RESEARCH ARTICLE | APRIL 10 2023

Natural virtual orbitals for the *GW* method in the randomphase approximation and beyond

Laurenz Monzel ; Christof Holzer ; Wim Klopper

Check for updates J. Chem. Phys. 158, 144102 (2023) https://doi.org/10.1063/5.0144469



Articles You May Be Interested In

A National Virtual Observatory (NVO) Science Case: Properties of Very Luminous IR Galaxies (VLIRGs)

AIP Conference Proceedings (May 2003)

Simple eigenvalue-self-consistent $\Delta^-\,G\,W$

- J. Chem. Phys. (November 2018)
- A "moment-conserving" reformulation of GW theory
- J. Chem. Phys. (March 2023)





Natural virtual orbitals for the *GW* method in the random-phase approximation and beyond

Cite as: J. Chem. Phys. 158, 144102 (2023); doi: 10.1063/5.0144469 Submitted: 30 January 2023 • Accepted: 23 March 2023 • Published Online: 10 April 2023



Laurenz Monzel,¹ (D) Christof Holzer,² (D) and Wim Klopper^{1,a}) (D)

AFFILIATIONS

¹ Institute of Physical Chemistry, Karlsruhe Institute of Technology (KIT), Fritz-Haber-Weg 2, 76131 Karlsruhe, Germany ² Institute of Theoretical Solid State Physics, Karlsruhe Institute of Technology (KIT), Wolfgang-Gaede-Straße 1, 76131 Karlsruhe, Germany

^{a)}Author to whom correspondence should be addressed: klopper@kit.edu

ABSTRACT

The increasingly popular *GW* method is becoming a convenient tool to determine vertical ionization energies in molecular systems. However, depending on the formalism used and the range of orbitals investigated, it may be hampered by a steep computational scaling. To alleviate this issue, correlated natural virtual orbitals (NVOs) based on second-order Møller–Plesset (MP2) and direct MP2 correlation energies are implemented, and the resulting correlated NVOs are tested on *GW* quasiparticle energies. Test cases include the popular *GW* variants $G_0 W_0$ and ev*GW*₀ as well as more elaborate vertex corrections. We find that for increasingly larger molecular systems and basis sets, NVOs considerably improve efficiency. Furthermore, we test the performance of the truncated (frozen) NVO ansatz on the GW100 test set. For the latter, it is demonstrated that, using a carefully chosen truncation threshold, NVOs lead to a negligible loss in accuracy while providing speedups of one order of magnitude. Furthermore, we compare the resulting quasiparticle energies to very accurate vertical ionization energies obtained from coupled-cluster theory with singles, doubles, and noniterative triples [CCSD(T)], confirming that the loss in accuracy introduced by truncating the NVOs is negligible compared to the methodical errors in the *GW* approximation. It is also demonstrated that the choice of basis set impacts results far more than using a suitably truncated NVO space. Therefore, at the same computational expense, more accurate results can be obtained using NVOs. Finally, we provide improved reference CCSD(T) values for the GW100 test set, which have been obtained using the def2-QZVPP basis set.

© 2023 Author(s). All article content, except where otherwise noted, is licensed under a Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/). https://doi.org/10.1063/5.0144469

I. INTRODUCTION

Correlated natural virtual orbitals (NVO)¹ are an emerging class of orbitals that diagonalize a correlated density matrix spanning the full virtual orbital space. Most prominently, second-order Møller–Plesset perturbation theory (MP2) has been used for the construction of the virtual part of the density matrix.^{2,3} Another viable alternative that can provide correlated density matrices at a rather low cost is the random-phase approximation (RPA).⁴ It has further been shown that using only a truncated subset of these MP2-based NVOs is a well-suited approach for obtaining correlation energies or properties from higher-order correlated methods, allowing for truncation of the space spanned by the virtual orbitals. MP2-based NVOs have since been applied to coupled-cluster (CC) theory with singles–and–doubles (CCSD),⁵ with perturbative triples

[CCSD(T)],⁶⁻¹¹ and in equation-of-motion-based CC (EOM-CC) and linear-response CC methods.¹²⁻¹⁴ Recently, MP2-based NVOs have also been applied to two- and four-component relativistic electron-correlation methods.¹⁵

Another interesting application for correlated NVOs is the GW approximation for molecules.^{16–18} In the GW approximation, the self-energy is calculated to correct a set of Kohn–Sham (KS) orbital energies to yield quasiparticle energies. The latter are closely related to vertical ionization energies (VIEs). Obtaining the correlation part of the self-energy is however not a trivial task. Within the random-phase approximation, a naive ansatz scales as N^6 with molecular size, which is roughly the same scaling behavior as CCSD. While certain techniques have been developed to lower this high computational effort, ^{19–24} they are usually specific to a single variant of GW, namely, with the response calculated in the direct RPA approximation

(dRPA), neglecting all exchange terms. Furthermore, these approximations are often limited to only a few quasiparticle states, either drastically losing either efficiency or accuracy if full quasiparticle spectra are to be extracted. Also, specific vertex corrections are not accessible using these low-scaling GW variants. In many molecular applications, a GW calculation is only the beginning of a larger series of calculations, including follow-up calculations. Typical examples of follow-up calculations are simulations of optical excitation spectra using the Bethe–Salpeter equation,^{22,25–29} transport simulations,^{30–33} or computations of further molecular properties.^{34–36} When these types of calculations are performed a posteriori, it is questionable to only use partial quasiparticle spectra, and instead full ones are desirable. Therefore, the GW step is often the time-limiting step, especially if more than a few quasiparticle states are to be computed. Reducing the virtual expansion space can then significantly speed up the GW step, which has been recognized early as a viable path to obtaining fast yet reliable results. Previous techniques employed to reduce the virtual expansion space in GW, for example, focused on directly optimizing the virtual orbital space.³⁷ In solid-state applications, attempts to reduce the expansion space in a GW calculations have also been applied successfully.^{38,39} A straightforward reduction of the virtual space is however connected to many caveats, as the GW self-energy generally converges only slowly with respect to the number of empty states. Therefore, truncation-based approaches are usually accompanied by extrapolation techniques or other error compensation approaches.^{37,38} Still, the convergence of truncated virtual spaces is still problematic.^{40,41} Like coupled-cluster theory, GW methods would see significant benefits from being able to truncate the virtual space drastically. In this study, we will demonstrate that correlated NVOs provide a convenient route toward truncating the virtual space. Viable paths to use truncated NVO spaces in GW calculations will be outlined, and the impact of certain truncation parameters will be discussed based on carefully assessed test sets.

II. THEORY

Unlike in correlated wave-function theory, as, for example, in coupled-cluster or configuration-interaction theory, the starting point for a *GW* calculation is usually a Kohn–Sham reference Slater determinant, not a Hartree–Fock (HF) reference determinant. Therefore, in Sec. II, all orbitals are assumed to be obtained from a converged KS calculation. While certainly there is a strong dependence of the *GW* results on the functional used to optimize the orbitals,^{42,43} in this study, we will show that the performance of NVOs is similar for all functional approximations.

A. Natural virtual orbitals from second-order perturbation theory

Natural orbitals in their most general form were introduced for the first time in 1955 by Löwdin as a set of molecular orbitals with the property of diagonalizing a one-electron density matrix.⁴⁴ First defined only for occupied orbitals, it was shown that post-HF calculations using natural orbitals could be brought to convergence significantly faster than with canonical orbitals. As shown by Davidson,¹ natural orbitals minimize the least square error between the canonical density matrix and its truncated approximate

$$f_{\rm err.} = \sum_{ab} |D_{ab} - \tilde{D}_{ab}|. \tag{1}$$

Therefore, they provide an optimal expansion space for truncation, as they guarantee the overall lowest possible error in the matrix elements of the density matrix. The choice of the density matrix is here decisive for the efficiency of the natural orbitals in the desired post-HF method for which there are various proposals in the literature. The most common examples of natural orbitals are the pair (or pseudo) natural orbitals (PNOs),^{45–47} natural transition orbitals (NTOs),⁴⁸ and MP2 natural orbitals.^{49,50} While NTOs are mainly used for the calculation of excited state properties,⁵¹ PNOs and MP2-NOs are primarily used for ground state calculations. Whereas PNOs are constructed for pairs of occupied HF orbitals, MP2-NOs are constructed from the total first-order wave function.

To obtain natural virtual orbitals (NVOs) at the MP2 level of theory, the construction of the virtual-virtual block of the (RI-)MP2 density matrix^{49,50,52,53} is the most central step. This virtual-virtual block is given by

$$D_{ab}^{\rm MP2} = \frac{1}{2} \sum_{ijc} \frac{\left[(ai|cj) - (ci|aj) \right] \left[(ib|jc) - (ic|jb) \right]}{(\varepsilon_{ij} - \varepsilon_{ac})(\varepsilon_{ij} - \varepsilon_{bc})}, \qquad (2)$$

where $\varepsilon_{pq} = \varepsilon_p + \varepsilon_q$ denotes the sum of two orbital energies from two canonical spin-orbitals, and (pq|rs) is a two-electron integral in Mulliken notation. We denote occupied spin-orbitals with the indices *i*, *j*, *k*, ..., virtual spin-orbitals with *a*, *b*, *c*, ..., and general spin-orbitals with *p*, *q*, *r*, ... Note that in our implementation, the computed occupation numbers (i.e., eigenvalues) refer to the spinorbital-based density matrix in Eq. (2) not only in the case of an unrestricted Hartree–Fock (UHF) or Kohn–Sham (UKS) reference determinant but also in the case of a restricted Hartree–Fock (RHF) or Kohn–Sham (RKS) reference determinant.

The NVOs are then obtained from the eigenvectors N of the virtual–virtual density matrix \mathbf{D}^{MP2} of Eq. (2). We keep all vectors $\widetilde{\mathbf{N}}$ with occupation numbers larger than some prescribed threshold. In the next step, we build semi-canonical NVOs by diagonalizing the Fock matrix in the truncated space. The essential steps for the determination of the semi-canonical NVOs, collected in the matrix $\widetilde{\mathbf{C}}$, are summarized in Table I. The orbital energies of the semi-canonical NVOs are the diagonal elements of the matrix $\widetilde{\mathbf{c}}$. The matrix C contains the original HF or KS virtual spin-orbitals.

Note that the density matrix becomes block diagonal if the wave function is chosen to be an eigenfunction of the spin operator \hat{S}_z , as it is done in restricted or unrestricted HF/KS calculations. Then, the off-diagonal α - β blocks are zero, i.e., ($\mathbf{D}_{\alpha\beta} = \mathbf{D}_{\beta\alpha} = \mathbf{0}$). Accordingly, the density matrix can be diagonalized separately for the α - α and β - β blocks. The computational effort can be reduced even further in the closed-shell RHF/RKS case by exploiting the symmetry $\mathbf{D}_{\alpha\alpha} = \mathbf{D}_{\beta\beta}$. Otherwise, steps (1)–(5) must be carried out separately for the two spin cases. If the wave function is neither an eigenfunction of \hat{S}_z nor \hat{S}^2 , as it is the case when spin–orbit effects are included, also the off-diagonal blocks must be included in the calculation of \mathbf{D}^{MP2} . In the latter case, the resulting complex NVOs are suitable to be applied in two- or four-component relativistic calculations, or calculations in external magnetic fields.^{22,54–57} Direct TABLE I. Outline of the most essential steps in the determination of MP2 semi-canonical natural virtual orbitals.

(1) Computation of the virtual-virtual MP2 density matrix	$\mathbf{D}^{\mathrm{MP2}}$
(2) Diagonalization of the density matrix	$\mathbf{N}^{\dagger}\mathbf{D}^{\mathrm{MP2}}\mathbf{N} = \mathbf{n}$
(3) Selection of a reduced number of eigenvectors	$N \to \widetilde{N}$
(4) Construction of Fock matrix in truncated space	$\widetilde{\mathbf{F}} = \widetilde{\mathbf{N}}^{\dagger} \boldsymbol{\varepsilon} \widetilde{\mathbf{N}}$
(5) Diagonalization of the Fock operator in the truncated space	$\mathbf{U}^{\dagger} \widetilde{\mathbf{F}} \mathbf{U} = \widetilde{\boldsymbol{\epsilon}}$
(6) Construction of semi-canonical orbitals	$\widetilde{\mathbf{C}} = \mathbf{C} \widetilde{\mathbf{N}} \mathbf{U}$

MP2-based NVOs (denoted as dNVOs) can be obtained similarly by simply neglecting the exchange integrals when generating the MP2 density matrix using Eq. (2).⁵⁸ Note that a prefactor of two has to be applied additionally in this case. Equation (2) also determines the overall scaling of the NVO generation step, scaling as N^5 . For dNVOs, the scaling can be reduced to N^4 using Laplace transformations.⁵⁹ While MP2 density matrices are a convenient choice in localized orbital simulations, they are less common in periodic frameworks. For the latter, density matrices based on the RPA may be better suited. Virtual–virtual RPA density matrices, for example, occur in self-consistent RPA,⁶⁰ and their calculation scales similar to Laplace-transformed direct MP2.⁶¹

B. Correlation self-energies from the *CW* method within the random-phase approximation

A rather general way to obtain correlation energies in the direct random-phase approximation (dRPA) is the "plasmon formula"

$$E_{\rm c}^{\rm RPA} = {\rm tr}(\boldsymbol{\Omega} - {\bf A}), \qquad (3)$$

where Ω is a matrix containing all positive dRPA excitation energies (see, for example, Ref. 62). The latter can be obtained from the eigenvalue problem

$$\begin{pmatrix} \mathbf{A} & \mathbf{B} \\ -\mathbf{B}^* & -\mathbf{A}^* \end{pmatrix} \begin{pmatrix} \mathbf{X} & \mathbf{Y}^* \\ \mathbf{Y} & \mathbf{X}^* \end{pmatrix} = \begin{pmatrix} \mathbf{X} & \mathbf{Y}^* \\ \mathbf{Y} & \mathbf{X}^* \end{pmatrix} \begin{pmatrix} \mathbf{\Omega} & \mathbf{0} \\ \mathbf{0} & -\mathbf{\Omega} \end{pmatrix}, \quad (4)$$

where **X** and **Y** satisfy the condition $\mathbf{X}^{\dagger}\mathbf{X} - \mathbf{Y}^{\dagger}\mathbf{Y} = \mathbf{1}$. Within the dRPA, the orbital rotation matrices **A** and **B** are given as

$$A_{ia,jb} = (\varepsilon_a - \varepsilon_i)\delta_{ab}\delta_{ij} + \nu_{ia,jb},$$
(5)

$$B_{ia,jb} = v_{ia,bj},\tag{6}$$

where ε_p refers to a KS orbital energy. Within the direct RPA, the interaction kernel $v_{pq,rs} = (pq|sr)$ is just the bare Coulomb interaction.

We now turn to the self-energy within the GW approximation. As in our previous work, ^{36,63} the GW self-energy⁶⁴

$$\Sigma(\mathbf{x}, \mathbf{x}'; \omega) = -\frac{1}{2\pi i} \int_{-\infty}^{\infty} \exp(i\omega' 0^+) W(\mathbf{x}, \mathbf{x}'; \omega') G(\mathbf{x}, \mathbf{x}'; \omega + \omega') d\omega'$$
(7)

is calculated from Green's function

$$G(\mathbf{x}, \mathbf{x}'; \omega) = \sum_{q} \frac{\phi_q(\mathbf{x})\phi_q^*(\mathbf{x}')}{\omega - \varepsilon_q + \mathrm{i}\delta \operatorname{sgn}\left(\varepsilon_q^{\mathrm{F}}\right)}$$
(8)

and the screened interaction

$$W(\mathbf{x},\mathbf{x}';\omega) = \int \varepsilon^{-1}(\mathbf{x},\mathbf{x}'';\omega) \,\nu(\mathbf{r}''-\mathbf{r}') \,\,\mathrm{d}\mathbf{x}'',\tag{9}$$

where $\varepsilon_q^{\rm F} = \varepsilon_q - \varepsilon^{\rm F}$ is the difference between the energy ε_q of the spin-orbital (or spinor) ϕ_q and the Fermi level $\varepsilon^{\rm F}$. 0⁺ describes an infinitesimally small, positive number. Note that the self-energy as given in Eq. (7) contains an exchange part as well as a correlation part. Using Eq. (4), and subtracting the Coulomb potential ν from the screened interaction W, the correlation part of the self-energy can be rewritten as

$$\Sigma_{c}(\mathbf{x}, \mathbf{x}', \omega) = \sum_{k} \sum_{m \neq 0} \frac{V_{m}(\mathbf{x}) V_{m}^{*}(\mathbf{x}') \phi_{k}(\mathbf{x}) \phi_{k}^{*}(\mathbf{x}')}{\omega + \omega_{m} - \varepsilon_{k} - i\eta} + \sum_{c} \sum_{m \neq 0} \frac{V_{m}^{*}(\mathbf{x}) V_{m}(\mathbf{x}') \phi_{c}(\mathbf{x}) \phi_{c}^{*}(\mathbf{x}')}{\omega - \omega_{m} - \varepsilon_{c} + i\eta}, \quad (10)$$

where $\eta = 2\delta$, and where the fluctuation potential $V_m(\mathbf{x})$ is determined for all excited states *m* as

$$V_m(\mathbf{x}) = \int_{-\infty}^{\infty} \nu(\mathbf{r} - \mathbf{r}') \rho_m(\mathbf{x}') \, \mathrm{d}\mathbf{x}', \qquad (11)$$

$$\rho_m(\mathbf{x}) = \sum_{ia} \left[\phi_a^*(\mathbf{x}) \phi_i(\mathbf{x}) X_{ia}^m + \phi_i^*(\mathbf{x}) \phi_a(\mathbf{x}) Y_{ia}^m \right].$$
(12)

Inserting Eqs. (11) and (12) into Eq. (10) yields the final expression for the expectation value of the correlation self-energy,

$$\langle p | \Sigma_{c}(\varepsilon_{p}) | p \rangle = \sum_{k} \sum_{m \neq 0} (pk|\rho_{m}) (pk|\rho_{m})^{*} D_{p,k,m}^{+} + \sum_{c} \sum_{m \neq 0} (cp|\rho_{m}) (cp|\rho_{m})^{*} D_{p,c,m}^{-},$$
(13)

with the element $D_{p,q,m}^{\pm}$ being defined as

$$D_{p,q,m}^{\pm} = \frac{\varepsilon_p - \varepsilon_q \pm \omega_m}{(\varepsilon_p - \varepsilon_q \pm \omega_m)^2 + \eta^2}.$$
 (14)

The two-electron integral $(pq|\rho_m)$ is computed as

$$(pq|\rho_m) = \sum_{ia} \left[(pq|ai) X_{ia}^m + (pq|ia) Y_{ia}^m \right].$$
(15)

GW correlation self-energies are tedious to compute, needing to fully solve Eq. (4). This severely limits the accessible molecular systems for GW.

C. Introducing natural virtual orbitals in the *CW* method

From Eq. (10), the GW quasiparticle energy can be calculated as

$$\varepsilon_p^{\rm QP}(\omega) = \varepsilon_p + Z \langle \phi_p | \Sigma^{\rm x} + \Sigma_{\rm c}(\omega) - V^{\rm KS} | \phi_p \rangle, \tag{16}$$

where $\varepsilon_p^{QP}(\omega)$ marks p-th quasiparticle state with an energy of ω . At the zeroth iteration ω corresponds to the KS orbital energy ε_p . Σ^X is the Hartree–Fock exchange energy of the p-th orbital, and V_p^{KS} the KS potential of the p-th orbital. *Z* is a linearization factor used to circumvent self-consistent solutions of Eq. (16), which would otherwise be necessary already for $G_0 W_0$.¹⁷ To reintroduce *G* and *W* explicitly, we rewrite Eq. (16) as

$$\varepsilon_{p}^{\text{QP}}(\omega) = \varepsilon_{p} + Z\left(\phi_{p}\left|\sum_{\substack{\Sigma^{x} \\ Variant \ 3}} \underbrace{\nabla_{\text{Variant } 2}}_{\text{Variant } 3} - V^{\text{KS}}\left|\phi_{p}\right\right)\right). \quad (17)$$

Depending on where NVOs are used for the calculation of the correlation self-energy, three different variants of the NVO-GW method can be derived as outlined in Eq. (17). In the first variant, we restrict ourselves by only representing the screened interaction W using NVOs. This approach is equivalent to using NVOs exclusively for the charge fluctuation $\rho_m(\mathbf{x})$ by substituting the virtual orbitals (or spinors) $\phi_c(\mathbf{x})$ by their NVO counterparts $\widetilde{\phi}_c(\mathbf{x})$ in Eq. (12), as well as in the eigenvalue problem in Eq. (4). This transformation of W is crucial to all approaches if NVOs are to be used since the solution of the eigenvalue problem in Eq. (4) is associated with the greatest computational effort. Note that in Eq. (13), in addition to the charge fluctuation $\rho_m(\mathbf{x})$, also the energy eigenvalues ω_m in $D_{p,q,m}^{\pm}$ must be replaced by the eigenvalues obtained from Eq. (4) using NVOs. If besides the screened interaction W also Green's function G is represented by NVOs, marked as variant 2 in Eq. (17), the whole correlation self-energy Σ_c is computed in the NVO space. In this case, also the sum over all virtual orbitals $\phi_c(\mathbf{x})$ in the second term on the right-hand side of Eq. (13) is replaced by NVOs as well as the corresponding orbital energies ε_c in $D^-_{p,c,m}$. In the third variant, in addition to the self-energy, also the orbitals or spinors $\phi_p(x)$, for which the self-energy shall be computed, are transformed into a basis of natural virtual orbitals. This variant is relevant only if the considered orbital $\phi_p(x)$ in Eq. (13) denotes an unoccupied orbital $\phi_c(\mathbf{x})$ because, otherwise, it is equivalent with the second variant. While the first and second variants require the parallel use of two basis sets during the GW calculation, the third approach considerably simplifies the technicalities of implementing the approach. In the latter, the virtual canonical orbitals are simply replaced by the NVOs throughout all necessary steps. However, the eigenvalues of the NVOs are considerably worse approximations of the quasiparticle energies, and therefore we mainly opt to use NVOs to evaluate Eqs. (4) and (12), that is, variant 1. The remaining terms in the correlation self-energy given in Eq. (13) are obtained using the canonical KS orbitals. Variant 1 is generally also compatible with self-consistent variants as for example, $evGW_0$, during which G is updated but W is kept at its initial state. Introducing

self-consistency in W through full eigenvalue self-consistent GW (evGW) is however troublesome using NVOs. Starting from the second iteration, the eigenvalues used in the construction of W are no longer the NVO eigenvalues, but the evGW ones, which are, in general, not compatible with the NVO eigenfunctions. This leads to significant errors in the unoccupied quasiparticle states. Due to the iterative nature, the errors in the unoccupied quasiparticle states subsequently also disturb the occupied quasiparticle states. Quasiparticle self-consistent GW (qsGW)⁶⁵ is currently beyond the scope of this work, as it would require to also update the NVOs during the iterative process. We note in passing that, contrary to evGW, the use of NVO for qsGW is nevertheless well-defined, as long as they are updated in each iteration. From a methodical scaling point of view, classical GW variants scale as N^6 with system size. Therefore, using NVOs pays off given that they can be determined with at most N^5 cost. For low-scaling GW algorithms, scaling as N^4 or better, NVO is less rewarding. Nevertheless, the possibility of applying low-scaling GW algorithms is mostly limited to direct RPA response, while NVOs in combination with Eqs. (4) and (13) are generally applicable. We further note that the construction of the (d)MP2 density matrix can be made local, reducing its effective scaling drastically.66

D. Vertex corrections

Vertex corrections are the next step on the GW hierarchy of methods, surpassing the simple approach of using solely Coulomb interactions from the direct random-phase approximation.^{67–70} Unfortunately, calculating vertex corrections is computationally demanding, yielding rather complicated expressions. While it is yet to be determined if vertex corrections improve VIEs obtained from the GW method,⁷¹ current investigations of transition-metal monoxides have revealed some importance for certain molecular systems.⁷² The high costs of calculating vertex corrections, however, severely hamper further investigations, being unfeasible for molecular systems of chemically relevant size. As NVOs provide a convenient route to reducing the cost of obtaining vertex corrections, we will expand our discussion by including NVOs in their computation.

In Ref. 68, Grüneis *et al.* use ordered second-order diagrams to obtain the vertex correction $\Delta \varepsilon_p(\omega)$ as

$$\Delta \varepsilon_{p}(\omega) = \sum_{i,a,b} \frac{(ia|W|pb)(ib|W|pa)^{*}}{\varepsilon_{a} + \varepsilon_{b} - \varepsilon_{i} - \omega} - \sum_{i,j,a} \frac{(ia|W|jp)(ip|W|ja)^{*}}{\varepsilon_{a} + \omega - \varepsilon_{i} - \varepsilon_{j}}.$$
(18)

W in Eq. (18) refers to the full static screened interaction as described in Eq. (9), where $\omega = 0$ is set. The latter choice is termed test-charge–test-charge (TC–TC) interaction, and labeled $W^{\text{TC-TC}}$ in Ref. 68. Evaluating Eq. (18) scales as N^6 , providing a steep increase in computational demands for *GW*. Using the resolution-of-the-identity approximation for the construction of $W^{\text{TC-TC}}$ can reduce this to N^5 ,²² and accordingly using NVOs can further accelerate the construction of vertex corrections. In our current implementation, we use NVOs for all orbital functionals as well as in the construction of $W^{\text{TC-TC}}$, yielding an efficient way to calculate vertex corrections.

scitation.org/journal/jcp

158, 144102-4

III. COMPUTATIONAL DETAILS

The NVO generation scheme outlined in Table I has been implemented into a locally modified version of TURBOMOLE 7.7, making use of OpenMP parallelization.73,74 The corresponding necessary modifications of the GW code of TURBOMOLE have been carried out in the course of this work. To test the performance of NVOs within the GW method, VIEs are obtained with NVOs using the linearized $G_0 W_0$ and the ev $G W_0$ approaches. As reference VIE values, the canonical results (obtained using canonical orbitals), will be used. This allows us to assess the deviations caused directly by truncating the NVO space. Furthermore, we compare the obtained VIE values, with and without using truncating the NVO space, to accurate CCSD(T) reference values. GW-based VIEs will be evaluated for three density-functional approximations (DFAs), namely, the global hybrid PBE0,75 the range-separated hybrid wB97X-D,⁷⁶ and the local hybrid TMHF.⁷⁷ Hybrid functionals, especially range-separated and local hybrid functionals, have earlier shown to yield excellent results for VIEs in combination with the $G_0 W_0$ This will allow one to conclude the transferability of method.44 NVOs between different reference functionals. All DFT calculations were tightly converged, with energy and density-matrix thresholds being set to $10^{-9} E_{\rm h}$ and 10^{-8} , respectively. Fine integration grids of size 4 or better were used to obtain fully converged results for all tested molecules and functionals.78 To determine the dependence of the NVO approximation on a given basis-set expansion, all $G_0 W_0$ and $ev G W_0$ computations were carried out using the triple- ζ def2-TZVPP basis set.⁷⁹ Additionally, for an initial investigation, the $G_0 W_0$ computations were also performed using the quadruple- ζ def2-QZVPP basis sets.⁷⁹ All DFT, NVO, and GW calculations were performed using a locally modified development version of TUR-BOMOLE 7.7.73 The resolution-of-the-identity approximation has been used in all NVO and GW calculations, and the corresponding def2-TZVPP and def2-QZVPP auxiliary basis sets optimized for post-HF calculations (denoted as cbas in TURBOMOLE jargon), were used.⁸⁰ See Ref. 80 and references therein for a detailed description of the auxiliary basis sets.

For the GW100 test set,⁸¹ we compare the computed VIEs obtained as GW quasiparticle energies with reference data obtained at the coupled-cluster level, including single and double substitutions as well as a perturbative correction for connected triple substitutions, that is, at the CCSD(T) level. The CCSD(T) reference values have been carefully re-assessed, and changes and new values are outlined in Sec. IV. All EOMIP-CCSD and EOMIP-CCSDT calculations have been performed using the CFOUR program.⁸² All of our primary reference data [geometries, UHF/def2-TZVPP, UHF/def2-QZVPP, CCSD(T)/def2-TZVPP, and CCSD(T)/def2-QZVPP energies, number of orbitals excluded from the correlation treatment] are published as supplementary data in the supplementary material.

IV. COUPLED-CLUSTER REFERENCE DATA

The CCSD(T)/def2-TZVPP reference data have already been published in an earlier work,⁸³ and in the present paper, we specifically report revised data for four molecules of the test set. Furthermore, CCSD(T)/def2-QZVPP reference data for the GW100 test set have been computed, and a full overview of the calculated VIEs is provided in the supplementary material. In our previous work,⁸³ the CCSD(T) computations were based on unrestricted Hartree-Fock (UHF) reference determinants, both for the cationic and neutral systems, and for all systems, we had attempted to find the UHF solution with the lowest energy. Recently, Bruneval, Dattani, and van Setten (BDvS) have repeated our CCSD(T)/def2-TZVPP calculations using a restricted Hartree-Fock (RHF) reference determinant for the neutral systems instead.⁸⁴ In the course of the latter work, BDvS found UHF solutions with lower energy for the cations of uracil (CAS Reg. No. 66-22.8), cytosine (71-30-7), and hydrogen cyanide (74-90-8). We have adopted these solutions in our present work. Furthermore, BDvS pointed out in their work that our previously reported value for carbonyl selenide (1603-84-5, OCSe) had erroneously been computed for carbonyl sulfide (in the OCSe geometry). In the present work, we report the correct values for the carbonyl selenide molecule.

After the above revisions, only two clear outliers remain when comparing our CCSD(T)/def2-TZVPP data with the data of BDvS. These outliers are sulfur dioxide (7446-09-5, SO₂) and magnesium oxide (1309-48-4, MgO), which will be discussed below. Excluding SO₂ and MgO from the comparison, the mean deviation of our values from those of BDvS amounts to 0.003 eV. The mean absolute deviation is 0.02 eV, and the sample standard deviation is 0.05 eV (note that data for the cesium dimer are missing from DBvS's supplementary data, such that the comparison is made for 97 molecules).

The problem with SO₂ is that we use the lowest-energy UHF solution for the SO_2^+ cation, which amounts to $-546.881967 E_h$, whereas BDvS use the solution with $E_{\text{UHF}} = -546.861914 E_{\text{h}}$. Admittedly, our lowest-energy solution is heavily spincontaminated $(\langle \hat{S}^2 \rangle = 1.13 \ \hbar^2)$ and the corresponding CCSD(T) energy does not seem accurate. The CCSD(T) energy obtained with BDvS's UHF solution is much closer to the CCSD(T) energy obtained with a restricted open-shell Hartree-Fock (ROHF) reference determinant for SO₂⁺ (Table II). Also the EOMIP-CCSD (equation-of-motion-ionization-potential coupled-cluster-singlesand-doubles) value⁸⁵ favors BDvS's CCSD(T) result.

The problem with MgO is that we use the lowest-energy UHF solution for the neutral molecule, whereas BDvS use the closedshell restricted Hartree-Fock (RHF) solution (Table III). Also in this case, the EOMIP-CCSD value of 8.172 eV for the VIE favors BDvS's CCSD(T) result (7.909 eV). We acknowledge that our value (7.487 eV) is very likely to be too small by about 0.4 eV and suggest using the EOMIP-CCSDT (EOMIP-CCSD-and-triples) values for the molecules SO₂ and MgO instead. Those VIE values amount to 12.225 and 7.877 eV for SO2 and MgO, respectively, as computed by the CFOUR program.⁸² In the def2-QZVPP basis set, the corresponding EOMIP-CCSDT values are 12.403 and 7.939 eV, respectively (cf supplementary material).

V. RESULTS

A. Ionization energies using NVOs

To test the capabilities of NVOs within the GW approximation, as a first test, we calculate quasiparticle energies for the GW100 test set. The latter consists of VIEs of 100 small molecules.

Figure 1 outlines the general convergence of the correlation self-energy as NVOs are subsequently added, starting from 20% to 100%, with the latter value exactly reproducing the untruncated

TABLE II.	Coupled-cluster	reference da	ta (total	energies ir	n E _h) for	SO_2^+	as	obtained	in the	def2-TZVPP	basis s	set. '	The
EOMIP-CO	CSD and EOMIP	-CCSDT ener	gies wei	re computed	with the	CFÕU	JR p	orogram. ⁸²	2				

Type of computation	Hartree-Fock	CCSD	CCSD(T)	VIE (eV)
UHF (this work) UHF (BDvS) ⁸⁴ ROHF EOMIP-CCSD EOMIP-CCSDT	-546.881 967 -546.861 914 -546.844 345	-547.461 26 -547.494 98 -547.493 36 -547.496 39	-547.488 62 -547.532 27 -547.532 89	13.486 12.298 12.281 12.373 12.225

TABLE III. Coupled-cluster reference data (total energies of in E_h) for MgO as obtained in the def2-TZVPP basis set. The EOMIP-CCSD and EOMIP-CCSDT energies were computed with the CFOUR program.⁸²

Type of computation	Hartree-Fock	CCSD	CCSD(T)	VIE (eV)
UHF (this work) RHF (BDvS) ⁸⁴ EOMIP-CCSD EOMIP-CCSDT	-274.464 527 -274.374 647	-274.813 34 -274.811 94	-274.82075 -274.83626	7.487 7.909 8.172 7.877

canonical value. First, it is visible that simply truncating the space spanned by the canonical orbitals in GW calculations is not a viable choice. The convergence toward the canonical result is slow, with any practically useable truncation parameter leading to unacceptable errors. Even including 80%-90% of the canonical orbitals does not yield converged results. Contrary, very good results are obtained if NVOs are used to calculate the self-energy Σ . In Fig. 1, NVOs are employed to evaluate Eqs. (4) and (12), consistent with variant 1 outlined in Sec. II A. Converged results are obtained if 30%-40% of the total number of virtual orbitals are included. Before this point, the error curve is less steep and the overall deviation therefore already largely reduced. This effect gets more prominent as more virtual orbitals are discarded. Note that the basis set size does not seem to be too important. The def2-QZVPP basis set shows deviations similar to those of the def2-TZVPP basis set, with the latter showing slightly reduced deviations in the case of a very low number of virtual orbitals taken into account.

Next, we will analyze the deviation introduced by NVOs dependent on the system size and the cardinality of the correlationconsistent basis set used.^{86,87} Therefore, the deviations of the VIEs in the acene series from benzene to hexacenes are shown in Fig. 2 for a 50% truncated virtual space. Furthermore, Fig. 2 also shows the deviations obtained for benzene with increasing basis set size.

Figure 2 exhibits that the deviation of the VIE normalized to Σ_C obtained in the fully intact virtual space decreases with the system size. The latter effect is to be expected as the virtual expansion size linearly increases in the row, providing more flexible combinations from which the NVOs can be generated. Moreover, Fig. 2 outlines how the deviation of the VIEs converges with an increasing number of basis functions for the benzene molecule. Again, raising the cardinal number of the basis set (adding a considerable amount of basis functions) provides a more flexible basis from which the NVOs can be generated. Discarding half of the virtual orbitals in each case therefore still leads to an increasingly efficient description of the

Deviation of VIE from full virtual space in (eV)



FIG. 1. Averaged deviation of the vertical ionization energy (VIE) for the GW100 test set of molecules, for a gradually decreasing truncated virtual orbital space. VIEs were obtained using the G_0W_0 @PBE0 method, using the def2-TZVPP and def2-QZVPP basis sets. The untruncated G_0W_0 @PBE0 values are used as references. All deviations in eV.

virtual space by the NVOs, significantly increasing the accuracy of NVO calculations. Overall, Fig. 2 points out at large molecules combined with large basis sets being a primary target for NVOs, with the latter working exceptionally well in these cases.

Table IV outlines the timings needed to build NVOs from scratch for the acenes shown in Fig. 2. Even on moderate computational resources, building NVOs for systems with up to 1000–1500 Cartesian basis functions (BFs) only takes seconds to a few minutes. Compared to classical N^6 scaling GW calculations, this is a vanishing



FIG. 2. Left: Deviation of vertical ionization energies (VIEs) for the series of acenes ranging from benzene to hexacene. Right: Deviation of the VIEs for benzene in the series of basis sets ranging from cc-pVDZ to cc-pV6Z. For the $G_0 W_0$ calculation, a PBE0 functional was used as a reference. The virtual orbital space was truncated consistently by 50%.

TABLE IV. Wall times needed to obtain the NVOs for the acenes from benzene to hexacene. Timings were measured on four cores of an AMD Ryzen 5500 CPU. All calculations were done using a PBE0/def2-TZVPP reference. Timings in seconds.

Acene	$N_{ m BF}$	T _{MP2} (s)
C ₆ H ₆	306	1.3
C10H8	480	11.3
$C_{14}H_{10}$	654	50.9
C ₁₈ H ₁₂	828	161
$C_{22}H_{14}$	1002	403
C ₂₆ H ₁₆	1176	910

effort. Still, for extensive systems, the overall $N^4 - N^5$ effort needed to construct canonical NVOs will be significant once 2000–3000 BF are surpassed. In this case, local methods to construct the (d)MP2 density matrix need to be employed to reduce the overall scaling in the NVO construction.

B. Ionization energies using NVOs for the GW100 test set

To analyze the overall deviation stemming from truncating the virtual orbitals space, we determine the error in the quasiparticle energies for the GW100 test set using various settings that determine how many virtual orbitals are neglected (frozen). Figures 3 and 4 outline a more detailed statistical analysis of the usage of NVOs in the GW100 test set for the def2-TZVPP and def2-QZVPP basis sets, respectively. Both plots show the obtained deviations for a 50% reduced virtual space and for three different occupation number thresholds that have been used to choose NVOs of lesser importance. Box plots were used to give a compact representation of the data distribution with respect to the full virtual orbital space. The horizontal orange line indicates the median and the upper and lower limits of the box (upper and lower quartiles) indicate that 25% of the data values are located above and below the quartile. The range of the box is the interquartile range (IQR). The whiskers extend to the data with values less than 1.5 times the interquartile range plus the lower or upper quartile.

First, comparing Figs. 3 and 4 suggests that a required occupation number threshold is dependent on the basis set size. For

example, for the def2-TZVPP basis set, a threshold of less than 10⁻⁴ below which NVOs are discarded leads to basically converged results. Contrary, the def2-QZVPP basis set requires a threshold of less than 5×10^{-5} to reduce the deviations found for the VIEs to a similar accuracy. For comparison, Figs. 3 and 4 also outline results with truncated canonical orbital spaces, where the number of active virtual orbitals has been chosen to match those of the NVOs. Figures 3 and 4 confirm the initial assumption that truncating canonical orbitals is a rather error-prone approach. For any truncation threshold, truncating canonical virtual orbitals yields a broad error distribution, being unsuitable to obtain quasiparticle energies. While truncating the NVO space is superior to truncating canonical virtual orbitals, both plots exhibit that the exact means of the generation of the NVOs are not too important. Quasiparticle energies obtained from NVOs generated from a direct-MP2 density matrix (dNVOs) are practically indistinguishable from their standard counterparts. Especially for larger molecules, where highperformance algorithms for Laplace-transformed direct MP2 have been developed,^{58,59} this can be translated in considerable gains while practically no accuracy is lost. Concerning the choice of the density functional approximation, no dependence of the NVOs on the underlying functional can be observed. Figure 5 depicts the fraction of the discarded virtual orbitals in dependence on the occupation number thresholds used in Figs. 3 and 4.

As outlined by Fig. 5, the tighter truncation thresholds for def2-QZVPP still lead to a higher relative number of NVOs that are discarded, translating into more pronounced speedups. To further check for the validity of NVOs in GW calculation, we also investigate the behavior in the $evGW_0$ method. In this method, specifically, the screened exchange W is kept at the KS level. Accordingly, W can be calculated using the NVOs obtained from the KS reference, and we update only the Greens function in every iteration.⁸⁸ Deviations found for the $evGW_0$ method, shown in Fig. 6, are in principle identical to the previously observed deviations for the $G_0 W_0$ approximation. The same conclusions drawn before for the truncated virtual spaces and difference density functional approximations therefore also apply to $evGW_0$. Using (d)NVOs can therefore help to considerably accelerate $evGW_0$. Notably, Fig. 6 shows that there is a single outlier for the smaller thresholds. This outlier is in every case helium, which reacts very sensitively to the omission of any orbitals in the GW treatment. For atomic cases, especially for helium, NVOs should therefore not be used.

Figure 7 depicts the statistical errors obtained for the VIE of the energetically lowest occupied orbital. This orbital corresponds to the 1s orbital when no effective core potential (ECP) is used, and to the first orbital not included in the ECP otherwise. For these vertical core ionization energies, a generally amplified magnitude of errors is observed. The latter is caused by the significantly raised ionization energies needed to expel core or non-valence electrons. Non-valence electron VIEs exhibit an increased sensitivity to the chosen occupation number threshold, with only the tighter thresholds yielding acceptable results. Notably, using canonical orbitals, the VIEs of these orbitals are strongly overestimated. In contrast, as outlined in Figs. 4-6, valence electron VIEs are strongly underestimated with truncated canonical orbitals. The exceedingly large error of VIEs when frozen canonical orbitals are used, reaching up to 10-20 eV, basically renders this approach useless for core ionization energies. Using NVOs, these errors can be largely reduced. The

ARTICLE

scitation.org/journal/jcp



FIG. 3. Deviation of the vertical ionization energies (VIEs) for a truncated virtual space using the GW100 test set of molecules. For the Kohn–Sham reference orbitals and the G_0W_0 calculations three functionals were considered (PBE0, TMHF, and ω B97X-D). The def2-TZVPP basis set was used. The damping parameter $\eta = 0.001$ was used.



FIG. 4. Deviation of the vertical ionization energies (VIEs) for a truncated virtual space using the GW100 test set of molecules. For the Kohn–Sham reference orbitals and the G_0W_0 calculations, three functionals were considered (PBE0, TMHF, and ω B97X-D). The def2-QZVPP basis set was used. The damping parameter $\eta = 0.001$ was used.

overall accuracy with frozen NVOs is then again comparable to the overall accuracy that can be reached with GW methods while speeding up the calculation by an order of magnitude or more. Notably, the outliers in Fig. 7 can be grouped into two classes. The first class is constituted by elements where ECPs have been employed, as e.g., I₂. The second class is built by elements with 1s VIEs exceeding 10 000 eV, for example, Br₂ or GeH₄. For orbitals close to an effective

core potential, using truncated NVOs is therefore not recommendable. For the second class, while larger than usual absolute errors are obtained, the relative error, which is also connected to the overall accuracy of *GW*, is very good. The presented approach can therefore also be recommended for core ionization energies.

Finally, the question remains of how much the VIEs are impacted at all by the use of a truncated set of virtual orbitals.





Figure 8 compares the VIEs obtained from CCSD(T)/def2-TZVPP, G_0W_0 @DFT/def2-TZVPP, and G_0W_0 @DFT/def2-QZVPP to the reference VIEs obtained from CCSD(T)/def2-QZVPP.

In Fig. 8, it is visible that NVOs are not lowering the overall accuracy of the $G_0 W_0$ method when compared to the results using the full virtual space. The overall error when compared to CCSD(T)/def2-QZVPP reference values is hardly affected at all, with the NVO/def2-TZVPP combination even exhibiting a slightly improved performance compared to a full virtual space. The latter effect can however safely be attributed to error cancellation. It is therefore important to point out two things. First, using NVOs one could accelerate a calculation using the same basis set, without a significant loss in accuracy. Second, using a truncated NVO space, quadruple- ζ results can be obtained basically at triple- ζ cost or less. As shown in Fig. 8, the accuracy improvement of going from triple- ζ to quadruple- ζ far outperforms any error introduced by the NVOs. Contrary, truncating the canonical orbital space significantly deteriorates the obtained VIEs. Figure 8 estimates that the average errors are 2-3 times larger in the latter case. Note that even CCSD(T), when used with a smaller basis set, is considerably in error. Deviations of up to 0.25 eV are found for the latter

basis set when compared to the reference values, with a mean error of ~0.1 eV.

C. Vertex corrections

Subsequently, to standard G_0W_0 and $evGW_0$ calculations, NVOs are especially interesting to speed up the calculation of otherwise rather expensive higher-order vertex corrections to the G_0W_0 or $evGW_0$ quasiparticle energies. Using NVOs, the virtual space can be truncated, which leads to large gains when evaluating Eq. (18). Especially the evaluation of the first term in Eq. (18), featuring a sum over two virtual indices, will benefit from truncating the virtual space. Of course, the question of how much NVOs will affect the results remains. It turns out that for vertex corrections the deviation introduced by using NVOs is again reduced quickly and smoothly with the occupation number threshold as outlined in Fig. 9.

Given that the magnitude of the vertex corrections $\Delta \varepsilon_p$ is usually much smaller than the initial correlation self-energy obtained from $G_0 W_0$ or ev $G W_0$, smaller deviations are obtained overall. Still, truncating the virtual space while using canonical orbitals leads to

ARTICLE

scitation.org/journal/jcp



FIG. 6. Absolute deviation of the vertical ionization energies (VIEs) for a truncated virtual space using the GW100 test set of molecules. For the Kohn–Sham reference orbitals and the evGW₀ calculations three functionals were considered (PBE0, TMHF, and ω B97X-D). The def2-TZVPP basis set was used. The damping parameter $\eta = 0.001$ was used.



FIG. 7. Deviation of the lowest calculated vertical ionization energies (VIEs) for a truncated virtual space using the GW100 test set of molecules. For organic molecules this refers to the 1s orbitals, if effective core potentials (ECP) were used, it refers to the first orbitals not enclosed in the ECP. For the Kohn–Sham reference orbitals and the $evGW_0$ calculations, three functionals were considered (PBE0, TMHF, and ω B97X-D). The def2-TZVPP basis set was used. The damping parameter $\eta = 0.001$ was used.

significant deviations. In the latter case, the incorporated error is of the same order of magnitude as the vertex corrections itself, yielding unacceptable deviations. Contrary, truncating the virtual space is possible without sacrificing much accuracy when NVOs are used. With NVOs the error is well under control, especially for lower occupation numbers. An occupation number threshold of 10^{-4} again seems to provide a reasonable balance between performance and accuracy, with the mean error on the vertex correction approaching 0.01–0.02 eV. The error distribution for vertex corrections is therefore in line with that of the quasiparticle energies themselves. Using the same thresholds for both, vertex corrections and quasiparticle energies, is therefore possible and advisable, yielding a consistent yet efficient and accurate method. As a final remark, it is interesting to note that for the lowest occupation number thresholds of 5×10^{-5} and 10^{-4} , the single largest outlier exhibiting double the error of the second largest one is again helium. The latter already showed to be troublesome for ev GW_0 and is, in general, known to be a challenging case for GW methods.⁸⁹ We, however, note that this is



FIG. 8. Comparison of vertical ionization energies (VIEs) from various methods to reference CCSD(T)/def2-QZVPP VIEs. For NVOs, a truncation threshold of 10⁻⁴ was chosen. For canonical orbitals, the same number of neglected virtual orbitals as in the truncated NVO space was assumed. Full virtual space labels untruncated results. All deviations in eV.



FIG. 9. Deviation of the vertex correction for vertical ionization energies (VIEs) for a truncated virtual space using the GW100 test set of molecules. TMHF was used as Kohn–Sham reference. The def2-TZVPP basis set was used. All deviations in eV.

only a minor setback, as truncating orbital spaces for a single atom is hardly ever necessary.

VI. CONCLUSION

We have outlined the principal capabilities of using the truncated natural virtual orbital spaces to accelerate the evaluation of quasiparticle energies from the *GW* approximation. For VIEs, we find that the NVO space can be considerably truncated without a significant loss in accuracy in the quasiparticle energies obtained from either $G_0 W_0$ or $evGW_0$. The truncation thresholds used to construct a set of NVOs are furthermore shown to be transferable between different classes of density functional approximations. However, it is not automatically transferable between basis sets, and most likely an optimal value needs to be determined for at least

J. Chem. Phys. **158**, 144102 (2023); doi: 10.1063/5.0144469 © Author(s) 2023 every cardinal number of orbital basis sets. We recommend truncating the space spanned by the NVOs by neglecting orbitals with occupation numbers of less than 10^{-4} for triple- ζ basis sets, and a slightly tighter threshold of 5 ×10⁻⁵ for quadruple- ζ basis sets. Using these thresholds leads to tightly converged quasiparticle energies, and also to tightly converged vertex corrections. The remaining error of using NVOs has then been shown to be insignificant compared to the overall error of the GW method. This also holds for non-valence orbitals, for example, 1s orbitals of lighter elements. For the latter, NVOs yield excellent quasiparticle energies, while largely reducing the cost of the associated GW calculation. It is furthermore interesting to note that the exact way the correlated NVOs are obtained is not too important. NVOs obtained from direct MP2 lead to practically indistinguishable result, while speeding up the process by an order of magnitude. For periodic GW calculations, where MP2 density matrices are not commonly available, one could resort to RPA density matrices, which have similar properties and are expected to work as well as dMP2 density matrices for NVO generation due to their principal similarity. Given the robustness of the NVO scheme, exhibiting no significant difference between MP2 and dMP2 density matrices, we also expect RPA density matrices to provide results of similar quality. These results are especially encouraging for vertex corrections, which are generally considered to be computationally demanding. The latter can be accelerated using NVOs to yield methods that are routinely applicable to molecular systems, allowing future investigations to take them into account when needed.

For electron attachment energies (not reported here), we find NVOs to be not suitable within the *GW* ansatz when the full virtual space is replaced by NVOs. The failure can be attributed to the fact that the virtual eigenvalues of the Fock matrix in the truncated space of the NVOs should not be used as starting points for the computation of the respective quasiparticle energies. The problem can be remedied by not replacing those virtual orbitals that are relevant for the electron attachment energies by NVOs. This must also be taken into account when computing excitation energies through the Bethe–Salpeter equation, in which quasiparticle energies of occupied and virtual levels are used as input.

SUPPLEMENTARY MATERIAL

See the supplementary material for the geometries used in the GW100 test set, the Hartree–Fock and CCSD(T) energies in the def2-TZVPP and def2-QZVPP basis sets for the cations as well as the neutral systems, and energies for the molecules SO₂, MgO, and their cations. Hartree–Fock and CCSD(T) energies in the def2-TZVPP and def2-QZVPP are additionally available as machine-readable ASCII files. Furthermore, the G_0W_0 and ev GW_0 results are also reported as machine-readable ASCII files.

ACKNOWLEDGMENTS

L.M. and W.K. gratefully acknowledge financial support by the Deutsche Forschungsgemeinschaft (DFG) through the Research Training Group 2450 on "Tailored Scale-Bridging Approaches to Computational Nanoscience." C.H. gratefully acknowledges the Volkswagen Stiftung for financial support.

AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

Laurenz Monzel: Data curation (lead); Formal analysis (lead); Software (lead); Visualization (lead); Writing – original draft (lead). Christof Holzer: Data curation (lead); Formal analysis (lead); Software (lead); Visualization (lead); Writing – original draft (lead). Wim Klopper: Conceptualization (equal); Data curation (equal); Formal analysis (equal); Project administration (equal); Supervision (equal); Writing – review & editing (equal).

DATA AVAILABILITY

The data that support the findings of this study are available within the article and its supplementary material.

REFERENCES

- ¹E. R. Davidson, Rev. Mod. Phys. 44, 451 (1972).
- ²C. Møller and M. S. Plesset, Phys. Rev. 46, 618 (1934).

³M. Pitoňák, F. Holka, P. Neogrády, and M. Urban, J. Mol. Struct.: THEOCHEM **768**, 79 (2006).

⁴B. Ramberger, Z. Sukurma, T. Schäfer, and G. Kresse, J. Chem. Phys. **151**, 214106 (2019).

⁵A. Grüneis, G. H. Booth, M. Marsman, J. Spencer, A. Alavi, and G. Kresse, J. Chem. Theory Comput. 7, 2780 (2011).

⁶W. Klopper, J. Noga, H. Koch, and T. Helgaker, Theor. Chem. Acc. **97**, 164–176 (1997).

⁷Z. Rolik, L. Szegedy, I. Ladjánszki, B. Ladóczki, and M. Kállay, J. Chem. Phys. 139, 094105 (2013).

⁸Z. Rolik and M. Kállay, J. Chem. Phys. 135, 104111 (2011).

⁹A. E. DePrince and C. D. Sherrill, J. Chem. Theory Comput. 9, 293 (2013).

¹⁰A. E. DePrince and C. D. Sherrill, J. Chem. Theory Comput. **9**, 2687 (2013).

¹¹ P. R. Nagy, L. Gyevi-Nagy, and M. Kállay, Mol. Phys. **119**, e1963495 (2021).

¹²A. Landau, K. Khistyaev, S. Dolgikh, and A. I. Krylov, J. Chem. Phys. 132, 014109 (2010).

¹³A. Kumar and T. D. Crawford, J. Phys. Chem. A **121**, 708 (2017).

¹⁴P. Pokhilko, D. Izmodenov, and A. I. Krylov, J. Chem. Phys. **152**, 034105 (2020).

¹⁵X. Yuan, L. Visscher, and A. S. P. Gomes, J. Chem. Phys. **156**, 224108 (2022).

¹⁶F. Aryasetiawan and O. Gunnarsson, Rep. Prog. Phys. 61, 237 (1998).

¹⁷ M. J. van Setten, F. Weigend, and F. Evers, J. Chem. Theory Comput. **9**, 232 (2013).

¹⁸ J. W. Knight, X. Wang, L. Gallandi, O. Dolgounitcheva, X. Ren, J. V. Ortiz, P. Rinke, T. Körzdörfer, and N. Marom, J. Chem. Theory Comput. **12**, 615 (2016).

¹⁹C. Friedrich, S. Blügel, and A. Schindlmayr, Phys. Rev. B 81, 125102 (2010).

²⁰ P. Liu, M. Kaltak, J. Klimeš, and G. Kresse, Phys. Rev. B **94**, 165109 (2016).

²¹ M. Govoni and G. Galli, J. Chem. Theory Comput. **11**, 2680 (2015).

²²C. Holzer and W. Klopper, J. Chem. Phys. **150**, 204116 (2019).

²³ I. Duchemin and X. Blase, J. Chem. Theory Comput. **16**, 1742 (2020).

²⁴I. Duchemin and X. Blase, J. Chem. Theory Comput. **17**, 2383 (2021).

²⁵F. Bruneval, S. M. Hamed, and J. B. Neaton, J. Chem. Phys. **142**, 244101 (2015).

²⁶K. Krause and W. Klopper, J. Comput. Chem. **38**, 383 (2017).

²⁷C. Holzer and W. Klopper, J. Chem. Phys. **149**, 101101 (2018).

²⁸X. Blase, I. Duchemin, and D. Jacquemin, Chem. Soc. Rev. 47, 1022 (2018).

²⁹X. Blase, I. Duchemin, D. Jacquemin, and P.-F. Loos, J. Chem. Phys. Lett. 11, 7371 (2020). ³⁰P. Darancet, A. Ferretti, D. Mayou, and V. Olevano, Phys. Rev. B **75**, 075102 (2007).

³¹ K. S. Thygesen and A. Rubio, Phys. Rev. B 77, 115333 (2008).

³²K. S. Thygesen and A. Rubio, J. Chem. Phys. **126**, 091101 (2007).

³³ M. Thoss and F. Evers, J. Chem. Phys. **148**, 030901 (2018).

³⁴ M. Kehry, Y. J. Franzke, C. Holzer, and W. Klopper, Mol. Phys. **118**, e1755064 (2020).

³⁵Y. J. Franzke, C. Holzer, and F. Mack, Chem. Theory Comput. 18, 1030 (2022).

³⁶C. Holzer, A. M. Teale, F. Hampe, S. Stopkowicz, T. Helgaker, and W. Klopper, J. Chem. Phys. **150**, 214112 (2019).

³⁷F. Bruneval, J. Chem. Phys. **145**, 234110 (2016).

³⁸F. Bruneval and X. Gonze, Phys. Rev. B 78, 085125 (2008).

³⁹J. Deslippe, G. Samsonidze, M. Jain, M. L. Cohen, and S. G. Louie, *Phys. Rev. B* 87, 165124 (2013).

⁴⁰ M. Stankovski, G. Antonius, D. Waroquiers, A. Miglio, H. Dixit, K. Sankaran, M. Giantomassi, X. Gonze, M. Côté, and G.-M. Rignanese, Phys. Rev. B 84, 241201 (2011).

⁴¹ M. J. van Setten, M. Giantomassi, X. Gonze, G.-M. Rignanese, and G. Hautier, Phys. Rev. B **96**, 155207 (2017).

⁴²F. Bruneval and M. A. L. Marques, J. Chem. Theory Comput. 9, 324 (2013).

⁴³C. Holzer, Y. J. Franzke, and M. Kehry, J. Chem. Theory Comput. 17, 2928 (2021).

⁴⁴P.-O. Löwdin, Phys. Rev. **97**, 1474 (1955).

⁴⁵C. Edmiston and M. Krauss, J. Chem. Phys. 45, 1833 (1966).

⁴⁶W. Meyer, J. Chem. Phys. 58, 1017 (1973).

⁴⁷R. Ahlrichs, H. Lischka, V. Staemmler, and W. Kutzelnigg, J. Chem. Phys. 62, 1225 (1975).

48 R. L. Martin, J. Chem. Phys. 118, 4775 (2003).

⁴⁹H. J. A. Jensen, P. Jørgensen, H. Ågren, and J. Olsen, J. Chem. Phys. 88, 3834 (1988).

⁵⁰ H. J. A. Jensen, P. Jørgensen, H. Ågren, and J. Olsen, J. Chem. Phys. 89, 5354 (1988).

⁵¹S. Höfener and W. Klopper, Chem. Phys. Lett. **679**, 52 (2017).

⁵²G. W. Trucks, E. A. Salter, C. Sosa, and R. J. Bartlett, Chem. Phys. Lett. 147, 359 (1988).

⁵³C. Hättig, A. Hellweg, and A. Köhn, Phys. Chem. Chem. Phys. 8, 1159 (2006).

⁵⁴L. Monzel, A. Pausch, L. D. M. Peters, E. I. Tellgren, T. Helgaker, and W. Klopper, J. Chem. Phys. **157**, 054106 (2022).

⁵⁵C. Holzer, A. Pausch, and W. Klopper, Front. Chem. **9**, 746162 (2021).

⁵⁶ A. Pausch, C. Holzer, and W. Klopper, J. Chem. Theory Comput. **18**, 3747 (2022).

⁵⁷A. Pausch and C. Holzer, J. Phys. Chem. Lett. 13, 4335 (2022).

⁵⁸Y. Jung, R. C. Lochan, A. D. Dutoi, and M. Head-Gordon, J. Chem. Phys. **121**, 9793 (2004).

⁵⁹M. Häser and J. Almlöf, J. Chem. Phys. **96**, 489 (1992).

⁶⁰F. Catara, G. Piccitto, M. Sambataro, and N. Van Giai, Phys. Rev. B 54, 17536 (1996).

⁶¹D. Graf, M. Beuerle, and C. Ochsenfeld, J. Chem. Theory Comput. 15, 4468 (2019).

⁶²K. Krause and W. Klopper, J. Chem. Phys. **139**, 191102 (2013).

⁶³C. Holzer, A. M. Teale, F. Hampe, S. Stopkowicz, T. Helgaker, and W. Klopper, J. Chem. Phys. 151, 069902 (2019).

⁶⁴L. Hedin, Nucl. Instrum. Methods Phys. Res., Sect. A 308, 169 (1991).

⁶⁵M. van Schilfgaarde, T. Kotani, and S. Faleev, Phys. Rev. Lett. **96**, 226402 (2006).

⁶⁶P. Pinski and F. Neese, J. Chem. Phys. **148**, 031101 (2018).

⁶⁷A. Schindlmayr and R. W. Godby, Phys. Rev. Lett. **80**, 1702 (1998).

⁶⁸A. Grüneis, G. Kresse, Y. Hinuma, and F. Oba, Phys. Rev. Lett. **112**, 096401 (2014).

⁶⁹M. Shishkin, M. Marsman, and G. Kresse, Phys. Rev. Lett. **99**, 246403 (2007).

- ⁷⁰E. Maggio and G. Kresse, J. Chem. Theory Comput. **13**, 4765 (2017).
- ⁷¹ A. M. Lewis and T. C. Berkelbach, J. Chem. Theory Comput. **15**, 2925 (2019).
- 72 Y. Wang and X. Ren, J. Chem. Phys. 157, 214115 (2022).

⁷³S. G. Balasubramani, G. P. Chen, S. Coriani, M. Diedenhofen, M. S. Frank, Y. J. Franzke, F. Furche, R. Grotjahn, M. E. Harding, C. Hättig, A. Hellweg, B. Helmich-Paris, C. Holzer, U. Huniar, M. Kaupp, A. Marefat Khah, S. Karbalaei Khani, T. Müller, F. Mack, B. D. Nguyen, S. M. Parker, E. Perlt, D. Rappoport, K. Reiter, S. Roy, M. Rückert, G. Schmitz, M. Sierka, E. Tapavicza, D. P. Tew, C. van Wüllen, V. K. Voora, F. Weigend, A. Wodyński, and J. M. Yu, J. Chem. Phys. **152**, 184107 (2020).

⁷⁴C. Holzer and Y. J. Franzke, OpenMP version of ridft, rdgrad, and egrad with contributions to mpshift, dscf, and grad; improved OpenMP version of aoforce and escf, released with TURBOMOLE V7.4 and further improved in TURBOMOLE V7.5.

⁷⁵J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996).

⁷⁶J.-D. Chai and M. Head-Gordon, Phys. Chem. Chem. Phys. **10**, 6615 (2008).

⁷⁷C. Holzer and Y. J. Franzke, J. Chem. Phys. 157, 034108 (2022).

78 O. Treutler and R. Ahlrichs, J. Chem. Phys. 102, 346 (1995).

⁷⁹F. Weigend and R. Ahlrichs, Phys. Chem. Chem. Phys. 7, 3297 (2005).

⁸⁰ A. Hellweg, C. Hättig, S. Höfener, and W. Klopper, Theor. Chem. Acc. 117, 587–597 (2007).

⁸¹ M. J. van Setten, F. Caruso, S. Sharifzadeh, X. Ren, M. Scheffler, F. Liu, J. Lischner, L. Lin, J. R. Deslippe, S. G. Louie, C. Yang, F. Weigend, J. B. Neaton, F. Evers, and P. Rinke, J. Chem. Theory Comput. **11**, 5665 (2015).

⁸²CFOUR, Coupled-Cluster techniques for Computational Chemistry—A Quantum-Chemical Program package by J. F. Stanton, J. Gauss, M. E. Harding, and P. G. Szalay with contributions from A. A. Auer, R. J. Bartlett, U. Benedikt, C. Berger, D. E. Bernholdt, Y. J. Bomble, L. Cheng, O. Christiansen, M. Heckert, O. Heun, C. Huber, T.-C. Jagau, D. Jonsson, J. Jusélius, K. Klein, W. J. Lauderdale, F. Lipparini, D. A. Matthews, T. Metzroth, L. A. Mück, D. P. O'Neill, D. R. Price, E. Prochnow, C. Puzzarini, K. Ruud, F. Schiffmann, W. Schwalbach, C. Simmons, S. Stopkowicz, A. Tajti, J. Vázquez, F. Wang, and J. D. Watts and the integral packages MOLECULE (J. Almlöf and P. R. Taylor), PROPS (P. R. Taylor), ABACUS (T. Helgaker, H. J. A. Jensen, P. Jørgensen, and J. Olsen), and ECP routines by A. V. Mitin and C. van Wüllen. For the current version, see http://www.cfour.de (accessed 02.11.2016).

83 K. Krause, M. E. Harding, and W. Klopper, Mol. Phys. 113, 1952 (2015).

⁸⁴F. Bruneval, N. Dattani, and M. J. van Setten, Front. Chem. 9, 749779 (2021).

⁸⁵M. F. Lange and T. C. Berkelbach, J. Chem. Theory Comput. 14, 4224 (2018).

⁸⁶K. A. Peterson, D. E. Woon, and T. H. Dunning, Jr., J. Chem. Phys. **100**, 7410 (1994).

⁸⁷A. K. Wilson, T. van Mourik, and T. H. Dunning, Jr., J. Mol. Struct.: THEOCHEM 388, 339 (1996).

⁸⁸T. Rangel, S. M. Hamed, F. Bruneval, and J. B. Neaton, J. Chem. Theory Comput. 12, 2834 (2016).

⁸⁹J. Li, M. Holzmann, I. Duchemin, X. Blase, and V. Olevano, Phys. Rev. Lett. 118, 163001 (2017).