

**High-temperature single-rod tests on the oxidation of Cr coated cladding tubes,
carried out at KIT within the IAEA project ATF-TS**

J. Stuckert, U. Peters, U. Stegmaier, K. Vizelkova, E. Vogel

Abstract

22 single rod oxidation tests with Cr coated zirconium alloy tubes were carried out within the framework of the IAEA project ATF-TS under DBA and BDBA transient conditions in an induction furnace. Eight M5[®] and four ZIRLO[™] claddings PVD coated to 7 µm thick Cr layer, four E110 claddings with arc ion Cr plating to 8 µm thick Cr layer, six opt. ZIRLO[™] claddings PVD coated to 5, 18 µm thick Cr layer and 18+8 µm thick Cr+CrN layer were tested.

The tests performed with ZIRLO claddings coated to 7 µm Cr layer showed that for $T_{\max} < 1300$ °C, only small oxidation of Cr layer with low hydrogen release (factor 15 in comparison to not coated claddings) occurred. For higher temperatures, formation of numerous local swellings at the outer cladding surface due to Zr-Cr eutectic at 1350 °C occurred and significant hydrogen release due to oxidation of Zr substrate was measured.

The tests performed with E110 claddings coated to 8 µm Cr layer showed significant decrease of hydrogen release under extended LOCA conditions (max 1250 °C) in comparison to not coated samples (factor 15). For peak cladding temperature of 1350 °C, numerous local swellings at the outer cladding surface formed and intensive hydrogen release (factor 18 in comparison to 1250 °C) was measured.

The tests performed with opt. ZIRLO claddings coated to three different layer thicknesses showed that an increase in the thickness of the Cr layer by 3 times (from 6 to 18 µm) reduced the hydrogen release by 30% under DBA conditions (transient to 1200 °C), however, only by 3% under BDBA conditions (1400 °C). Barrier layer CrN has only moderate influence on hydrogen release under DBA, but strong reduced hydrogen release under BDBA (reduction factor 4). A local temperature escalation and formation and relocation of (Zr,Cr)+Zr melt for Cr/CrN coating occurred perhaps due to the lack of intermediate oxidation of part of the zirconium substrate, which could not react with steam due to the barrier effect of CrN layer.

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29th QWS

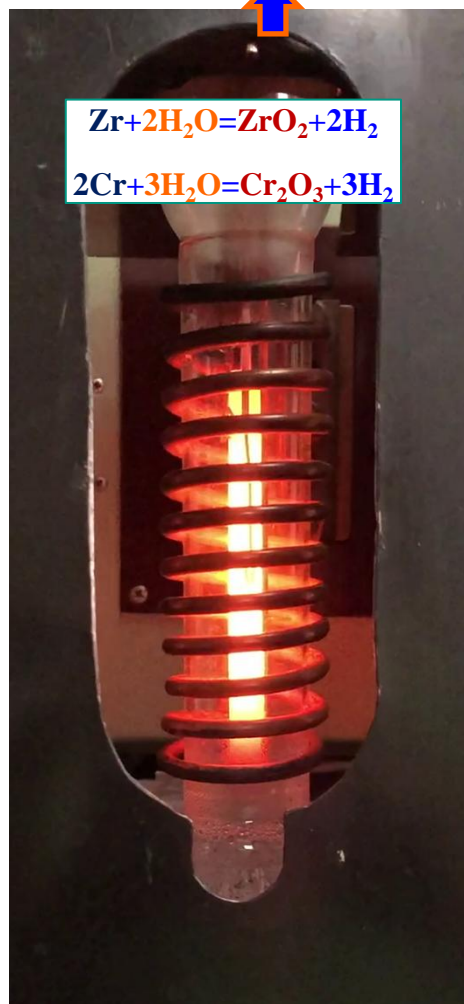
Institute for Applied Materials; Program NUSAFE



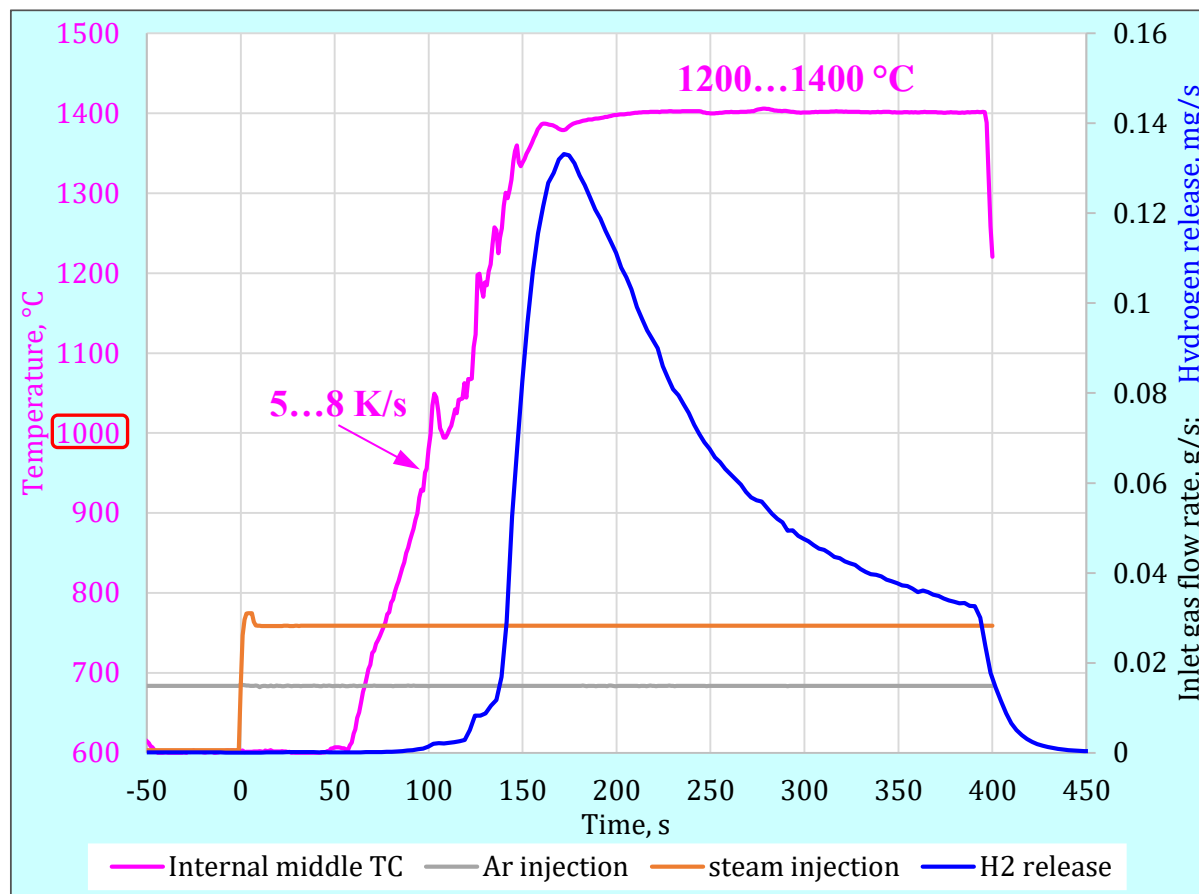
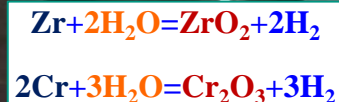
Three sets of HT oxidation tests

- I. ZIRLO and M5[®] tubes, PVD coated by Oerlikon/Balzers GmbH with Cr to the layer thickness of 7 μm .
- II. ZIRLO tubes, PVD coated by UPM/Prague with Cr to the layer thicknesses of 5 and 18 μm as well as with Cr/CrN to the layer thickness 18/8 μm (Cr/CrN).
- III. E110 tubes, treated by AEOI with arc ion coated method to the Cr layer thickness of 8 μm .

Transient tests in inductive furnace: cladding oxidation in steam



mass spectrometer

 $\text{H}_2 + \text{Ar} + \text{steam}$


Instrumentation: 1) TC inside the sample; 2) TC at the sample surface;
3) pyrometer; 4) mass spectrometer at furnace outlet

Ar (30 l/h) + steam (100 g/h)

Control of generator according to internal TC or pyrometer;
oscillations above 1000 °C

Tests with 12 KIT samples coated by OERLIKON to 7 μm Cr

MC tubes: OD=10.75 mm, wall 725 μm ; MCs and ZOs tubes: OD=9.5 mm, wall 570 μm

Sample	Heating rate [K/s]	Maximal clad temperature T_{max} [°C]	Duration of oxidation at T_{max} [min]	Cooldown	Comment
MC-4	7	1200	10	steam + Ar	performed
MC-5	7	1200	15	steam + Ar	performed
MC-6	8	1300	5	steam + Ar	performed
MC-7	6	1250	10	steam + Ar	performed
MCs-2	8	1360	6	steam + Ar	performed
MCs-3	8	1360	0	steam + Ar	performed
MCs-4	5	1250	6 (1250°→900°)	water	performed
MCs-5	5	1250	0	steam + Ar	performed
ZOs-2	5	1200	4	steam + Ar	performed
ZOs-3	5	1380	0	steam + Ar	performed
ZOs-4	5	1200	4	steam + Ar	performed
ZOs-5	5	1400	4	steam + Ar	performed

MC – for coated M5[®]
ZO – for coated ZIRLO[™]

Appearance of coated cladding samples oxidized in inductive furnace



MC-4: 1200 °C
during 10 min



MC-5: 1200 °C
during 15 min



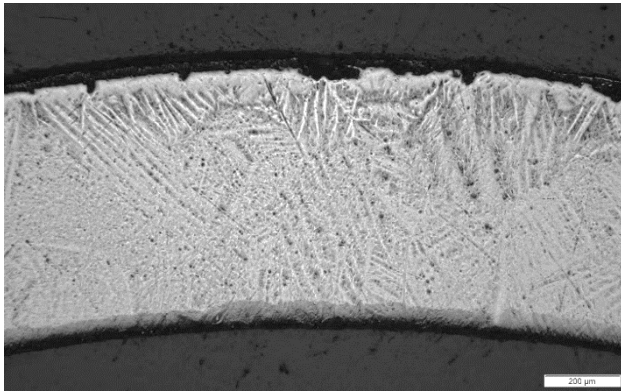
MC-7: 1250 °C
during 10 min



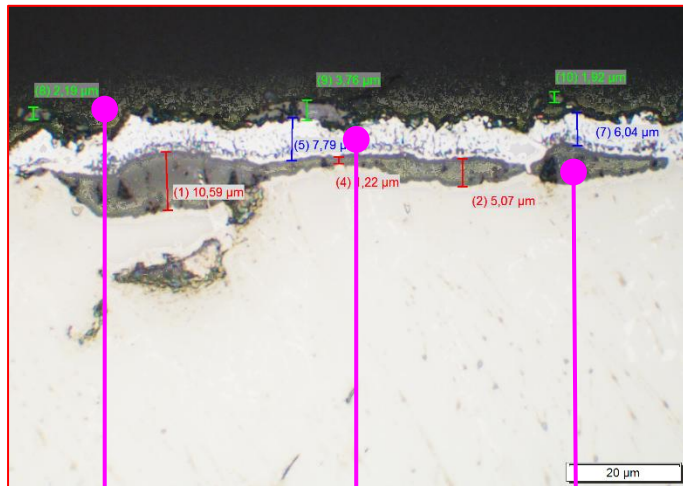
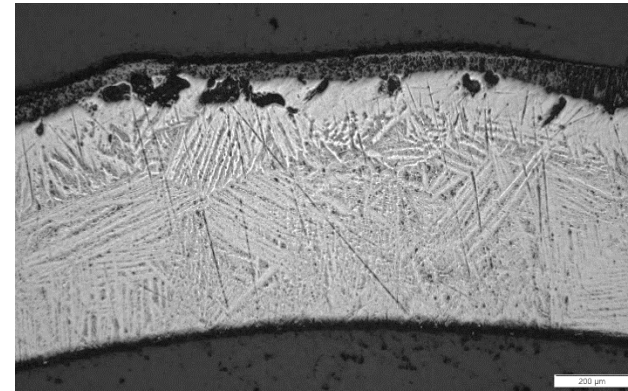
MC-6: 1350 °C
during 5 min

Cladding microstructure after heating with 7 K/s and oxidation at 1200 °C

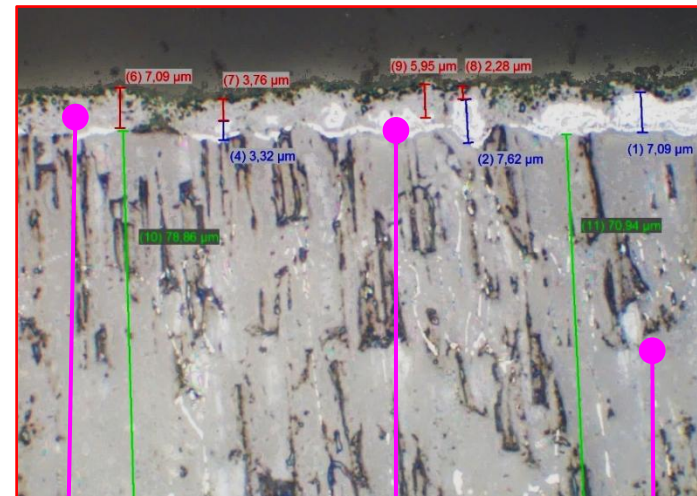
MC-4: 1200 °C during 10 min after heating



MC-5: 1200 °C during 15 min after heating

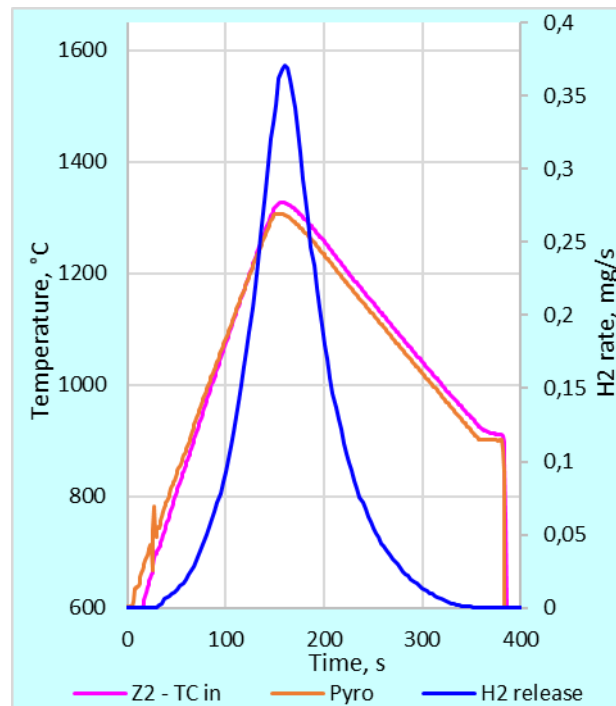
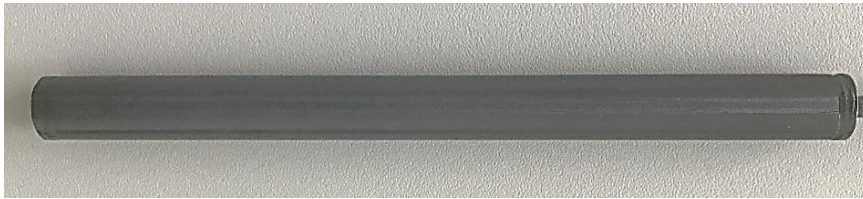


Cr₂O₃: 5µm **Cr: 5...8 µm** **ZrO₂: 2...25 µm**

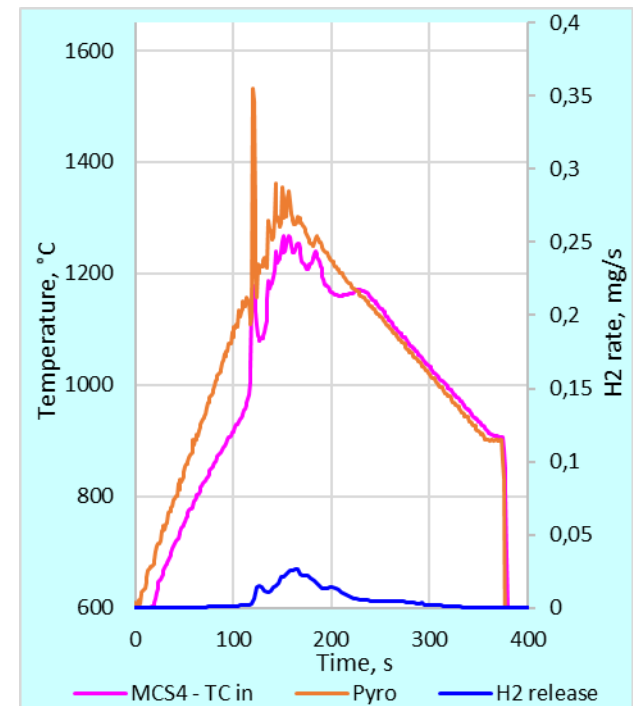


Cr₂O₃: 3...7 µm **Cr: 0...7 µm** **ZrO₂: 75 µm**

Influence of coating on hydrogen release (LOCA transient 5 K/s from 600 to 1250 °C, then cooldown to 900 °C and quenching)



not coated sample Z2 (Zry-4): significant hydrogen release



coated sample MCs-4 (M5®): strong reduced hydrogen release (factor 15)

LOCA transient followed by a constant temperature of 1400 °C: influence of duration of oxidation in steam

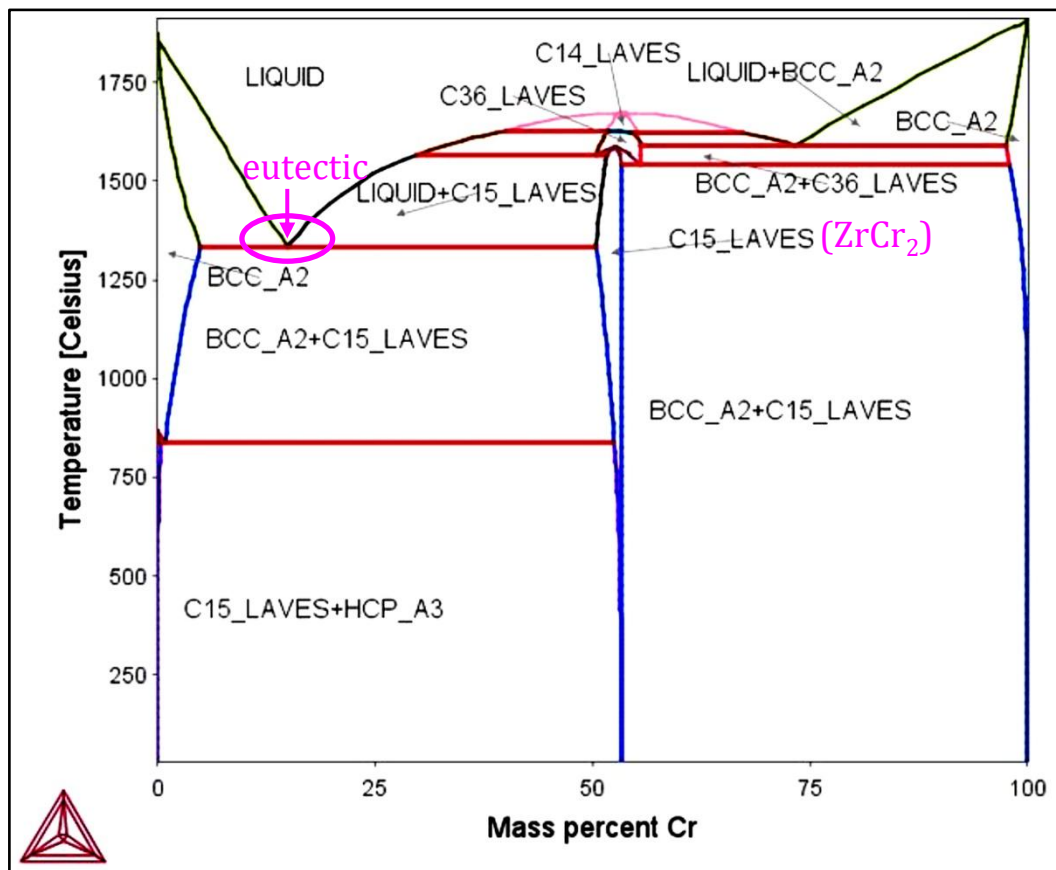


sample MCs-3: transient with 8 K/s, then 1400 °C during 3 s



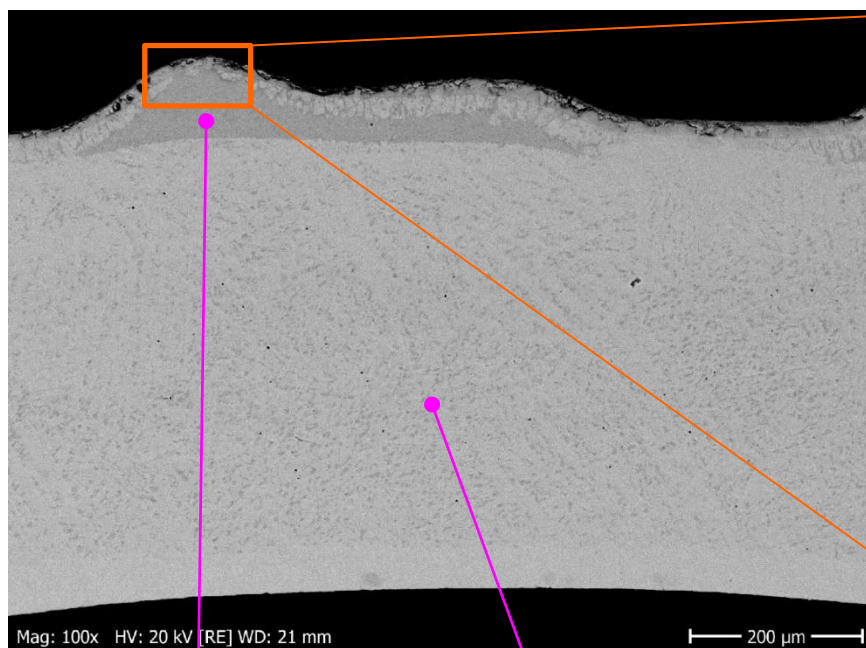
sample MCs-2: transient with 8 K/s, then 1400 °C during 240 s

Binary Zr-Cr phase diagram



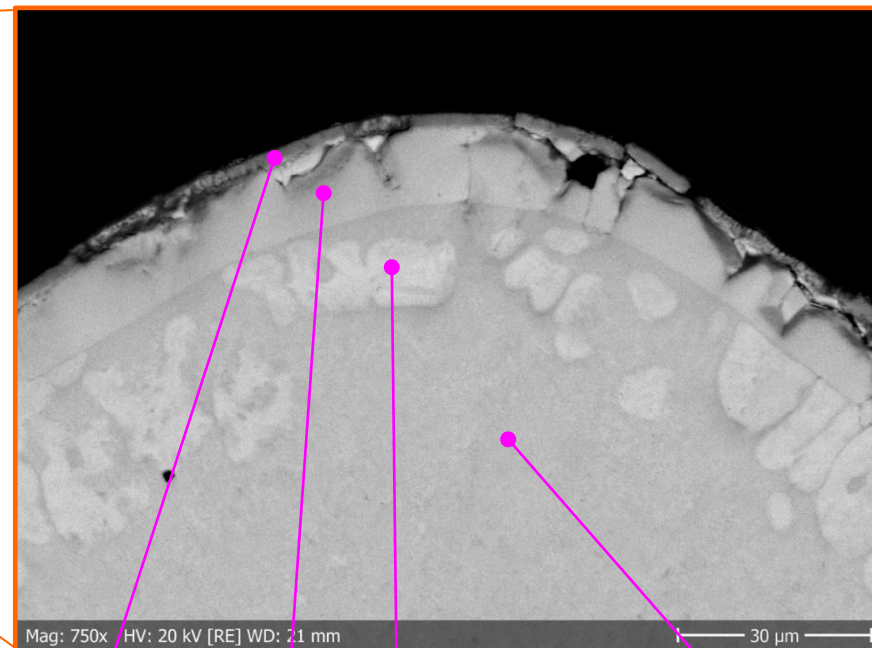
The eutectic Zr/Cr forms at ≈ 1330 °C

Result of SEM/EDX analysis of MCs3 sample: cladding microstructure immediately after LOCA transient from 600 to 1400 °C: formation of surface swellings containing Laves phase ZrCr_2



blister

prior β -Zr



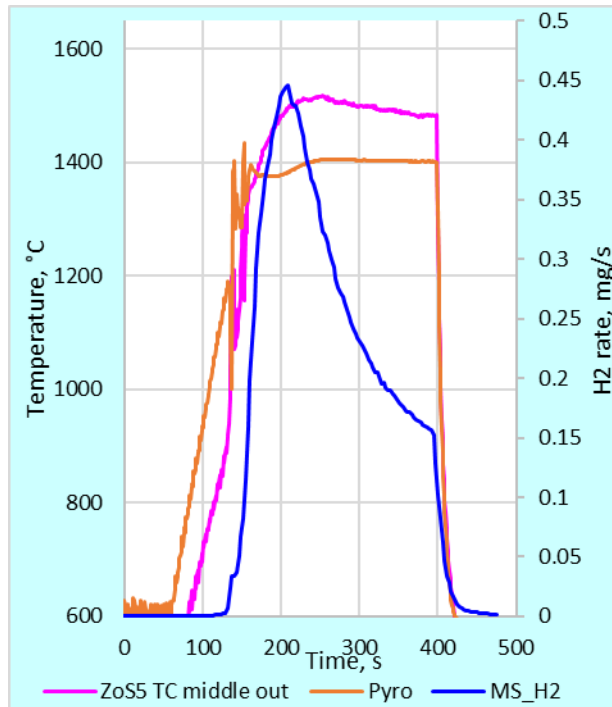
Cr_2O_3

α -Zr(O)

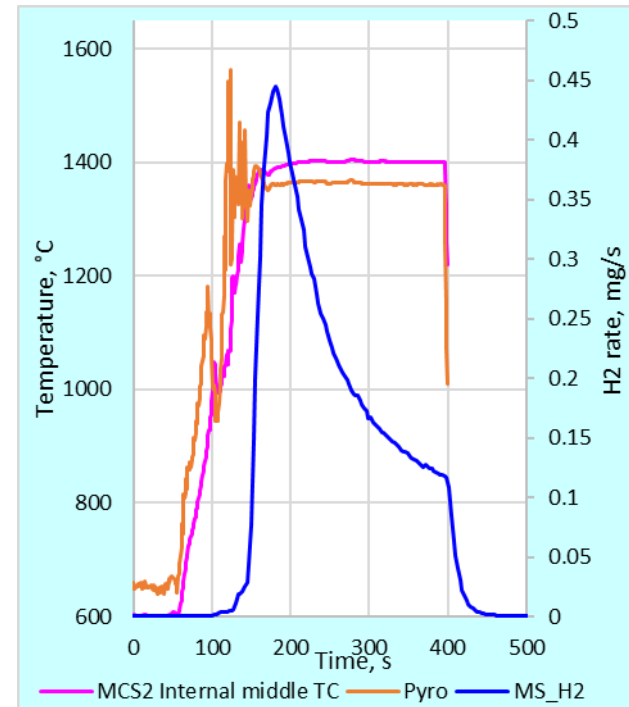
α -Zr(O) + Cr

α -Zr(O) + ZrCr_2

No influence of Zr alloy (as substrate) on hydrogen release for coated samples tested under similar high temperature conditions

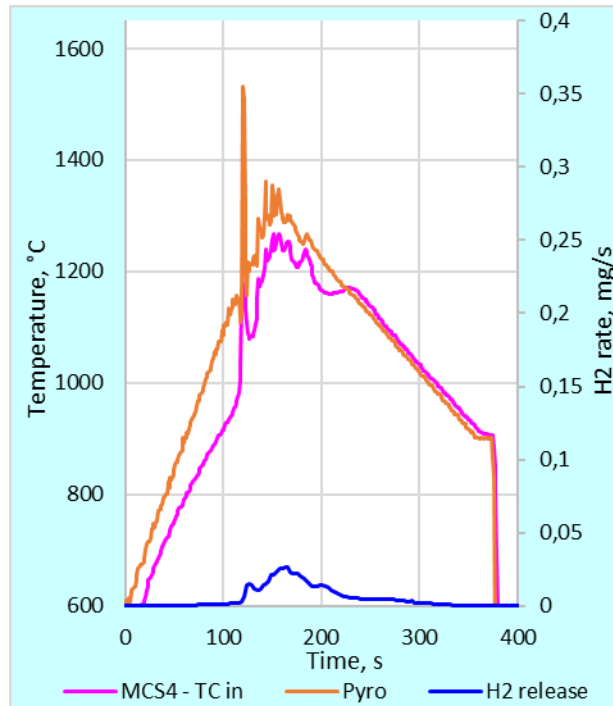


sample ZO5s (ZIRLO™)

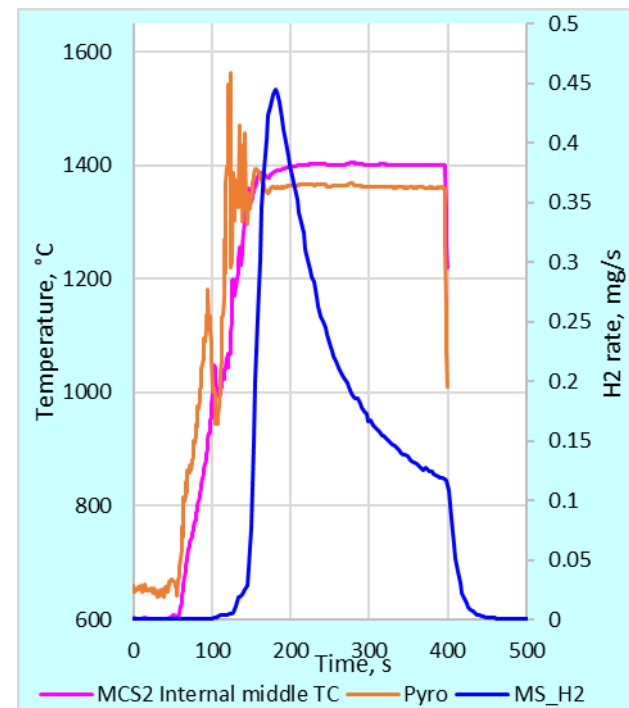


sample MCs2 (M5®)

Influence of peak cladding temperature on hydrogen release



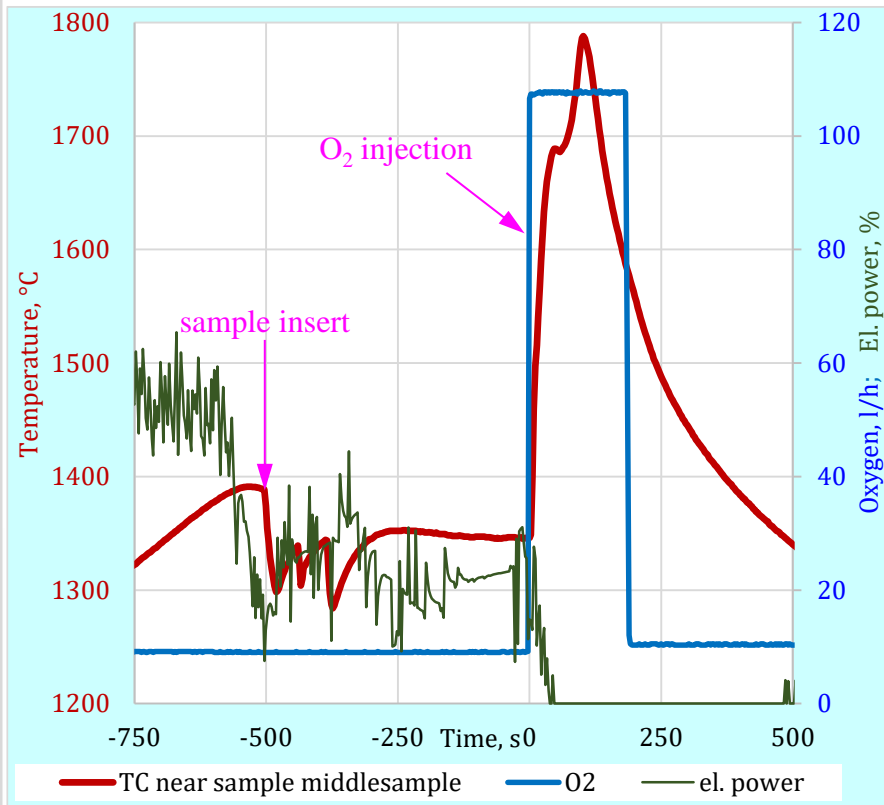
Cr coated sample MCs-4 oxidized with $T_{\max}=1250\text{ }^{\circ}\text{C}$
(only oxidation of Cr layer)



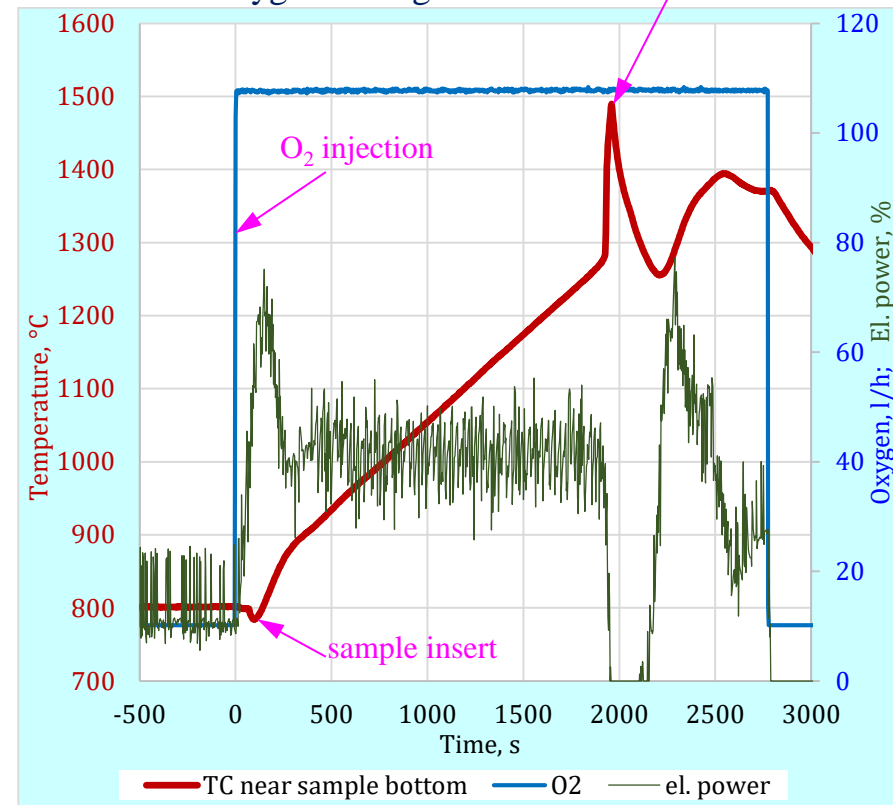
Cr coated sample MCs-2 oxidized with $T_{\max}=1400\text{ }^{\circ}\text{C}$: factor 20 for H_2
(additional oxidation of Zr substrate)

Catastrophic oxidation and temperature escalation at $T > 1300^\circ\text{C}$ in tube furnace (without radiation heat loss)

sample MC-1: injection of oxygen at 1350°C



sample ZOs-1: temperature escalation
in oxygen during slow transient



sample completely destroyed

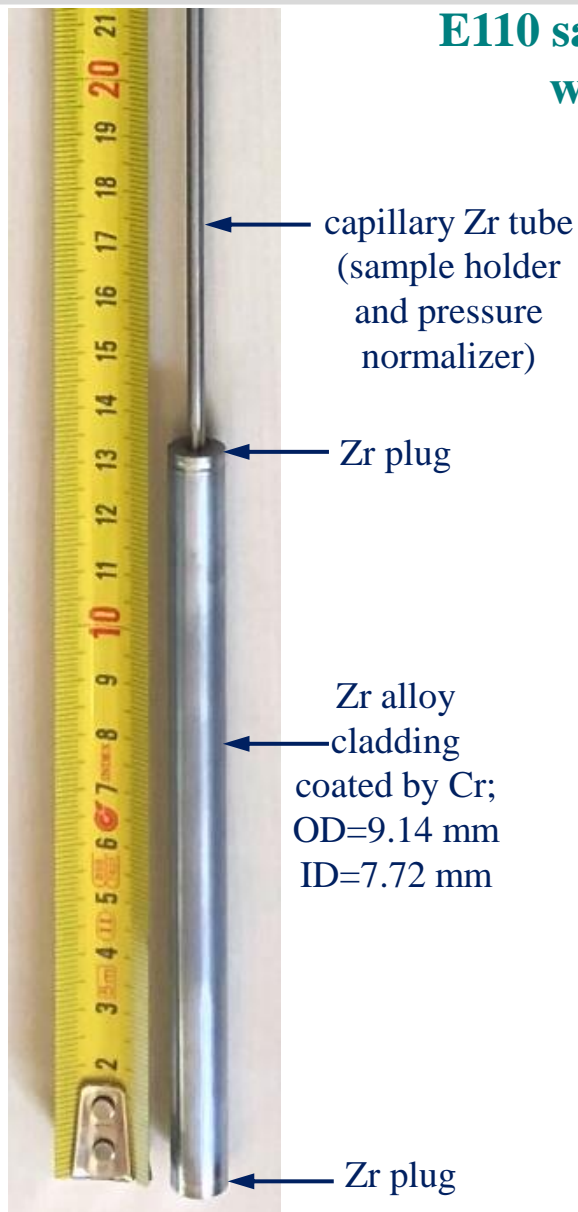
Summary on KIT samples

➤ Inductive furnace:

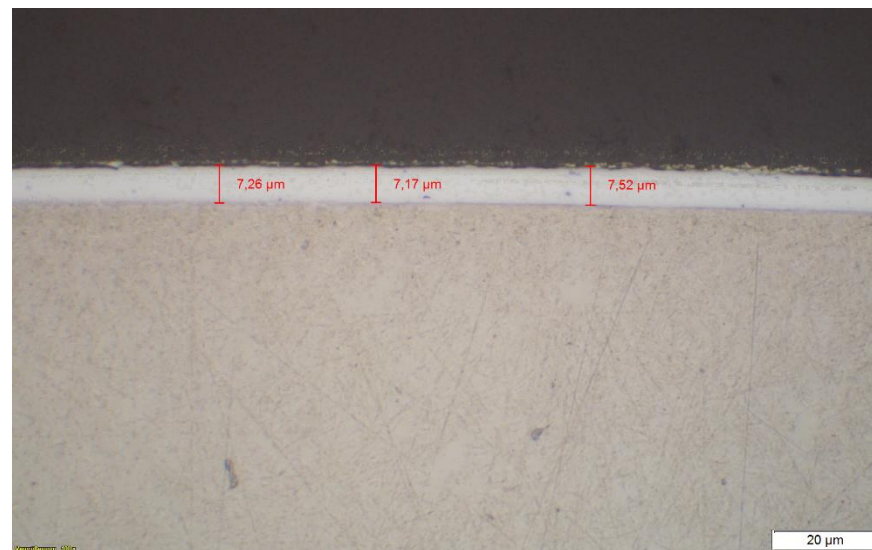
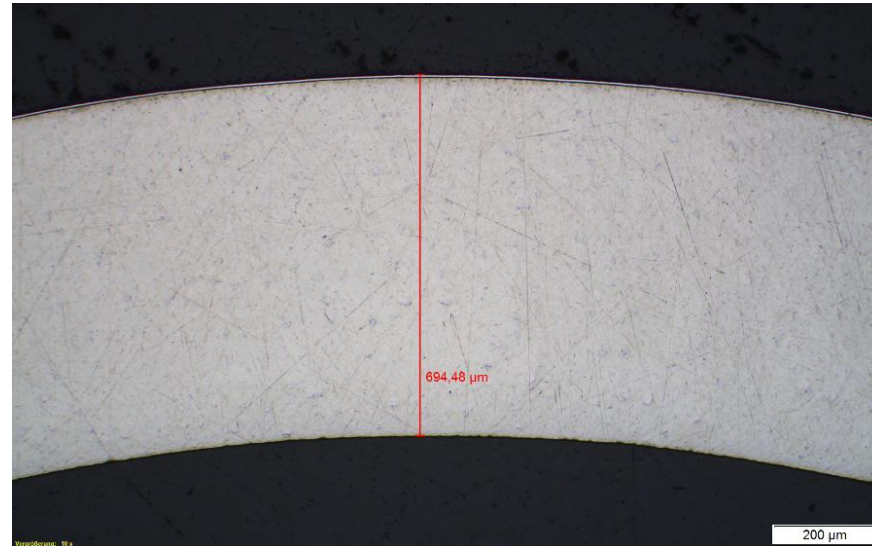
- 1) for $T_{\max} < 1300\text{ °C}$, only small oxidation of Cr layer with low hydrogen release (factor 15 in comparison to not coated claddings),
- 2) For higher temperatures, formation of numerous blisters (local swellings) at the outer cladding surface due to Zr-Cr eutectic at 1350 °C and significant hydrogen release due to oxidation of Zr substrate.

➤ Tube furnace: catastrophic oxidation at $T > 1300\text{ °C}$ due to the absence of radiation heat loss.

E110 sample with arc ion Cr plating (AEOI) with a layer thickness of 7...8 μm



coated empty sample welded at KIT
and prepared for SETs



Cr layer

optical microscopy



EC2: 5 K/s to $T_{\max}=1250\text{ }^{\circ}\text{C}$
cooldown 2 K/s, quench at $900\text{ }^{\circ}\text{C}$



EC4: 5 K/s to $1250\text{ }^{\circ}\text{C}$,
5 s $1330\text{ }^{\circ}\text{C}$, imm. cooldown

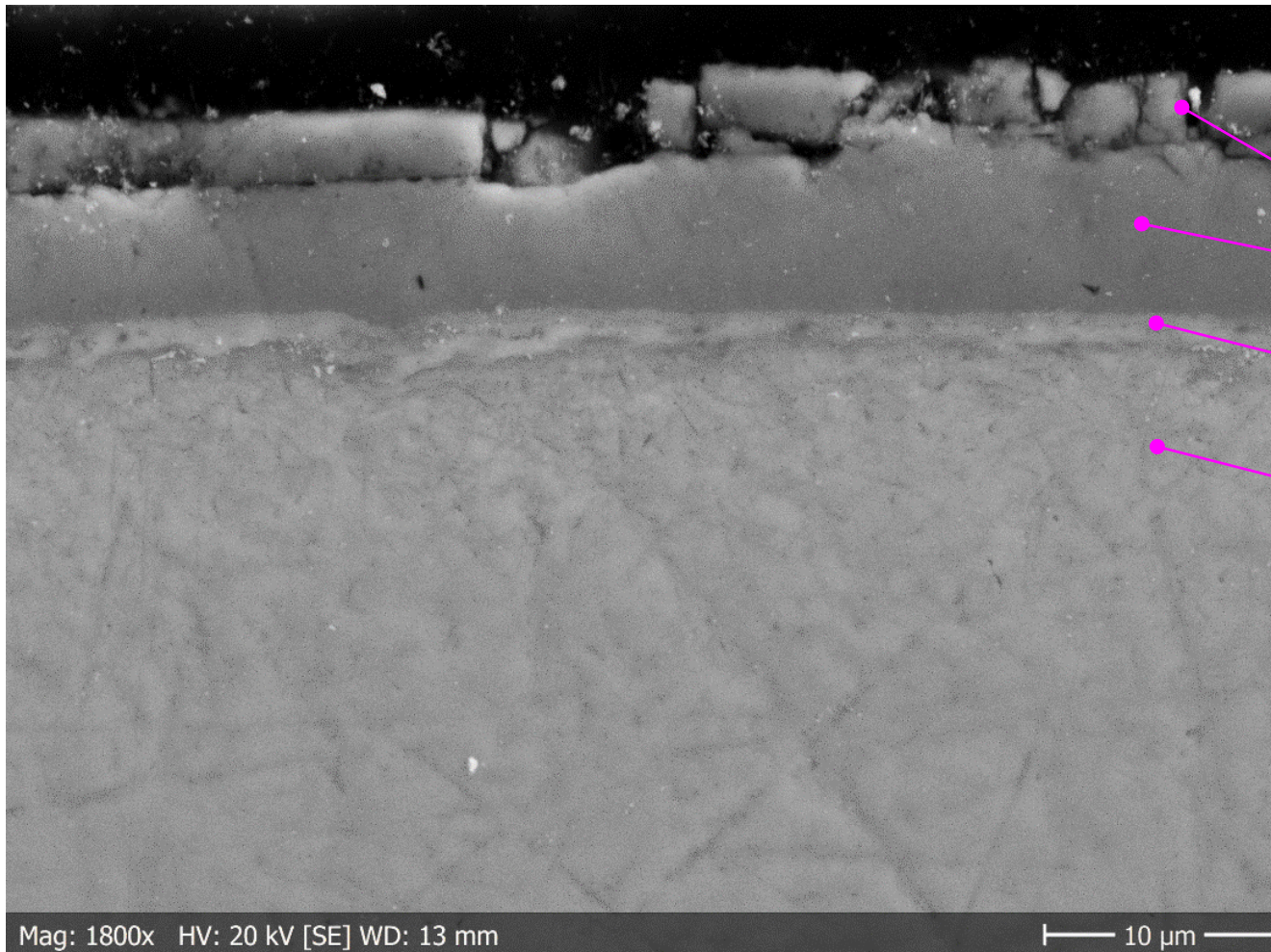


EC1: 8 K/s to $T_{\max}=1390\text{ }^{\circ}\text{C}$,
immediate cooldown



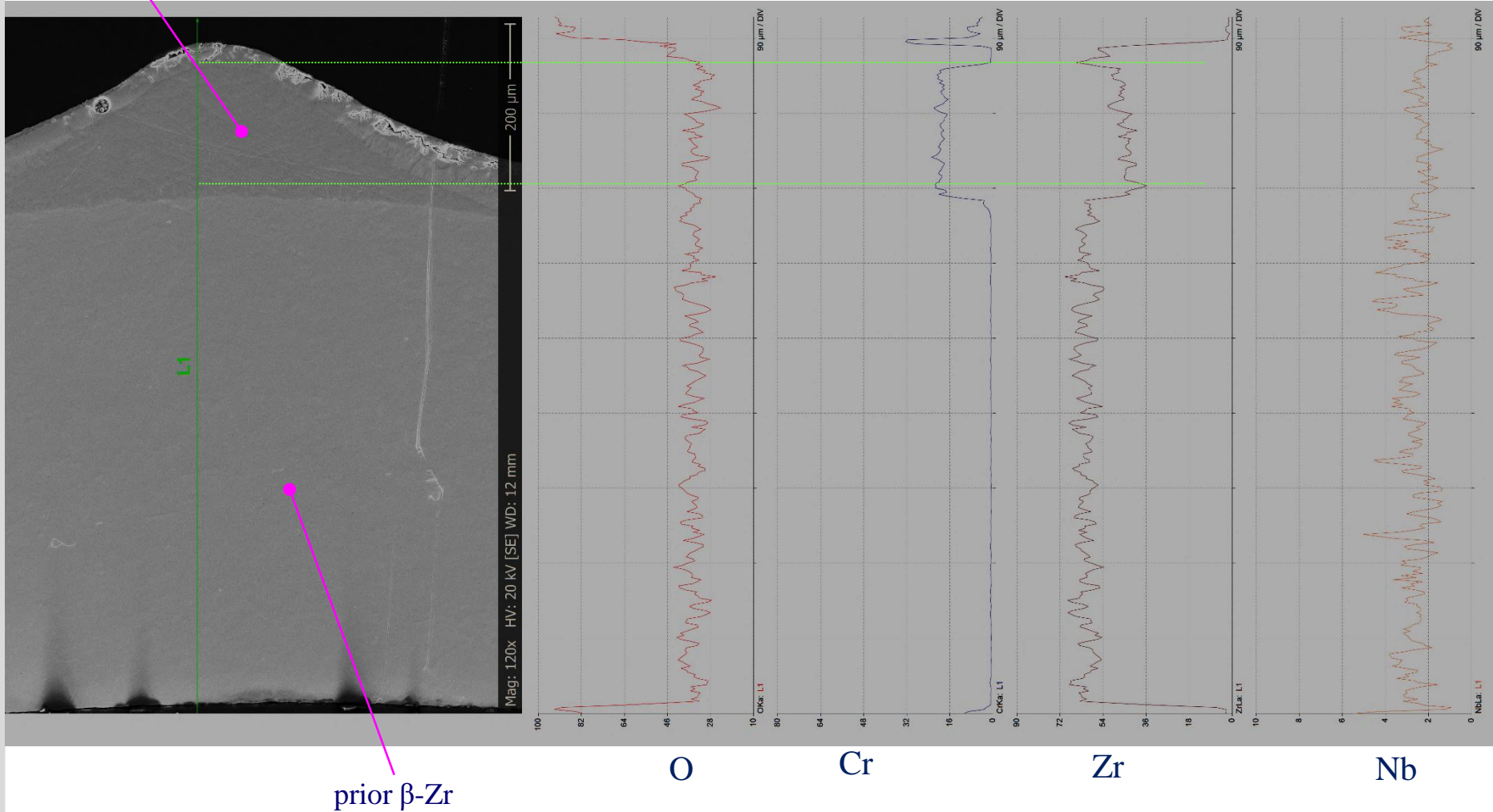
EC3: 8 K/s to $T_{\max}=1400\text{ }^{\circ}\text{C}$,
240 s $1400\text{ }^{\circ}\text{C}$, imm. cooldown

EC4 (1250 °C, immediate cool-down), SEM/EDX

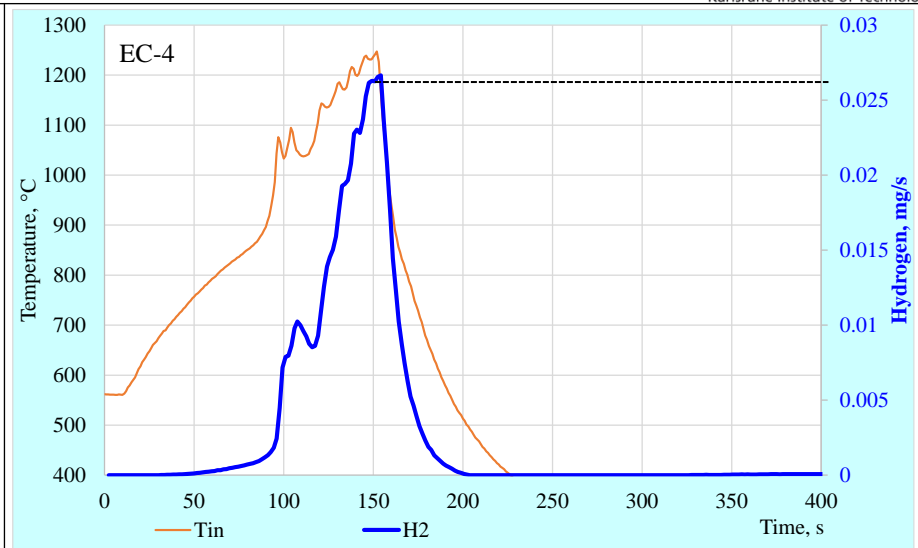
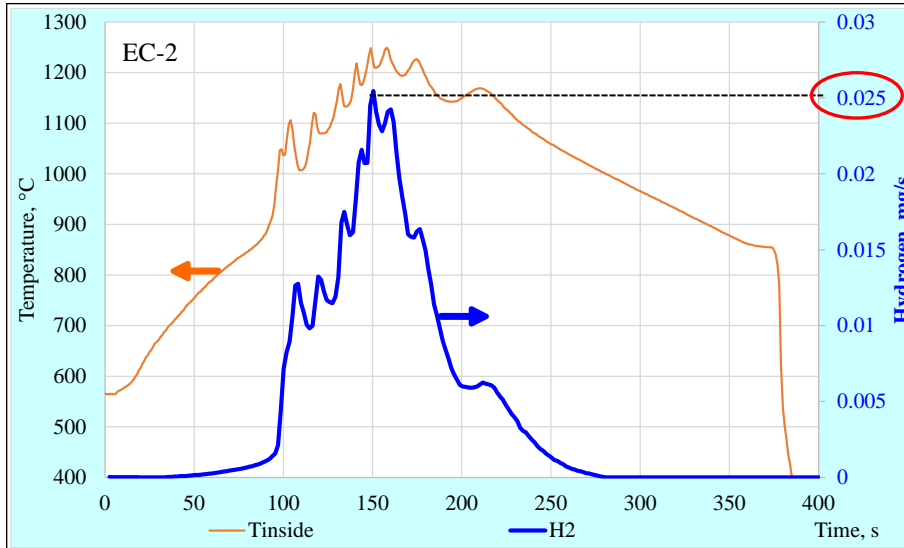


EC1 (1400 °C, immediate cool-down), SEM line scan: formation of Laves phase inside the blister

$\alpha\text{-Zr(O)} + \text{ZrCr}_2$

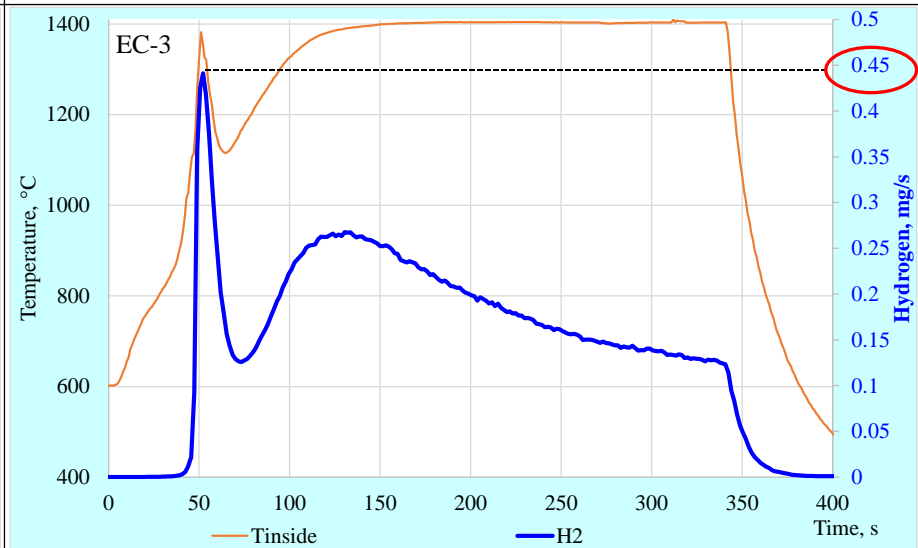
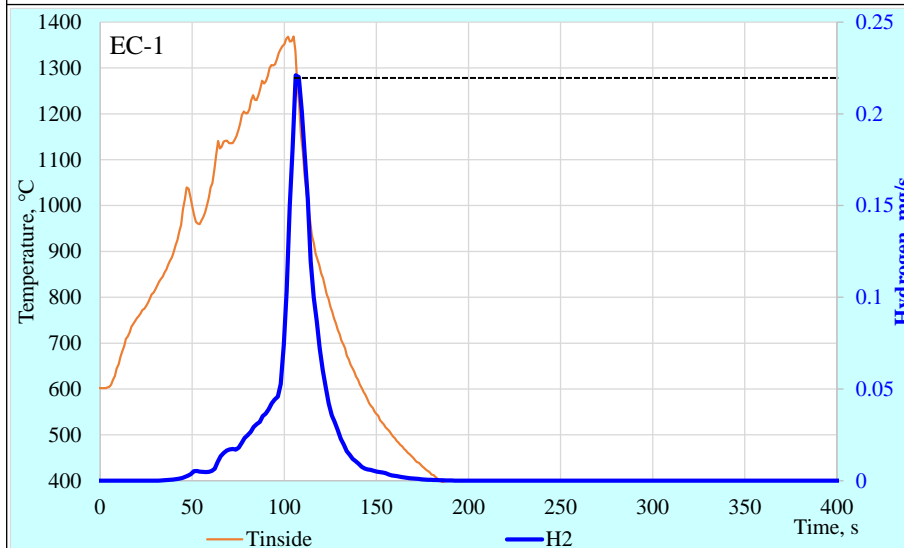


Coated AEOI samples: hydrogen release



5 K/s, T_{max}=1250 °C, cool-down 1.5 K/s to 900 °C, quench

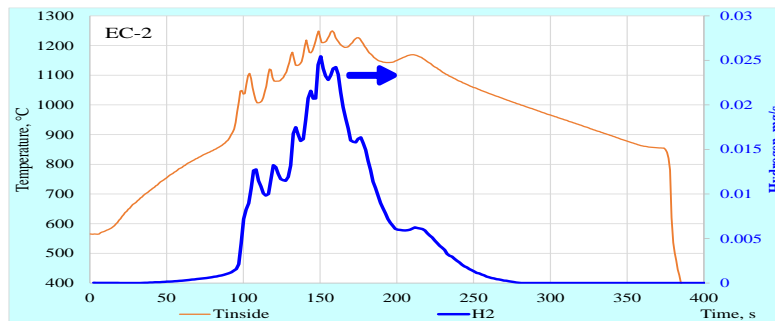
5 K/s, T_{max}=1250 °C, immediate cool-down



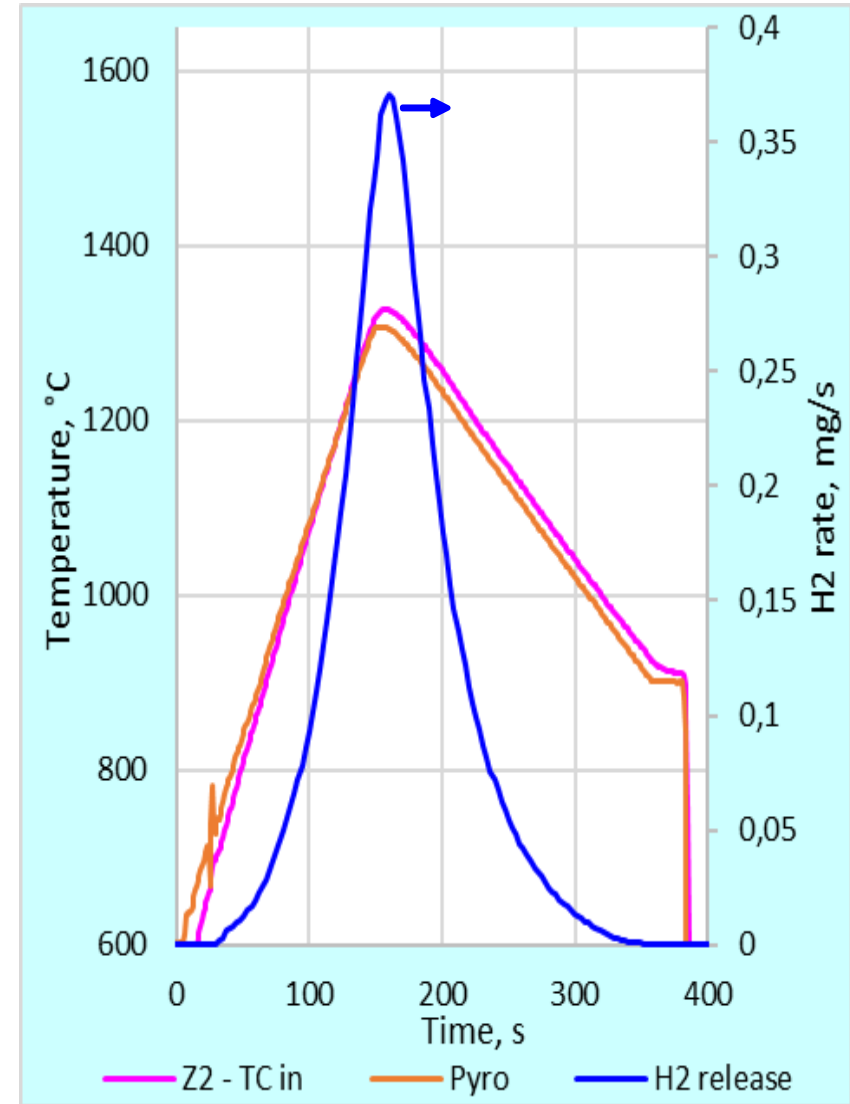
8 K/s, T_{max}=1390 °C, immediate cool-down

8 K/s, T_{max}=1400 °C during 240 s, cool-down

Comparison of coated E110 (2022) and not coated Zry-4: significant lower hydrogen release for coated cladding under extended LOCA (max 1250 °C, cool-down, quench)



coated E110



not coated Zry-4

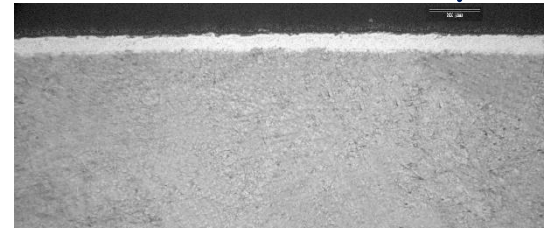
E110

Summary on AEOI samples

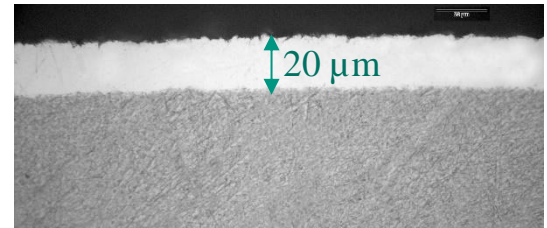
- significant decrease of hydrogen release under extended LOCA conditions (max 1250 °C) in comparison to not coated samples (factor 15).
- Numerous blisters (local swellings) at the outer cladding surface.
Zr-Cr eutectic at 1350 °C and formation of Laves phase ZrCr_2 .

Opt. ZIRLO claddings with OD=9.15 mm; wall thickness 650 μm :

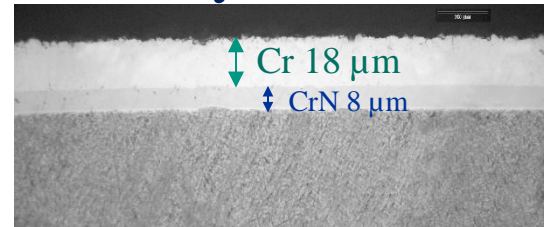
- 2 PVD coated samples with Cr layer thickness of $\approx 5 \mu\text{m}$



- 2 PVD coated samples with Cr layer thickness of $>17 \mu\text{m}$



- 2 PVD coated samples with CrN/Cr multilayer


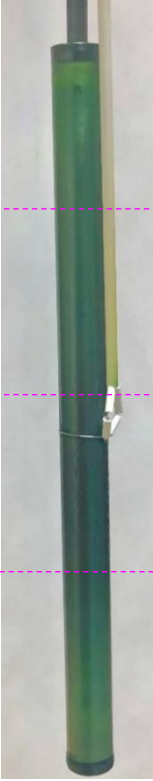


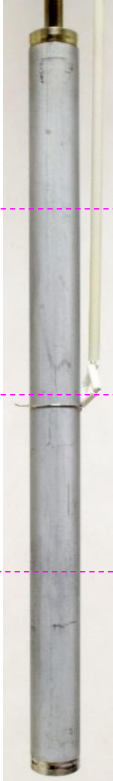
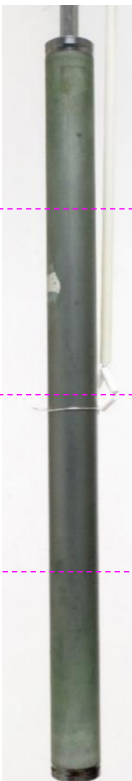





- *Samples are welded with Zr bottom and top plugs*
- *Top plug connected with Zr capillary tube*

Test matrix and mass gains

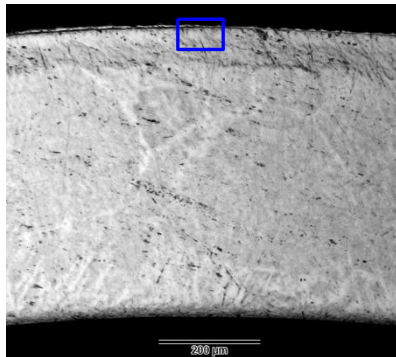
Sample	1.1 (230418a)	1.2 (230419a)	2.1 (230419c)	2.2 (230420a)	3.1 (230421a)	3.2 (230420b)
coating	Cr 5 μm		CrN/Cr		Cr 17 μm	
heat-up, K/s	5	8	5	8	5	8
T_{max} , °C	1250	1400	1250	1400	1250	1400
oxidation duration at T_{max} , s	0	240	0	240	0	240
cool-down, K/s	1.5 to 900 °C	immediately	1.5 to 900 °C	immediately	1.5 to 900 °C	immediately
quench	water	in Ar+steam	water	in Ar+steam	water	in Ar+steam
m_0 , g	26.0662	25.5293	26.2996	26.1545	26.1584	26.1517
m_{pt} , g	26.1036	26.1948	26.3287	26.3265	26.1865	26.7329
Δm , mg	37.4	665.5 - (TC wires)	29.1	172 + (spalled scales) - (TC wires)	28.1	581 - (TC wires)

Outer views of samples oxidized during DBA transients ($T_{max}^{surf} = 1200\text{ °C}$)

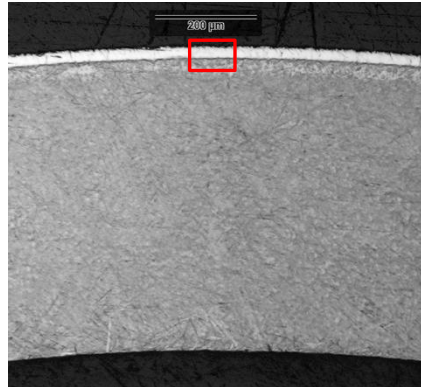
1.1 <i>pre-test</i>	1.1 (Cr≈5 μm) (sample 230418a)	3.1 <i>pre-test</i>	3.1 (Cr>17 μm) (sample 230421a)	2.1 <i>pre-test</i>	2.1 (CrN/Cr; 18/8 μm) (sample 230419c)
					
100 mm					
65 mm hottest elevation					
35 mm					
					
	wrinkled surface		glossy surface		partial oxide spalling

Cross-sections (at hottest elevation of 65 mm) of samples oxidized during DBA transients ($T_{max}^{surf} = 1200\text{ °C}$)

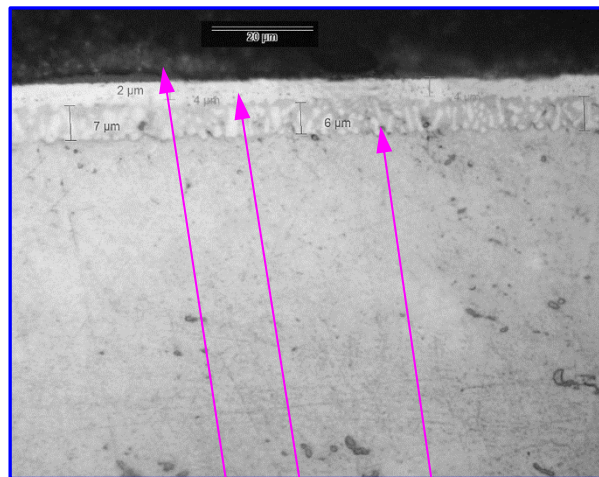
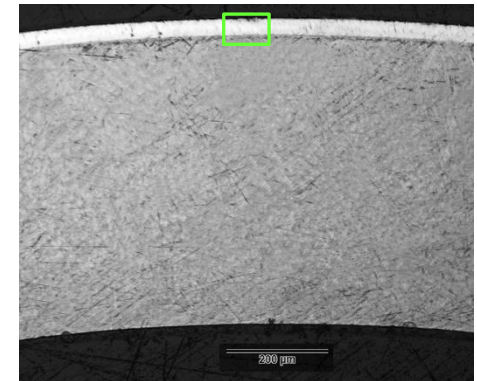
sample 1.1, Cr 5 μm



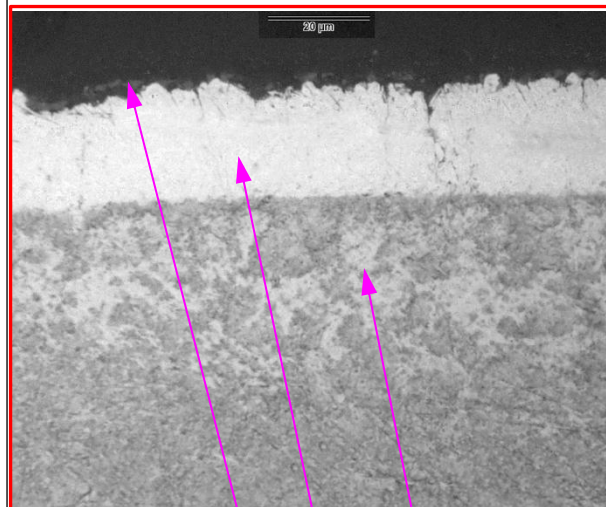
sample 3.1, Cr 20 μm



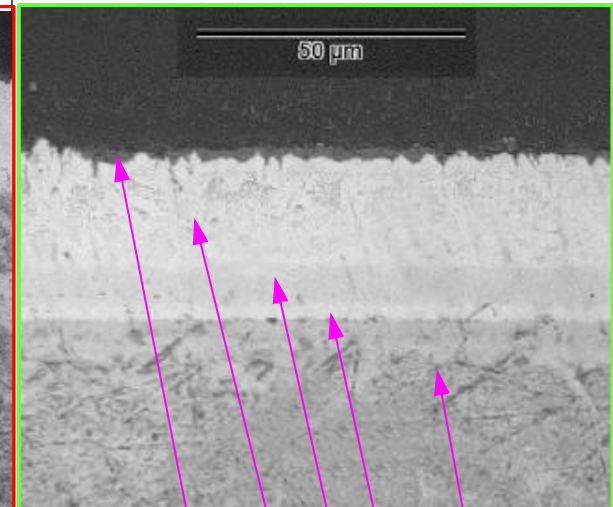
sample 2.1, CrN/Cr multilayer 18/8 μm



3 layers: Cr_2O_3 , Cr, diffusion zone



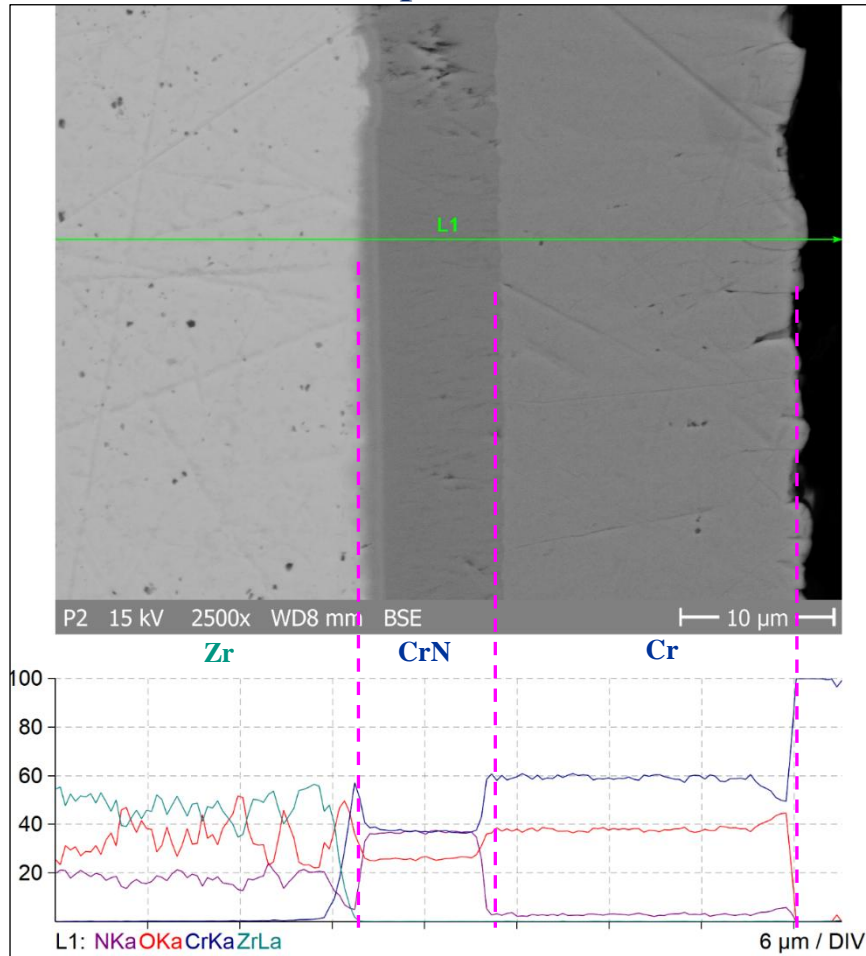
3 layers: Cr_2O_3 , Cr, diffusion zone



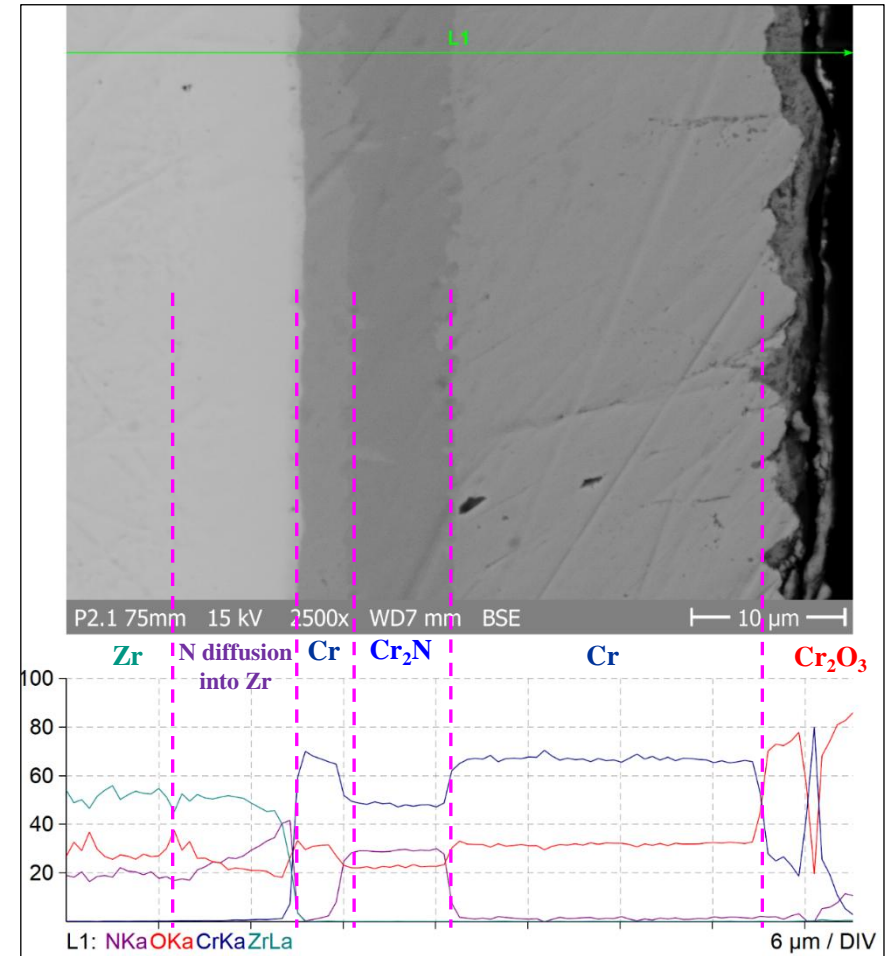
5 layers: Cr_2O_3 , Cr, CrN, Cr, diffusion zone

Features of the behavior of the CrN/Cr sample under transient to $T_{max}^{surf} = \underline{1200\text{ °C}}$



pre-test



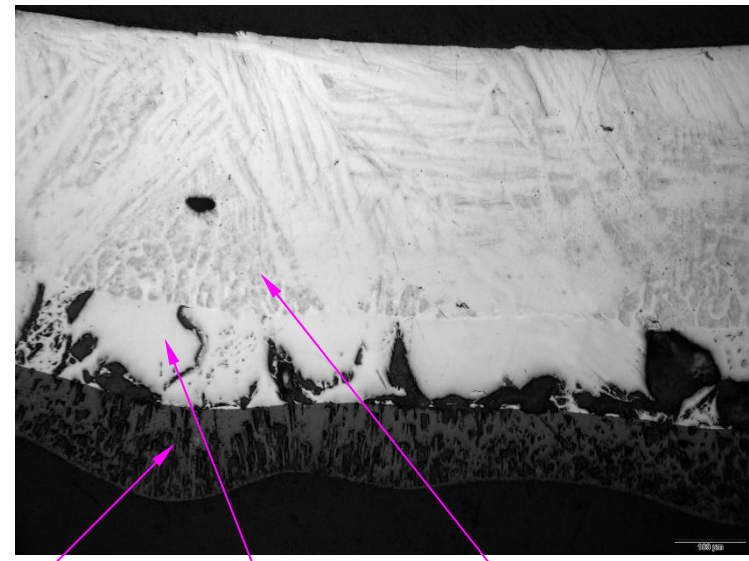
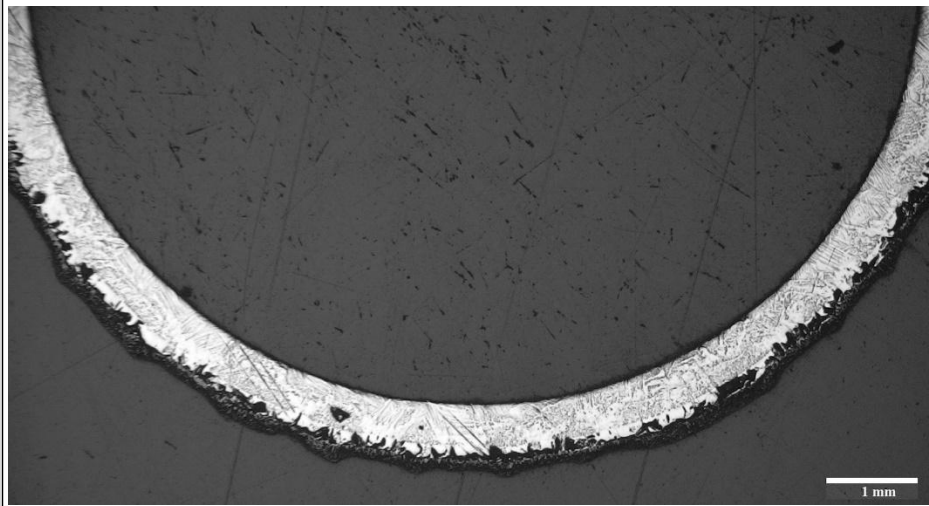
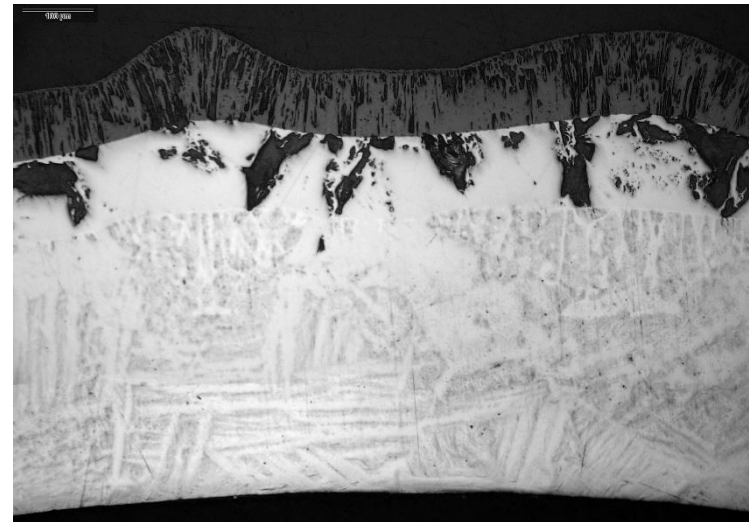
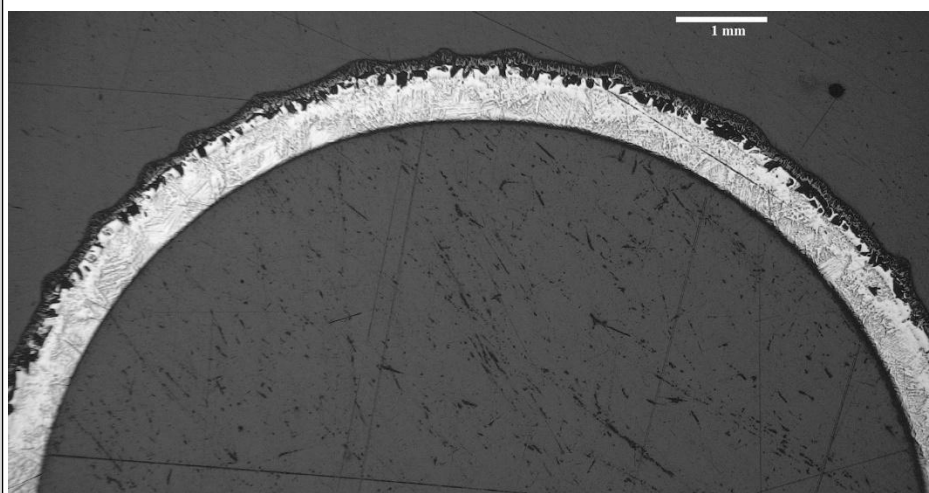
post-test



Outer views of samples oxidized at $T_{max}^{surf}=1400\text{ }^{\circ}\text{C}$

1.1 <i>pre-test</i>	1.2 (Cr \approx 5 μm) (230419a)	3.1 <i>pre-test</i>	3.2 (Cr $>$ 17 μm) (230420b)	2.1 <i>pre-test</i>	2.2 (CrN/Cr) (230420a)
					
					
	fine surface blisters		coarse surface blisters		leaked (Zr, Cr) eutectic melt

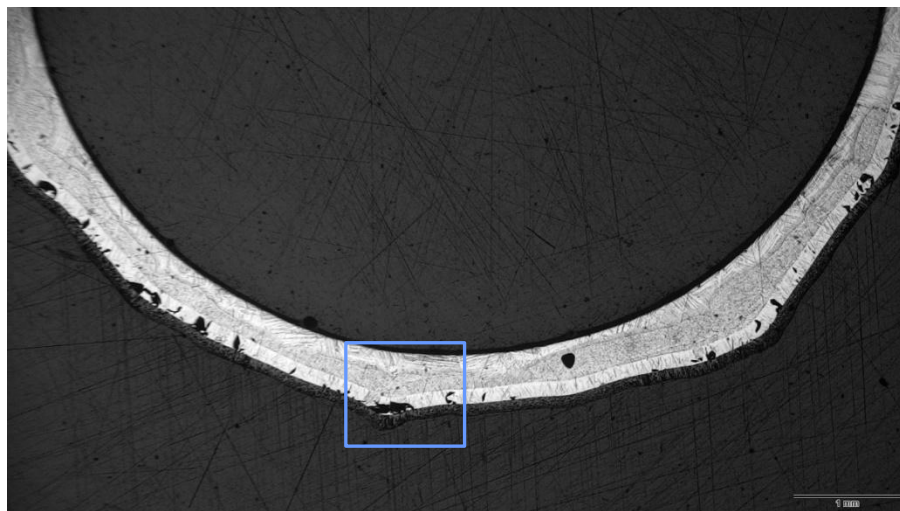
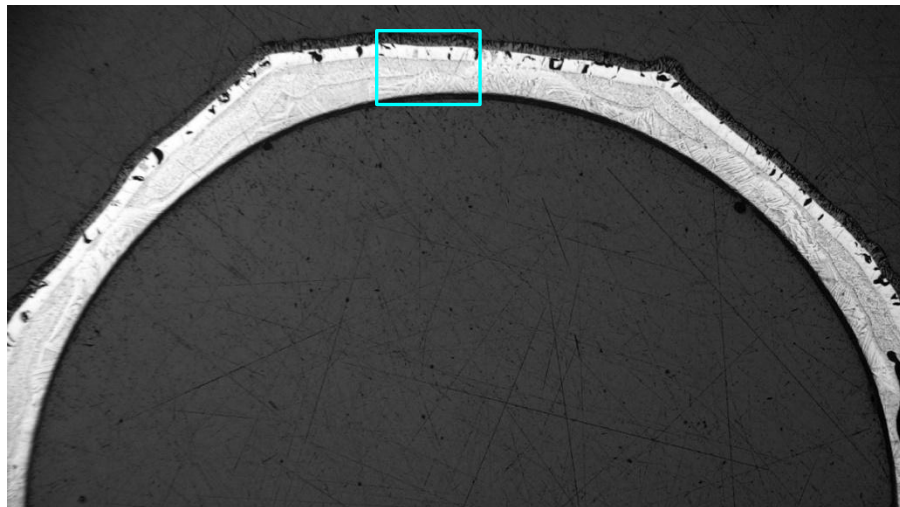
Cross-sections of sample 1.2 (5 μm Cr) at 65 mm
oxidized under BDBA conditions ($T_{\text{max}}^{\text{surf}} = 1400\text{ }^{\circ}\text{C}$)



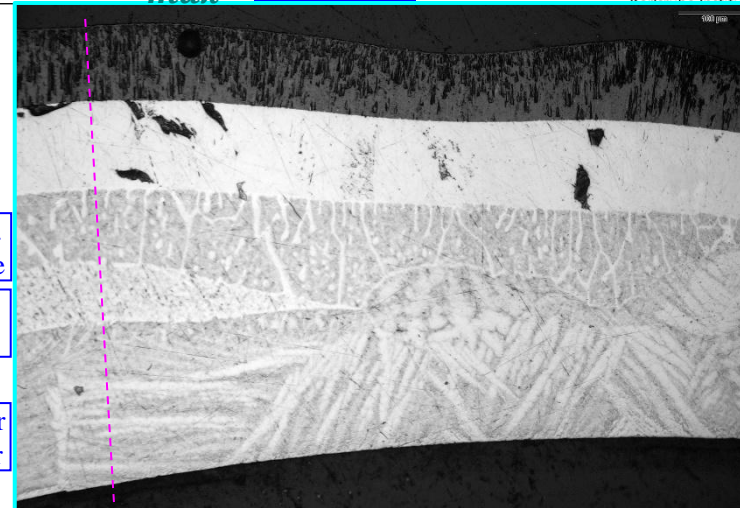
overview: ≈ 1.2 blisters per mm of circumference

ZrO_2 95-130 μm ; $\alpha\text{-Zr(O)}$ 125-140 μm ; $\text{Cr}_{\text{diffusion}}$ in Zr

Cross-sections of sample 3.2 (>17 μm Cr) at 65 mm oxidized under BDBA conditions ($T_{\text{max}}^{\text{surf}} = 1400\text{ }^{\circ}\text{C}$)



overview: ≈ 0.5 blisters per mm of circumference



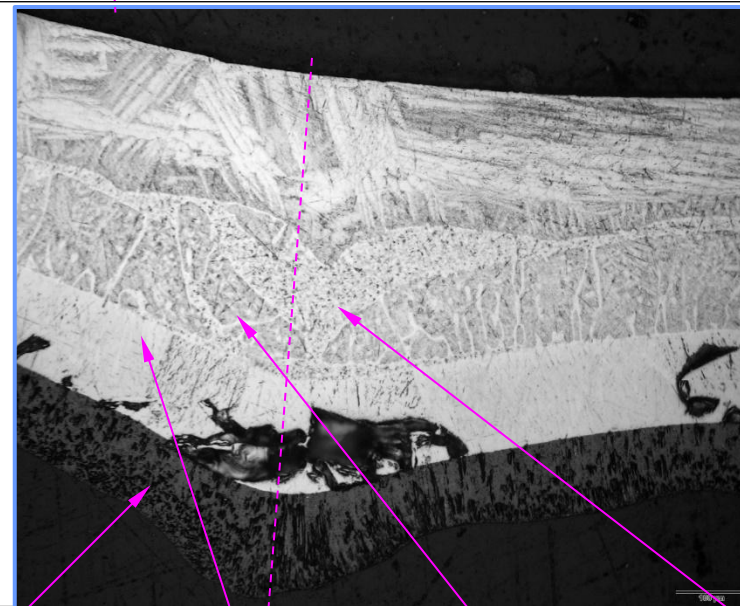
ZrO₂

α -Zr(O)

diff.
zone

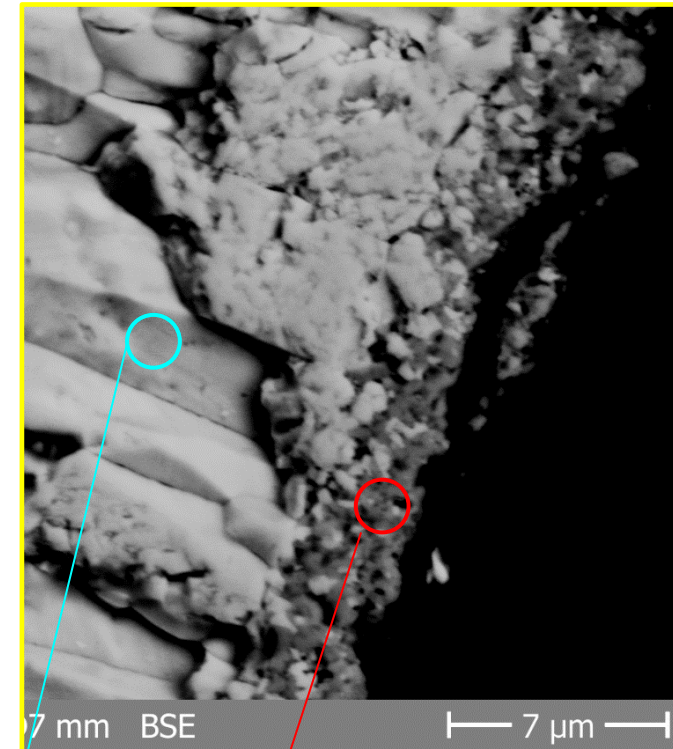
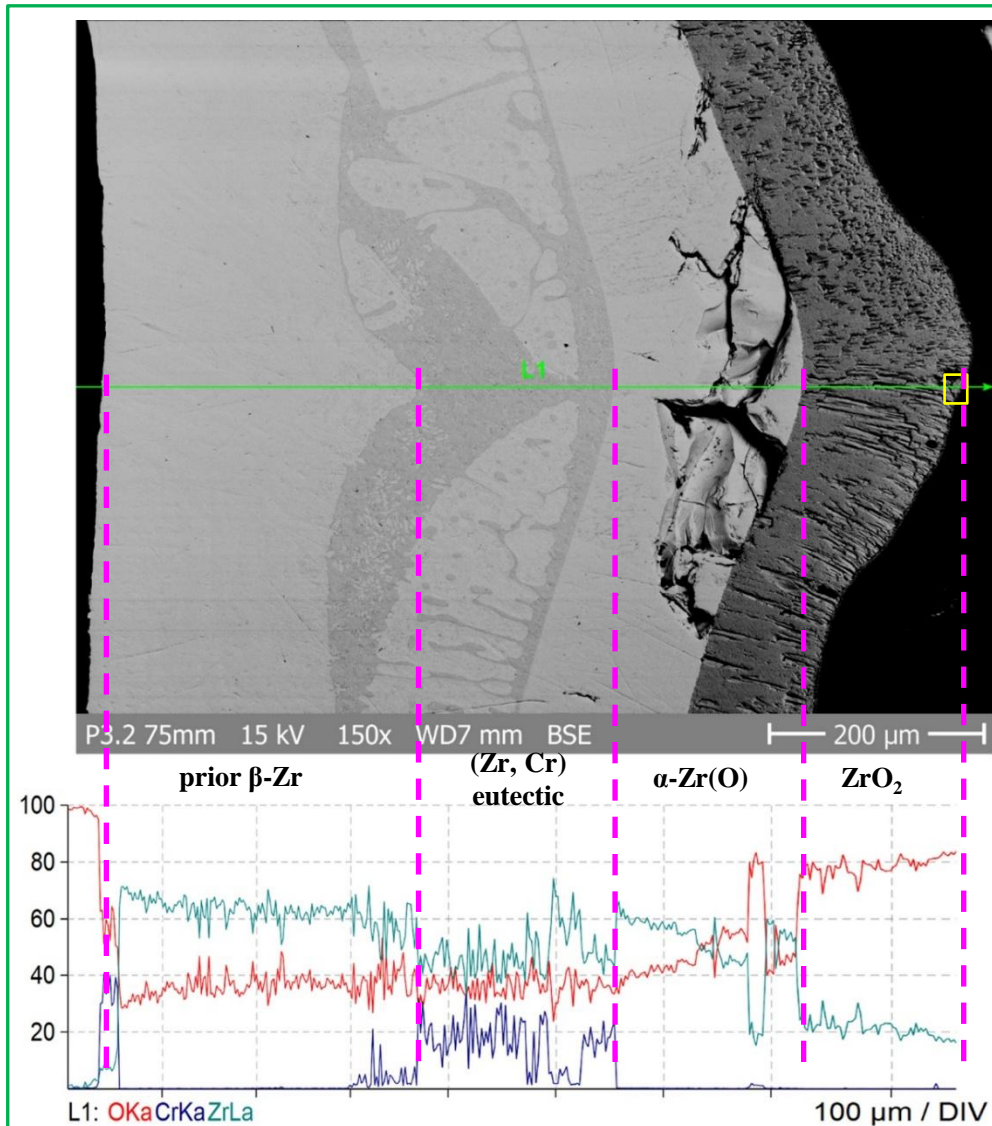
(Zr,
Cr)

prior
 β -Zr



ZrO₂ 95-155 μm ; α -Zr(O) 130-160 μm ; Cr_{diffusion in Zr} 120 μm ; (Zr, Cr)_{max} 120 μm

Cross-sections of sample 3.2 (>17 μm Cr) at 65 mm oxidized under BDBA conditions ($T_{max}^{surf}=1400\text{ °C}$)



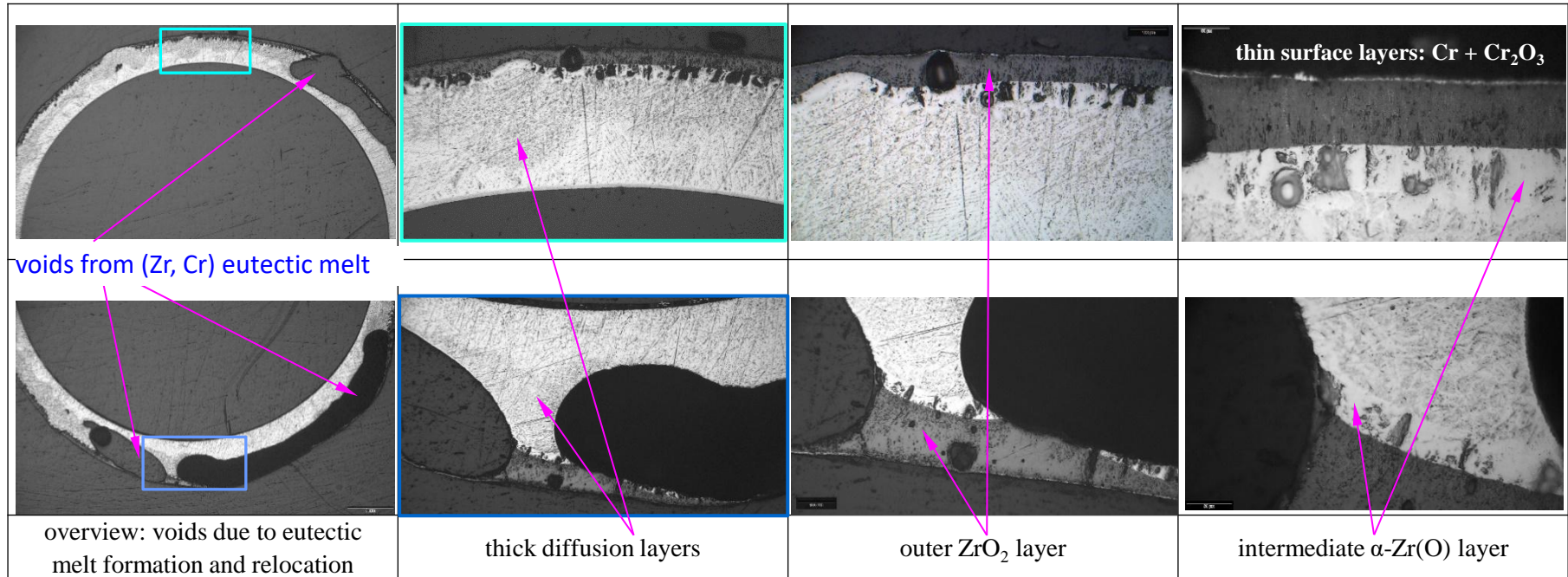
Cr₂O₃

Elt	XRay	Int	Error	K	Kratio	W%	A%
O	Ka	2563.5	64.5061	0.5452	0.4266	58.06	82.78
Cr	Ka	1630.6	1.9068	0.3948	0.3089	35.70	15.66
Zr	La	400.8	1.8673	0.0600	0.0469	6.24	1.56
				1.0000	0.7825	100.00	100.00

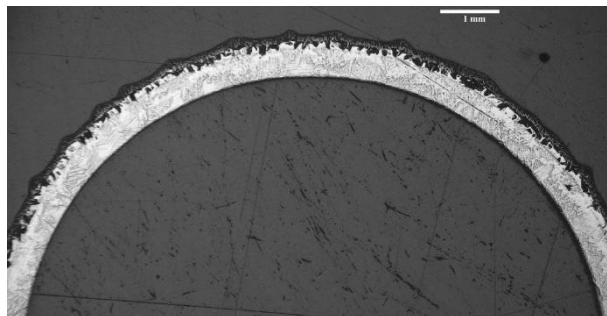
ZrO₂

Elt	XRay	Int	Error	K	Kratio	W%	A%
O	Ka	673.1	1.3042	0.2139	0.1256	46.00	82.91
Cr	Ka	3.0	0.1818	0.0011	0.0006	0.07	0.04
Zr	La	3508.4	6.5569	0.7850	0.4609	53.93	17.05
				1.0000	0.5872	100.00	100.00

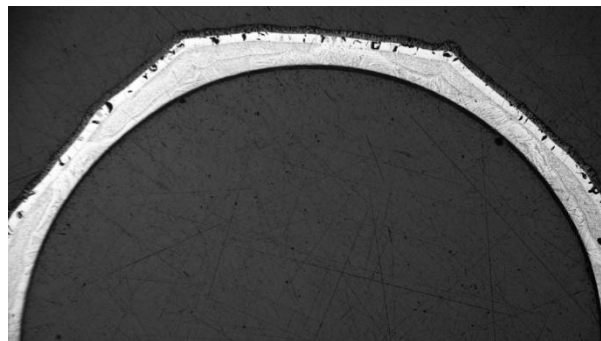
Cross-sections of sample 2.2 (CrN/Cr) at the hottest elevation 65 mm oxidized under BDBA conditions ($T_{max}^{surf} = \underline{1400\text{ °C}}$)



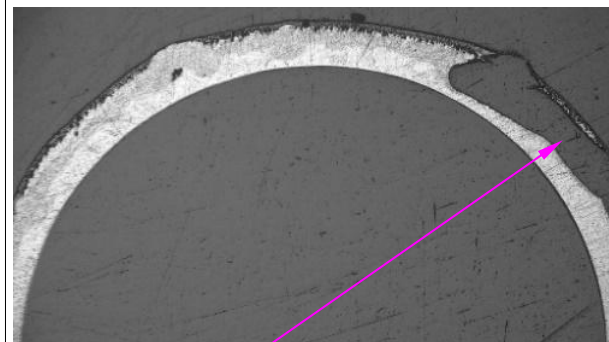
Cross-sections of the UPM/Prague samples at the hottest elevation 65 mm oxidized under BDBA conditions ($T_{max}^{surf} = \underline{1400\text{ °C}}$)



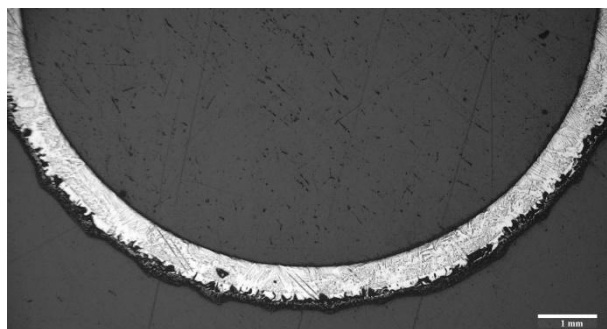
≈1.2 blisters per mm of circumference



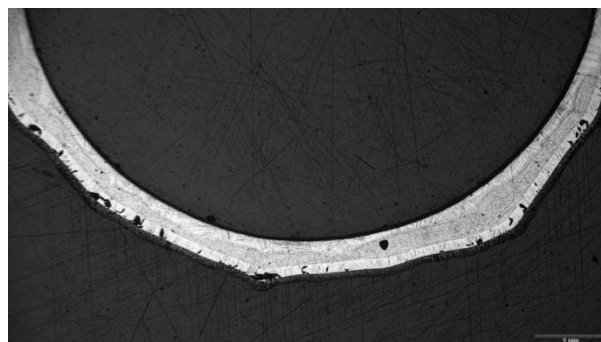
≈0.5 blisters per mm of circumference



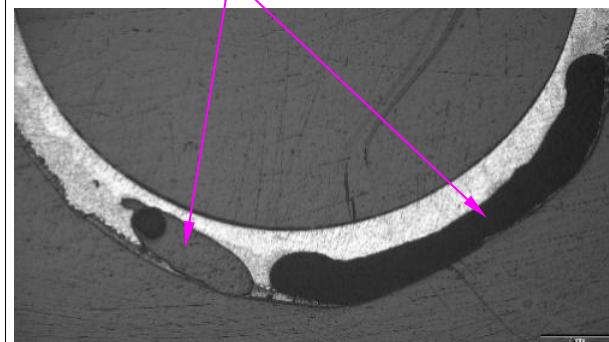
voids from (Zr, Cr) eutectic melt



sample 1.2: 5 μm Cr



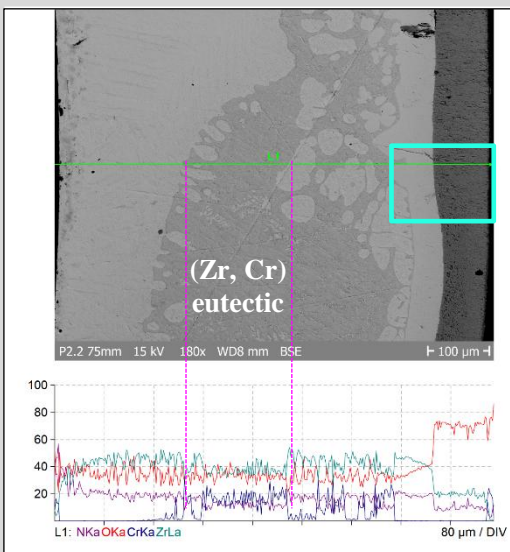
sample 3.2: >17 μm Cr



sample 2.2: Cr (18 μm)/CrN (8 μm)

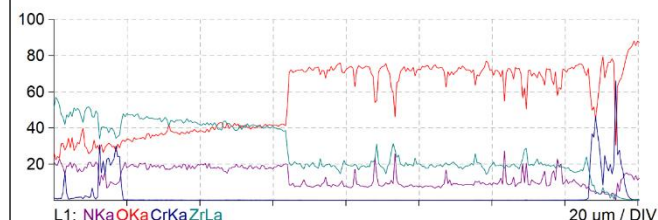
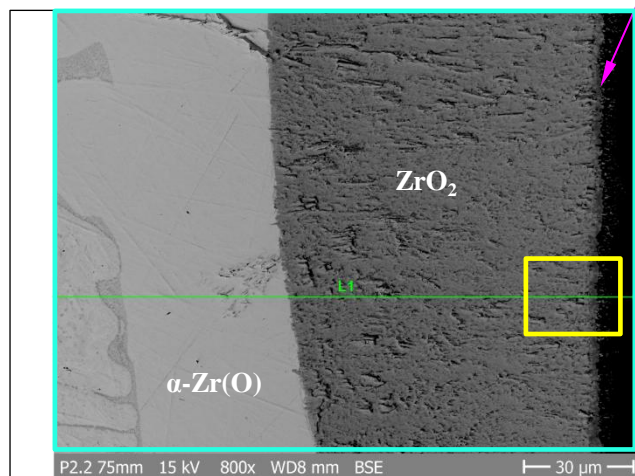
- Formation and relocation of (Zr,Cr) eutectic melt for Cr/CrN coating perhaps due to the lack of intermediate oxidation of part of the zirconium, which could not react with steam due to the barrier effect of CrN layer.

Cross-sections of BDBA sample 2.2 (Cr/CrN) at hottest elevation of 65 mm ($T_{max}^{surf} = 1400\text{ °C}$), circumferential position outside voids

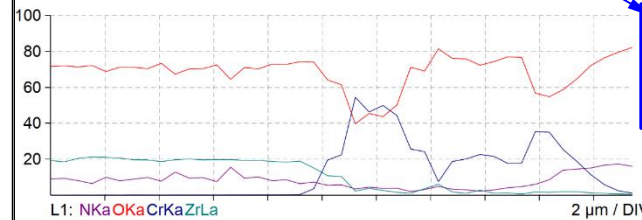
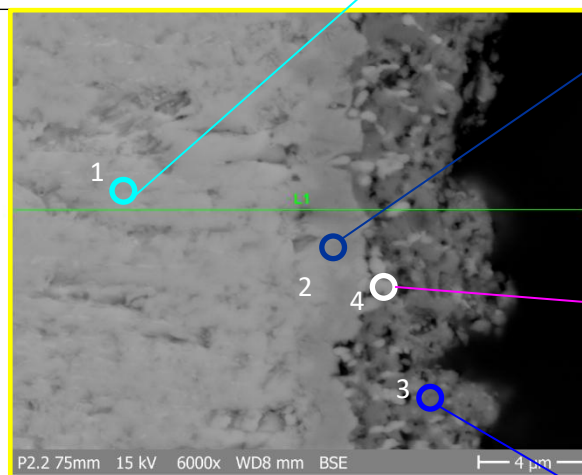


ZrO₂ with ZrN inclusions

Elt	XRaY	Int	Error	K	Kratio	W%	A%
N	Ka	69.1	158.8994	0.0174	0.0103	5.37	11.41
O	Ka	866.9	158.8994	0.1519	0.0901	37.63	69.96
Cr	Ka	14.1	0.3705	0.0028	0.0017	0.18	0.11
Zr	La	6711.8	115.6367	0.8279	0.4910	56.81	18.52



Cr₂O₃



Cr

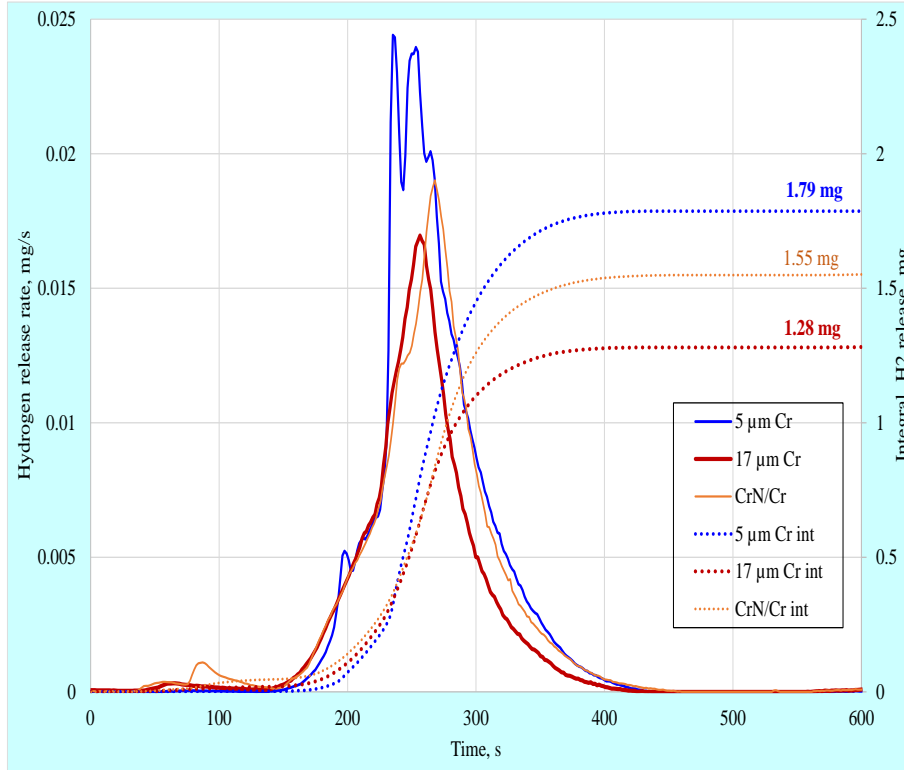
Elt	XRaY	Int	Error	K	Kratio	W%	A%
N	Ka	39.5	235.8792	0.0075	0.0065	1.35	3.46
O	Ka	1126.7	235.8792	0.1482	0.1299	19.46	43.74
Cr	Ka	5248.9	7.2788	0.7861	0.6889	72.62	50.21
Zr	La	629.0	11.3931	0.0582	0.0510	6.57	2.56

Cr₂O₃ with Zr inclusions

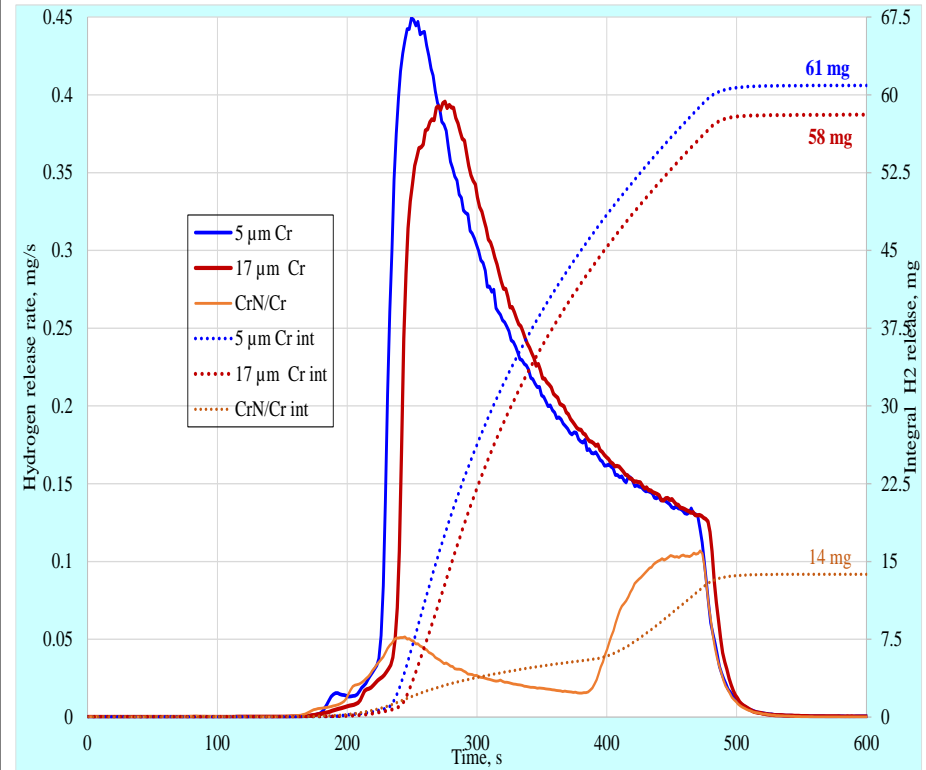
Elt	XRaY	Int	Error	K	Kratio	W%	A%
N	Ka	50.5	205.6407	0.0111	0.0074	1.94	3.43
O	Ka	2452.0	205.6407	0.3755	0.2497	49.74	77.09
Cr	Ka	2351.1	1.9388	0.4099	0.2726	30.95	14.76
Zr	La	1886.9	37.5481	0.2034	0.1353	17.38	4.72

Cr₂O₃

Elt	XRaY	Int	Error	K	Kratio	W%	A%
N	Ka	73.4	390.9179	0.0122	0.0098	1.69	2.88
O	Ka	4089.2	390.9179	0.4723	0.3791	50.62	75.79
Cr	Ka	3692.4	9.0070	0.4855	0.3897	44.47	20.49
Zr	La	370.5	12.8343	0.0301	0.0242	3.22	0.65



hydrogen release under DBA conditions ($T_{max}^{surf} = 1200\text{ °C}$)



hydrogen release under BDBA conditions ($T_{max}^{surf} = 1400\text{ °C}$)

- Increasing the thickness of the chromium layer by 3 times (from 5 to $\approx 17\text{ }\mu\text{m}$) reduced the hydrogen release by 30% under DBA conditions (transient to 1200 °C), however, only by 3% under BDBA conditions (1400 °C). *The barrier CrN layer has only a moderate influence on hydrogen release under DBA, but strong reduced hydrogen release under BDBA (reduction factor 4).*

Summary on Prague samples

- An increase in the thickness of the chromium layer by 3 times (from 6 to 18 μm) reduced the hydrogen release by 30% under DBA conditions (transient to 1200 $^{\circ}\text{C}$), however, only by 3% under BDBA conditions (1400 $^{\circ}\text{C}$). Barrier layer CrN has only moderate influence on hydrogen release under DBA, but strong reduced hydrogen release under BDBA (reduction factor 4).
- Formation of blisters (local swellings) at the outer cladding surface under BDBA conditions ($T_{\text{max}}=1400$ $^{\circ}\text{C}$): fine blisters (1.2/mm) for thin Cr layer (6 μm) and coarse blisters (0.5/mm) for thick Cr layer (18 μm).
- Cr transport from the cladding surface to the boundary of α/β -Zr phases.
- Formation and relocation of (Zr,Cr) eutectic melt for CrN/Cr coating perhaps due to the lack of intermediate oxidation of part of the zirconium, which could not react with steam due to the barrier effect of CrN layer.

Thank you for your attention

<http://www.iam.kit.edu/awp/163.php>
<http://quench.forschung.kit.edu/>