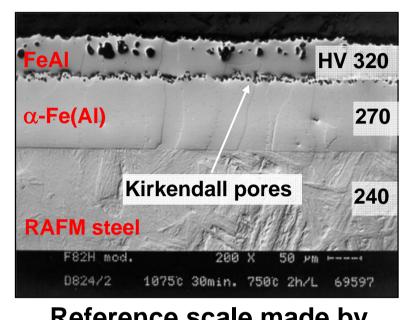


Influence of deposition conditions on the microstructure of Al-based coatings for applications as corrosion and anti-permeation barrier

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Motivation



Reference scale made by **HDA** after heat treatment $(1040^{\circ}C/0.5 h + 750^{\circ}C /1 h)$

Previous research revealed that the application of aluminum-based barriers is suitable to minimize corrosion rates of Eurofer steel in Pb-15.7Li (picture to the right) and tritium-permeation from the liquid breeder into the cooling system (HCLL) in an envisaged future fusion reactor.

Developed deposition techniques to electroplate aluminum are based on water-free electrolytes. Either ECA or ECX process can provide several advantages compared to the established Hot-Dip Aluminization (HDA) process:

- Better thickness distribution
- Controllable coating thicknesses
- Reduced amounts of aluminum applicable (low activation)

However, the main challenge is to convert the electroplated Al-coatings during a subsequent heat treatment, to enable the formation of protective scales on Eurofer for applications as corrosion and anti-permeation barriers in Pb-15.7Li.

Al-Surface ECA-plated

10 mA/cm², 1 h, 100°C

P2

25 mA/cm²

80%

AI-Surface ECX-plated

Electrodeposition of AI on Eurofer steel samples by ECX

pulse plated (P) samples \rightarrow j_{mean} and t were held constant

20 mA/cm²

80 mA/cm²

25%

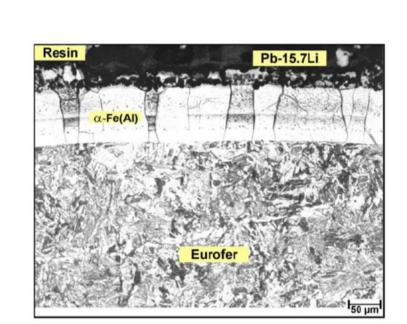
30 minutes

Direct current (DC) plated samples were compared to

Theoretical Al thickness should be about 12.5 µm

From an imidazolium-based ionic liquid

20 mA/cm², 0.5 h, 100°C



HDA coated sample after exposure in flowing Pb-15.7Li, taken from [1]

Electrodeposition of Aluminum

Aluminum is highly electronegative ($E_0 = -1.6 \text{ V vs. NHE}$)

- Al cannot be deposited from aqueous electrolytes
- Water-free / aprotic electrolytes are needed

Principle types of electrolytes to electrodeposit Al at moderate temperatures (RT-100°C)

- Toluene-based electrolytes → ECA process
- 2. Ionic liquids → ECX process

Electrochemical processes have in common

- No reaction with the substrate material Adjustable AI thickness (time, current density)
- Good adherence to the substrate
- Industrial relevance is given

Why ECX?

- ► Process is more flexible than ECA process
 - > Deposition parameters adjustable in a wider range (T, j, pulse plating,...)

Electrodeposition of AI by ECX

Electrolytes not oxygen sensitive → non-inflammable

Parameter

Jmean

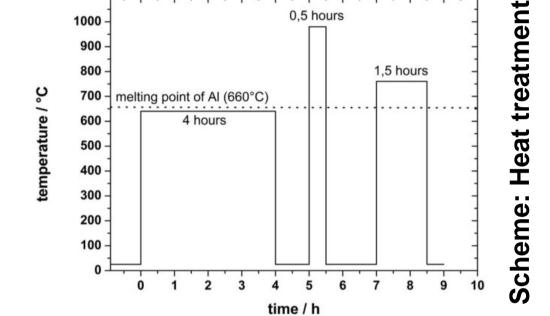
Jpeak

Duty Cycle

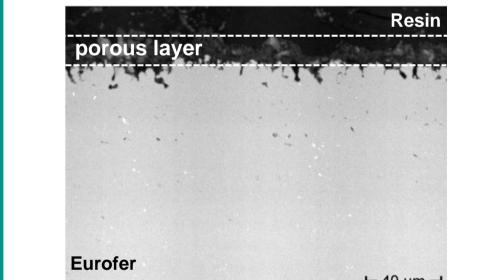
Heat treatment behavior

Experimental

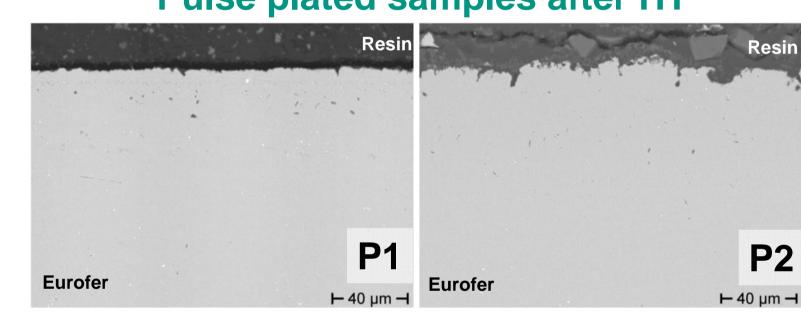
- Heat treatment (HT) procedure according to [1]
- 640°C, 4 h \rightarrow 980°C, 0.5 h \rightarrow 760°C, 1.5 h
- In flowing argon atmosphere
- Reduction of oxidation effects
- Samples cooled down under flowing Ar

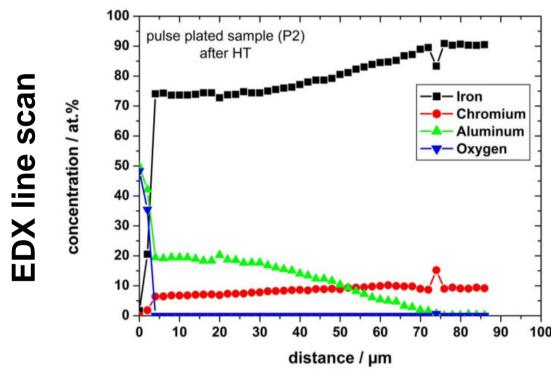


DC plated sample after HT

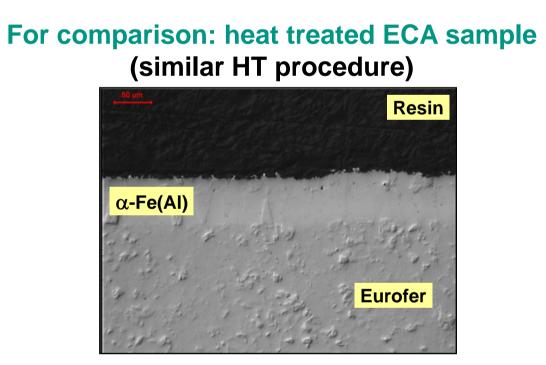


Pulse plated samples after HT





Pulse plated (P2) sample after HT



picture taken from [1]

Results

- Quality depends on surface morphology of Al-layers on Eurofer after electroplating
- Pulse plated samples by ECX: uniform formation of Al-based scales, comparable to scales produced by ECA (see [1])
- ► Rough and jagged surface structure (in this case DC plated): formation of relatively thick porous, oxygen containing layer on top of the sample
- ► All amount in the reacting zone of ca. 20 at.% (decreasing with distance from surface)

Influence of deposition parameters

100%

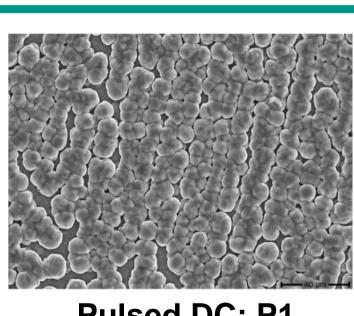
Scheme: Pulse plating

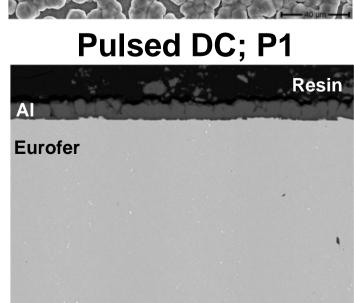
[Emim]Cl:AlCl₃;

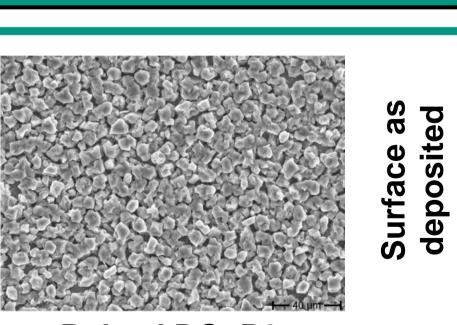
T=100°C;

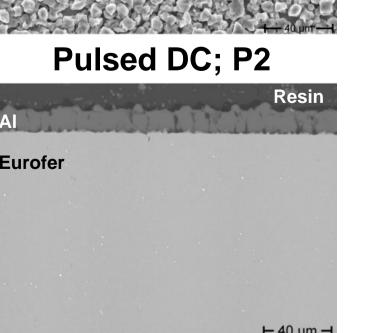
Agitation

Direct current









Surface morphology is strongly influenced by deposition conditions

- ▶ DC plated sample : Jagged and rough surface; crystallites not grown together
- Pulse plated samples: Dense, even and homogeneous morphology

Conclusions

ECX process show industrial relevance and deposit Al scales on Eurofer steel samples with controllable and reproducible thickness

- Morphologies of Al coatings made by ECX process depend on deposition parameters
- **▶** All layers have to be converted into protective scales by heat treatments (HT)
- HT behavior depends on the morphology of the coating produced by ECX
 - Formation of Fe-Al scales during HT is influenced by morphology Rough and jagged surface structures cannot be recommended for further HT
 - Even, dense and homogeneous morphologies are recommended for HT
- ➡ Heat treated samples with recommended morphology (pulse plated) were comparable to known ECA or HDA produced Al-based scales
- The more flexible ECX process has proven to be a promising alternative to ECA

References:

[1] Konys, J. et al.: Impact of heat treatment on surface chemistry of Al-coated Eurofer for application as anti-corrosion and T permeation barriers in flowing Pb-15.7 environment, Fus. Eng. Des., 87 (2012), 1483.