

# NUTECH-2020

Warsaw, 4-7 October 2020

International Conference on Development and  
Applications of Nuclear Technologies



Mirco Grosse



## Thermal limits of potential ATF cladding materials

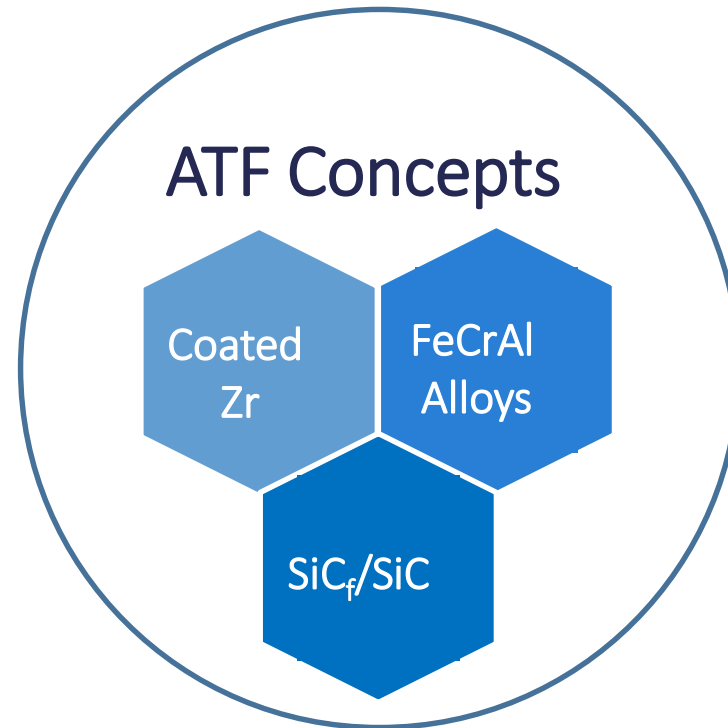


# Promising ATF concepts

Various coating systems:

- Metals
- Oxides or silicates
- MAX phases

Cr coated Zry is the most promising system



less developed system

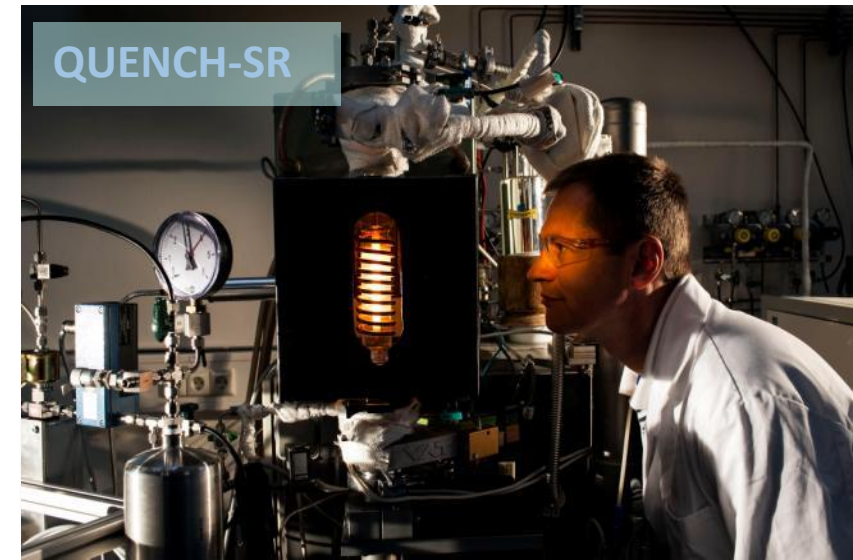
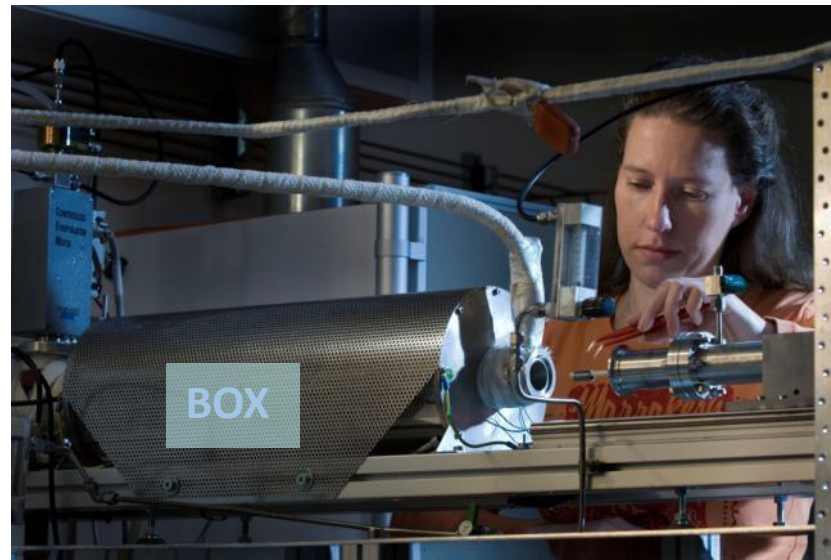
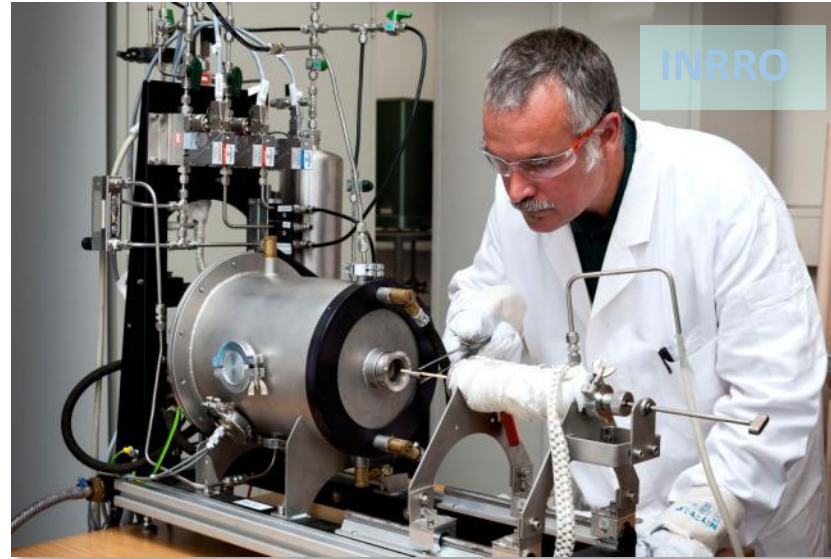
best HT properties

- issues:
- water corrosion
  - joining

Cr in FeCrAl is boon and bane at once:

- irradiation induced embrittlement due to the formation of Cr rich  $\alpha'$  precipitates
- Increase of the corrosion and HT oxidation resistance

# Several facilities for high and very high temperature tests



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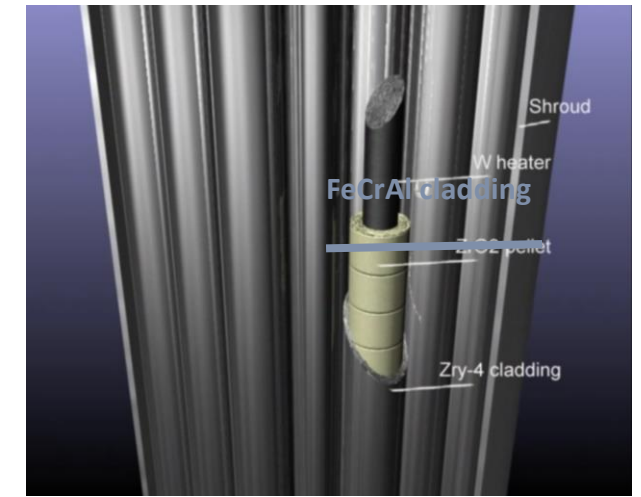
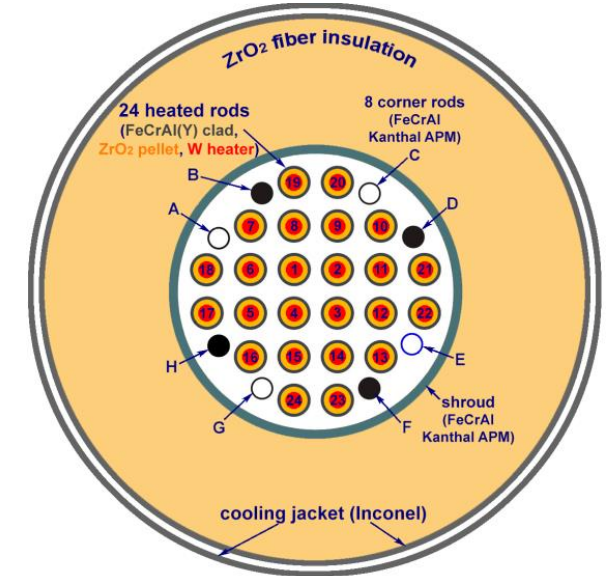
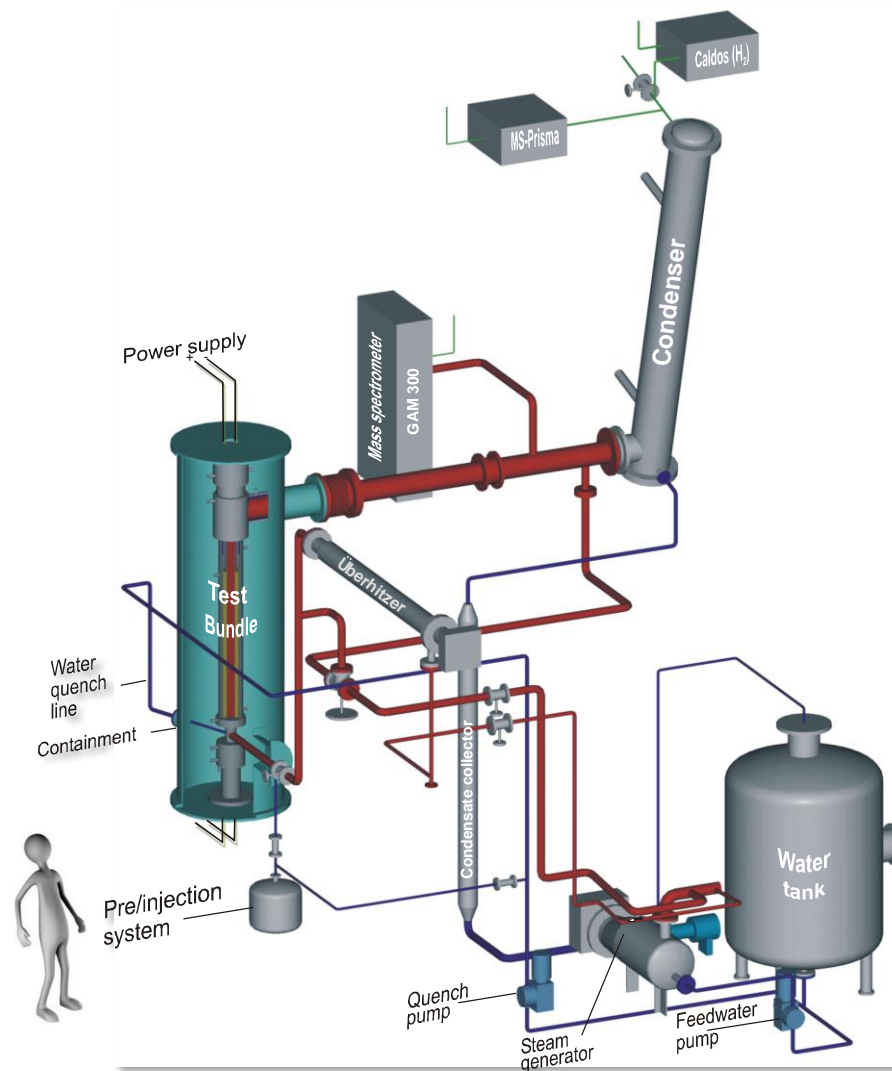
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# The QUENCH facility for bundle tests

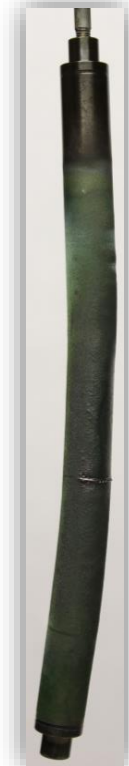
- Bundle with 21 - 32 fuel rod simulators of ~2.5 m length
- Electrically heated: ~2 m;
- Fuel simulator: ZrO<sub>2</sub> pellets
- Quenching from bottom
- Off-gas analysis by mass spectrometer (H<sub>2</sub>, steam ...)
- Extensive instrumentation for T, p, flow rates, water level, etc. (140 measurement channels)
- corner rods, can be removed during test



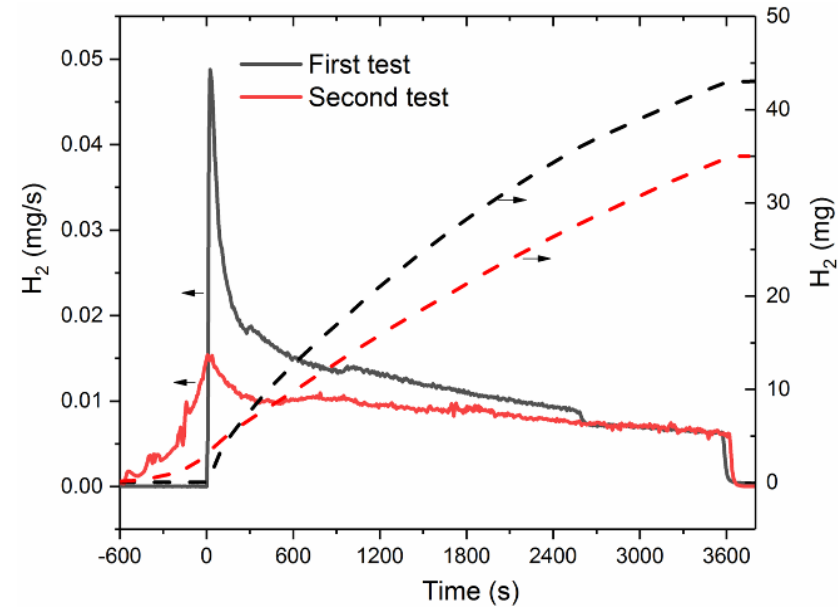
# Results

# Issues of chromium coatings at Zircalloys

# I) bending

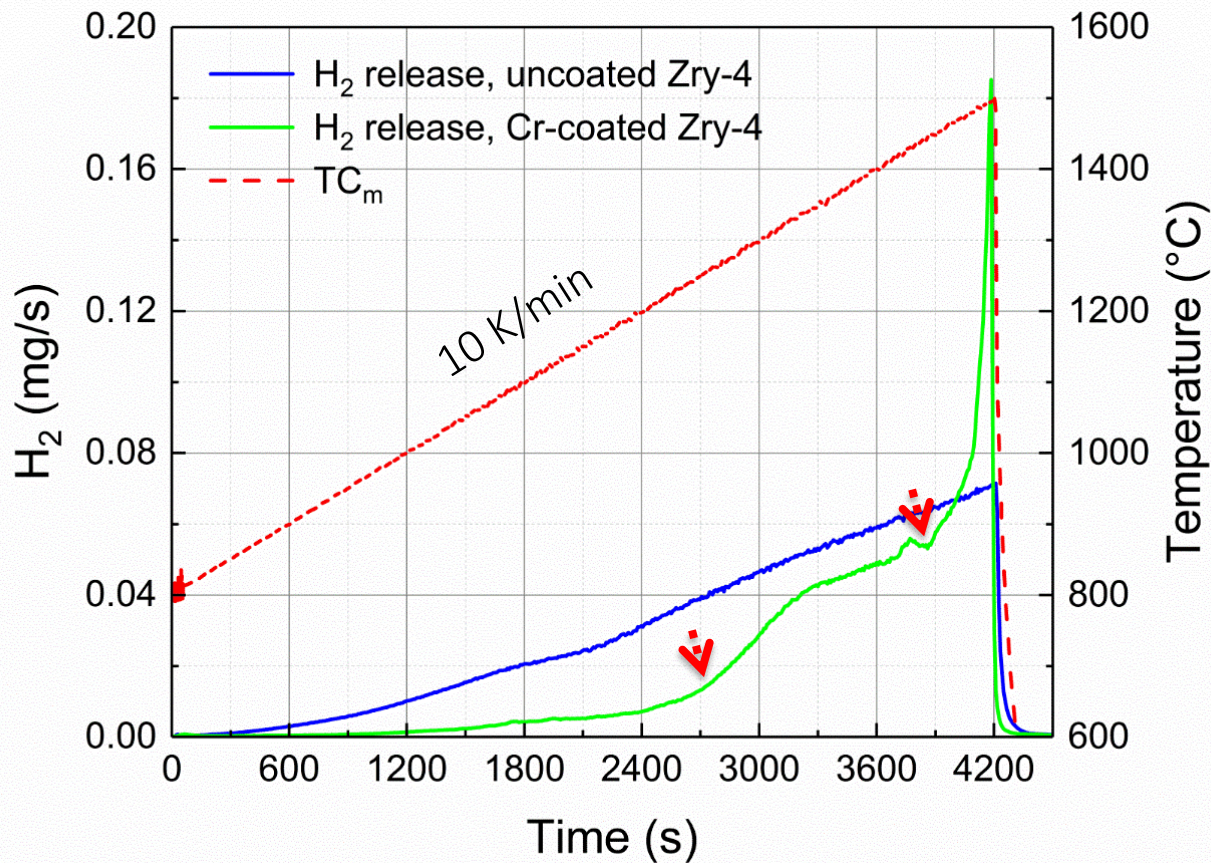


Repeated test  
Pre-oxidation from 800°C  
Same level bending  
First test



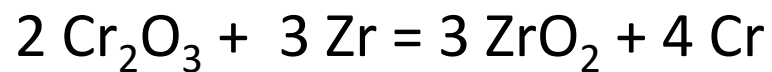
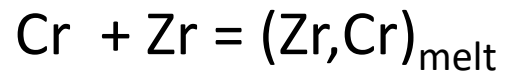
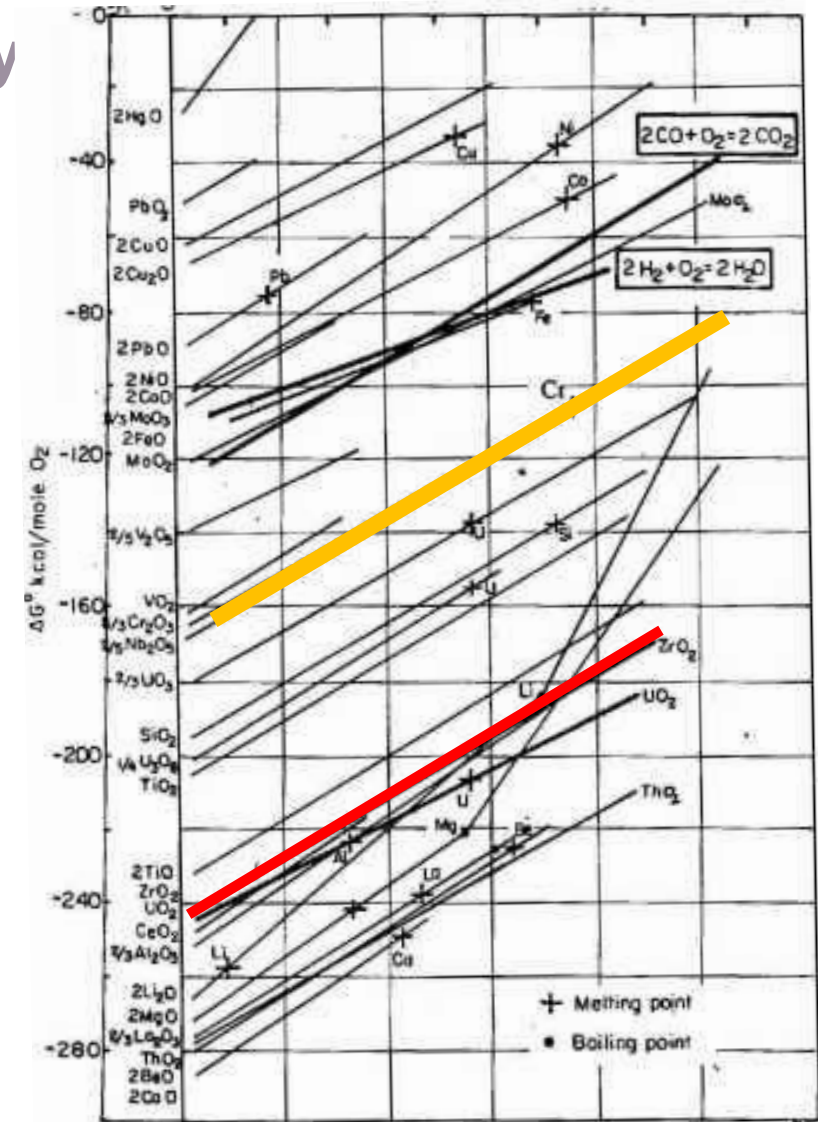
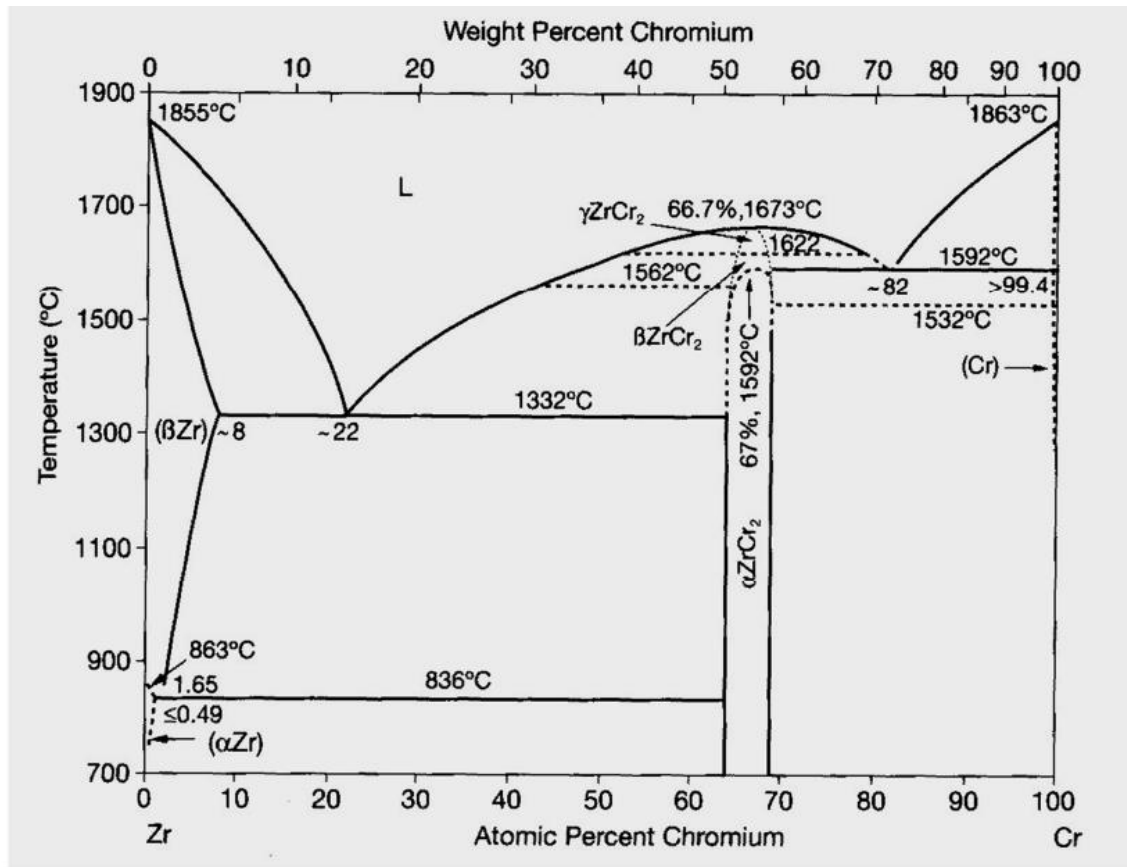
- Bending: High-level of residual stress and stress concentration (inhomogeneity) via cold spraying
- Local variations in the coating thickness

## Transient test from 800 to 1500°C

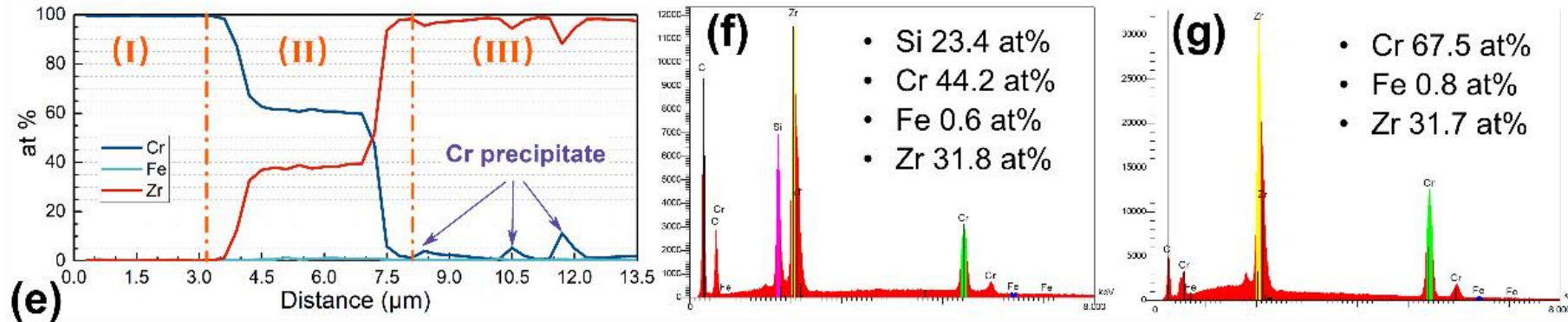
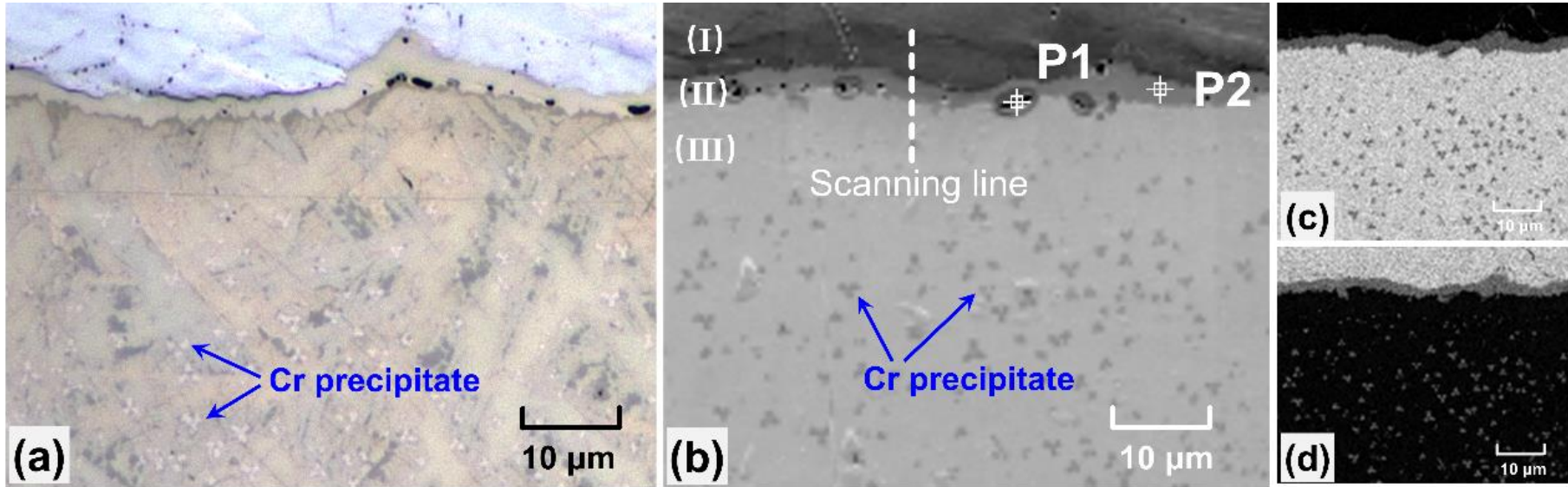


- The hydrogen release rate accelerated at temperature around 1250°C.
- A steeply increased hydrogen release rate above approx. 1410°C, then surpassed that of the uncoated reference Zircaloy-4.

# Results Issues of chromium coatings at Zircaloy



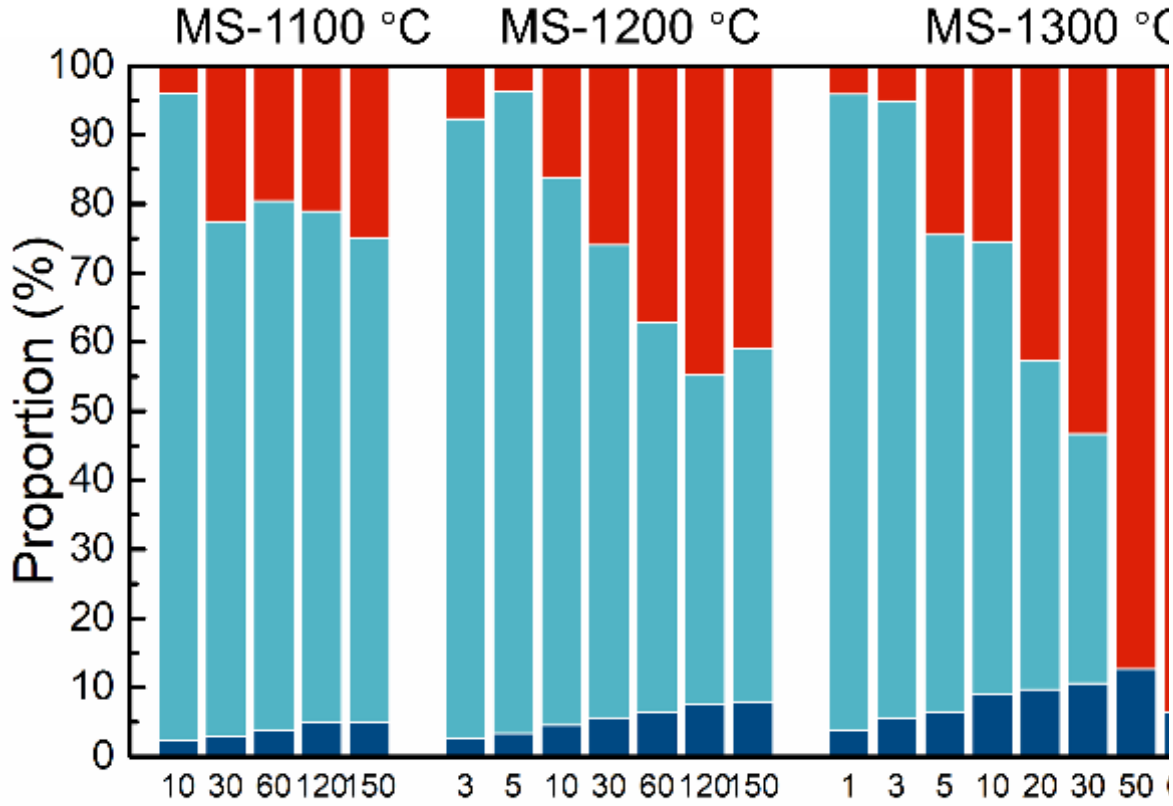
# Results      Dissolution of the Cr coating



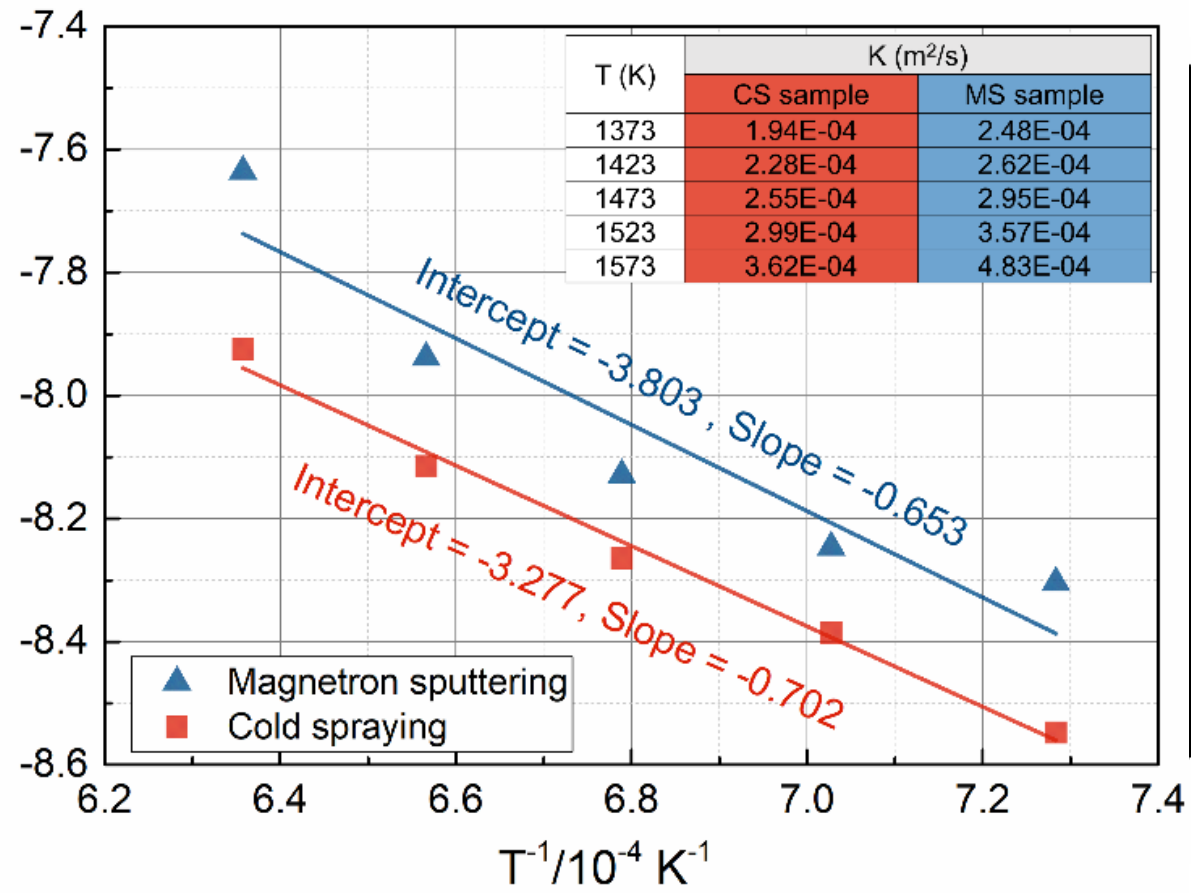


# Results

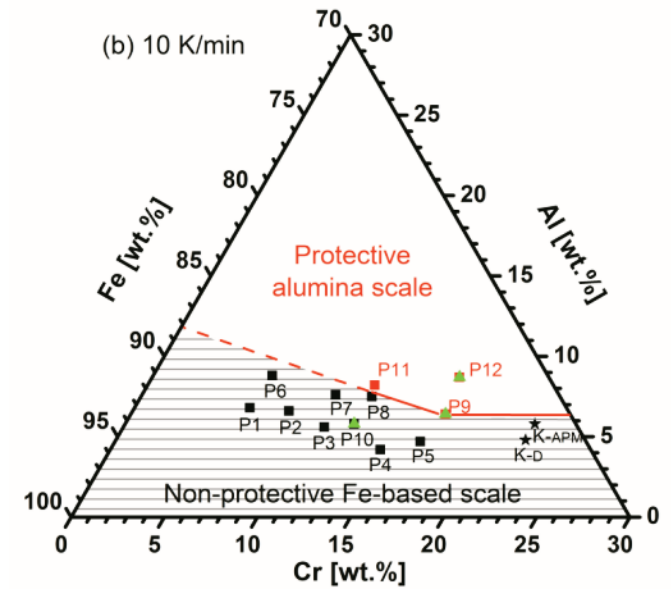
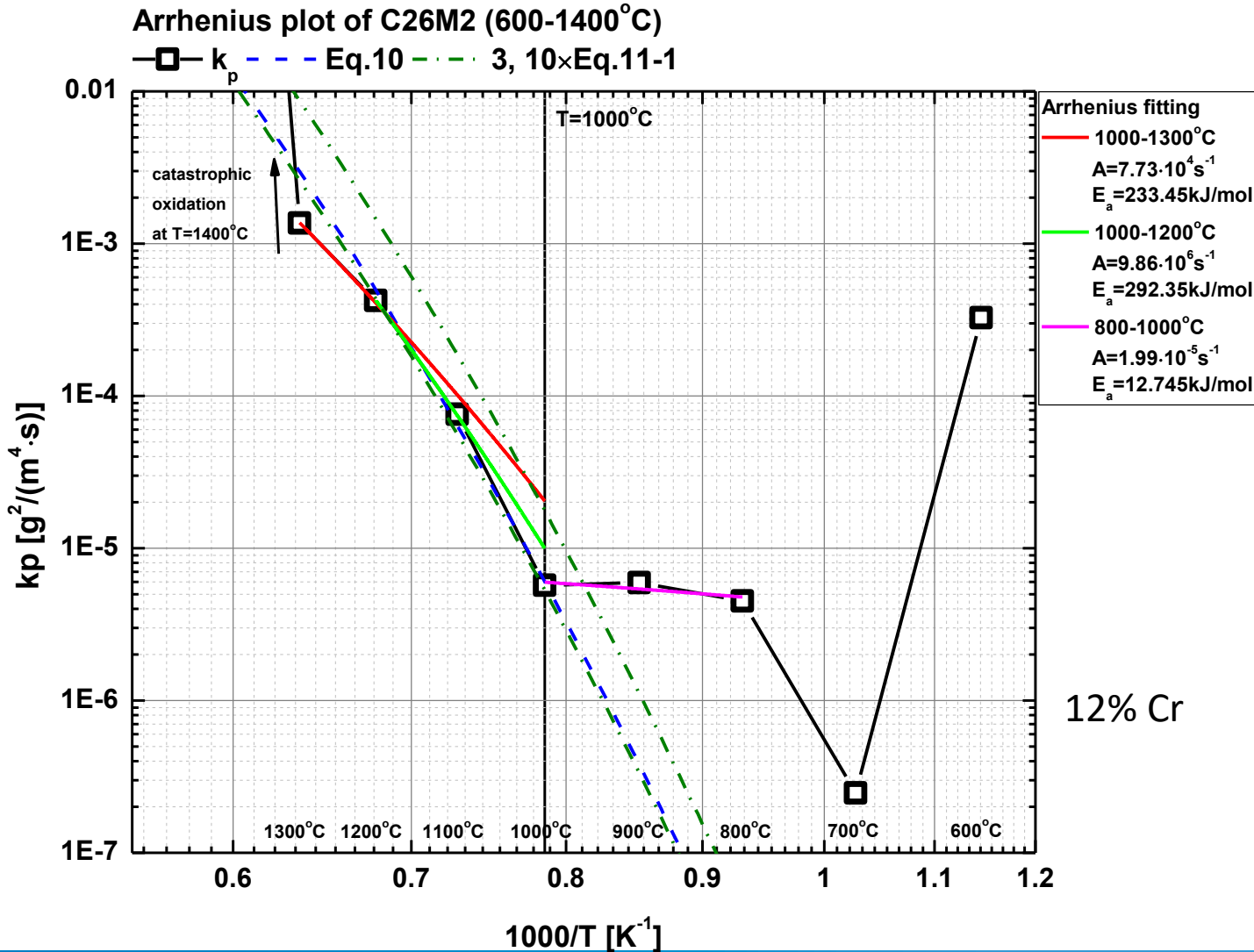
# Dissolution of the Cr coating



MS = Magnetron Sputtering coating thickness ~13 μm  
 CS = Cold Spraying coating thickness ~31 μm



# Results Issues of FeCrAl I) kinetic limitations



Oxidation behaviour of FeCrAl-based alloys during the transient tests from 500 to 1450°C with subsequent holding 1 h at 1450°C in steam. ■ and ★: alloy specimens showed catastrophic oxidation with non-protective Fe-based oxide scale

■ and ★: alloy specimens formed protective alumina scale

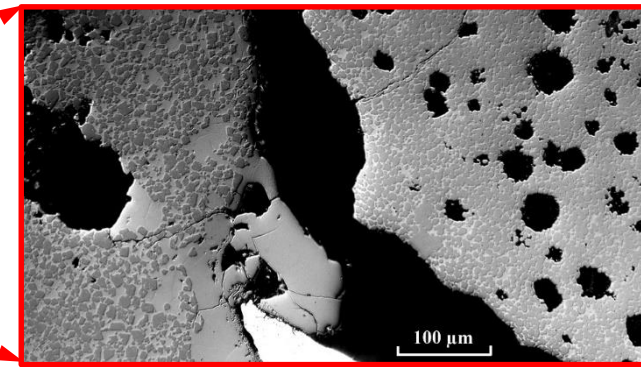
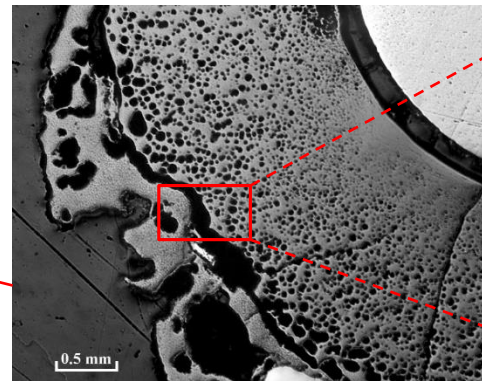
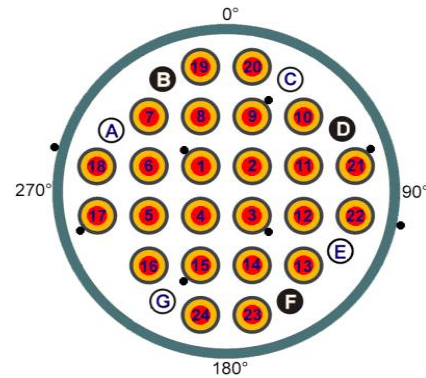
▲: alloy specimens with Y addition formed protective alumina scale

# Results

## Issues of FeCrAl II) eutectic interactions

### QUENCH-19 test with FeCrAl claddings

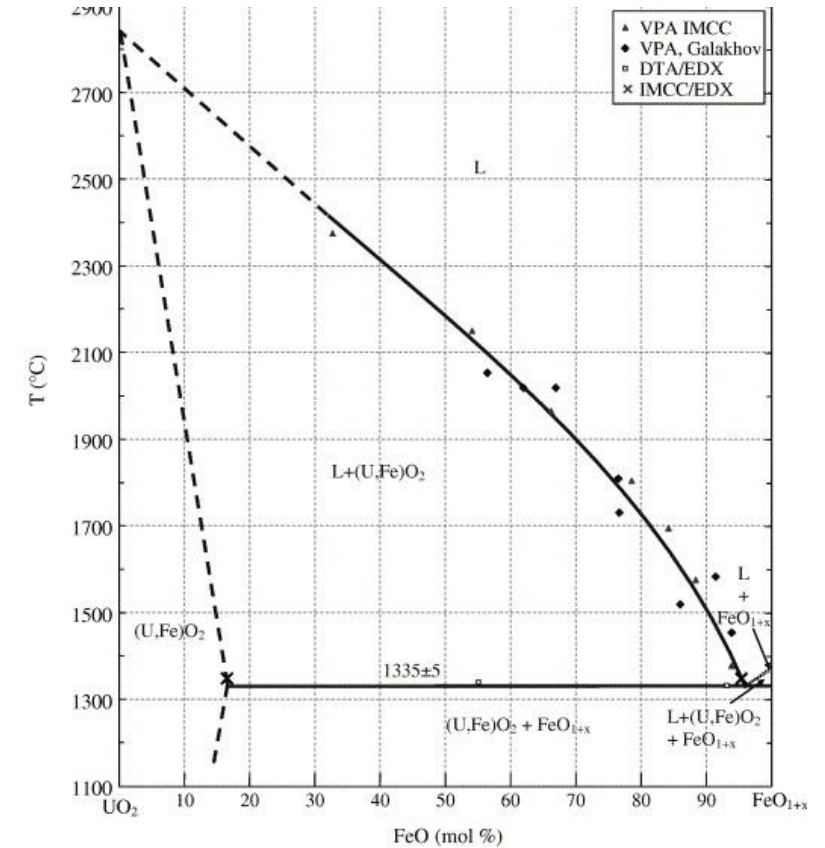
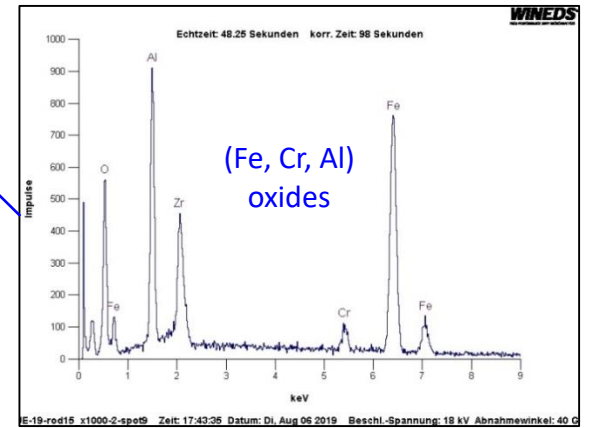
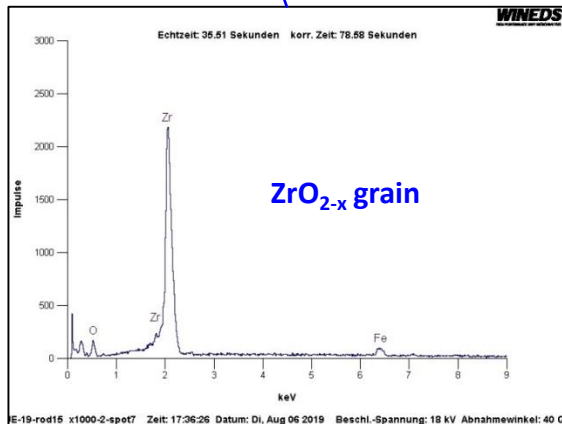
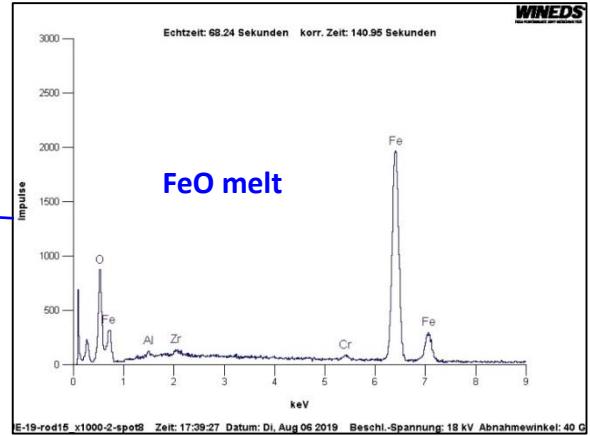
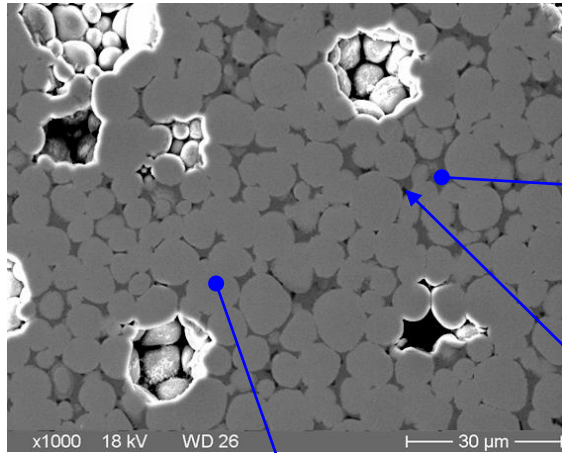
### QUENCH-19, 950 mm, rod 2: metallographic investigations, optical observation



frozen cladding melt

melt-interacting  
ZrO<sub>2</sub> pellet

QUENCH-19, 950 mm, rod 2: SEM/EDX analysis of pellet



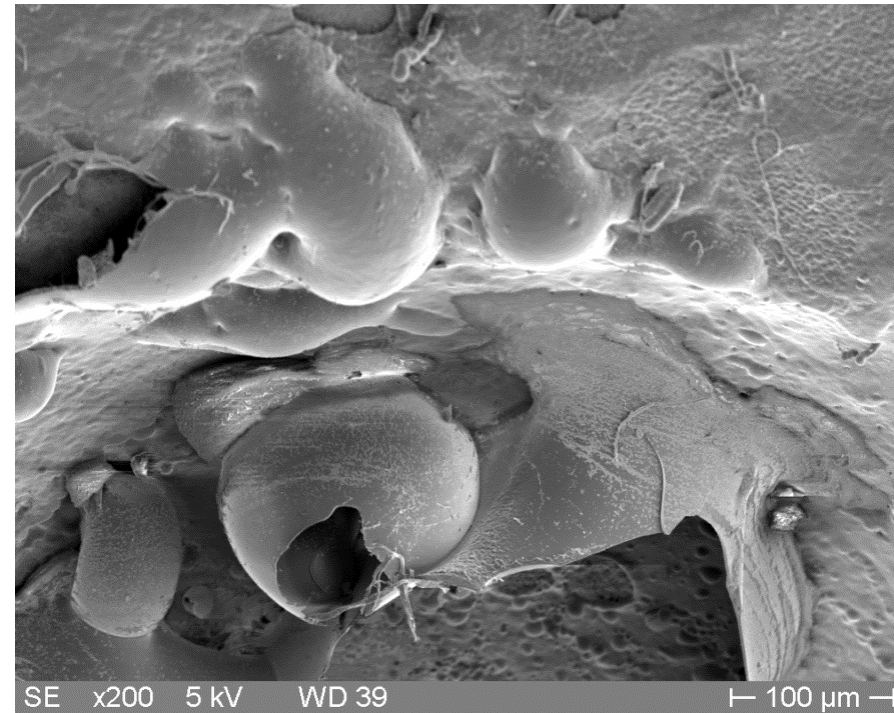
## Results

## Issues of SiC

## I) accelerated oxidation at $T > 1730^{\circ}\text{C}$



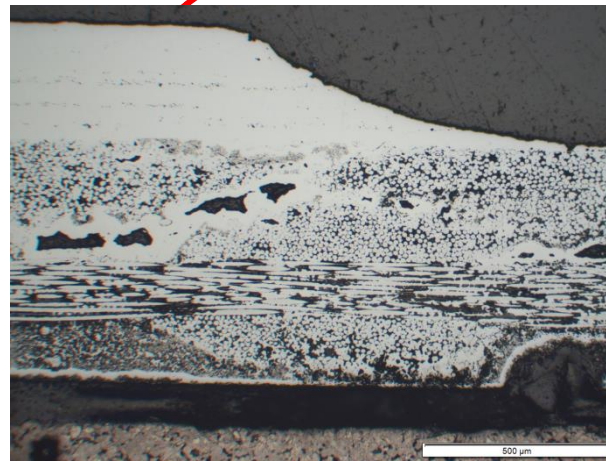
Bubble formation at  $T > 1713^{\circ}\text{C}$  (melting point of  $\text{SiO}_2$ )



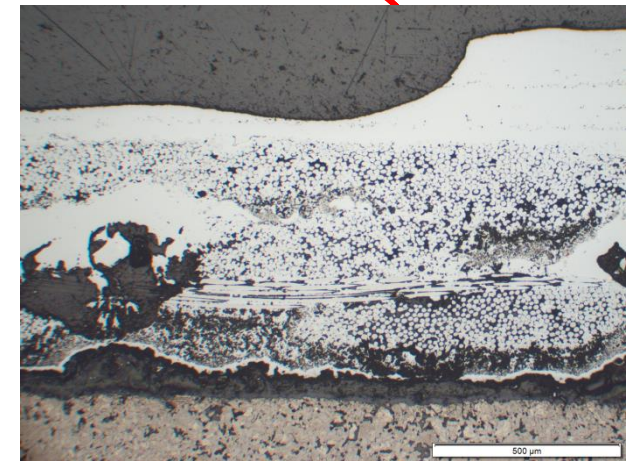
# Results

## Issues of SiC-CMC

II) very large reacting surface if steam can penetrate the fibers



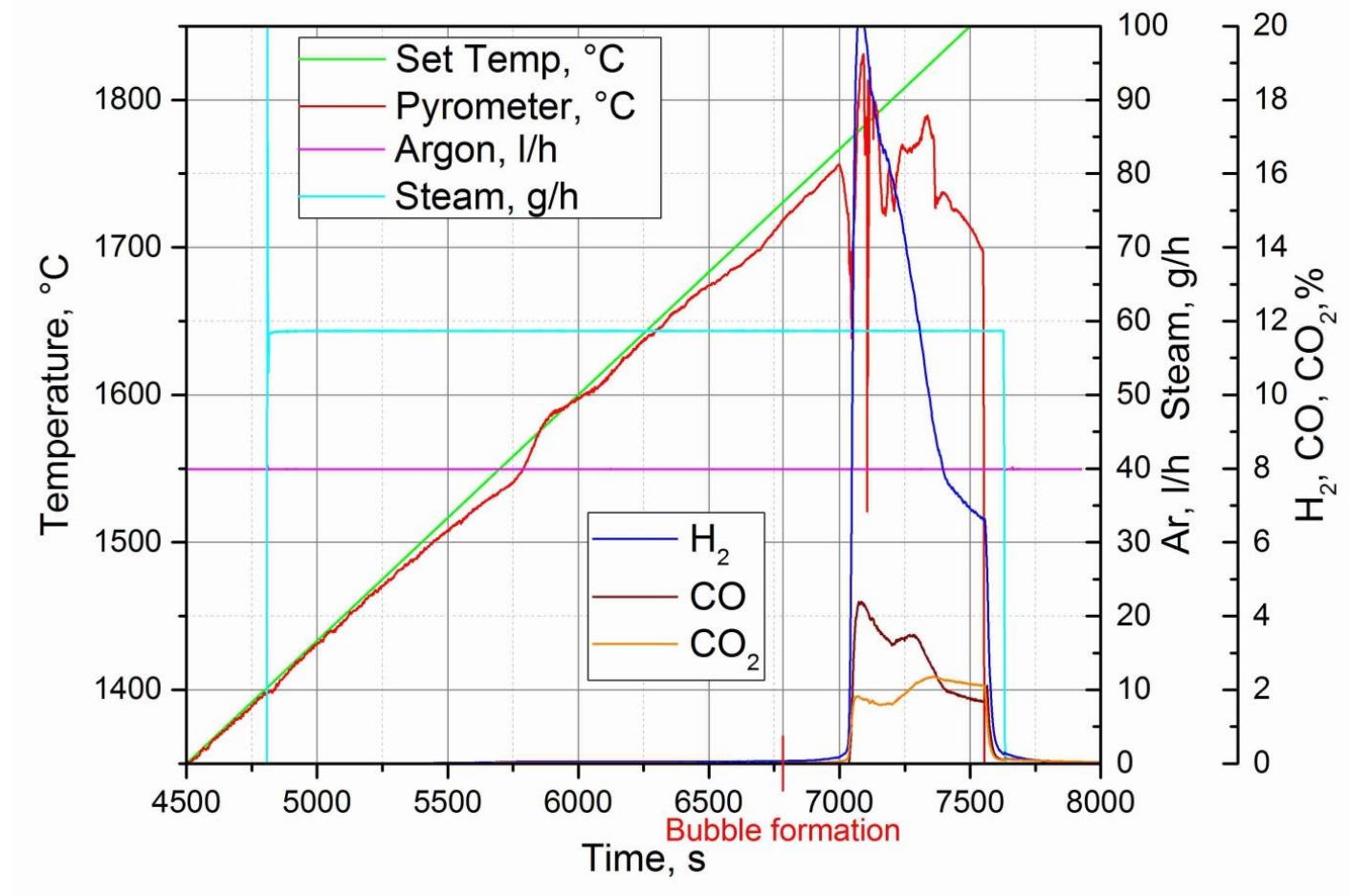
Rapid increase of the reacting surface at 1750°C when the monolithic outer layer is consumed.



# Results

## Issues of SiC-CMC

II) very large reacting surface if steam can penetrate the fibers



## Results

## Issues of SiC-CMC

II) very large reacting surface if steam can penetrate the fibers





## Summary and Conclusions

- ❖ The maximum temperatures for applications are often less than a-priori expected.
- ❖ Cr-coated Zry-4: eutectic interaction between Cr and Zr results in a higher oxidation rate at temperatures above 1330°C. Catastrophic oxidation if the coating fails.
- ❖ FeCrAl:
  - ❖ Eutectic interaction between molten FeO and  $UO_2$  at 1335°C can result in much earlier core melting.
  - ❖ Material performance depends on heating rate. Time is needed to form a protective  $Al_2O_3$  layer. Catastrophic oxidation occurs above 1330°C if no protective  $Al_2O_3$  layer is formed.
- ❖ SiC: faster oxidation above the melting temperature of  $SiO_2$  at 1720°C. Catastrophic oxidation occurs if the monolithic outer layer fails (even at 1200°C).
- ❖ **A lot of work has to be done to achieve accident tolerance. However, it is worth to do this work because the improvement of safety would be great.**

