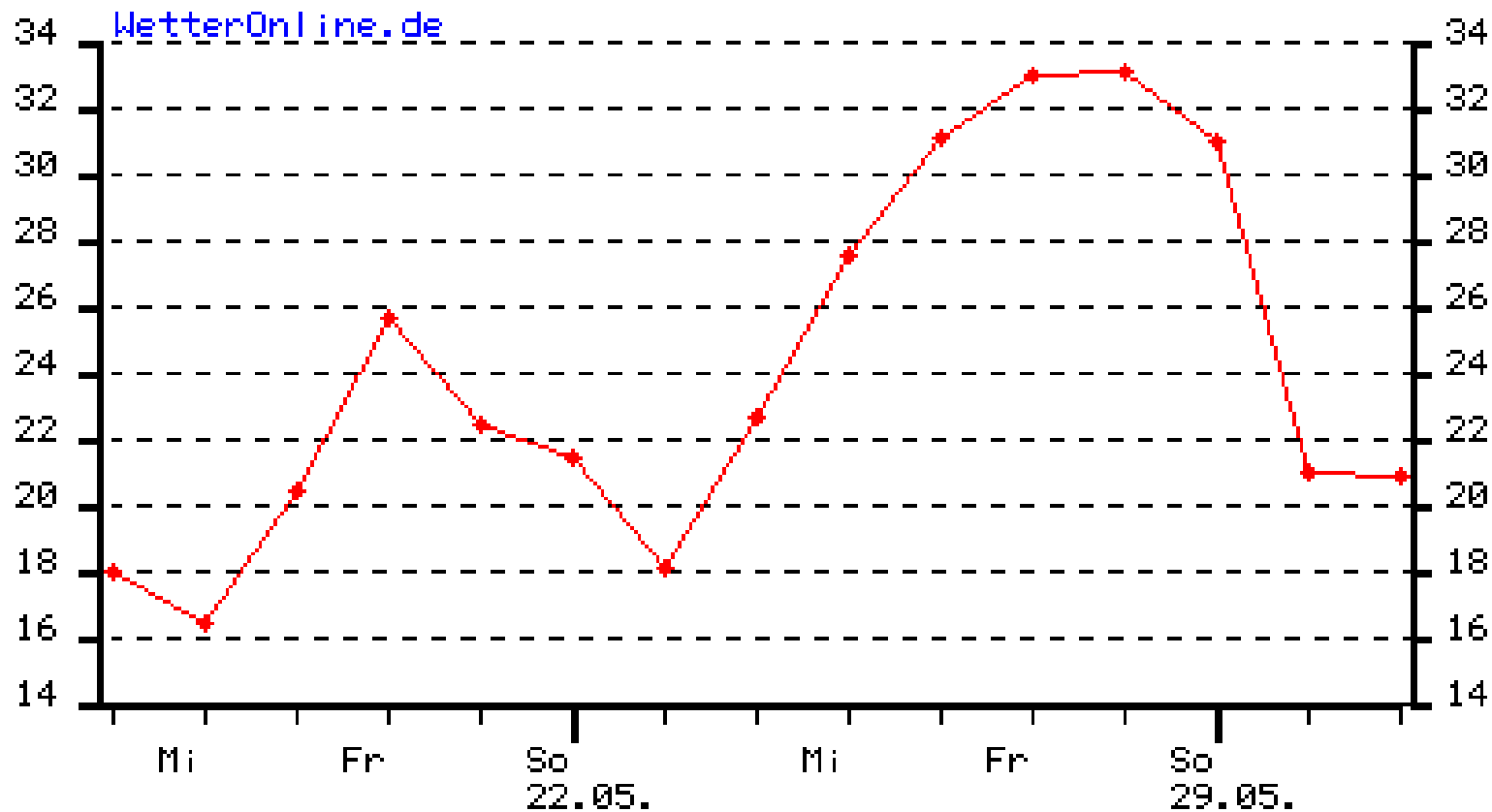




Differential Scanning (micro)Calorimetry





Outline

- DSC as method of thermal analysis (definitions, principle)
 - Instrumentation
 - DSC outcome (thermograms, determined parameters)
 - Applications
 - Pharmaceuticals
 - Proteins
 - Lipids
-



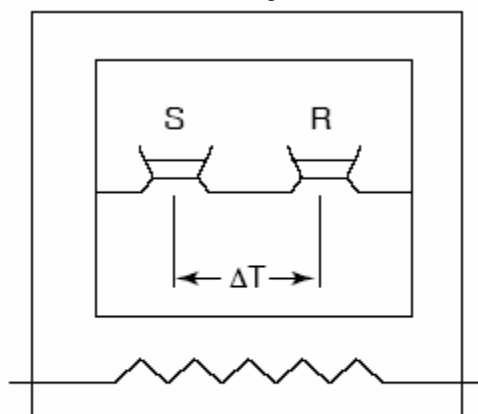
Thermoanalytical methods

<i>Method</i>	<i>Symbol</i>	<i>Measure</i>
Thermogravimetry	TG(TGA)	Mass (change)
Derivative Thermogravimetry	DTG	1 st derivative of mass change
Differential Thermal Analysis	DTA	Temperature difference vs. reference
Simultane Thermal Analysis STA (combination of DTA & TG)		see DTA & TG
Differential Scanning Calorimetry	DSC	Heat flow difference of sample vs. reference
Dynamic Differential Calorimetry	DDC	
Thermomechanical Analysis	TMA	Mechanical properties
Dynamic Mechanical Analysis	DMA	Visco-elastic properties
Dilatometry	-	Linear thermal expansion
Evolved Gas Analysis	EGA	Kind & amount of released gas
Thermomagnetometry	TM	Magnetic properties
Thermosonimetry	TS	Acoustic properties
Thermooptic Analysis	TOA	Optic properties



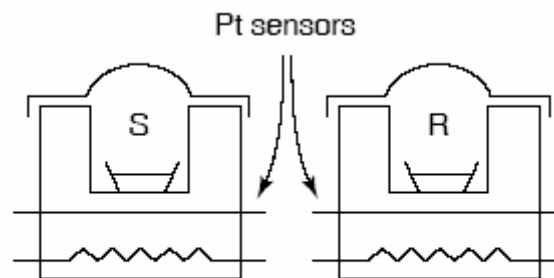
DTA vs. DSC

Differential Thermal Analysis



Single heater

Differential Scanning Calorimetry



Dual heaters



DSC: definitions

Heat flow = heat/time = dq/dt

Heating rate = temperature increase/time = dT/dt

Heat capacity = $C = \text{heat flow/heating rate} = dq/dT$

Adiabatic conditions: $C_p = (dq/dT)_p = dH/dT$

$$\Delta C_p = (C_p)_{\text{sample}} - (C_p)_{\text{reference}}$$

$$\Delta dH/dt = (dH/dt)_{\text{sample}} - (dH/dt)_{\text{reference}}$$

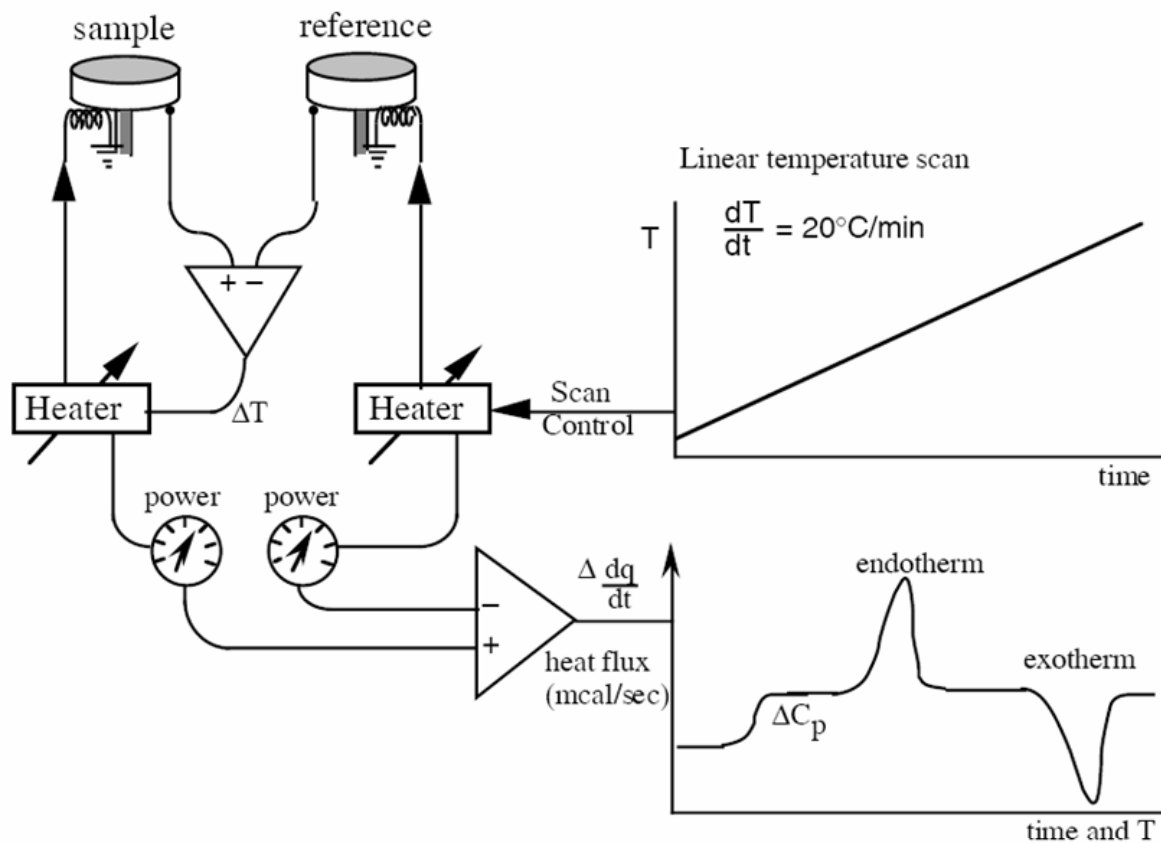


Instrumentation



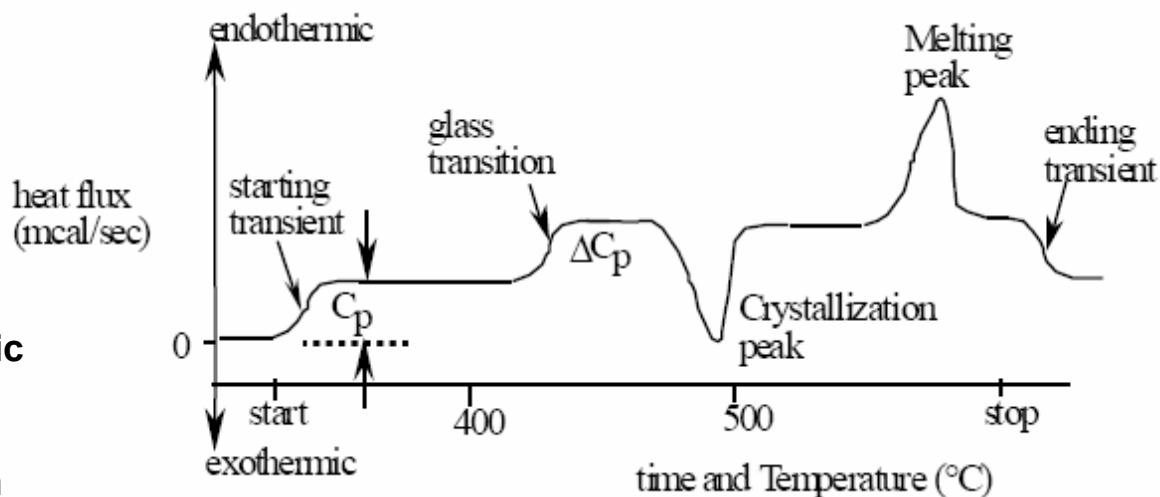


DSC output





Transitions observed in DSC



Endothermic

- Fusion
- Vaporisation
- Sublimation
- Desorption
- Desolvation
- Decomposition
- Reduction
- Degradation
- Glass transition
- Relaxation of glasses

Endothermic/Exothermic

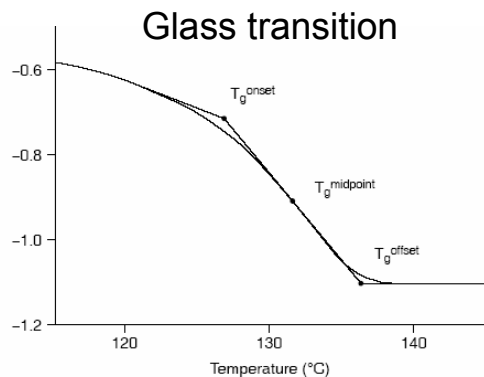
- Decomposition
- Degradation

Exothermic

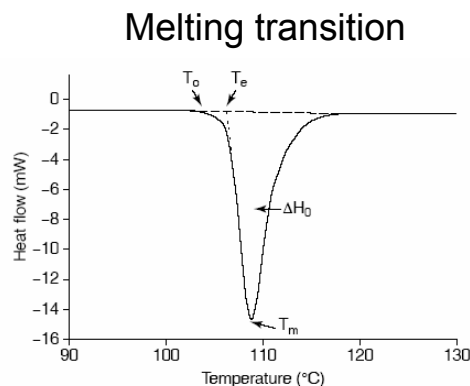
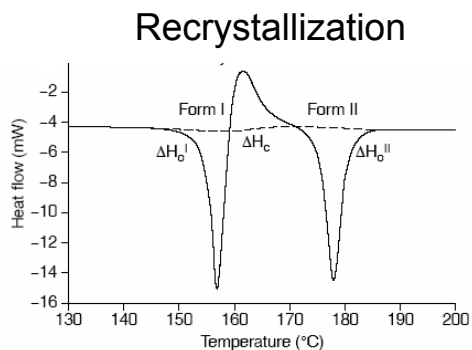
- Crystallisation
- Condensation
- Solidification
- Adsorption
- Chemisorption
- Solvation
- Decomposition
- Oxidation
- Curing of resins



DSC parameters



- T_g – glass transition temperature
- T_c – recrystallisation exotherm temperature
- T_m – melting endotherm temperature
- T_o – transition onset
- T_e – transition extrapolated onset
- $\Delta T_{1/2}$ – sharpness of the transition
- ΔH_c – recrystallization enthalpy
- ΔH_o – melting enthalpy





DSC parameters

T_m – transition maximum
 $\Delta T_{1/2}$ – sharpness of the transition
 ΔH_{cal} – transition enthalpy

ΔH_{cal} – directly measured

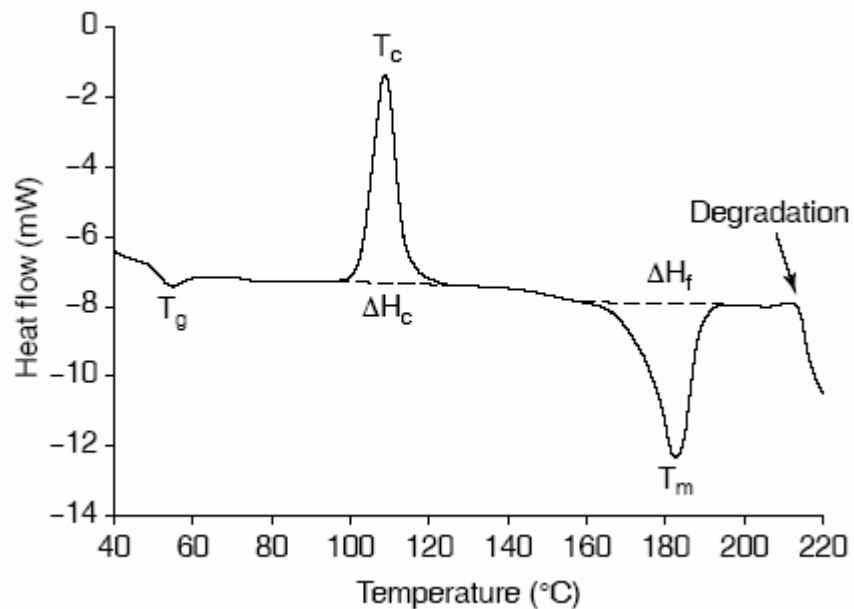
$$@T_m \Delta G=0 \rightarrow \Delta S = \Delta H_{cal} / T_m$$

$$\Delta H_{vH} \sim 4RT_m^2 / \Delta T_{1/2}$$

Cooperative Unit (**CU**) $\sim \Delta H_{vH} / \Delta H_{cal}$



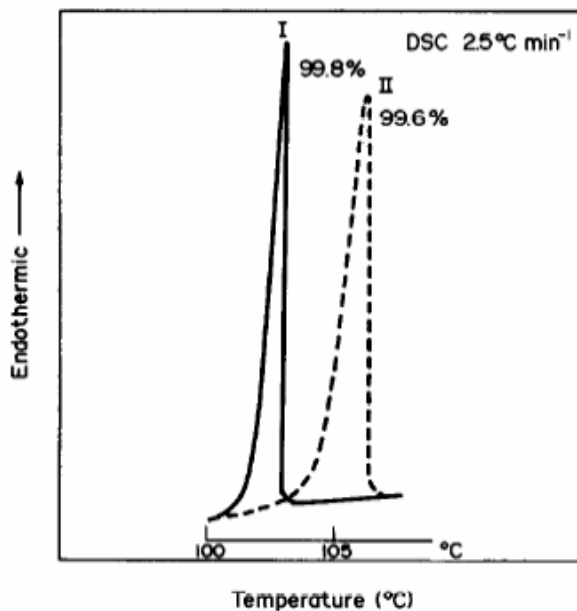
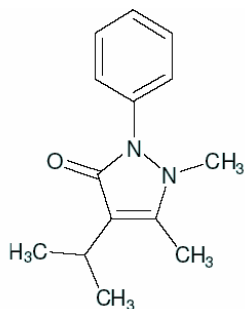
DSC of glucose



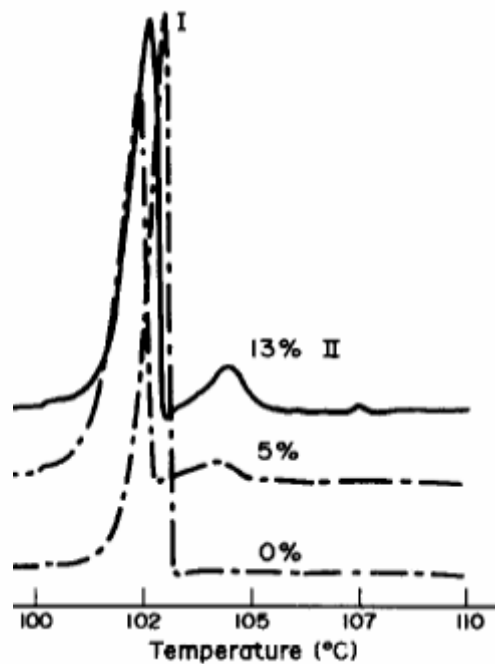
← Endothermic/Exothermic →



DSC of prophenazone

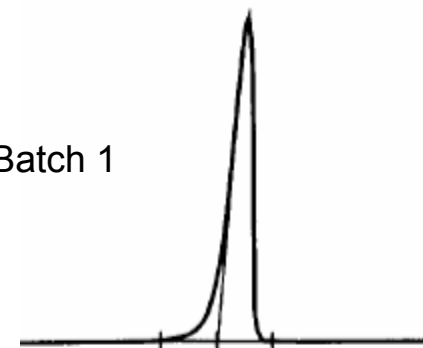


Different crystalites: I & II

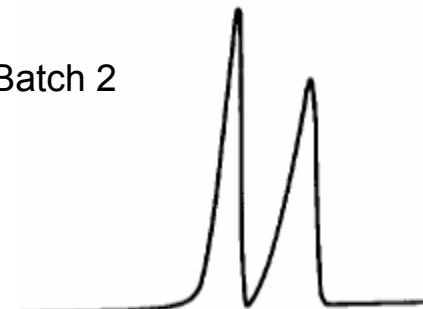


Different batches

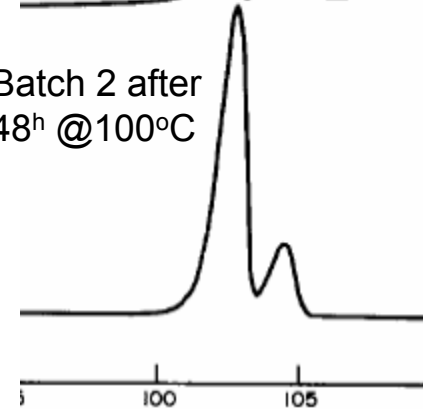
Batch 1



Batch 2



Batch 2 after
48^h @100°C





DSC of pharmaceuticals

Characterization of single components

Purity

Quantification of crystallinity

Polymorphism

Degradation, decomposition, stability determination

Glass transitions

Characterization of multicomponent mixtures

Drug-excipient compatibility

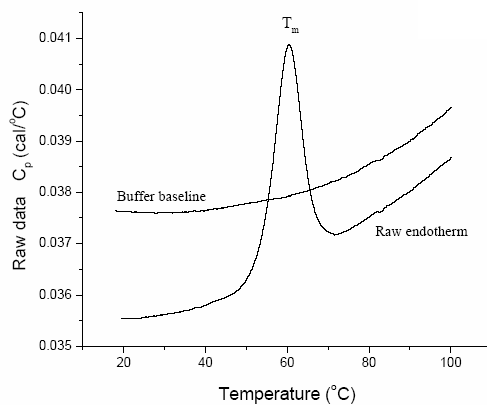
Glass transitions in mixtures

Water & hydrates

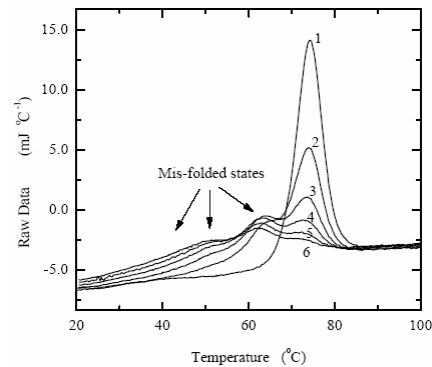
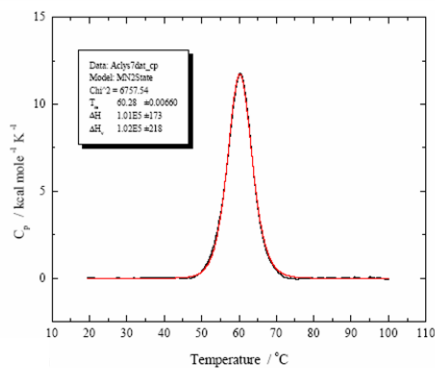
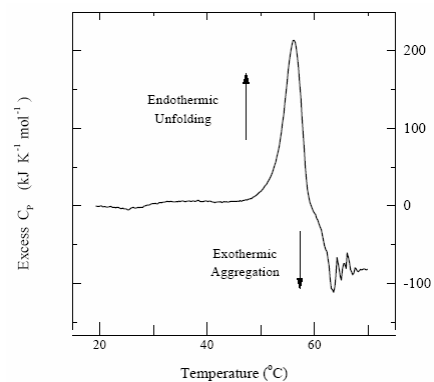


DSC of proteins: denaturation

„ideal case“



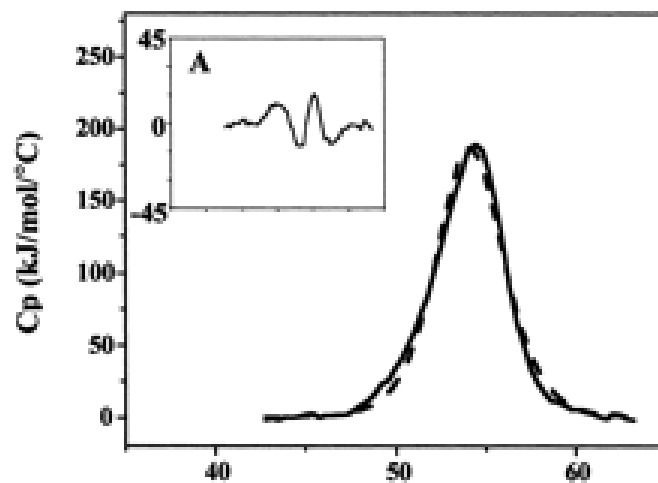
„real case“



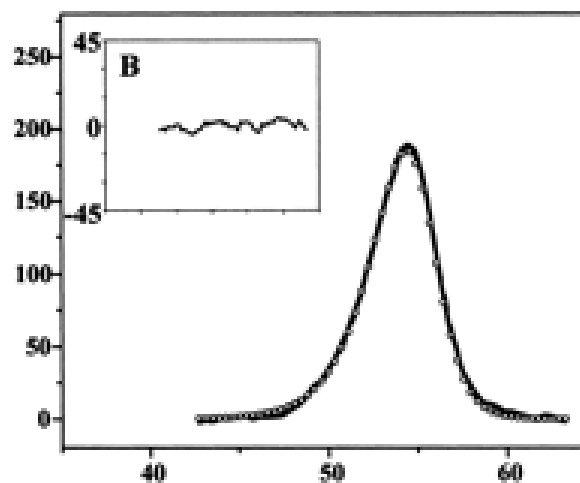
2-state model: $F \rightleftharpoons U$



DSC of NAD⁺ synthetase unfolding



$D_f \rightleftharpoons 2U$



$D_f \rightleftharpoons 2M_f \rightleftharpoons 2U$



DSC of proteins: protein stability

T_m is temperature midpoint
-Related to protein stability

ΔC_p almost always positive
- Denaturated protein has higher heat capacity
(hydration of the solvent exposed side chains)

Denaturated proteins susceptible to
-Proteolysis
- Oxidation
- Deamidation
- Aggregation

Factors responsible for folding and stability
-pH, salt, temperature, H-bonding
non-covalent interactions, conformational entropy



DSC of ligand binding

If ligand binds preferentially

- Protein unfolding less favourable as [L] increases
- New T_m occurs at higher temperature than in the absence of ligand (T_o)

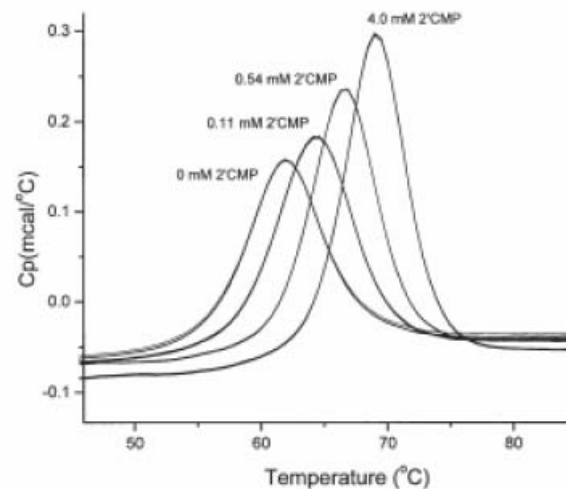
Binding constant (K_B) may be estimated from

- T_m , T_o , ΔH_{cal} and [L]

Used to study protein interactions with

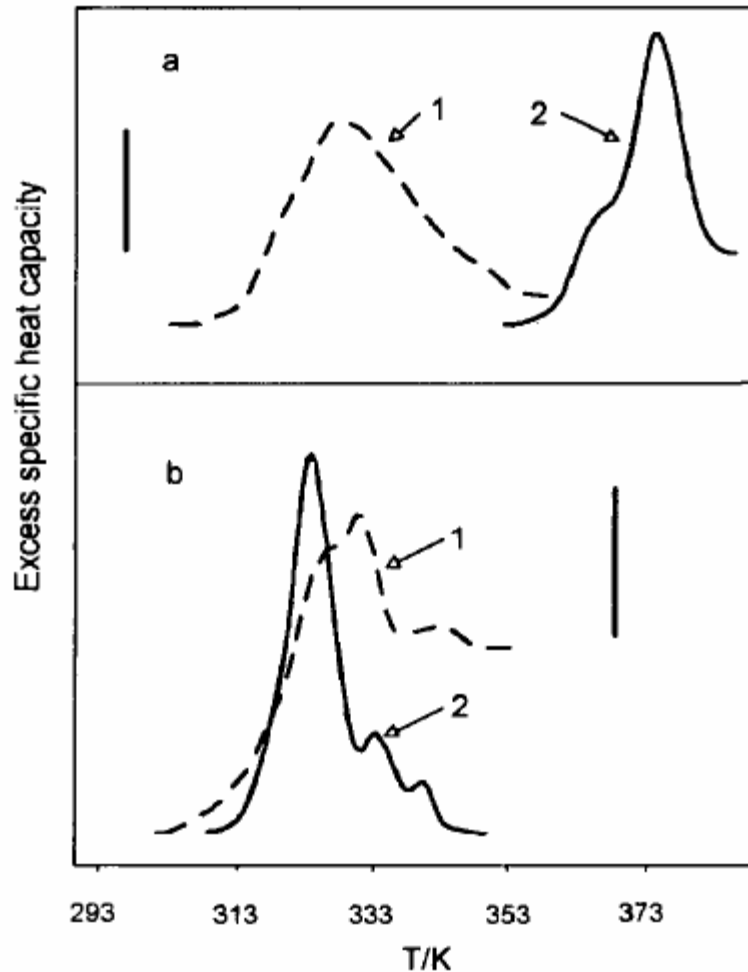
- small ligands – drugs, inhibitors, peptides

RNase A + 2'CMP





DSC of Troponin complex interaction with Ca^{2+}



Troponin C:

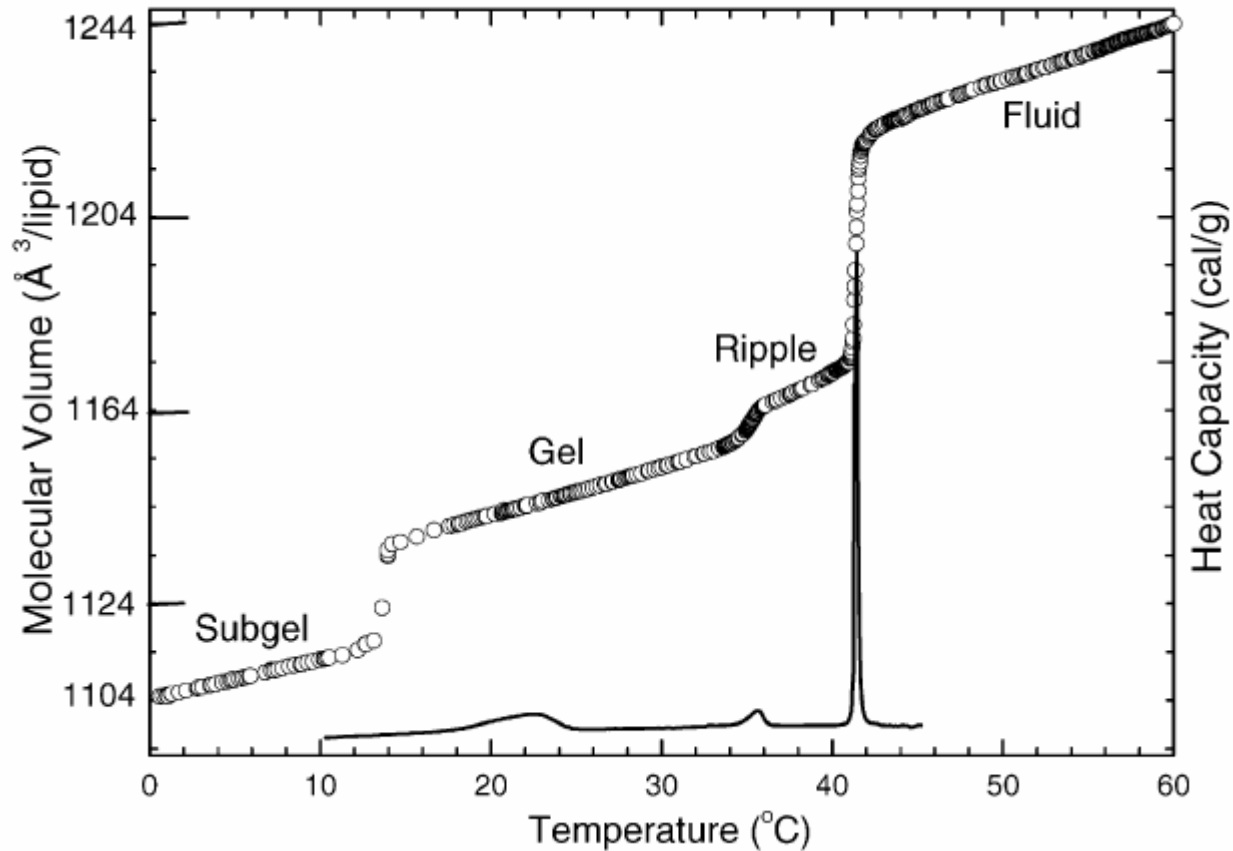
- 1 – no CaCl_2
- 2 – added CaCl_2

Complex of Troponins C/T/I:

- 1 – no CaCl_2
- 2 – added CaCl_2

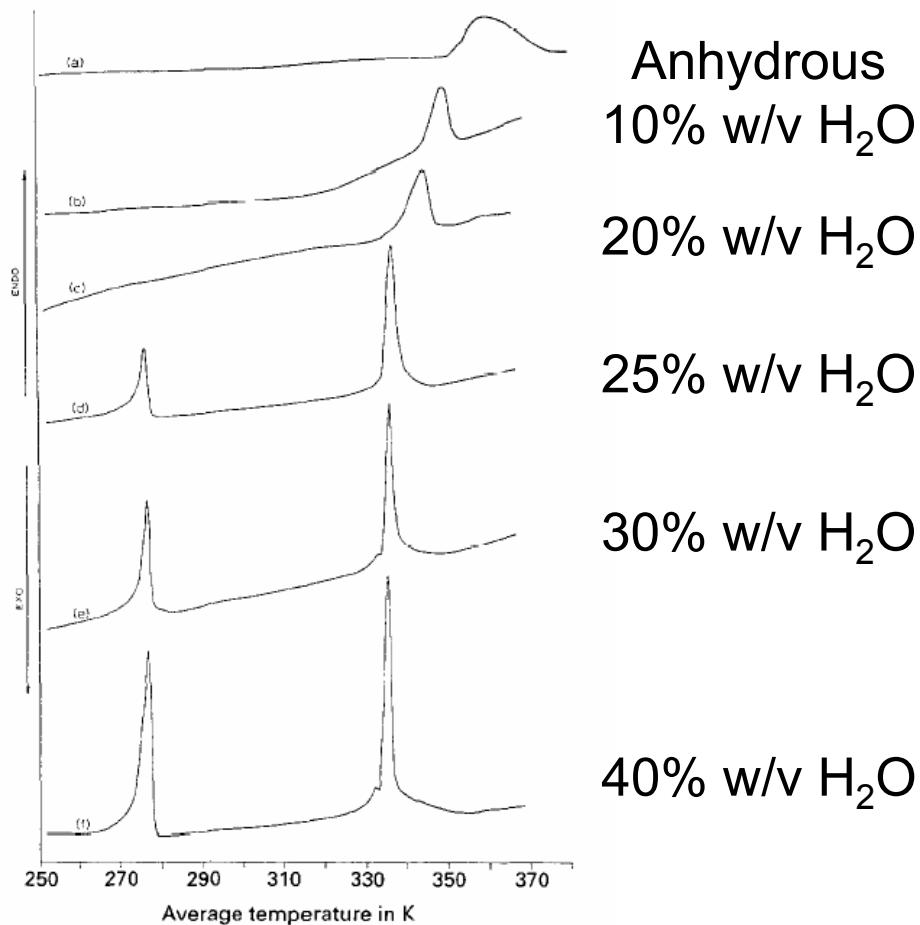


DSC of synthetic phospholipid dispersion



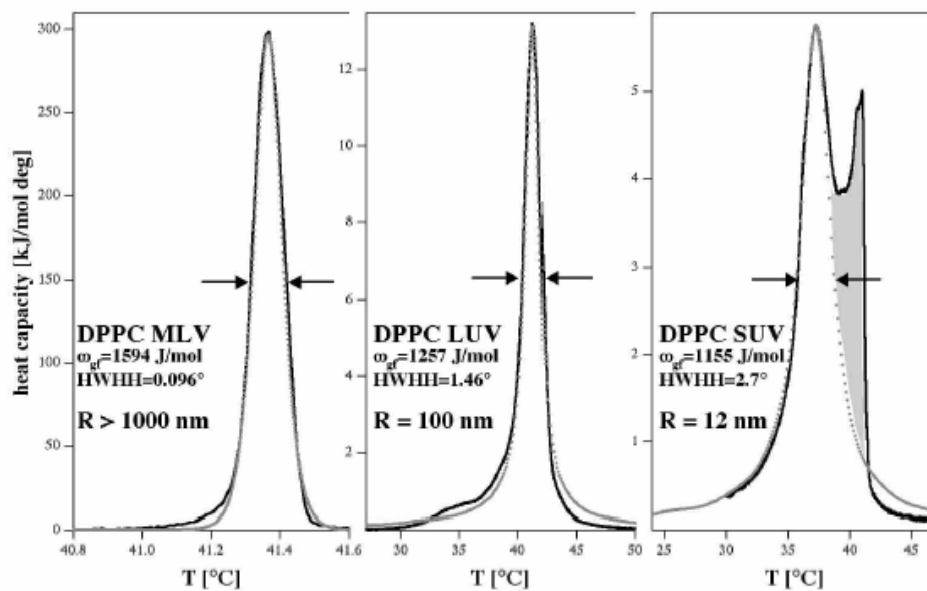


DSC of DSPC: hydration





DSC of phospholipids: lyosomes

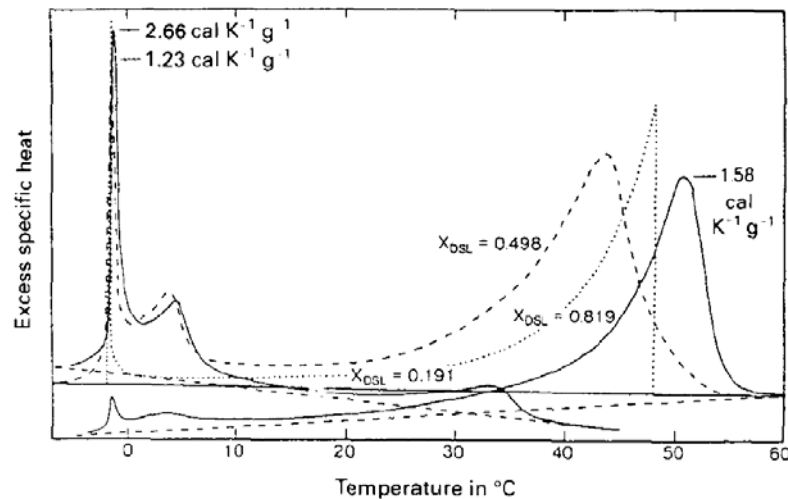
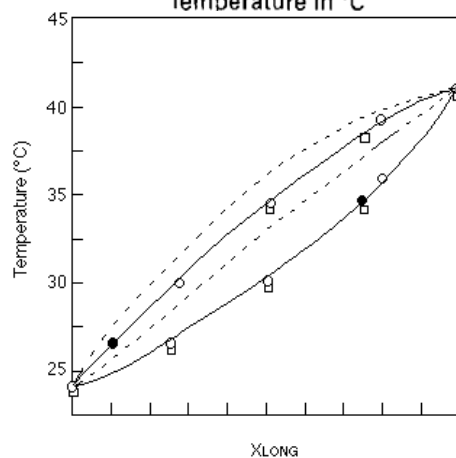
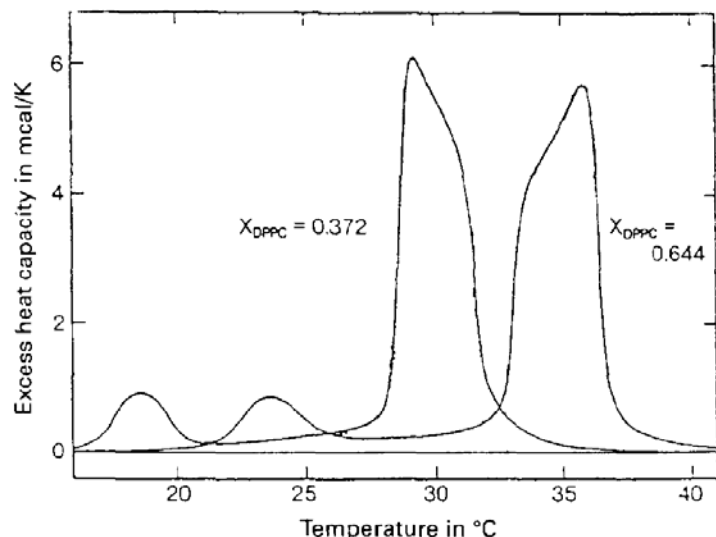


	DMPC		DPPC	
	Unsonicated	Sonicated	Unsonicated	Sonicated
Transition temperature/ $^\circ\text{C}$	23	12	41	32
Transition enthalpy/ kcal mol^{-1}	6.6	3.3	8.6	6.0
Transition entropy/ $\text{cal mol}^{-1} \text{ } ^\circ\text{C}^{-1}$	22	11	28	19

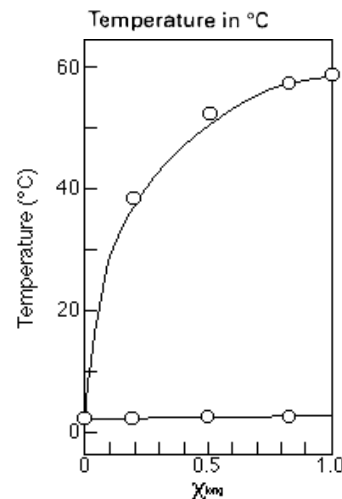


DSC of phospholipids: mixtures, phase diagrams

DPPC/DMPC



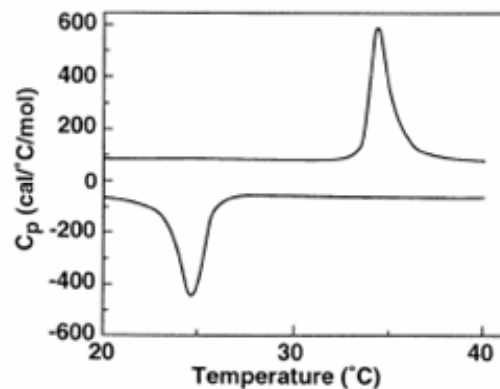
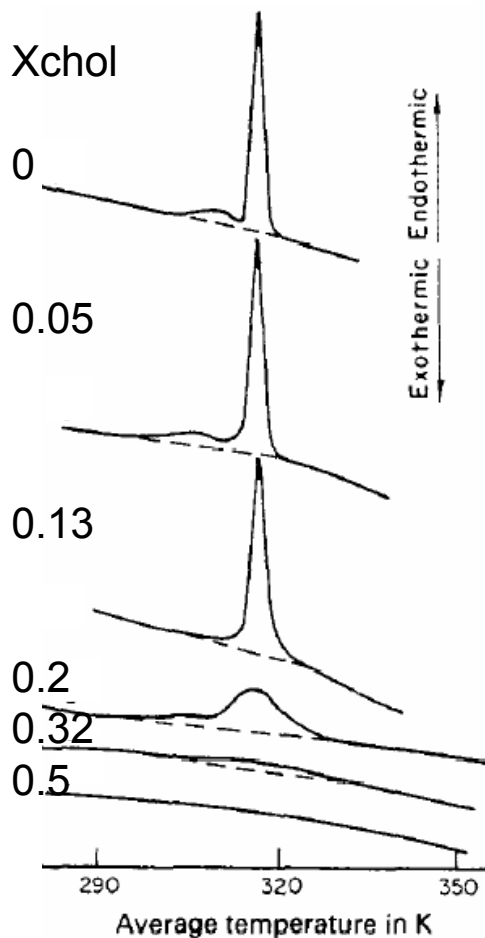
DLPC/DSPC





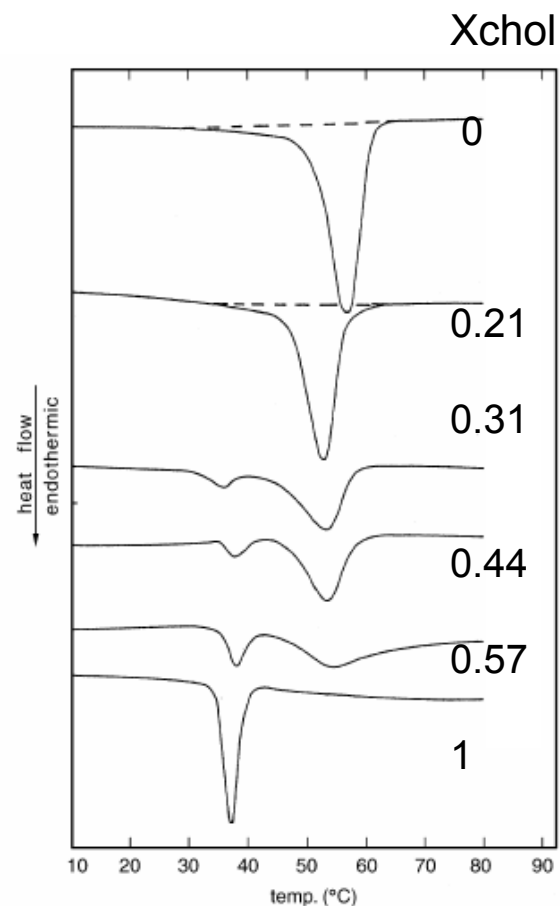
DSC of phospholipids: cholesterol

DPPC



DSC of cholesterol

DPPS





DSC of phospholipids: cholesterol

Onset of phase separation of cholesterol from phospholipid/cholesterol mixtures

Lipid	Charge	Presence of hydrogen bonds	Hydrocarbon chain length	Number of double bonds/hydro-carbon chain	Onset of cholesterol phase separation X(chol) ^a	
					Gel phase	I.c. phase
DMPC	(±)	no	14	0		0.45 0.5
DPPC	(±)	no	16	0	0.5	
DSPC	(±)	no	18	0	>0.5	>0.5
SAPC	(±)	no	18,20	0,4		0.5
DAPC	(±)	no	20	4		0.17
Egg PC ^b	(±)	no	variable	variable		0.45–0.55
DDPC	(±)	no	22	6		0.10
DMPE	(±)	yes	14	0	0.43	0.43
DEPE	(±)	yes	18	1 ^c		0.35–0.4
DOPE	(±)	yes	18	1		0.33 ^d
Egg PE ^b	(±)	yes	variable	variable		0.35–0.4
DMPG	(–)	no	14	0	0.4	0.44
DPPG	(–)	no	16	0	0.38	0.37
DSPG	(–)	no	18	0	0.32	0.35
POPG	(–)	no	16,18	0,1		0.45
DMPS	(–)	yes	14	0	0.33	0.37
DPPS	(–)	yes	16	0	0.3	
POPS	(–)	yes	16,18	0,1		0.36
SOPS	(–)	yes	16,18	0,1		0.2
PS ^{b,c}	(–)	yes	variable	variable		0.3



DSC of phospholipids with small molecules (peptides?)

Type A: down shift of T_m , increase in $T_{1/2}$ constant ΔH_{cal}
Partially buried in the bilayer, interact with C_2 - C_8 methylenes

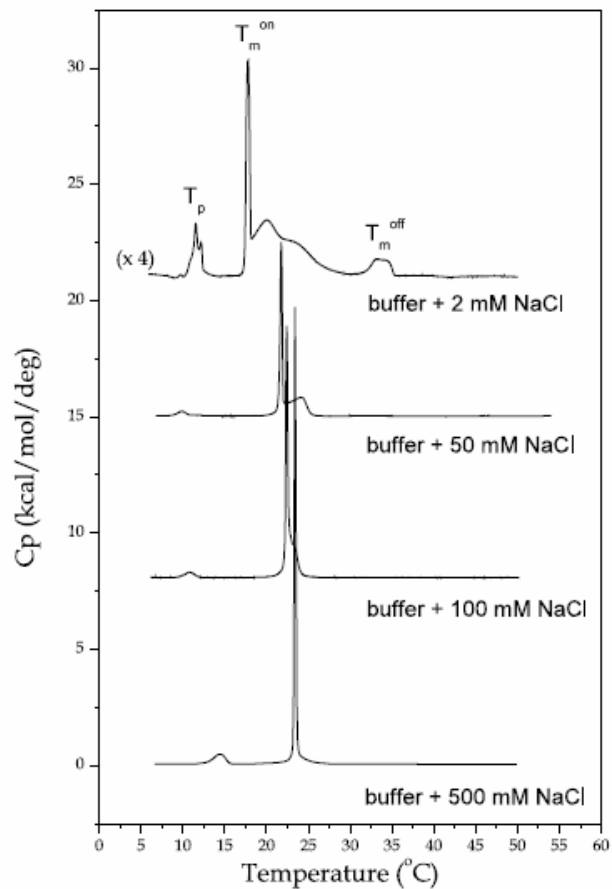
Type B: appear of the shoulder, that grows into new peak
with [additive], constant ΔH_{cal} of both peaks
Located on the interface, interact with glycerol backbone

Type C: down shift of T_m , constant $T_{1/2}$, ΔH_{cal}
Located in the center of bilayer, interact with C_9 - C_{16} methylenes

Type D: discrete new peak that grows in exence of the previous,
Located on the surface, interact with the headgroups

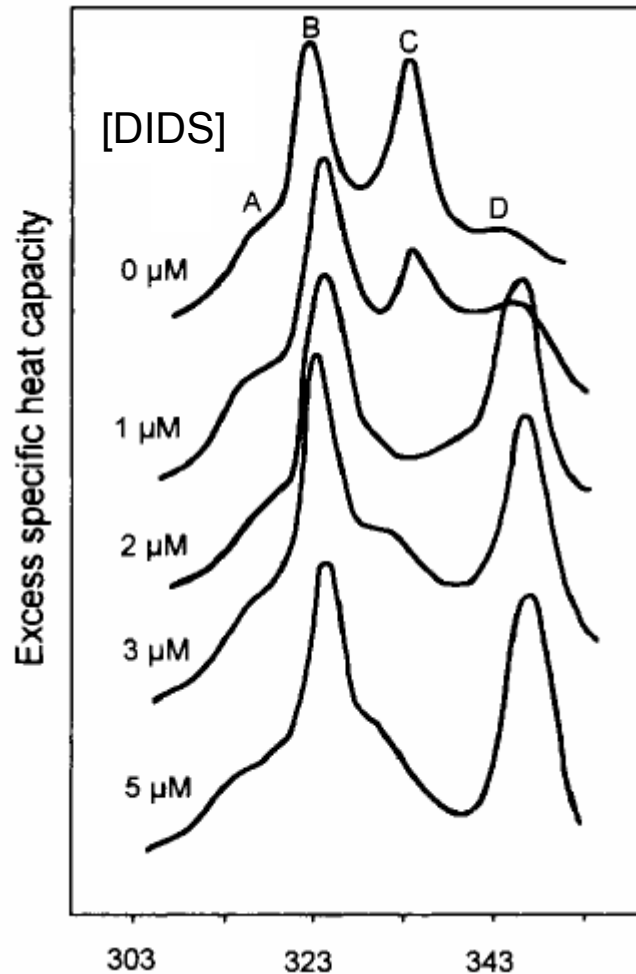


DSC of phospholipids: DMPG





DSC of a biological membrane



Erythrocyte ghosts:

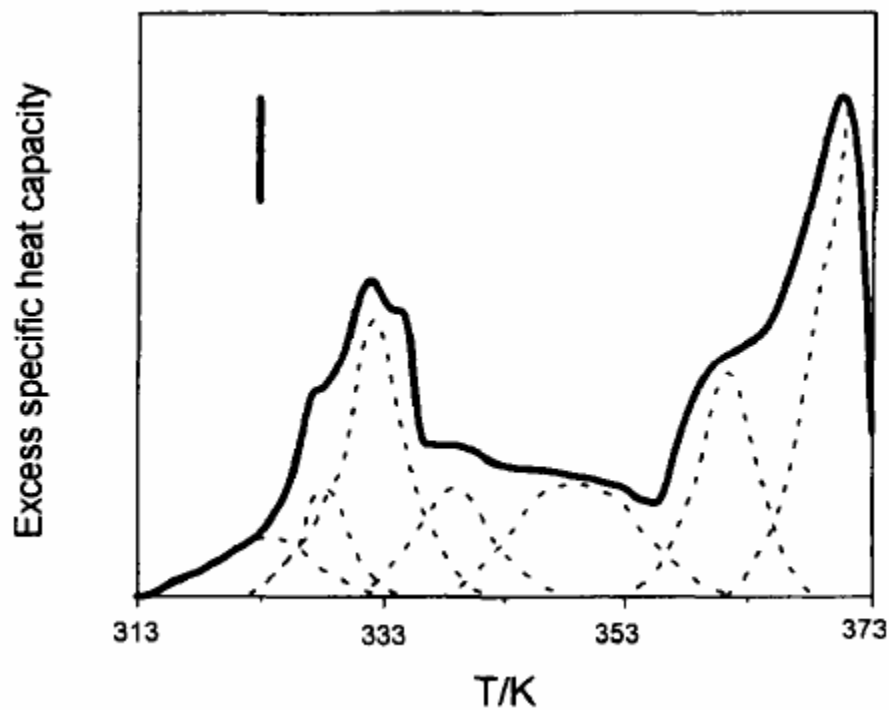
A: Spectrin

B: Extramembrane domains

C,D: Band 3



DSC of a cell



Cock sperm suspension



Thanks!
