

# Titanium substituted $\text{LiCoTi}_x\text{Mn}_{1-x}\text{O}_4$ ( $0.04 \leq x \leq 0.16$ ): *High-Voltage cathode materials for Li-Ion Batteries*

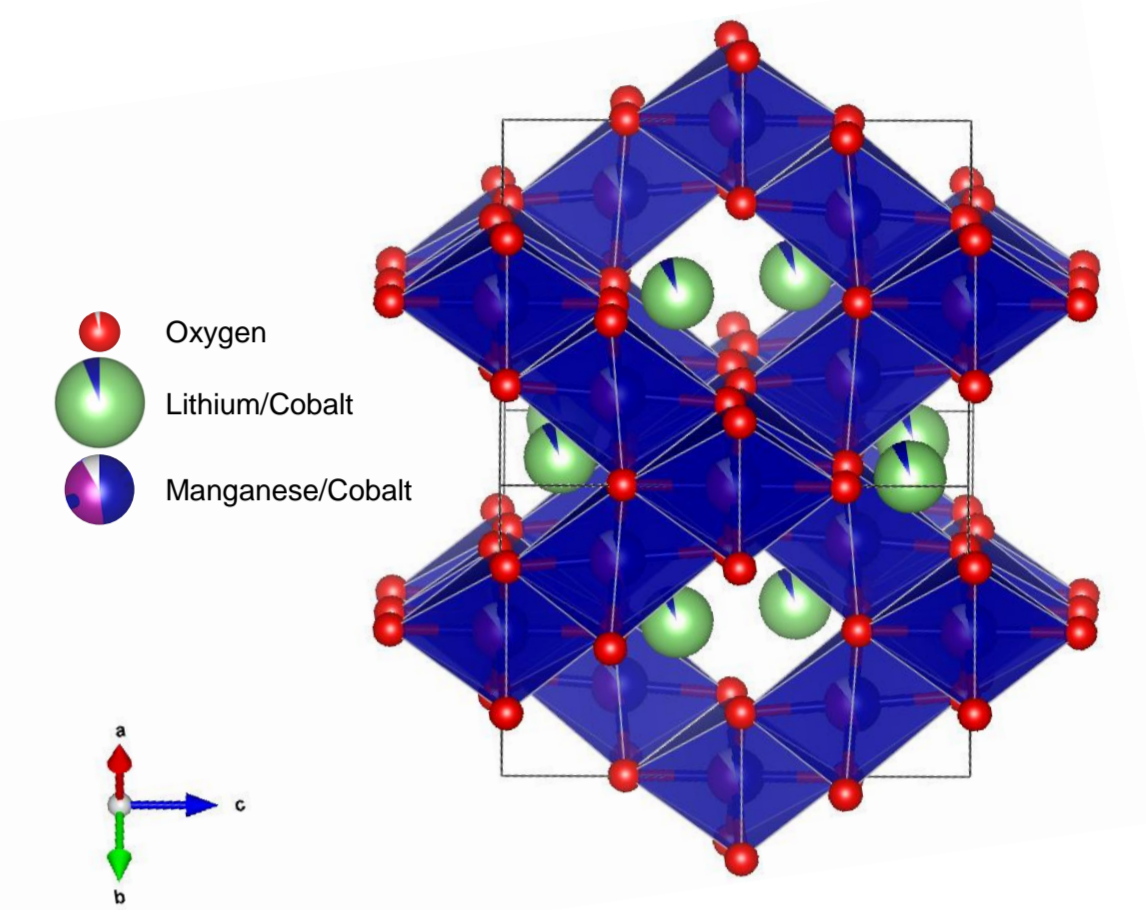
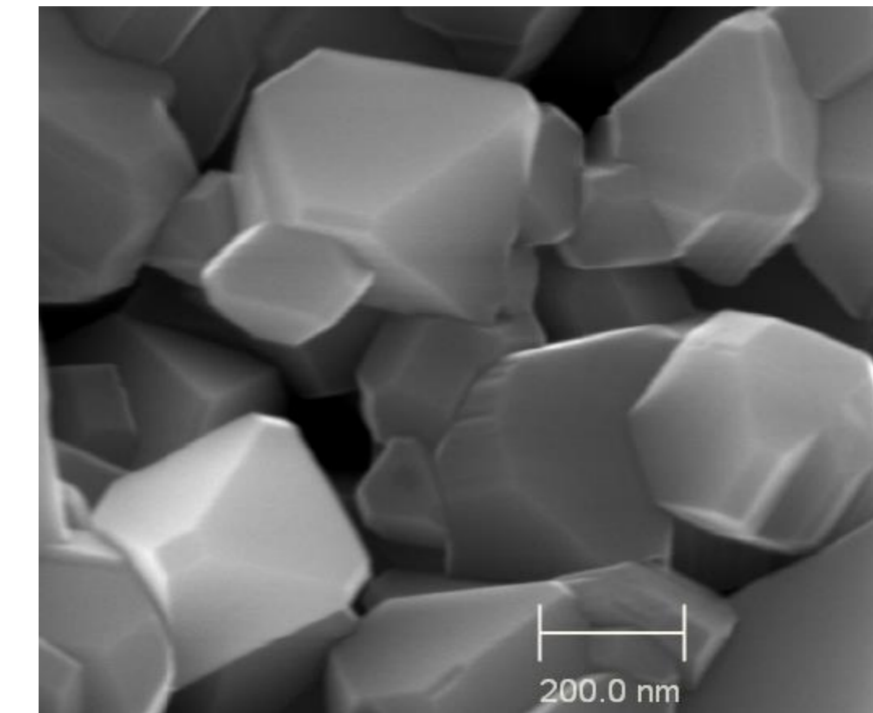
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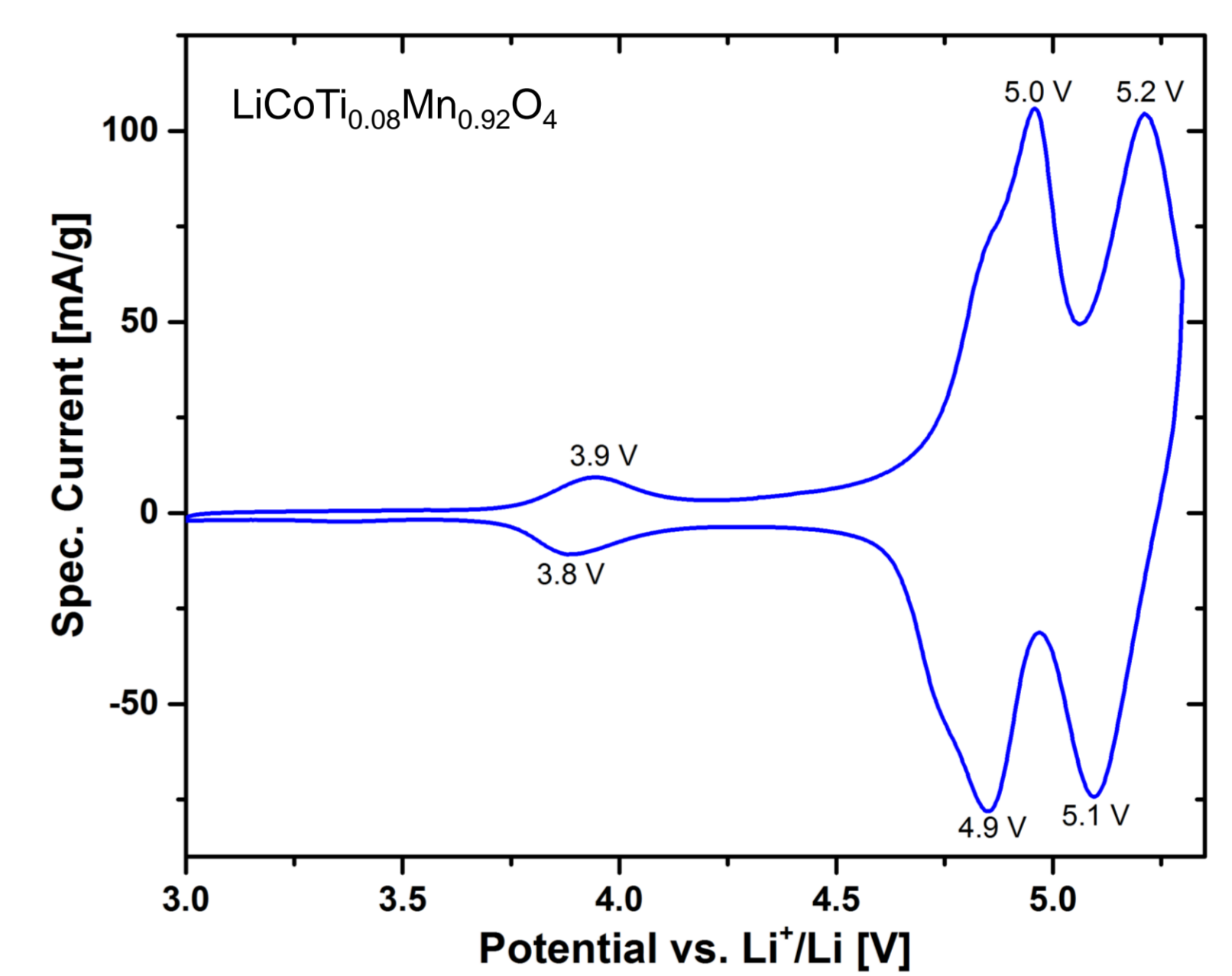
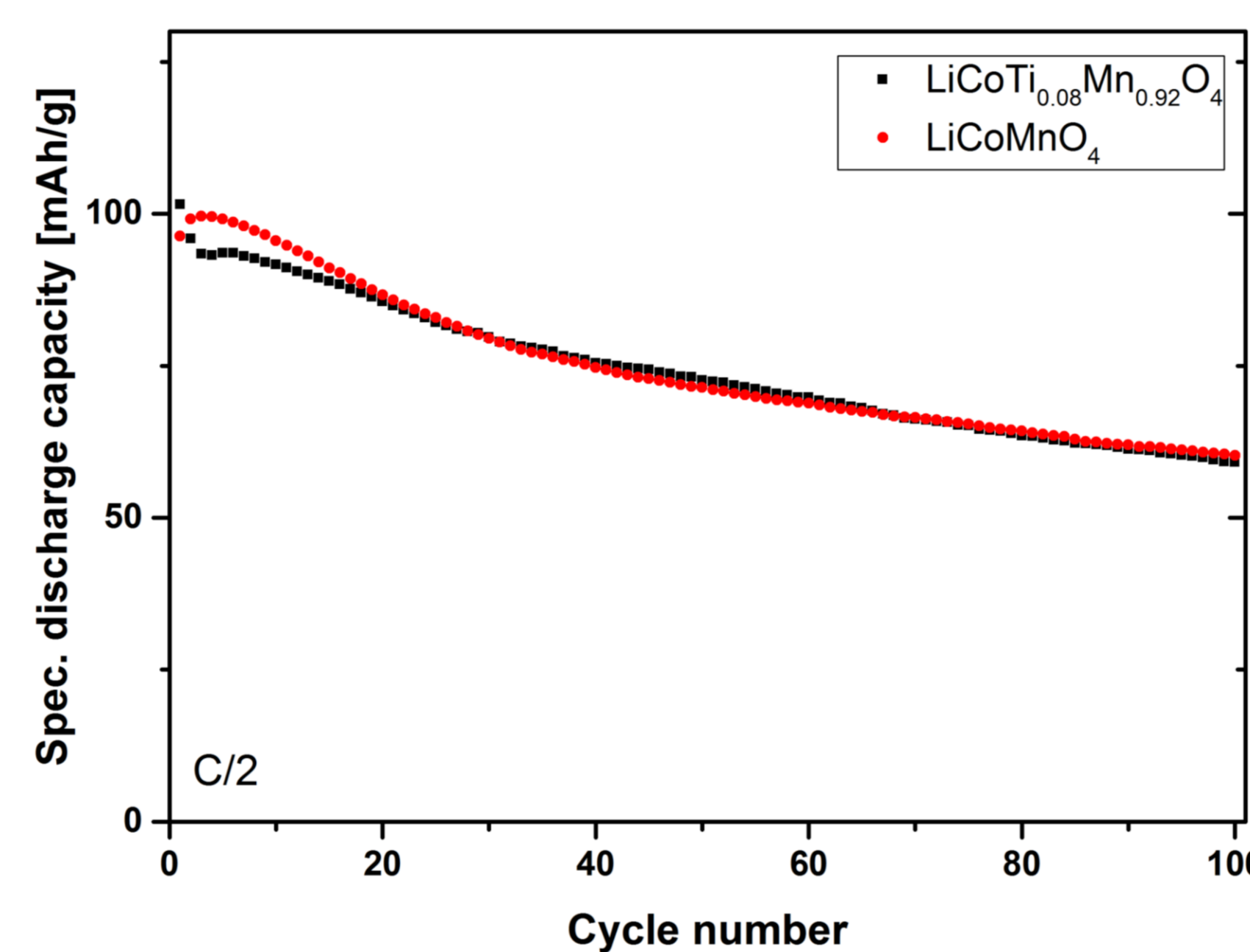
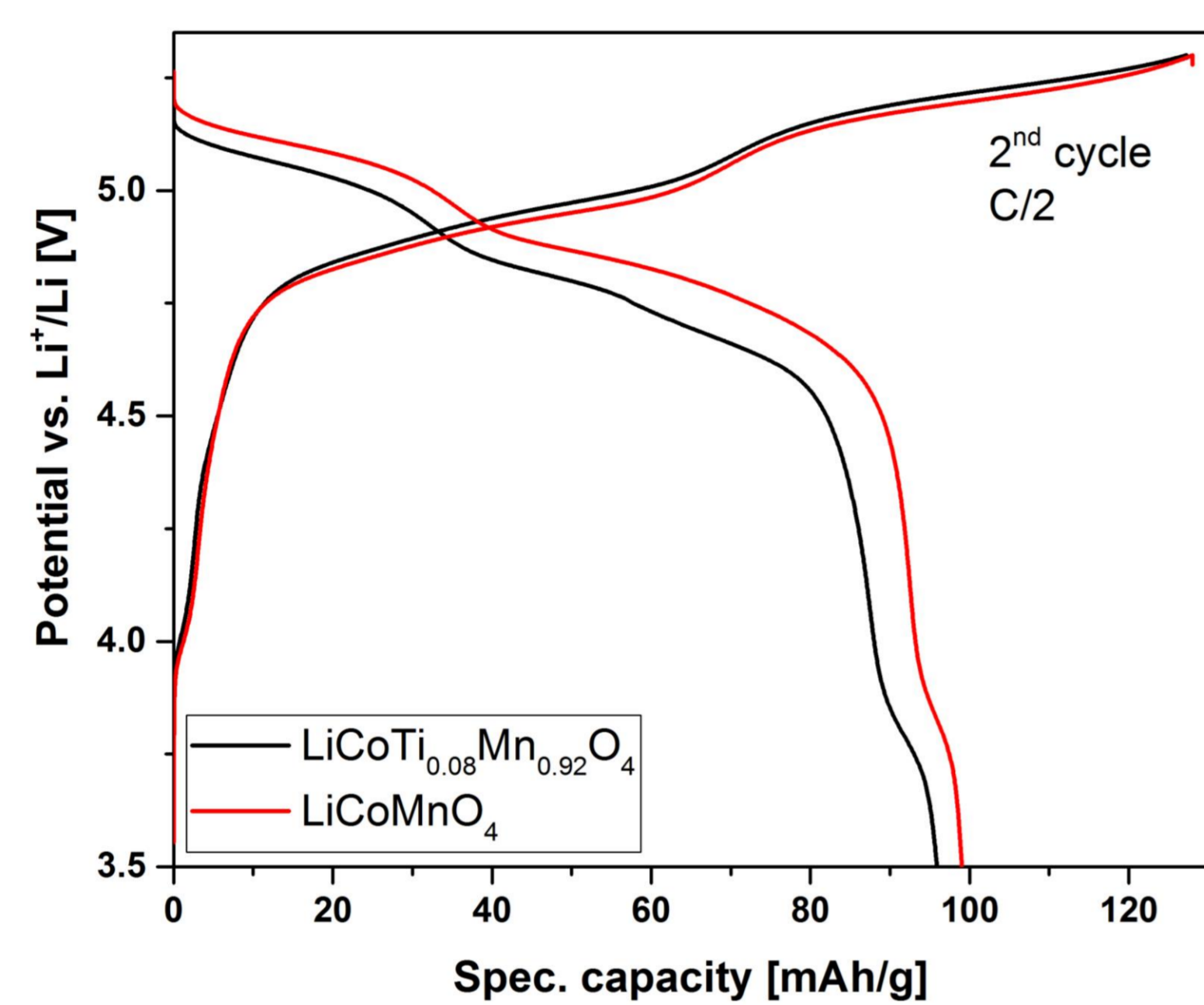
## Motivation

- The specific capacity and operating potential of commonly used high-rate  $\text{LiMn}_2\text{O}_4$  spinel cathode material is low which leads to an inferior energy density when combined in a LIB.
- The operating potential and specific capacity of a Mn-based spinel can be raised by substituting  $\text{Mn}^{3+}$  with  $\text{Co}^{3+}$ .
- Substituting  $\text{Mn}^{4+}$  with  $\text{Ti}^{4+}$  could reduce the formation of  $\text{Mn}^{3+}$  and therefore could reduce the oxygen loss during cycling, which improves the structural stability of the active material.

## Morphology

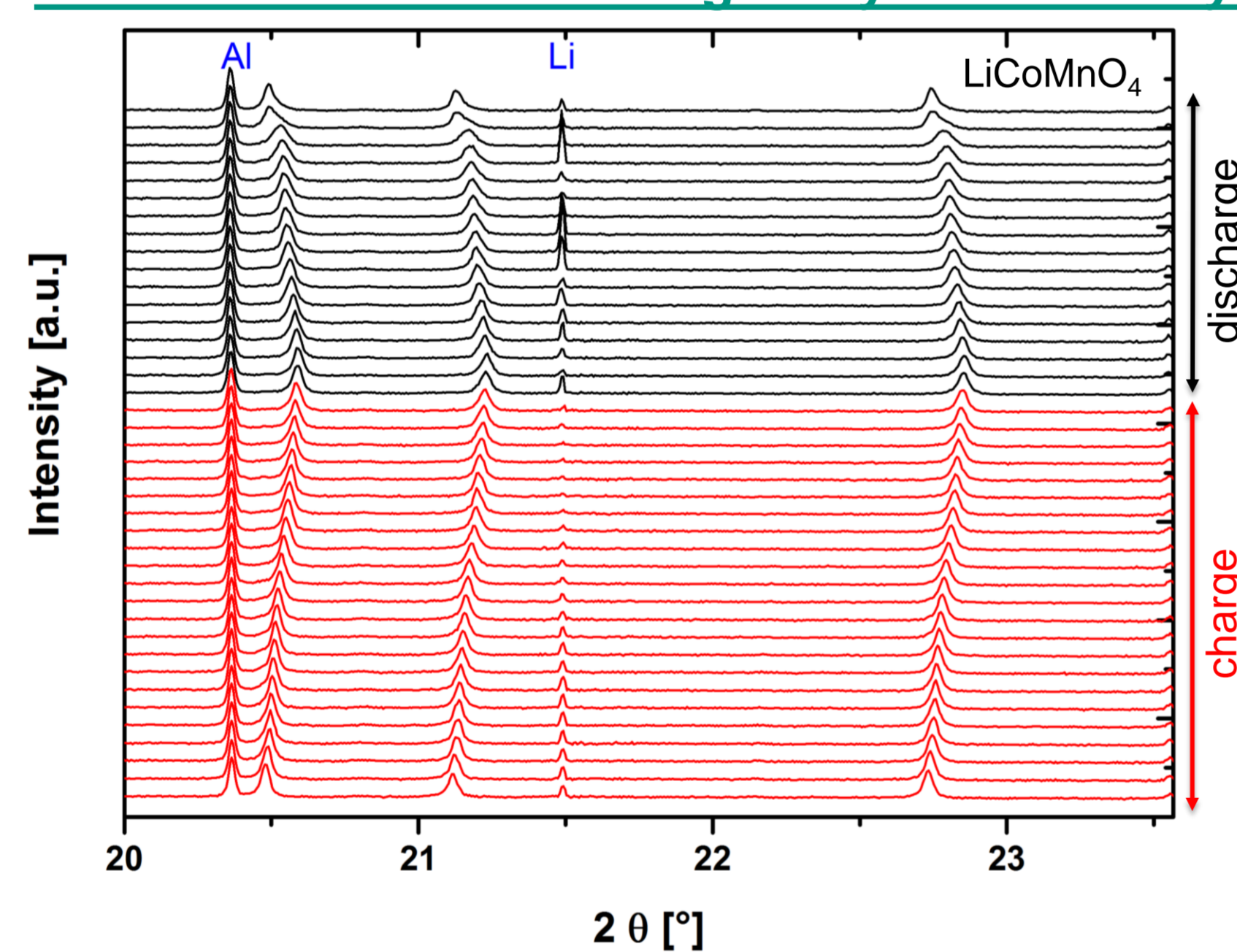


## Electrochemical investigations

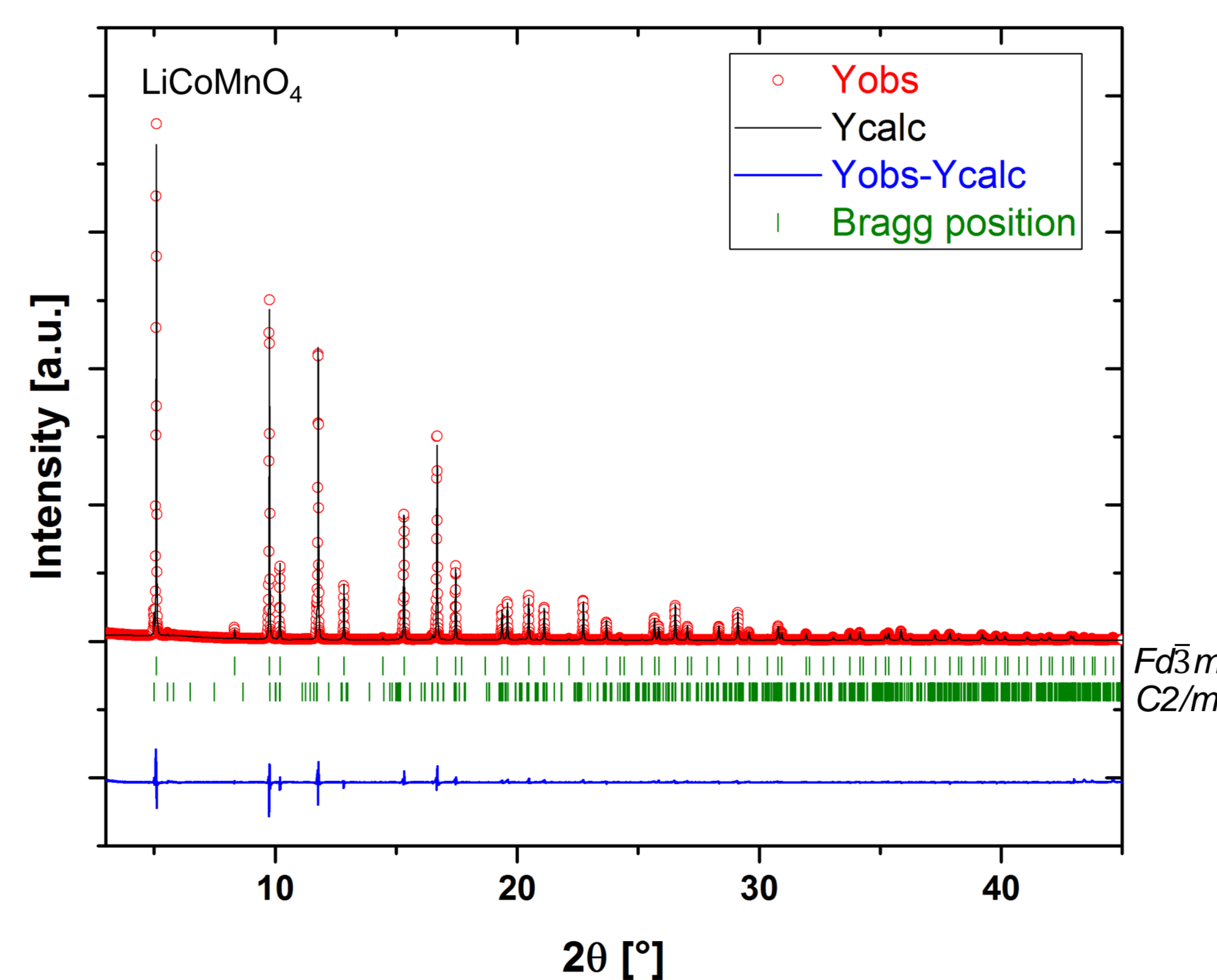
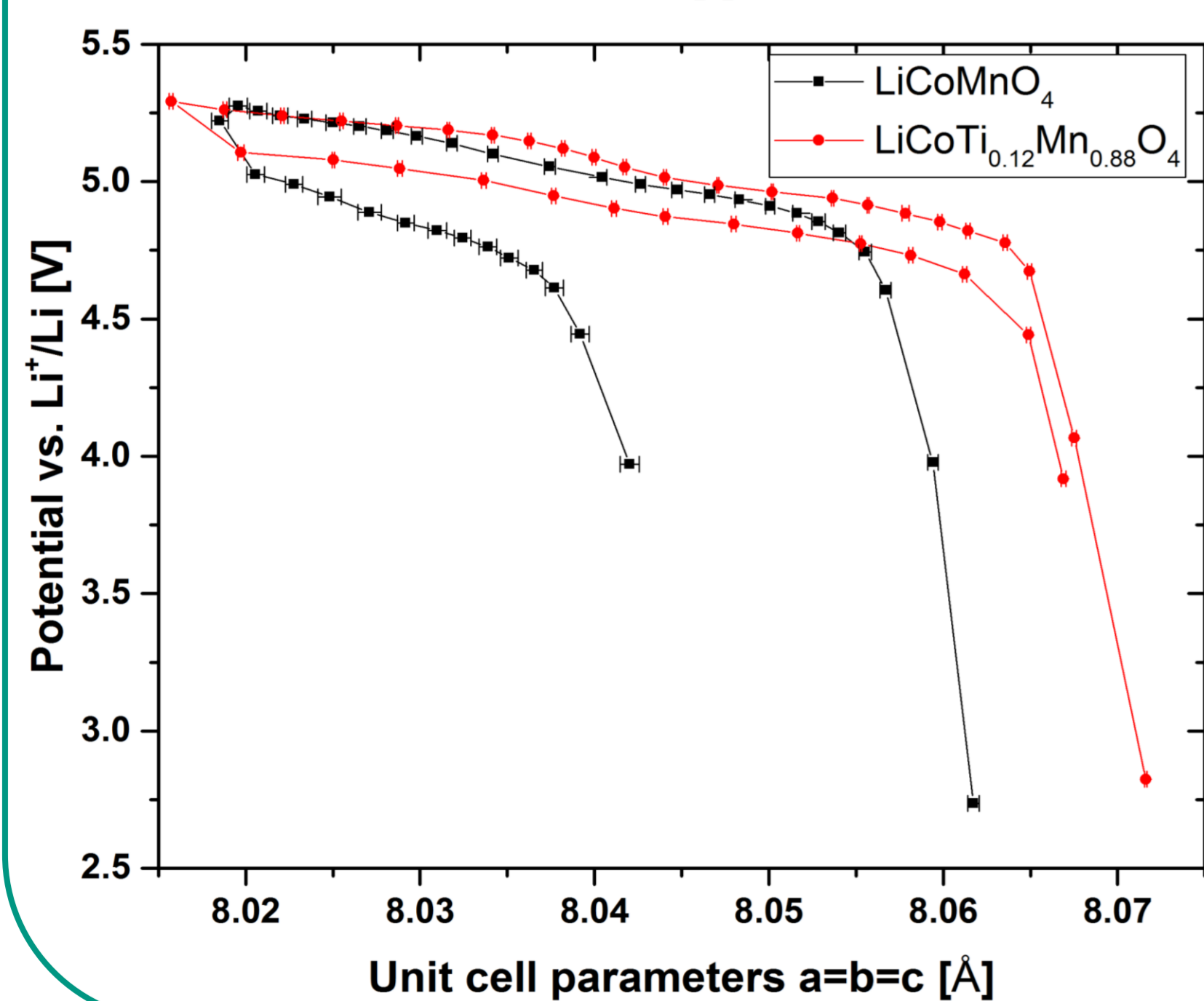


- Galvanostatic cycling against a Li-anode reveals a reversible discharge potential around 5.0 V with a spec. capacity of up to 100 mAh/g.
- CV measurements reveal a polarization of 0.1 V.

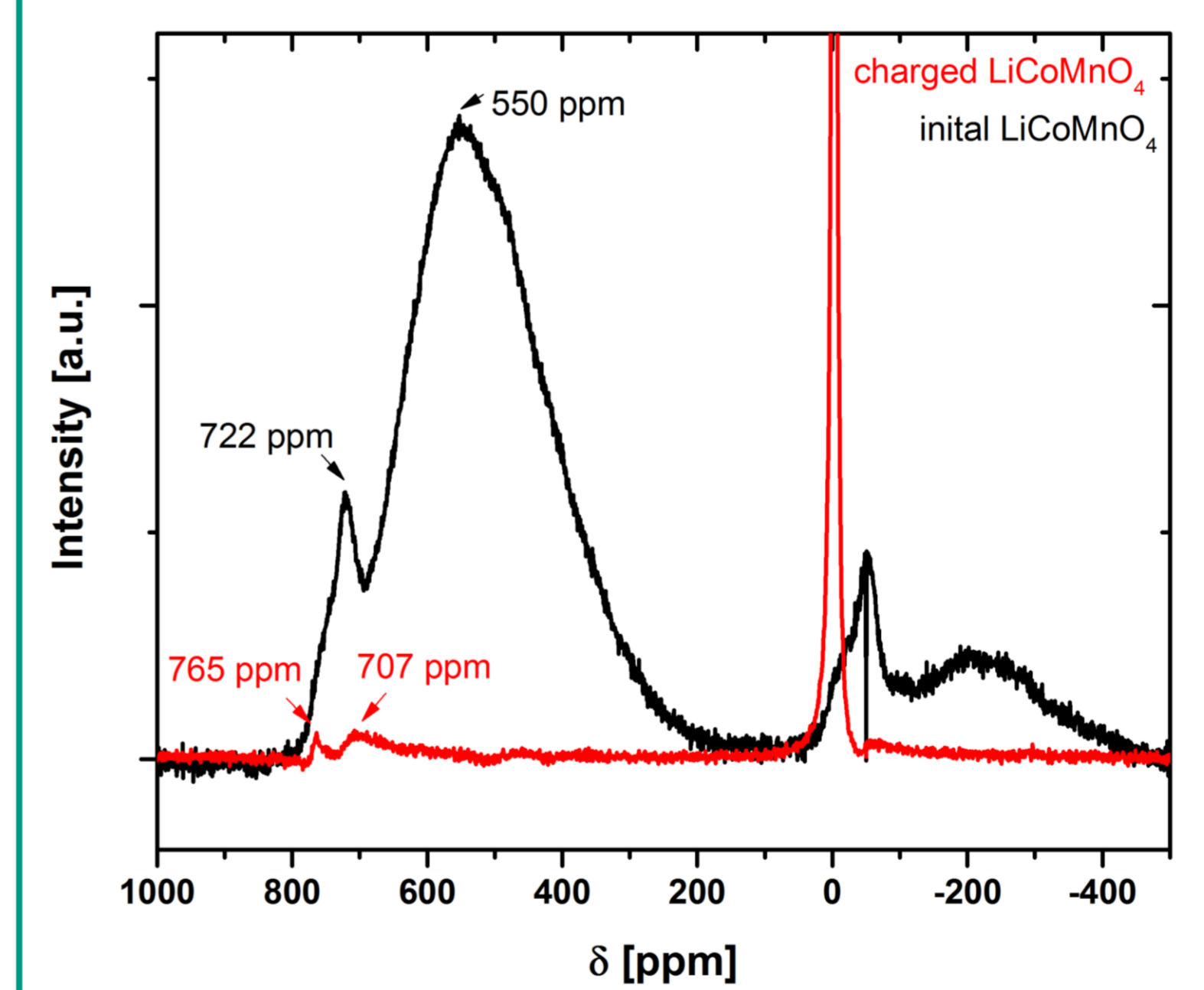
## Structural evolution during 1<sup>st</sup> cycle: *In situ* synchrotron powder diffraction



- $\text{LiCoMnO}_4$ -based cathode materials were investigated via *in situ* synchrotron powder diffraction. All materials show the same solid-solution behavior during cycling.
- The lattice parameters do not return to their initial state after one full cycle and indicates irreversible changes in the spinel lattice which in turn leads to capacity losses.
- Rietveld refinement results of the initial materials confirm the spinel phase (SG  $Fd\bar{3}m$ ) as the main phase. Small amounts of impurity can be identified as  $\text{Li}_2\text{MnO}_3$  (SG:  $C2/m$ ).



## <sup>7</sup>Li-MAS NMR



<sup>7</sup>Li-MAS-NMR of  $\text{LiCoMnO}_4$  shows 2 separated peaks. After the charge process (delithiation), the intensity of NMR-signal is drastically reduced and the peaks are slightly shifted.

## Acknowledgment

Financial support from the Federal Ministry of Education and Research (BMBF) within the DESIREE project, grant no. 03SF0477B is gratefully acknowledged.

## Summary and conclusions

- $\text{Ti}^{4+}$ -substituted  $\text{LiCoTi}_x\text{Mn}_{1-x}\text{O}_4$  ( $0.04 \leq x \leq 0.16$ ) can be successfully synthesized by a sol-gel method.
- Electrochemical cycling reveals a reversible capacity above 5.0 V with and without Ti-substitution. No electrochemical activity regions corresponding to Ti are observed.
- The  $\text{LiCoTi}_x\text{Mn}_{1-x}\text{O}_4$  ( $0.04 \leq x \leq 0.16$ ) materials show a solid-solution mechanism of Li intercalation/deintercalation irrespective of the dopant amounts.