Institute of Functional Interfaces

Helmholtz Research School "Energy-related catalysis"

Karlsruhe Institute of Technology Combined NEXAFS and XPS Study of Tetrahydroxybenzene Adsorption on Cu(111) and Ni(111)

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Motivation

Surface functionalization in the form of two-dimensional networks is a method of growing interest for the development of functional, periodic nanometer-scale structures with. great potential in catalysis, for example. A common problem in generation of such structures is that more stable covalent assemblies are prone to structural defects limiting the achievable degree of order, while highly ordered hydrogen-bonded structures often lack the necessary stability. Metal Organic Networks are intermediate between those two types of structures. They posses quite interesting 3D-analgoues, Metal Organic Frameworks.

Tetrahydroxybenzene (THB) was recently found to dehydrogenate on Cu(111) [1] and form a regular network consisting of a completely dehydrogenated THB species and Cu-adatoms [2]. Interestingly, STM and XPS investigations showed that the nodes of this network consist of Cu-trimers, where each Cu-atom is bonded to two dehydrogenated THB-molecules via their oxygen atoms. With such a complex node, this network is a very realistic truly two-dimensional MOF analogue that allows to exploit the surface-science toolkit to investigate MOF-like structures. Furthermore, the generation of monodispersed and very small metal (oxide) clusters is of great interest in the area of catalysis.

Introduction and Aims



The network formed by the dehydrogenated THB on Cu(111) is interesting due to its unusual coordination motif: the Cu-trimers may be useful in catalysis and the structure effectively mimics a truly 2-D MOF. For these reasons, a more detailed characterization of the Cu-nodes is obviously highly desirable.





References

[1] F. Bebensee, K. Svane, C. Bombis, F. Masini, S. Klyatskaya, F. Besenbacher, M. Ruben, B. Hammer, T. Linderoth, Adsorption and Dehydrogenation of Tetrahydroxybenezene on Cu(111), Chem. Commun. 2013, 49, 9308-9310.

[2] F. Bebensee, K. Svane, C. Bombis, F, Masini, S. Klyatskaya, F. Besenbacher, M. Ruben, B. Hammer, T. Linderoth, A Surface Coordination Network Based on Copper Adatom Trimers, Angew. Chem. Int. Ed. 2014, 53, 12955-12959.

Figure 2: DFT-derived network structure and coordination geometry of individual molecular building block.

Aims:1. Details of the electronic structure of the network on Cu(111)2. Explore other surfaces for network formation, particularly Ni(111)

Experimental and Results



- The Ni(111) and Cu(111) substrates were prepared via sputter-anneal cycles until XPS showed no carbon contamination and a clear LEED pattern was observed
- THB molecules (provided by Dr. M Tsotsalas) were sublimated onto clean Ni(111) and Cu(111) substrates held at 110 K from a home-built Knudsen cell at 330 K.
- The sample transfer to the analysis chamber after preparation lead to only a moderate increase of the sample temperature (below 15 K).
 After characterization of the as prepared samples, the respective samples were annealed for 20 minutes at the stated temperatures in order to achieve the desired dehydrogenation. The subsequent measurements were performed after re-cooling the sample to 110 K.

All experiments reported were carried out at the HESGM-endstation shown above.

The endstation provides a preparation chamber for sputtering and annealing. It is furthermore equipped with a LEED optics to check the surface structure.

The attached analysis chamber allows for NEXAFS and XPS experiments using a home-build detector and a hemispherical analyzer, respectively.





- Molecular adsorption at low temperature, even formation of multilayers
 Progressing dehydrogenation with increasing temperature (cf. [1], [2])
- Should provide a more detailed picture in conjunction with modelling
 Angular dependence suggests planar adsorption geometry

Conclusions and Outlook

- Different dehydrogenation states can be studied using NEXAFS
 Complete with simulations and experiments on high density phase on Cu(111)
- Ni(111) is not a suitable substrate for network formation!
 Investigate other promising candidates, e.g. Pt(111)

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