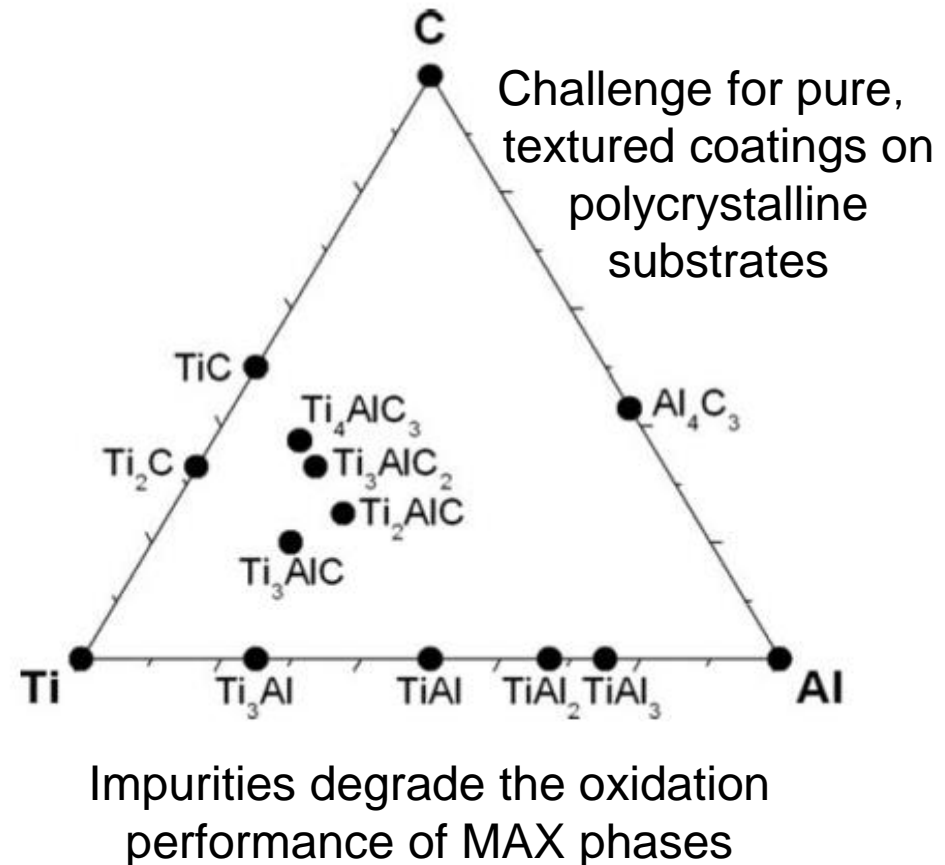
## ➤ MAX phase coatings

### ■ Spraying

- High velocity oxy-fuel spray
- Cold spraying

### ■ Physical vapor deposition (PVD)

- Cathodic arc evaporation
- Pulsed laser deposition
- Magnetron sputtering
  - sputtering with 3 element sources
  - sputtering with compound targets
  - reactive sputtering
  - sputtering-solid state reactions

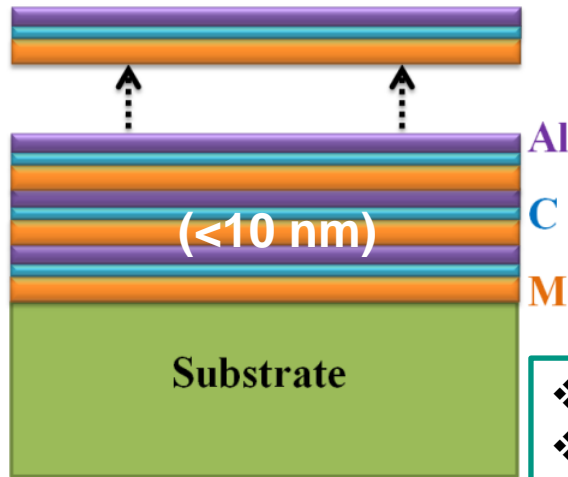


Magnetron sputtering and evaluation of Ti<sub>2</sub>AlC, Cr<sub>2</sub>AlC and Zr<sub>2</sub>AlC as oxidation resistant coatings (phase-pure)

# Experimental arrangement

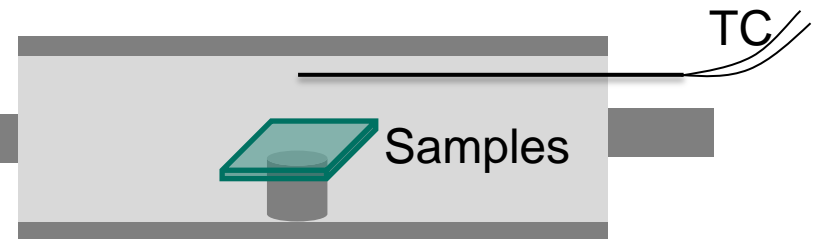
## A two-step approach

### Film design



Square: 10 × 10 mm

### Ex-situ annealing



- ❖ Reduced diffusion length
- ❖ Precisely stoichiometry control

### Magnetron Sputtering

Film thickness:

- ~ 3  $\mu\text{m}$  on  $\text{Al}_2\text{O}_3$ ,  $\text{SiO}_2/\text{Si}$
- ~ 5.5  $\mu\text{m}$  on Zircaloy-4 (~98wt%Zr)  
(fuel claddings in nuclear reactors)

### Annealing parameters

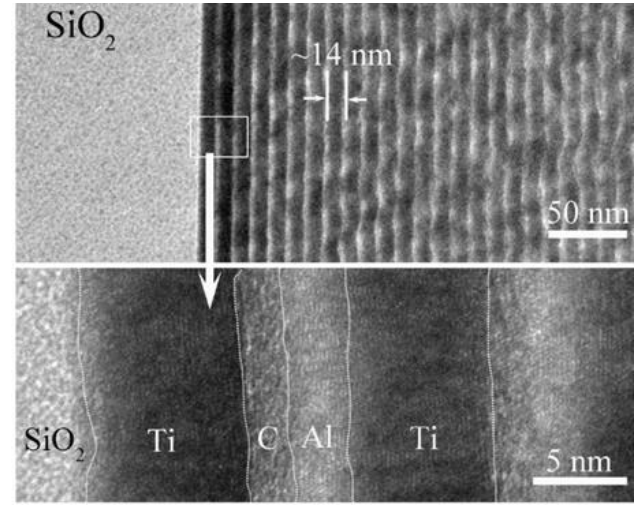
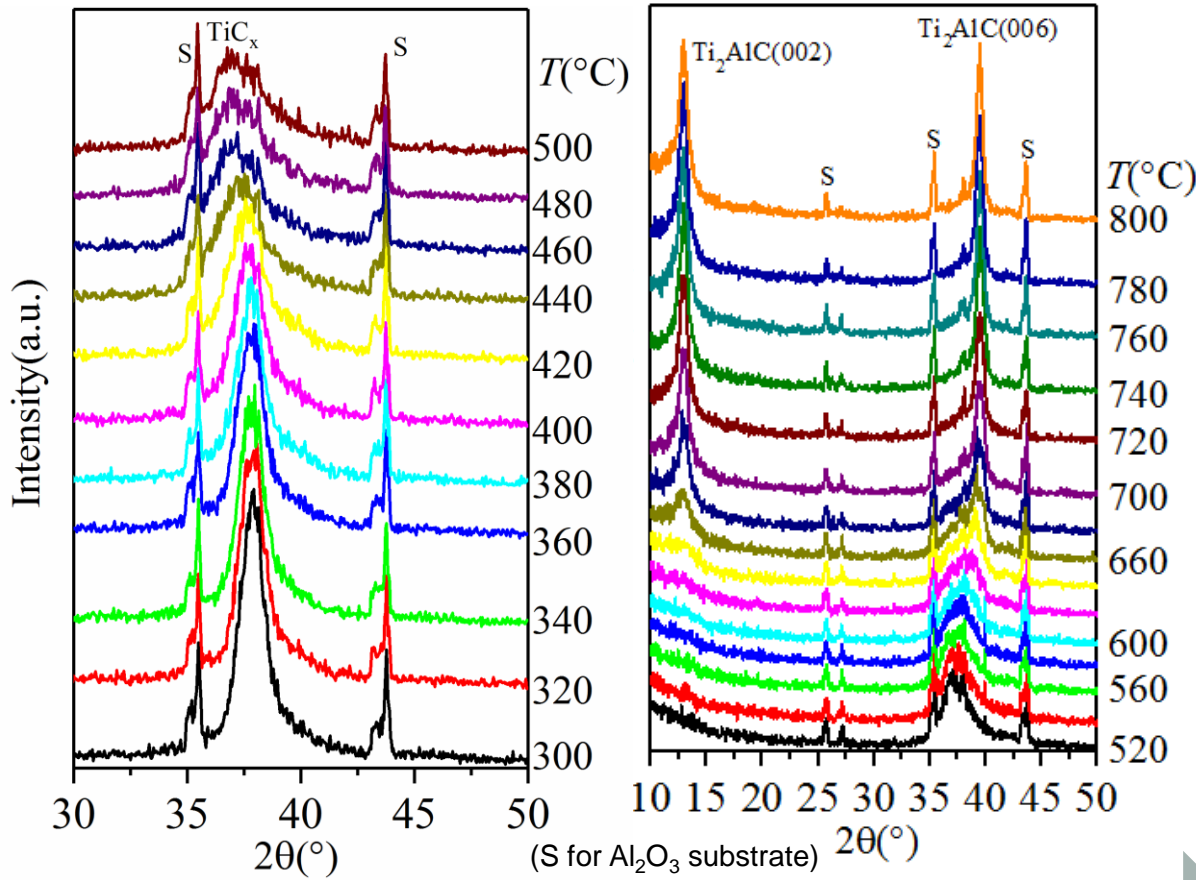
Annealing in pure Ar

400°C-1200°C @10 min

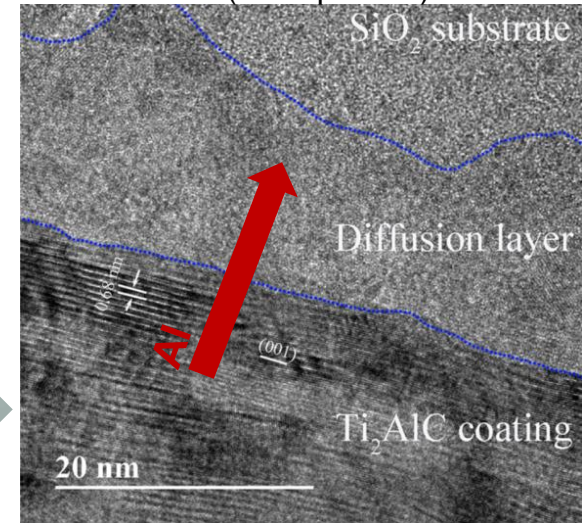
Heating/cooling rate: 10 K/min

# Synthesis of MAX phase coatings

## ➤ $Ti_2AlC$



(as-deposited)



(800°C, 10min)

As-deposited

440°C

660°C

800°C

Ti/C/Al multilayers

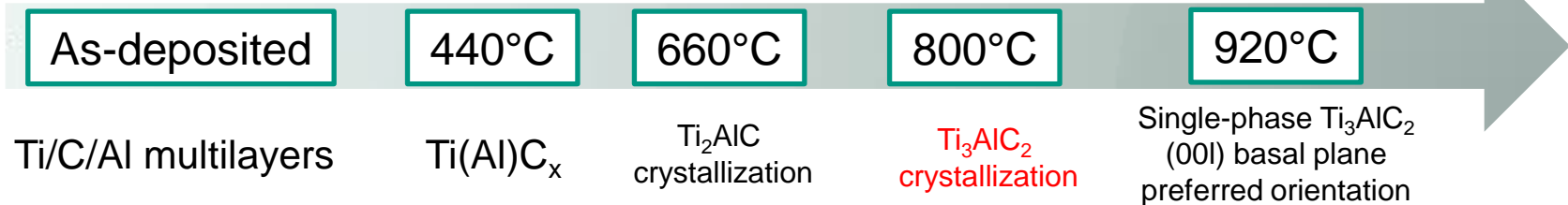
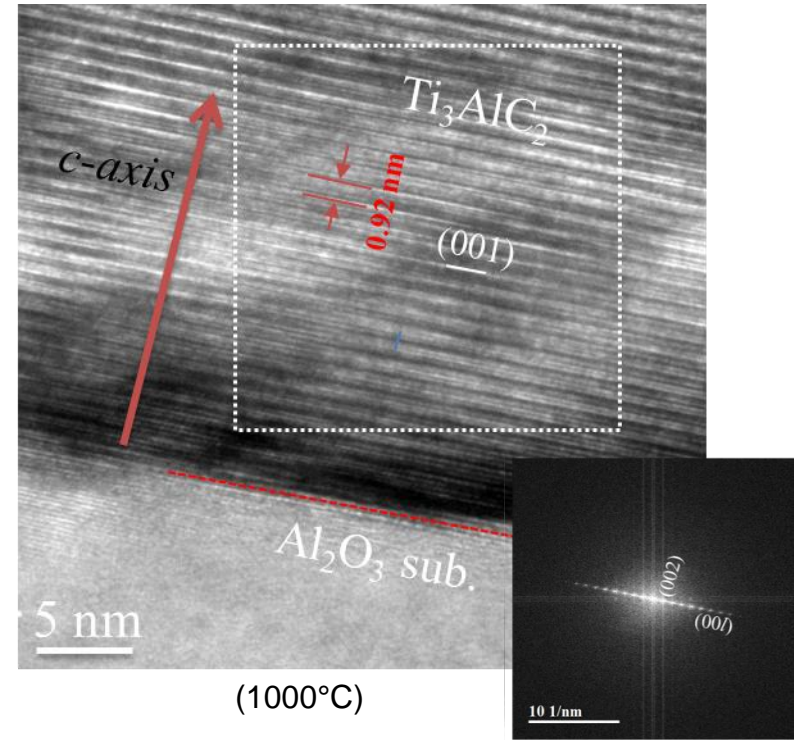
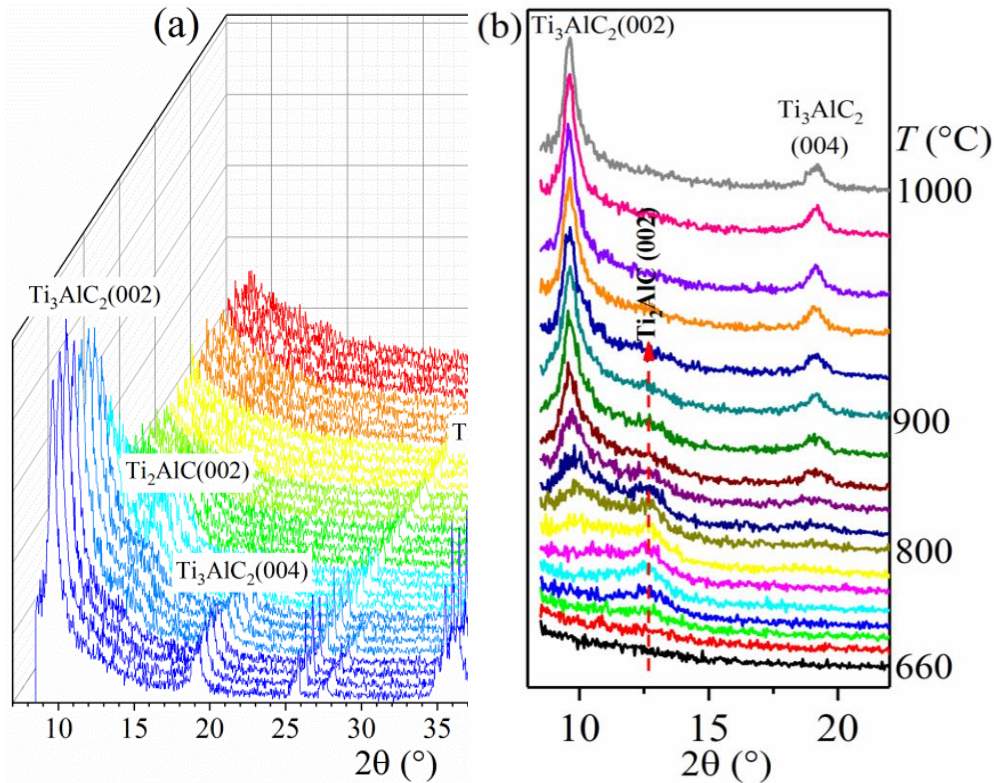
$Ti(Al)C_x$

$Ti_2AlC$   
crystallization

Single-phase  $Ti_2AlC$   
(001) basal plane  
preferred orientation

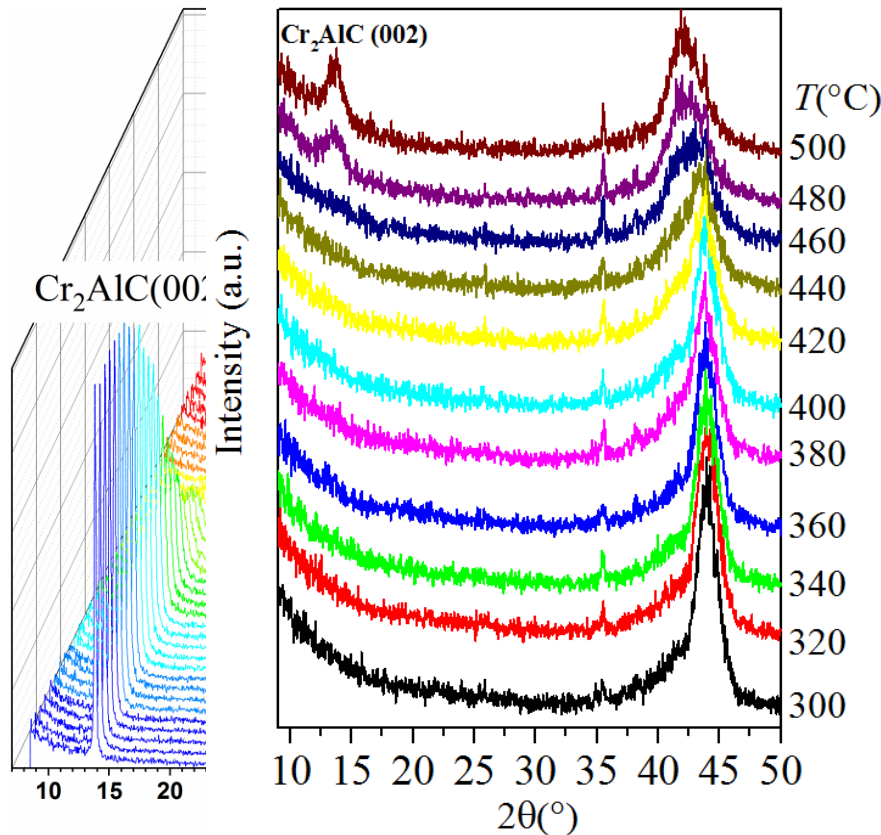
# Synthesis of MAX phase coatings

## ➤ $Ti_3AlC_2$



# Synthesis of MAX phase coatings

## ➤ Cr<sub>2</sub>AlC coating



## ➤ Texture analysis

As-deposited



M layers: Textured growth

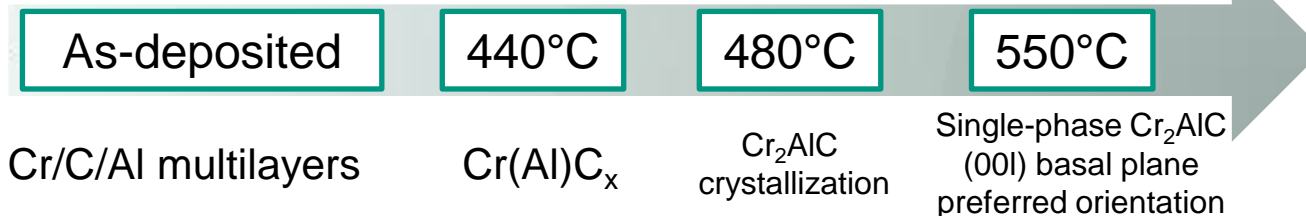
Ti, hexagonal, (002)

Cr, cubic, (110)



d-spacing misfit

Ti(002)/Ti<sub>2</sub>AlC(00l): 3.3%  
Cr(110)/Cr<sub>2</sub>AlC(00l): 4.5%

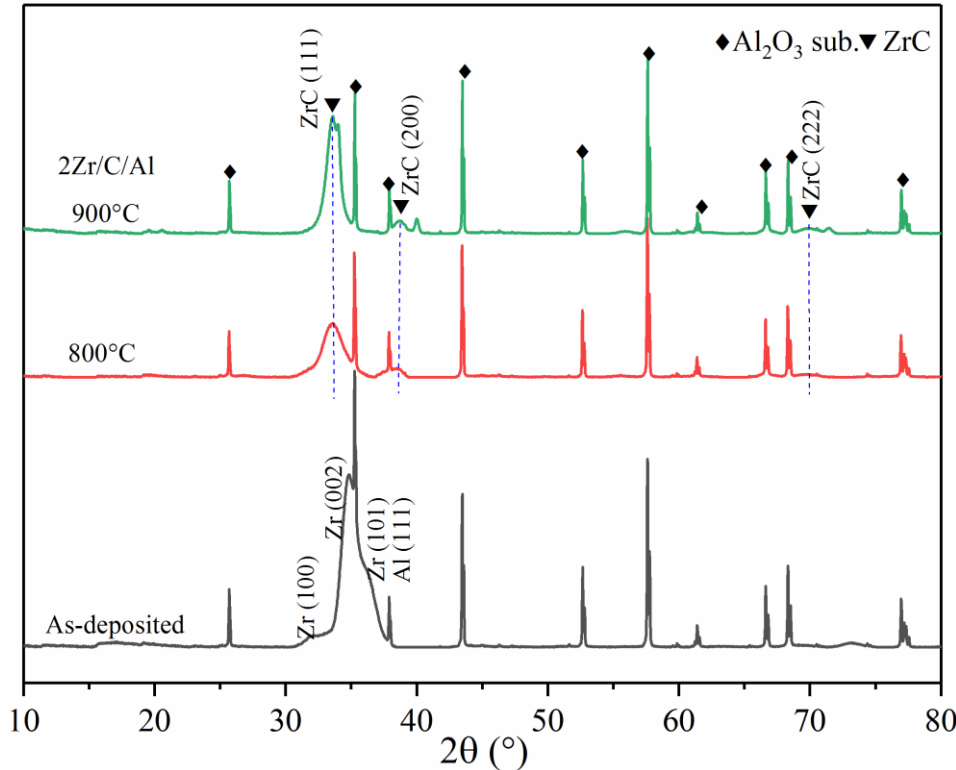


Crystallographic orientation relationship prompts textural growth of MAX phase grains.

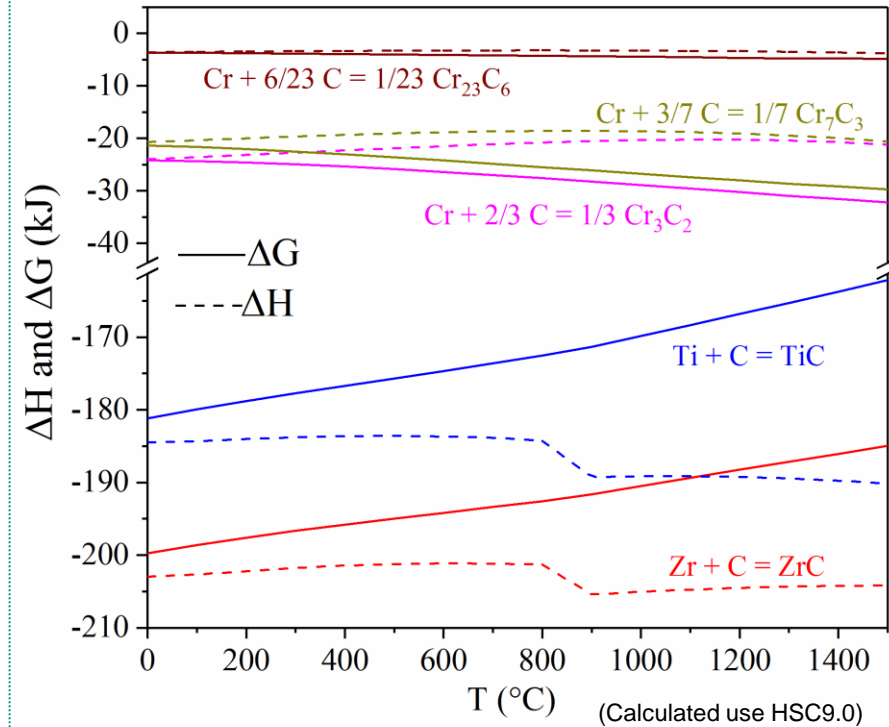


# Synthesis of MAX phase coatings

## ➤ Zr-C-Al coating



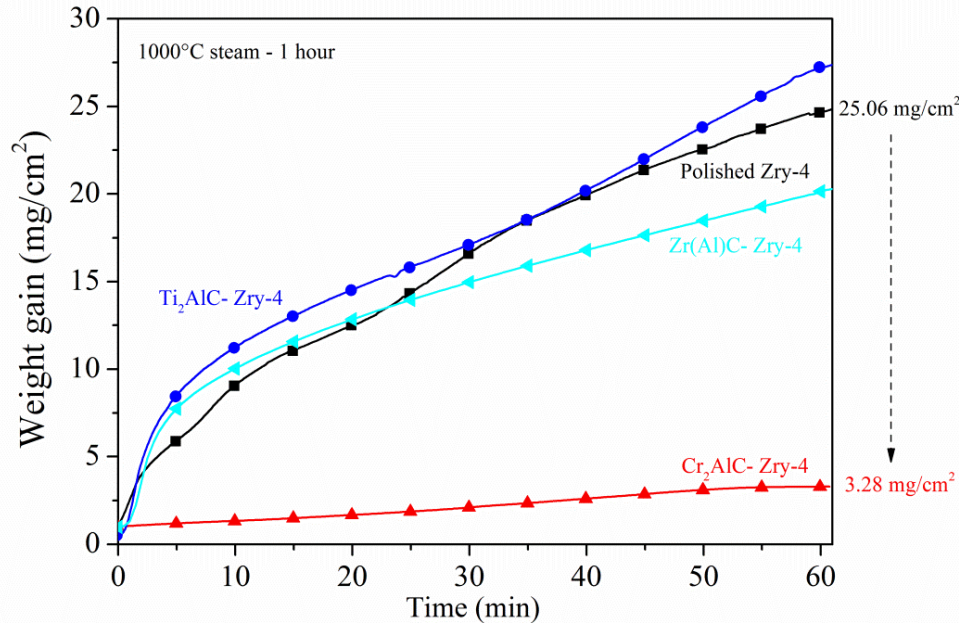
- Zr(Al)C carbide formation, rather than ternary nanolaminated MAX phase after 600-1200°C annealing.



- Thermodynamic metastable of  $\text{Zr}_2\text{AlC}$
- Stability:  $\text{ZrC} > \text{TiC} > \text{CrC}_x$
- Crystallization temperature of MAX phases
  - $\text{Ti}_2\text{AlC}$  (~ 700°C) >  $\text{Cr}_2\text{AlC}$  (~ 500°C)
  - $\text{Ti}_3\text{AlC}_2$  (~ 800°C) >  $\text{Ti}_2\text{AlC}$  (~ 700°C)

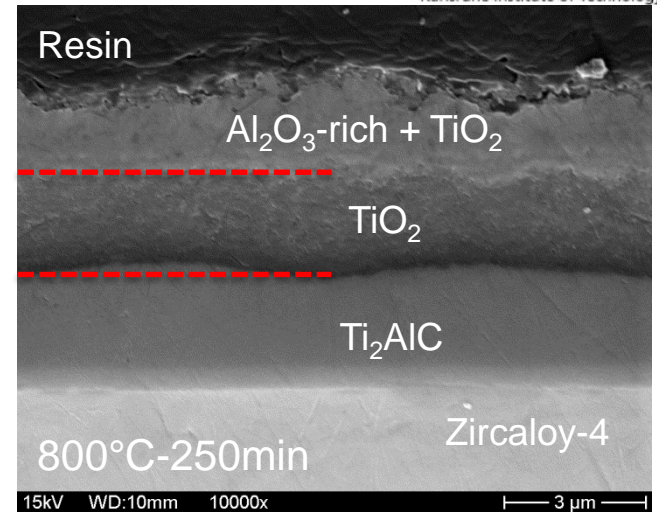
# Oxidation of coated Zircaloy-4

➤ Isothermal 1000°C 60 min in steam

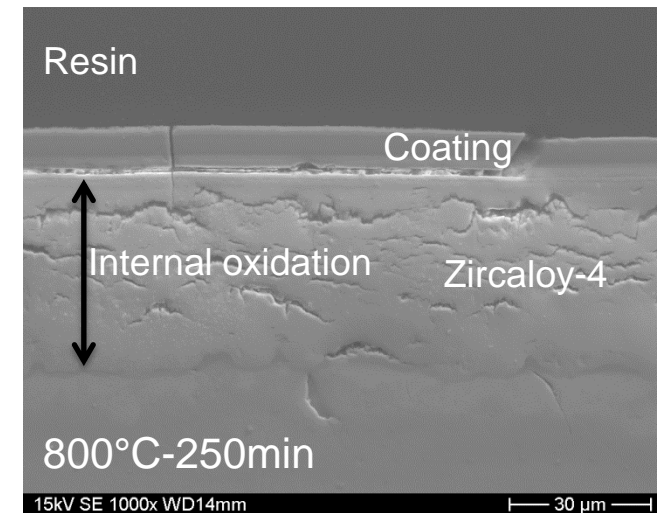


(Annealing: 800°C for Ti<sub>2</sub>AlC, 550°C for Cr<sub>2</sub>AlC, 600°C for Zr(Al)C)

- Significant spallation of Ti<sub>2</sub>AlC and Zr(Al)C coatings after oxidation at 1000°C.
- Formation of Al<sub>2</sub>O<sub>3</sub>-rich layer with TiO<sub>2</sub> or ZrO<sub>2</sub> layer beneath at low temperatures, rather than dense alumina layer.



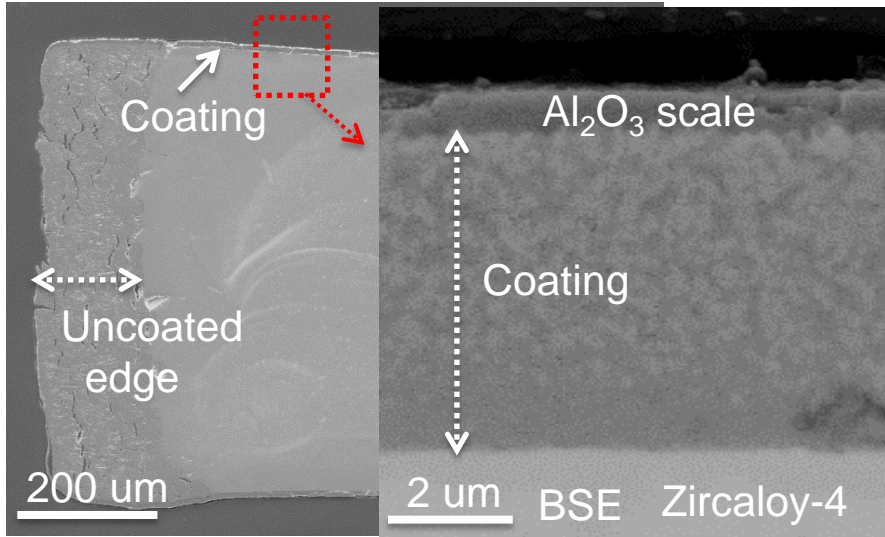
Ti<sub>2</sub>AlC-Zircaloy 4



Zr(Al)C-Zircaloy 4

# Oxidation of coated Zircaloy-4

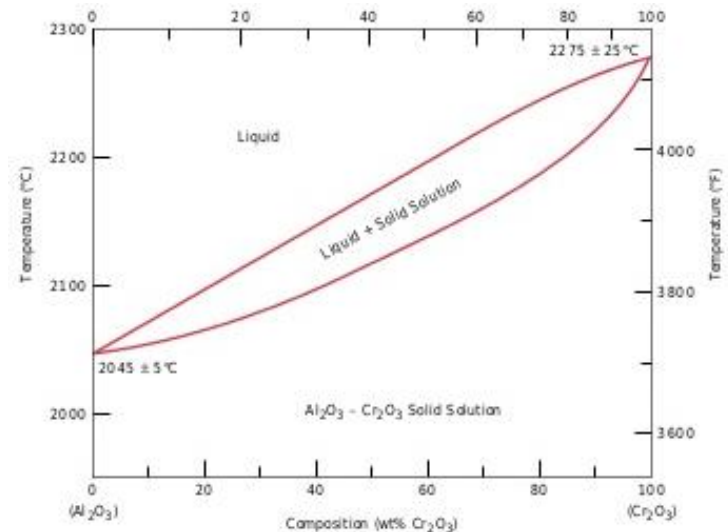
➤ Isothermal 1000°C 60 min in steam



Cr<sub>2</sub>AlC-Zircaloy 4

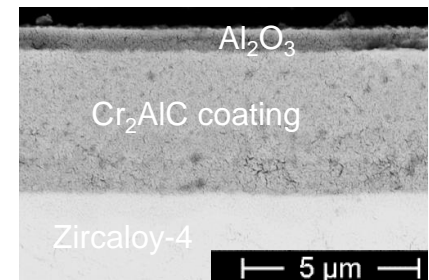
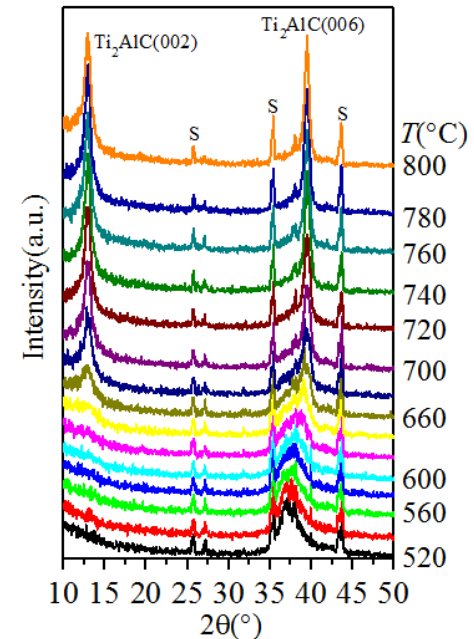
- Growth of dense alumina scale during high-temperature oxidation of Cr<sub>2</sub>AlC coating.
- High adherence without spallation.

- Low protective effect of TiO<sub>2</sub> and ZrO<sub>2</sub> scale
- Phase transformation of TiO<sub>2</sub> and ZrO<sub>2</sub>, prone to cracking
  - TiO<sub>2</sub>: anatase, rutile
  - ZrO<sub>2</sub>: monoclinic, tetragonal, cubic
- Cr<sub>2</sub>O<sub>3</sub> and α-Al<sub>2</sub>O<sub>3</sub>: similar crystal structure, completely miscible
  - Cr<sub>2</sub>O<sub>3</sub> as nucleation sites for α-Al<sub>2</sub>O<sub>3</sub>



# Conclusions

- A two-step approach to synthesize MAX phase coatings
  - Nanoscale multilayer stacks+ thermal annealing in Ar
  - Pure  $Ti_2AlC$ (800°C),  $Ti_3AlC_2$  (900°C) and  $Cr_2AlC$ (550°C) MAX phases obtained with textured structure independent of substrates
  - Zr(Al)C formation in the system Zr-C-Al
- Oxidation of coated Zircaloy-4
  - Growth of  $Al_2O_3$ -rich layer with  $TiO_2$  or  $ZrO_2$  layer beneath for  $Ti_2AlC$  and Zr(Al)C coatings with low oxidation resistance
  - Growth of a dense  $Al_2O_3$  scale for  $Cr_2AlC$  coatings with excellent HT oxidation resistance
- Phase-pure and highly textured MAX phase coatings can offer benefits for some specific applications



## Many thanks to:

High-temperature chemistry group, Thin film group  
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# Thanks for your attention !