# Stable Organic Passivated Carbon Nanotube-Silicon Solar Cells with an Efficiency of 22\% 

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#### Abstract

The organic passivated carbon nanotube (CNT)/silicon (Si) solar cell is a new type of low-cost, high-efficiency solar cell, with challenges concerning the stability of the organic layer used for passivation. In this work, the stability of the organic layer is studied with respect to the internal and external (humidity) water content and additionally long-term stability for low moisture environments. It is found that the organic passivated CNT/Si complex interface is not stable, despite both the organic passivation layer and CNTs being stable on their own and is due to the CNTs providing an additional path for water molecules to the interface. With the use of a simple encapsulation, a record power conversion efficiency of $22 \%$ is achieved and a stable photovoltaic performance is demonstrated. This work provides a new direction for the development of high-performance/low-cost photovoltaics in the future and will stimulate the use of nanotubes materials for solar cells applications.


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## 1. Introduction

Silicon solar cells are clearly the most successful and prolific photovoltaic(PV) technology of our time. Ninety seven percent of the global PVs market can be accredited to silicon based cells, ${ }^{[1,2]}$ recently industrialized solar cell architectures such as the passivated emitter and rear cell (PERC) have reached power conversion efficiencies (PCEs) of $\approx 23 \%$, ${ }^{[3]}$ and new generation scaled silicon heterojunction cells have pushed past $24 \%$ on industrial production lines ${ }^{[4]}$ with interdigitated back contacts reporting $26.7 \%$ in the laboratory. ${ }^{[5]}$ However, these impressive PCEs are coupled to increasingly complex device designs and these have correspondingly high manufacturing costs. The dominant feature for the recent high-efficiencies is the "passivating-contact," which consists of a carrier-selective contact (phosphorous, $\mathrm{n}^{+}$or boron, $\left.\mathrm{p}^{+}\right)^{[6]}$ overlaid by a passivation layer ( $\mathrm{SiO}_{2}$ or intrinsic hydrogenated amorphous Si (a-Si:H(i)). ${ }^{[7,8]}$ These silicon layers are energy intensive to fabricate, require high vacuum processes, and are deposited serially on the production line. Therefore, in order to reduce the levelized cost of production, new and simplified approaches are urgently required. For example, water-based dye-sensitized solar cells and perovskites solar cells have attracted the great attention as a cheap and efficient alternative. ${ }^{[9-14]}$

In this direction, hybrid materials are being developed that can passivate interfacial electron traps whilst also being carrier selective. These materials have been named "conductive passivating contacts (c-PCs)" and one such approach is to disperse carrier selective nanoparticles in an organic passivation agent to form an ink that can be spin coated in a single step and at room temperature. ${ }^{[15]}$ On their own, organic thin films with the sulfonic acid group ( $-\mathrm{SO}_{3} \mathrm{H}$ ), such as polystyrenesulfonate, ${ }^{[16]}$ poly(2-acrylamido-2-methylpropanesulfonic acid), polystyrene-block-poly(ethylene-ran-butylene)-block-polystyrene-sulfonated-cross-linkable, ${ }^{[17]}$ and Nafion have all been shown to effectively passivate interfacial trap states on silicon by the electrochemical grafting of $-\mathrm{SO}_{3} \mathrm{H}$ onto Si surface dangling bonds to form Si suboxides at the polymer/Si interface. ${ }^{[18]}$ More specifically, minority carrier lifetimes of $9.6-28.6 \mathrm{~ms}$ have been reported for organic passivation and these rival industry standard hydrogenated
amorphous $\mathrm{Si}(\mathrm{a}-\mathrm{Si}: \mathrm{H})$ or $\mathrm{SiO}_{2}$ film-passivation schemes. ${ }^{[19,20]}$ For c-PCs such as PEDOT:Nafion and CNT:Nafon, high PCEs of $18.8 \%$ and $21.4 \%$ for industrial p-type polycrystalline and n-type monocrystalline silicon solar cells, respectively, have been demonstrated. ${ }^{[15,21]}$ In terms of simplicity and the resultant PCE, this approach compares favorably with the recently industrialized PERC but allowed for the use of n-type silicon to form a new heterojunction and was predicted to be compatible with other nanoparticles including flakes of graphene, ${ }^{[22-25]}$ MXenes, ${ }^{[26]}$ or transition metal dichalcogenides. ${ }^{[27]}$

From the CNT-based PV device perspective, the c-PC approach provided a breakthrough in performance for CNT-Si solar cells, but also for the area ( $245.71 \mathrm{~cm}^{2}$ ) achieved. ${ }^{[21]}$ However, the downside of the approach was the long term stability of the CNT:Nafion composite layer. Although CNTs have excellent stability toward degradation in ambient, humid, hot, or ultraviolet radiation conditions, ${ }^{[28-31]}$ the hygroscopic nature of Nafion is known to lead to performance reductions ${ }^{[32]}$ and must be addressed. Primarily, these issues are related to reductions in the open circuit voltage ( $V_{o c}$ ) from degradation of the Nafion based passivation. In similar work by Qian et al., ${ }^{[33]}$ who also combined Nafion and CNTs, device stability was reported to be stable but this is likely due to the lower PCE/ $V_{\text {oc }}$ of $14.4 \% / 549 \mathrm{mV}$. In other work we have shown that the stability of the passivation effect afforded by a $-\mathrm{SO}_{3} \mathrm{H}$ based organic thin film alone can be increased to 430 days with the use of an ALD- $\mathrm{Al}_{2} \mathrm{O}_{3}$ encapsulation layer, but such a layer is incompatible with c-PC solar cells because of the insulating property of $\mathrm{Al}_{2} \mathrm{O}_{3}$. Nevertheless, it does suggest that organic passivation strategies alone can become stable via encapsulation. ${ }^{[15]}$ For a solar cell, the addition of CNTs is required to provide the conductive and carrier selective component of the c-PC and here the method of their inclusion appears to be important. When Nafion and the CNTs were combined together into a single ink, the high PCE (21\%) of our devices was found to be stable for only 20 min , whereas when a layer of CNTs was first laminated onto the silicon wafer and then coated with Nafion in a second step, performance was found to be stable for 2 days. ${ }^{[34]}$ Clearly, a mixed system that can be deposited in a single step is more attractive for industry if stability can be improved. In this work, the stability of CNT:Nafion mixed systems are studied, both by varying the content of water within the film, but also the humidity of the external environment. Device compatible encapsulation is then employed to demonstrate a stability trend and record high PCE of $22 \%$. In addition, a high fill factor (FF) of $83.4 \%$ and external quantum efficiency (EQE) which can rival the current a-Si:H(i/p+) passivating contact, are obtained, revealing the potential advantage of the CNT-organic c-PC technology.

## 2. Results and Discussion

Using the technique of the quasi steady-state and transient photoconductance decay (PCD), the minority carrier lifetime ( $\tau_{\text {eff }}$ ) for silicon wafers ( $400 \mu$ m-thick n-type (100)-oriented slightly doped, with the resistivity of $3-5 \mathrm{k} \Omega \bullet \mathrm{cm}$ ) coated symmetrically with a Nafion film alone were measured and are shown in Figure 1a. Nafion films were spin cast from an alcohol/water mix ( $2.94 \mathrm{wt} \%$ Nafion/92.06 wt\% aliphatic alcohols/5 wt\% $\mathrm{H}_{2} \mathrm{O}$ ) and the relative humidity ( RH ) was increased from $17-33 \%$ over a period of 800 min . To aid comparison, the initial value of $\tau_{\text {eff }}(4-20.6 \mathrm{~ms})$,
measured immediately after coating, was used for normalization. The initial increase in lifetime (up to $\approx 146 \%, 20.6 \mathrm{~ms}$ ) is related to the electrochemical passivation effect of Nafion. This effect is enhanced by $\mathrm{O}_{2}$ in the atmosphere (the basic principle of the $\mathrm{O}_{2}$ enhanced passivation can be found elsewhere ${ }^{[32]}$ ) but the ingress of small quantities of water into the film has a negative effect and $\tau_{\text {eff }}$ was reduced to $\approx 26 \%$ for a RH of $33 \%$. In contrast, when the RH was held at $17 \%$ or when an $\mathrm{Al}_{2} \mathrm{O}_{3}$ encapsulation layer was used, $\tau_{\text {eff }}$ was found to be stable for 10 h . Figure 1 b shows the normalized $\tau_{\text {eff }}$ evolution for Si wafers passivated by the Nafion thin-films with different water content (5-20\%) in the solution to be spin coated. Once again, for a RH $<17 \%$ no obvious degradation in lifetime is observed for each of the films. However, for a RH $>17 \%$ and a water content of $20 \%$, degradation of the lifetime begins to occur. The passivation effect of Nafion can be summarized in the following way: i) the passivation is degraded by the increase of the water within film and RH in the environment; ii) when the water content is $<20 \%$ and RH is $<17 \%$, long term stability can be achieved. This is unlike conventional chemical passivation or field-effect passivation mechanisms and depends on the electrochemical interaction between $-\mathrm{SO}_{3} \mathrm{H}$ and Si surface, according to the Equation (1), ${ }^{[18]}$
$\equiv \mathrm{Si} \cdot+\mathrm{X}-\mathrm{SO}_{3} \mathrm{H} \underset{e^{-}}{\stackrel{h^{+}}{\rightleftharpoons}} \mathrm{Si}-\mathrm{O}-\mathrm{R}^{\prime}$
where X represents a normal constituent other than a sulfonic functional group, $-\mathrm{SO}_{3} \mathrm{H}$, and $\mathrm{R}^{\prime}$ is the surplus O -containing constituent in the $\mathrm{X}-\mathrm{SO}_{3} \mathrm{H}$ molecule. The water molecule invasion will contribute electrons to the Si surface via,
$2 \mathrm{Si}-\mathrm{O}-\mathrm{R}^{\prime}+\mathrm{H}_{2} \mathrm{O} \rightarrow 2\left(\mathrm{X}-\mathrm{SO}_{3} \mathrm{H}\right)+\frac{1}{2} \mathrm{O}_{2}+2 e^{-}+2 \equiv \mathrm{Si}$.
which favors the reverse reaction in Equation (1) and results in the degradation of passivation.

Figure 1c shows the normalized $\tau_{\text {eff }}$ evolution of the Si wafers passivated by a CNT:Nafion thin-film spin cast from a solution with a water content of $5 \%$. Samples with and without $\mathrm{Al}_{2} \mathrm{O}_{3}$ encapsulation at $23 \% \mathrm{RH}$ are shown along with the first 60 min of the Nafion only trace in Figure 1b for reference. Unexpectedly, the addition of the CNTs leads to a decrease in the stability with reduction of $\approx 69 \%$ of the initial lifetime value already occurring within this short time period. This suggests that CNTs accelerate the ingress of water, most likely by increasing the porosity of the film but possibly also due to endohedral water transport to the interface. ${ }^{[35]}$ This physiochemical mechanism is sketched in Figure S1, Supporting Information. A $\mathrm{MoO}_{3}$ nanoparticle ( $\approx 5 \mathrm{~nm}$ ) was used to fill the holes in the CNT:Nafion layer and stop the ingress of water and this lead to a slight increase in stability (Figure S2, Supporting Information), but a denser $\mathrm{Al}_{2} \mathrm{O}_{3}$ layer completely eliminated water, as shown in Figure 1c. Note that before and after the addition of $\mathrm{MoO}_{3}$ nanoparticles, large difference in the electrical conductivity (Table S1, Supporting Information) and work function (Figure S3, Supporting Information) were not observed. However, the use of an insulating $\mathrm{Al}_{2} \mathrm{O}_{3}$ layer for encapsulation prevents carrier transport when used in a back-junction architecture and when full rear contact with the c-PC layer is desired. For solar cells, a conductive encapsulation material is required.


Figure 1. Lifetime determined from transient photoconductance decay measurements of a) a Nafion film with increasing relative humidity (RH), b) varied internal water content (5-20\%) in dry and humid environments, and c) in combination with carbon nanotubes. d) Implied $V_{\text {oc }}$ calculated from at peak passivation and after 80 min .

The technique of PCD allows for the implied $V_{\text {oc }}$ to be calculated using the following equation:
implied- $V_{\text {oc }}=\frac{k T}{q} \ln \left(\frac{\Delta n\left(\Delta n+N_{D}\right)}{n_{i}^{2}}\right)$
where $\Delta n$ is the excess carrier density, $n_{\mathrm{i}}$ is the intrinsic carrier concentration and $N_{\mathrm{D}}$ is the effective bulk donor concentration. Using this approach, the best possible $V_{\text {oc }}$ that can be expected from a real solar cell can be predicted based on the peak passivation effect and before degradation of the lifetime. Figure 1d shows $\tau_{\text {eff }}$ and implied $V_{\text {oc }}$ as a function of the time for an unencapsulated CNT:Nafion film with $10 \%$ water content. As shown earlier, a Nafion film with $5 \%$ water content has the least water and therefore the most stable passivation effect, but it also results in the thinnest films from spin coating. For the final use with CNTs in a solar cell, a $10 \%$ water Nafion film was the best compromise between the required thickness and passivation, see Figure S4 and Table S2, Supporting Information. Taking the initial peak value of 7.75 ms affords a high implied $V_{\text {oc }}$ of 685 mV . Back junction solar cells were then built from such a CNT:Nafion film and a schematic of the complete device is shown in Figure 2a. Such high-quality passivation of a CNT:Nafion c-PC layer can also be shown by spatially resolved PL measurements of the samples before and after passivation (Figure S5, Supporting Information). An electron-selective passivation contact comprising $n$-doped hydrogenated amorphous silicon stack thin film
(a-Si:H(i/n $\left.\mathrm{n}^{+}\right)$) with a transparent conductive ITO antireflection layer is used on the front side. A CNT:Nafion c-PC layer is spin coated on the back side, followed by the evaporation of a thin Ag film (eAg). In this design, the CNT-Si heterojunction is responsible for hole-selective transport like a p-n junction and the Nafion-Si organic-inorganic interface is responsible for the passivation of the interfacial electron traps. Encapsulation was performed with an electrical Ag paste and forms a dense coating on the eAg film. Figure 2 b shows the illuminated $J-V$ curve of the champion device. A record high PCE of 22.04\% (Table S3, Supporting Information), a $V_{\text {oc }}$ of 683.4 mV , a $J_{\mathrm{sc}}$ of $40.38 \mathrm{~mA} \mathrm{~cm}^{-2}$, and a FF of $79.9 \%$ were obtained. Figure 2c shows the EQE and reflectance curves. A p-doped hydrogenated amorphous silicon stack thin film ( $\mathrm{a}-\mathrm{Si}: \mathrm{H}\left(\mathrm{i} / \mathrm{p}^{+}\right)$) passivating contact cell is shown as a reference to demonstrate the improvement of the infrared light management in the c-PC cell, which has the benefit of rear surface chemical polishing and low refractive index of the back c-PC layer ${ }^{[23]}$ in this kind of solar cell. Note that only $V_{\text {oc }}$ needs to be further improved compared with the a-Si:H $\left(\mathrm{i} / \mathrm{p}^{+}\right)$passivating contact cell. As shown in Figure 2d PV performance values were found to be stable over a period of 1600 min . $V_{\text {oc }}$ exhibits an increase in a short initial period, which is ascribed to the $\mathrm{O}_{2}$ enhanced operating mechanism by the Nafion passivation. ${ }^{[32]}$

In Figure 3 the performance values for 44 devices are presented and a comparison is made between encapsulated and unencapsulated cells, Table S4, Supporting Information showed the PV parameters with the error analysis. In our laboratory, the process of CNT:Nafion thin film coating, metallization, and finally


Figure 2. a) Schematic of the c-PC based device. The eAg layer is an evaporated Ag layer as a back electrode. b) J-V curve of the champion solar cell device with performance values inset. c) The external quantum efficiency (EQE) measurements and reflectance spectra comparison between conductive passivating contact cell and a-Si:H $\left(\mathrm{i} / \mathrm{p}^{+}\right)$passivating contact cell with performance values inset. d) Stability of $V_{o c}, J_{s c}$, FF, and PCE over a period of 1600 min in ambient conditions. Note: the solar cells were not illuminated continuously during this time. Photographs of the completed devices showing the front and rear of the cell.
measurement requires at least 1 h of time, during which the solar cell is exposed to water. This results in a reduction in $V_{\text {oc }}(683 \mathrm{mV})$ compared to the encapsulated devices (Figure 3a), but it compares well to the lower implied $V_{\text {oc }}$ of $<655 \mathrm{mV}$ in Figure 1d at 1 h . As it can be seen in Figure 3b,c, encapsulation generally does not affect the $J_{\mathrm{sc}}$ or FF, the former is due to the stable and consistent optical property on the front of devices and the later suggests that the electrical contact (i.e, $\mathrm{CNT}: \mathrm{Si}$ ) on the rear of the device is stable. The small variation in parameters can be seen and is attributed to the device stability and the preparation of the ink with the good dispersion. In fact, the FF can reach $83.4 \%$, as shown in Figure 3c, which is unusually high and revels an advantage of the CNT-based c-PC technology compared other nanomaterial based passivating contact approaches. We attribute this result to the high mobility of the CNTs ${ }^{[36,37]}$ and high-quality passivation of Nafion. Figure 2c compares the EQE response of a-Si:H(i/p+) passivating contact solar cell to ours. Both cells have the same geometry except for the back-side c-PC or a-Si:H(i/p+) passivating contact. Both have a similar EQE, suggesting that the c-PC can have the same level performance potential with the current best a-Si:H(i/p+) passivating contact. Further work is now required to increase the $V_{\text {oc }}$, of c-PC based solar cells, but now with a proven environmental stability, the combination of nanomaterials with organic passivation layers appears to provide a promising new avenue for low-cost and high-PCE PV devices in the future.

## 3. Conclusion

The a-Si:H passivating contact has been a great success to the PV community due to the super passivation of a-Si:H(i) and good carrier-selective transport properties of a a-Si:H(p)/a-Si:H(i)/n-Si p-i-n junction. In this work, a single composite c-PC layer with dual (carrier selective and passivating) has been shown to be long term stable with simple encapsulation. Following encapsulation, a new record in CNT:Si solar cell performance was obtained and a PCE of $22.04 \%$ reported. An EQE that is similar to industry standard passivated junctions also showed this approach is becoming competitive with existing technology. In the long term, because stability is limited by water and final solar cells are always encapsulated in a PV module, a c-PC approach is highly attractive for the vacuum-free, room temperature, low-cost fabrication.

## 4. Experimental Section

Solution and Thin Film Preparation: Nafion solutions with different water content ( $5 \%, 10 \%$, and $20 \%$ ) were obtained by mixing an original $20 \mathrm{wt} \%$ Nafion (Sigma Aldrich, $20 \mathrm{wt} \%$ in a mixture of lower aliphatic alcohols and $34 \%$ water) with ethanol ( $>99.7 \mathrm{wt} \%$ ). Simple magnetic stirring for 12 h was used to obtain the uniform precursor solution. The cPC ink was prepared by dispersing raw electric-arc single-wall CNT powder (XFS28, XFNano) with Nafion solution in the ratio of $1.8 \mathrm{mg} \mathrm{mL}^{-1}$ by shear force mixing for 3 h . The CNT:Nafion composite thin-films were


Figure 3. Comparison of the solar cell performance values a) $V_{o c}$, b) $J_{s c}, c$ c $F F$, and d) PCE with and without encapsulation.
spin-coated at 5000 rpm for 60 s . The two-side coated silicon wafer was placed into the ALD chamber and $99 \%$ trimethylaluminum $\left(\mathrm{Al}\left(\mathrm{CH}_{3}\right)_{3}\right.$, TMA) used as the aluminum source of ALD reaction with deionized (DI) water used as the oxidizer. High purity nitrogen was used as the carrier gas and to purge the chamber. A 20 ms pulse time for the TMA and DI Water was used and a 20 s purge time. The substrate temperature for deposition was $90^{\circ} \mathrm{C}$ and the number of cycles was used to control the thickness. A cycle of $40 \mathrm{~s}, 45$ times afforded an alumina film thickness of 10 nm , the thickness of alumina film was $\approx 10 \mathrm{~nm}$.

Solar Cells: Device fabrication involved several steps: 1) wet chemistry, including texturing and cleaning of wafer surface with the method found elsewhere; ${ }^{[38,39]} 2$ ) back surface polishing using a $\mathrm{HNO}_{3} / \mathrm{HF} / \mathrm{H}_{2} \mathrm{O}$ (4:1:2) mixed solution, which was highly associated with the final CNT:Nafion film morphology and hence important to reproducibly achieve high performance; 3) the formation of the front electron-selective passivating contact, antireflection layer, and metallization; 4) spin coating of a c-PC CNT:Nafion thin-film on the relatively flat back side using a CNT/Nafion ink under $\mathrm{N}_{2}$ atmosphere at room temperature. 5) thermal evaporation of a 300 nm full contact Ag electrode on CNT:Nafion thin film, coated by a Ag paste (SPI Supplies, 05001) was applied for encapsulation. On the front side, a-Si:H( $\left.\mathrm{n}^{+}\right) / \mathrm{a}-\mathrm{Si}: \mathrm{H}$ stack layer was deposited using a plasma-enhanced chemical vapor deposited (PECVD) equipment on the textured front surface of n-type wafers, and then an ITO layer was deposited by magnetron sputtering on the top of the a- $\mathrm{Si}: \mathrm{H}\left(\mathrm{n}^{+}\right) \cdot{ }^{[40,41]}$ Subsequent metallization of front surface was performed by screen-printing with an Ag paste, followed by a curing process at $170^{\circ} \mathrm{C}$ for 40 min . This reproducibility of the final device performances was enabled by the spin-coating method, stable back surface wet chemistry, the preparation of the ink with the accurate CNTNafion ratio, and the water resistance encapsulation.

Characterization: The lifetime was measured by spin-coating CNT:Nafion films on both sides of floated-zone Si wafer ( $400 \mu \mathrm{~m}$, n-type,
$3-5 \mathrm{k} \Omega \bullet \mathrm{cm})$ double-sided mirror-polished surfaces with an Sinton WCT-120 instrument. A "transparent plastic box" with gas switches that could be directly mounted onto the WCT-120 measurement stage to provide a stable dry air atmosphere. ${ }^{[32]}$ The performances of the solar cells were characterized by current density-voltage ( $J-V$ ) measurements under standard test conditions (AM $1.5,100 \mathrm{~mW} \cdot \mathrm{~cm}^{-2}$ and $25^{\circ} \mathrm{C}$ ) and the EQE (R3011, Enlitech).

## Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

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## Conflict of Interest

The authors declare no conflict of interest.

## Data Availability Statement

Research data are not shared.

## Keywords

organic passivation, photovoltaics, silicon solar cells, single-walled carbon nanotube (SWCNT)

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