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## Development of an H-Sensor system for lithium melts

### Non-metal Impurity Measurement in Liquid Lithium :

**Motivation:** Within the DONES facility to be built in Granada, Spain, a deuteron irradiated Li-flow target generates DEMO-relevant neutrons, by which the foreseen fusion power plant materials can be tested and evaluated under DEMO conditions. A critical aspect will be the hydrogen impurity content and its control in the liquid lithium systems of IFMIF-DONES => **important issue due to corrosion and safety risks.**

**Objective and approach:** So far, a consequential EUROFUSION task was the development of an ElectroChemical Hydrogen Sensor for Liquid Lithium (ECHSLL).

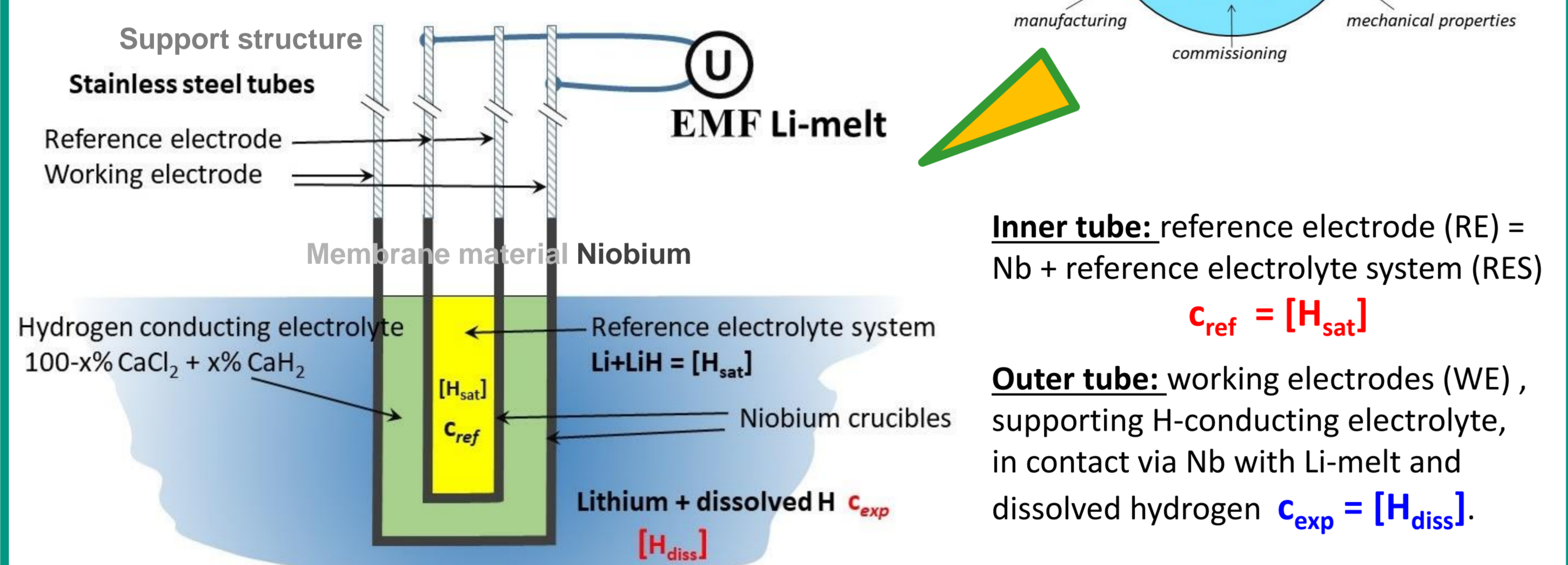
## Design of ECHSLL

### Electrochemical conception of ECHSLL:

#### Cross-section task requirements:

- High permeable for hydrogen
- + stable in liquid lithium +
- + mech stability vs. shear forces
- + insulation for potential measurements
- + conducting voltage contacts,.....

#### Solution: ECHSLL as "single-rod measuring cell"



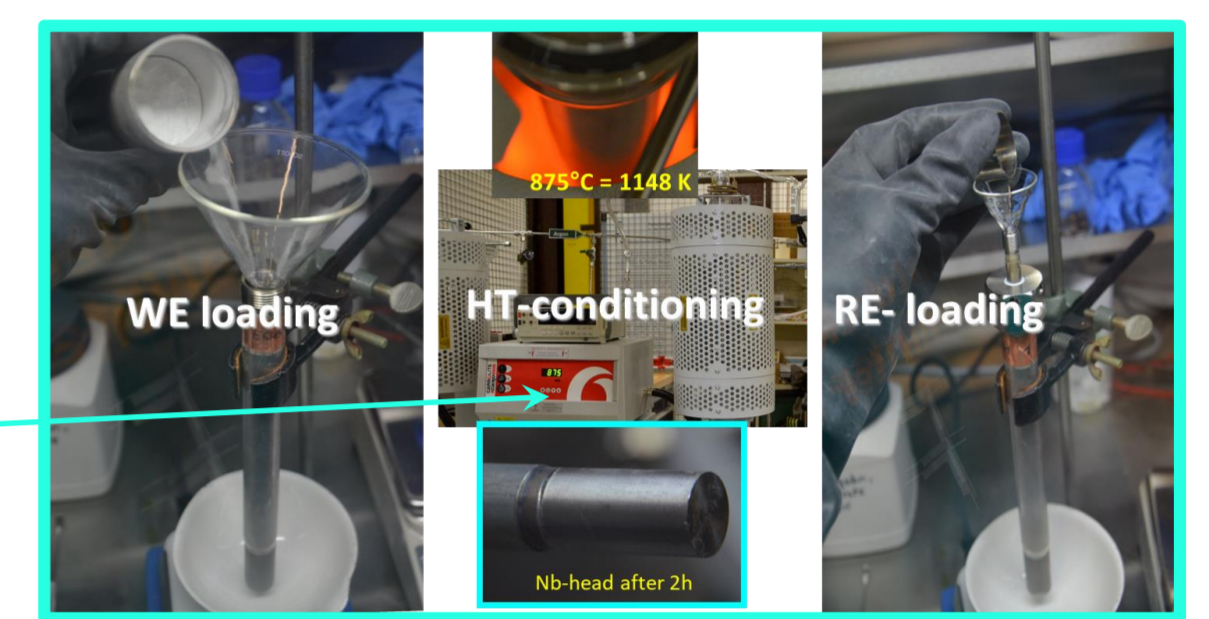
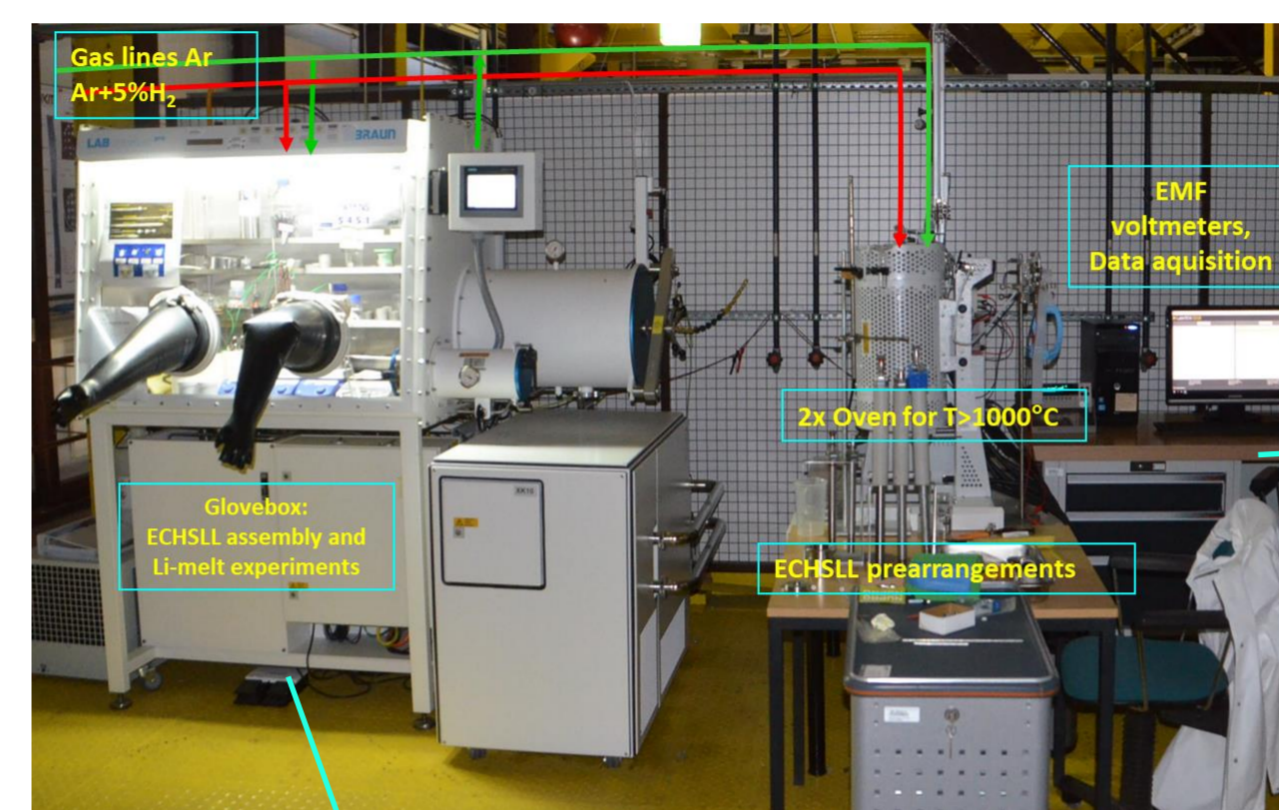
### ECHSLL as single-rod measuring cell

EMF as result of voltage between potentials of hydrogen dissolved in Li-melt and the concentration in the RE-system

## ECHSLL-application and tests

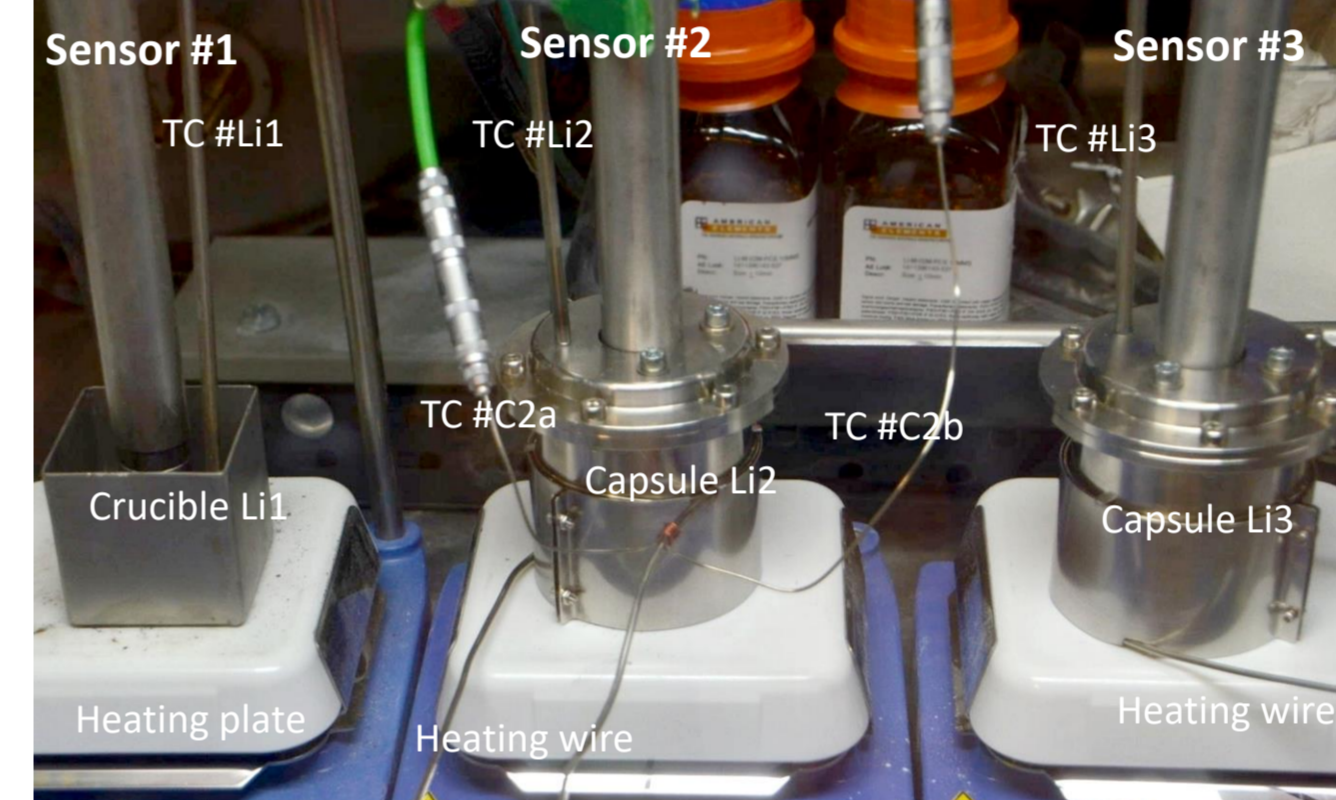
### Preparation + Investigation was carried out in inert atmosphere:

ECHSLL chemical systems/substances are sensitive to water/moisture, air, and nitrogen.



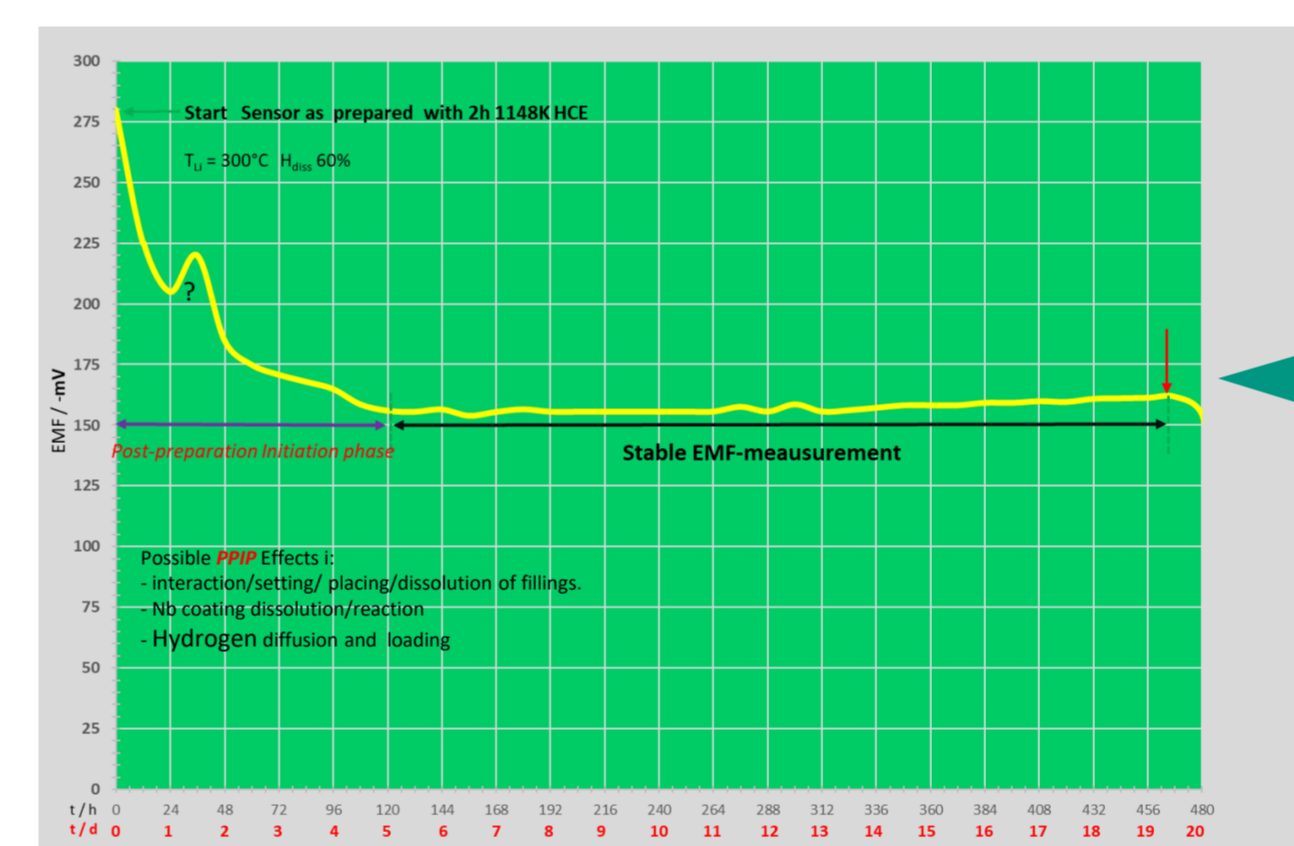
### Main steps of sensor preparation:

filing of outer tube (WE) with  $CaCl_2/CaH_2$ , HCE HT-conditioning, RE filling with Li/LiH



### Measurements set-ups :

Besides crucibles for open visible short-term experiments with 10-50 ml melt, also use of heating capsules with Li melt amounts up to 100 ml; three independent contemporary systems (containment material: steel). Due to prevent observed electrical charging effects, each sensor connected to an own voltmeter.



### Measurements :

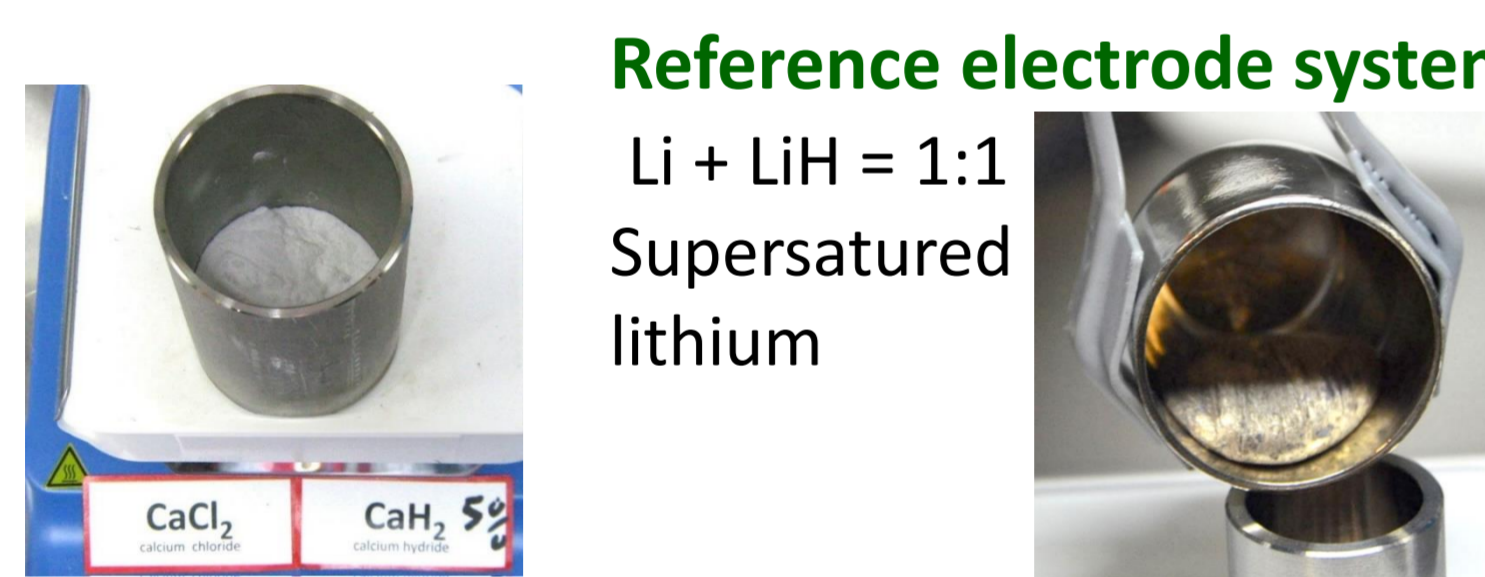
Long-term measurement > 300h with 60% of  $H_{sat}$ .

- Observed Initiation phases (PIIP) of sensor, incipiently showing arbitrary parasitic behaviour. To be avoided by post preparation T-t-procedures to reach initial identical balance conditions.
- Stable behaviour for several hundred hours (slight V increase caused by H loss).
- No corrosive affects on Nb-measure heads: functional and structural materials not disabled!

## ECHSLL materials: membranes, electrolyte, reference

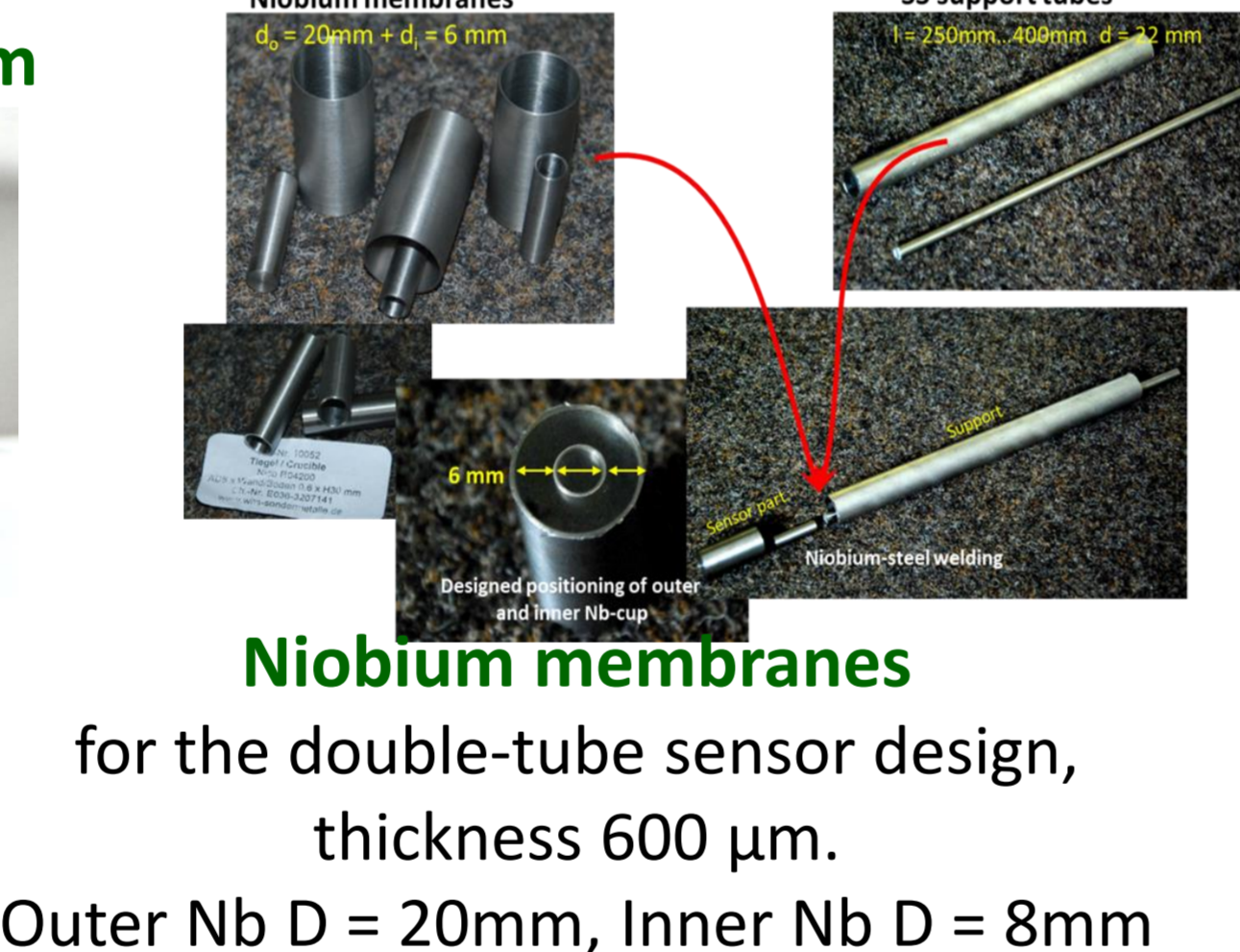
### Reference electrode system

Li + LiH = 1:1  
Supersaturated lithium



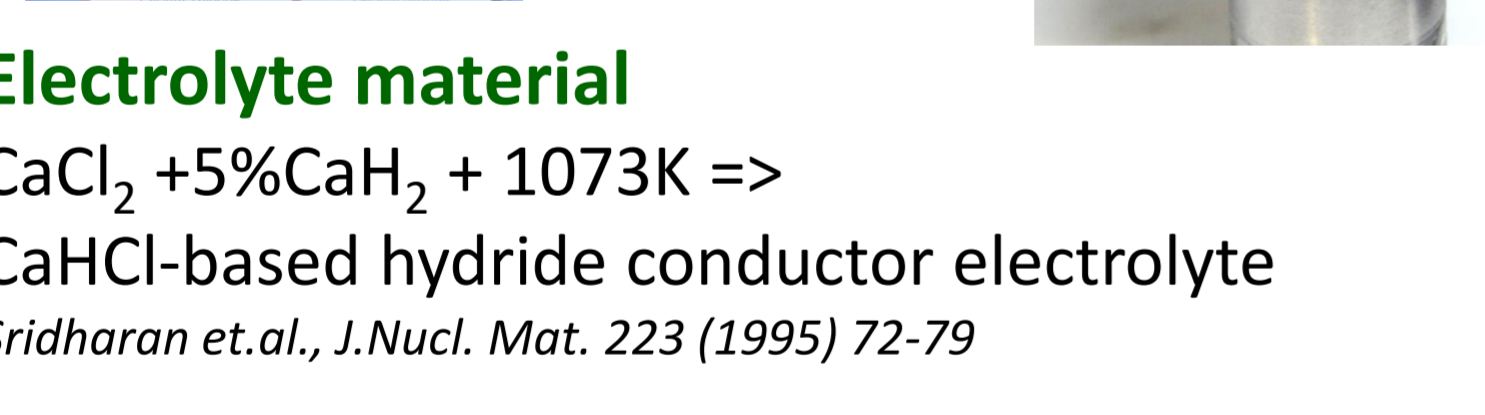
### Niobium membranes

for the double-tube sensor design, thickness 600  $\mu m$ .  
Outer Nb D = 20mm, Inner Nb D = 8mm



### Electrolyte material

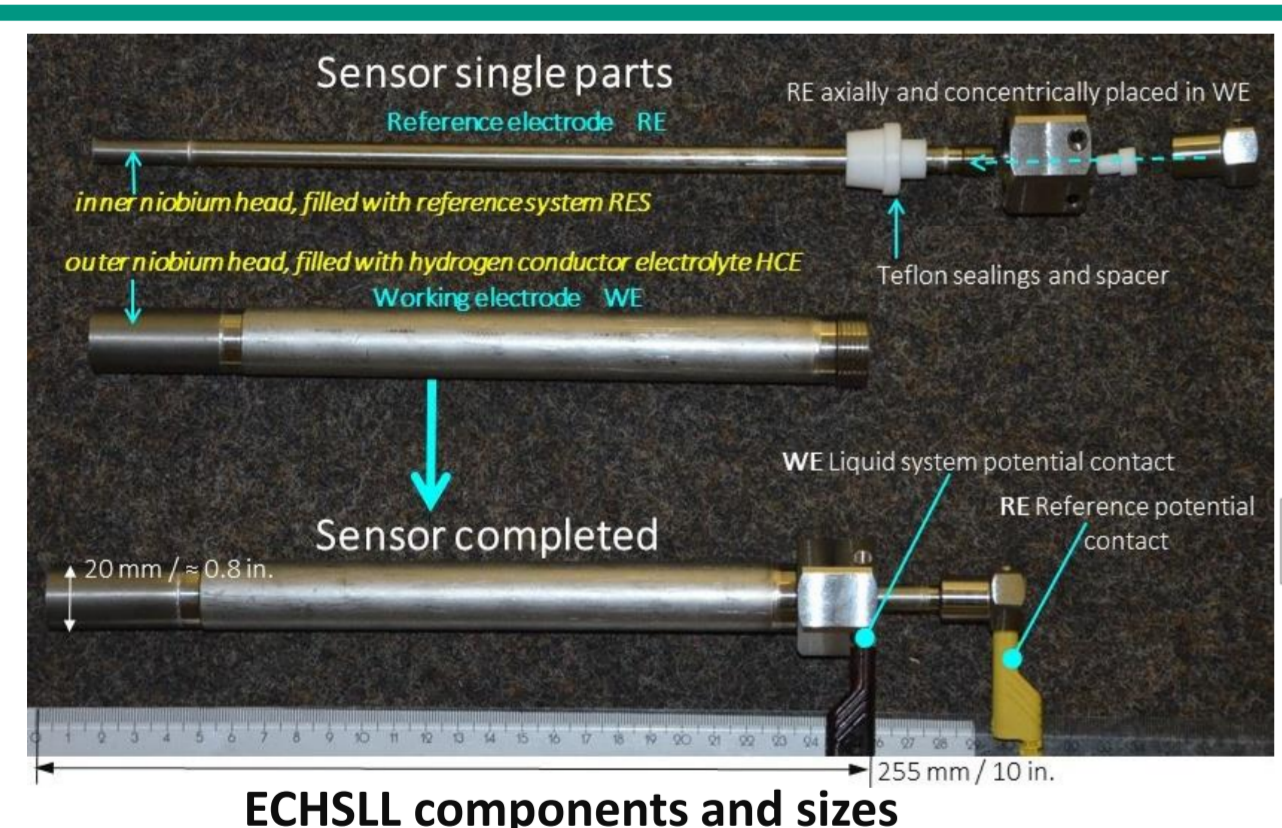
$CaCl_2 + 5\%CaH_2 + 1073K \Rightarrow$   
CaHCl-based hydride conductor electrolyte  
Sridharan et.al., J.Nucl. Mat. 223 (1995) 72-79



## ECHSLL-set-ups

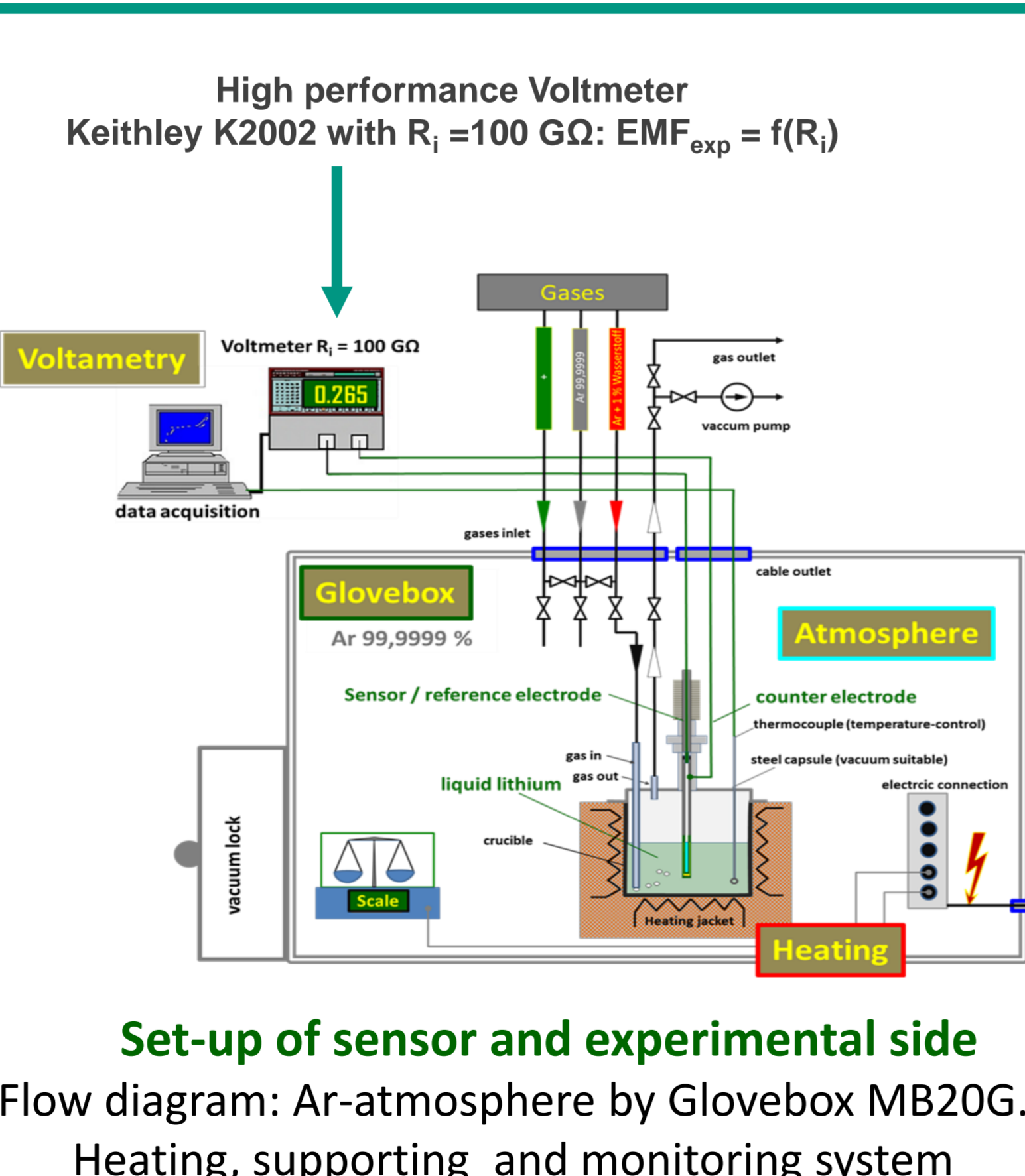
### Sensor single parts

Reference electrode RE  
Working electrode WE



### High performance Voltmeter

Keithley K2002 with  $R_i = 100 \text{ G}\Omega$ :  $EMF_{exp} = f(R_i)$



Set-up of sensor and experimental side  
Flow diagram: Ar-atmosphere by Glovebox MB20G.  
Heating, supporting and monitoring system

## Conclusions and outlook

### The developed design of the ECHSLL and the quantification of sensor

- As membrane metal was selected (due to harsh limiting conditions): Niobium
- Selected Ca-chloride based supporting electrolyte acts as solid-electrolyte.
- Internal RE double-tube sensor system: single-rod measuring cell.  
Fixed defined electrode distances; practically handling under GB conditions.  
Manufacturing of bigger sensor quantities
- Results of measurements in accordance with calculations.
- Very good long-term stability;
- Effects of Y (to reduce H content/measuring time constants) in work;
- Outlook: - thinner membrane materials (300-400  $\mu m$ )  
- tests of further liq.Li-protection materials (Ti as  $N_2$ -getter)

## Acknowledgment

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The views and opinions expressed herein do not necessarily reflect those of the European Commission.