



# Global impact of anthropogenic NH<sub>3</sub> emissions on upper tropospheric aerosol formation

Christos Xenofontos<sup>a,1</sup>, Matthias Kohl<sup>b</sup>, Samuel Rühl<sup>b</sup>, João Almeida<sup>c</sup>, Lucía Caudillo-Plath<sup>d</sup>, Romulo Cruz-Simbron<sup>e</sup>, Lubna Dada<sup>f</sup>, Jonathan Duplissy<sup>g</sup>, Sebastian Ehrhart<sup>b</sup>, Henning Finkenzeller<sup>g</sup>, Kristina Höller<sup>h</sup>, Weimeng Kong<sup>i</sup>, Felix Kunkler<sup>b</sup>, Clara J. Lietzke<sup>e</sup>, Bernhard Mentler<sup>i</sup>, Aleksandra Morawiec<sup>k</sup>, Antti Onnela<sup>c</sup>, Pedro Rato<sup>c,d</sup>, Birte Röhrup<sup>e</sup>, Douglas M. Russell<sup>d</sup>, Meredith Schervish<sup>l</sup>, Wiebke Scholz<sup>j</sup>, Milin Kaniyodical Sebastian<sup>h</sup>, Mario Simon<sup>d</sup>, Eva Sommer<sup>c,k</sup>, Yandong Tong<sup>e</sup>, Nsikanabasi Silas Umo<sup>h</sup>, Gabriela R. Unfer<sup>m</sup>, Lejish Vettikkat<sup>n</sup>, Boxing Yang<sup>f</sup>, Wenjuan Yu<sup>g</sup>, Imad Zgheib<sup>o</sup>, Zhensen Zheng<sup>p</sup>, Joachim Curtius<sup>d</sup>, Neil M. Donahue<sup>q,r,s</sup>, Richard C. Flagan<sup>i</sup>, Hamish Gordon<sup>q</sup>, Imad El Haddad<sup>f</sup>, Armin Hansel<sup>l</sup>, Hartwig Harder<sup>b</sup>, Xu-Cheng He<sup>g,t</sup>, Jasper Kirkby<sup>c,d</sup>, Markku Kulmala<sup>u,v</sup>, Katrianne Lehtipalo<sup>g,w</sup>, Ottmar Möhler<sup>h</sup>, Tuukka Petäjä<sup>g</sup>, Mira L. Pöhlker<sup>m</sup>, Siegfried Schobesberger<sup>n</sup>, Dominik Stolzenburg<sup>x</sup>, Mingyi Wang<sup>y</sup>, Paul M. Winkler<sup>k</sup>, Douglas R. Worsnop<sup>g,z</sup>, Michael Höpfner<sup>aa</sup>, Rainer Volkamer<sup>e</sup>, Andrea Pozzer<sup>ab</sup>, Jos Lelieveld<sup>a,b,1</sup>, and Theodoros Christoudias<sup>a,1</sup>

Affiliations are included on p. 9.

Edited by Mark Thieme, University of California San Diego, La Jolla, CA; received March 26, 2025; accepted September 17, 2025

**Anthropogenic ammonia (NH<sub>3</sub>) emissions have significantly increased in recent decades due to enhanced agricultural activities, contributing to global air pollution. While the effects of NH<sub>3</sub> on surface air quality are well documented, its influence on particle dynamics in the upper troposphere-lower stratosphere (UTLS) and related aerosol impacts remain unquantified. NH<sub>3</sub> reaches the UTLS through convective transport and can enhance new particle formation (NPF). This modeling study evaluates the global impact of anthropogenic NH<sub>3</sub> on UTLS particle formation and quantifies its effects on aerosol loading and cloud condensation nuclei (CCN) abundance. We use the EMAC Earth system model, incorporating multicomponent NPF parameterizations from the CERN CLOUD experiment. Our simulations reveal that convective transport increases NH<sub>3</sub>-driven NPF in the UTLS by one to three orders of magnitude compared to a baseline scenario without anthropogenic NH<sub>3</sub>, causing a doubling of aerosol numbers over high-emission regions. These aerosol changes induce a 2.5-fold increase in upper tropospheric CCN concentrations. Anthropogenic NH<sub>3</sub> emissions increase the relative contribution of water-soluble inorganic ions to the UTLS aerosol optical depth (AOD) by 20% and increase total column AOD by up to 80%. In simulations without anthropogenic NH<sub>3</sub>, UTLS aerosol composition is dominated by sulfate and organic species, with a marked reduction in ammonium nitrate and aerosol water content. This results in a decline of aerosol mass concentration by up to 50%. These findings underscore the profound global influence of anthropogenic NH<sub>3</sub> emissions on UTLS particle formation, AOD, and CCN production, with important implications for cloud formation and climate.**

new particle formation | anthropogenic NH<sub>3</sub> emissions | UTLS | CCN | AOD

Ammonia (NH<sub>3</sub>) plays a critical role in atmospheric new particle formation (NPF) by stabilizing acid-base nucleation (1). It also contributes to particle growth through condensation (2, 3). The resulting particles can affect climate by scattering and absorbing solar radiation (4, 5), while they can also act as cloud condensation nuclei (CCN) that seed cloud droplets (6–8). It is estimated that approximately half of all CCN—and nearly all in the upper troposphere (UT)—originate from NPF (9, 10). The specific contribution of anthropogenic NH<sub>3</sub> emissions to particle formation in the upper troposphere-lower stratosphere (UTLS), CCN production, and their influence on atmospheric aerosol remains unquantified. This study aims to address this knowledge gap.

NH<sub>3</sub> is released into the atmosphere from biogenic and anthropogenic sources (11). While biogenic sources such as soil microbial activity (12), decomposition of animal waste (13), and vegetation (14) contribute to global NH<sub>3</sub> emissions, anthropogenic activities remain the dominant source. Agricultural practices, including fertilizer application (15) and livestock farming (16), account for 80 to 90% of global NH<sub>3</sub> emissions (17), with additional contributions from industrial processes (18), vehicle emissions (19), and biomass burning (20). These emissions are concentrated in major agricultural regions, primarily Europe (21), the United States (22), South and East Asia (23, 24), with Central Africa (25) and South America (26) also identified as high-emission regions. Previous studies indicated that NH<sub>3</sub> emissions can increase CCN concentrations by up to 80%

## Significance

Ammonia (NH<sub>3</sub>) emissions from human activities can significantly influence aerosol processes in the upper troposphere and lower stratosphere (UTLS). Using an Earth system model, we show that anthropogenic NH<sub>3</sub> strongly enhances new particle formation and growth, leading to substantial changes in UTLS aerosol composition and abundance. These changes can enhance cloud condensation nuclei concentrations by a factor of 2.5 in the upper troposphere over high-emission regions. In addition, aerosol optical depth can increase by up to 80%, potentially affecting climate. Our findings underscore the need to account for UTLS NH<sub>3</sub>-driven aerosol processes in Earth system models to improve predictions of atmospheric composition and cloud effects in climate scenarios.

The authors declare no competing interest.

This article is a PNAS Direct Submission.

Copyright © 2025 the Author(s). Published by PNAS. This open access article is distributed under [Creative Commons Attribution License 4.0 \(CC BY\)](https://creativecommons.org/licenses/by/4.0/).

PNAS policy is to publish maps as provided by the authors.

<sup>1</sup>To whom correspondence may be addressed. Email: c.xenofontos@cyi.ac.cy, jos.lelieveld@mpic.de, or t.christoudias@cyi.ac.cy.

This article contains supporting information online at <https://www.pnas.org/lookup/suppl/doi:10.1073/pnas.2506658122/-DCSupplemental>.

Published October 27, 2025.

over the Asian monsoon region (27). However, the exact contribution of anthropogenic  $\text{NH}_3$  and its global impact on CCN was not addressed. As anthropogenic  $\text{NH}_3$  emissions are projected to double by 2100 (28), it is crucial to understand their global impact in terms of UTLS particle formation and climatic influence through CCN and aerosol optical depth (AOD) changes.

As most  $\text{NH}_3$  sources are situated at the Earth's surface, only a small fraction can reach the UTLS due to the short atmospheric lifetime (29). This limited longevity results from interaction with aerosol and clouds, which promotes efficient removal through scavenging and condensation processes (30). Consequently, gas-phase  $\text{NH}_3$  concentrations can be expected to decrease sharply with altitude, resulting in low levels in the UTLS, particularly over marine regions. This is supported by various in situ observations (31, 32). However, due to the semivolatile nature, hydrophilicity, and highly variable ambient concentrations of  $\text{NH}_3$ , in situ observations in the UTLS are subject to considerable uncertainty (33). Analysis of the MIPAS (Michelson Interferometer for Passive Atmospheric Sounding) spaceborne instrument infrared limb-emission spectra has revealed elevated concentrations of  $\text{NH}_3$  (reaching up to 30 pptv) in the UTLS over emission hotspots (34). Höpfner et al. (34) further noted that limited modeled outputs are available for comparison with these observations.

Accurately representing  $\text{NH}_3$ -induced particle formation in the UTLS remains challenging with atmospheric models. This is partly because most of the models do not include the recently discovered (35) synergistic interaction of  $\text{NH}_3$  with nitric acid ( $\text{HNO}_3$ ) and sulfuric acid ( $\text{H}_2\text{SO}_4$ ), which results in NPF, particularly in typical UTLS conditions. Furthermore, atmospheric models often do not accurately represent the dynamics of particle growth to CCN sizes, a critical step linking NPF processes to cloud formation and climate (7).

The contribution of  $\text{NH}_3$  to the formation of ammonium nitrate ( $\text{NH}_4\text{NO}_3$ ) and its effects on AOD are another significant source of uncertainty in atmospheric models (36). This is primarily due to the variability and limitations in emission inventories (37, 38). It is suggested that  $\text{NH}_4\text{NO}_3$  may become the largest contributor to anthropogenic AOD by the end of the 21st century due to increasing agricultural  $\text{NH}_3$  emissions (39). Aircraft observations have revealed high concentrations of  $\text{NH}_3$  as well as  $\text{NH}_4\text{NO}_3$  aerosols in the UTLS (40, 41), where a low scavenging rate allows them to persist longer than in the lower troposphere (42). This extended residence time can enhance their contribution to AOD. Representing all these processes in atmospheric models is critical for accurately predicting UTLS particle formation processes, their contribution to atmospheric composition, and their broader implications for climate (43, 44).

This study quantifies the global influence of anthropogenic  $\text{NH}_3$  on UTLS particle formation and its implications for CCN concentrations and AOD. Uniquely, we evaluate the modeled UTLS  $\text{NH}_3$  precursor concentrations against MIPAS observations, and we take into account the interactions between  $\text{NH}_3$ ,  $\text{HNO}_3$ , and  $\text{H}_2\text{SO}_4$  in NPF. Recently published NPF parameterizations from the CERN CLOUD (Cosmics Leaving Outdoor Droplets) chamber (2, 35, 45–48) are implemented into the state-of-the-art EMAC (ECHAM/MESSy Atmospheric Chemistry) Earth system model (49). We note that the recently discovered UTLS organic nucleation (50) is not yet included in our study due to the lack of available parameterizations. Isoprene-derived oxidation products can significantly increase NPF in convective outflow regions over tropical rainforests (51). We quantify the contribution of convective updrafts to NPF rates and particle number concentrations in the UTLS. Furthermore,

by comparing multiyear simulations with and without global anthropogenic  $\text{NH}_3$  emissions, we estimate the changes induced by anthropogenic  $\text{NH}_3$  in the UTLS aerosol composition, number, and mass concentrations. The relative importance of the various UTLS aerosol components to the total AOD is calculated, critical for simulating the aerosol direct climatic effect (52). Finally, we quantify the impact of anthropogenic  $\text{NH}_3$  emissions on UT CCN number concentrations, revealing notable changes that could influence cloud formation.

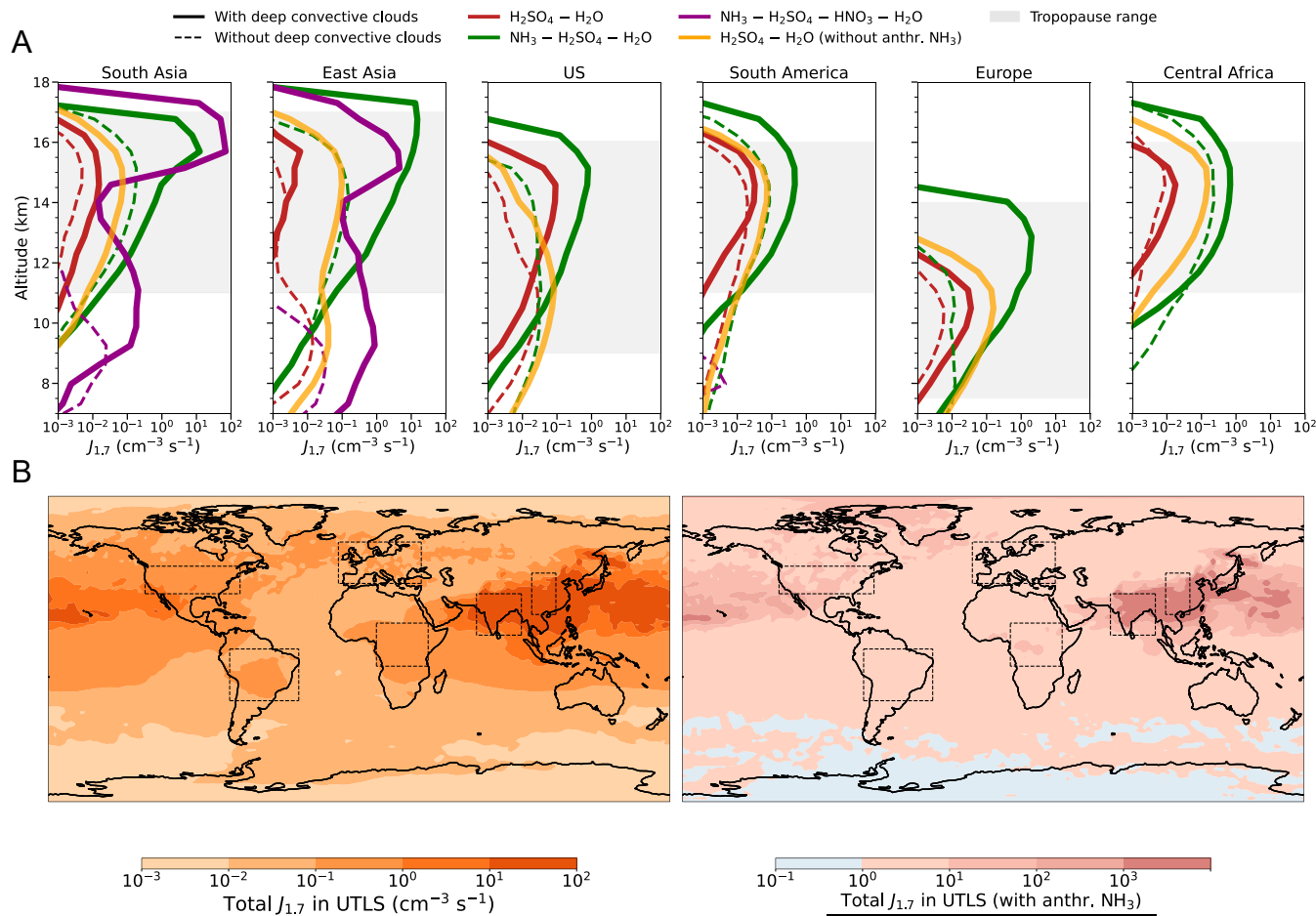
## Results and Discussion

**Convection-Induced NPF.** Convection is thought to influence NPF by transporting precursor vapors from the boundary layer to higher altitudes in the UTLS (2). We investigate the impact of convection-induced NPF by simulating nucleation rates at 1.7 nm diameter ( $J_{1.7}$ ) in the UTLS, showing average global and regional profiles from 2003–2019 (Fig. 1). Here, the UTLS is defined as approximately 7 to 18 km altitude, encompassing the variable tropopause region. The simulated time period enables a direct comparison of  $\text{NH}_3$  concentrations with satellite data from MIPAS, IASI (Infrared Atmospheric Sounding Interferometer), and AIRS (Atmospheric Infrared Sounder). The model reproduces the observed  $\text{NH}_3$  concentrations with a normalized mean bias within  $\pm 15\%$  (Methods and SI Appendix).

Fig. 1A shows the vertical profiles of  $J_{1.7}$  over the  $\text{NH}_3$  strong-emission regions. Composites of days with deep convective clouds are compared to clear-sky days for six major  $\text{NH}_3$  emission hotspot regions: South and East Asia, the United States (US), Europe, South America, and Central Africa. The results indicate that ternary  $\text{NH}_3 - \text{H}_2\text{SO}_4 - \text{H}_2\text{O}$  nucleation is overall the dominant mechanism across all regions, with rates significantly enhanced in the presence of deep convective clouds. Binary  $\text{H}_2\text{SO}_4 - \text{H}_2\text{O}$  nucleation is, on average, two to three orders of magnitude lower than ternary nucleation. Synergistic  $\text{NH}_3 - \text{H}_2\text{SO}_4 - \text{HNO}_3 - \text{H}_2\text{O}$  nucleation is more prominent over South and East Asia. These regions are characterized by frequent convective updraft events, in particular during the Asian monsoon, substantial anthropogenic  $\text{NH}_3$  emissions and abundant nitrogen oxides ( $\text{NO}_x$ ), e.g., from lightning (53).

Zhao et al. (10) incorporated the  $\text{NH}_3$ -enhanced NPF CLOUD parameterizations to estimate the contributions of different nucleation pathways to the total  $J_{1.7}$ . In our results, the peak  $J_{1.7}$  is, on average, an order of magnitude higher than that reported by Zhao et al. over  $\text{NH}_3$  hotspots. This difference arises from distinct averaging approaches. Zhao et al. averaged over all days throughout their simulation period, whereas we separately average over composites of days with and without deep convection. This allows us to focus on and quantify the specific enhancement in  $J_{1.7}$  associated with deep convection (27). Under clear-sky conditions over  $\text{NH}_3$  hotspots, our peak  $J_{1.7}$  values are comparable to those in Zhao et al.

In our additional baseline simulation, we switch off anthropogenic  $\text{NH}_3$  emissions globally, with all other conditions kept constant. This ensures that any differences between the two simulations arise solely from the absence of anthropogenic  $\text{NH}_3$ . In this scenario, the binary  $\text{H}_2\text{SO}_4 - \text{H}_2\text{O}$  nucleation mechanism dominates. However, on average, the peak  $J_{1.7}$  is one to three orders of magnitude lower than  $\text{NH}_3$ -enhanced nucleation. The spatial distribution highlights that with anthropogenic  $\text{NH}_3$ , there are strong enhancements in the total UTLS  $J_{1.7}$  over industrialized and agricultural regions, particularly South and East Asia and the United States, compared to the scenario without anthropogenic  $\text{NH}_3$  (Fig. 1B).



**Fig. 1.** (A) Average simulated nucleation rates at 1.7 nm diameter ( $J_{1.7}$ ), calculated at ambient temperature and pressure, as a function of altitude. The six major  $\text{NH}_3$  emission regions are indicated with dashed outlines over the map in (B): South and East Asia, the United States, Europe, South America, and Central Africa. The nucleation mechanisms included are synergistic  $\text{NH}_3\text{--H}_2\text{SO}_4\text{--HNO}_3\text{--H}_2\text{O}$  (purple); ternary  $\text{NH}_3\text{--H}_2\text{SO}_4\text{--H}_2\text{O}$  (green); binary  $\text{H}_2\text{SO}_4\text{--H}_2\text{O}$  without anthropogenic  $\text{NH}_3$  emissions (orange); and binary  $\text{H}_2\text{SO}_4\text{--H}_2\text{O}$  in the presence of anthropogenic  $\text{NH}_3$  emissions (brown). The solid lines denote conditions with deep convective clouds, whereas the dashed lines represent instances of a quiescent atmosphere without convective clouds. In some regions, the purple lines representing synergistic nucleation are not visible because the contribution of this mechanism is very low, resulting in  $J_{1.7}$  values below the range shown in the plot. In contrast, synergistic nucleation is enhanced in Asia. The gray shaded area represents the tropopause altitude range in each region. (B) Global map of total simulated UTLS  $J_{1.7}$  with anthropogenic  $\text{NH}_3$  (Left) and the enhancement ratio due to  $\text{NH}_3$  (Right). The enhancement is defined as the ratio of UTLS  $J_{1.7}$  with anthropogenic  $\text{NH}_3$  to UTLS  $J_{1.7}$  without. Values greater than 1 indicate an increase in  $J_{1.7}$  due to  $\text{NH}_3$  (in red), while values below 1 indicate a reduction (in blue). The results are averaged over the entire period from 2003 to 2019.

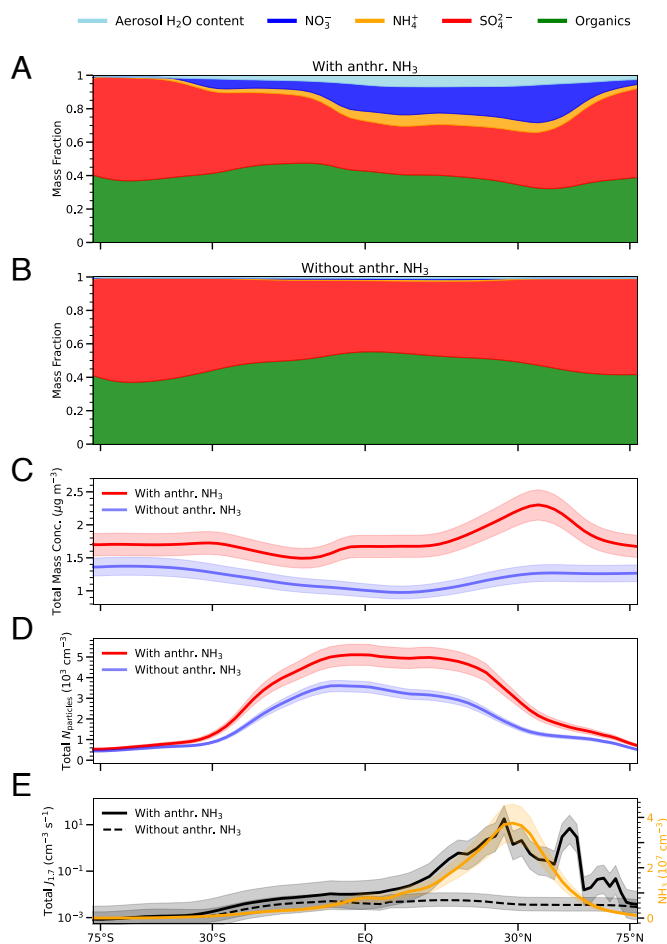
**Influence on UTLS Aerosol Composition and Abundance.** Understanding the impact of anthropogenic  $\text{NH}_3$  on aerosol composition and abundance in the UTLS is important for assessing its influence on climate (34, 40). The coincidence of latitudinal profiles of  $\text{NH}_3$  concentration, total  $J_{1.7}$ , aerosol mass fraction, total aerosol mass, and number concentrations ( $N_{\text{particles}}$ ) in the UTLS can be seen in Fig. 2.

The influence of anthropogenic  $\text{NH}_3$  emissions on the average mass fractions of key aerosol components—sulfate ( $\text{SO}_4^{2-}$ ), nitrate ( $\text{NO}_3^-$ ), ammonium ( $\text{NH}_4^+$ ), water ( $\text{H}_2\text{O}$ ), and organics is shown in Fig. 2A and B. Over emission hotspots,  $\text{NH}_3$  markedly increases  $\text{NO}_3^-$  and  $\text{NH}_4^+$  aerosol mass fractions in the UTLS, while its absence drastically reduces  $\text{NO}_3^-$  and nearly eliminates  $\text{NH}_4^+$  and aerosol  $\text{H}_2\text{O}$  content.  $\text{NH}_4\text{NO}_3$  evaporates at the higher temperatures of the lower troposphere but remains entirely in the particulate phase in the UTLS (40). Below roughly 233 K, it is essentially impossible for  $\text{NH}_3$  and  $\text{HNO}_3$  to coexist in the gas phase, and the formation of  $\text{NH}_4\text{NO}_3$  is

practically irreversible (54). We find that, on average, organics constitute 40% of the simulated UTLS aerosol mass. Secondary organic aerosols from volatile organic compound oxidation account for 90% of this organic mass, while primary organic aerosols from biomass and fossil fuel combustion account for the remaining 10%.

As shown in Fig. 2C, the total simulated aerosol mass concentration is increased in the UTLS when anthropogenic  $\text{NH}_3$  is present, with the largest increase observed from 0 to 40°N. This increase in mass concentration due to  $\text{NH}_4\text{NO}_3$  coincides with the increase in  $\text{NH}_3$  concentration. In addition,  $N_{\text{particles}}$  (Fig. 2D) are increased by up to 2,000  $\text{cm}^{-3}$  between 30°S and 30°N with anthropogenic  $\text{NH}_3$  relative to the scenario without anthropogenic  $\text{NH}_3$  emissions. The presence of  $\text{NH}_3$  enhances  $J_{1.7}$  (Fig. 2E) in the tropics and midlatitudes by up to three orders of magnitude. The alignment of the peak  $J_{1.7}$  and aerosol mass concentrations with high  $\text{NH}_3$  concentration is evidence of its importance in both NPF and the growth of particles in the UTLS.





**Fig. 2.** Simulated latitudinal profiles of UTLS aerosol components by mass (aerosol H<sub>2</sub>O content, NO<sub>3</sub><sup>−</sup>, NH<sub>4</sub><sup>+</sup>, SO<sub>4</sub><sup>2−</sup>, and organics) (A) with, and (B) without anthropogenic NH<sub>3</sub> emissions. Latitudinal profiles of simulated total particle (C) mass and (D) number concentrations ( $N_{\text{particles}}$ ) in the UTLS with (red line) and without (blue line) anthropogenic NH<sub>3</sub>. Particle size ranges from 2 nm to 1,000 nm. (E) Simulated latitudinal profiles of the total UTLS nucleation rate at 1.7 nm diameter ( $J_{1.7}$ ) with (solid black line) and without (dashed black line) anthropogenic NH<sub>3</sub>. The UTLS concentrations of anthropogenic NH<sub>3</sub> are shown in orange. All concentrations and  $J_{1.7}$  are calculated at ambient temperature and pressure. The lines and the shaded areas represent the mean and the interannual variability (temporal SD) for the corresponding latitude, respectively. Modeled outputs are averaged within the 10 to 15 km altitude range over the entire period from 2003 to 2019.

**Effects on AOD.** The influence of anthropogenic NH<sub>3</sub> emissions on the total AOD at 550 nm and the different UTLS aerosol components of the AOD for each case is presented in Fig. 3. The individual aerosol components considered include black and organic carbon, dust, H<sub>2</sub>O-soluble inorganic ions, and the aerosol H<sub>2</sub>O content.

In Fig. 3A and B, we compare the changes in the UTLS AOD components in the presence and absence of anthropogenic NH<sub>3</sub>, respectively. The contribution of the H<sub>2</sub>O-soluble inorganic ions to the total UTLS AOD (550 nm) increases and reaches an average 40 to 50% in the presence of anthropogenic NH<sub>3</sub>. This increase is especially pronounced over regions with intense agricultural and industrial activities, such as South and East Asia. The peaks in the AOD enhancement ratio due to anthropogenic NH<sub>3</sub> align with enhanced contributions from H<sub>2</sub>O-soluble inorganic ions. On average, this peak enhancement ratio varies from 20 to 40% over high-emission regions. As a strong base, NH<sub>3</sub> facilitates the partitioning of HNO<sub>3</sub> and H<sub>2</sub>O into the

aerosol phase, leading to increased formation of H<sub>2</sub>O-soluble compounds (such as NH<sub>4</sub>NO<sub>3</sub>), which in turn enhances AOD (55). In the absence of anthropogenic NH<sub>3</sub>, the contribution of the H<sub>2</sub>O-soluble inorganic ions to the total AOD is reduced on average by 20% in the UTLS.

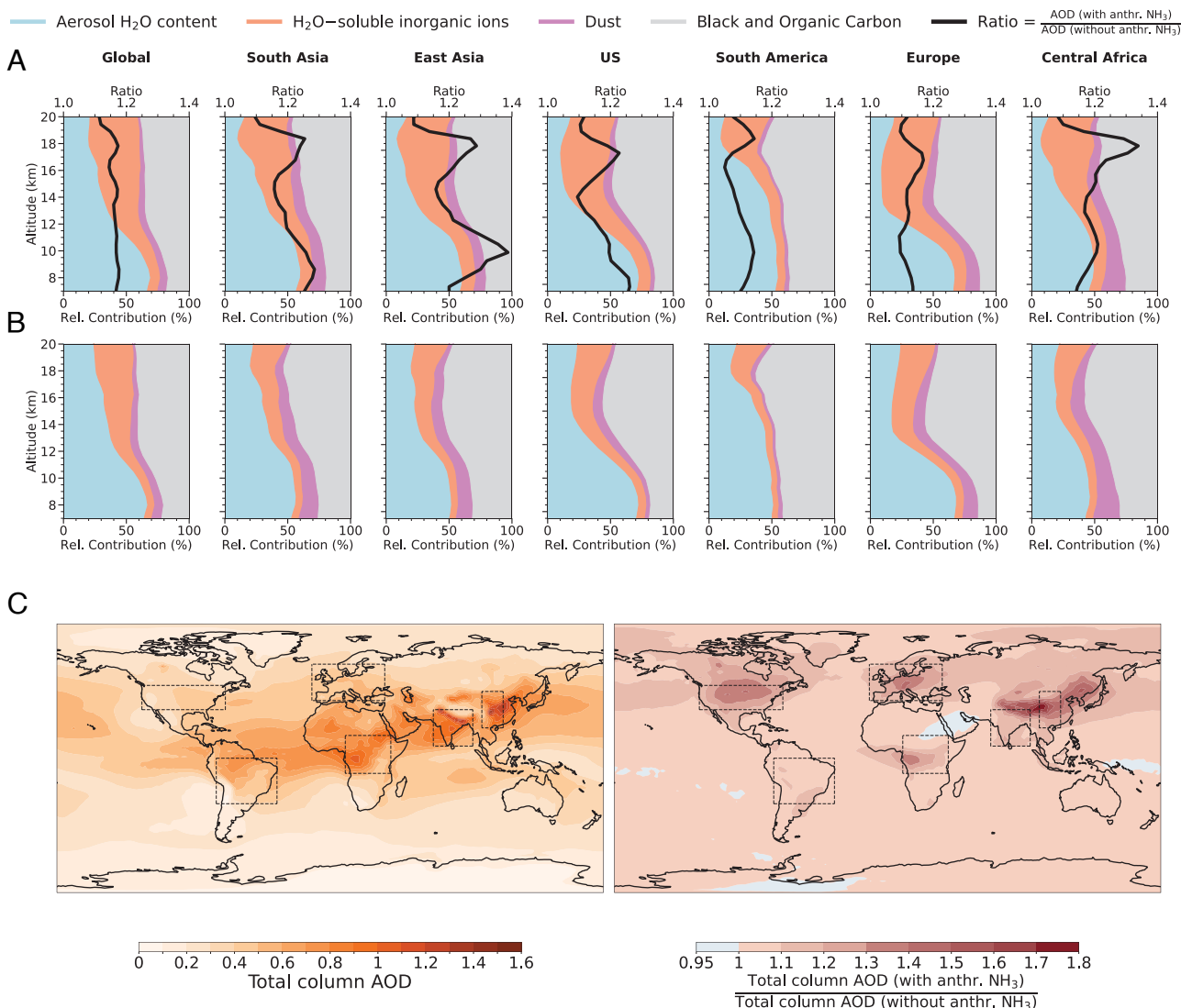
The total atmospheric column AOD (from the surface to the top of the atmosphere) at 550 nm exhibits a global increase in response to anthropogenic NH<sub>3</sub> emissions (Fig. 3C). The strongest enhancements of AOD are observed over major NH<sub>3</sub> source regions. The increase in AOD can reach up to 60 to 80% in the presence of anthropogenic NH<sub>3</sub> in these regions compared to its absence. This enhancement results from NH<sub>3</sub>-driven aerosol formation in the UTLS and lower altitudes, indicating a significant influence on aerosol optical properties with potential implications for atmospheric radiative transfer and climate.

**Global Impact of Anthropogenic NH<sub>3</sub> on UT CCN.** We model the global impact of anthropogenic NH<sub>3</sub> emissions on CCN concentrations at 0.4% supersaturation (CCN<sub>0.4%</sub>) in the UT. Our simulations show that particle number concentrations ( $N_{\text{particles}}$ ) in the UTLS (Fig. 2D) exhibit a strong response to anthropogenic NH<sub>3</sub> emissions, especially in the Northern Hemisphere, where these emissions are most concentrated. This increase in  $N_{\text{particles}}$  can contribute to the total CCN<sub>0.4%</sub> in the UT, with potential implications for cloud formation (7). As these particles continue to grow, they may be transported downward with descending air masses. Upon reaching lower altitudes, they can further contribute to cloud formation in the lower troposphere (7).

In the UT, the CCN<sub>0.4%</sub> concentration is substantially enhanced by anthropogenic NH<sub>3</sub> (Fig. 4B). This enhancement results from the doubling of UTLS aerosol abundance and is most pronounced over regions with strong agricultural and industrial emissions, where CCN<sub>0.4%</sub> concentrations exceed 200 cm<sup>−3</sup>—up to a 2.5-fold increase. In addition, over the oceans, downwind of NH<sub>3</sub> emissions and in the absence of local anthropogenic sources, CCN<sub>0.4%</sub> concentrations increase substantially. A significant portion of cirrus ice crystals can originate from cloud droplets that first condense onto CCN before freezing (56–58). Changes in CCN concentrations at the altitudes of convective anvils, contrails, or cirrus clouds influence ice crystal formation and concentrations (56, 59) that can significantly affect climate by altering the radiative properties of cirrus clouds (60, 61).

**Sensitivity Analysis.** Anthropogenic emissions in our simulations are derived from the Community Emissions Data System (CEDS) (62). Uncertainties in CEDS arise from inaccuracies in activity data (e.g., energy consumption), incomplete or outdated emission factors, and limited information on the enforcement of control technologies, particularly in emerging economies. These uncertainties vary across different compounds, sectors, regions, and time periods. For NH<sub>3</sub>, the main source of uncertainty is the variability in agricultural practices and emission reporting.

To assess the sensitivity of our results to this uncertainty, we performed additional simulations in which global NH<sub>3</sub> emissions were perturbed by  $\pm 15\%$ . This range was based on the global normalized mean bias (NMB) between EMAC-simulated and IASI-retrieved NH<sub>3</sub> concentrations, which is approximately 15% (Methods and SI Appendix). We find that perturbing global NH<sub>3</sub> emissions by  $\pm 15\%$  leads to regional variations in UT  $J_{1.7}$ , CCN<sub>0.4%</sub>, and AOD within 20% in spatial distributions and vertical profiles (SI Appendix, Figs. S1–S3).



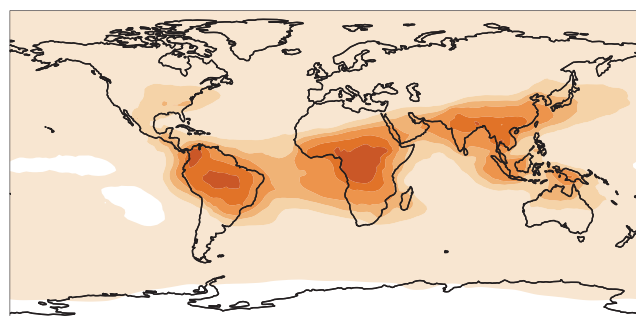
**Fig. 3.** Vertical distribution of the contribution of different UTLS AOD components to the total AOD over NH<sub>3</sub> emission hotspots (A) with and (B) without anthropogenic NH<sub>3</sub>. The relative contributions of aerosol components (H<sub>2</sub>O, H<sub>2</sub>O-soluble inorganic ions, dust, black and organic carbon) to the total AOD (550 nm) are shown as stacked colored regions. The black line represents the ratio of AOD with anthropogenic NH<sub>3</sub> to AOD without, highlighting the relative enhancement due to NH<sub>3</sub> emissions at each altitude. The tropopause altitude range for each region is specified in Fig. 1A. (C) Total atmospheric column AOD at 550 nm in the presence of anthropogenic NH<sub>3</sub> (Left). AOD total column enhancement ratio is defined as the ratio of total column AOD with anthropogenic NH<sub>3</sub> to without (Right). Values greater than 1 indicate an increase in AOD due to NH<sub>3</sub> (in red), while values below 1 indicate a decrease (in blue). NH<sub>3</sub> emission hotspots are marked on the map. The results are calculated at ambient conditions and averaged over the entire period from 2003 to 2019.

In South and East Asia, the NMB is 27% and 50%, respectively, and up to 200% at certain altitudes. To quantify the influence of this uncertainty on our results, we carried out region- and season-specific perturbation sensitivity simulations in which NH<sub>3</sub> emissions were scaled based on the EMAC-to-IASI ratio in each hotspot region for each month. During the summer monsoon season in South and East Asia, variations in UT  $J_{1.7}$ , CCN<sub>0.4%</sub>, and AOD due to NH<sub>3</sub> emission perturbations can reach up to 30%. Elevated uncertainties are also observed during spring, while winter and autumn show minimal sensitivity across all hotspot regions (SI Appendix, Figs. S1 and S2). Since AIRS shows better agreement with EMAC than IASI, these seasonal and regional uncertainties should be regarded as conservative estimates, and the actual uncertainties may be smaller.

To examine the sensitivity of UTLS particle formation to long-term emission trends, we analyzed time series of anthropogenic

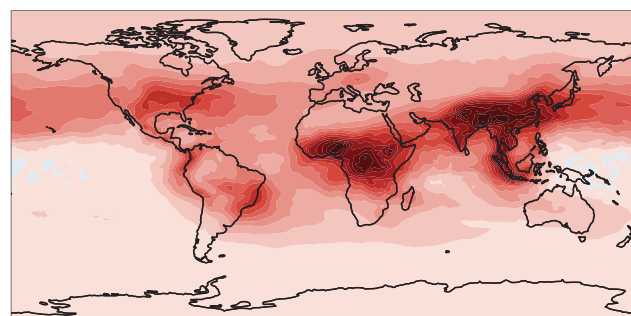
emissions (from CEDS) and total UTLS  $J_{1.7}$  over NH<sub>3</sub> hotspots (SI Appendix, Fig. S4). Individual species emissions in certain regions exhibit temporal trends, with increasing NH<sub>3</sub> fluxes in South Asia, and declining SO<sub>2</sub> and NO<sub>x</sub> in Europe and the United States due to emission controls. The response to these trends of  $J_{1.7}$  is influenced by a complex interplay of the subsequent reactions producing multiple nucleation precursors (e.g. oxidation of SO<sub>2</sub> to H<sub>2</sub>SO<sub>4</sub>), the presence of natural emissions (such as lightning NO<sub>x</sub>), and transport dynamics, in particular convective updrafts. The variability in  $J_{1.7}$  is thus not solely modulated by the trend of individual species emissions over each region, and without a consistent long-term trend across all regions. The presence of deep convective clouds substantially enhances  $J_{1.7}$  values across all regions; thus it is expected that, other than emission trends, meteorological factors such as monsoon duration and intensity, play an important role in year-to-year temporal variability.

## With anthr. NH<sub>3</sub>



25 50 75 100 150 200 250 500  
UT CCN<sub>0.4%</sub> (cm<sup>-3</sup>)

## Effect of anthr. NH<sub>3</sub> relative to without



1 1.2 1.4 1.6 1.8 2 2.2 2.4  
UT CCN<sub>0.4%</sub> (with anthr. NH<sub>3</sub>)  
UT CCN<sub>0.4%</sub> (without anthr. NH<sub>3</sub>)

**Fig. 4.** Global maps of cloud condensation nuclei concentrations at 0.4% supersaturation (CCN<sub>0.4%</sub>) in the UT, averaged within the 10 to 13 km altitude range. The *Left* panel illustrates the absolute UT CCN<sub>0.4%</sub> values in the presence of anthropogenic NH<sub>3</sub>. The *Right* panel shows the enhancement ratio, defined as the ratio of UT CCN<sub>0.4%</sub> values with anthropogenic NH<sub>3</sub> to those without. Values greater than 1 indicate an increase due to NH<sub>3</sub> (in red), while values below 1 indicate a reduction (in blue). The results are calculated at ambient conditions and averaged over the entire period from 2003 to 2019.

## Conclusions

This study underscores the critical role of anthropogenic NH<sub>3</sub> emissions in driving particle formation in the UTLS. By incorporating recently published NPF parameterizations derived from the CERN CLOUD experiment into the EMAC model, we quantified the impact of anthropogenic NH<sub>3</sub> emissions on UTLS particle formation by comparing a simulation with anthropogenic NH<sub>3</sub> emissions to a baseline simulation without anthropogenic NH<sub>3</sub>.

Our simulations show that convective transport of anthropogenic NH<sub>3</sub> enhances UTLS NPF rates by one to three orders of magnitude over NH<sub>3</sub> hotspots. This NH<sub>3</sub>-driven particle formation and growth leads to a doubling of UTLS particle numbers over emission hotspots compared to a scenario without anthropogenic NH<sub>3</sub>. These particles can contribute to a 2.5-fold enhancement of UT CCN concentrations over high-emission regions, and smaller but significant changes over the oceans. Eliminating anthropogenic NH<sub>3</sub> would nearly deplete the UTLS aerosol NH<sub>4</sub><sup>+</sup> and H<sub>2</sub>O content, and substantially reduce NO<sub>3</sub><sup>-</sup> levels. These chemical composition changes lead to a global reduction in the total UTLS aerosol mass concentration, reaching up to 50% over Asia. This significant decrease in global aerosol loading also translates to a lower AOD without NH<sub>3</sub>, up to 80% in the Northern Hemisphere and up to 10% in the Southern Hemisphere. These findings highlight the role of anthropogenic NH<sub>3</sub> in affecting UTLS aerosol abundance and composition, cloud formation, and aerosol optical properties, with potential implications for climate.

Our study opens future research directions by highlighting the need to integrate UTLS NH<sub>3</sub>-driven aerosol formation into Earth system models to improve the predictive accuracy of atmospheric composition and cloud effects in climate scenarios. While this study focuses on global CCN and AOD, further research should quantify the radiative forcing of UTLS NH<sub>3</sub>-driven aerosol changes. In addition, anthropogenic NH<sub>3</sub> may influence ice-nucleating particle activity, with potential consequences for cloud glaciation and mixed-phase cloud processes (63). A deeper understanding of these processes is essential for assessing the role of NH<sub>3</sub> in aerosol–cloud–climate interactions. In particular,

more targeted observations and improved emission estimates over South and East Asia during the summer monsoon season are needed to better constrain the role of deep convection in NH<sub>3</sub> transport and UTLS aerosol formation.

Current policies primarily target NH<sub>3</sub> emissions to address air quality concerns (64), with recent directives encouraging additional NH<sub>3</sub> measurements (65). This will provide more data for further analysis that connects air quality and climate. However, these policies often overlook the indirect climate implications of NH<sub>3</sub>-driven aerosol formation in the UTLS. While reducing agricultural NH<sub>3</sub> emissions can improve air quality, our findings indicate that it also leads to a depletion of the UTLS aerosol layer, which can have significant implications for the Earth's radiation balance. Incorporating our findings into IPCC (The Intergovernmental Panel on Climate Change) socio-economic models can help refine projections of aerosol–cloud–climate interactions and guide policies that balance agricultural productivity with climate goals, contributing to the objectives of the UNFCCC Paris Agreement and promoting environmental and societal sustainability (66, 67).

## Methods

**EMAC Model.** The EMAC (ECHAM/MESSy Atmospheric Chemistry) model is a numerical framework designed to simulate global atmospheric chemistry and climate interactions. It integrates various submodels that represent atmospheric processes and their interactions with land, oceans, and anthropogenic influences (68). The dynamical core is the ECHAM5 atmospheric circulation model, which is linked with the second version of the Modular Earth Submodel System (MESSy2) to facilitate the integration of multi-institutional computer codes (49). To ensure realistic atmospheric transport conditions, meteorological prognostic variables are nudged toward ECMWF ERA-5 reanalysis data through Newtonian relaxation. Atmospheric chemical kinetics are computed online at each model time step using the MIM chemistry mechanism (69). EMAC follows the 1957 World Meteorological Organization definition of the tropopause, which is the lowest altitude at which the temperature lapse rate decreases to 2 K km<sup>-1</sup> or less, provided that the average lapse rate within the next 2 km does not exceed this threshold (70).

For this study, we use EMAC (ECHAM5 version 5.3.02, MESSy version 2.55.2) at a T63L90 resolution to simulate the period from January 2003 to



December 2019. The simulation period allows for direct comparison with the satellite retrievals by the MIPAS and IASI instruments. The model employs 90 vertical hybrid levels extending from the surface to approximately 80 km altitude (0.01 hPa) and applies a spherical truncation of T63, corresponding to a horizontal grid resolution of  $1.875^\circ \times 1.875^\circ$  at the equator. A 10-min time step is used, with model output at hourly intervals. All conditions are identical in both simulations, with the only difference being that anthropogenic  $\text{NH}_3$  emissions are turned off in one of them.

The simulation code encompasses multiple submodels: i) GMXe that handles aerosol microphysics (71), ii) NAN that is responsible for nucleation mechanisms (72), iii) IONS that simulates ion pair production due to cosmic rays and radon decay (72), iv) AEROPT that calculates aerosol optical properties (73), v) MECCA, dedicated to gas-phase chemistry (74), vi) JVAL for photochemistry rates (75), and vii) DRYDEP, which simulates dry deposition using the big-leaf approach (76, 77). Dry deposition of  $\text{NH}_3$  introduces additional uncertainty due to the complexities of bidirectional exchange. Sensitivity simulations were carried out to account for the uncertainties in  $\text{NH}_3$  budget (SI Appendix, Figs. S1–S3).

In this study, the CONVECT submodel is employed to parameterize convection using the Tiedtke scheme (78) with Nordeng closure (79), a standard configuration for T63 resolution (80). The LNOX submodel calculates real-time  $\text{NO}_x$  emissions from lightning activity (81), applying the parameterization developed by Grewe et al. (82), which correlates flash frequency with updraft velocity.

Our model computes the molality of semivolatile compounds and the equilibrium states of binary solutions, considering both stable and metastable phases (83).  $\text{HNO}_3$  is produced by the oxidation of lightning  $\text{NO}_x$ . In the UT, transport processes primarily influence production rather than local reactions (52).  $\text{H}_2\text{SO}_4$  is mainly produced through the oxidation of sulfur dioxide (49).

In Fig. 5,  $\text{NH}_3$  distributions across space, time, and altitude generated by the EMAC model are compared against satellite measurements, with a focus on the UTLS. The MIPAS (Michelson Interferometer for Passive Atmospheric Sounding) instrument aboard the ENVISAT satellite retrieves vertical profiles of  $\text{NH}_3$  by detecting its infrared emissions. MIPAS is a high-resolution mid-infrared emission limb sounder designed to measure  $\text{NH}_3$  volume mixing ratios, along with a broad range of other atmospheric trace species (84). We construct synthetic averaging kernels to account for vertical resolution effects when averaging model data for comparison with MIPAS retrieved values. We use the vertical resolution, defined as the altitude grid spacing over the averaging kernel diagonal, to generate Gaussian-shaped weighting functions at different altitudes. Each Gaussian function is centered at a given altitude, with its half-width determined by the corresponding vertical resolution value. The model profile is then convolved with these Gaussian functions, effectively smoothing it to match the vertical sensitivity of the retrieval. We also compare total column concentrations against retrievals by the IASI (Infrared Atmospheric Sounding Interferometer) instrument aboard the MetOp satellites that measures  $\text{NH}_3$  total column concentrations globally using infrared spectroscopy (85–88).

Fig. 5A compares the spatial distribution of global  $\text{NH}_3$  total column concentrations from EMAC simulations and IASI observations, averaged over 2017–2019. The comparison identifies major hotspots in South and East Asia, Central Africa, Europe, South America, and the United States. Fig. 5B illustrates monthly averages of total column  $\text{NH}_3$  from EMAC and IASI. While the model overestimates  $\text{NH}_3$  concentrations over South and East Asia, the normalized mean biases are within  $\pm 10\%$  in all other hotspot regions. Both globally and across all hotspot regions, Pearson correlation coefficients range from 0.6 to 0.85, and 60 to 91% of simulated data fall within a factor of two of the observations (SI Appendix, Table S1). Fig. 5C compares the vertical profiles of EMAC kernel-adjusted  $\text{NH}_3$  median profiles and the equivalent MIPAS retrievals for 2003–2011 in the UTLS over hotspot regions. Simulated values broadly encompass the observations and their associated uncertainties.

The model has also been evaluated against AIRS (Atmospheric Infrared Sounder)  $\text{NH}_3$  global measurements (89–91) from the Aqua satellite within the 4 to 5.5 km altitude range. EMAC reproduces the observed concentrations with normalized mean biases within  $\pm 8\%$ , Pearson correlation coefficients above 0.7 across all regions, and more than 70% of simulated data falling within a factor of two of the observations (SI Appendix, Table S1 and Fig. S5).

Simulated  $\text{NH}_4^+$  concentrations were also compared with aircraft observations from the NASA ATom (Atmospheric Tomography) mission (92), covering

continental North America, the Pacific, Atlantic, and Southern Ocean (SI Appendix, Fig. S6). The model captures the observed vertical distributions of  $\text{NH}_4^+$  across ATom campaigns, with simulated concentrations within the observed interquartile range (SI Appendix, Fig. S7).

Finally, the simulated AOD and CCN values were compared to MODIS (MODerate Resolution Imaging Spectroradiometer) (93) and E3SM (Energy Exascale Earth System Model) (10), respectively, with consistent regional patterns between models (SI Appendix, Figs. S8 and S9). The model has also been evaluated extensively over the Asian monsoon region (27).

**CEDS Inventory.** Trace gas emissions, including  $\text{NH}_3$ , are adopted from the CEDS inventory (62). CEDS includes default global emission estimates for each compound, based on a bottom-up approach. These estimates are scaled to match global inventories by sector and fuel type and subsequently refined using regional or national inventories that incorporate more accurate local data. CEDS involves normalized spatial distribution proxies to scale yearly country-level emission estimates onto a global  $0.5^\circ \times 0.5^\circ$  grid. These gridded emission fluxes are aggregated into sectors and distributed over a 12-mo period. The spatial distribution and magnitude of the simulated  $\text{NH}_3$  emissions are shown in SI Appendix, Fig. S10, and the sectoral contributions are illustrated in SI Appendix, Fig. S11.

Compared to other inventories, such as EDGAR and ECLIPSE, CEDS adopts a mosaic approach that integrates activity and emission input data from multiple sources—including EDGAR, ECLIPSE, and detailed regional and national inventories, such as SMOG-India. This approach reduces uncertainty and produces global emission estimates that are historically consistent and aligned with contemporary country-level data from 1970 to 2017, particularly in regions undergoing rapid energy and policy shifts (62, 94, 95).

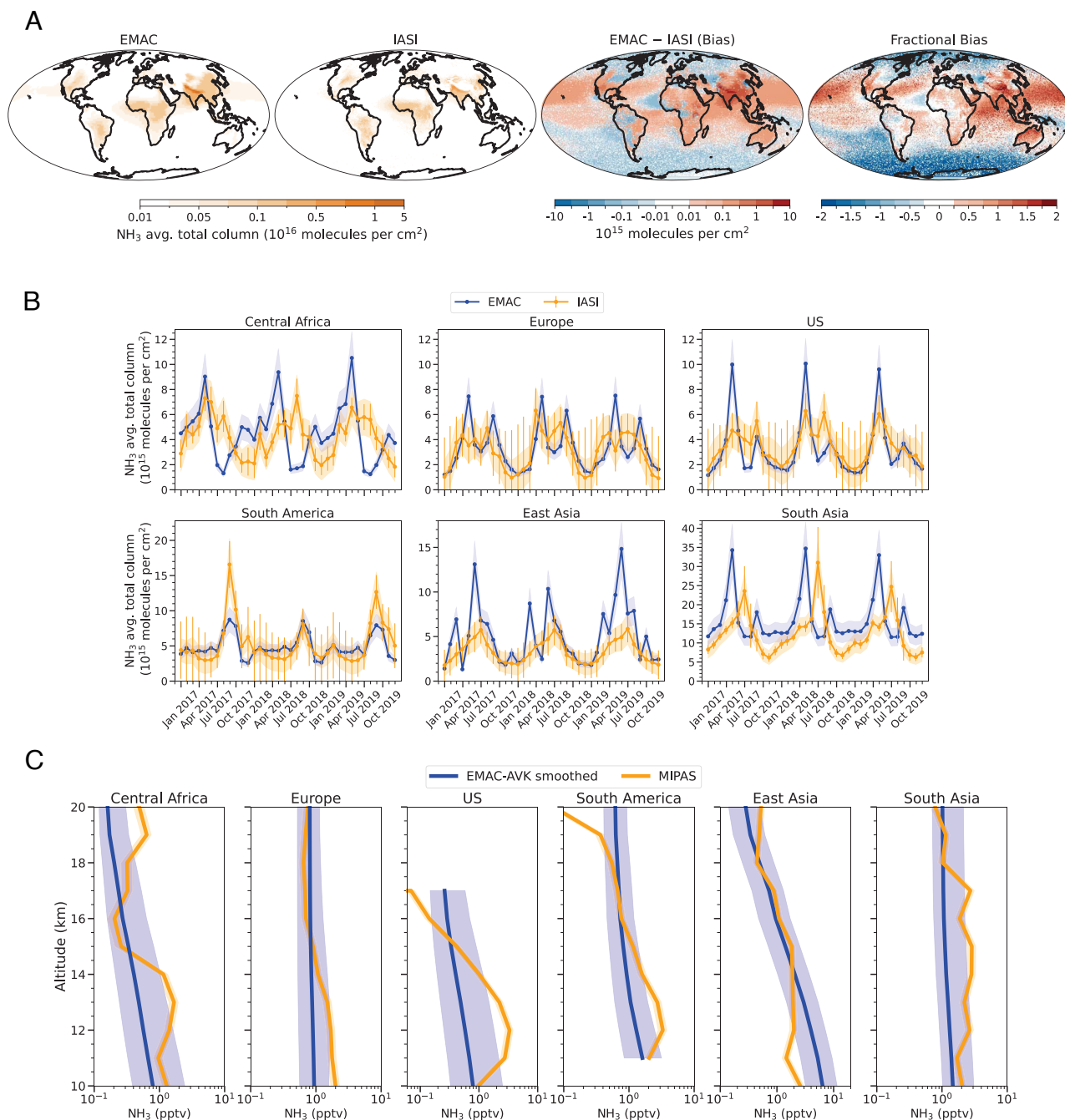
**NPF and Aerosol Submodels.** The EMAC model incorporates the NAN (New Aerosol Nucleation) submodel (72) to represent NPF. NAN estimates nucleation rates using parameterizations derived from the CERN CLOUD experiment, specifically, neutral and ion-induced binary nucleation involving  $\text{H}_2\text{SO}_4$  and  $\text{H}_2\text{O}$  (47) and ternary nucleation of  $\text{NH}_3$ – $\text{H}_2\text{SO}_4$ – $\text{H}_2\text{O}$  (47), and a synergistic nucleation mechanism involving  $\text{NH}_3$ – $\text{H}_2\text{SO}_4$ – $\text{HNO}_3$ – $\text{H}_2\text{O}$  (35). The integration of these NPF parameterizations within EMAC is described in detail by Ehrhart et al. (72). Details on the implementation of the most recently published CLOUD parameterization for synergistic nucleation (35) are provided by Xenofontos et al. (27).

The IONS submodel calculates atmospheric ion pair production rates and steady-state ion concentrations, considering galactic cosmic rays and radon decay. It provides real-time ion pair production rates for ion-induced nucleation while accounting for losses through ion-ion recombination and aerosol scavenging. The NAN and IONS submodels have been evaluated by Ehrhart et al. (72).

The GMXe (Global Modal-aerosol eXtension) submodel (71) simulates aerosol dynamics through a full thermodynamic treatment of gas/aerosol partitioning using the ISORROPIA-II model (83). GMXe represents the aerosol size distribution as seven log-normal modes—four hydrophilic and three hydrophobic. Aerosol number concentration and mass for each component are prognostically calculated, assuming a constant geometric SD for each mode. The aerosol size distribution follows:

$$n(\ln r) = \sum_{i=1}^7 \frac{N_i}{\sqrt{2\pi} \ln \sigma_i} \exp \left( -\frac{(\ln r - \ln \tilde{r}_i)^2}{2 \ln^2 \sigma_i} \right),$$

where each mode ( $i$ ) is characterized by its number concentration ( $N_i$ ), number median radius ( $\tilde{r}_i$ ), and geometric SD ( $\sigma_i$ ). The four hydrophilic modes span the full aerosol size range: i) nucleation ( $< 10$  nm), ii) Aitken (10–100 nm), iii) accumulation (100 to 1,000 nm), and iv) coarse ( $> 1,000$  nm). The three hydrophobic modes mirror these size ranges, encompassing Aitken, accumulation, and coarse modes (71). In our simulations,  $\sigma = 1.59$  for the nucleation mode and both (hydrophilic and hydrophobic) Aitken modes,  $\sigma = 1.49$  for both accumulation modes, and  $\sigma = 1.7$  for both coarse modes. Composition is uniform (internal mixing) within modes, but different modes exhibit compositional variation (external mixing).



**Fig. 5.** (A) Spatial distribution of  $\text{NH}_3$  total column concentrations (0 to 20 km) from EMAC simulations compared to IASI satellite retrievals, averaged over January 2017–December 2019. (B) Temporal validation of monthly  $\text{NH}_3$  column averages at key emission hotspot regions, showing consistency between EMAC and IASI observations from January 2017–December 2019. The light blue shaded areas represent the temporal SD in EMAC. The orange vertical lines and shaded areas represent random and systematic uncertainties in IASI observations, respectively. (C) Vertical  $\text{NH}_3$  profiles from EMAC simulations, smoothed using averaging kernels, compared against MIPAS retrievals. The blue lines and shaded areas represent the medians and the interquartile ranges for 2003–2011 in EMAC, respectively. The orange lines and shaded areas represent the median and the uncertainty in MIPAS measurements, respectively. According to ref. 40, the uncertainty in MIPAS data is estimated as  $\pm 5$  pptv (single profile precision) and  $\pm 15\%$  (accuracy).

Coagulation in GMx follows the method in ref. 96, with coagulation coefficients derived from Brownian motion theory based on the original work of Fuchs (97). The coagulation matrix accommodates varying species numbers within each mode. Coagulation processes transfer aerosol particles from smaller to larger modes and hydrophobic to hydrophilic modes (71).

**Data, Materials, and Software Availability.** The EMAC (ECHAM/MESSy Atmospheric Chemistry) model is continuously developed and used by a consortium of institutions. Members of institutions within the MESSy

consortium are granted a license to use MESSy and access its source code. Institutions can join the consortium by signing the MESSy Memorandum of Understanding. Further information is available on the MESSy consortium website (<https://www.messy-interface.org>) (98). The results presented in this paper were produced using MESSy version 2.55.2. A permanent identifier (DOI: [10.5281/zenodo.14875637](https://doi.org/10.5281/zenodo.14875637)) (99) has been assigned in Zenodo under the "CERN CLOUD experiment community." This includes the EMAC configuration files, namelist setup, chemical mechanisms, and details on the emissions setup. Additionally, the complete dataset used in the figures will be made available



upon finalization for publication to ensure long-term accessibility and facilitate reproducibility.

**ACKNOWLEDGMENTS.** We express our appreciation to the European Organization for Nuclear Research (CERN) for providing Cosmics Leaving Outdoor Droplets (CLOUD) with important financial and technical resources. Funding: This work was supported by the European Union (EU)'s Horizon 2020 research and innovation programme under grant agreement no. 856612 (EMME-CARE); EU Marie Skłodowska-Curie Actions Doctoral Network CLOUD-DOC 101073026; Atmosphere and Climate Competence Center Flagship funded by the Academy of Finland grant nos. 337549 (University of Helsinki), 337552 (Finnish Meteorological Institute), and 337550 (University of Eastern Finland); Academy professorship funded by the Academy of Finland (grant no. 302958); Academy of Finland projects no. 1325656, 311932, 334792, 316114, 325647, 325681, 347782, and 346371, "Quantifying carbon sink, CarbonSink+ and their interaction with air quality;" Academy of Finland funding no. 359331; Institute for Atmospheric and Earth System Research project funded by Jane and Aatos Erkkö Foundation; "Gigacity" project funded by Wihuri foundation; European Research Council project ATM-GTP Contract No. 742206; Research Council of Finland project no. 349659; German Federal Ministry of Education and Research project CLOUD-22 (01LK2201A, 01LK2201B, and 01LK2201C); Horizon Europe project FOCL (no. 101056783); Swiss NSF (SNF): 200021\_213071; Vienna Science and Technology Fund (WWTF) through project VRG22-003; US NSF Grant ATM-2215527; US NSF awards AGS-2215522, 2027252, 2215489, 2431817, and 2132089; H.G. acknowledges funding from NASA under grant 80NSSC19K0949; P.M.W. acknowledges funding from the Austrian National Bank, donation 2023-5223.

1. F. Yu, Effect of ammonia on new particle formation: A kinetic  $\text{H}_2\text{SO}_4 - \text{H}_2\text{O} - \text{NH}_3$  nucleation model constrained by laboratory measurements. *J. Geophys. Res. Atmos.* **111**, e2005JD005968 (2006).
2. M. Wang *et al.*, Rapid growth of new atmospheric particles by nitric acid and ammonia condensation. *Nature* **581**, 184–189 (2020).
3. D. Stolzenburg *et al.*, Atmospheric nanoparticle growth. *Rev. Mod. Phys.* **95**, 045002 (2023).
4. W. W. Kellogg, *Aerosols and Climate*, W. Bach, J. Pankrath, J. Williams, Eds. (Springer Netherlands, Dordrecht, 1980), pp. 281–303.
5. N. Bellouin, O. Boucher, J. Haywood, M. S. Reddy, Global estimate of aerosol direct radiative forcing from satellite measurements. *Nature* **438**, 1138–1141 (2005).
6. J. R. Pierce, Cosmic rays, aerosols, clouds, and climate: Recent findings from the CLOUD experiment. *J. Geophys. Res. Atmos.* **122**, 8051–8055 (2017).
7. C. J. Williamson *et al.*, A large source of cloud condensation nuclei from new particle formation in the tropics. *Nature* **574**, 399–403 (2019).
8. C. Rose *et al.*, CCN production by new particle formation in the free troposphere. *Atmos. Chem. Phys.* **17**, 1529–1541 (2017).
9. H. Gordon *et al.*, Causes and importance of new particle formation in the present-day and preindustrial atmospheres. *J. Geophys. Res. Atmos.* **122**, 8739–8760 (2017).
10. B. Zhao *et al.*, Global variability in atmospheric new particle formation mechanisms. *Nature* **631**, 98–105 (2024).
11. N. Anderson, R. Strader, C. Davidson, Airborne reduced nitrogen: Ammonia emissions from agriculture and other sources. *Environ. Int.* **29**, 277–286 (2003).
12. A. A. Swelum *et al.*, Ammonia emissions in poultry houses and microbial nitrification as a promising reduction strategy. *Sci. Total. Environ.* **781**, 146978 (2021).
13. J. M. Paillat, P. Robin, M. Hassouna, P. Leterme, Predicting ammonia and carbon dioxide emissions from carbon and nitrogen biodegradability during animal waste composting. *Atmos. Environ.* **39**, 6833–6842 (2005).
14. S. Krupa, Effects of atmospheric ammonia ( $\text{NH}_3$ ) on terrestrial vegetation: A review. *Environ. Pollut.* **124**, 179–221 (2003).
15. N. Anderson, R. Strader, C. Davidson, Airborne reduced nitrogen: Ammonia emissions from agriculture and other sources. *Environ. Int.* **29**, 277–286 (2003).
16. M. Hassouna *et al.*, *Measuring Emissions from Livestock Farming: Greenhouse Gases, Ammonia and Nitrogen Oxides* (INRA-ADEME, 2016).
17. D. P. Van Vuuren *et al.*, The representative concentration pathways: An overview. *Clim. Change* **109**, 5–31 (2011).
18. M. Van Damme *et al.*, Industrial and agricultural ammonia point sources exposed. *Nature* **564**, 99–103 (2018).
19. W. W. Walters *et al.*, Quantifying the importance of vehicle ammonia emissions in an urban area of northeastern USA utilizing nitrogen isotopes. *Atmos. Chem. Phys.* **22**, 13431–13448 (2022).
20. A. Hegg, L. F. Radke, P. V. Hobbs, P. J. Riggan, Ammonia emissions from biomass burning. *Geophys. Res. Lett.* **15**, 335–337 (1988).
21. J. Webb *et al.*, Managing ammonia emissions from livestock production in Europe. *Environ. Pollut.* **135**, 399–406 (2005).
22. C. L. Heald *et al.*, Atmospheric ammonia and particulate inorganic nitrogen over the United States. *Atmos. Chem. Phys.* **12**, 10295–10312 (2012).
23. J. Kuttippurath *et al.*, Record high levels of atmospheric ammonia over India: Spatial and temporal analyses. *Sci. Total. Environ.* **740**, 139986 (2020).

Author affiliations: <sup>a</sup>Climate and Atmosphere Research Center, The Cyprus Institute, Nicosia 1645, Cyprus; <sup>b</sup>Department of Atmospheric Chemistry, Max Planck Institute for Chemistry, Mainz 55128, Germany; <sup>c</sup>The European Organization for Nuclear Research (CERN), Geneva 1211, Switzerland; <sup>d</sup>Institute for Atmospheric and Environmental Sciences, Goethe University Frankfurt, Frankfurt am Main 60438, Germany; <sup>e</sup>Cooperative Institute for Research in Environmental Sciences, University of Colorado Boulder, Boulder, CO 80309; <sup>f</sup>Laboratory of Atmospheric Chemistry, Paul Scherrer Institute, Villigen 5232, Switzerland; <sup>g</sup>Institute for Atmospheric and Earth System Research/Physics, Faculty of Science, University of Helsinki, Helsinki 00014, Finland; <sup>h</sup>Institute of Meteorology and Climate Research Atmospheric Aerosol Research, Karlsruhe Institute of Technology, Karlsruhe 76021, Germany; <sup>i</sup>Division of Chemistry and Chemical Engineering, California Institute of Technology, Pasadena, CA 91125; <sup>j</sup>Institute for Ion Physics and Applied Physics, University of Innsbruck, Innsbruck 6020, Austria; <sup>k</sup>Faculty of Physics, University of Vienna, Vienna 1090, Austria; <sup>l</sup>Irvine Department of Chemistry, University of California, Irvine, CA 92697; <sup>m</sup>Atmospheric Microphysics Department, Leibniz Institute for Tropospheric Research, Leipzig 04318, Germany; <sup>n</sup>Department of Technical Physics, University of Eastern Finland, Kuopio 70211, Finland; <sup>o</sup>TOPWERK, Thun 3645, Switzerland; <sup>p</sup>Ionicon Analytik GmbH, Innsbruck 6020, Austria; <sup>q</sup>Department of Chemical Engineering/Center for Atmospheric Particle Studies, Carnegie Mellon University, Pittsburgh, PA 15213; <sup>r</sup>Department of Engineering and Public Policy, Carnegie Mellon University, Pittsburgh, PA 15213; <sup>s</sup>Department of Chemistry, Carnegie Mellon University, Pittsburgh, PA 15213; <sup>t</sup>Yusuf Hamied Department of Chemistry, University of Cambridge, Cambridge CB2 1EW, United Kingdom; <sup>u</sup>School of Atmospheric Sciences, Joint International Research Laboratory of Atmospheric and Earth System Sciences, Nanjing University, Nanjing 210023, China; <sup>v</sup>Aerosol and Haze Laboratory, Beijing Advanced Innovation Center for Soft Matter Science and Engineering, Beijing University of Chemical Technology, Beijing 100029, China; <sup>w</sup>Finnish Meteorological Institute, Helsinki 00101, Finland; <sup>x</sup>Institute for Materials Chemistry, Technische Universität (TU) Wien, Vienna 1060, Austria; <sup>y</sup>Department of the Geophysical Sciences, The University of Chicago, Chicago, IL 60637; <sup>z</sup>Aerodyne Research Inc., Billerica, MA 01821; and <sup>aa</sup>Institute of Meteorology and Climate Research Atmospheric Trace Gases and Remote Sensing, Karlsruhe Institute of Technology, Karlsruhe 76021, Germany

Author contributions: C.X., A.P., J.L., and T.C. designed research; C.X., M. Kohl, S.R., J.A., L.C.-P., R.C.-S., L.D., J.D., S.E., H.F., K.H., W.K., F.K., C.J.L., B.M., A.M., A.O., P.R., B.R., D.M.R., M. Schervish, W.S., M.K.S., M. Simon, E.S., Y.T., N.S.U., G.R.U., L.V., B.Y., W.Y., I.Z., Z.Z., J.C., N.M.D., R.C.F., H.G., I.E.H., A.H., H.H., X.-C.H., J.K., M. Kulmala, K.L., O.M., T.P., M.L.P., S.S., D.S., M.W., P.M.W., D.R.W., M.H., R.V., A.P., J.L., and T.C. performed research; C.X., M. Kohl, M.H., R.V., A.P., J.L., and T.C. analyzed data; and C.X., J.L., and T.C. wrote the paper.

24. X. Zhang *et al.*, Ammonia emissions may be substantially underestimated in China. *Environ. Sci. Technol.* **51**, 12089–12096 (2017).
25. J. Vira, P. Hess, M. Ossouhou, C. Galy-Lacaux, Evaluation of interactive and prescribed agricultural ammonia emissions for simulating atmospheric composition in CAM-chem. *Atmos. Chem. Phys.* **22**, 1883–1904 (2022).
26. T. Li, Z. Wang, Increasing  $\text{NH}_3$  emissions in high emission seasons and its spatiotemporal evolution characteristics during 1850–2060. *Atmosphere* **14**, e1056 (2023).
27. C. Xenofontos *et al.*, The impact of ammonia on particle formation in the Asian Tropopause Aerosol Layer. *NPJ Clim. Atmos. Sci.* **7**, 215 (2024).
28. M. Prather *et al.*, "IPCC, 2013: Annex II: Climate system scenario tables" in *Climate Change 2013* (2013). (2014).
29. F. J. Dentener, P. J. Crutzen, A three-dimensional model of the global ammonia cycle. *J. Atmos. Chem.* **19**, 331–369 (1994).
30. P. J. Adams, J. H. Seinfeld, D. M. Koch, Global concentrations of tropospheric sulfate, nitrate, and ammonium aerosol simulated in a general circulation model. *J. Geophys. Res. Atmos.* **104**, 13791–13823 (1999).
31. H. Ziereis, F. Arnold, Gaseous ammonia and ammonium ions in the free troposphere. *Nature* **321**, 503–505 (1986).
32. H. W. Georgii, W. J. Müller, On the distribution of ammonia in the middle and lower troposphere. *Tellus* **26**, 180–184 (1974).
33. C. L. Heald *et al.*, Atmospheric ammonia and particulate inorganic nitrogen over the United States. *Atmos. Chem. Phys.* **12**, 10295–10312 (2012).
34. M. Höpfner *et al.*, First detection of ammonia ( $\text{NH}_3$ ) in the Asian summer monsoon upper troposphere. *Atmos. Chem. Phys.* **16**, 14357–14369 (2016).
35. M. Wang *et al.*, Synergistic  $\text{HNO}_3 - \text{H}_2\text{SO}_4 - \text{NH}_3$  upper tropospheric particle formation. *Nature* **605**, 483–489 (2022).
36. R. S. Park, S. Lee, S. K. Shin, C. H. Song, Contribution of ammonium nitrate to aerosol optical depth and direct radiative forcing by aerosols over East Asia. *Atmos. Chem. Phys.* **14**, 2185–2201 (2014).
37. A. Beusen, A. Bouwman, P. Heuberger, G. Van Dreht, K. Van Der Hoek, Bottom-up uncertainty estimates of global ammonia emissions from global agricultural production systems. *Atmos. Environ.* **42**, 6067–6077 (2008).
38. C. A. Beale *et al.*, Large sub-regional differences of ammonia seasonal patterns over India reveal inventory discrepancies. *Environ. Res. Lett.* **17**, 104006 (2022).
39. D. A. Hauglustaine, J. Balkanski, M. Schulz, A global model simulation of present and future nitrate aerosols and their direct radiative forcing of climate. *Atmos. Chem. Phys.* **14**, 11031–11063 (2014).
40. M. Höpfner *et al.*, Ammonium nitrate particles formed in upper troposphere from ground ammonia sources during Asian monsoons. *Nat. Geosci.* **12**, 608–612 (2019).
41. O. Appel *et al.*, Chemical analysis of the Asian tropopause aerosol layer (ATAL) with emphasis on secondary aerosol particles using aircraft-based in situ aerosol mass spectrometry. *Atmos. Chem. Phys.* **22**, 13607–13630 (2022).
42. P. J. Rasch *et al.*, An overview of geoengineering of climate using stratospheric sulphate aerosols. *Philos. Trans. R. Soc. A Math. Phys. Eng. Sci.* **366**, 4007–4037 (2008).
43. S. E. Bauer *et al.*, Historical (1850–2014) aerosol evolution and role on climate forcing using the GISS ModelE2.1 contribution to CMIP6. *J. Adv. Model. Earth Syst.* **12**, e2019MS001978 (2020).
44. S. E. Bauer *et al.*, The turning point of the aerosol era. *J. Adv. Model. Earth Syst.* **14**, e2022MS003070 (2022).

45. J. Kirkby *et al.*, Role of sulphuric acid, ammonia and galactic cosmic rays in atmospheric aerosol nucleation. *Nature* **476**, 429–433 (2011).
46. J. Kirkby *et al.*, Ion-induced nucleation of pure biogenic particles. *Nature* **533**, 521–526 (2016).
47. E. M. Dunne *et al.*, Global atmospheric particle formation from CERN CLOUD measurements. *Science* **354**, 1119–1124 (2016).
48. J. Kirkby *et al.*, Atmospheric new particle formation from the CERN CLOUD experiment. *Nat. Geosci.* **16**, 948–957 (2023).
49. P. Jöckel *et al.*, Development cycle 2 of the modular earth submodel system (MESSy2). *Geosci. Model. Dev.* **3**, 717–752 (2010).
50. J. Shen *et al.*, New particle formation from isoprene under upper-tropospheric conditions. *Nature* **636**, 115–123 (2024).
51. J. Curtius *et al.*, Isoprene nitrates drive new particle formation in Amazon's upper troposphere. *Nature* **636**, 124–130 (2024).
52. J. H. Seinfeld, S. N. Pandis, K. J. Noone, Atmospheric chemistry and physics: From air pollution to climate change. *Phys. Today* **51**, 88–90 (1998).
53. J. Lelieveld *et al.*, The South Asian monsoon-pollution pump and purifier. *Science* **361**, 270–273 (2018).
54. N. M. Donahue *et al.*, Low temperature growth of sub 10 nm particles by ammonium nitrate condensation. *Environ. Sci. Atmos.* **5**, 67–81 (2025).
55. A. Pozzer *et al.*, AOD trends during 2001–2010 from observations and model simulations. *Atmos. Chem. Phys.* **15**, 5521–5535 (2015).
56. P. Taylor, H. R. Pruppacher, J. D. Klett, Microphysics of clouds and precipitation. *Bound. Layer Meteorol.* **86**, 187–188 (1998).
57. C. R. Hoyle, B. P. Luo, T. Peter, The origin of high ice crystal number densities in cirrus clouds. *J. Atmos. Sci.* **62**, 2568–2579 (2005).
58. B. Kärcher, Homogeneous ice formation in convective cloud outflow regions. *Q. J. R. Meteorol. Soc.* **143**, 2093–2103 (2017).
59. J. E. Kay, R. Wood, Timescale analysis of aerosol sensitivity during homogeneous freezing and implications for upper tropospheric water vapor budgets. *Geophys. Res. Lett.* **35**, e2007GL032628 (2008).
60. E. J. Jensen *et al.*, Ice nucleation and dehydration in the Tropical Tropopause Layer. *Proc. Natl. Acad. Sci. U.S.A.* **110**, 2041–2046 (2013).
61. R. Wagner *et al.*, Solid ammonium nitrate aerosols as efficient ice nucleating particles at cirrus temperatures. *J. Geophys. Res. Atmos.* **125**, e2019JD032248 (2020).
62. E. E. McDuffie *et al.*, A global anthropogenic emission inventory of atmospheric pollutants from sector- and fuel-specific sources (1970–2017): An application of the Community Emissions Data System (CEDS). *Earth Syst. Sci. Data* **12**, 3413–3442 (2020).
63. B. Bertozzi *et al.*, Influence of the neutralization degree on the ice nucleation ability of ammoniated sulfate particles. *J. Geophys. Res. Atmos.* **129**, e2023JD040078 (2024).
64. R. Ma *et al.*, Mitigation potential of global ammonia emissions and related health impacts in the trade network. *Nat. Commun.* **12**, 6308 (2021).
65. European Parliament and Council of the European Union, Directive (EU) 2024/2881 of 23 October 2024 on ambient air quality and cleaner air for Europe (recast). Official Journal of the European Union, L 2024/2881, 20 November 2024. ELI: <http://data.europa.eu/eli/dir/2024/2881/oj>.
66. IPCC, *Climate Change 2021: The Physical Science Basis. Contribution of Working Group I to the Sixth Assessment Report of the Intergovernmental Panel on Climate Change* (Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 2021).
67. IPCC, *Summary for Policymakers*, V. Masson-Delmotte *et al.*, Eds. (Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 2021), pp. 3–32.
68. E. Roeckner *et al.*, Sensitivity of simulated climate to horizontal and vertical resolution in the ECHAM5 atmosphere model. *J. Clim.* **19**, 3771–3791 (2006).
69. U. Pöschl, R. von Kuhlmann, N. Poisson, P. J. Crutzen, Development and intercomparison of condensed isoprene oxidation mechanisms for global atmospheric modeling. *J. Atmos. Chem.* **37**, 29–52 (2000).
70. W. Meteorology, A three-dimensional science: Second session of the Commission for Aerology. *WMO Bull.* **4**, 134–138 (1957).
71. K. Pringle *et al.*, Description and evaluation of GMX: A new aerosol submodel for global simulations (v1). *Geosci. Model. Dev.* **3**, 391–412 (2010).
72. S. Ehrhart *et al.*, Two new submodels for the Modular Earth Submodel System (MESSy): New Aerosol Nucleation (NAN) and small ions (IONS) version 1.0. *Geosci. Model. Dev.* **11**, 4987–5001 (2018).
73. S. Dietmüller *et al.*, A new radiation infrastructure for the Modular Earth Submodel System (MESSy, based on version 2.51). *Geosci. Model. Dev.* **9**, 2209–2222 (2016).
74. R. Sander *et al.*, The community atmospheric chemistry box model CAABA/MECCA-4.0. *Geosci. Model. Dev.* **12**, 1365–1385 (2019).
75. R. Sander *et al.*, The photolysis module JVAL-14, compatible with the MESSy standard, and the JVal PreProcessor (JVPP). *Geosci. Model. Dev.* **7**, 2653–2662 (2014).
76. A. Kerkweg *et al.*, Technical Note: An implementation of the dry removal processes DRY DEPosition and SEDimentation in the Modular Earth Submodel System (MESSy). *Atmos. Chem. Phys.* **6**, 4617–4632 (2006).
77. L. Ganzeveld, J. Lelieveld, Dry deposition parameterization in a chemistry general circulation model and its influence on the distribution of reactive trace gases. *J. Geophys. Res. Atmos.* **100**, 20999–21012 (1995).
78. M. Tiedtke, A comprehensive mass flux scheme for cumulus parameterization in large-scale models. *Mon. Weather Rev.* **117**, 1779–1800 (1989).
79. N. T. Erik, Extended versions of the convective parametrization scheme at ECMWF and their impact on the mean and transient activity of the model in the tropics. *Res. Rep. Tech. Memo.* **206**, 1–41 (1994).
80. H. Tost, P. Jöckel, J. Lelieveld, Influence of different convection parameterisations in a GCM. *Atmos. Chem. Phys.* **6**, 5475–5493 (2006).
81. H. Tost *et al.*, Global cloud and precipitation chemistry and wet deposition: Tropospheric model simulations with ECHAM5/MESSy1. *Atmos. Chem. Phys.* **7**, 2733–2757 (2007).
82. V. Grewe *et al.*, Origin and variability of upper tropospheric nitrogen oxides and ozone at northern mid-latitudes. *Atmos. Environ.* **35**, 3421–3433 (2001).
83. C. Fountoukis, A. Nenes, ISORROPIA II: A computationally efficient thermodynamic equilibrium model for  $K^+$ - $Ca^{2+}$ - $Mg^{2+}$ - $NH_4^+$ - $Na^+$ - $SO_4^{2-}$ - $NO_3^-$ - $Cl^-$ - $H_2O$  aerosols. *Atmos. Chem. Phys.* **7**, 4639–4659 (2007).
84. H. Fischer *et al.*, MIPAS: An instrument for atmospheric and climate research. *Atmos. Chem. Phys.* **8**, 2151–2188 (2008).
85. D. Blumstein *et al.*, "IASI instrument: Technical overview and measured performances" in *Infrared Spaceborne Remote Sensing XII*, M. Strojnik, Ed. (International Society for Optics and Photonics, SPIE, 2004), vol. 5543, pp. 196–207.
86. L. Clarisse *et al.*, The IASI  $NH_3$  version 4 product: Averaging kernels and improved consistency. *Atmos. Meas. Tech.* **16**, 5009–5028 (2023).
87. S. Whitburn *et al.*, A flexible and robust neural network IASI- $NH_3$  retrieval algorithm. *J. Geophys. Res. Atmos.* **121**, 6581–6599 (2016).
88. B. Franco *et al.*, A general framework for global retrievals of trace gases From IASI: Application to methanol, formic acid, and PAN. *J. Geophys. Res. Atmos.* **123**, 13963–13984 (2018).
89. M. T. Chahine *et al.*, AIRS: improving weather forecasting and providing new data on greenhouse gases. *Bull. Am. Meteorol. Soc.* **87**, 911–926 (2006).
90. J. X. Warner, Z. Wei, L. L. Strow, R. R. Dickerson, J. B. Nowak, The global tropospheric ammonia distribution as seen in the 13-year AIRS measurement record. *Atmos. Chem. Phys.* **16**, 5467–5479 (2016).
91. J. X. Warner *et al.*, Increased atmospheric ammonia over the world's major agricultural areas detected from space. *Geophys. Res. Lett.* **44**, 2875–2884 (2017).
92. C. R. Thompson *et al.*, The NASA atmospheric tomography (ATom) mission: Imaging the chemistry of the global atmosphere. *Bull. Am. Meteorol. Soc.* **103**, E761–E790 (2022).
93. S. Platnick *et al.*, MODIS cloud optical properties: User guide for the Collection 6 Level-2 MOD06/MYD06 product and associated Level-3 Datasets. *Version 1*, 145 (2015).
94. P. Kumar *et al.*, Global atmospheric inversion of the  $NH_3$  emissions over 2019–2022 using the LMDZ-INCA chemistry-transport model and the IASI  $NH_3$  observations. *EGU sphere* **2025**, 1–32 (2025).
95. X. Wang *et al.*, The influence of ammonia emissions on the size-resolved global atmospheric aerosol composition and acidity. *EGU sphere* **2025**, 1–35 (2025).
96. E. Vignati, J. Wilson, P. Stier, M7: An efficient size-resolved aerosol microphysics module for large-scale aerosol transport models. *J. Geophys. Res. Atmos.* **109**, e2003JD004485 (2004).
97. N. A. Fuchs, The mechanics of aerosols. By N. A. Fuchs. Translated by R. E. Daisley and Marina Fuchs; Edited by C. N. Davies. London (Pergamon Press), 1964. Pp. xiv, 408; 82 Figures; 40 Tables. £6. *Q. J. R. Meteorol. Soc.* **91**, 249–249 (1965).
98. Modular Earth Submodel System (MESSy), MESSy – Modular Earth Submodel System. <https://messy-interface.org/>. Accessed 10 September 2025.
99. C. Xenofontos, T. Christoudias, Global impact of anthropogenic  $NH_3$  emissions on upper tropospheric aerosol formation: Data sources [Data set]. Zenodo. <https://doi.org/10.5281/zenodo.14875637>. Deposited 30 September 2025.