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## LETTER

## Sources of nitrous oxide from intensively managed pasture soils: the hole in the pipe

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Supplementary material for this article is available [online](#)

**Abstract**

Rainfall and irrigation trigger large pulses of the powerful greenhouse gas N<sub>2</sub>O from intensively managed pastures, produced via multiple, simultaneously occurring pathways. These N<sub>2</sub>O pulses can account for a large fraction of total N<sub>2</sub>O losses, demonstrating the importance to determine magnitude and source partitioning of N<sub>2</sub>O under these conditions. This study investigated the response of different pathways of N<sub>2</sub>O production to wetting across three different textured pasture soils. Soil microcosms were fertilised with an ammonium nitrate (NH<sub>4</sub>NO<sub>3</sub>) solution which was either single or double <sup>15</sup>N labelled, wetted to four different water-filled pore space (WFPS) levels, and incubated over two days. The use of a <sup>15</sup>N pool mixing model together with soil N gross transformations enabled the attribution of N<sub>2</sub>O to specific pathways, and to express N<sub>2</sub>O emissions as a fraction of the underlying N transformation. Denitrification and nitrification mediated pathways contributed to the production of N<sub>2</sub>O in all soils, regardless of WFPS. Denitrification was the main pathway of N<sub>2</sub>O production accounting for >50% of cumulative N<sub>2</sub>O emissions even at low WFPS. The contribution of autotrophic nitrification to N<sub>2</sub>O emissions decreased with the amount of wetting, while the contribution of heterotrophic nitrification remained stable or increased. Following the hole-in-the-pipe model, 0.1%–4% of nitrified N was lost as N<sub>2</sub>O, increasing exponentially with WFPS, while the percentage of denitrified N emitted as N<sub>2</sub>O decreased, providing critical information for the representation of N<sub>2</sub>O/WFPS relationships in simulation models. Our findings demonstrate that the wetting of pasture soils promotes N<sub>2</sub>O production via denitrification and via the oxidation of organic N substrates driven by high carbon and N availability upon wetting. The large contribution of heterotrophic nitrification to N<sub>2</sub>O emissions should be considered when developing N<sub>2</sub>O abatement strategies, seeking to reduce N<sub>2</sub>O emissions in response to rainfall and irrigation from intensively managed pastures.

**1. Introduction**

Pasture soils are a major source of nitrous oxide (N<sub>2</sub>O), a powerful greenhouse gas with a global warming potential 298 times higher than that of carbon dioxide (CO<sub>2</sub>) (Myhre *et al* 2013) and the single most depleting substance of stratospheric

ozone in this century (Ravishankara *et al* 2009). High carbon (C) (Li *et al* 2005, Morley and Baggs 2010) and nitrogen (N) substrate availability (Kim *et al* 2013, Van Lent *et al* 2015) promote emissions of N<sub>2</sub>O from pasture soils. In intensively managed pasture systems such as dairy pastures, N substrate availability for N<sub>2</sub>O production is further increased by N

inputs in the form of urine and dung (Clough *et al* 2020) and N fertiliser (Stott and Gourley 2016). Large pulses of N<sub>2</sub>O from pasture soils are triggered by rainfall (Rowlings *et al* 2015) and irrigation (Mumford *et al* 2019). Rainfall variability in subtropical regions is high (Murphy and Ribbe 2004, Rowlings *et al* 2015), and over the coming decades, pastures in these regions will be subjected to further increasing drying and wetting cycles due to the predicted changes in global climate. The cascade of N transformations triggered by the wetting of dry soil (Borken and Matzner 2009) produces N<sub>2</sub>O via a multitude of different production pathways, fuelled by the sudden increase in soil water content, microbial activity and N substrate availability. Predictions on the processes contributing and dominating N<sub>2</sub>O productions under these conditions are highly uncertain despite an increasingly well-defined mechanistic understanding of N<sub>2</sub>O production (Bakken and Frostegård 2017, Yoon *et al* 2019), and the representation of N<sub>2</sub>O pulses triggered by wetting in simulation models remains challenging (Bessou *et al* 2010, Fuchs *et al* 2020). These challenges reflect the lack of systematic research including different N<sub>2</sub>O production pathways, the reduction of N<sub>2</sub>O to dinitrogen (N<sub>2</sub>), and underlying gross N transformations in response to wetting pulses.

In the conceptual hole-in-the-pipe (HIP) model (Firestone and Davidson 1989, Zhang *et al* 2015), N<sub>2</sub>O emissions are depicted as the fraction  $RN_2O$  of the underlying N transformation. This conceptual framework is widely used in simulation models such as DAYCENT (Necpálová *et al* 2015), DNDC (Li *et al* 2000), LDNDC (Haas *et al* 2013) and NOE (Hénault *et al* 2005). The size of the pipe represents the rate of the N transformation, while the hole is the respective fraction ( $R$ ) that is emitted as N<sub>2</sub>O, defined by physical and chemical factors such as soil moisture, temperature, and soil pH. The main pools considered are the ammonium (NH<sub>4</sub><sup>+</sup>) and nitrate (NO<sub>3</sub><sup>-</sup>) pool, and the model attributes N<sub>2</sub>O formation to these pools, and the respective N transformations, i.e. nitrification and denitrification. Following the HIP model, N<sub>2</sub>O production pathways via nitrification include the chemical decomposition of hydroxylamine (Heil *et al* 2015) and the reduction of nitrite (NO<sub>2</sub><sup>-</sup>) by autotrophic nitrifiers, i.e. nitrifier-denitrification, (Wrage-Mönnig *et al* 2018), while NO<sub>3</sub><sup>-</sup> is regarded as the sole source pool of N<sub>2</sub>O formation via denitrification. Analogue to the HIP model, stable isotope tracing methods, based on the <sup>15</sup>N labelling of the NH<sub>4</sub><sup>+</sup> and/or NO<sub>3</sub><sup>-</sup> pool have been widely used to quantify N gross transformation rates (Kirkham and Bartholomew 1954, Müller *et al* 2004) and associated N<sub>2</sub>O emissions (Müller *et al* 2014) based on a two-source model (Stevens *et al* 1997). In contrast to the use of inhibitors (Berg *et al* 1982, Hynes and Knowles 1982), or analysis of the

isotopic composition of N<sub>2</sub>O without the addition of stable isotopes (Decock and Six 2013, Yu *et al* 2020), this approach does not account for specific microbial processes but aligns with the representation of N<sub>2</sub>O production in biogeochemical models, recommending its use to establish the contribution of specific N transformation to N<sub>2</sub>O production.

Two pathways of N<sub>2</sub>O production linked to organic N pool have recently gained more attention due to the inclusion of the organic N pool in <sup>15</sup>N<sub>2</sub>O tracing models: heterotrophic nitrification of organic N (Zhang *et al* 2015) and co-denitrification (Clough *et al* 2017, Rex *et al* 2019). The formation of N<sub>2</sub>O via heterotrophic nitrification is thought to occur via the oxidation of organic N to NO<sub>2</sub><sup>-</sup> and its subsequent reduction to N<sub>2</sub>O (Braker and Conrad 2011). Although heterotrophic nitrifiers can use a wide range of substrates including NH<sub>4</sub><sup>+</sup> (Stein 2011), we refer to this pathway in the context of <sup>15</sup>N source partitioning as N<sub>2</sub>O production from organic N compounds only. Besides classic denitrification, co-denitrification can also contribute to N<sub>2</sub>O production, forming hybrid N<sub>2</sub>O by combining an inorganic N compound such as NO<sub>2</sub><sup>-</sup> with a co-metabolised organic N-substrate (Spott *et al* 2011).

Source partitioning of N<sub>2</sub>O from temperate pastures has demonstrated the significance of both heterotrophic nitrification (Müller *et al* 2014, Jansen-Willems *et al* 2016, Moser *et al* 2018) and co-denitrification (Selbie *et al* 2015, Rex *et al* 2019) for N<sub>2</sub>O production, yet their significance in response to different degrees of wetting remains largely unknown. More importantly, assessments of N<sub>2</sub>O production pathways in response to soil water content (Bateman and Baggs 2005, Mathieu *et al* 2006, Loick *et al* 2021) mostly lack data on N<sub>2</sub>O reduction to N<sub>2</sub>. This hinders the quantification of overall denitrification, and thus the assessment of denitrification as source and sink of N<sub>2</sub>O upon wetting.

The aim of this study was therefore to establish the response and significance of nitrification and denitrification mediated pathways of N<sub>2</sub>O production across three different textured pasture soils exposed to different degrees of wetting, and to quantify what fraction of the underlying N transformation is emitted as N<sub>2</sub>O. We combined <sup>15</sup>N<sub>2</sub>O analysis and a <sup>15</sup>N<sub>2</sub>O pool mixing model with soil N transformation and N<sub>2</sub> data presented in Friedl *et al* (2018), allowing the calculation of the contribution of nitrification and denitrification (N<sub>2</sub> + N<sub>2</sub>O) emitted as N<sub>2</sub>O following the HIP model, and to derive the response curves of these fractions across different soil water contents. As such, this study addresses a major uncertainty in biochemical models simulating the N cycle: the fraction of N<sub>2</sub>O emitted from denitrification ( $RN_2O_d$ ), the magnitude of overall denitrification (Del Grosso *et al* 2020), and the fraction of N<sub>2</sub>O

**Table 1.** Selected soil characteristics (0–10 cm) for three intensively managed pasture sites under dairy production in subtropical Australia.

Soil property	Clay	Loam	Sandy clay loam
Site	Casino	Gympie	Kerry
Latitude	−28.865	−26.19	−28.109
Longitude	152.874	152.74	153.031
Mean annual rainfall	1107 mm	1127 mm	906.7 mm
Soil type (ASC)	Vertosol	Dermosol	Tenosol
Soil type (FAO)	Pellic Vertisol	Ferric Acrisol	Mollic Fluvisol
Texture (USDA)	Clay	Loam	Sandy clay loam
pH (water, 1:5)	6.3	6.1	5.9
Organic Carbon (%)	4.2	4.9	4.1
Total Nitrogen (%)	0.36	0.5	0.4
C:N ratio	11.4	9.8	10.4

emitted via nitrification mediated pathways ( $RN_2O_n$ ) (Chen *et al* 2008).

## 2. Material and methods

Soil samples (0–10 cm) were collected from three intensively managed dairy pastures in subtropical Australia. Emissions of  $N_2O$  from these pasture sites were previously quantified in both laboratory-based (Friedl *et al* 2016, 2020) and field-based experiments (Friedl *et al* 2017, Mumford *et al* 2019, De Rosa *et al* 2020). The site location and characteristics, including physical and chemical soil properties, are shown in table 1. The soils were classified as pellic Vertisol, ferric Acrisol and mollic Fluvisol, respectively (IUSS Working Group 2015), and are henceforth referred to as clay, loam and sandy clay loam (sandy CL), according to their texture from 0 to 10 cm. The organic C content of the soils ranges from 4.1% to 4.9%, following the order sandy CL < clay < loam (table 1). The soil pH measured in water (1:5, v:w) is 6.3, 6.1 and 5.9 for the clay, the loam, and the sandy CL, respectively.

### 2.1. $^{15}N$ tracing experiment

The experiment was set up in a full factorial design with four different water-filled pore space (WFPS) levels across three different textured pasture soils and four replicates using a triple  $^{15}N$  labelling approach combined with a  $^{15}N$  tracing model (Friedl *et al* 2018). In the study presented here, we used additional  $^{15}N_2O$  gas analysis together with gross N transformation data presented in Friedl *et al* (2018) to attribute  $N_2O$  losses to specific  $N_2O$  production pathways.

The experimental setup is described in Friedl *et al* (2018). Briefly, soil collected from the three pasture sites was partially airdried (10% gravimetric water content) and sieved to 4 mm. Soil microcosms were established in 50 ml centrifuge tubes using the equivalent of 8 g oven-dry soil. One milliliter of  $NH_4NO_3$  solution containing the equivalent of 35  $\mu g N g^{-1}$  soil was applied to each microcosm, either single ( $NH_4^{15}NO_3$ ) labelled (a) or double ( $^{15}NH_4^{15}NO_3$ )

labelled (b) at 10 atom %. Soil microcosms were wetted to 40, 60, 80 and 95% WFPS and the soil was compacted to a volume of 8 ml using a plunger, resulting in an adjusted bulk density of 1  $g cm^{-3}$ . Homogenous labelling was ensured by applying water and fertiliser solution dropwise on two layers of 4 g of soil. Centrifuge tubes were then closed with suba-seals (Sigma Aldrich) and kept closed in an incubator at a constant temperature of 25 °C between gas sampling events.

### 2.2. Gas samples

Ambient background air samples ( $n = 4$ ) were taken each day before closing the centrifuge tubes. The headspace atmosphere of treatment (a) and (b) was sampled 24 and 48 h after closure using a gastight syringe. The suba-seals were removed after the first sampling event for 10 min to allow gas exchange with the headspace atmosphere. Gas samples were transferred into pre-evacuated 12 ml Exetainer tubes with a double wadded Teflon/silicon septa cap (Labco Ltd, Buckinghamshire, UK) and stored until  $N_2O$  and  $CO_2$  analysis by gas chromatography (Shimadzu GC-2014). Gas samples were analysed for isotopologues of  $N_2O$  ( $^{14}N^{14}N$ ,  $^{14}N^{15}N$  and  $^{15}N^{15}N$ , and  $^{14}N^{14}N^{16}O$ ) using an automated isotope ratio mass spectrometer (IRMS) (Sercon Limited, 20–20, UK). Due to instrument malfunction, headspace samples from the sandy clay loam from treatment (a) and (b) at 95% WFPS were lost and could not be analysed for  $^{15}N_2O$ .

### 2.3. Gross N transformations

Data on gross N transformations were derived from Friedl *et al* (2018). In brief: soil mineral N was extracted using 40 ml of 2 M KCl 30 min ( $t = 0$ ) and 48 h ( $t = 2$  d) after fertiliser application. Four additional soil microcosms per  $^{15}N$  treatment were used for  $t = 0$  extractions, and all soil microcosms (treatment a and b) were extracted at  $t = 2$  d. The  $^{15}N$  enrichment of the soil mineral N pool was determined by the diffusion method (Stark and Hart 1996). Gross N transformations were quantified using the  $^{15}N$  tracing tool Ntrace (Müller *et al* 2007, Zaman *et al* 2021), using a Markov Chain Monte Carlo method to optimise the fit of the data to the conceptual N cycling model

shown in figure S1 (available online at [stacks.iop.org/ERL/16/065004/mmedia](https://stacks.iop.org/ERL/16/065004/mmedia)). This model comprises five N pools linked by ten different N transformations. Autotrophic nitrification is defined as the oxidation of  $\text{NH}_4^+$  to  $\text{NO}_3^-$  ( $O_{\text{NH}_4}$ ) and heterotrophic nitrification as the oxidation of recalcitrant organic N to  $\text{NO}_3^-$  ( $O_{\text{Nrec}}$ ), given in  $\mu\text{g N g}^{-1}$  soil.

#### 2.4. Fluxes of $\text{N}_2\text{O}$ and $\text{CO}_2$ , and $\text{N}_2\text{O}$ source partitioning

Fluxes of  $\text{N}_2\text{O}$  and  $\text{CO}_2$  were determined using the closed chamber method and calculated from the slope of the assumed linear increase in gas concentration during the closure period, corrected for temperature and air pressure (Scheer et al 2014). The  $^{15}\text{N}$  fraction of  $\text{N}_2\text{O}$  was then used to attribute  $\text{N}_2\text{O}$  production to source pools of a specific  $^{15}\text{N}$  enrichment using a pool mixing model. The model assumes uniform  $^{15}\text{N}$  labelling of the soil mineral N pools and negligible isotopic discrimination for all  $\text{N}_2\text{O}$  production pathways.

IRMS analysis of the headspace samples of treatment (a) and (b) at  $t = 24$  h and  $t = 48$  h provided the ion currents ( $I$ ) at  $m/z$  44,45 and 46 enabling the molecular ratios  $^{45}\text{R}$  ( $^{45}\text{I}/^{44}\text{I}$ ) and  $^{46}\text{R}$  ( $^{46}\text{I}/^{44}\text{I}$ ) to be calculated. Possible  $\text{N}_2\text{O}$  species include also  $^{47}\text{R}$  and  $^{48}\text{R}$  and were calculated as

$$^{47}\text{R} = (^{15}\text{R})^2 * ^{17}\text{R} + 2 * ^{15}\text{R} * ^{18}\text{R} \quad (1)$$

$$^{48}\text{R} = ^{18}\text{R} * (^{15}\text{R})^2 \quad (2)$$

with  $^{17}\text{R}$  ( $^{17}\text{O}/^{16}\text{O}$ ) = 0.00037795 and  $^{18}\text{R}$  ( $^{18}\text{O}/^{16}\text{O}$ ) = 0.002079 as  $^{17}\text{R}$  and  $^{18}\text{R}$  are assumed to be at natural abundance.

The  $^{15}\text{N}$  fraction of  $\text{N}_2\text{O}$  is then calculated as

$$a_{\text{N}_2\text{O}} = \frac{(^{45}\text{R} + 2 * ^{46}\text{R} - ^{17}\text{R} - 2 * ^{18}\text{R})}{(2 + 2 * ^{45}\text{R} + 2 * ^{46}\text{R})}. \quad (3)$$

Emissions of  $\text{N}_2\text{O}$  from treatment *a* and *b* were attributed to four different source pathways including three different N pools: denitrification derived from the  $\text{NO}_3^-$  pool at enrichment  $a_d$ , autotrophic nitrification ( $O_{\text{NH}_4}$ ), the oxidation of the  $\text{NH}_4^+$  pool at enrichment  $a_n$ , heterotrophic nitrification ( $O_{\text{Nrec}}$ ), defined as the oxidation of organic N at natural abundance  $a_o$ , and hybrid formation of  $\text{N}_2\text{O}$  attributed to co-denitrification. Hybrid  $\text{N}_2\text{O}$  is formed by one N atom from the  $\text{NO}_3^-$  pool and one from the organic N pool. The  $^{15}\text{N}$  fraction of  $\text{N}_2\text{O}$  is given by equation (4) as

$$a_{\text{N}_2\text{O}} = d * a_d + n * a_n + h * a_o + cd * (0.5 * (a_d + a_o)) \quad (4)$$

where  $a_{\text{N}_2\text{O}}$  is the  $^{15}\text{N}$  enrichment of the respective headspace sample, and  $d$ ,  $n$ ,  $h$  and  $cd$  are the

fraction of  $\text{N}_2\text{O}$  emitted via denitrification, autotrophic nitrification, heterotrophic nitrification and co-denitrification, respectively. Daily values for  $a_d$  and  $a_n$  were calculated as daily average, assuming a linear increase/decrease of the respective  $^{15}\text{N}$  enrichment over the time of the experiment (from  $t = 0$  to  $t = 2$  d). For each headspace sample,  $d$ ,  $n$ , and  $cd$  were quantified using the EXCEL SOLVER (Microsoft Excel 2016) by minimising the absolute difference between measured and calculated  $^{15}\text{N}$  enrichment of  $\text{N}_2\text{O}$ , using all possible combinations between the different replicates of treatment (a) and (b) ( $n = 16$ ). The fraction  $h$  was calculated as

$$h = 1 - d - n - cd \quad (5)$$

with all the  $\text{N}_2\text{O}$  assumed to come from one of the four  $\text{N}_2\text{O}$  production pathways. Average contributions were calculated using the values for  $d$ ,  $n$ ,  $h$  and  $cd$  when the SOLVER solution satisfied all the constraints. The SOLVER function did not find a feasible solution for 18% of all the combinations. Multiplying the  $\text{N}_2\text{O}$  flux with  $d$ ,  $n$ ,  $h$ , and  $cd$  gave the amount of  $\text{N}_2\text{O}$  emitted via denitrification ( $\text{N}_2\text{O}_d$ ), autotrophic nitrification ( $\text{N}_2\text{O}_a$ ), heterotrophic nitrification ( $\text{N}_2\text{O}_h$ ) and co-denitrification ( $\text{N}_2\text{O}_{cd}$ ), respectively. Emissions of  $\text{N}_2\text{O}$  produced by all nitrification mediated pathways ( $\text{N}_2\text{O}_n$ ) were calculated as the sum of  $\text{N}_2\text{O}_a$  and  $\text{N}_2\text{O}_h$ . Following the hole in the pipe model, the fraction of  $\text{N}_2\text{O}$  lost via denitrification was expressed as

$$\text{RN}_2\text{O}_d = \frac{\text{N}_2\text{O}_d}{\text{Den}} \quad (6)$$

where Den was overall denitrification, calculated using the product ratio of denitrification  $\text{N}_2\text{O}/(\text{N}_2 + \text{N}_2\text{O})$  obtained in Friedl et al (2018) (treatment c) and the  $\text{N}_2\text{O}_d$  from treatment (a) and (b). The  $\text{RN}_2\text{O}$  factors for autotrophic nitrification ( $O_{\text{NH}_4}$ ) and heterotrophic nitrification ( $O_{\text{Nrec}}$ ) were calculated accordingly using equations (7) and (8):

$$\text{RN}_2\text{O}_a = \frac{\text{N}_2\text{O}_n}{O_{\text{NH}_4}} \quad (7)$$

$$\text{RN}_2\text{O}_h = \frac{\text{N}_2\text{O}_h}{O_{\text{Nrec}}}. \quad (8)$$

The factor  $\text{RN}_2\text{O}_n$  for total nitrification was calculated as

$$\text{RN}_2\text{O}_n = \frac{\text{N}_2\text{O}_n}{O_{\text{Nrec}} + O_{\text{NH}_4}}. \quad (9)$$

#### 2.5. Statistical analysis

Statistical analyses were conducted with SPSS 22.0 (SPSS Inc. 2013). The effects of soil texture and WFPS on cumulative emissions of  $\text{N}_2\text{O}$ ,  $\text{N}_2\text{O}_d$ ,  $\text{N}_2\text{O}_{cd}$ ,  $\text{N}_2\text{O}_a$ ,  $\text{N}_2\text{O}_h$ ,  $\text{N}_2\text{O}_n$  and  $\text{CO}_2$  were examined by analysis of variance (ANOVA) ( $P < 0.05$ ). Normal distribution

of the data was assessed by the Shapiro–Wilk test for normality. Tukey’s honest significant difference (HSD) test was used to determine differences between pasture soils within a WFPS treatment, and within a pasture soil across different WFPS. The response of cumulative  $N_2O_d$ ,  $N_2O_n$ ,  $N_2O_a$ , and  $N_2O_h$  emissions to WFPS was quantified with generalised additive models (GAM) utilising the R package mgcv (Wood 2015). GAMs are semi-parametric models and can test and quantify the non-linear relationship between response and explanatory variables. The Akaike Information Criterion (AIC) and the deviance explained aided for model selection (Akaike 1974). Due to analytical problems,  $N_2O$  source partitioning for the sandy clay loam at 95% WFPS is missing, and therefore not considered when evaluating differences between soils or the response to different WFPS. The relationship between N transformations and  $RN_2O$  values vs. WFPS was evaluated by regression analysis using SPSS 22.0 and SigmaPlot Version 13.0. Results of the regression analysis, including the best-fit model and parameters for figures 2 and 3 are given in table S1. Values in the text, tables and figures represent means  $\pm$  standard error of the mean.

### 3. Results

#### 3.1. Cumulative $N_2O$ emissions

Cumulative  $N_2O$  emissions ranged from 0.04 to  $>7 \mu\text{g } N_2O\text{-N g}^{-1}$  soil and differed between soil type, following the order clay  $>$  loam  $>$  sandy CL, regardless of WFPS (figure 1, table 2). Emissions of  $N_2O$  increased exponentially with increasing WFPS, except for the loam, where  $N_2O$  emissions increased exponentially up to 80% WFPS and decreased at WFPS  $>$  80%.

#### 3.2. Contribution of different $N_2O$ production pathways to cumulative $N_2O$ emissions

Denitrification, autotrophic nitrification, and heterotrophic nitrification contributed to the production of  $N_2O$  in all soils, regardless of WFPS (table 2, figure 1). Denitrification was the main  $N_2O$  production pathway in the clay and the loam at all WFPS levels and at 40 and 80% WFPS in the sandy CL, accounting for 40%–82% of cumulative  $N_2O$  emissions (table 2). Emissions of  $N_2O_d$  were highest in the clay, followed by the loam and the sandy CL, and increased exponentially with increasing WFPS, except for the loam, where  $N_2O_d$  emissions decreased at WFPS  $>$  80% following an exponential increase at WFPS levels  $<$  80%. The magnitude of  $N_2O_{cd}$  emissions was smaller than the combined error of GC analysis,  $^{15}N_2O$  analysis and the quantification of the fraction of  $^{15}N$  in the soil  $NH_4^+$  and  $NO_3^-$  pool, and therefore assumed to be below the method detection limit.

Emissions of  $N_2O_a$  increased with soil WFPS in the clay and the loam, with highest  $N_2O_a$  emissions observed at 80 and 95% WFPS in the clay, and at 95% in the loam. Emissions of  $N_2O_a$  from

the sandy CL were negatively correlated to increasing soil WFPS and accounted for less than 23% of cumulative  $N_2O$  emissions. Across soils,  $N_2O_h$  emissions increased with increasing WFPS reaching a plateau at WFPS  $\geq$  80%. Heterotrophic nitrification exceeded autotrophic nitrification as an  $N_2O$  production pathway in the loam and the sandy CL regardless of WFPS, but its contribution to  $N_2O$  emissions remained below the one of autotrophic nitrification in the clay. Emissions of  $N_2O$  from all nitrification mediated pathways, calculated as the sum  $N_2O_a$  and  $N_2O_h$  increased with soil WFPS, with highest  $N_2O_n$  emissions observed  $\geq$  80% WFPS.

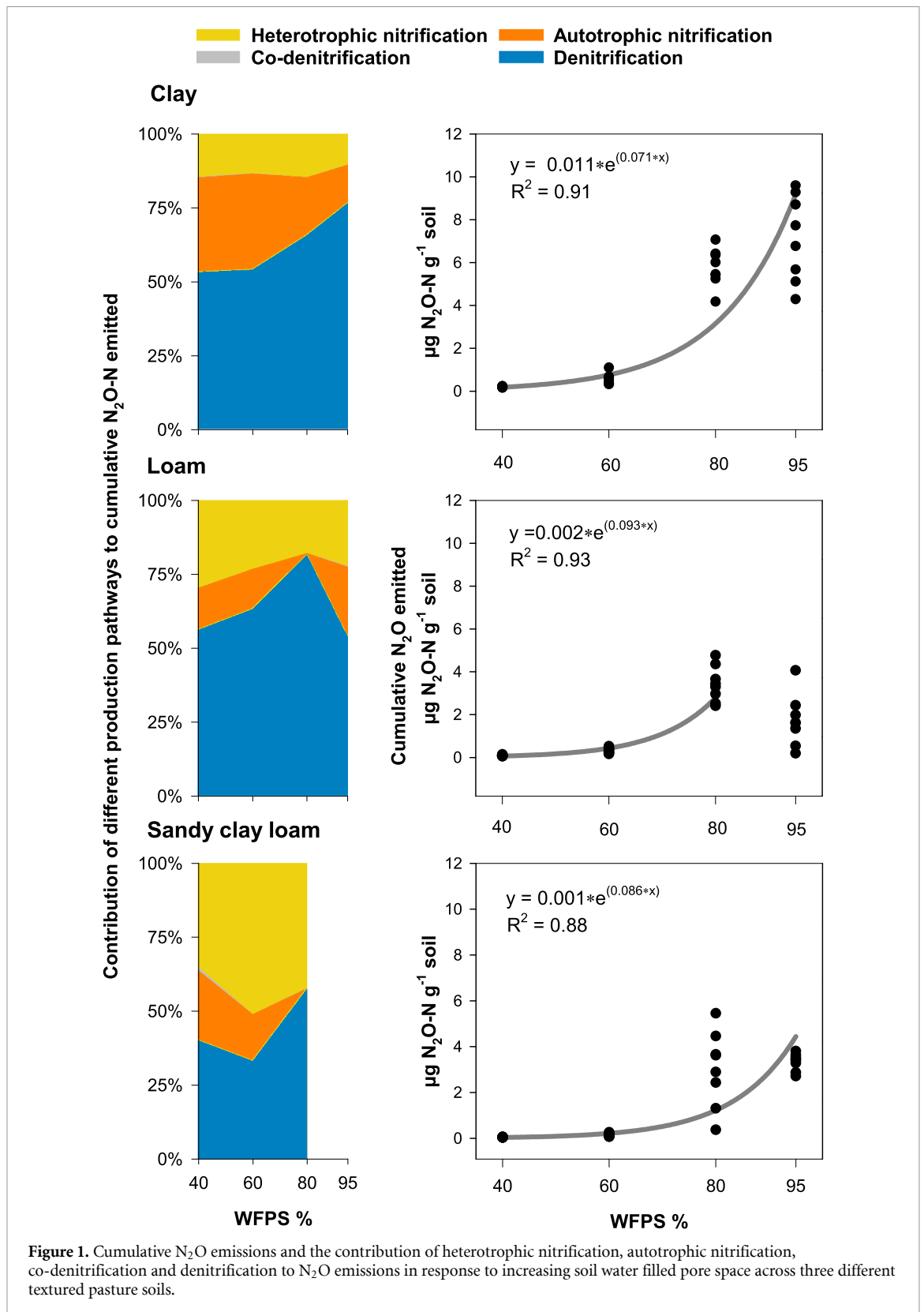
#### 3.3. The fraction of different N transformations emitted as $N_2O$

Denitrification and total nitrification (the sum of autotrophic and heterotrophic nitrification) and the fraction of nitrified and denitrified N emitted as  $N_2O$  ( $RN_2O$ ) are shown in figure 2. Denitrification ( $N_2 + N_2O_d$ ) increased exponentially with increasing WFPS (table S2), with no difference in magnitude between soils. The fraction of denitrified N emitted as  $N_2O$  ( $RN_2O_d$ ) followed a linear decrease with increasing WFPS in the loam and the sandy CL, and in the clay after increasing from 40% to 60% WFPS.

The response of total nitrification to WFPS differed between soils (figure 2), showing a slight increase from 40% to 80% WFPS in the clay and the sandy CL, while nitrification rates peaked at 80% WFPS in the loam at  $>40 \mu\text{g } NO_3^-N \text{ g}^{-1} \text{ d}^{-1}$  soil and decreased thereafter. Across all soils,  $RN_2O_n$  ranged from 0.002 to 0.042 and increased exponentially with increasing WFPS.

The response of autotrophic and heterotrophic nitrification to WFPS and respective  $RN_2O$  values is shown in figure 3. In the clay and the sandy CL, autotrophic nitrification peaked between 60 and 80% WFPS at  $>20 \mu\text{g } NO_3^-N \text{ g}^{-1} \text{ soil day}^{-1}$ , and decreased to less than  $2 \mu\text{g } NO_3^-N \text{ g}^{-1} \text{ d}^{-1}$  at 95% WFPS. In the loam, autotrophic nitrification remained  $<1.5 \mu\text{g } NO_3^-N \text{ g}^{-1} \text{ d}^{-1}$  across all WFPS levels. The fraction  $RN_2O_a$  from the clay and the loam responded with an exponential increase to increasing WFPS, with values up to 1 and 0.4 observed at 95% WFPS in the clay and the loam, respectively. The fraction  $RN_2O_a$  from the sandy CL decreased from 0.013 to  $<0.001$  from 40% to 80% WFPS, following an exponential decay curve.

Heterotrophic nitrification was the dominant nitrification pathway in the loam at all WFPS levels and at 95% WFPS in the clay and the loam (figure 3). Rates of heterotrophic nitrification followed a quadratic function in the clay and the sandy clay loam, with the lowest rates observed at 60% WFPS, and a subsequent increase up to 8 and  $25 \mu\text{g } NO_3^-N \text{ g}^{-1} \text{ soil day}^{-1}$  at 95% WFPS, respectively. Heterotrophic nitrification peaked between 60 and 80% WFPS in



the loam at >than 40  $\mu\text{g NO}_3^- \text{N g}^{-1} \text{ soil day}^{-1}$  and decreased thereafter. Highest values for  $\text{RN}_2\text{O}_h$  were observed from the sandy CL peaking at 0.95 at 60% WFPS, dropping to 0.31 at 80% WFPS. In the clay and the loam,  $\text{RN}_2\text{O}_h$  remained below 0.04, peaking at 80 and 95% WFPS, respectively.

### 3.4. Labile C availability and cumulative CO<sub>2</sub> emissions

Labile C availability prior incubation was derived from Friedl *et al* (2018). Labile C measured as permanganate-oxidisable C (Weil *et al* 2003) was highest in the loam ( $1196 \pm 14.9 \mu\text{gC g}^{-1} \text{ soil}$ ),

**Table 2.** Cumulative N<sub>2</sub>O emissions from denitrification (N<sub>2</sub>O<sub>d</sub>), autotrophic nitrification N<sub>2</sub>O<sub>a</sub>, heterotrophic nitrification (N<sub>2</sub>O<sub>h</sub>), total nitrification (N<sub>2</sub>O<sub>n</sub>) and total N<sub>2</sub>O from three different textured pasture soils at four different soil moisture levels and their relative contribution in % to overall N<sub>2</sub>O emissions.

	Clay			Loam			Sandy clay loam					
WFPS	↓	↔	%	↓	↔	%	↓	↔	%			
N <sub>2</sub> O <sub>d</sub> N μg N g <sup>-1</sup> soil												
40%	0.108 ± 0.007	c	A	53	0.048 ± 0.008	c	B	57	0.019 ± 0.002	b	C	40
60%	0.307 ± 0.08	c	A	55	0.207 ± 0.040	c	A	63	0.047 ± 0.011	b	B	33
80%	3.798 ± 0.288	b	A	66	2.795 ± 0.341	b	B	82	1.732 ± 0.47	a	C	57
95%	5.481 ± 0.767	a	A	77	0.927 ± 0.321	a	B	54				
N <sub>2</sub> O <sub>a</sub> N μg N g <sup>-1</sup> soil												
40%	0.064 ± 0.004	b	A	32	0.012 ± 0.002	b	B	14	0.011 ± 0.001	b	B	23
60%	0.178 ± 0.034	b	A	32	0.044 ± 0.009	b	B	13	0.022 ± 0.005	a	B	16
80%	1.13 ± 0.104	a	A	20	0.02 ± 0.004	b	B	1	0.003 ± 0.001	c	B	0
95%	0.922 ± 0.129	a	A	13	0.41 ± 0.142	a	B	24				
N <sub>2</sub> O <sub>h</sub> N μg N g <sup>-1</sup> soil												
40%	0.029 ± 0.002	b	A	14	0.025 ± 0.004	c	A	30	0.017 ± 0.002	b	B	36
60%	0.073 ± 0.015	b	A	13	0.076 ± 0.015	c	A	23	0.073 ± 0.018	b	A	52
80%	0.838 ± 0.058	a	AB	15	0.612 ± 0.083	a	B	18	1.288 ± 0.354	a	A	43
95%	0.739 ± 0.103	a	A	10	0.381 ± 0.132	b	B	22				
N <sub>2</sub> O <sub>n</sub> = N <sub>2</sub> O <sub>h</sub> + N <sub>2</sub> O <sub>a</sub> N μg N g <sup>-1</sup> soil												
40%	0.094 ± 0.006	a	A	47	0.037 ± 0.006	b	B	44	0.028 ± 0.003	b	B	60
60%	0.251 ± 0.049	a	A	45	0.12 ± 0.023	b	B	37	0.095 ± 0.022	b	B	67
80%	1.968 ± 0.156	b	A	34	0.632 ± 0.087	a	C	18	1.291 ± 0.355	a	B	43
95%	1.662 ± 0.232	b	A	23	0.791 ± 0.273	a	B	46				
Total N <sub>2</sub> O-N μg N g <sup>-1</sup> soil												
40%	0.202 ± 0.013	c	A	100	0.084 ± 0.014	c	B	100	0.047 ± 0.004	b	C	100
60%	0.559 ± 0.129	c	A	100	0.327 ± 0.063	c	B	100	0.141 ± 0.033	b	B	100
80%	5.769 ± 0.444	b	A	100	3.427 ± 0.414	a	B	100	3.026 ± 0.825	a	B	100
95%	7.148 ± 0.999	a	A	100	1.724 ± 0.596	b	B	100	3.151 ± 0.186	a	B	100

Letters denote homogenous groups (Tuckey HSD ( $p < 0.05$ )) for cumulative N<sub>2</sub>O emissions ( $\mu\text{g N g}^{-1}\text{soil}$ ).

↓ Within a soil type across different WFPS (small letters) for cumulative N<sub>2</sub>O emissions ( $\mu\text{g N g}^{-1}\text{soil}$ ).

↔ Across soil types, within a WFPS treatment (capital letters) for cumulative N<sub>2</sub>O emissions ( $\mu\text{g N g}^{-1}\text{soil}$ ).

exceeding values for the clay ( $856.0 \pm 39.4 \mu\text{gC g}^{-1}\text{soil}$ ) and the sandy CL ( $701.4 \pm 11.7 \mu\text{gC g}^{-1}\text{soil}$ ). Cumulative CO<sub>2</sub> emissions (table S2) as a relative measure of heterotrophic soil respiration were highest in the loam, followed by the clay and the sandy CL, regardless of WFPS and were positively correlated ( $P < 0.05$ ) to labile C availability.

#### 4. Discussion

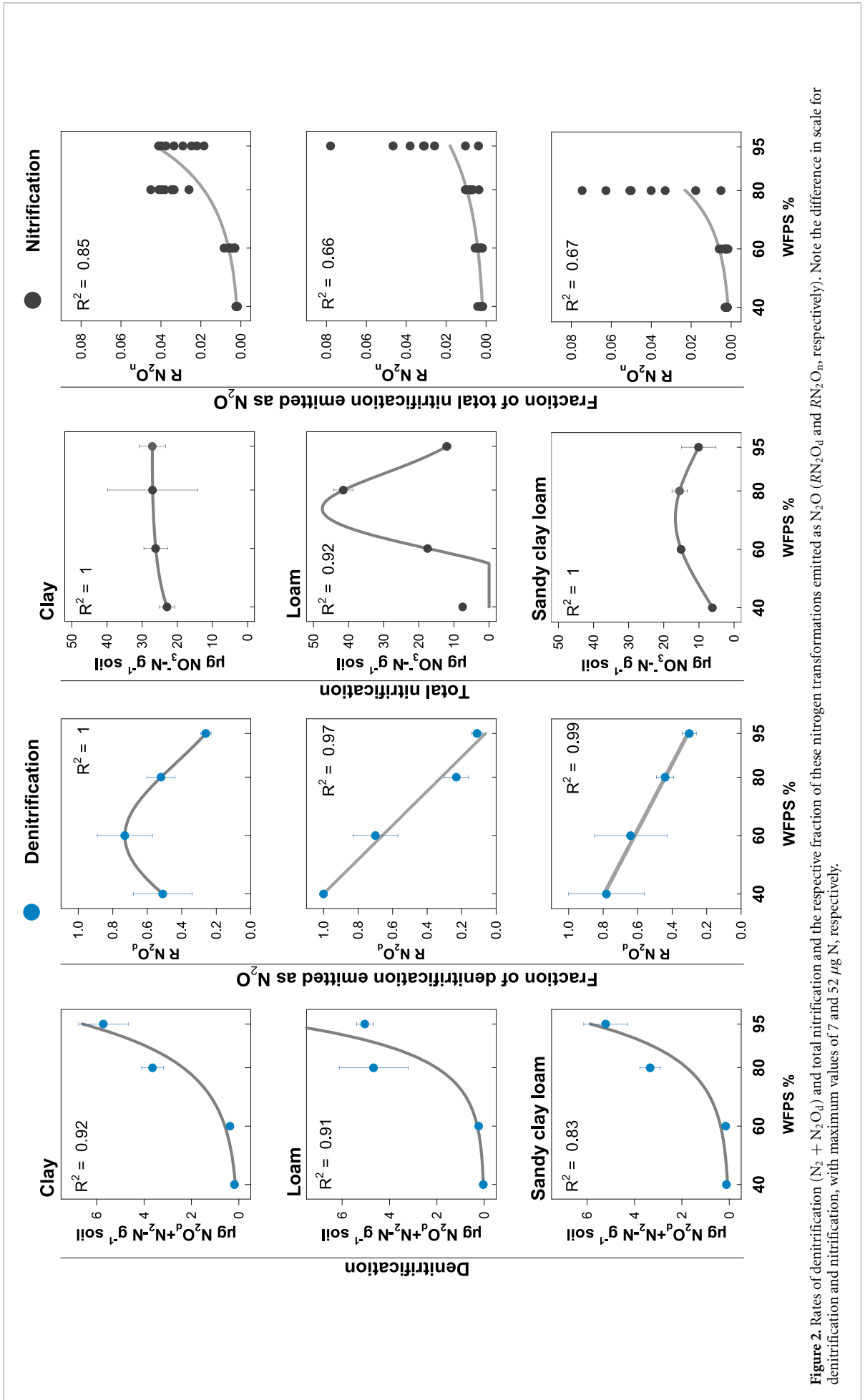
Wetting events trigger pulses of N<sub>2</sub>O emissions from soils, accounting for a large proportion of overall N<sub>2</sub>O emissions. The relative change in soil water content together with the antecedent soil moisture, rather than absolute amounts of water in soil define the magnitude of these N<sub>2</sub>O pulses (Bergstermann *et al* 2011, Harris *et al* 2021). The short-term response of different N<sub>2</sub>O production pathways to wetting of pasture soils provides therefore critical information to constrain magnitude and source partitioning of N<sub>2</sub>O pulses. This study demonstrates the simultaneous occurrence of N<sub>2</sub>O emissions via denitrification and nitrification mediated pathways across three different pasture soils regardless of the amount of wetting. Partitioning of N<sub>2</sub>O emission

shows the large contribution of heterotrophic nitrification to N<sub>2</sub>O production, highlighting the oxidation of organic N as a major source of N<sub>2</sub>O from pasture soils. Following the HIP model, the fraction of total nitrification lost as N<sub>2</sub>O ( $RN_2O_n$ ) ranged from 0.001 to 0.04 and increased exponentially with soil WFPS, while the respective fraction of denitrification ( $RN_2O_d$ ) decreased. Based on these findings, we postulate that (a) in pasture soils with high organic C and N content, the cascade of physical, chemical and biological processes triggered by wetting promotes N<sub>2</sub>O production via denitrification and via the oxidation of organic N substrates and (b) that the exponential increase of the hole in the pipe, i.e. the amount of N<sub>2</sub>O lost from nitrification mediated pathways is driven by the denitrification of nitrified N.

##### 4.1. Production of N<sub>2</sub>O from denitrification and co-denitrification in response to wetting

The use of the <sup>15</sup>N pool mixing model showed that denitrification dominated N<sub>2</sub>O production in all soils (figure 1), accounting for 40%–80% of cumulative N<sub>2</sub>O losses. Under oxic conditions ( $\leq 60\%$  WFPS), production of N<sub>2</sub>O is assumed to occur





**Figure 2.** Rates of denitrification ( $N_2 + N_2O_d$ ) and total nitrification and the respective fraction of these nitrogen transformations emitted as  $N_2O$  ( $RN_2O_d$  and  $RN_2O_n$ , respectively). Note the difference in scale for denitrification and nitrification, with maximum values of 7 and 52  $\mu g N$ , respectively.

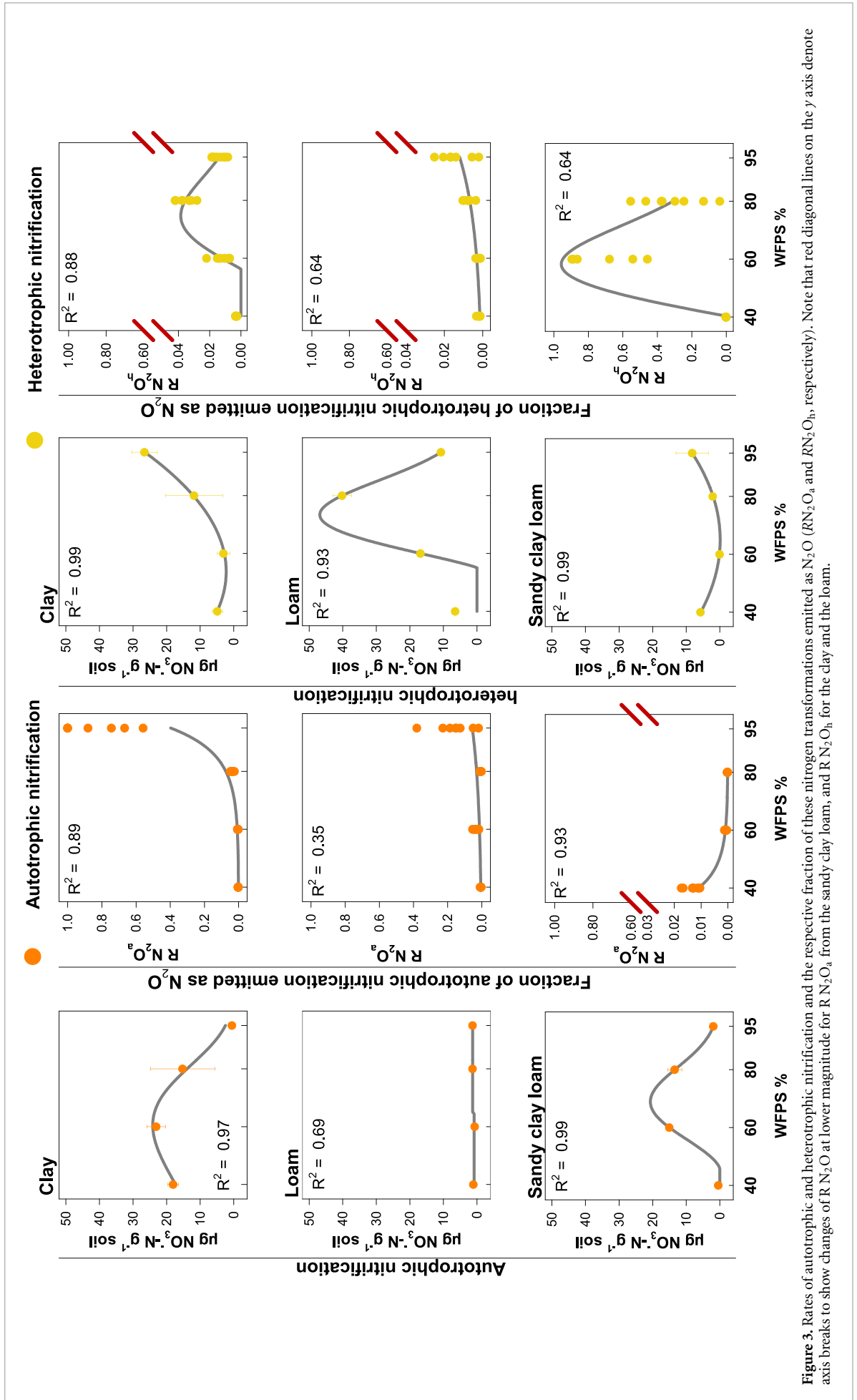


Figure 3. Rates of autotrophic and heterotrophic nitrification and the respective fraction of these nitrogen transformations emitted as N<sub>2</sub>O (RN<sub>2</sub>O<sub>a</sub> and RN<sub>2</sub>O<sub>f</sub>, respectively). Note that red diagonal lines on the y axis denote axis breaks to show changes of RN<sub>2</sub>O<sub>a</sub> at lower magnitude for R N<sub>2</sub>O<sub>a</sub> from the sandy clay loam, and RN<sub>2</sub>O<sub>a</sub> for the clay and the loam.

mainly via nitrification mediated pathways (Bollmann and Conrad 1998, Bateman and Baggs 2005). Partitioning of  $N_2O$  in response to wetting in this study shows however denitrification as the main  $N_2O$  source even at WFPS  $\leq 60\%$  in the clay and the loam, and accounting for  $>30\%$  of  $N_2O$  emissions in sandy CL. Dry periods induce the build-up of N bearing substrates in grassland soils (Harris *et al* 2021). Microbial activity increases rapidly upon wetting of dry soils (Congreves *et al* 2018), resulting in increased microbial  $O_2$  consumption, which creates favourable conditions for denitrification. This effect is likely to be more pronounced in high organic C pasture soils with high microbial activity (Friedl *et al* 2020), where the release of solutes from microbial cells (Schimel *et al* 2007) and the decomposition of the microbial necromass (Kieft *et al* 1987) supplies low C:N substrate in response to wetting. Source partitioning of  $N_2O$  demonstrated in the study here integrates the above-mentioned effects of wetting on  $N_2O$  production, creating conditions conducive for denitrification even at low soil water contents.

The magnitude of  $N_2O_d$  emissions followed the same pattern as cumulative  $N_2O$  emissions (figure 1, table 2). In the clay and the sandy CL,  $N_2O_d$  increased exponentially with WFPS, with peak  $N_2O$  emissions  $>5 \mu\text{g } N_2O\text{-N}$  emitted from the clay. In the loam however, the exponential increase was only observed up to 80% WFPS, with a subsequent decrease in  $N_2O_d$  emissions. Emissions of  $N_2O$  are generally expected to decrease when soil moisture reaches saturation, as hypoxic conditions favour the reduction of  $N_2O$  to  $N_2$  (Morley and Baggs 2010). The exponential increase of  $N_2O_d$  in the clay and sandy CL however denotes residual  $O_2$  in the soil matrix and a delay in the development of anaerobiosis in these soils. The exponential increase in  $N_2O$  emissions in response to wetting and fertilisation may be further caused by high  $NO_3^-$  availability, promoting preferential  $NO_3^-$  reduction, and thus limiting the reduction of  $N_2O$  to  $N_2$  (Senbayram *et al* 2019, Friedl *et al* 2020). The effect of wetting is amplified by the high labile C availability in pasture soils, supplying an energy source to heterotrophic denitrifiers, but also increasing microbial  $O_2$  consumption (Azam *et al* 2002, Meyer *et al* 2010) and the subsequent formation of anaerobic microsites. The loam had the highest labile C availability across soils, explaining the decrease of  $N_2O_d$  emissions  $>80\%$  WFPS as a consequence of increased microbial  $O_2$  consumption. Limited  $O_2$  availability in the soil matrix induces  $NO_3^-$  consumption by DNRA (Friedl *et al* 2018, Putz *et al* 2018), and the reduction of  $N_2O$  to  $N_2$ , both limiting  $N_2O_d$  emissions. The effect of labile C availability on denitrification is further reflected in the fraction of denitrification ( $N_2 + N_2O$ ) lost as  $N_2O$  ( $RN_2O_d$ ) (figure 2). Across soils,  $RN_2O_d$  followed a similar pattern decreasing at

WFPS levels  $\leq 60\%$ , yet the rate of decrease increased with labile C concentration, with lowest values for  $RN_2O_d$  observed from the loam near saturation. These findings highlight the soil specific response of  $N_2O$  formation via denitrification in pasture soils and show the combined effect of soil moisture and microbial  $O_2$  consumption on  $N_2O_d$  production.

Co-denitrification was negligible for  $N_2O$  production in this study. Large  $N_2O$  fluxes from co-denitrification of  $>80 \text{ mg } N_2O\text{-N } m^{-2} d^{-1}$  have been reported from urine patches (Selbie *et al* 2015), while  $NH_4NO_3$  additions comparable to the study here induced only minor  $N_2O_{cd}$  fluxes ( $\leq 5\%$  of cumulative  $N_2O$ ) (Jansen-Willems *et al* 2016). The production of  $N_2O$  via co-denitrification maybe, therefore, closely linked to the chemical and biological reactions triggered by high urine/urea N deposition in pasture soils (Spott *et al* 2011, Breuillin-Sessoms *et al* 2017, Clough *et al* 2020).

#### 4.2. Production of $N_2O$ from autotrophic and heterotrophic nitrification in response to wetting

The  $^{15}N_2O$  pool mixing model used in this study attributes  $N_2O$  without  $^{15}N$  label to the unlabelled organic N pool, and therefore to heterotrophic nitrification. Other pathways of  $N_2O$  production using organic N substrates include chemo and co-denitrification (Butterbach-Bahl *et al* 2013). These pathways form however hybrid  $N_2O$  by combining labelled  $NO_2^-$  with organic N compounds at natural abundance, differing in their isotopic composition from  $N_2O_h$  at natural abundance.

The observed heterotrophic nitrification rates are amongst the highest reported from agricultural soils (Chen *et al* 2015, Zhang *et al* 2015) and exceed reported rates from grassland soils (Rütting *et al* 2010, Müller *et al* 2014, Jansen-Willems *et al* 2016). Production of  $N_2O_h$  is thought to occur via the oxidation of reduced N in organic matter to  $NO_2^-$  and  $NO_3^-$ , and the subsequent reduction to  $N_2O$  (Braker and Conrad 2011). Similar to nitrifier-denitrification, the process links an aerobic metabolism with denitrification (Blagodatsky *et al* 2006), and is likely more adapted to fluctuating redox conditions in soils triggered by wetting events. This is consistent with the production of  $N_2O_h$ , remaining constant or increasing with increasing WFPS. Peak emissions of  $N_2O_h$  denote 80% WFPS as optimum across soils, with subsequent emissions decreasing. Production of  $N_2O$  via heterotrophic nitrification has been suggested in waste-water treatment reactors (Zhao *et al* 2012), its significance for soil gas exchange remains however unknown. Regardless of its source,  $N_2O$  is more likely reduced to  $N_2$  at high soil water content, due to low  $O_2$  availability and prolonged retention of  $N_2O$  in the soil (Hansen *et al* 2014), explaining the decrease of  $N_2O_h$  at soil moisture contents  $>80\%$  WFPS.

The fraction  $RN_2O_a$  shows a decoupling between the respective rates of  $NO_3^-$  production and magnitude of  $N_2O$  emitted derived from autotrophic nitrification. Autotrophic nitrification rates were either  $b < 5 \mu g NO_3^- N g^{-1}$  soil or decreased below this threshold when WFPS  $> 80\%$  (figure 3), denoting the sensitivity of  $NO_3^-$  production by autotrophic nitrifiers to increasing anaerobiosis. Emissions of  $N_2O_a$  however increased in the clay and the loam, resulting in an exponential response of  $RN_2O_a$  to increasing soil moisture. Nitrification as a source of  $N_2O$  has been traditionally attributed to aerobic conditions in agricultural soils, yet the response to wetting shown here implies increasing  $N_2O$  production along the ammonia oxidising pathway under  $O_2$  limited conditions. The response of  $RN_2O_a$  to wetting is consistent with the reported increase of  $RN_2O_a$  with decreasing  $O_2$  availability (Zhu *et al* 2013), which has been attributed to nitrifier-denitrification, i.e. the ability of ammonia oxidisers to denitrify (Wrage *et al* 2001, Prosser *et al* 2020). The  $^{15}N_2O$  pool mixing model used in our study links  $N_2O$  production to the respective N source pools and this representation of  $N_2O$  production corresponds with the one in the HIP model. The N substrate supplying process may however differ from the microbial process of  $N_2O$  production and can respond differently to wetting. Besides autotrophic nitrifiers, heterotrophic denitrifiers are also able to use  $NO_2^-$  produced via autotrophic nitrification as a substrate for  $N_2O$  production (Liu *et al* 2013), which could also explain continued emissions of  $N_2O_a$  at high WFPS. High WFPS and therefore reduced soil gas diffusivity is also likely to have delayed  $N_2O_a$  surface emissions, which were produced when residual  $O_2$  was still abundant, facilitating autotrophic nitrification at microsites. The contribution of nitrifier-denitrification and denitrification to  $N_2O_a$  production has been debated (Bakken and Frostegård 2017) as methodological constraints hinder accurate  $N_2O$  partitioning (Prosser *et al* 2020). Analogue to biogeochemical models, the source partitioning model used in the study here summarises these processes under  $N_2O_a$  according to the source of the N substrate, and suggests that heterotrophic and/or autotrophic denitrification of  $NO_2^-$  derived from nitrification drives the increase of  $RN_2O_a$  in response to wetting.

The fraction  $RN_2O_h$  did not follow a common pattern across soils, reflecting vastly different  $N_2O$  and  $NO_3^-$  production via heterotrophic nitrification. Production of  $N_2O$  via heterotrophic nitrification has been shown to increase with decreasing soil pH (Zhang *et al* 2018b), which is consistent with the observed negative correlation of  $N_2O_h$  and soil pH in the study presented here. The highest emissions of  $N_2O_h$  were observed from the sandy CL, together with the lowest rates of heterotrophic nitrification (figure 3), implying different factors driving  $NO_3^-$  and related  $N_2O_h$  production. The

sandy CL has the lowest pH across soils and has been characterised as a pasture soil under extensive management (De Rosa *et al* 2020), receiving less mineral N fertiliser inputs than the clay and the loam. Heterotrophic nitrifiers comprise a large variety of phylogenetically unrelated bacteria and fungi (Braker and Conrad 2011) and fungi have been shown to dominate heterotrophic  $NO_3^-$  production under more acidic conditions (Zhu *et al* 2015). Specific adaption of the soil microbial community to soil chemical properties and N fertilisation (Zhang *et al* 2018a) may explain the differences in partitioning of  $N_2O$  production between heterotrophic and autotrophic nitrification with important implication for practical  $N_2O$  abatement strategies: Nitrification inhibitors such as 3,4-dimethylpyrazole phosphate (DMPP) inhibit the activity of the ammonia monooxygenase (AMO), facilitating ammonia oxidation to hydroxylamine. Heterotrophic nitrification by fungi is however thought to lack AMO (Wood 1990), suggesting that nitrification inhibitors are ineffective in mitigating N substrate supply for  $N_2O$  production via this pathway. The contribution of heterotrophic nitrification to  $N_2O$  production in response to wetting across soils shows that this pathway is not restricted to strong acidic conditions and needs to be considered when managing N substrate availability for  $N_2O$  formation in intensively managed pasture soils.

Fractions of  $N_2O$  emitted from autotrophic and heterotrophic nitrification differed in their response to wetting. However, the fraction of overall nitrification emitted as  $N_2O$  increased exponentially across all soils, with an inflection point between 60 and 80% WFPS. These findings demonstrate the importance of both processes for N substrate availability for  $N_2O$  production. The increase of  $RN_2O_n$  however suggests denitrification of nitrified N as the process of  $N_2O$  production, driven by the rapid depletion of soil  $O_2$ . Our findings are consistent with the response of  $RN_2O_n$  to decreasing  $O_2$  availability (Khalil *et al* 2004, Zhu *et al* 2013), and confirm the exponential relationship implemented in models such as NOE (Hénault *et al* 2005) and DAYCENT (Yang *et al* 2017). The use of a constant for  $RN_2O_n$  independent of soil WFPS is however likely to underestimate  $N_2O$  derived from nitrification when simulating large rainfall events.

### 4.3. The HIP

The response of  $N_2O$  production demonstrates an exponential increase of  $N_2O$  emissions with the amount of wetting, dominated by denitrification across pasture soils. The immediate reduction of  $N_2O$  to  $N_2$  even at low soil water content denotes increased microbial activity and therefore  $O_2$  consumption following the wetting of dry pasture soils, driving the  $N_2O:N_2$  ratio towards  $N_2$ . The inclusion of the organic N pool into the  $^{15}N_2O$  mixing model revealed a significant contribution of heterotrophic nitrification to  $N_2O$  production, which appears to

be less sensitive to increasing amounts of wetting than autotrophic nitrification. Losses of  $N_2O_d$  and  $N_2O_h$  suggest that build up and sudden release of C and N substrate upon wetting stimulates  $N_2O$  production via denitrification and via the oxidation of organic N. Continuous  $N_2O$  emissions derived from autotrophic nitrification even at high WFPS highlight the decoupling between the pipe and the hole, i.e. between the rate of the N transformation and the respective fraction lost as  $N_2O$ . This is shown by the exponential increase of  $RN_2O_n$  with increasing soil water content, suggesting denitrification of nitrified  $NO_2^-$  causing  $N_2O$  losses from nitrification mediated pathways at high soil water content. The consideration of the  $NO_2^-$  pool as a central N pool in models such as in LDNDC may therefore help to accurately simulate  $N_2O$  production via nitrification mediated pathways, and further research tracing  $^{15}N$  in the  $NO_2^-$  pool is needed to deliver experimental evidence and validation data for this approach. Our findings demonstrate that the proportion of nitrified N lost as  $N_2O$  ( $RN_2O_n$ ) is not constant as assumed in some models but increases exponentially with the degree of wetting. These results provide important experimental evidence for the relationship of soil water with  $N_2O_n$  production and corroborate the exponential response of  $RN_2O_n$  in models such as SWAT and NOE. Importantly, the  $RN_2O$  response curves established in our study combine physical, chemical, and biological effects of wetting on  $N_2O$  production pathways from pasture soils, and their implementation in modelling approaches may help to increase model performance when simulating drying and wetting cycles. The large contribution of heterotrophic nitrification to  $N_2O$  production suggests an opportunity to improve models by adding further pathways of  $N_2O$  production. The resulting complexity however needs to be weighed against potential benefits, ensuring accurate  $N_2O$  forecasting for intensively managed pasture systems.

### Data availability statement

The data that support the findings of this study are available upon reasonable request from the authors.

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