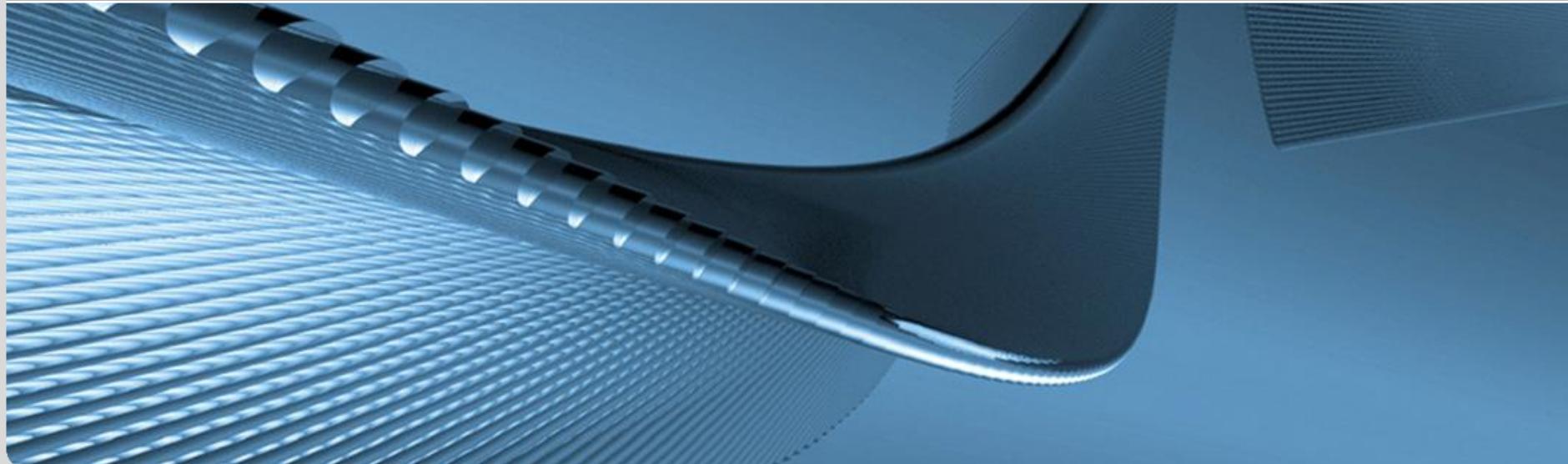


# International tungsten development efforts and the ~~German~~ gas cooled divertor design

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# CONTENTS (→ PRE-DEFINED BY LANCE)

- **Introduction (or focusing on the not quite hopeless problems)**
- **Gas cooled divertor designs**
- **Identification of issues  
(fabrication, properties, environment, irradiation effects)**
- **Material system mapped to design window  
(Where do we look good, where are still gaps?)**
- **Summary and path forward**

# INTRODUCTION

The life time of a power plant divertor is TWO years – hopefully!

## High heat flux

- High operating temperatures (peaks > 1800°C)
- Microstructural stability, aging, ...
- Thermo-shocks, cyclic loading, fatigue, ...

## Heavy ion bombardment

- Sputtering, cracking/fracturing
- Surface interactions and modifications
- Microstructural changes

## Neutron load 15 dpa/year (in W)

- 30 dpa (in W) until end of service
- Transmutation effects (Re, Os)
- Swelling (order of magnitude: 3% for pure W)

**QUESTION:** What can we do about this? How can we improve **ARMOR** materials?

**ANSWER:** There is not much we can do now! We have to live with most properties and effects. A final assessment would require real in-service conditions.

# “REASONABLE” ALLOYING ELEMENTS FOR W

M <sub>2</sub> W		MW	MW <sub>2</sub>	M <sub>4</sub> W			
Be	Mg	B	C M <sub>1-x</sub> W	Al		Y	La
Ti	V	MW <sub>3</sub>	Mn	MW, M <sub>7</sub> W <sub>6</sub>	M <sub>7</sub> W <sub>6</sub>	MW	
>3wt.% >300°C		Cr		Fe	Co	Ni	Cu
MW <sub>2</sub>				Ru	Rh	M <sub>3</sub> W	
Zr	Nb	Mo		< 3 wt. %	< 2 wt. %	Pd	Ag
MW <sub>2</sub>			MW		MW	MW	Cd
Hf	Ta		Re < 26 %	Os < 5 %	Ir	Pt	Au

Insoluble

Intermetallic Phases

Line Compounds

Solid Solution

## Pure Tungsten

Grain Stabilized Tungsten  
„ODS Tungsten“

„Heavy Metals“  
(Two Phases)

Potassium  
Doping

Oxides &  
Carbides

e.g. WVM, WVMW  
→ Bulb Wire

- $\text{La}_2\text{O}_3$  (e.g. WL10, WL15, WL20)
- $\text{CeO}_2$  (e.g. WC20)
- $\text{ThO}_2$  (e.g. WT20)
- → Weld Electrodes
- $\text{Y}_2\text{O}_3$ ,  $\text{ZrO}_2$ ,  $\text{TiC}$ ,  $\text{HfC}$ , etc.

- W-Ni-Fe (e.g. Dense, Inermet)
- W-Cu
- → Functional Applications

- Not suitable for structural divertor applications**
- W-Re (<26%)  
→ only commercial alloy
  - W-V
  - W-Ta
  - W-Mo
  - W-Ti
  - (W-Nb)  
→ Even more brittle as pure tungsten !!!

# PRODUCTION ROUTES

## Powder Metallurgy

- Blending
- Pressing
- Sintering
- Hot Forming

- Blending with Binder
- Injection Molding
- Debinder/Sintering
- HIP

- Ball Milling
- Encapsulation
- HIP/Hot Forming

This is so far the only large-scale production route which could handle the 500 tons of W needed for one divertor !

+ Mass Production  
+ Near Net Shaped Products  
+ Heterogeneous Microstructure

→ S. Antusch, KIT  
→ J. Opschoor, ECN

+ Novel Structure  
+ Diverse  
+ Small Scale  
+ Production Route

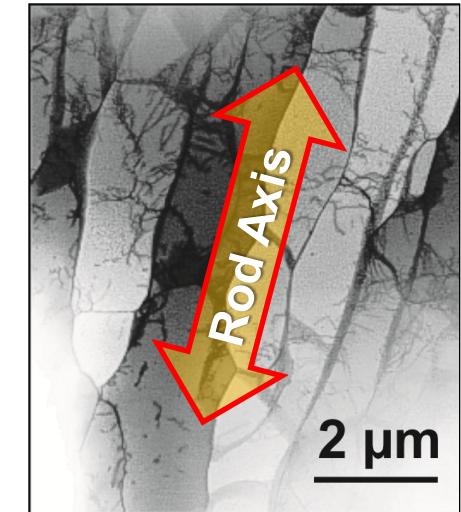
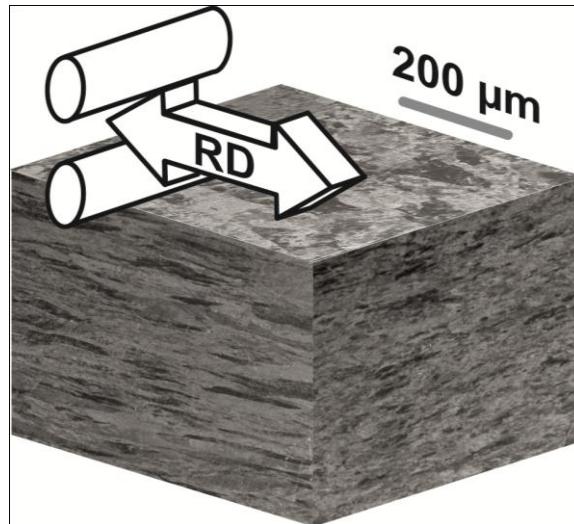
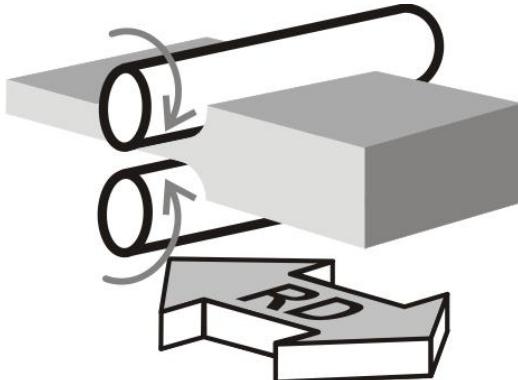
→ H. Kirushita, IMR  
→ N. Baluc, PSI  
→ A. Muñoz, CIEMAT

**Functional Applications**

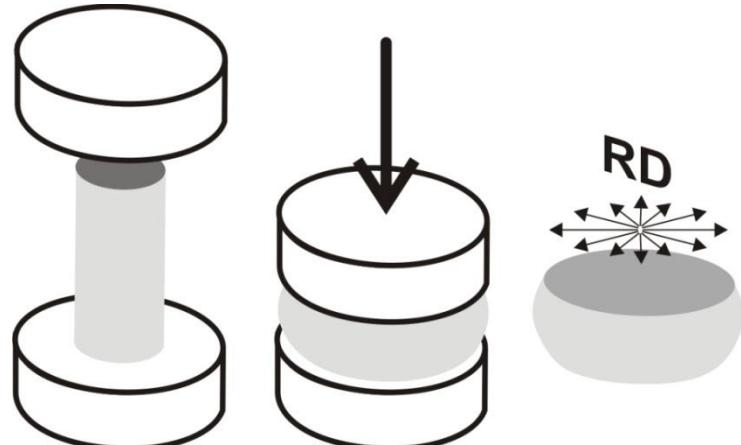
**Still Under Investigation**

# COMMERCIAL SEMI-FINISHED W PRODUCTS

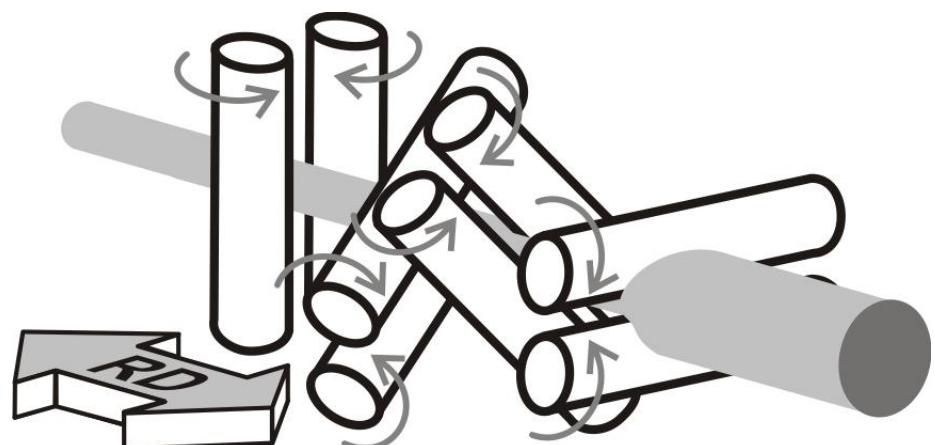
## Rolling Plates



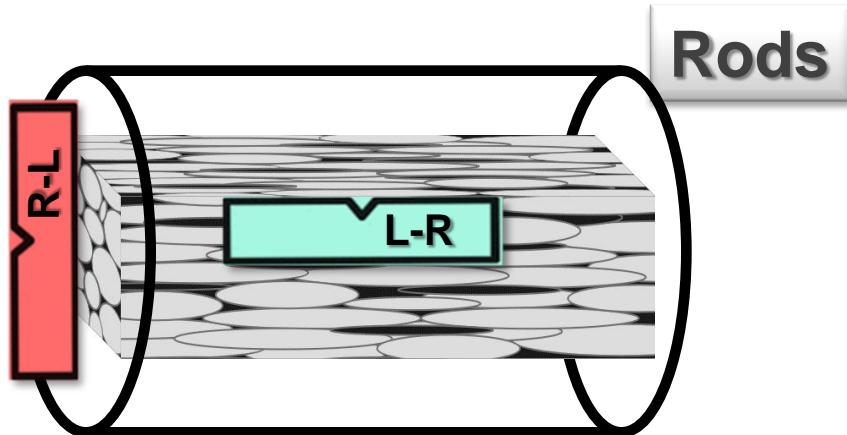
## Forging Round Blanks



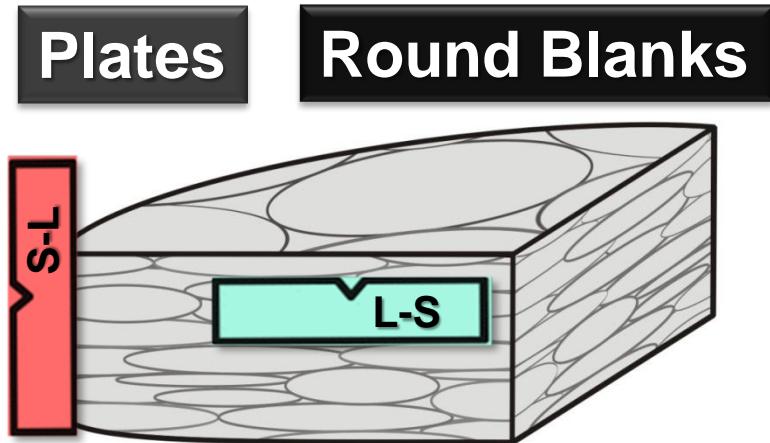
## Rolling/Swaging Rods



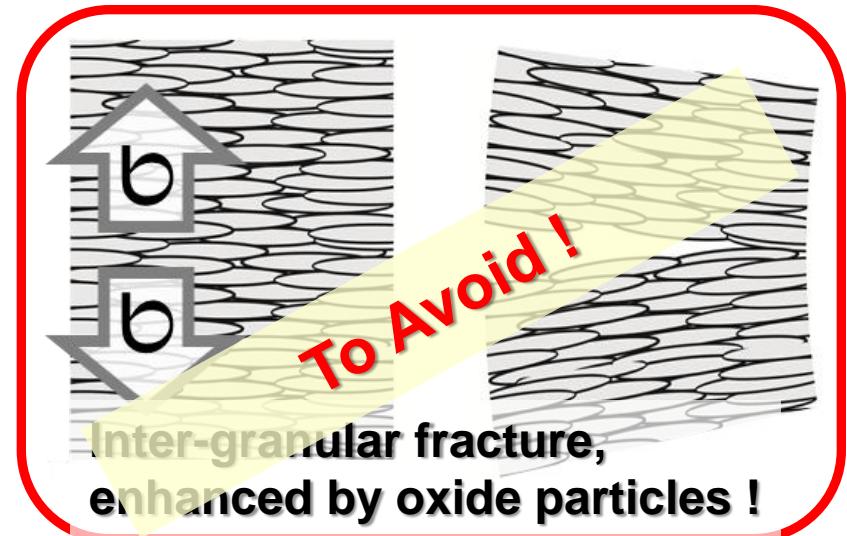
# MICROSTRUCTURE AND RELATED PROPERTIES



Bundle of „Fibres“



Stack of „Pancakes“



**So far, the best suitable tungsten materials for structural applications (divertor or other large scale components) are**

**Thin Plates, Thickness < 4 mm**

**Produced by Sintering (Hydrogen Atmosphere) and Cross-Rolling**

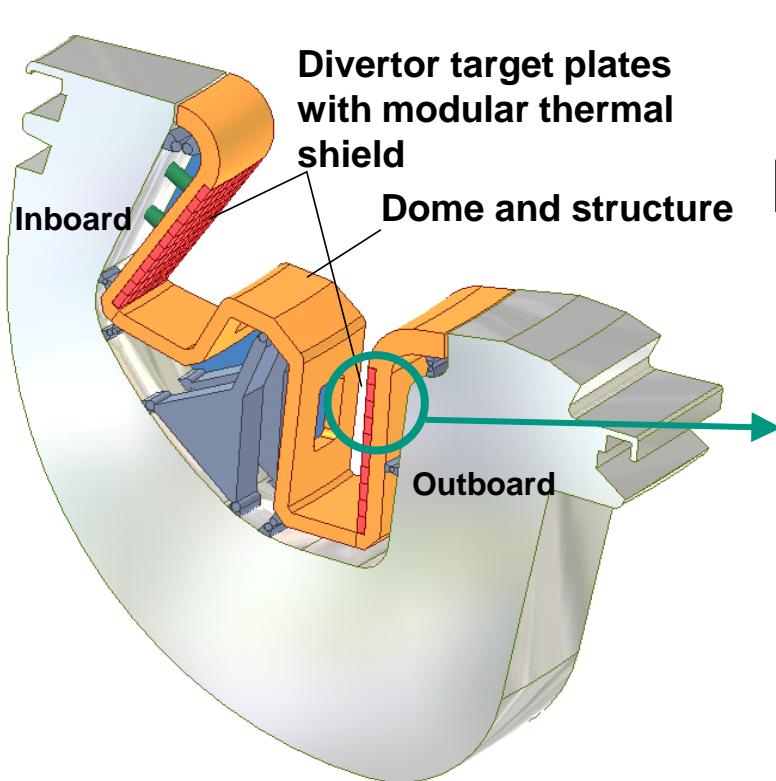
**Pure Tungsten (maybe small amounts of grain stabilizers, like  $\text{La}_2\text{O}_3$ )**

**Costs for 500 t → over 100 Mio. US \$**

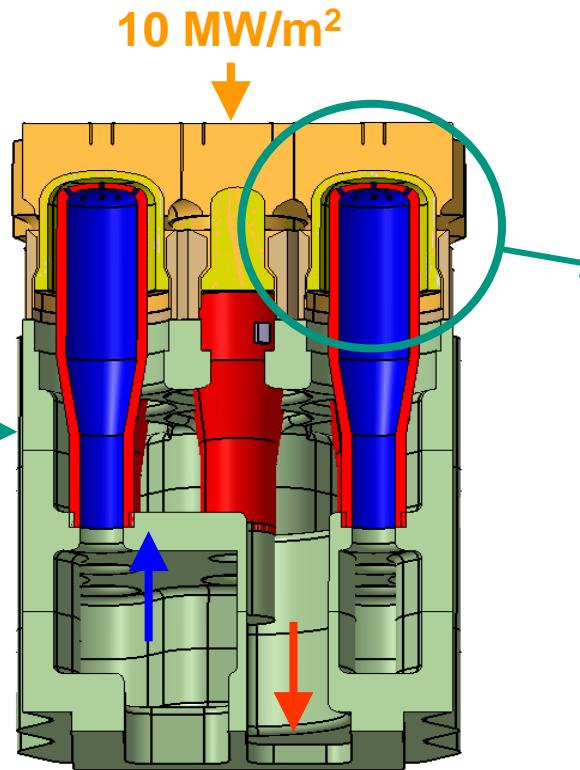
**based on PLANSEE online catalogue, September 2011, including discount**

# 2 HELIUM COOLED DIVERTOR DESIGNS

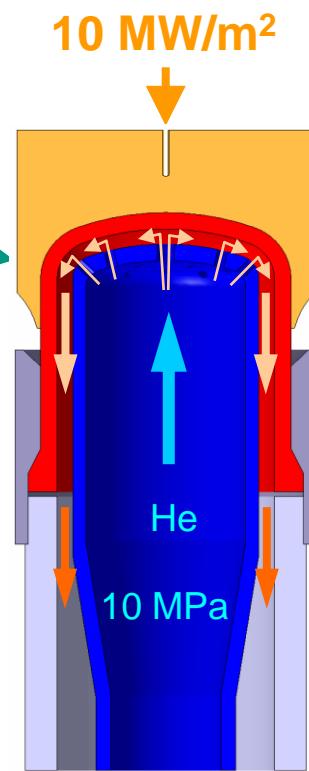
# FINGER DESIGN, JET COOLING



**Divertor  
Cassette**



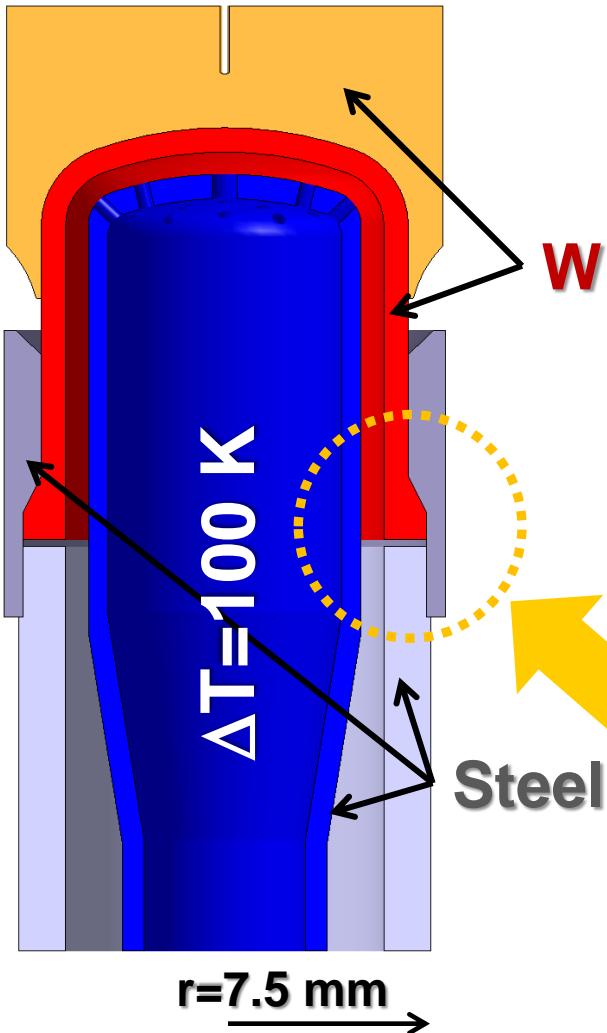
**9-Finger  
Module**



**Finger**

→ P. Norajitra, T. Ihli *et al.*, 2003-2009

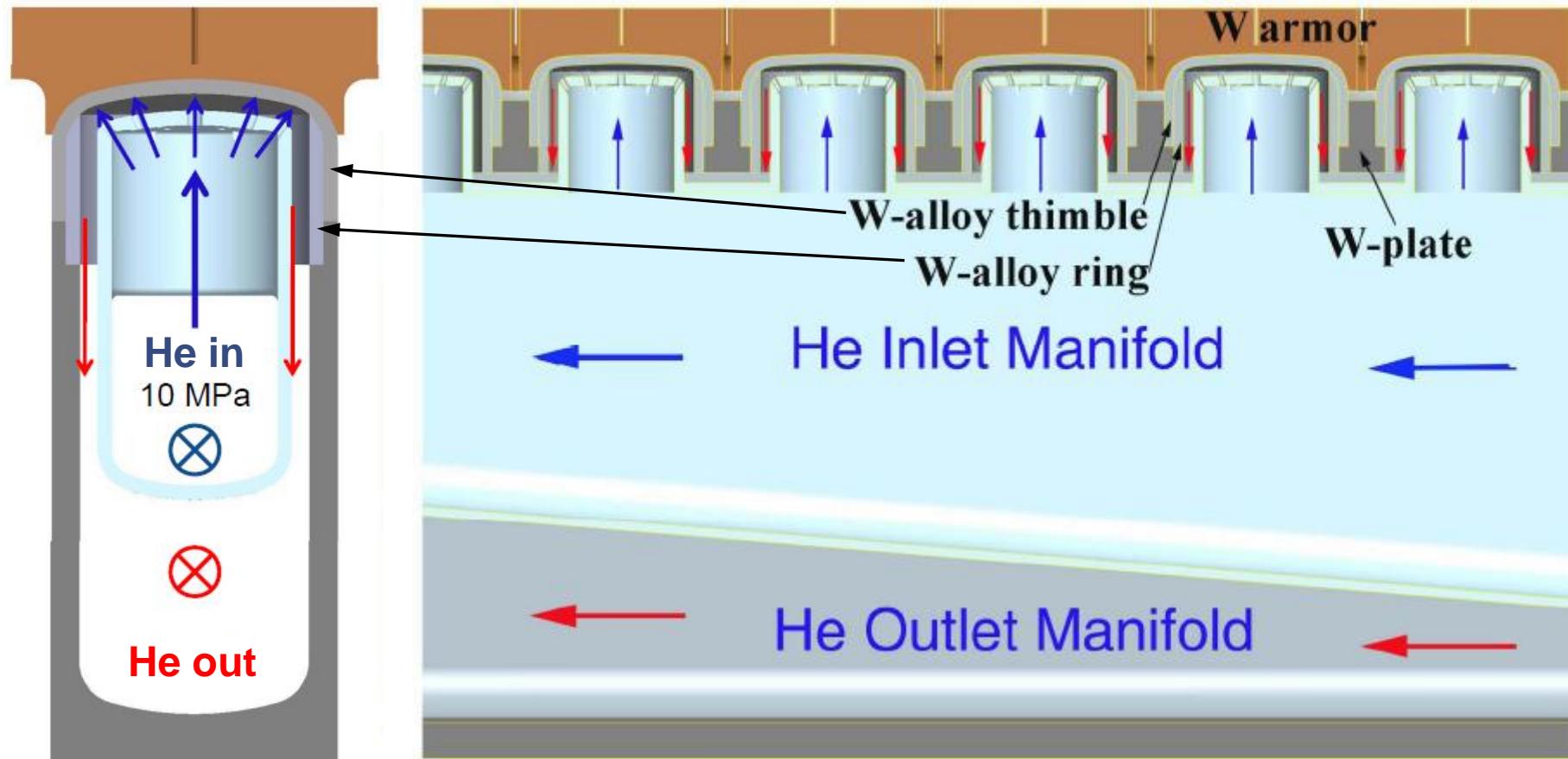
# FINGER DESIGN, JET COOLING



- + High heat flux:  $>10 \text{ MW/m}^2$
- + Small Size, Thin Walls
- Large Numbers:  $\sim 500\,000$
- Joints between W and Steel

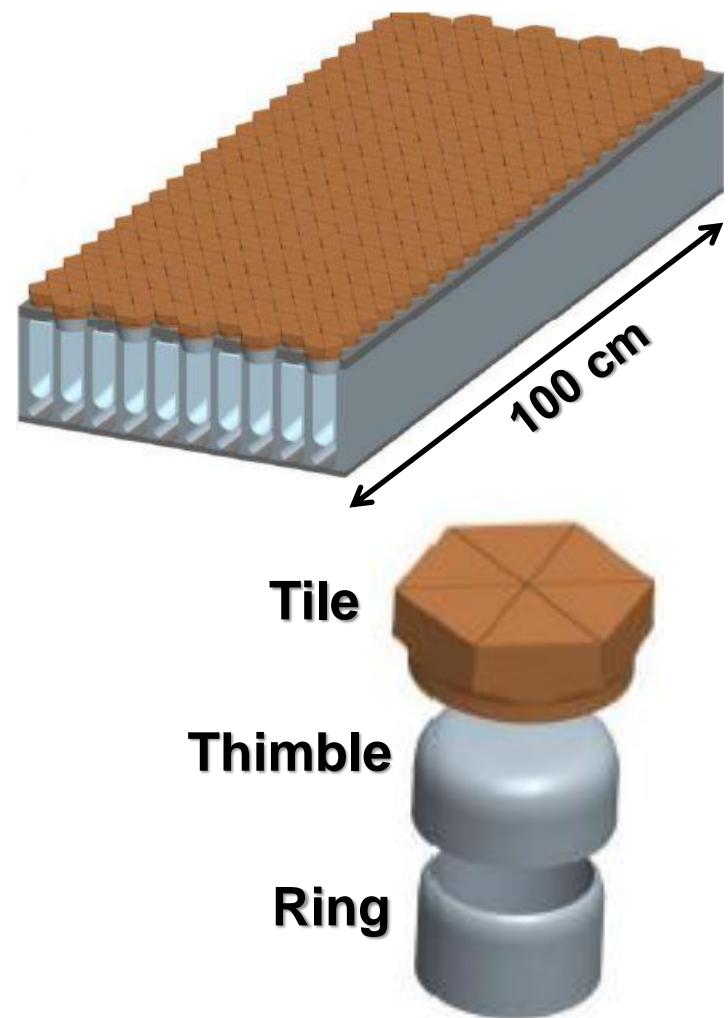
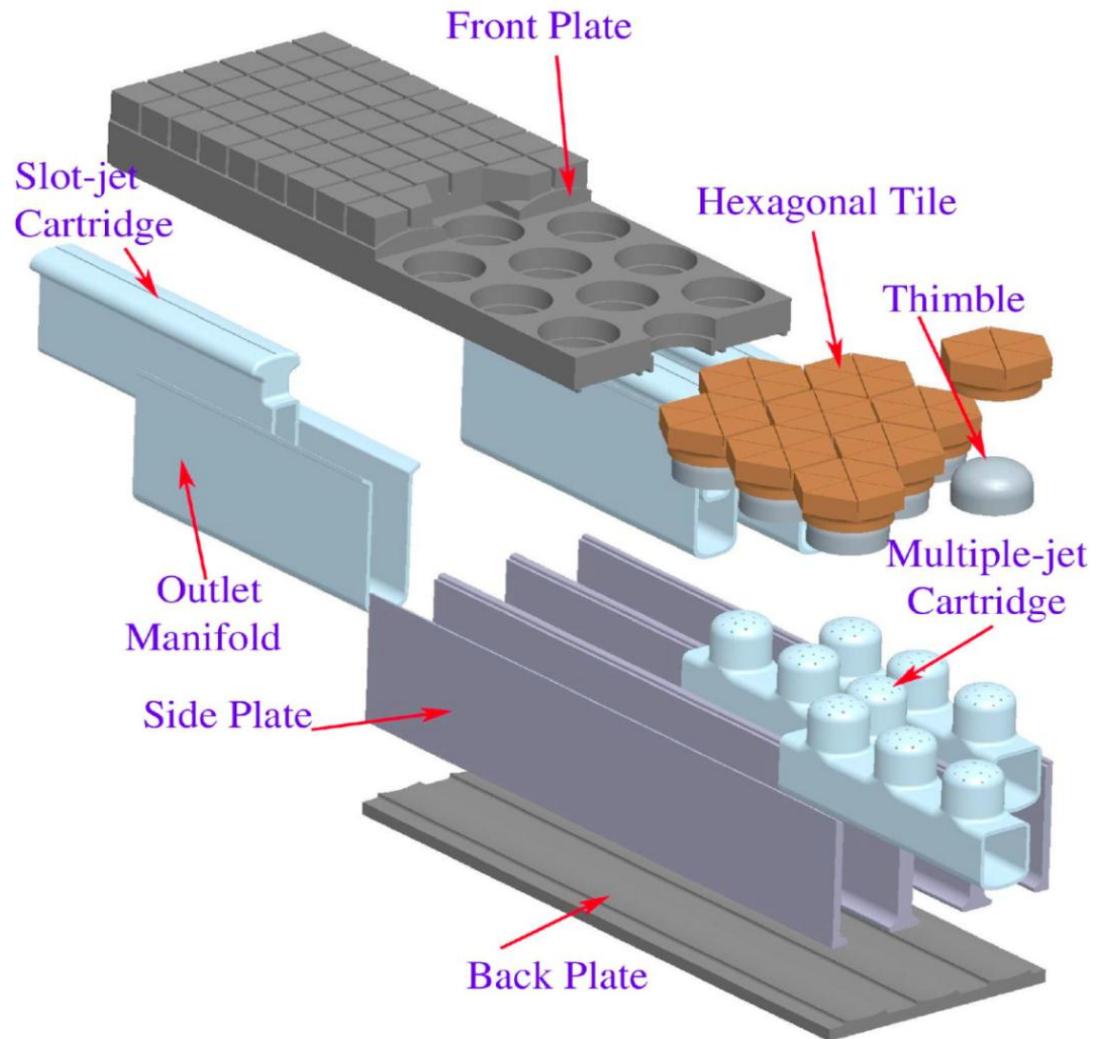
He and W temperature limited by steel  
Ferritic ODS  $\rightarrow 750^\circ\text{C}$  ???  
Eurofer ODS, 9Cr ODS  $\rightarrow 650^\circ\text{C}$   
Eurofer 97, F82H  $\rightarrow 550^\circ\text{C}$

# PLATE DESIGN (ARIES), JET COOLING



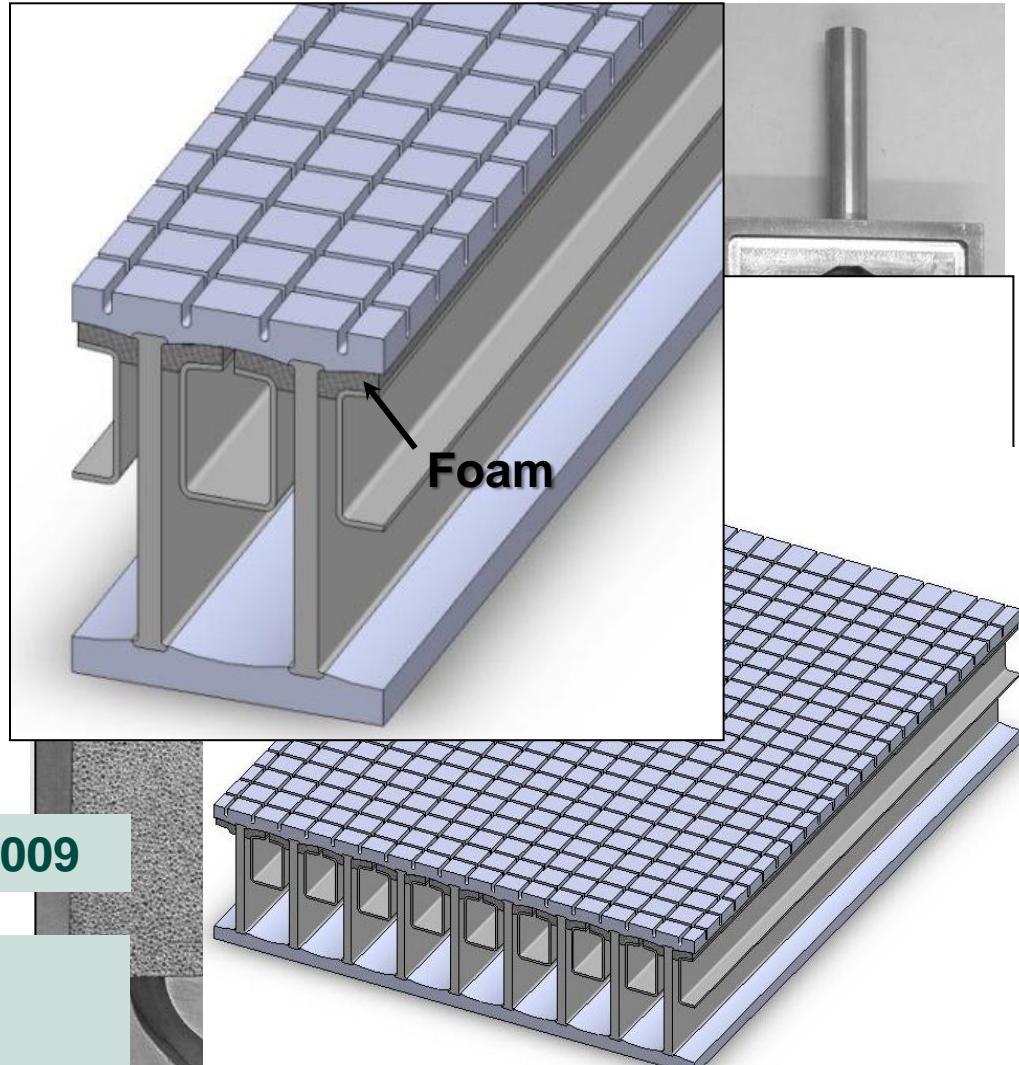
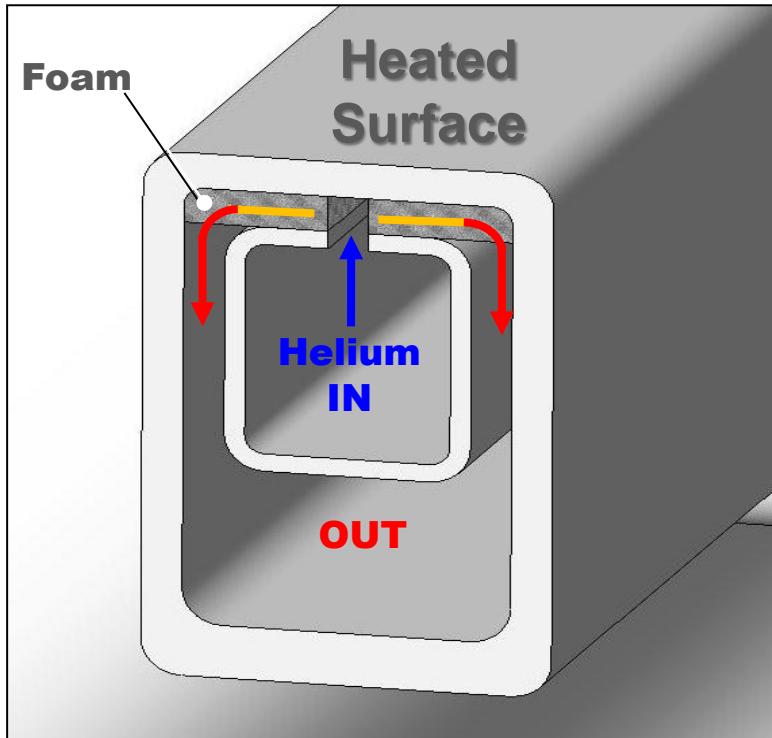
X.R. Wang, S. Malang, M.S. Tillack & ARIES Team, 2008-2011

# PLATE DESIGN (ARIES), JET COOLING



X.R. Wang, S. Malang, M.S. Tillack & ARIES Team, 2008-2011

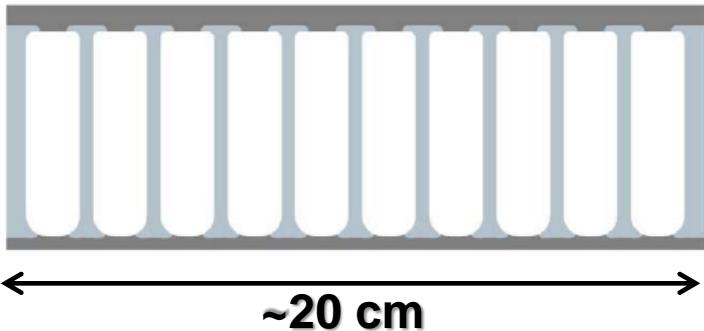
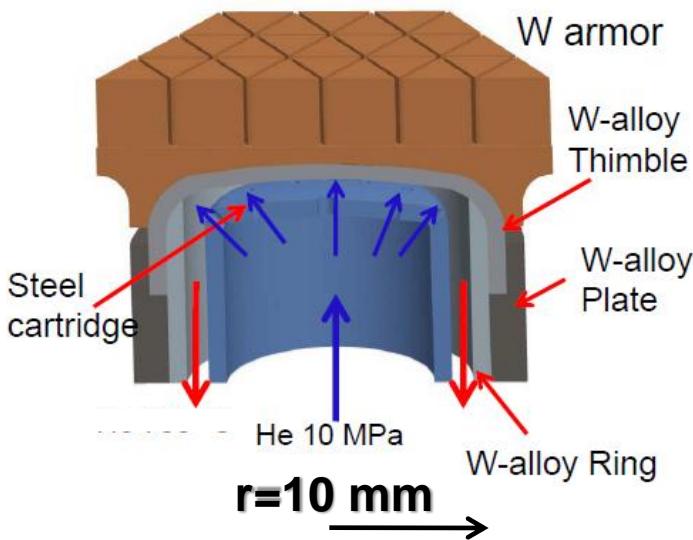
# PLATE DESIGN, FOAM PROMOTER



→ S. Sharafat *et al.*, UCLA, 2005-2009

→ Mo, Nb, SiC Foam:  
D. Youshison *et al.*, SNL, 2011

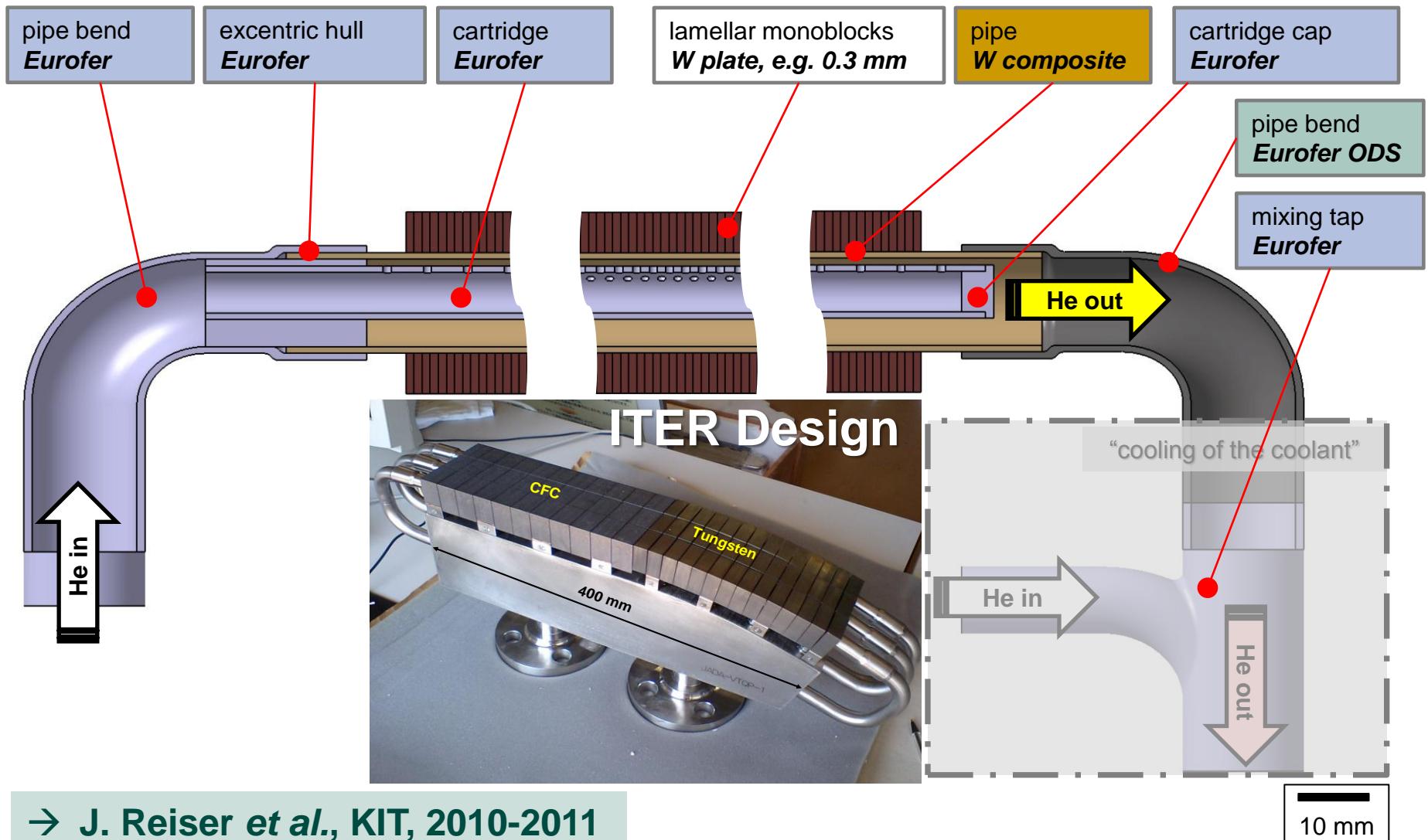
# PLATE DESIGN (ARIES), JET COOLING



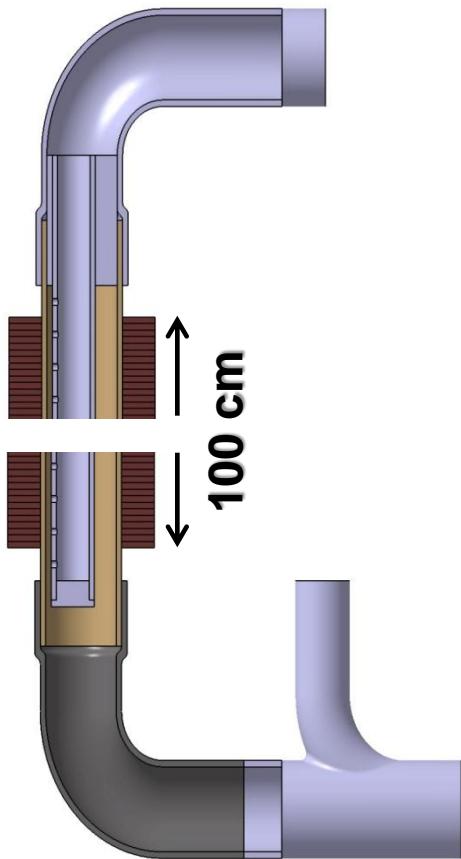
- + High heat flux:  $\sim 15 \text{ MW/m}^2$
- + Small Size, Thin Walls
- + “Double Containment”
- Large Numbers:  $\sim 300\,000$
- Many/Long Struct. W-W Joints

He temperature limited by steel  
(multiple jet cartridges)  
Ferritic ODS  $\rightarrow 750^\circ\text{C}$  ???  
Eurofer ODS, 9Cr ODS  $\rightarrow 650^\circ\text{C}$   
Eurofer 97, F82H  $\rightarrow 550^\circ\text{C}$

# PIPE/MONOBLOCK DESIGN, JET COOLING



# PIPE/MONOBLOCK DESIGN, JET COOLING



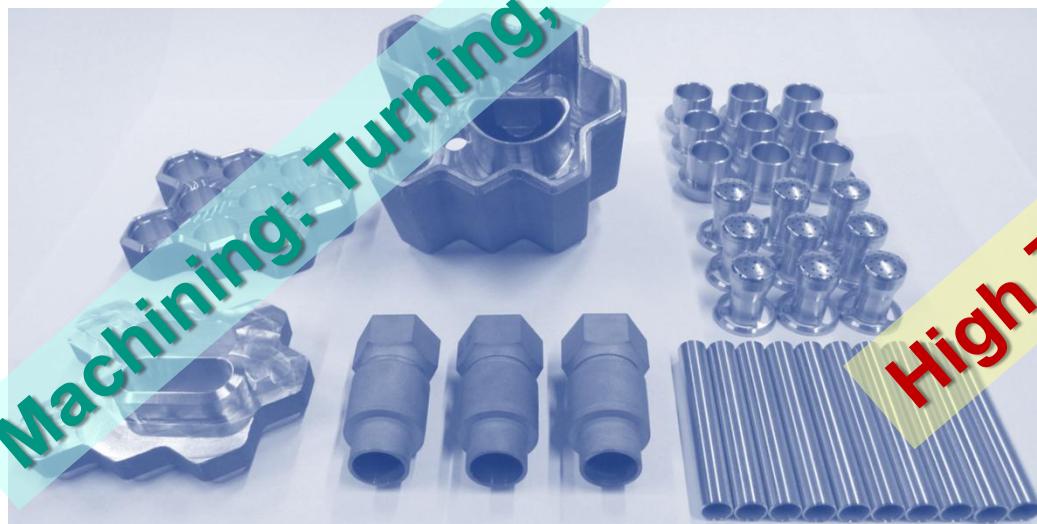
- + Simple Design
- + “Easy Joints”
- + “Small Number of Parts”
- + Inexpensive

- Low Temperatures for W
- Low Performance
- Speculative Pipe Material

He temperature limited by steel  
Eurofer ODS → 650°C (upper)  
Eurofer 97 → 350°C (lower)

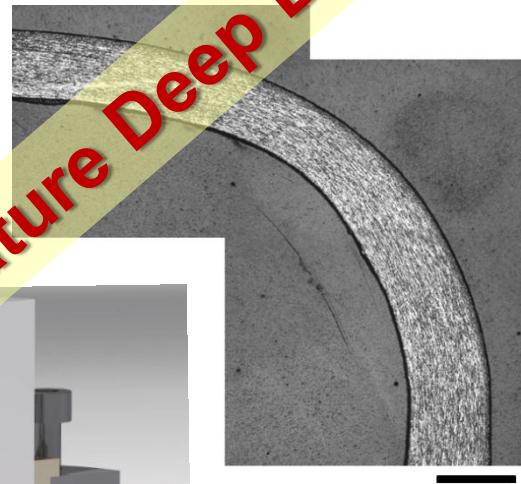
# 3 IDENTIFICATION OF ISSUES

# FABRICATION



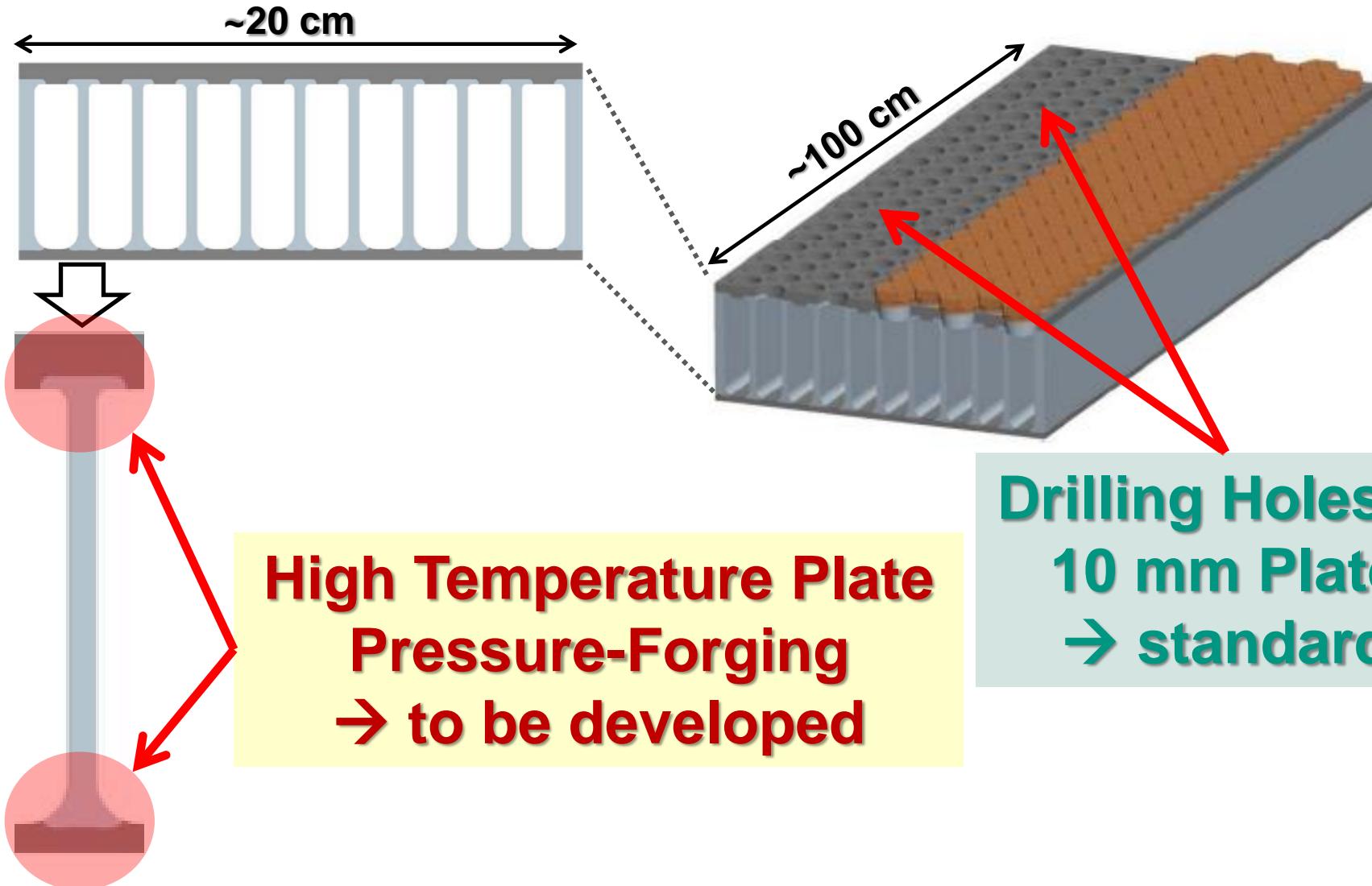
Machining: - Turning, Milling, Grinding, ...

High Temperature Deep Drawing



P. Norajitra, J. Reiser, S. Baumgärtner et al., KIT

# FABRICATION



**High Temperature Plate  
Pressure-Forging  
→ to be developed**

**Drilling Holes in  
10 mm Plate  
→ standard**

**IMPORTANT: We have to consider joints for STRUCTURAL applications!!!**  
**Joining W tiles to W timbles is NOT considered as structural application!**

## Brazing Material for W-W Joints

Brazing temperature must be  $>>900\text{-}1200\text{ }^{\circ}\text{C}$  (operating temp.)

Brazing temperature must be  $<1800\text{ }^{\circ}\text{C} \rightarrow$  Grain growth

Formation of brittle compounds cannot be tolerated

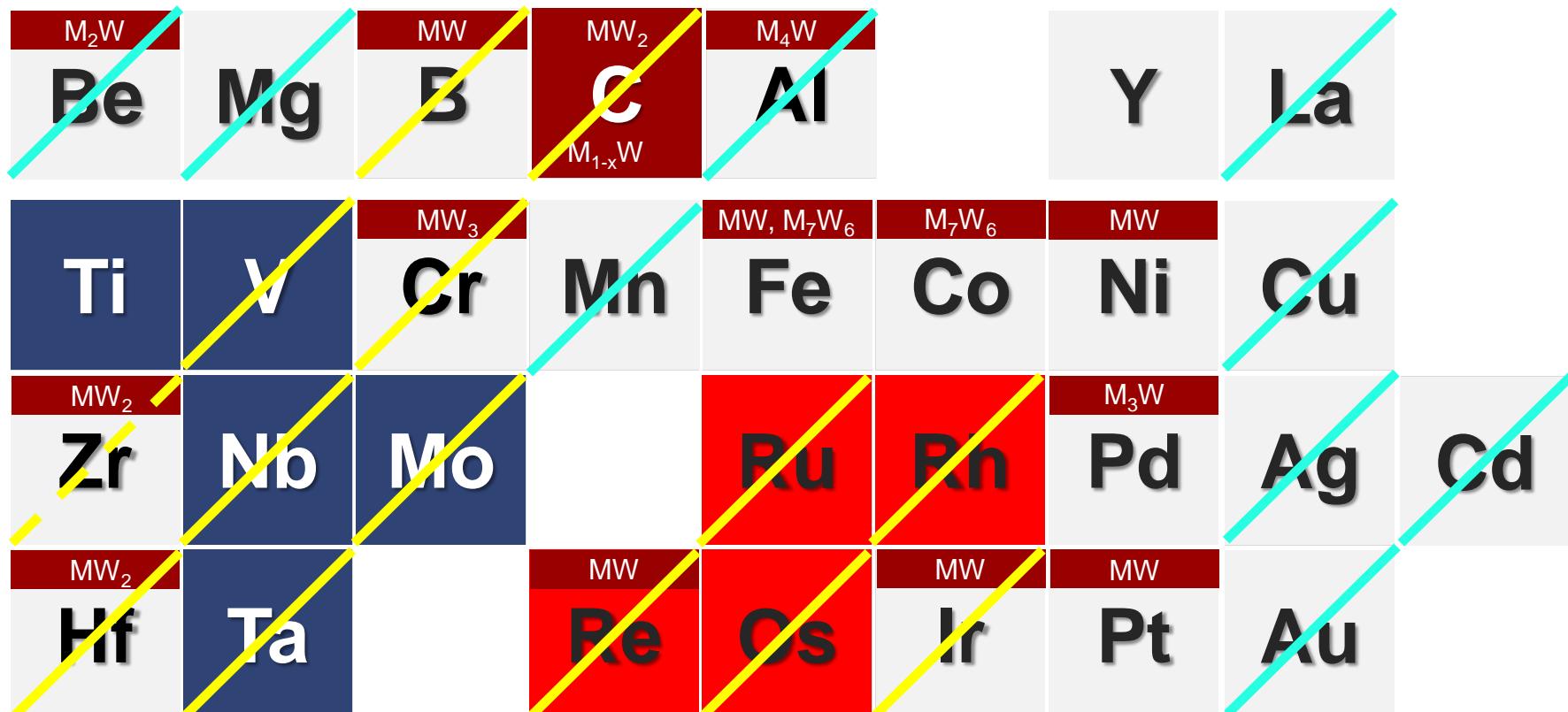
## Brazing Material for W-Steel Joints

Brazing temperature  $<1100\text{ }^{\circ}\text{C} \rightarrow$  Grain growth (in steel)

Brazing temperature must be  $>>550\text{-}750\text{ }^{\circ}\text{C}$  (operating temp.)

Formation of brittle compounds cannot be tolerated

# BRAZING W $\rightarrow$ W



**W Insoluble**

**Intermetallic Phases**

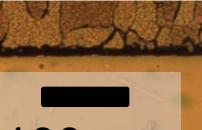
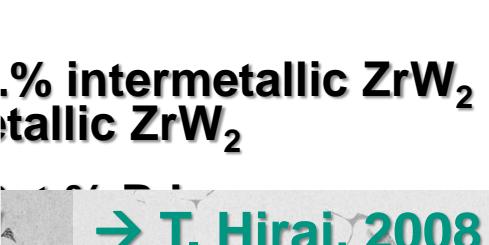
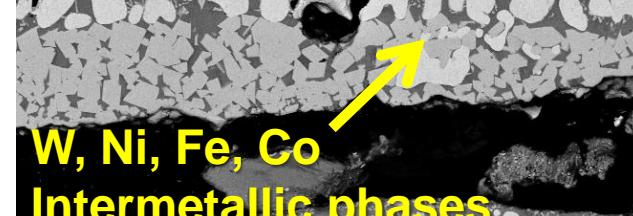
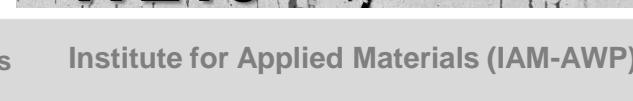
**W Rich Line Compounds**

**Solid Solution**

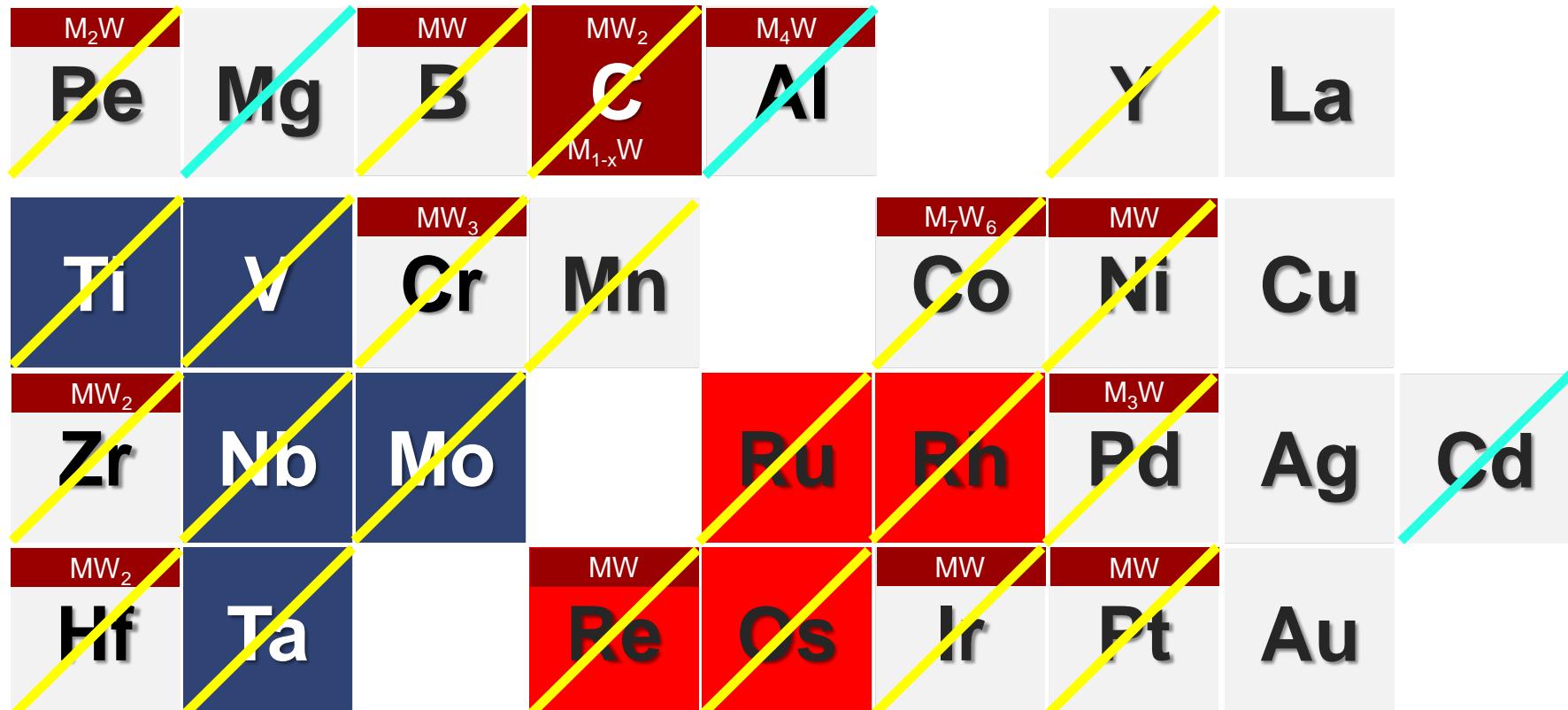
$T_m > 1800^\circ$

$T_m < 1400^\circ$

# BRAZING MATERIALS, W→W

1670°C	Ti	<740°C: W <b>Difficult, ... ?</b>	 100 µm
1520°C	Y	Strong <1570°C	40°C 100°C: alpha-Y + Y <sub>2</sub> O <sub>3</sub> 
1850°C	Zr	<2160°C: W → ZrW <sub>2</sub> <1700°C: ZrW <sub>2</sub> <860°C: ZrW <sub>2</sub> <b>Brittle Joint</b>	.% intermetallic ZrW <sub>2</sub> metallic ZrW <sub>2</sub> 
1550°C	Pd	<1800°C: W → Pd <sub>3</sub> W <900°C: intermetallic Pd <sub>3</sub> W ???	→ T. Hirai, 2008 
1770°C	Pt	<2400°C: intermetallic Pt <sub>3</sub> W	Pt <sub>3</sub> W 
1540°C	Fe	<1700°C: $\mu$ phase Fe <sub>7</sub> W <sub>6</sub> (Fe,W) <sub>4</sub> <1000°C: Laves phase Fe <sub>2</sub> W	
1500°C	Co	<1700°C: $\mu$ phase Co <sub>7</sub> W <sub>6</sub> (Co,W) <sub>4</sub> <1000°C: Laves phase Co <sub>3</sub> W	
1450°C	Ni	<b>Brittle Joints</b> <1000°C: peritectoid intermetallics NiW <950°C: peritectoid intermetallic Ni <sub>4</sub> W	W, Ni, Fe, Co Intermetallic phases WL10 20 µm

# BRAZING W→STEEL



**W Insoluble**

**Intermetallic Phases**

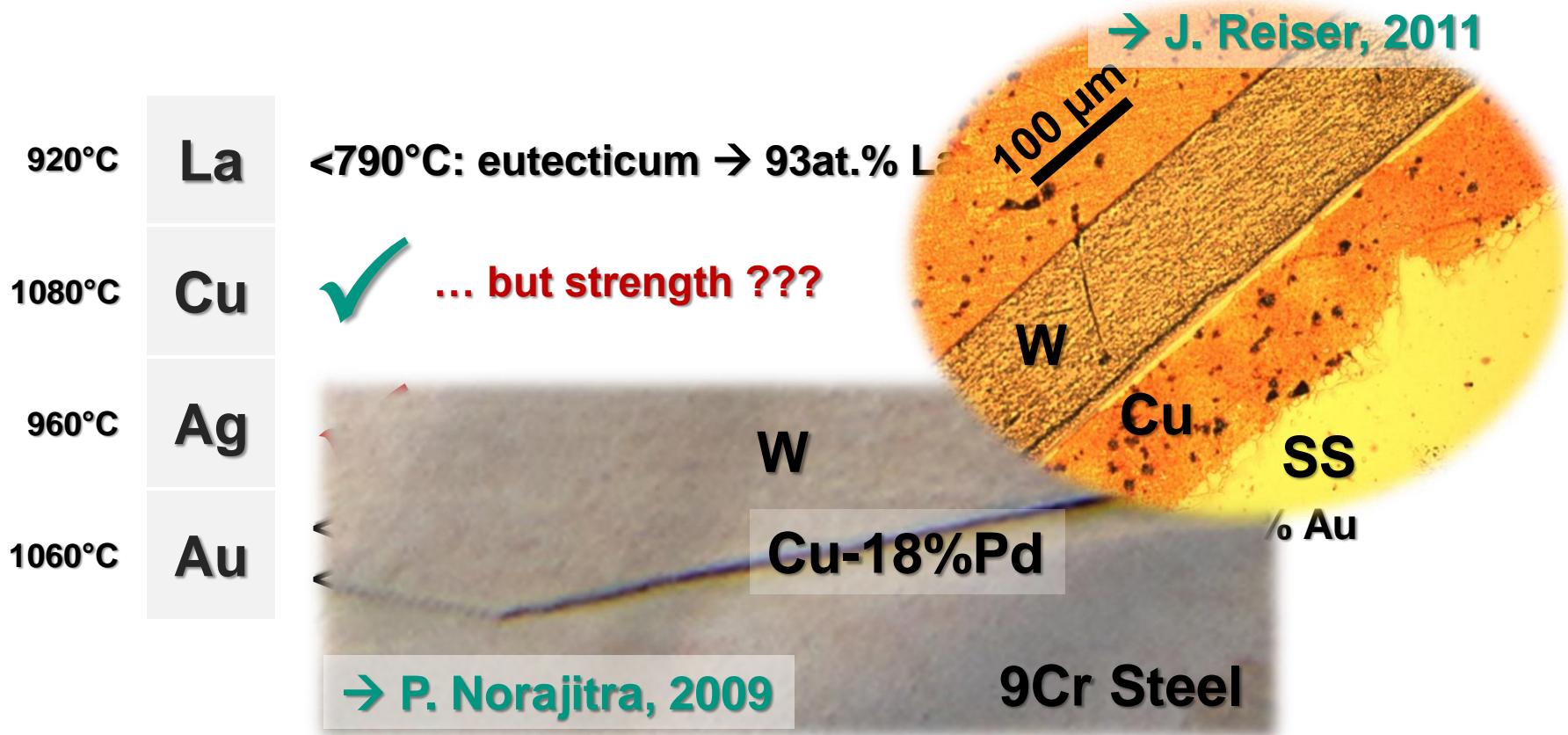
**W Rich Line Compounds**

**Solid Solution**

$T_m > 1200^\circ$

$T_m < 900^\circ$

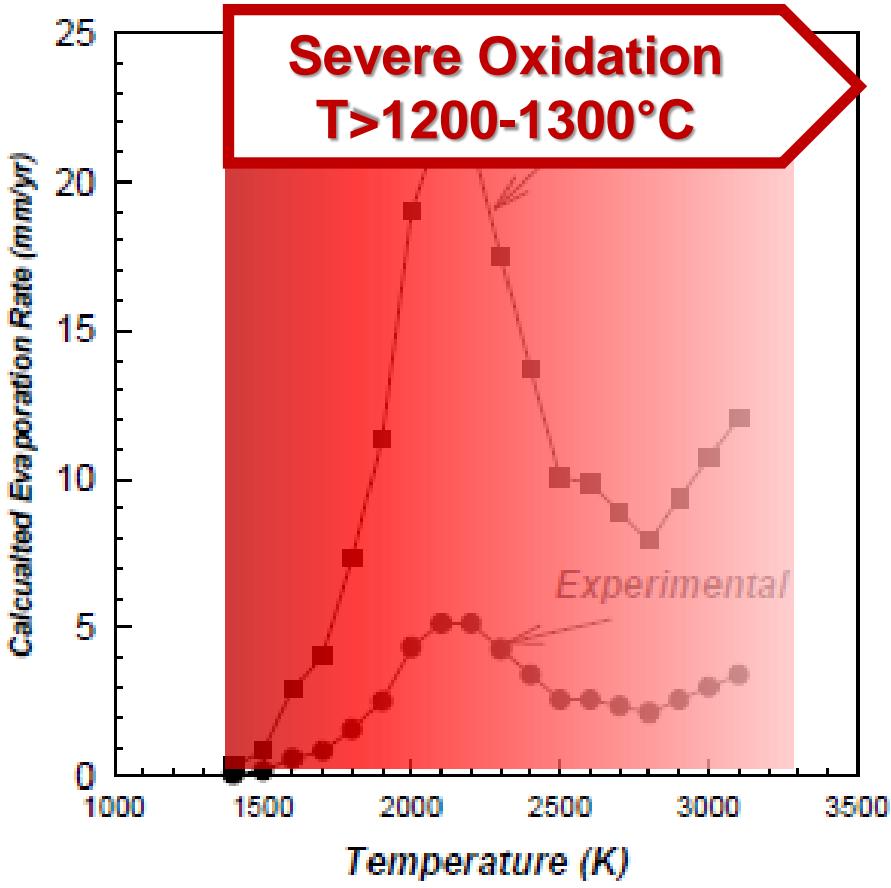
# BRAZING MATERIALS, W→STEEL



**CONCLUSION: Copper has to be used as sealing rather than as a braze material !!!**

# ENVIRONMENT → OXIDATION

$\text{WO}_3$  yellow,  $T_m=1470^\circ\text{C}$ ,  $7200 \text{ kg/m}^3$ , volatile in vacuum



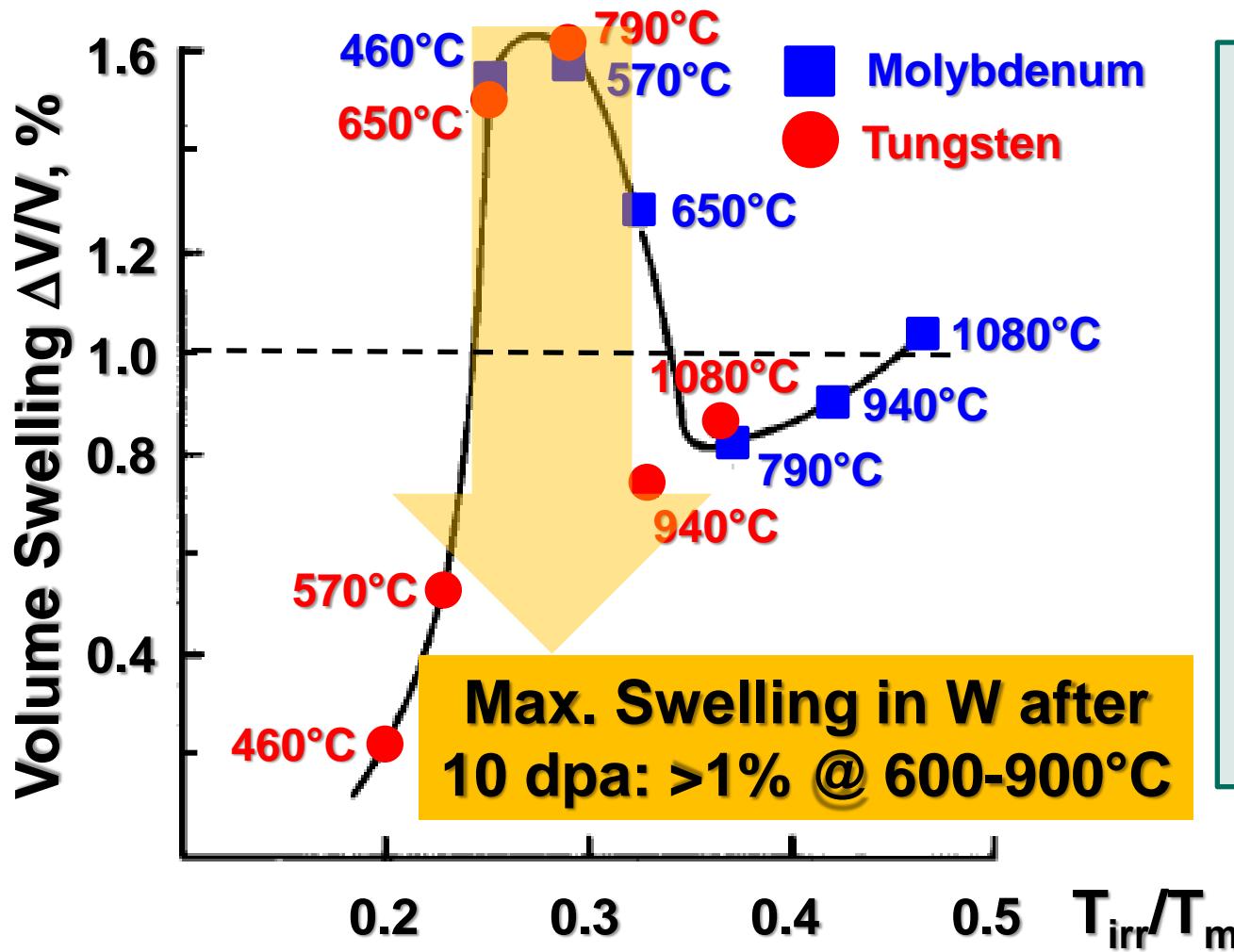
→ N.M. Ghoniem, 1998



## CONCLUSION

For tungsten operating at 50 atm. He coolant, at 0.1 ppm oxygen, the upper temperature is estimated at  $1200-1300^\circ\text{C}$ .

# IRRADIATION EFFECTS → SWELLING



## EBR-II

$E_n > 1 \text{ MeV}$   
 $1 \times 10^{22} \text{ n/cm}^2$

$E_n > 0.1 \text{ MeV}$   
 $1.6 \times 10^{22} \text{ n/cm}^2$

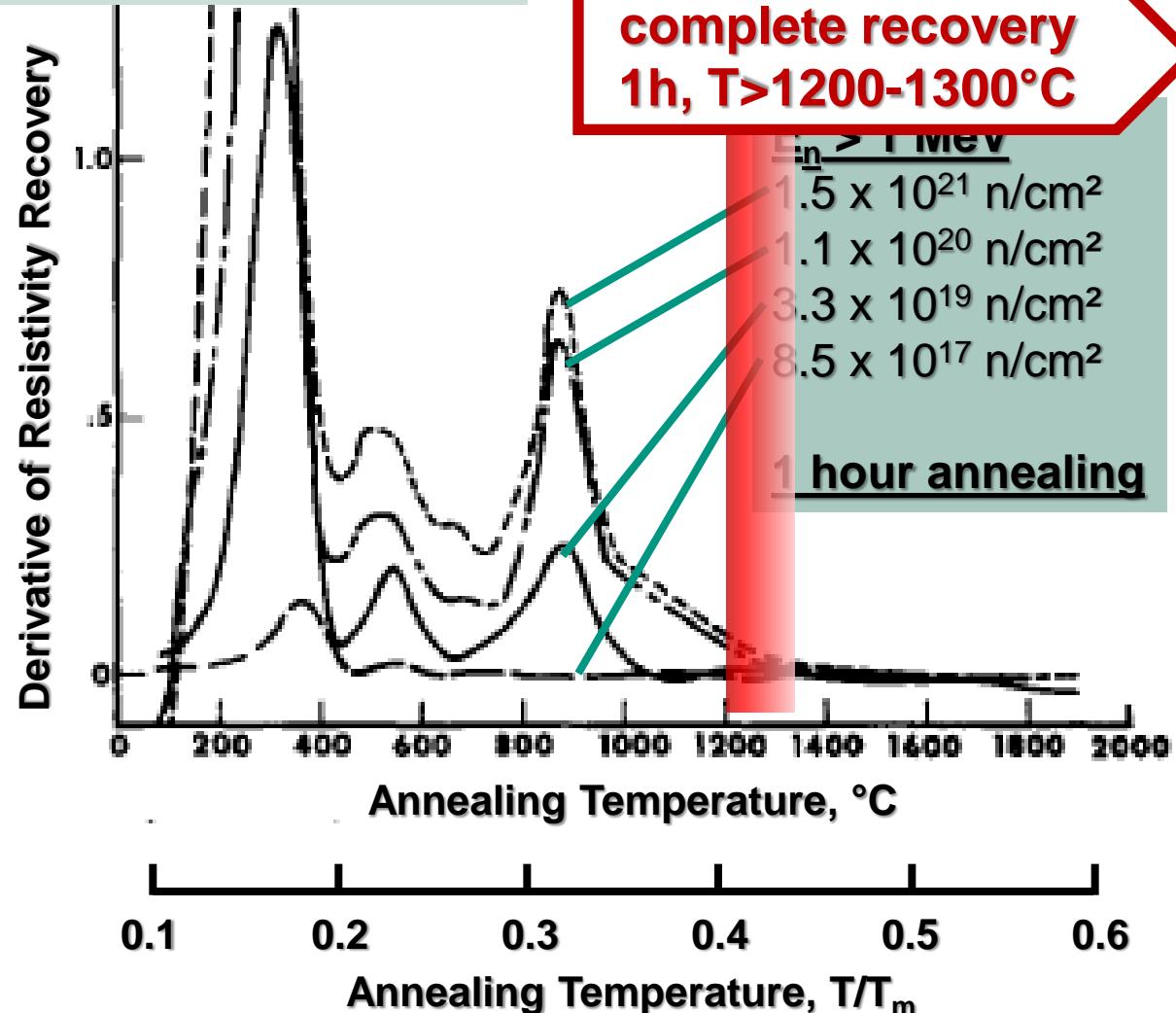
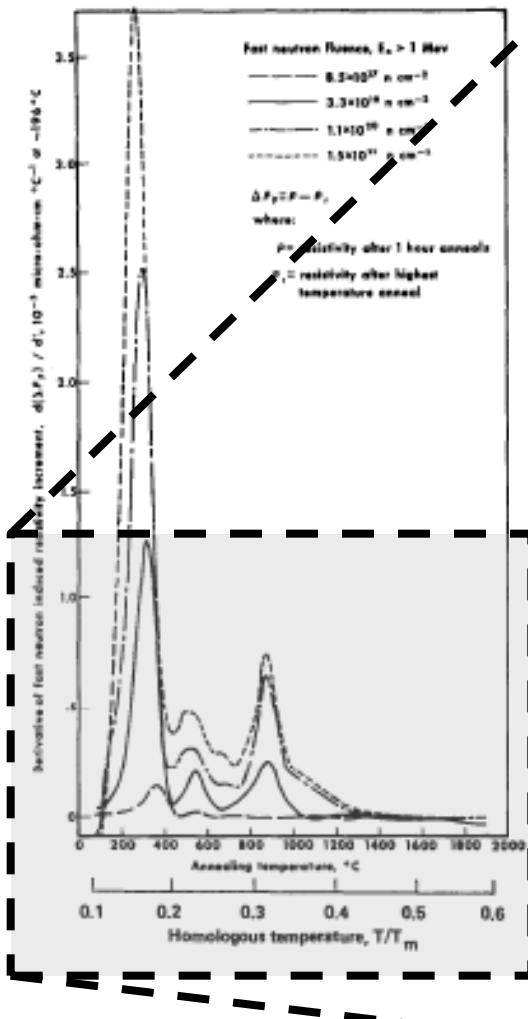
→ 29 dpa in Mo

→ 9.6 dpa in W

→ F. Lee, J. Matolich, J. Motteff, JNM 62 (1976) 115-117

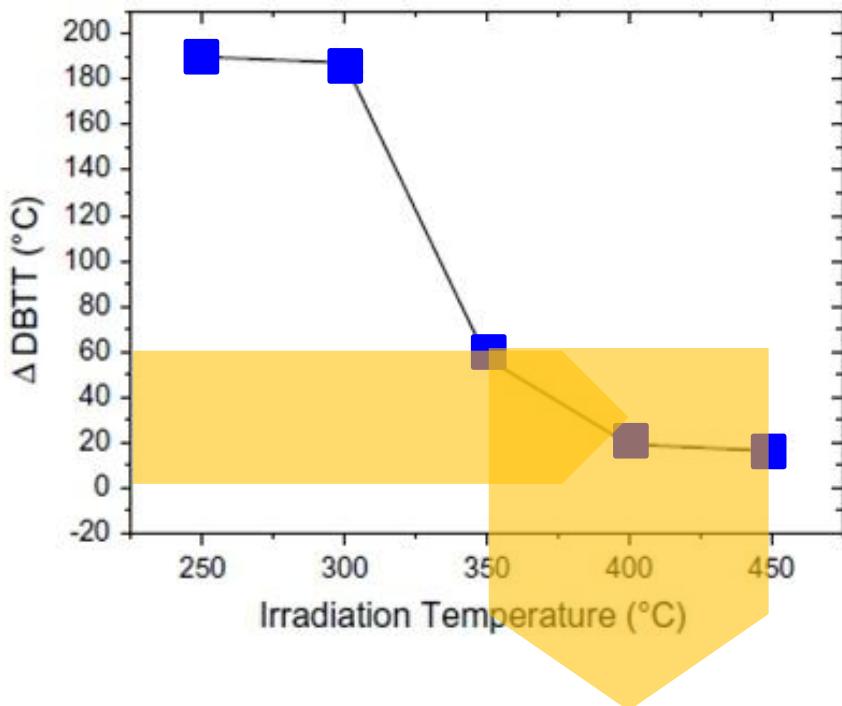
# IRRADIATION EFFECTS → RECOVERY

→ L.K. Keys, J. Moteff, JNM 34 (1970) 260-280



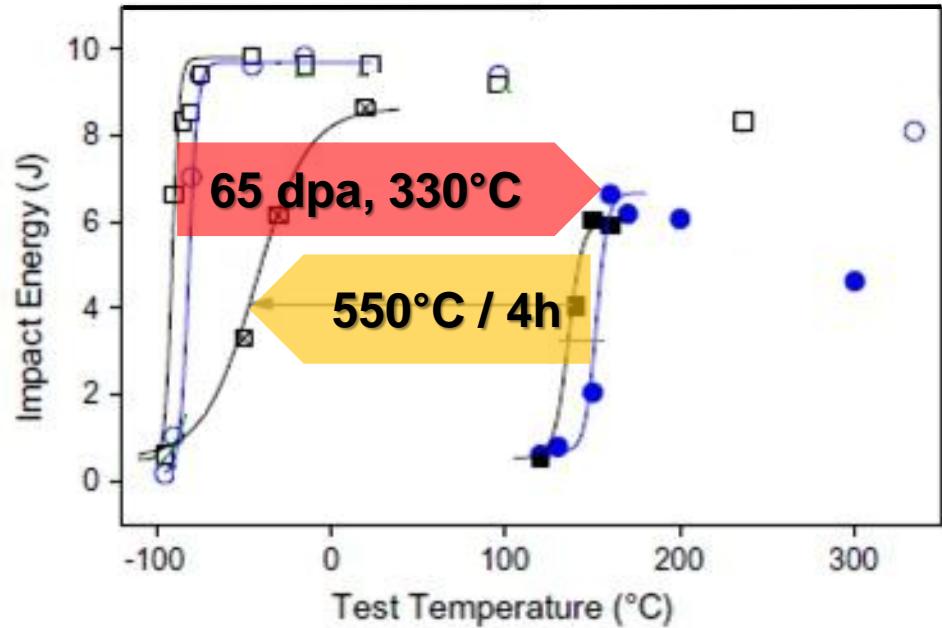
# IRRADIATION EFFECTS → EMBRITTLEMENT

**EUROFER:** In-service irradiation embrittlement after ~10 dpa



**Possible Operating Temperature  $T_{op} > 350^\circ\text{C}$**

**EUROFER:** Recovery of 65 dpa irradiation embrittlement

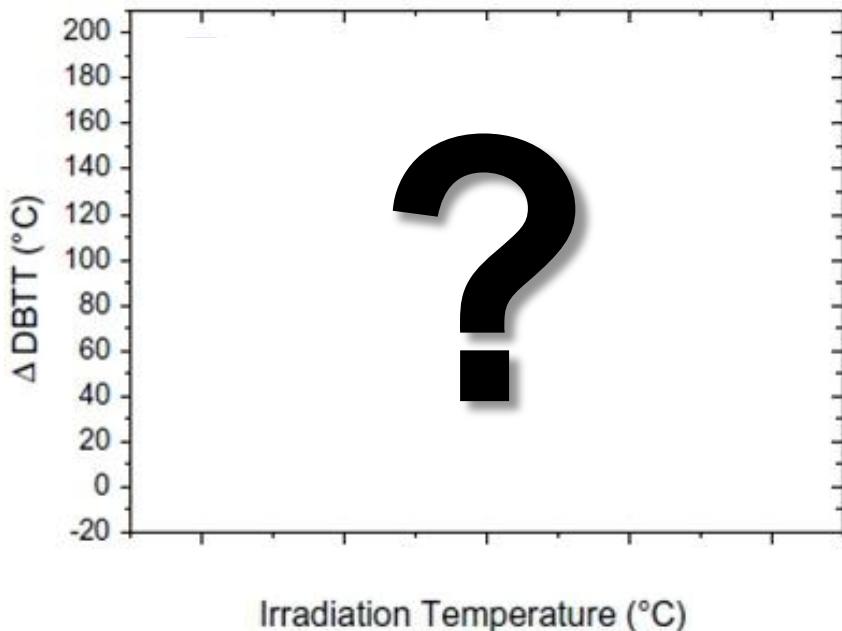


**Possible Recovery Temp.**  
 $T_{rec} > 550^\circ\text{C}$

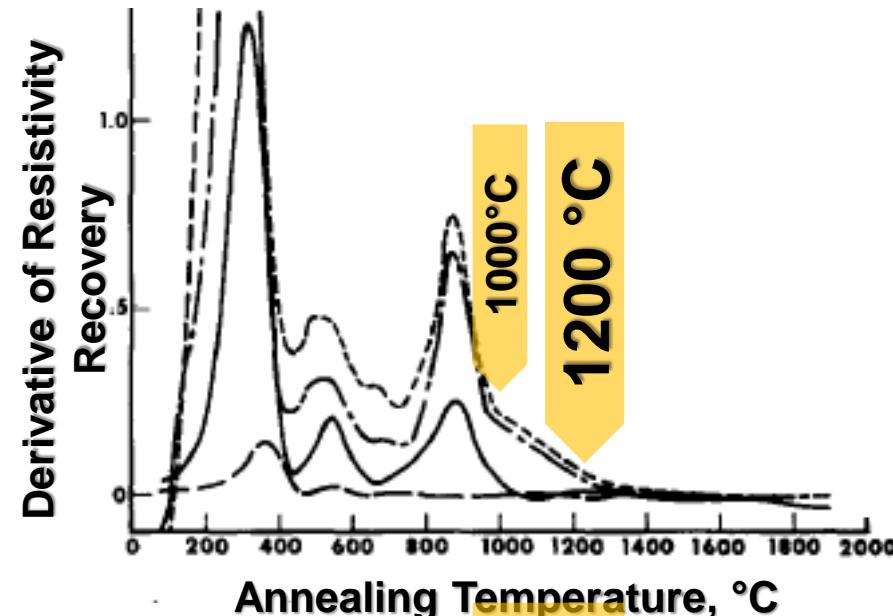
→ E. Gaganidze et al., KIT

# IRRADIATION EFFECTS → EMBRITTLEMENT

**TUNGSTEN:** In-service irradiation embrittlement after 10-20 dpa



**Tungsten:** Recovery of ~2 dpa stage IV irradiation hardening

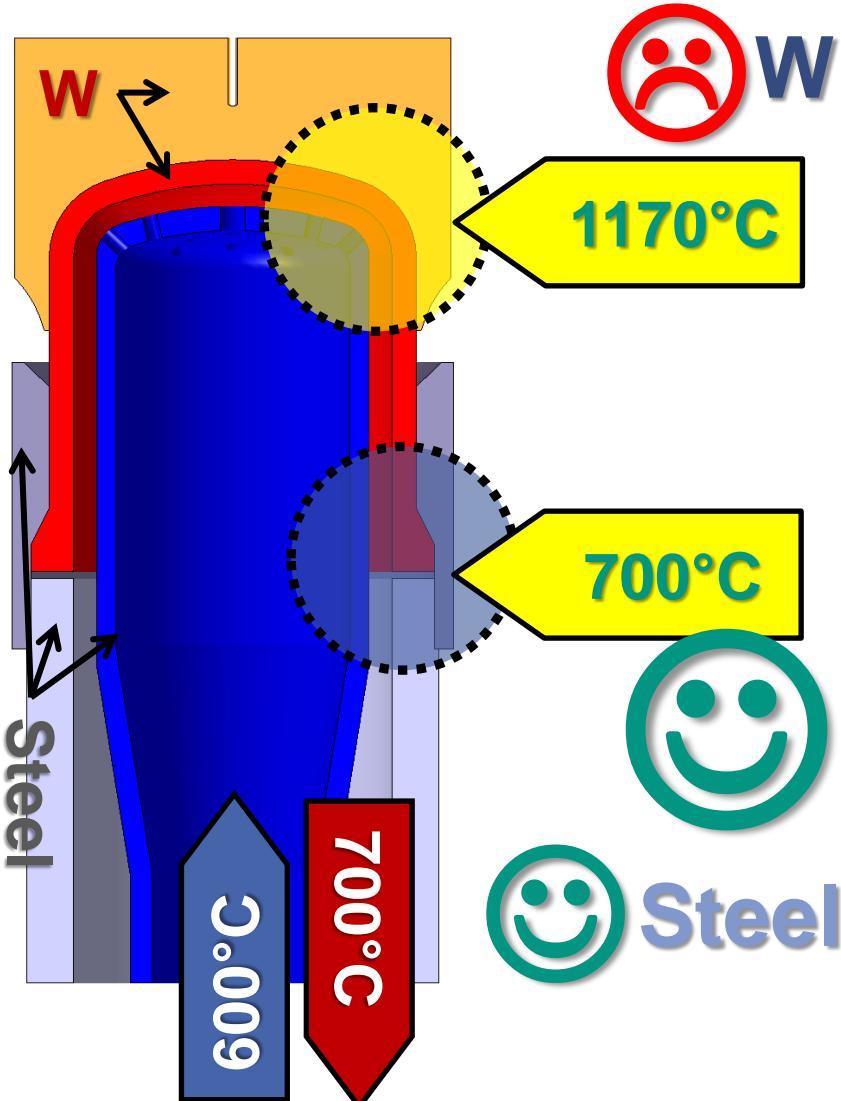


**Possible Operating Temp.**  
 $T_{\text{op}} > 800^{\circ}\text{C} \dots 1000^{\circ}\text{C}$



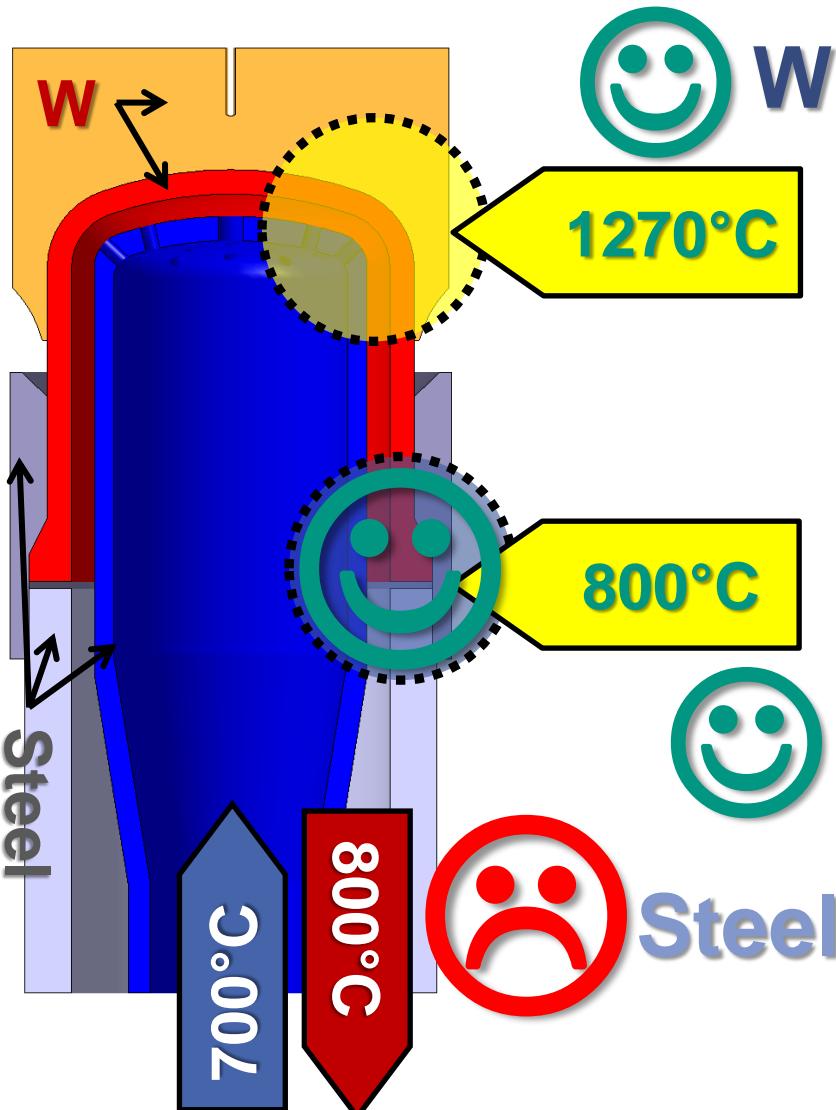
MATERIAL  
MEETS  
DESIGN

# MATERIALS / DESIGN WINDOW → FINGER



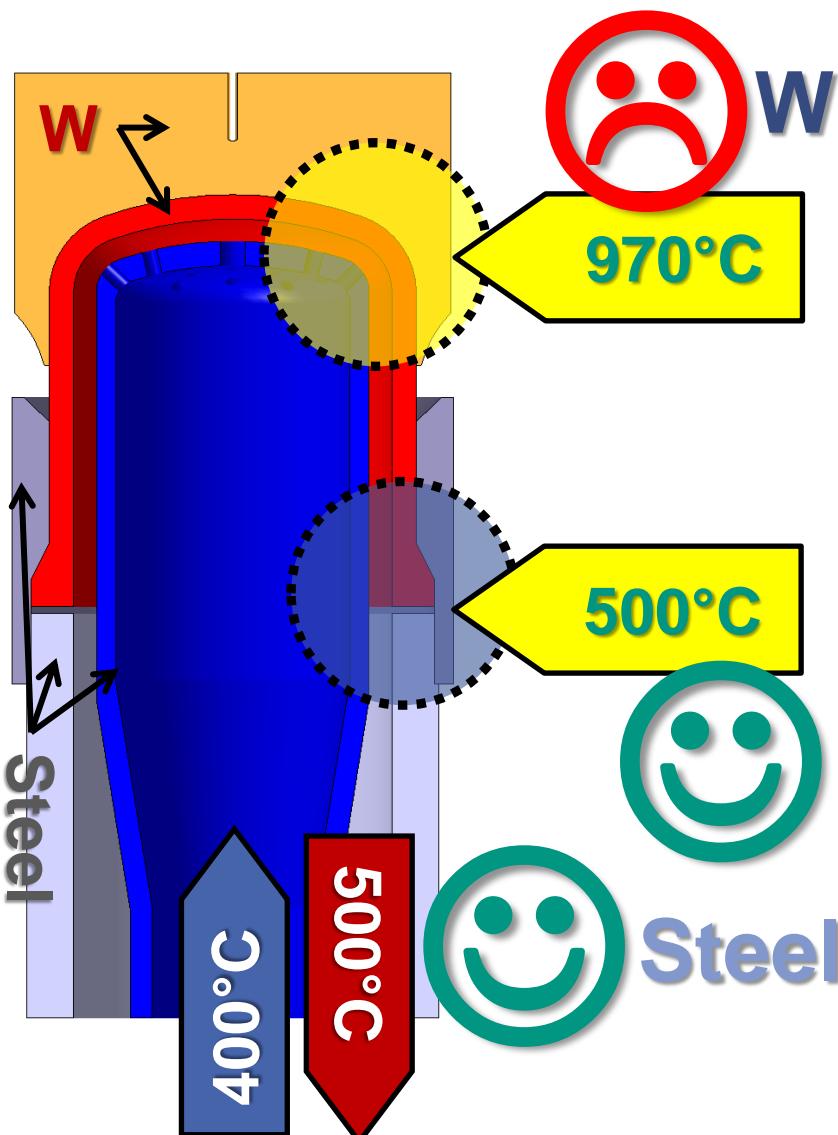
Oxidation	→	OK
Swelling ~3%	→	?
Embrittlement	→	OK
Grain Growth	→	? (ODS)
Swelling ~5%	→	?
Embrittlement	→	<b>NO GO (?)</b>
Brazing (Cu)	→	OK (...)
Embrittlement	→	OK
Strength, ...	→	? (ODS)

# MATERIALS / DESIGN WINDOW → FINGER



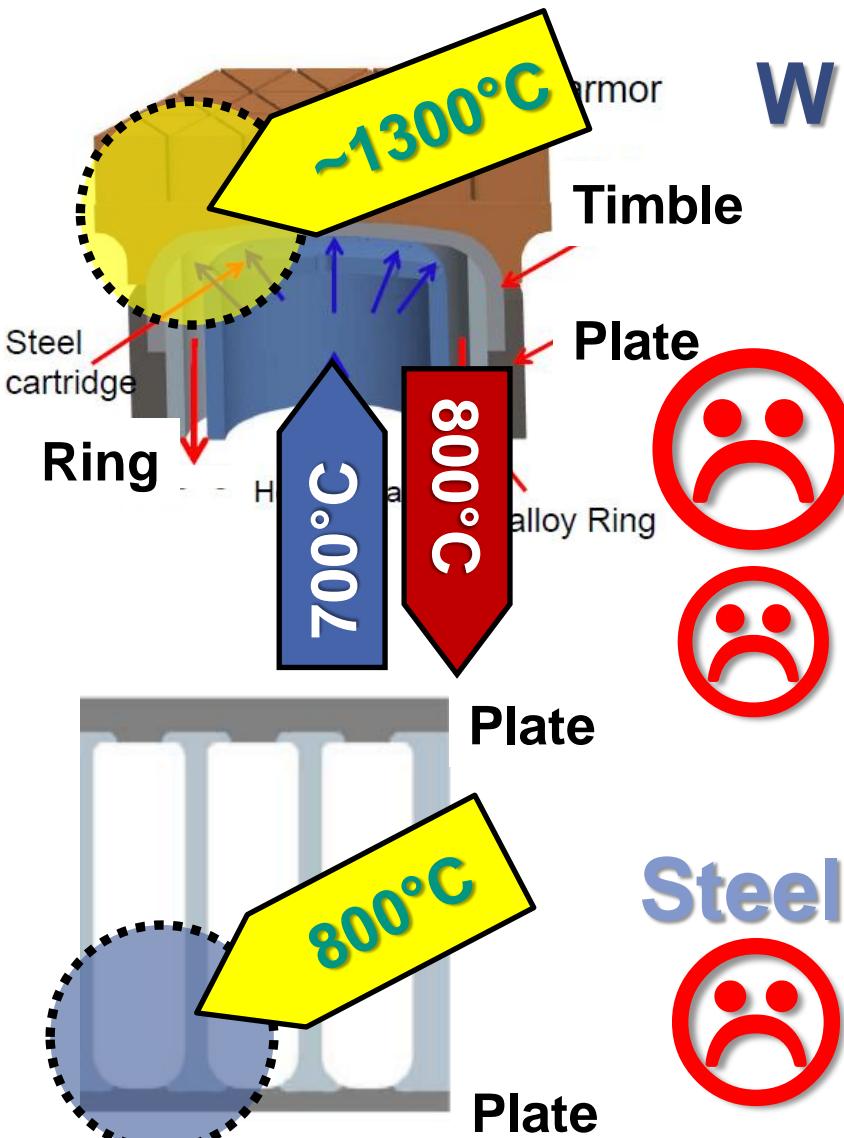
Oxidation	→	OK
Swelling ~3%	→	?
Embrittlement	→	OK
Grain Growth	→	ODS
Swelling ~5%	→	?
Embrittlement	→	OK (tbc)
Brazing (Cu)	→	OK (...)
Embrittlement	→	OK
Strength, ...	→	ODS ???

# MATERIALS / DESIGN WINDOW → FINGER



Oxidation	→	OK
Swelling <3%	→	OK (?)
Embrittlement	→	OK
Grain Growth	→	OK
Swelling <2%	→	OK (?)
Embrittlement	→	<b>NO GO</b>
Brazing (Cu)	→	OK (...)
Embrittlement	→	OK
Strength, ...	→	OK

# MATERIALS / DESIGN WINDOW → PLATE



**W**

Oxidation → OK  
 Swelling ~3% → ???  
 Embrittlement → OK  
 Grain Growth → ODS



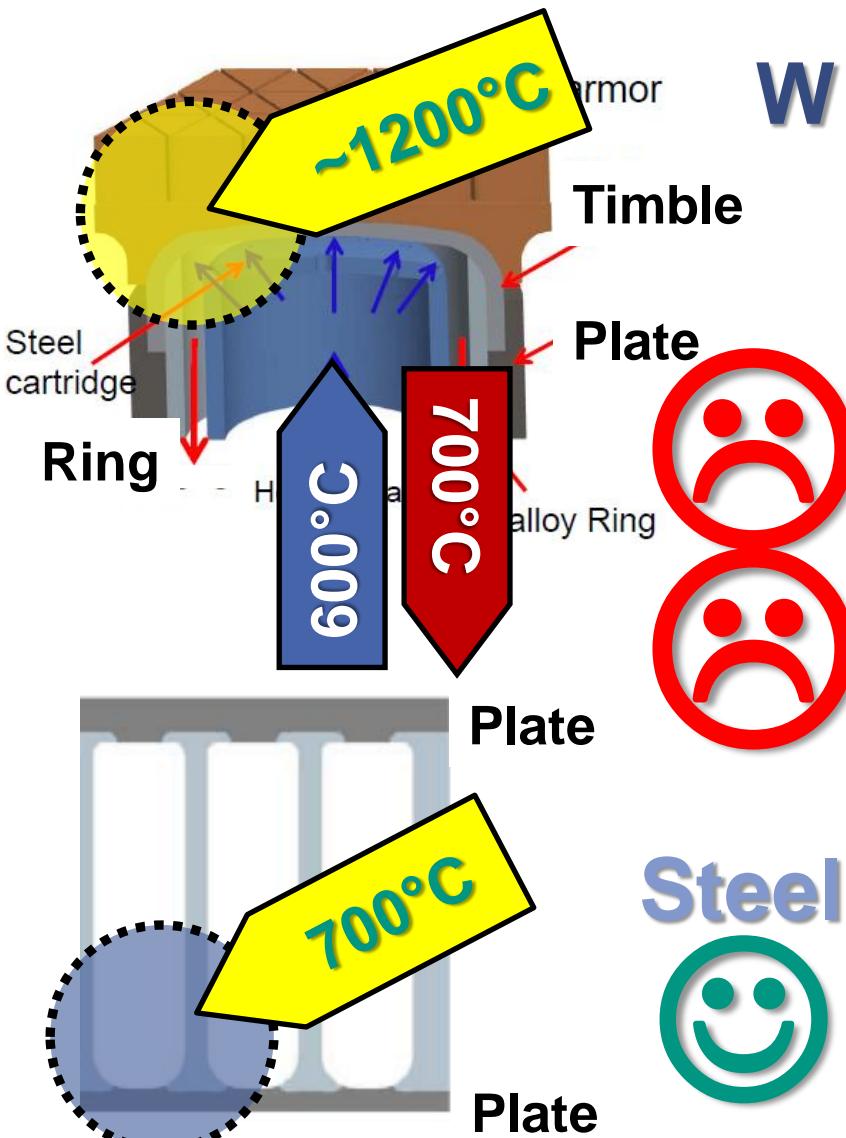
Brazing W→W → Pd, Pt ???

Swelling ~5% → ?????  
 Embrittlement → OK (tbc)

**Steel**

Embrittlement → OK  
 Strength, ... → ODS ???

# MATERIALS / DESIGN WINDOW → PLATE



**W**

Oxidation	→	OK
Swelling ~3%	→	???
Embrittlement	→	OK
Grain Growth	→	? (ODS)

**Brazing W→W** → Pd, Pt ???

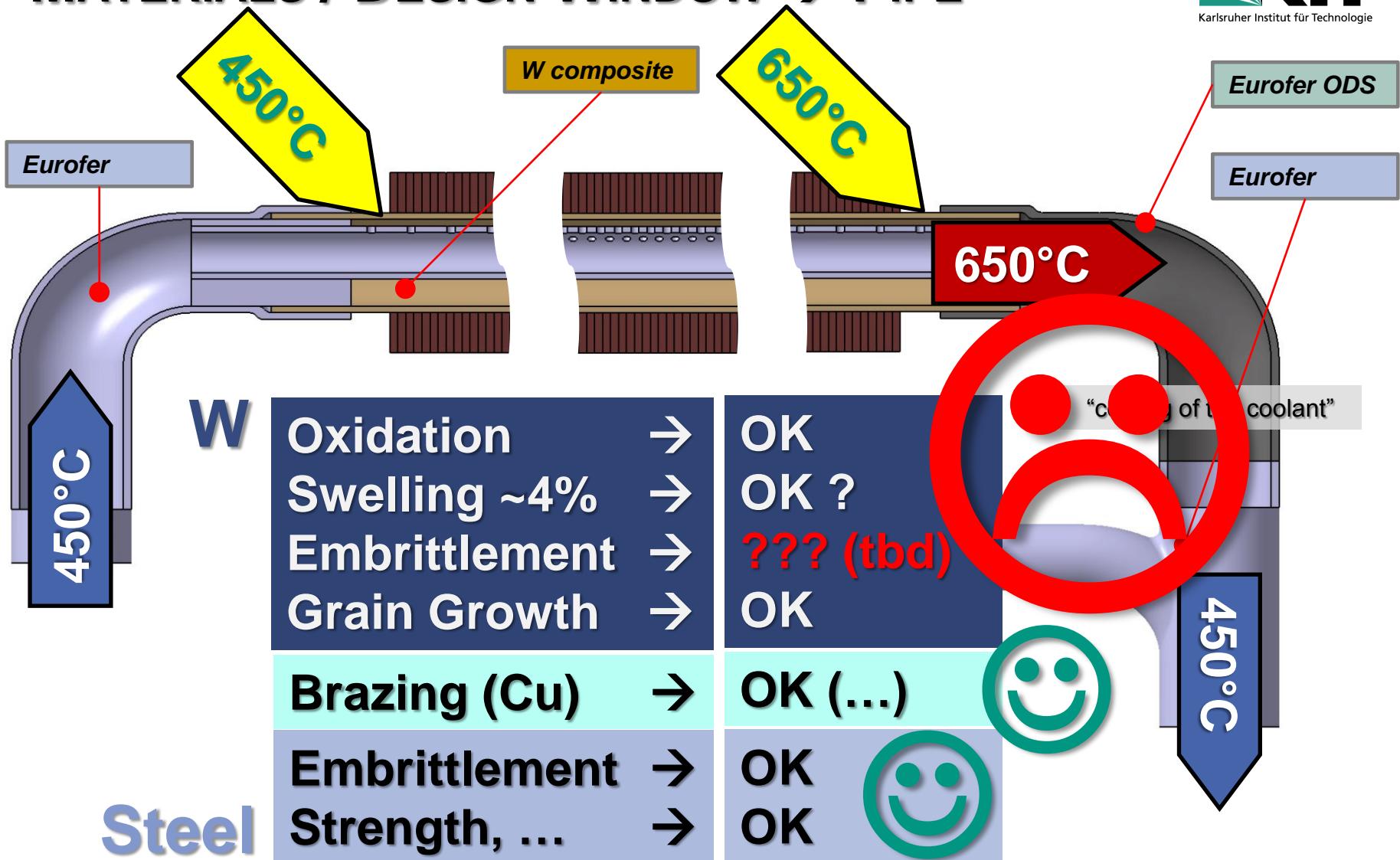
Swelling ~5% → ?????  
Embrittlement → NO GO (?)

**Steel**



Embrittlement	→	OK
Strength, ...	→	? (ODS)

# MATERIALS / DESIGN WINDOW → PIPE



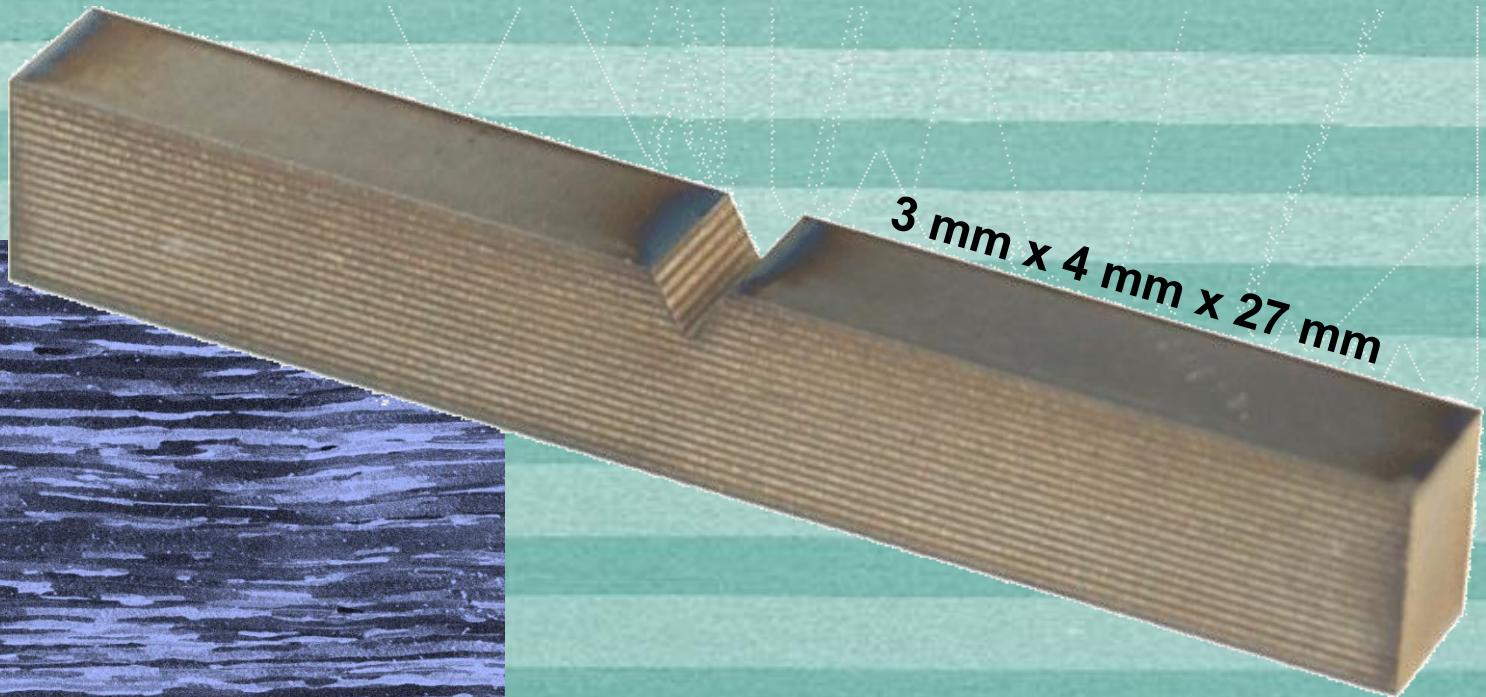
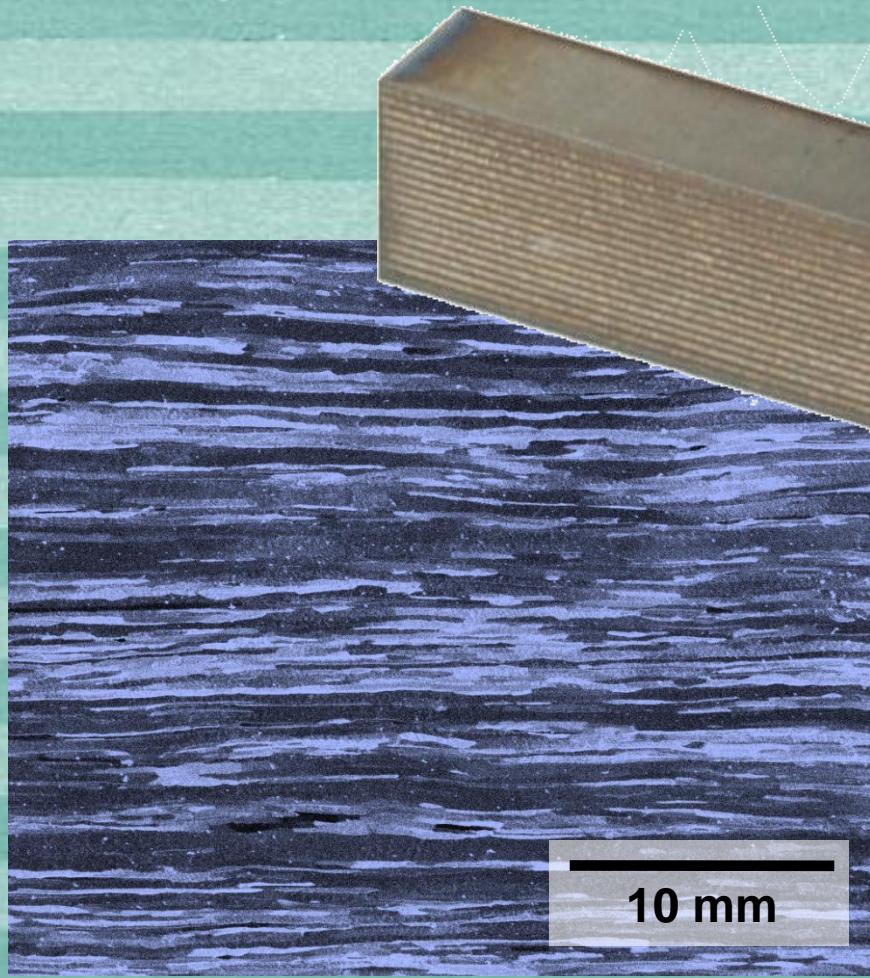
5

# summary and path forward

# SUMMARY

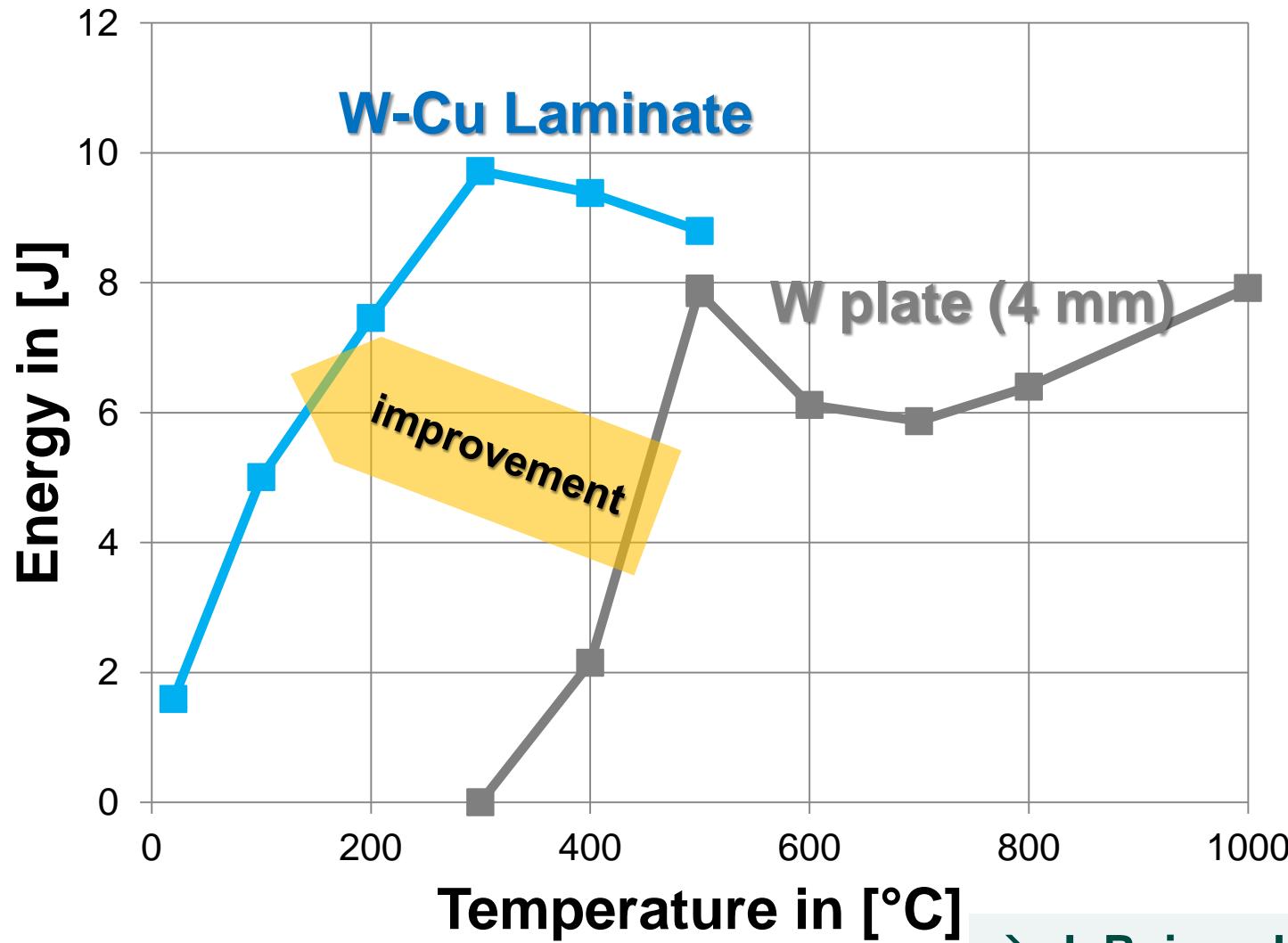
- Thin plates (<4 mm) of pure W (including small amounts of grain stabilizers, if necessary) are the most suitable semi-finished products for structural applications !
  - The microstructure of the according components must be developed !
  - Copper can be used as brazing material for W-steel joints, but needs additional strengthening by desing !
  - Suitable structural W-W joints are not yet demonstrated and characterized ! → Pt, Pd, ...?
  - W irradiation data needed for design (determination of lowest possible operating temperature) !!!
- 
- 

## W Laminate Material



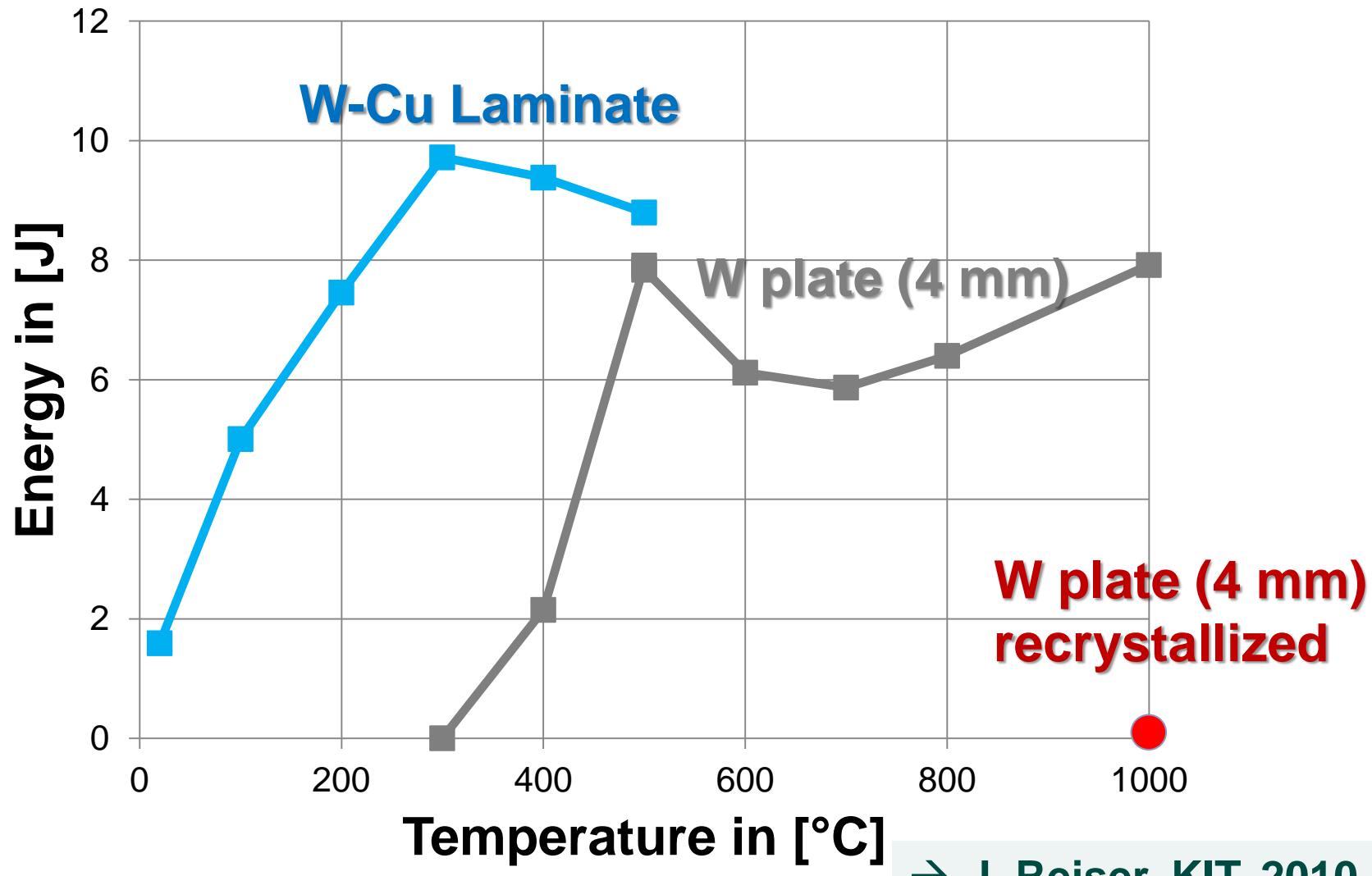
→ J. Reiser, KIT, 2010-2011

# PATH FORWARD → TUNGSTEN ALTERNATIVES

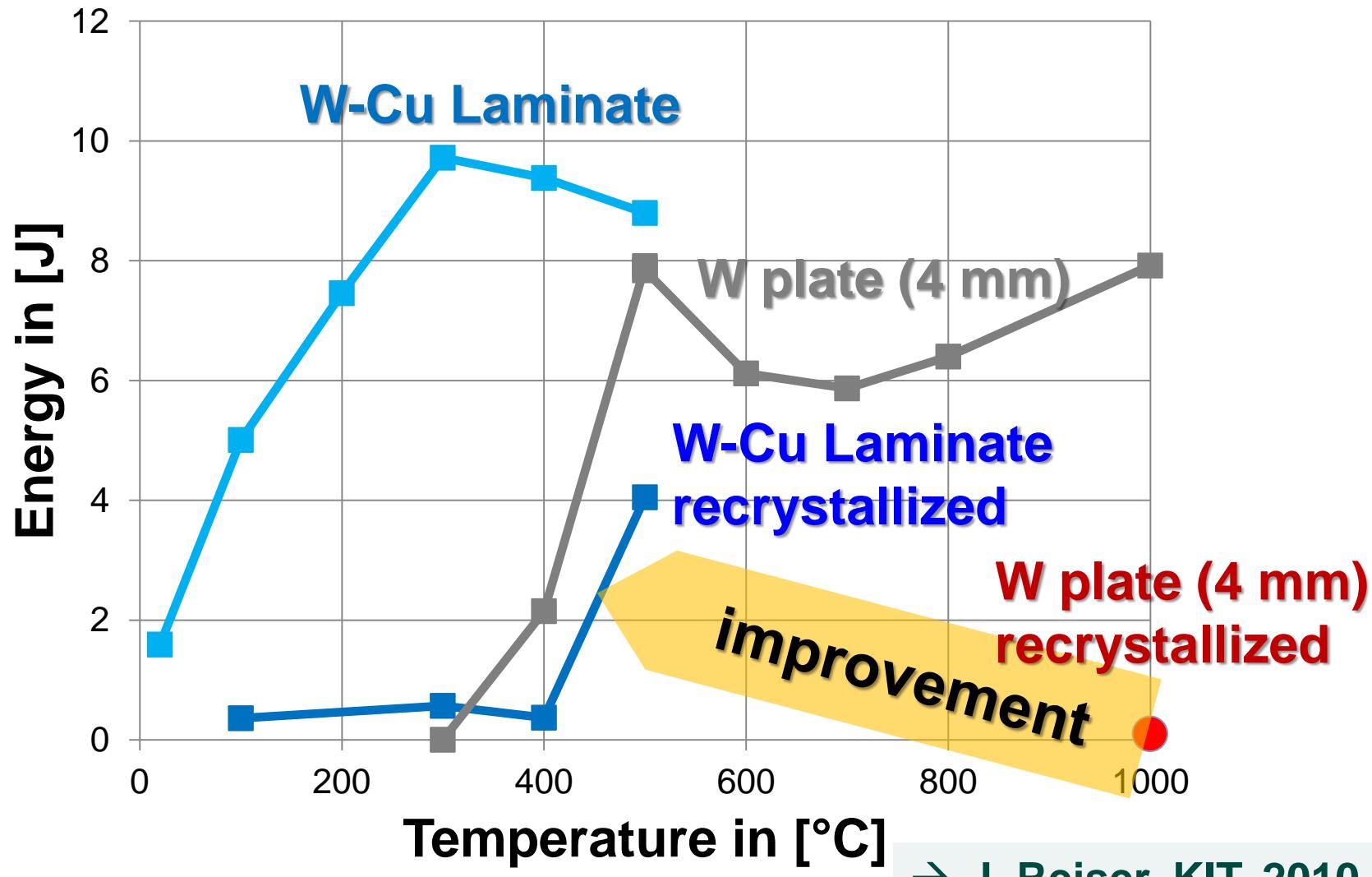


→ J. Reiser, KIT, 2010-2011

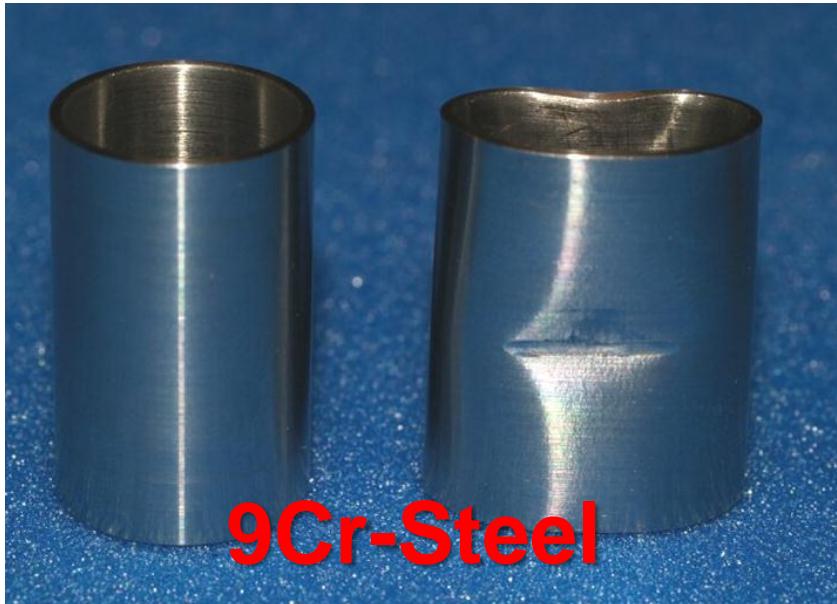
# PATH FORWARD → TUNGSTEN ALTERNATIVES



# PATH FORWARD → TUNGSTEN ALTERNATIVES



# PATH FORWARD → TUNGSTEN ALTERNATIVES



9Cr-Steel



W (rod)



→ J. Reiser, KIT, 2010-2011

# THE ULTIMATE DIVERTOR → GENERIC DESIGN

