



Radiation Effects in Americium Ceramic Compounds

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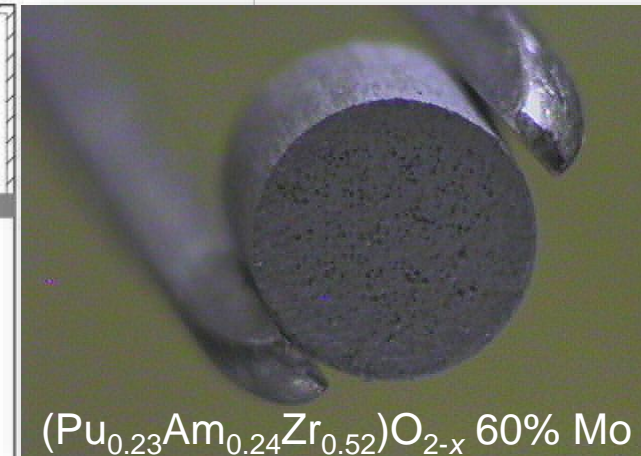
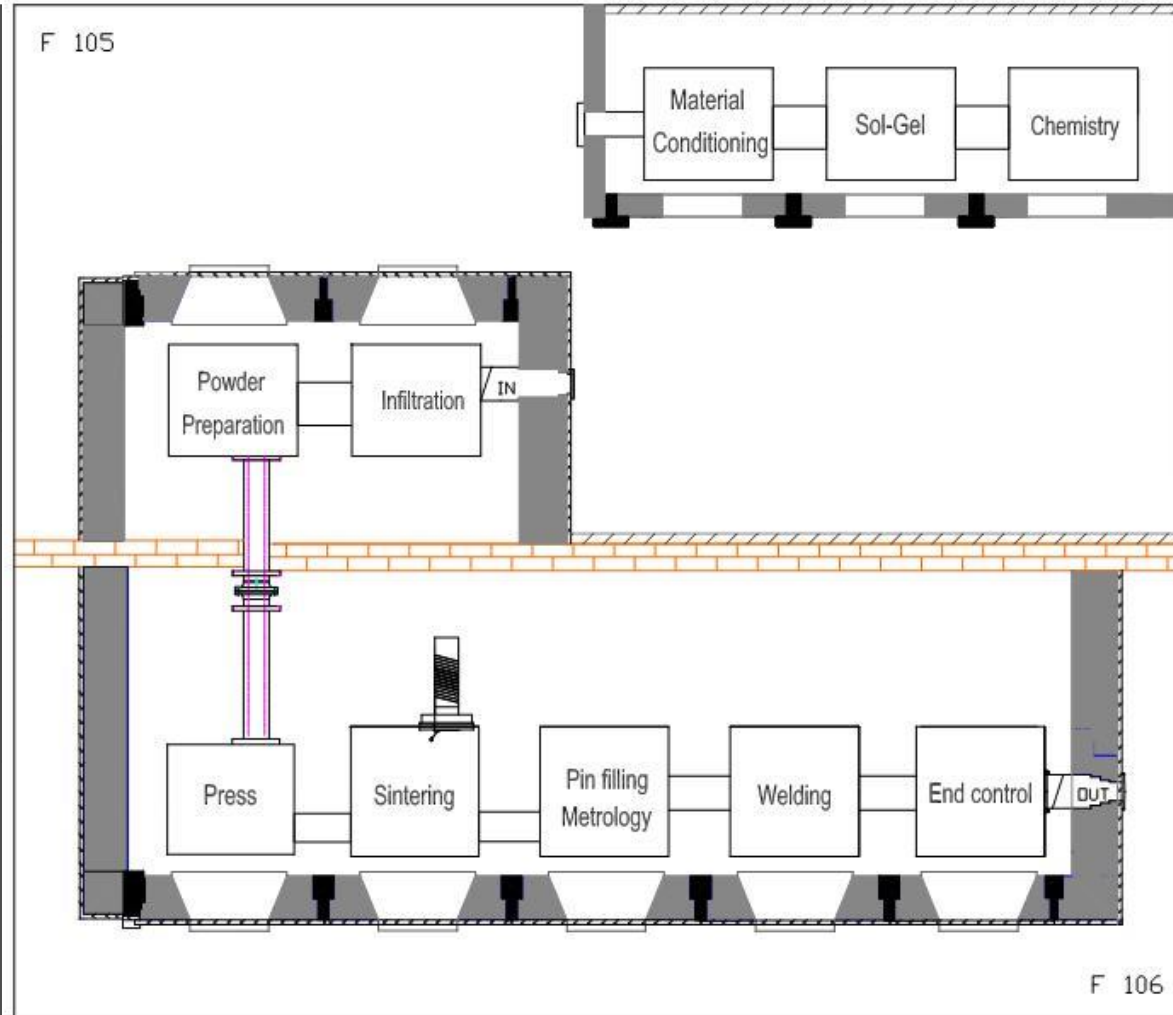
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The JRC Minor Actinide Laboratory

Isotope	Limiting mass (g) ^a	Criterion
²³¹ Pa	10	Shielding
²³⁷ Np	∞ ^b	
²⁴¹ Am	50	License ^c
^{242m} Am	0.1	Shielding
²⁴³ Am	65	License ^c
²⁴⁴ Cm	Isotope	Shielding & licence ^c

^a to yield max 2 μSv/h at 1 metre
^b no practical limit
^c corresponding to the dose equivalent of 200 g Pu in the form of powder (oxide)



Optimisation of uranium-doped americium oxide synthesis for space application

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Graphical Abstract

Uranium

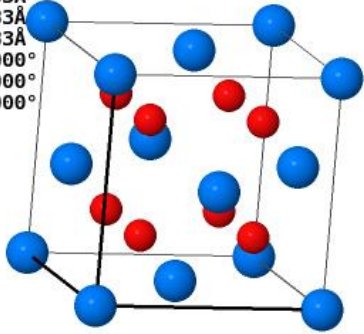
²⁴¹Am

100%

80%

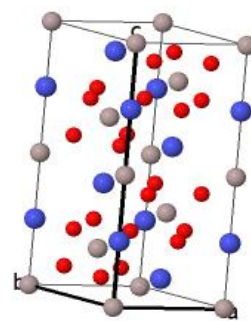
Am-ceramic compounds for radioisotope heater units and radioisotope thermoelectric generators

HM:F m -3 m #225
a=5.433Å
b=5.433Å
c=5.433Å
α=90.000°
β=90.000°
γ=90.000°



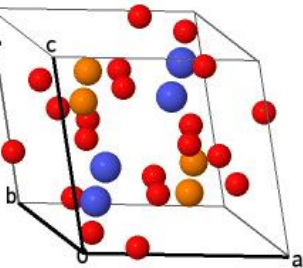
$\text{Am}_y\text{U}_{1-y}\text{O}_2$ (fluorite)

HM:R -3 c H #167
a=5.333Å
b=5.333Å
c=12.926Å
α=90.000°
β=90.000°
γ=120.000°



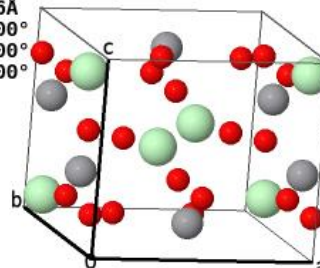
AmAlO_3 , AmVO_3 (perovskites)

HM:P 1 21/n 1 #14
a=6.753Å
b=6.967Å
c=6.433Å
α=90.000°
β=103.787°
γ=90.000°



AmPO_4 (monazite)

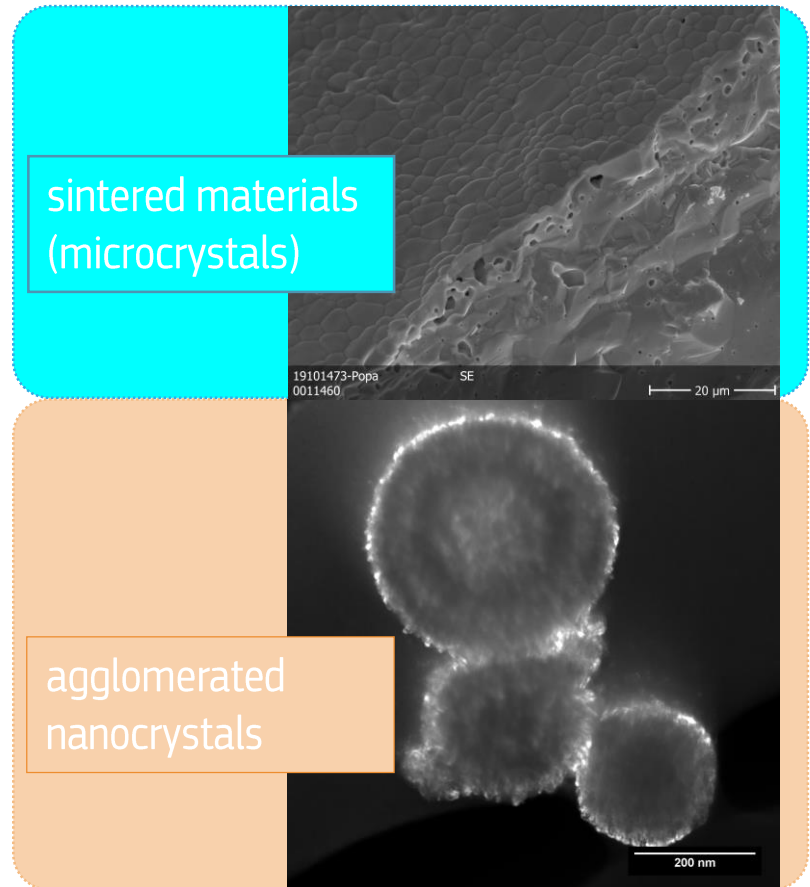
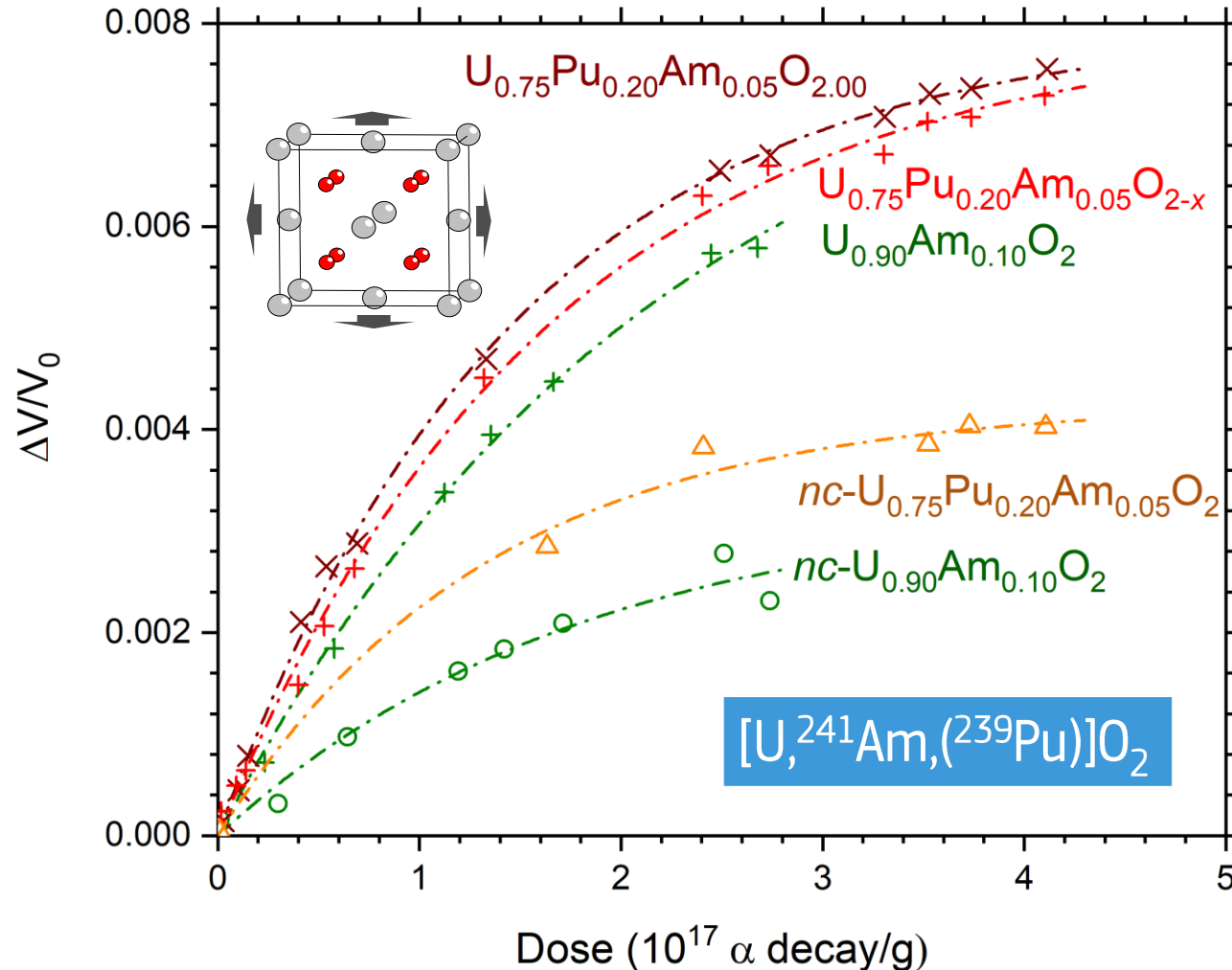
HM:I 41/a m d Z #141
a=7.331Å
b=7.331Å
c=6.436Å
α=90.000°
β=90.000°
γ=90.000°



AmVO_4 (zircon)

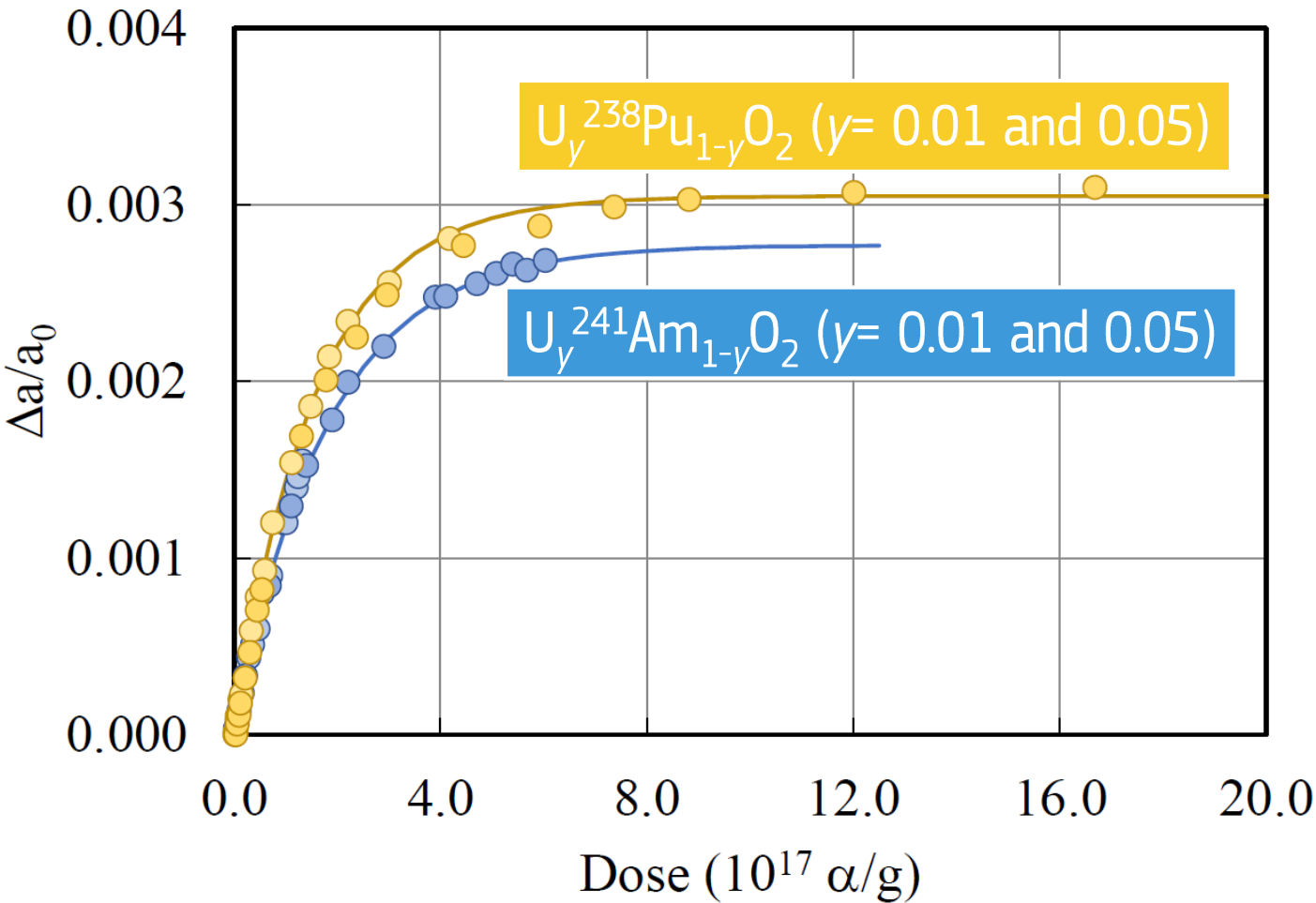
- Am-content
- **amorphization behaviour**
- **swelling degree**
- ^{237}Np -distribution
- natural analogue
- stability (HT/LT; air, inert atmosphere, vacuum, ...)

Swelling of the fluorite structure: influence of the particle size

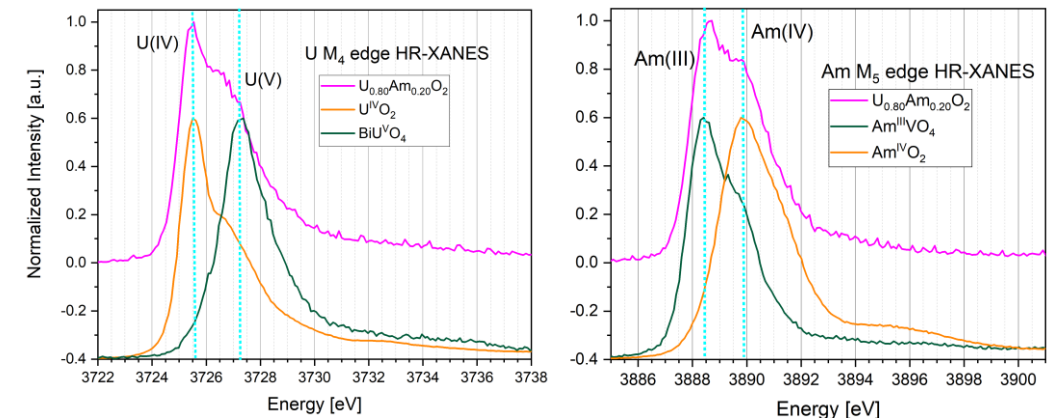
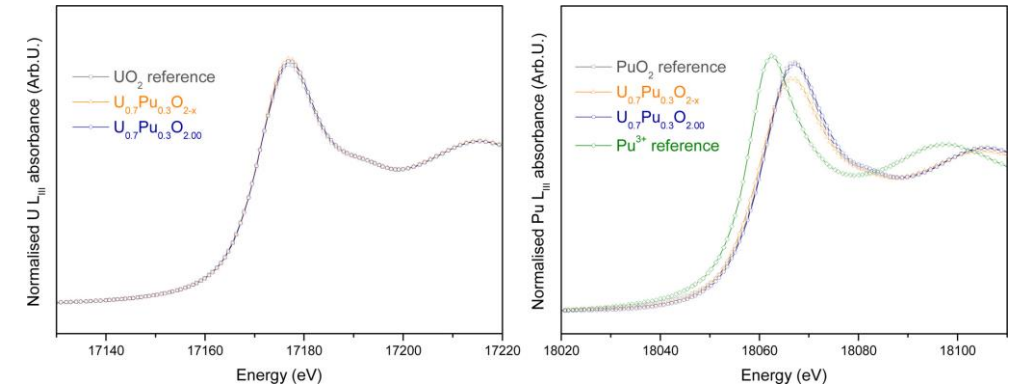


shorter diffusion path of α particles to the grain boundaries

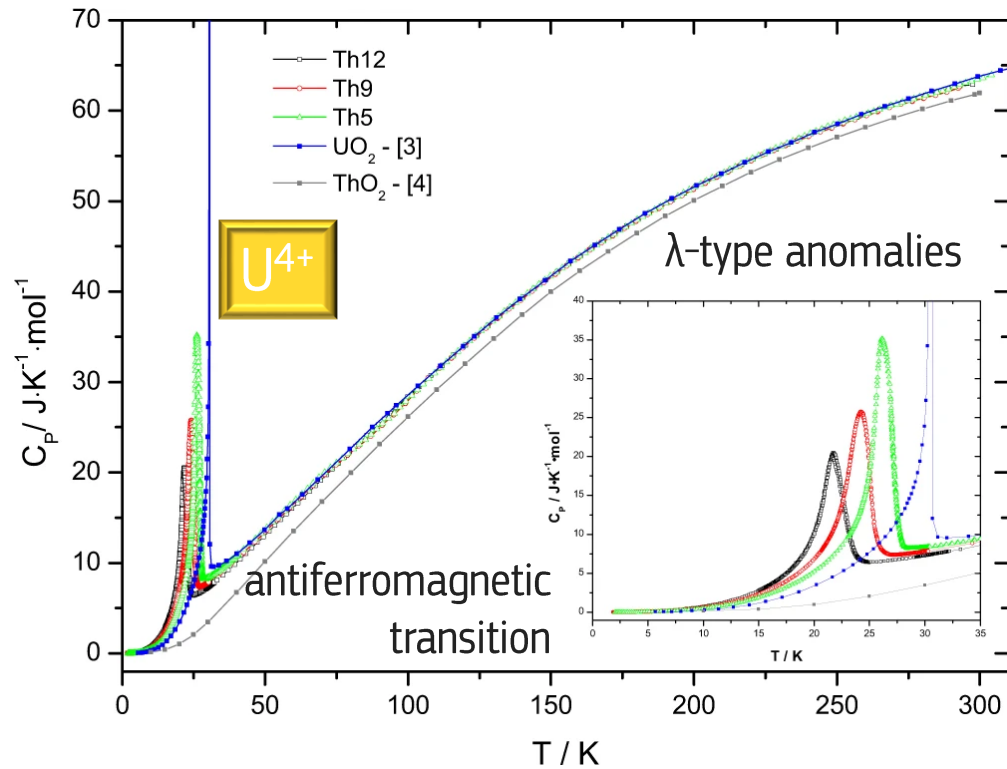
Swelling of the fluorite structure: influence of the nature of dopant



(Vigier *et al.*, *Inorg. Chem.* **54** (2015) 5358–5365)



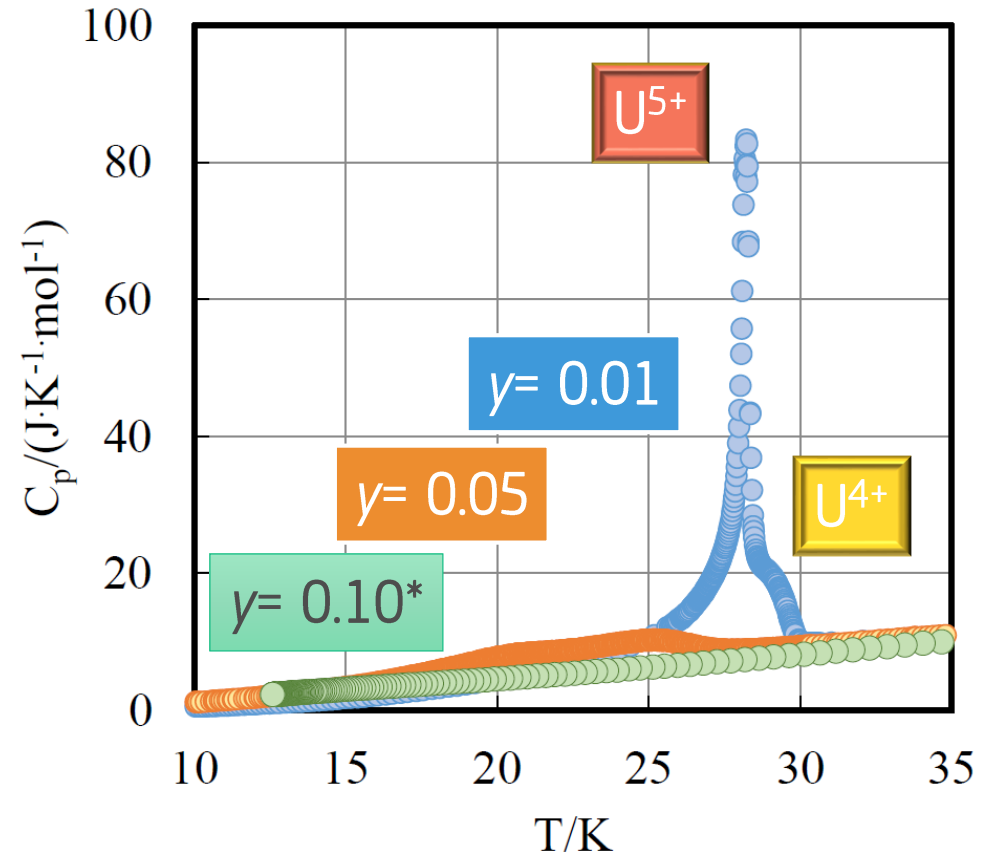
Low temperature heat capacity of $U_{1-y}An_yO_2$ ($An= Th, {}^{241}Am$)



(Vălu *et al.*, *Sci. Rep.* **9** (2019) 5358–5365)

[3] J. Huntzicker & E. Westrum, *J. Chem. Thermodyn.* **3** (1971) 61–67

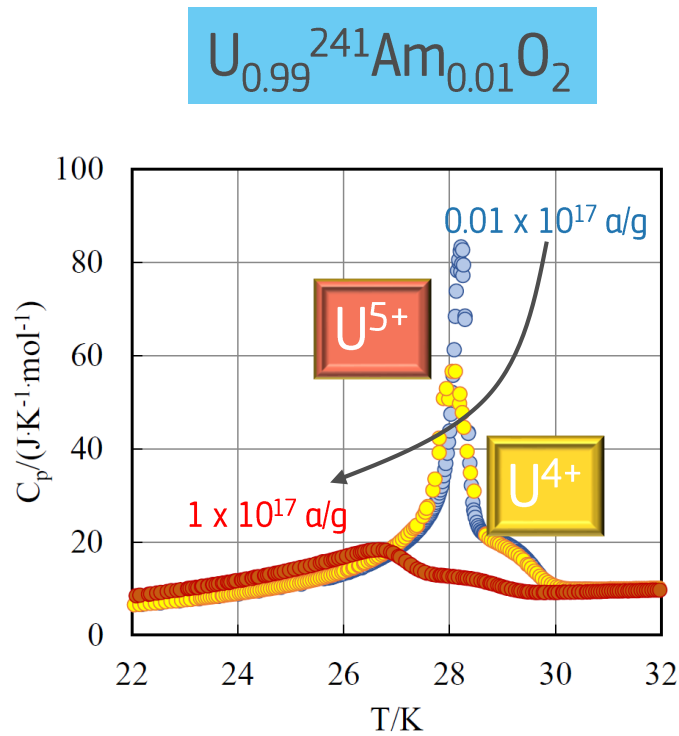
[4] D. Osborne, & E. Westrum, *J. Chem. Phys.* **21** (1953) 1884–1887)



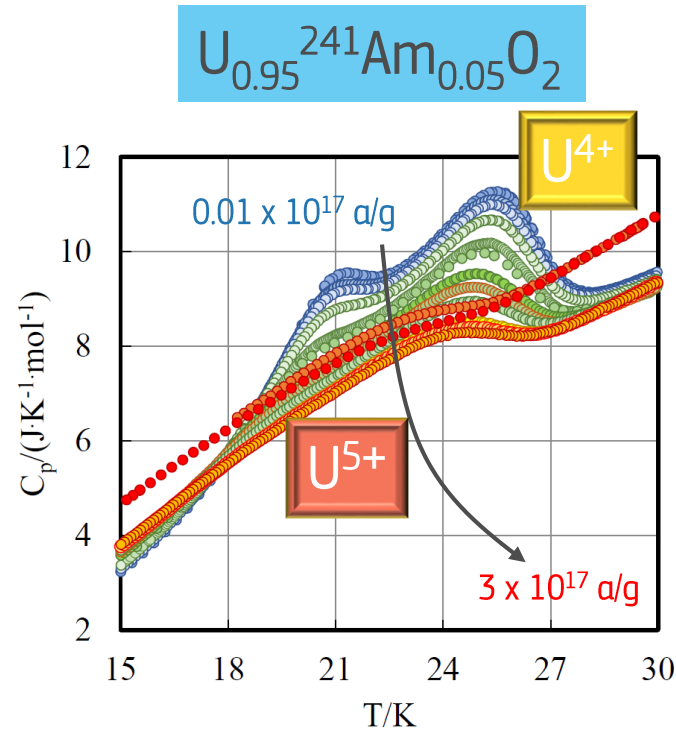
(J.-C. Griveau *et al.*, *J. Appl. Phys.* (2022) submitted;

* from Vălu *et al.*, *J. Nucl. Mater.* **507** (2018) 126–134)

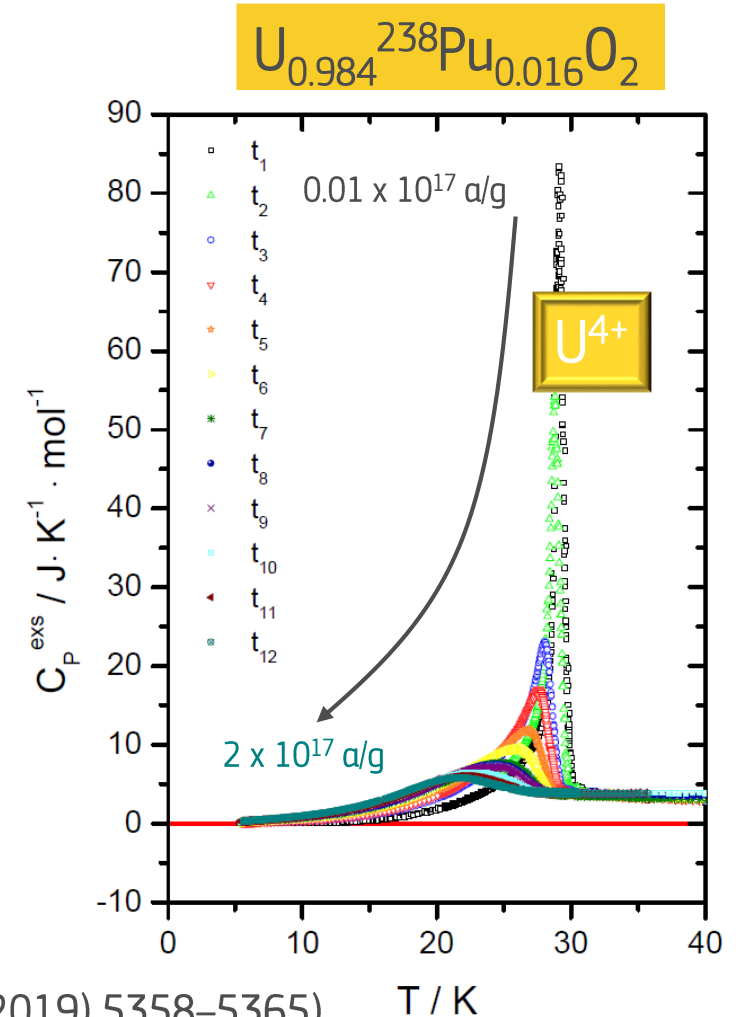
Low temperature heat capacity of $U_{1-y}An_yO_2$ ($An = {}^{241}\text{Am}$, ${}^{238}\text{Pu}$) – aging behaviour



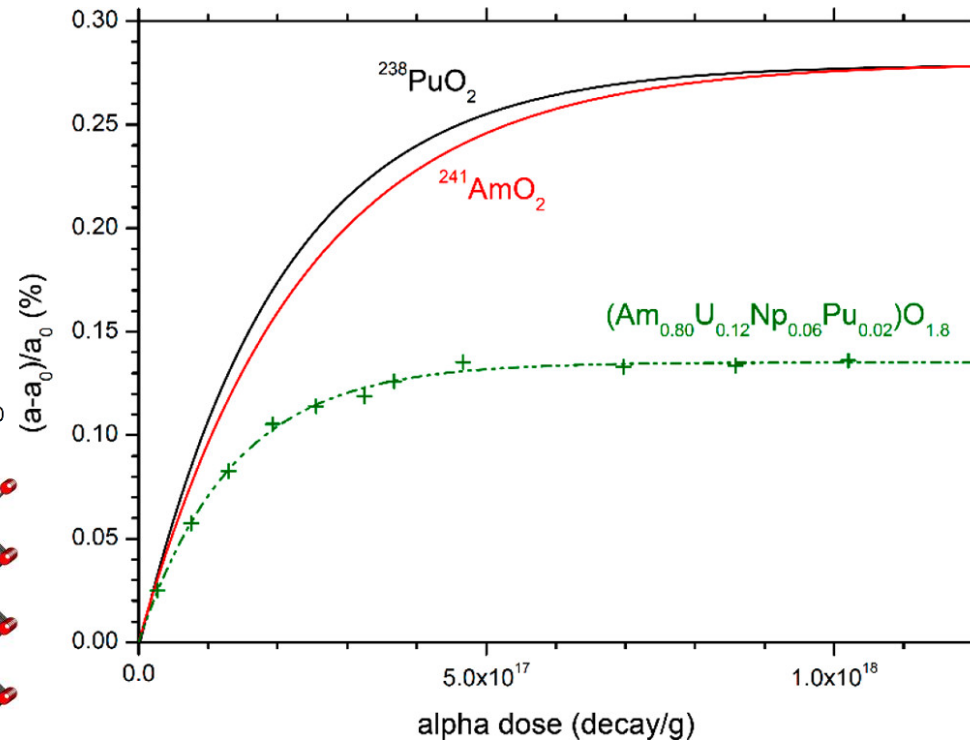
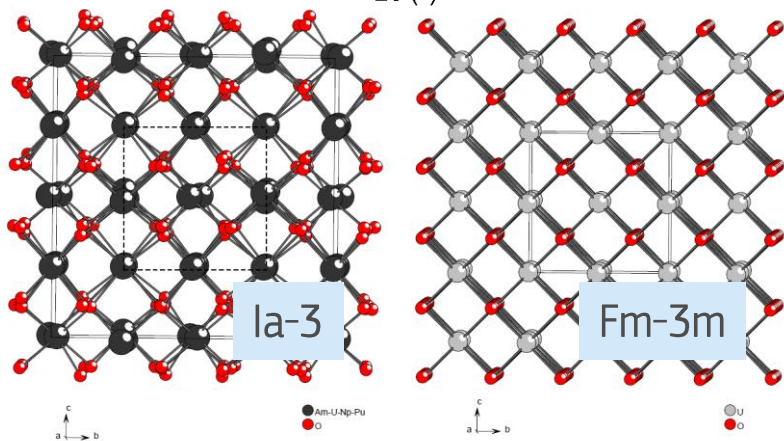
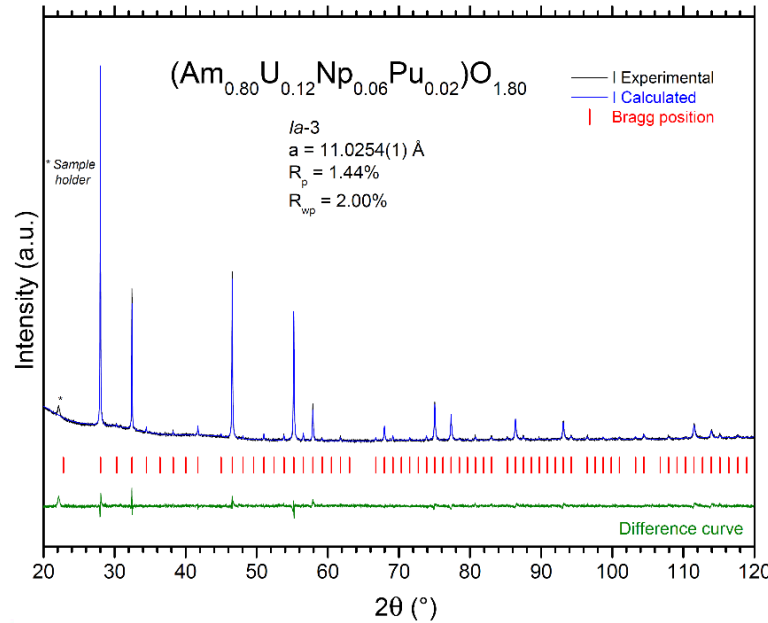
(J.-C. Griveau *et al.*, *J. Appl. Phys.* (2022) submitted)



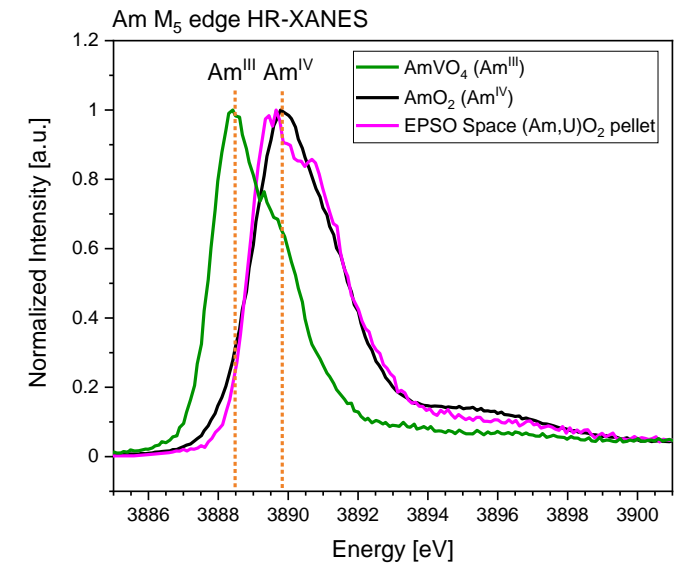
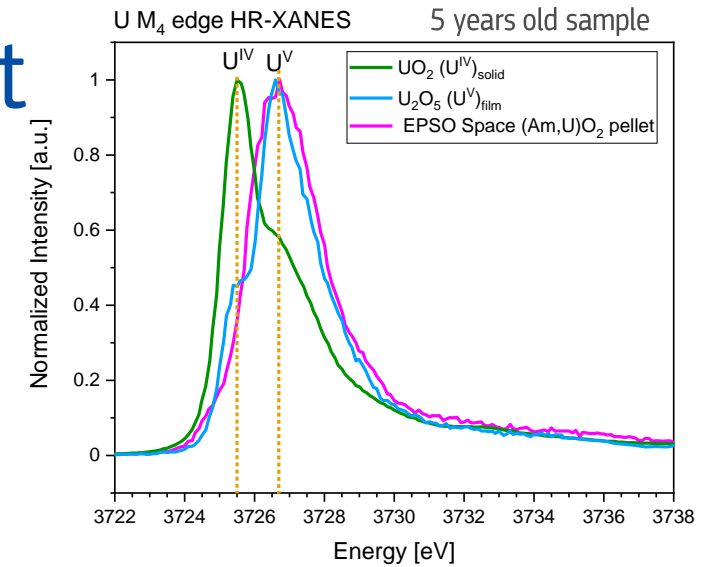
(Vălu *et al.*, *Sci. Rep.* **9** (2019) 5358–5365)



Self-irradiation behaviour of the fluorite and fluorite-related structures: influence of the nature of dopant

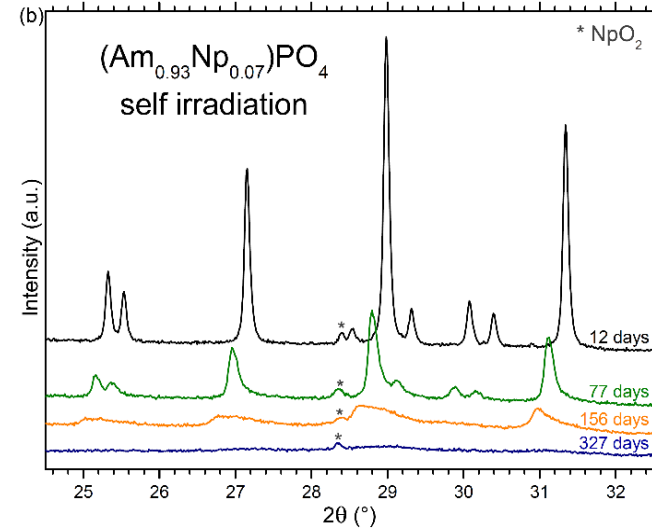
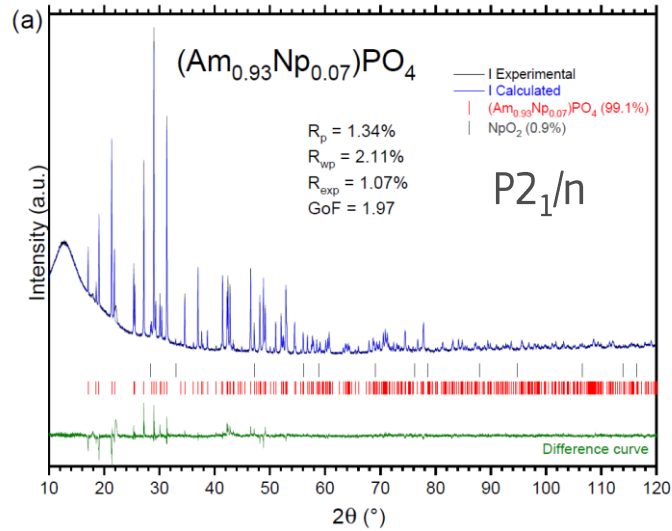


(Vigier *et al.*, *Inorg. Chem.* **57** (2018) 4317-4327)

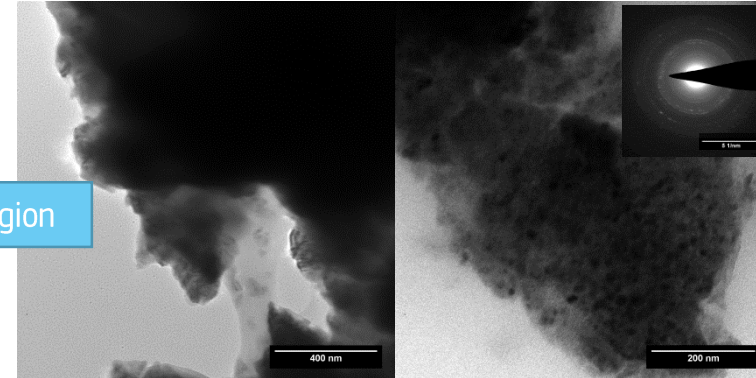


Self irradiation behaviour of AmPO_4

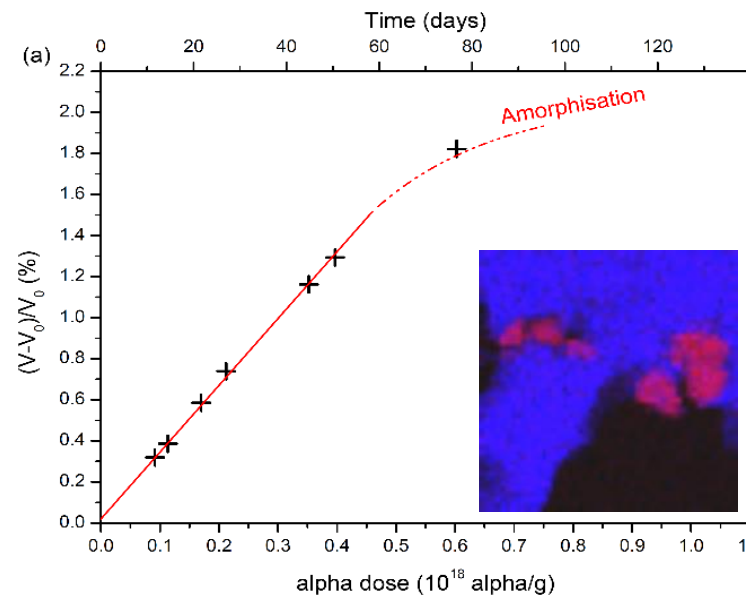
TEM, 320 days after synthesis



Perfectly amorphous region



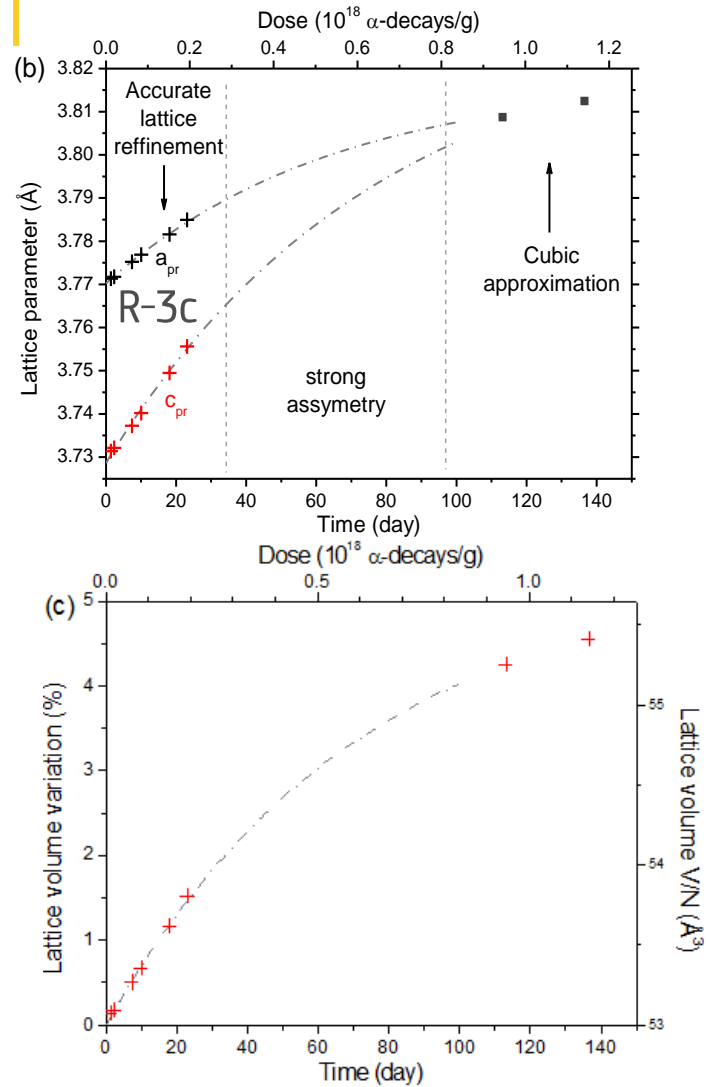
Polycrystalline particle



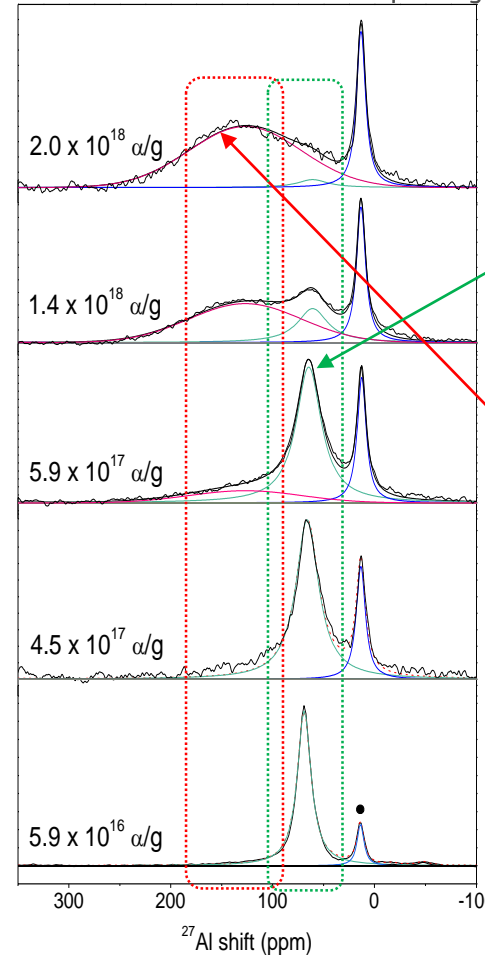
Dark field of the previous region proving its polycrystalline nature

(K. Popa *et al.*, *Inorg. Chem.* **59** (2020) 6595-6602)

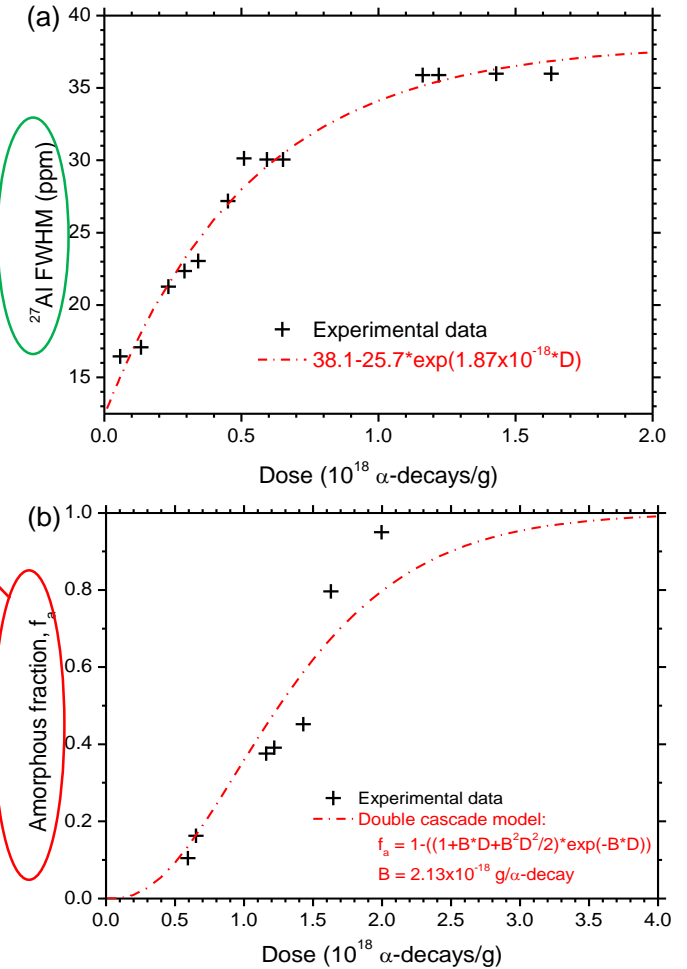
Self-irradiation behaviour of AmAlO₃



Variation of the ²⁷Al MAS-NMR spectra central transition with time and their corresponding fits

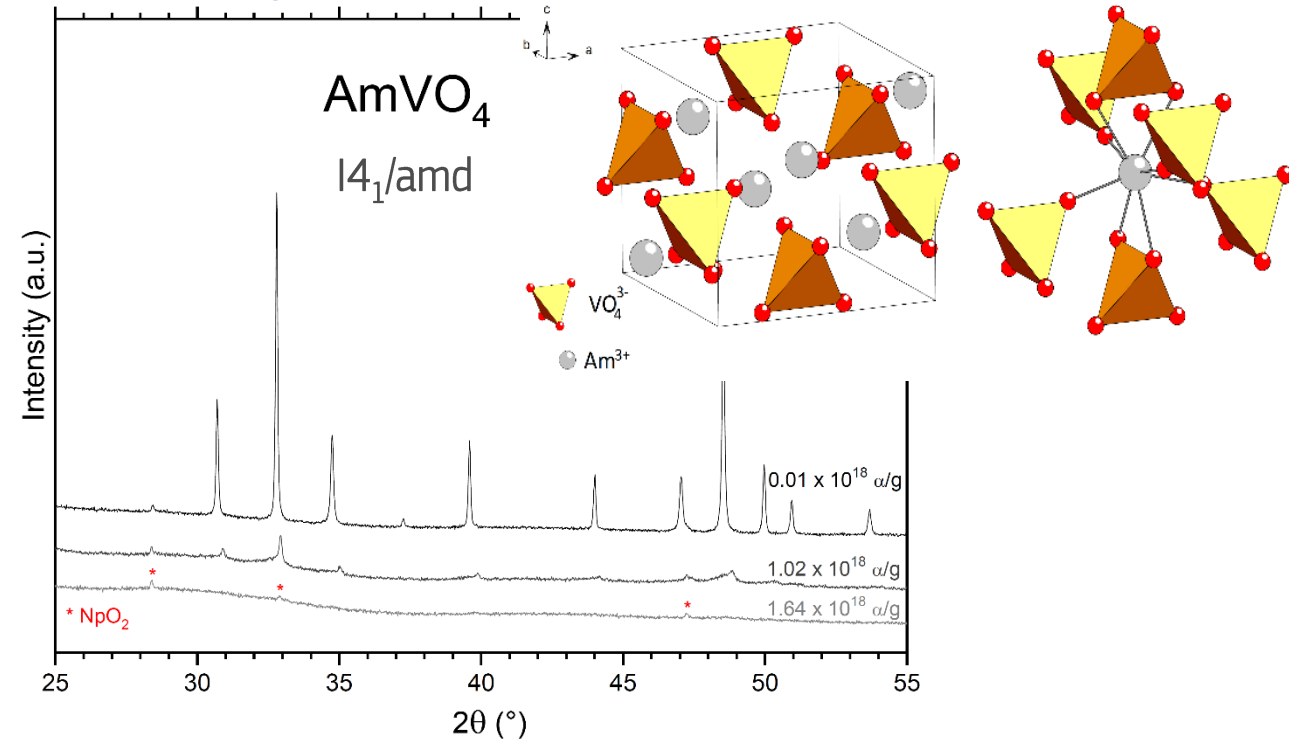
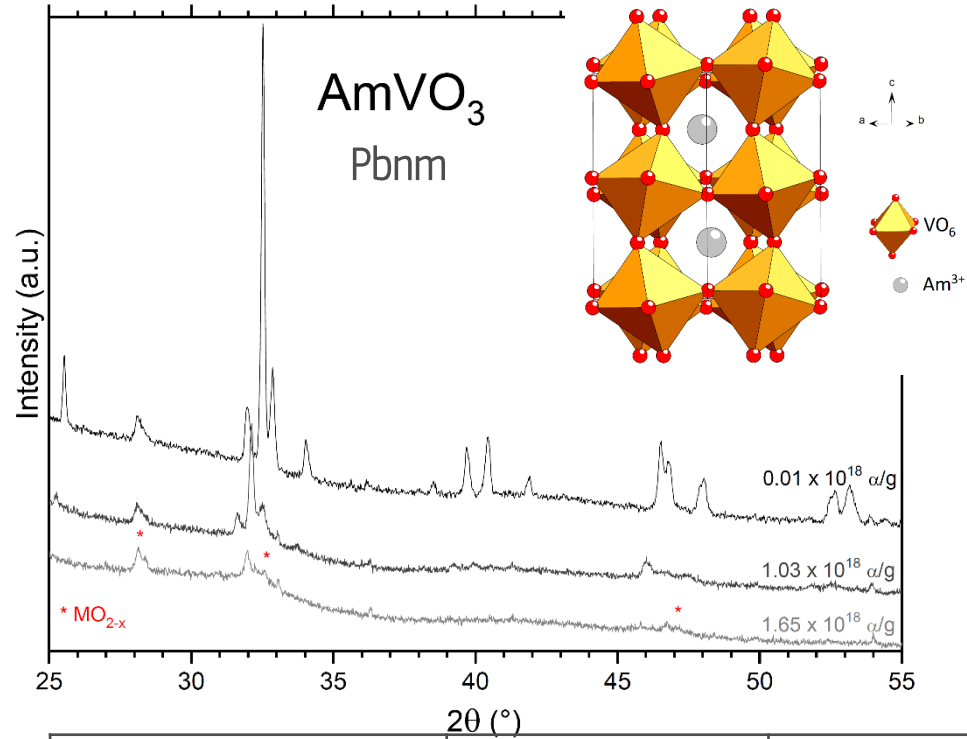


Variation of the ²⁷Al FWHM of crystalline phase (a) and the amorphous phase (b) content, with the alpha decay dose D



(J.-F. Vigier *et al.*, *Inorg. Chem.* **58** (2019) 9118-9126)

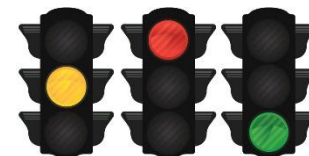
Self irradiation behaviour of AmVO₃ and AmVO₄



a, Å	b, Å	c, Å	V, Å ³	Volume variation, %	Dose (α/g)	dpa
AmVO₃ - Pbnm - Perovskite						
5.444	5.590	7.755	235.982	-	0.01 x 10 ¹⁸	0.0016
5.513	5.661	7.882	245.994	4.2	1.03 x 10 ¹⁸	0.158
AmVO₄ - I4₁/amd - Zircon						
7.294	7.294	6.428	341.952	-	0.01 x 10 ¹⁸	0.0013
7.243	7.243	6.405	335.964	-1.8	1.02 x 10 ¹⁸	0.134

Comparison of the Am-containing ceramics

		AmVO₃	AmVO₄	(Am,U)O₂	AmPO₄	AmAlO₃
Structure		perovskite	zircon	fluorite	monazite	perovskite
²⁴¹ Am-content (wt. %)		71	68	75	72	76
Volume variation (%)		+ 4.2	- 1.8	+ 0.4	+ 1.5	+ 5
Amorphization		fast	fast	no	slow	fast
Stability	oxidising	RT - 280 °C	RT - 1500 °C	RT - 1500 °C	RT - 1500 °C	RT - 1500 °C
	inert				RT - 1500 °C	
	reducing	RT - 1500 °C	RT - 280 °C	RT - 1500 °C		
²³⁷ Np-distribution		segregation	segregation	atomic scale	atomic scale + segregation	segregation



References

J.-C. Griveau *et al.*, “Low-temperature heat capacity and magnetism in $(U_{1-y}Ln_y)O_2$ and $(U_{1-y}Am_y)O_2$ ($y= 0.01 - 0.05$) solid solutions: effects of substitution and self-irradiation”, *J. Appl. Phys.* (2022) submitted.

J.-F. Vigier *et al.*, “Synthesis and characterization of homogeneous $(U,Am)O_2$ and $(U,Pu,Am)O_2$ nanopowders”, *CrystEngComm* (2022) accepted, doi.org/10.1039/D2CE00527A.

J.-F. Vigier *et al.*, “Synthesis, characterization, and stability of two americium vanadates, $AmVO_3$ and $AmVO_4$ ”, *Inorg. Chem.* (2022) in progress.

K. Popa *et al.*, “Synthesis, characterization, and stability of $AmPO_4$ ”, *Inorg. Chem.* **59** (2020) 6595-6602.

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J.-F. Vigier *et al.*, “Uranium-doped americium oxide synthesis for space applications”, *Inorg. Chem.* **57** (2018) 4317-4327.

Thank you

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