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X-ray photoelectron spectroscopy of osmium tetroxide stained solid electrolyte interphase on graphite electrodes

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Introduction

Aging of the solid electrolyte interphase (SEI) is one of the major degradation process in a lithium ion batteries with graphite anodes.

- The detailed structure of the SEI still remains largely unknown 📫 investigation with high resolution electron microscopy.
- Instability of the SEI in the electron beam (volatile) complicates analysis \implies osmium tetroxide (OsO₄) staining stabilizes & enhances contrast [1].



Experiment

Sample preparation

- Electrodes were always handled and transferred under inert gas.
- Graphite electrodes were cycled with C/20 to build the SEI layer.
- As reference a sample was put into LP30 and dried without cycling.
- Samples were cut in the middle
 - one half directly to XPS
 - one exposed to OsO_4 atmosphere (12 h < t < 20 h) \implies XPS
- Samples were investigated with SEM after XPS analysis.

XPS evaluation

Evaluation of XPS measurements is difficult, due to the overlaps of Table 1: Measured averages of four samples the Li 1s with Os 4f - doublet and C 1s with Os $4d_{5/2}$ - spectra.

Results XPS

- There exists four Osmium doublets with Os $4f_{7/2}$ at 53.0 eV, 54.4 eV, 55.2 eV and 56.8 eV.
- Different binding energy for ethylene carbonate (EC) in LP30 (291.3 eV) and SEI (290.1 eV).
- After staining, the total amount of Os is about 0.2 at % for LP30 and about 4 at % in SEI
- OsO₄ does not react with EC
- After staining the SEI all the Li is bound to O as Li₂O₂ (55.0 eV, O1s 531.6 eV).
- Os seams to react with phosphorus and fluorine, ଳୁ causing a shift to higher binding energy.



- At first, Os 4f spectrum was fitted without any Li 1s peak. Then the relative intensity were kept the same for Os $4d_{5/2}$ and Os $4p_{3/2}$.
- Therafter, Li 1s peaks were added, and the Os heights were fixed, due to the very large difference of the photoionization cross section of lithium (~ 0.06) and osmium (6.96).



Fig. 1: SEM image of a graphite electrode with

	Measured value
FWHM Os 4f _{7/2}	1.67 ± 0.24 eV
FWHM Os 4d _{5/2}	3.32 ± 0.24 eV
ΔE (Os 4d _{5/2} – Os 4f _{7/2})	227.29 ± 0.47 eV
ΔE (Os 4p _{3/2} – Os 4f _{7/2})	419.25 ± 0.46 eV

Results SEM

- Dried LP30 (Fig. 1a)) is more stable under the electron beam than the SEI.
- There were thin layers of LP30 confirming the $XPS - analysis \implies graphite is still visible.$
- LP30 is dried in small islands.
- After staining (Fig. 1b)) the size of the islands increased.
- Fig. 1 c) shows a graphite surface with SEI.
- Thin SEI layer (marked with white arrows) is vapored away, after the first image. can not be sure to get a real image of the SEI.
- After staining, some areas are brighter (Fig. 1d)) in the electron beam and they have a nice nanostructure.
- These structure is very stable (6 to 10 images no big change) under SEM investigation.
- EDX measurements reveal O at $\% \gg$ Os, P, F.
- Solo and Aspect Binding energy (eV) F1s P2p PF_{x} Li_xPF_v SEI Os-P-LiPF₆ LiPF

a) dried electrolyte LP30 b) same stained with OsO₄ c) SEI from four cycles C/20 d) SEI stained with OsO₄

 \longrightarrow Li₂O₂



Conclusion

- XPS analysis point out, that OsO_4 (Os(VIII)) is reduced to Os(VI) Os(V) and Os(IV): 56.8 eV, 55.2 eV, 54.4 eV and 53.0 eV.
- The oxygen, which is released from OsO_4 , mainly reacts with the lithium in the SEI to Li_2O_2 (55.0 eV, O1s 531.6 eV).
- There is found only PF_x (136.7 eV, 686.8 eV) in the stained SEI, because the whole Li is bound in Li₂O₂. The PF_x accumulates around the OsO_x .
- It can be confirmed that the stained SEI is more stable under electron beam investigations.
- SEI nano structures could be observed on graphite surfaces by SEM.

Reference

[1] Martin Zier, Frieder Scheiba, et al., Journal of Power Sources, "Lithium dendrite and solid electrolyte interphase investigation using OsO4", Issue 266, pp198 – 207, 2014

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