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# Structural Influence on Exciton Formation and the Critical Role of Dark Excitons in Polymeric Carbon Nitrides

Changbin Im, Radim Beranek, and Timo Jacob\*

Polymeric carbon nitrides (PCNs) exhibit intriguing optical properties and exceptional performance in (photo)catalysis, optoelectronics, and energy storage. Nevertheless, the intricate phenomena involving light absorption, formation of long-lived excitons, photo-charging, and photochemical processes observed in PCNs remain poorly understood. This theoretical investigation elucidates the origin of distinct dark and bright excitons, their stability and lifetimes, and their correlation with the microstructural attributes of PCNs. Based on these results, the decisive role of dark excitons in photocatalytic reactivity is proposed, which underlies the experimentally observed differences in the photocatalytic performance of various PCN derivatives. This study thus establishes novel insights into the factors governing the light-driven processes in PCNs that can provide essential guidelines for rational design of PCNs with enhanced performance.

1. Introduction

Polymeric carbon nitrides (PCNs), commonly referred to as graphitic carbon nitrides or melon, along with their (semi)crystalline derivatives such as poly(heptazine imide) (PHI) have emerged as compelling materials for photocatalysis, optoelectronics, and energy storage applications. [1–5] This can be attributed to their straightforward synthesis, robust chemical stability, and optical properties that span from the UV to visible light range. [5] Research on PCNs has demonstrated their efficacy in photocatalytic applications such as hydrogen production

C. Im, R. Beranek, T. Jacob Institute of Electrochemistry Ulm University

Albert-Einstein-Allee 47, 89081 Ulm, Germany

E-mail: timo.jacob@uni-ulm.de

T. Jacob

Helmholtz-Institute-Ulm

Helmholtzstr. 11, 89081 Ulm, Germany

T. Jacob

Karlsruhe Institute of Technology (KIT) P.O. Box 3640, 76021 Karlsruhe, Germany



The ORCID identification number(s) for the author(s) of this article can be found under https://doi.org/10.1002/aenm.202405549

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under near-visible light, water splitting, CO<sub>2</sub> reduction, and H<sub>2</sub>O<sub>2</sub> production.<sup>[6]</sup> Highly crystalline forms of carbon nitrides are typically obtained using molten salt methods, resulting in highly ordered 2D structures.[7] Particularly, these derivatives, containing cations, are of significant interest due to enhanced photocatalytic activity and the photocharging behavior that can be utilized in time-delayed photocatalysis or boosted photocurrent response.[8-10] Despite the plethora of useful and intriguing properties, the knowledge of the relationship between the structural and optical properties in PCN materials is still largely underdeveloped.<sup>[5]</sup> Merschiann et al. reported that charge transport in PCN structures occurs between interplanar layers.[11] Godin et al. discussed

the charge trapping phenomena in conventional PCNs, and Yang et al. reported that the excess electron accumulation in ionic PCNs leads to lower quantum yields due to accelerated electronhole (e-h) recombination.[12,13] However, any clear conclusions from experimental studies of exciton dynamics in PCNs using time-resolved spectroscopy remain limited due to the lack of detailed understanding of the structure-induced excited state properties.[12-14] For example, the frequently observed trapped excitons in PCNs are typically attributed to structural defects. Nonetheless, there is a notable absence of detailed discussion on the nature of these structural defects or their impact on photocatalytic activity. Utilizing many-body perturbation theory based on Green's function formalism can provide accurate descriptions of excited state properties such as accurate bandgap, exciton energy, and absorption spectra. Our previous studies reported strong exciton binding energy,<sup>[15]</sup> and have examined the effects of PCN microstructures, such as corrugation, degree of condensation, and stacking, focusing on their thermodynamic stability, electronic structure, and optical properties.<sup>[16]</sup> We particularly noted that the interlayer interactions in PCNs play a significant role in both thermodynamic and electronic properties.<sup>[17]</sup> Herein, we perform first-principles calculations based on GW and the Bethe-Salpeter equation (GW-BSE),<sup>[18]</sup> to elucidate the crucial link between the various structural motifs in PCNs and the corresponding optical properties, formation of bright and dark excitons, exciton stability and lifetime, as the resulting factors governing the photocatalytic activity.[19-22] We demonstrate that the strongly localized electrons in PCNs induce active interlayer transitions due to the stacked structure of PCNs.

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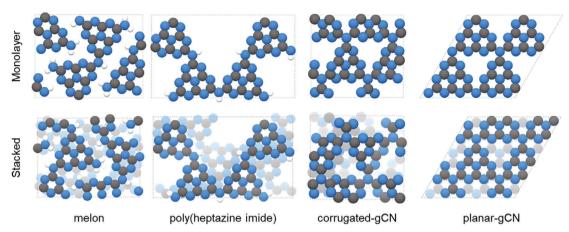


Figure 1. Structures of polymeric carbon nitrides.

Furthermore, we show that the interactions between heptazine units vary with the microstructure of PCNs, affecting the formation and the lifetimes of the dark excitons more significantly than of the bright excitons. Our findings thus establish the pronounced structure-related variations of the dark excitons as a key factor determining the experimentally encountered optical properties and photocatalytic performance of different types of PCNs, providing an alternative perspective to the already established knowledge on the important role of defects and structural heterogeneity. [23,24]

# 2. Results and Discussion

## 2.1. Models

Figure 1 illustrates the different PCN structures considered in this work. These models encompass variations in the degree of poly-condensation of heptazine units, corrugations, and stacking configurations.[16] Increased poly-condensation leads to higher carbon and nitrogen contents, but a reduced hydrogen ratio. The planar-graphitic carbon nitride (p-gCN) is known to be energetically less stable compared to corrugated-graphitic carbon nitride (c-gCN), which represents a local minimum in the energy landscape. [25] PCN structures, composed of these heptazine units, typically exhibit highly-localized  $\pi$  electrons. Thus, we primarily present results for optimized geometries using the Perdew-Zunger parameterization of the Ceperley-Alder Monte Carlo correlation (CA) functional, which is categorized under the local density approximation (LDA) functionals. These functionals provide an efficient description of structures with highlylocalized electrons, considered as key properties. To address the limitations of the MBD@rsSCS dispersion correction, which depends on the exchange-correlation energy from the LDA functionals, we also included results obtained with the PBE-D3 xcfunctional. This approach offers a better description of interlayer properties. Results of the optimized geometry with PBE-D3 are available in the Supporting Information for comparison (Figures \$9-\$20, Supporting Information). The lattice parameters of the optimized geometries obtained with both functionals are summarized in Table S1 (Supporting Information).

# 2.2. Optical Spectra

The sole use of density functional theory (DFT) to accurately describe optical properties has numerous limitations.<sup>[26,27]</sup> To address these challenges, we use the BSE to address the energy originating from *e*–*h* interactions on top of the quasi-particle energy derived from GW calculations. This GW-based quasi-particle energy incorporates precisely screened *e–e* interactions, thereby enhancing the accuracy of single excitation property predictions. Notably, the quasi-particle energy adjusted by the BSE correction typically exhibits a red shift compared to those predicted by the independent quasi-particle (IP) approximation, which omits *e*–*h* interactions. By analyzing the imaginary part of the macroscopic dielectric functions from both GW@BSE and GW@IP calculations, we can estimate the exciton binding energy in 2D and 3D PCN structures, as illustrated in Figure 2. Spatially-projected absorption coefficients reveal distinct patterns depending on the microstructural variations (degrees of condensation and stacking) of PCNs. First, when transitioning from 2D to 3D PCNs, the positions of the main absorption peaks in the 2D structures remain largely unchanged, while additional peaks emerge. This indicates that the strong absorption characteristics of a heptazine unit at around 4 eV,[16] attributed to the strong electronic localization of heptazine units, are preserved regardless of the PCN structure. However, the absorption edges and the xx and yy components of the absorption coefficients at energy above 4 eV show significant changes between 2D and 3D configurations. This suggests that the stacking of heptazine units induces diverse optical interactions depending on the microstructure. Furthermore, 2D PCN structures show strong exciton binding energy (>2 eV).[15] However, this significant exciton binding energy diminishes to less than 1 eV in 3D stacked structures. Moreover, the transition from a 2D monolayer to a stacked 3D configuration alters the absorption edges and reduces the optical anisotropy observed in melon-2D and c-gCN-2D structures. Since the optical anisotropy is attributed to the in-plane joint patterns of the heptazine units formed during poly-condensation, the reduced optical anisotropy implies that the fractions of the 2D PCN micro-structure can indeed interact with each other throughout the 3D assembly. In brief, this means that diverse forms of PCN fragments within the PCN structure interact electronically and optically through their

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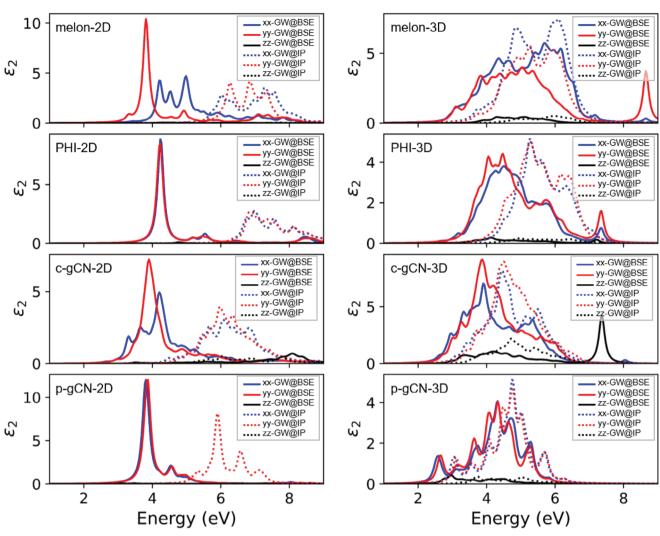


Figure 2. Imaginary part of dielectric functions. The solid and the dashed lines are obtained by the GW@BSE and the GW@IP respectively.

stacking configurations, regardless of size or shape. Our findings provide clear evidence of strong interlayer interactions, resulting in dramatically decreased exciton binding energy and changes in absorption patterns including altered optical anisotropy and absorption edges.

To better elucidate the origins of the absorption features near the absorption edge, in **Figure 3** we depict the oscillator strengths and averaged absorption spectra along spatial directions for PCN 2D and 3D structures obtained from *GW@BSE* calculations. First, we observed that each 2D structure of PCNs exhibits distinct absorption spectra patterns. When these structures are stacked to form a 3D structure, additional peaks arise due to interlayer interactions. These emergent peaks are more pronounced in flat structures (melon and p-gCN), where interlayer interactions result in more absorption patterns. In contrast, in buckled structures (PHI and c-gCN), small absorption peaks appear due to interlayer interactions with the red-shifted 2D peaks. The strong intensity of absorption peaks near 200 nm in 3D PCNs is due to the stacking of infinite 2D models, indicating that finite-sized ribbon structures exhibit a reasonable decrease in the inten-

sity of these absorption peaks. [17] The peaks appearing beyond 400 nm originate from the interactions between layers and the in-plane  $\pi$ – $\pi$ \* transitions, which will be discussed below. The absorption edges of the different PCN structures are obtained as 2.93 eV for melon-3D, 2.60 eV for PHI-3D, 2.62 eV for c-gCN-3D, and 1.69 eV for p-gCN-3D. These PCN structures and bandgaps are completely in line with the experimentally observed bandgaps of PCNs, which range from 2.6 to 2.8 eV. [5.28–30] Especially, the frequently observed long absorption tails can not only be attributed to thermal vibrational motion, but can have a distinct structural origin, primarily from the graphitic (i.e., fully condensed) domains. [31]

#### 2.3. Exciton Formation

Figure 3 demonstrates the oscillator strengths, which, according to the Franck–Condon factor and Fermi's golden rule, are strongly correlated with the overlap of electronic densities and thus represent the probabilities of electronic transitions at each

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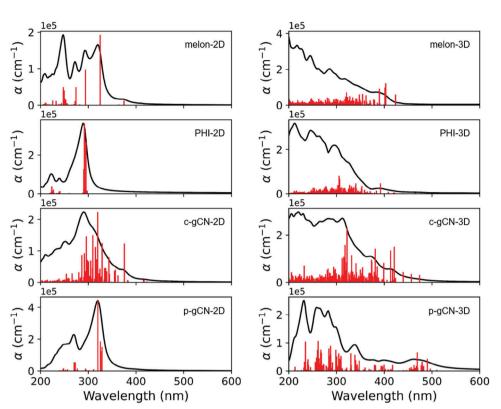


Figure 3. Illustration of the calculated absorption spectra for 2D and 3D PCN models along averaged spatial directions. The oscillator strengths in relation to wavelength are presented in red.

energy level. Cases of high oscillator strengths with relatively low absorption peaks in the spectrum can arise due to the influence of dipole momentum and selection rules. More specifically, a strong oscillator strength accompanied by a low absorption coefficient indicates that the corresponding transitions occur infrequently. We propose that such transitions are primarily driven by dark excitons involved in interlayer charge transfer. Interlayer transitions along the *z*-direction are more selective and less frequent compared to in-plane transitions, which may cause them to appear optically inactive or forbidden in computational results. This explanation also aligns with the discrepancy between the calculated and experimentally measured absorption spectra. While computational models predict a gradual decrease in the absorption coefficient at longer wavelengths, experimental measurements mostly show strong absorption peaks near the absorption edge. This can be attributed to thermal excitation or phonon interactions, which are not accounted for in static calculations but significantly enhance interlayer transitions in real systems. Notably, most electronic transitions near the absorption edge in PCN 3D structures are combined with the dark states. Our results from the BSE calculations provide the coupling coefficients corresponding to transitions between energy levels at each high symmetry point, as shown in Figure 4. The coupling constants of the other lowest 32 excited states are presented in the Supplemental Material. The coupling coefficients (radius of circles) at the band edges for each PCN 3D structure are displayed at the top line of the figure, and the coupling coefficients that are most pronounced for interlayer interactions are displayed at the bottom line. At the absorption edges, stable excitons are related to

transitions between the band edges in melon, c-gCN, and p-gCN structures while relatively more degenerate excitons are formed in the PHI structure. The coupling constants at the bottom line of Figure 4 represent the exciton formation at the Z-point in each structure, indicating a direction toward the other layer from the  $\Gamma$ -point. These interlayer transitions are identified as the electron transitions from the sub-levels of the valence band to near the conduction band edge, indicating that these transitions involve different  $\pi$  electrons than those associated with band edge transitions. Our previous study found that absorption peaks of PCN structure can be sorted as in-plane  $\pi$ - $\pi$ \* transitions and  $\pi$ - $\pi$ \* interlayer transitions. [17] When comparing the relative stability of excitons formed in the PCN structures at the interlayer, the c-gCN structure exhibits the most stable exciton with an energy gap of 2.96 eV. Melon and PHI also show stabilization of interlayer excitons at 3.11 and 2.97 eV, respectively, each showing roughly a 0.3 eV difference from their absorption edges. The differences in exciton stability at the interlayer are determined by the degree of overlap between orbitals of different layers, specifically being regulated by the magnitude of the transition momentum off-diagonal components. These off-diagonal components turn out to be the degree of interaction of the nitrogen non-bonding (*n*) electrons, which impact the  $\pi$  electrons at the interlayers. This explains how the interlayer distances are related to the configurations of PCN structures. This is more clearly understood when comparing the coupling constant plots of LDA-optimized geometries with different interlayer distances (see Table S1) to those of the generalized gradient approximation (GGA)-optimized geometry results (Figures S11 and S13-S20, Supporting Information).

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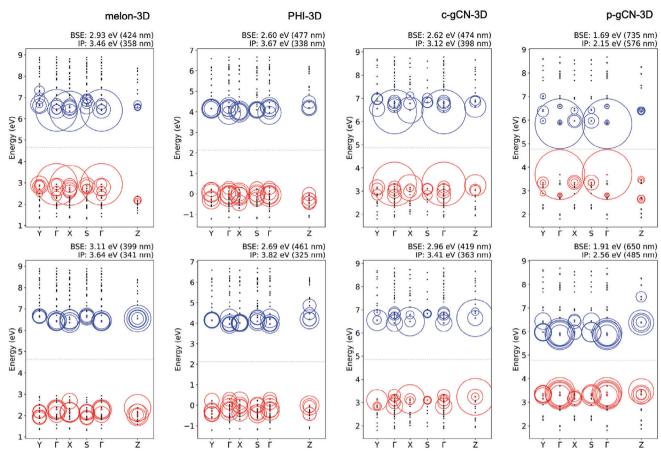


Figure 4. Representation of eigenvalues (BSE and IP) and coupling coefficients (radius of circles) of bright states for both 2D and 3D PCN structures, corresponding to the absorption edges (top) and interlayer interactions (bottom).

Furthermore, our previous study found that the thermodynamic contributions from vibrational motion due to the flexible layered structure of PCNs are non-negligible. [16] Thus, we expect that enhanced interlayer interactions may be evident due to interlayer vibration modes.

#### 2.4. Prolonged Exciton Lifetime and Photocatalytic Activity

We have calculated the exciton lifetimes and effective lifetimes at 300 K based on the energy required for each excited state, transition probability, and transition dipole momentum. The obtained behavior is shown in Figure 5 and summarized in Table 1. Here, we define the energy states with the top 5% oscillator strengths as bright states (blue) and the rest as dark states (red) (see Experimental Section).[32] First, the effective lifetimes of bright excitons in 3D PCN structures are slightly increased compared to those in 2D structures. This suggests that most bright excitons in PCN structures are formed by the direct in-plane transitions of strongly localized  $\pi$  electrons and the interlayer  $\pi$  transitions. Therefore, the stacking configuration helps stabilize these bright excitons. In contrast, the dark excitons formed in 3D structures are destabilized by interlayer interactions. Most dark excitons are related to in-plane  $n-\pi^*$  or distorted  $\pi-\pi^*$  transitions and interlayer interactions cannot sufficiently stabilize these states

due to the lack of interaction. Notably, the effective lifetime of dark excitons increases dramatically in graphitic structures (cgCN and p-gCN). This indicates the presence of a new form of excitons in g-CNs (cg-CN and pg-CN), characterized by deeply trapped dark excitons with remarkably long lifetimes, which distinguish them from the shallowly trapped dark excitons typically observed in melon and PHI structures. We propose that these long-lived excitons are closely related to the trapped excitons reported in the literature.[12] Furthermore, we suggest that the calculated lifetimes of dark excitons shed light on the photo-charging phenomenon observed in ionic-PHI.[8-10,13,14,33] Notably, the photocharging phenomenon persists in the protonexchanged structure (H-PHI), indicating that the phenomenon is more attributed to the high symmetry crystalline regions rather than the influence of existing cations. However, the principle and the nature of electron accumulation in PHIs have not been elucidated so far. Our results demonstrate that the extended lifetime of excitons is observed in both 3D stacked and 2D graphitic structures (c-gCN and p-gCN), indicating that these long lifetimes are primarily due to structural features rather than interlayer interactions. [34–36] In contrast, the PHI structures exhibit the shortest dark exciton lifetime even shorter than those of melon structures. This might explain the general experimental observation that the photo(electro)catalytic activity and photoinduced electron accumulation in ionic (PHI-type) PCNs are enabled only

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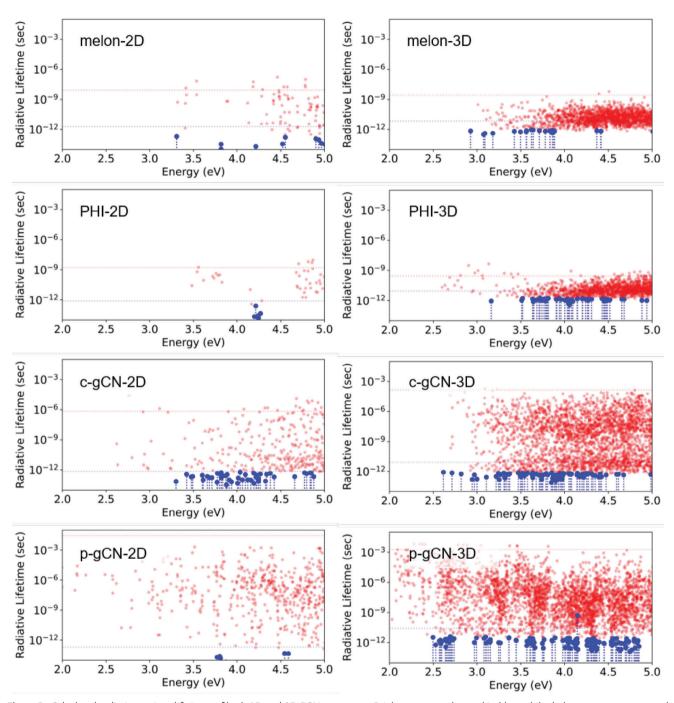


Figure 5. Calculated radiative exciton lifetimes of both 2D and 3D PCN structures. Bright states are denoted in blue, while dark states are represented in red. Their respective effective lifetimes at 300 K are indicated by dashed lines.

in the presence of effective electron donors, such as alcohols, that can quickly extract the hole from the short-lived excitons in PHI-type PCNs.<sup>[8–10,13,14,37]</sup>

In order to further elucidate the nature of these dark states, **Figure 6** shows the electronic density corresponding to the electron and the electron hole for the dark excitons of the longest lifetime in the PHI and the c-gCN structure. It clearly shows that these dark excitons (*e*–*h* densities) are different in PHI and c-gCN structures. This suggests that the repulsive force between hep-

tazine units—specifically, the interaction between the n-electron of nitrogen atoms—causes a distorted orientation of the n or  $\pi$  orbitals through the structural corrugation. The varying degrees of this distortion control the selection rules, resulting in allowed transitions in PHI and forbidden transitions in c-gCN for dark excitons. This forbidden in-plane transition further hinders the interlayer interaction due to the lack of orbital overlaps. Moreover, this distortion also affects the energy levels of  $\pi$  states, resulting in a larger energy gap between  $\pi^*$  and  $\pi^*_{\rm dist}$  in c-gCN than in the

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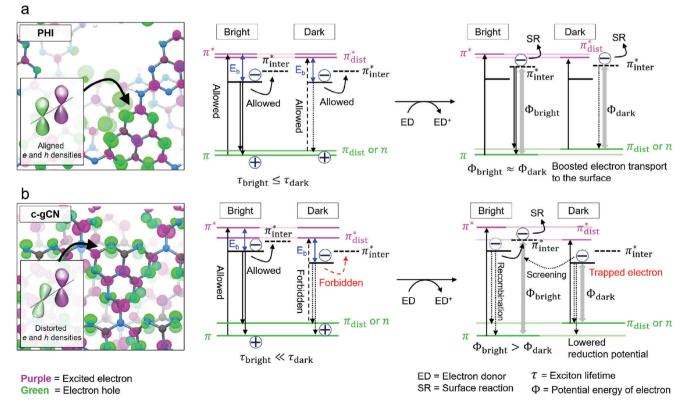
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**Table 1.** Bandgaps of 2D and 3D PCN structures (optimized using CA functionals) were calculated via independent particle approximations  $(G_0W_0@1P)$  and BSE  $(G_0W_0@BSE)$  methods, alongside the effective exciton lifetime at 300 K. (D) and (I) denote the direct and indirect bandgaps, respectively.

	Structure	G₀W₀@IP [eV]	G <sub>0</sub> W <sub>0</sub> @BSE [eV]	Exiton lifetime <	τ <sub>300K</sub> > dark
2D	melon	5.50 (D)	3.31	1.96 ps	8.70 ns
	PHI	5.83 (I)	3.48	0.78 ps	1.65 ns
	c-gCN	4.48 (D)	2.62	8.36 ps	141.00 μs
	p-gCN	4.26 (I)	1.95	0.20 ps	28.73 ms
3D	melon	3.46 (I)	2.93	6.96 ps	2.72 ns
	PHI	3.67 (D)	2.60	8.77 ps	277.34 ps
	c-gCN	3.12 (D)	2.62	8.36 ps	141.00 μs
	p-gCN	2.15 (I)	1.69	26.04 ps	1.92 ms

case of PHI. Therefore, the larger distortion of the orbitals can exhibit prolonged exciton lifetimes and lower energy gaps for the dark states. (See Table 1) In the case of melon, the presence of hydrogen bonding slightly alters one of the  $\pi$  electrons, resulting in dark states with lifetimes similar to but slightly longer than those in PHI.<sup>[17]</sup> In p-gCN, the strongly forced forbidden n– $\pi$ \* transitions due to the high symmetry lead to the detrimental effect of dark states, similar to those in c-gCN.

Furthermore, these findings shed light on the experimentally observed detrimental effects of deep charge trap states on photocatalytic activity in both conventional and ionic (PHI) carbon nitrides. [12,13] The significant distortion of  $\pi$  or n orbitals can create deeply trapped dark excitons compared to the bright states (Figure 6b). Due to their longer exciton lifetimes, these trapped excitons are less affected by hole-quenching rates and can effectively accumulate electrons. However, these electrons, being spatially distorted and trapped, can not directly contribute to the interlayer interactions for the carrier transport, resulting in significantly reduced quantum yields due to their strongly diminished reductive potential, as reported in the literature.[12] Moreover, the presence of these trapped electrons might shield interlayer charge transport of the bright states, which could further promote the recombination of bright excitons. The presence of these deeply trapped electrons also suggests that they can be used as a descriptor to indicate whether conventional or ionic (PHI) carbon nitrides obtained from the synthesis process contain graphitic domains. On the other hand, in PHI structures (Figure 6a), the dark states of the less affected  $\pi$  electrons exhibit similar energy levels to the bright states, forming shallow trapped states. Due to this similarity, these dark states can have lifetimes as short as the bright states. Although a fast electron donor rate is required for hole-quenching due to these short lifetimes, once the hole of the dark exciton is filled, the nearly undistorted orientation of the  $\pi$  electrons enables efficient interlayer carrier transport, leading to effective surface catalytic reactions. We assume that these dark-state-induced phenomena (e.g., enhanced photocatalytic



**Figure 6.** (left) Configurations of the electron density for electron holes and excited electrons in PHI and c-gCN and (right) the proposed mechanisms of photoexcitation process for a) PHI and b) c-gCN-3D structures, where  $\pi_{dist}$  and  $\pi_{dist}^*$  are the energy levels of distorted  $\pi$  electrons,  $\pi_{inter}$  is the  $\pi^*$  state at the other layer, and E<sub>b</sub> is the exciton binding energy, respectively.

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activity, electron accumulation enabling photodoping and/or time-delayed photocatalysis) in PHI can be more pronounced in the presence of cations, as the stabilization of these excess electrons is also a key factor in preventing recombination.<sup>[8,9,33]</sup>

## 3. Conclusion

Our study elucidates the principles underlying the exciton formation, transitions, and lifetimes within PCN structures. The results establish a valuable link between the structural and optical properties of PCN materials, which is a key prerequisite for rational tuning of their electronic structures for optimal performance. Utilizing GW@BSE calculations, we find the decisive role of dark excitons for effective photocatalytic performance. Notably, the structure-dependent factors (e.g., interlayer interactions, variations in exciton binding energy resulting from stacking configurations) governing the dark exciton formation and dynamics elucidate the experimentally observed differences in photo(electro)catalytic performance of different types of carbon nitrides. Specifically, the effective interlayer transfer of electrons in ionic (PHI-type) carbon nitrides predicted by our calculations is in line with their typically higher photo(electro)catalytic activity and possibility of electron accumulation phenomena (e.g., dark or delayed photocatalysis, photodoping). On the other hand, very short dark exciton lifetimes in PHIs (as compared to conventional carbon nitrides containing graphitic domains) shed light on the experimentally encountered fact that photocatalysis with PHIs typically requires the presence of readily reacting reducing agents such as alcohols, whereby slower reactions, such as water oxidation, are not accessible even with the use of co-catalysts (which, in contrast, is feasible with gCNbased conventional carbon nitrides). Finally, our study also establishes a valuable and coherent theoretical framework for further spectroscopic investigations and provides several guidelines for the knowledge-driven design of PCN-based materials with tailored properties. For example, promising future experimental efforts might include synthesis of i) few-layered PCN structures to achieve combination of low exciton binding energy with efficient interlayer charge transfer; ii) PCN structures favoring AA stacking configurations to enhance interlayer electronic coupling; iii) mixed or hybrid PCN materials with combined microstructures (e.g., PHI and gCN) to combine the distinct advantages of both classes of materials and to fine-tune the exciton trapping depth, lifetimes, interlayer transfer and charge storage capacity.

# 4. Experimental Section

All ab initio calculations were performed using VASP (6.4.1), which uses the projector augmented wave potentials (PAWs). [38] We considered PCN structures optimized by the LDA and the GGA functionals, respectively, to compare the description of the highly-localized  $\pi$ -electrons within the heptazine units. [17] The CA functional [39,40] with the many-body dispersion energy method (MBD@rsSCS)[41,42] is used within LDA, while PBE-D3 is adopted for the GGA calculations. [43,44] To describe the single-particle excitations of the PCN structure, a single-shot GW calculation was conducted to avoid bandgap overestimation. [26,45] In brief, the GW approximation neglects the vertex functions in Hedin's formalisms, and the self-energy  $(\Sigma)$ 

is given as the product of the Green function (G) with the screened self interactions (W). Then, the quasi-particle energy is given by:

$$E_{nk}^{QP} = E_{nk} + Z_{nk} \langle \psi_{nk} | \Sigma - V_{xc} | \psi_{nk} \rangle \tag{1}$$

where **k** and *n* are the wave vector and band index,  $Z_{nk}$  is the renormalization function obtained from the self-energy with DFT eigenvalues.<sup>[46]</sup> In the  $G_0W_0$  calculations, 256 bands were included for the 2D PCN-structure and 512 bands for 3D PCN-structure, respectively. For the 2D and 3D PCN structures,  $\Gamma$ -centered 2 × 2 × 1 and 2 × 2 × 2 **k**-space meshes are used, respectively. The quasi-particle energy is updated by using BSE to include e-h interactions within the Tamm–Dancoff approximation (TDA),<sup>[47]</sup> given by:

$$(E_{ck}^{QP} - E_{\nu k}^{QP}) A_{\nu ck} + \sum_{\nu'c'k'} \langle \nu ck | K_{e-h} | \nu'c'k' \rangle A_{\nu'c'k'} = \Omega A_{\nu ck}$$
 (2)

where c and v are indices of conduction and valance bands, respectively,  $A_{\nu ck}$  are e-h coupling coefficients,  $\Omega$  are the BSE eigenvalues, and  $K_{e-h}$  is the e-h interaction kernel. [48,49] For the BSE calculations, 32 eigenvalues were considered. Then, the IP energy considering electron–electron (e-e) interactions is obtained as GW@IP, and the BSE is used to include e-h interactions, finally denoted as GW@BSE. The real and the imaginary parts of the macroscopic dielectric functions are computed including local field effect. [27] From the calculated oscillator strengths, the bright excitons are selected having the 5% of the highest intensity, and the rest are classified as the dark states. The VASP source code was modified to obtain the transitional dipole momentum kernel ( $\mu_{\rm S}$ ) to calculate the radiative decay rate of the exciton state S ( $\gamma_{\rm S}$ ) at wavevector  ${\bf Q}=0$  using the equation, given by: [50]

$$\gamma_{\rm S}(0) = \tau_{\rm S}(0)^{-1} = \frac{8\pi e^2 E_{\rm S}(0)}{\hbar^2 c} \frac{\mu_{\rm S}^2}{A}$$
 (3)

where  $E_{\rm S}$  is the exciton energy at  ${\bf Q}$ , A is the area of the unit cells, c is the speed of light, and  $\tau_{\rm S}$  is the exciton lifetime. Furthermore, the effective radiative lifetime  $\langle \tau_{\rm eff} \rangle$  was calculated at temperature (T) assuming a neglectable exciton momentum in PCN structures at high symmetry points. [51] Therefore, the effective radiative lifetime using the Boltzmann average of  $\tau_{\rm S}$  is given by:

$$\langle \tau_{\text{eff}} \rangle^{-1} = \frac{\sum_{S} \tau_{S}^{-1} e^{-E_{S}(0)/k_{B}T}}{\sum_{S} e^{-E_{S}(0)/k_{B}T}} \tag{4}$$

All the geometries, the raw-data, and the patch for VASP are available via  $Zenodo.^{[52]}$ 

### **Supporting Information**

Supporting Information is available from the Wiley Online Library or from the author

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#### **Conflict of Interest**

The authors declare no conflict of interest.

# **Data Availability Statement**

The data that support the findings of this study are available in the supplementary material of this article.

# **Keywords**

carbon nitride, charge transfer, dark exciton, photocatalysis

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