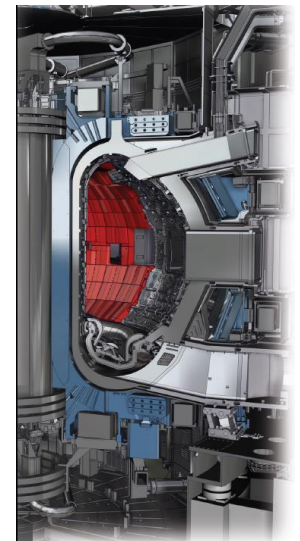
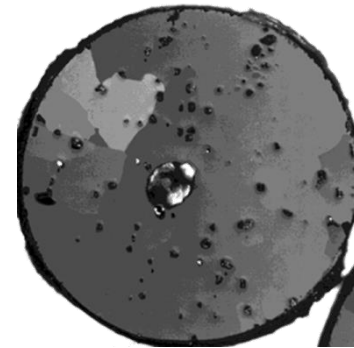
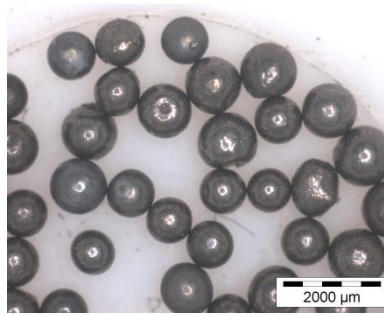
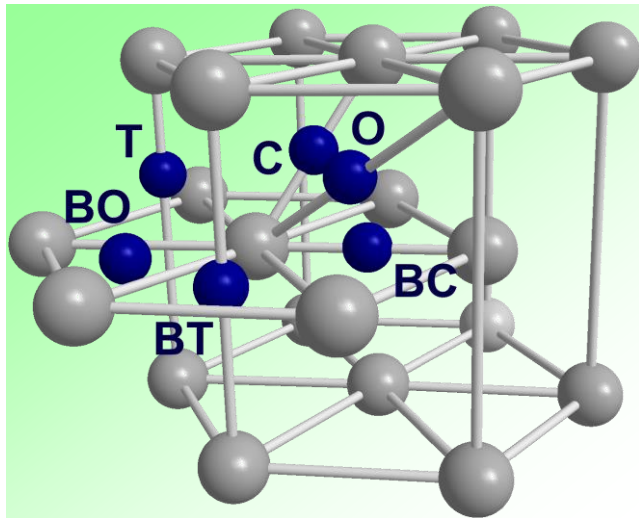


Modelling of Hydrogen Interactions with Beryllium Surfaces in Fusion Reactor

P. Vladimirov, D. Bachurin, Ch. Stihl

INSTITUTE OF APPLIED MATERIALS (IAM-AWP), Atomistic Modeling and Validation Group,
Department of Metallic Alloys



Outline

- **Beryllium in Fusion Reactor**
 - Be as plasma facing material
 - Be in T-breeding pebble bed
- **Simulation method**
- **Adsorption and desorption**
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 - H₂ adsorption on precovered surface
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 - Hydrogen associative desorption
- **Surface energy modification**
 - Clean surface energies
 - Surface energy with hydrogen
 - Faceting of H-covered bubble
- **Conclusions**

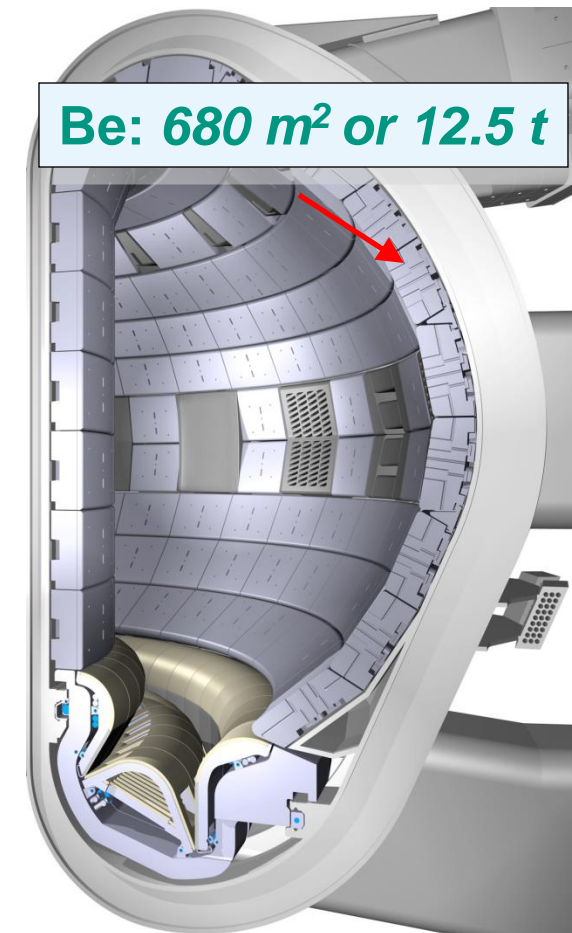
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Beryllium in fusion reactor

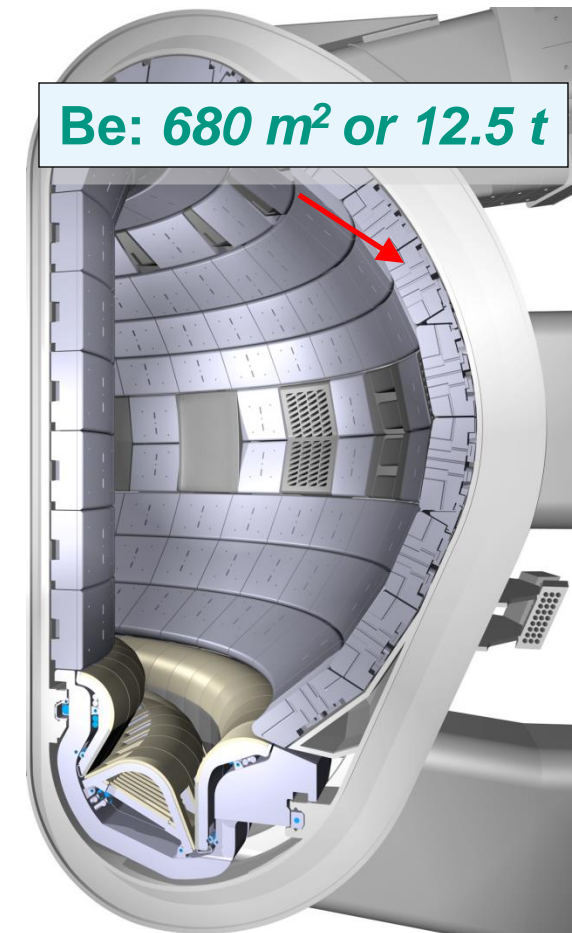


ITER Torus
cross-section

K.loki et al. Nucl.Fusion 41(3) 2001 265-275

Beryllium in fusion reactor

- Be is considered as **plasma facing material** and as effective **neutron multiplier** for tritium breeding blanket (HCPB)

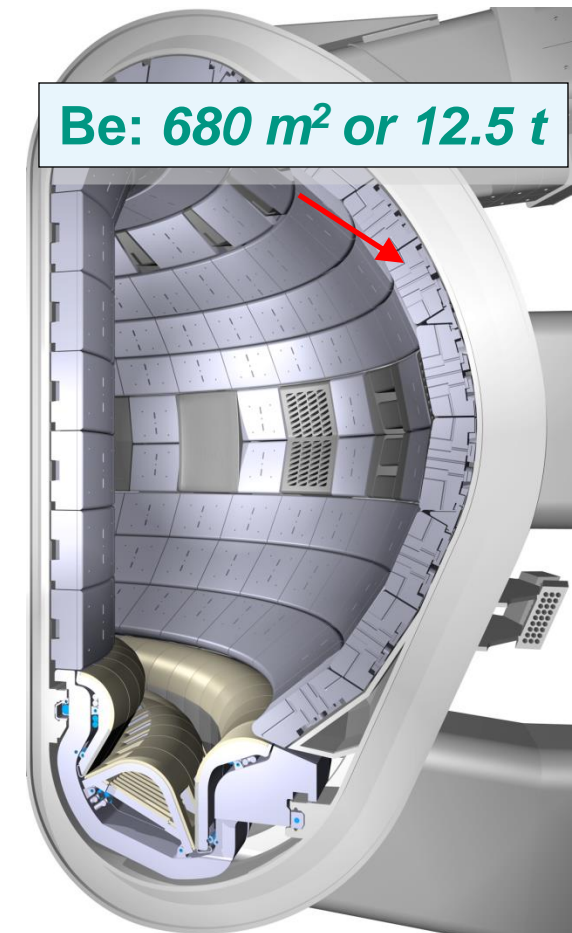


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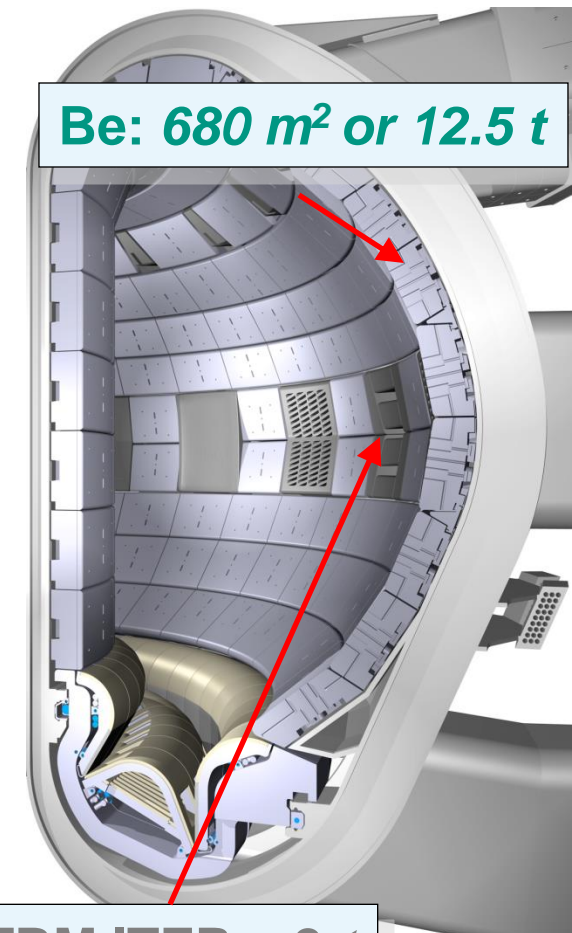


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Be: 680 m² or 12.5 t

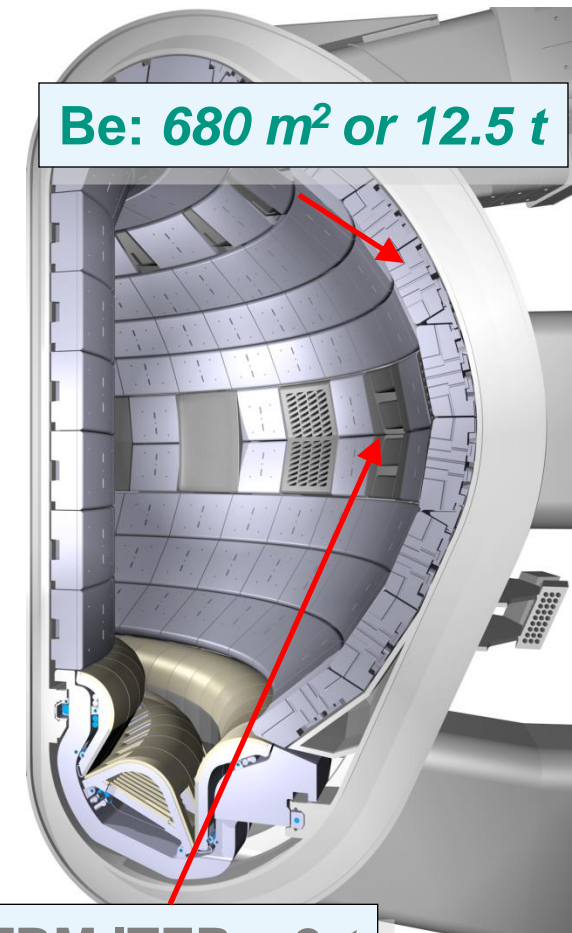
TBM ITER – 2 t
DEMO – 300 t

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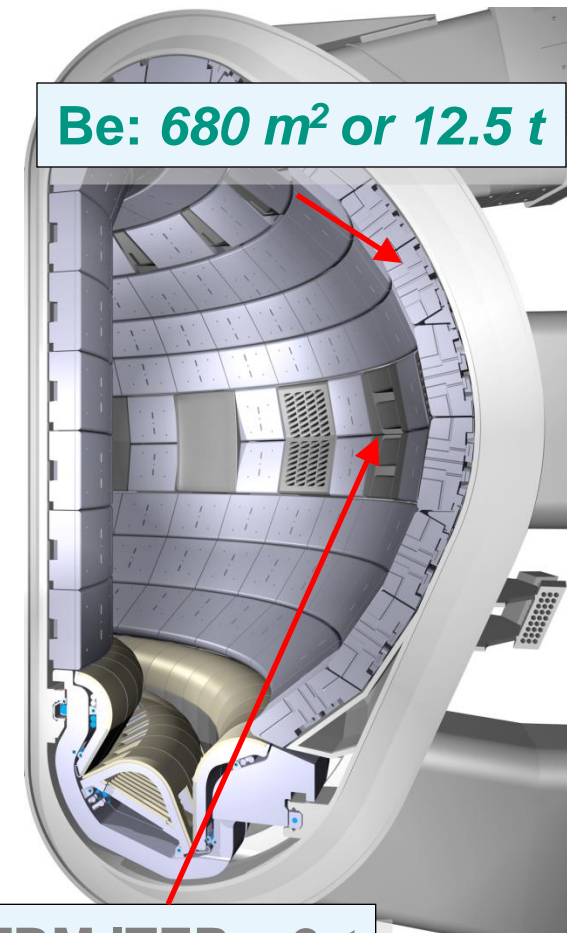
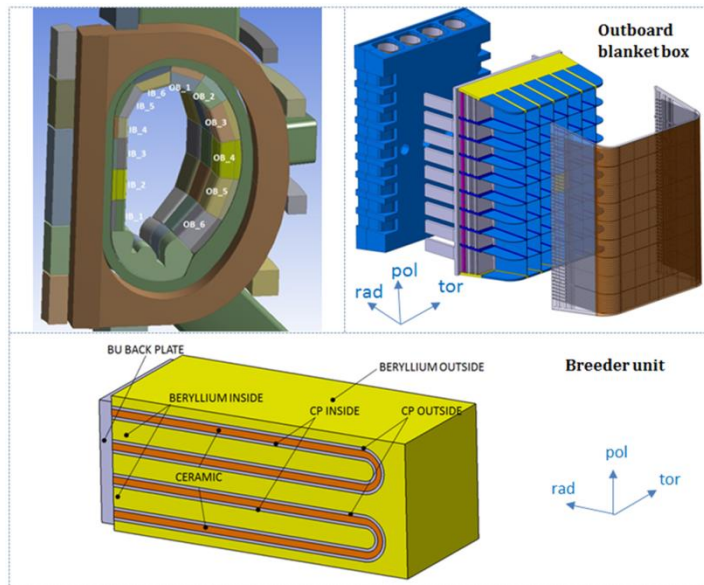
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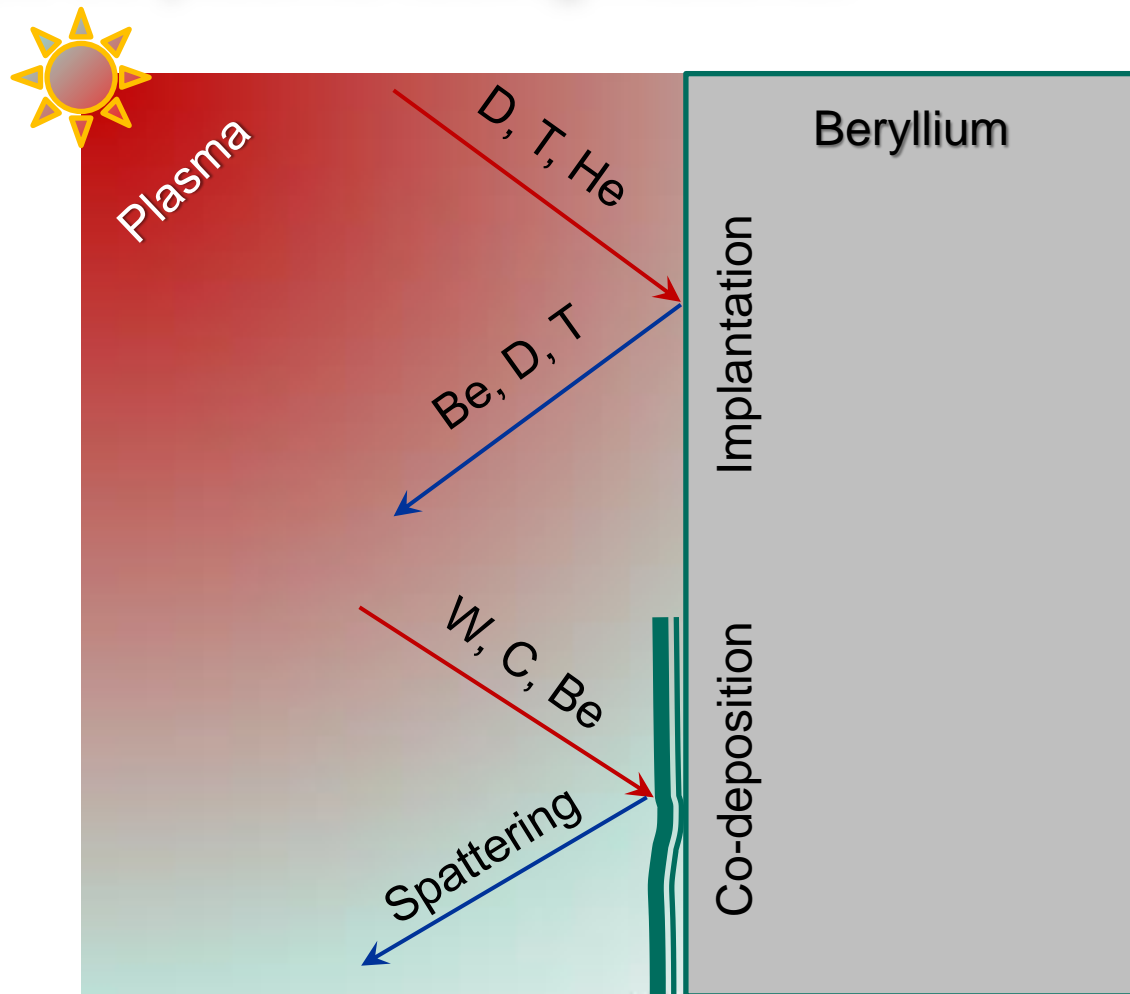
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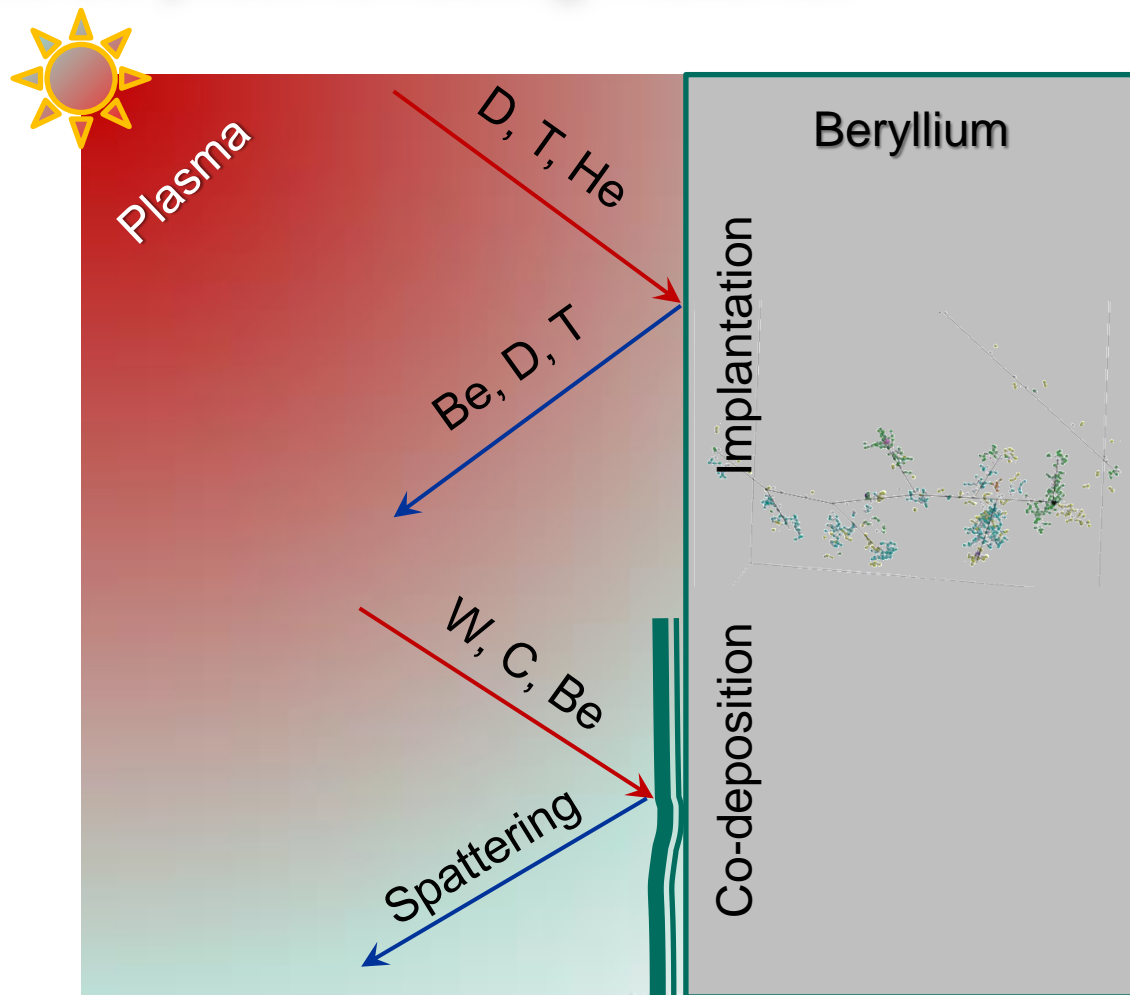
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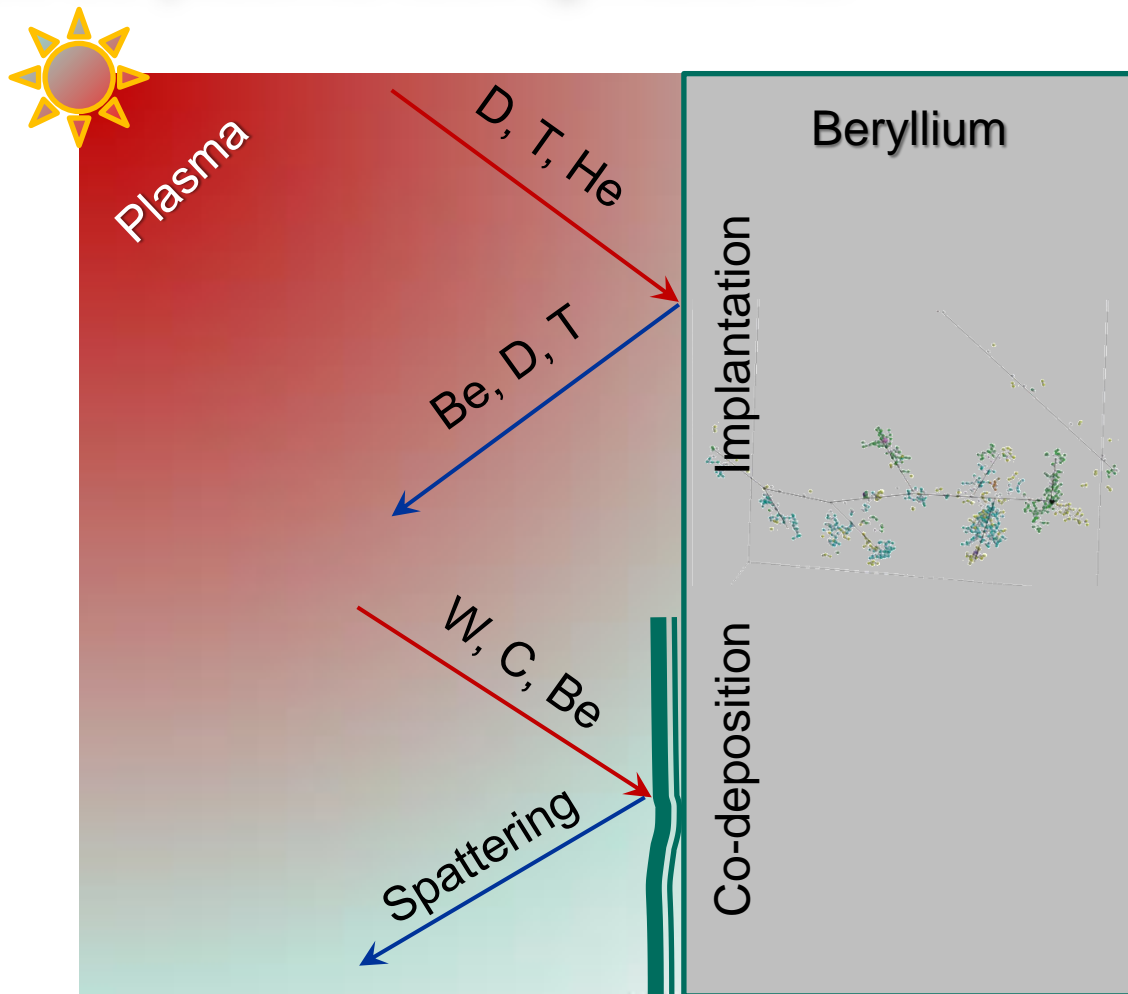
Be as plasma facing material



Be as plasma facing material



Be as plasma facing material



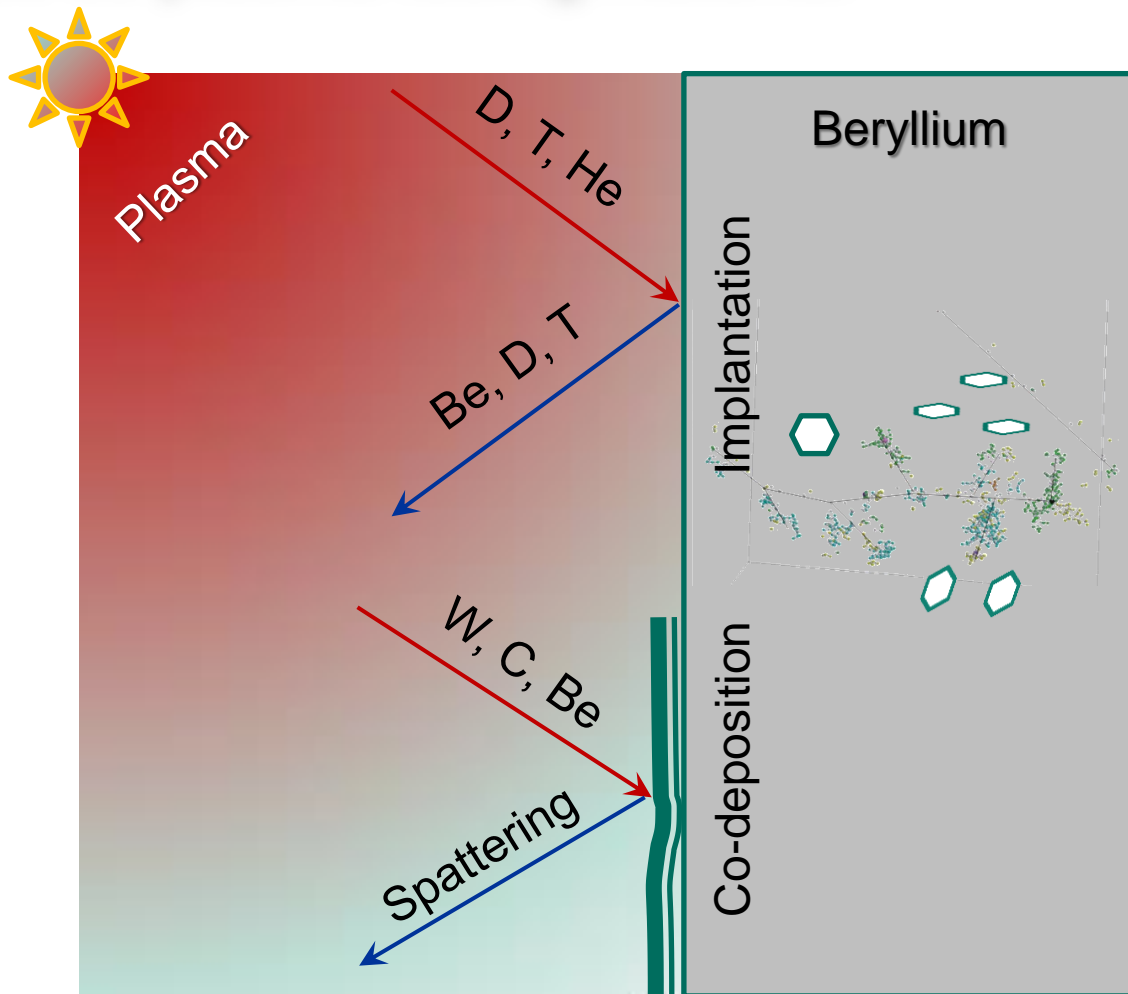
$E_n \geq 2.7 \text{ MeV}$:



$E_n \geq 0.71 \text{ MeV}$



Be as plasma facing material



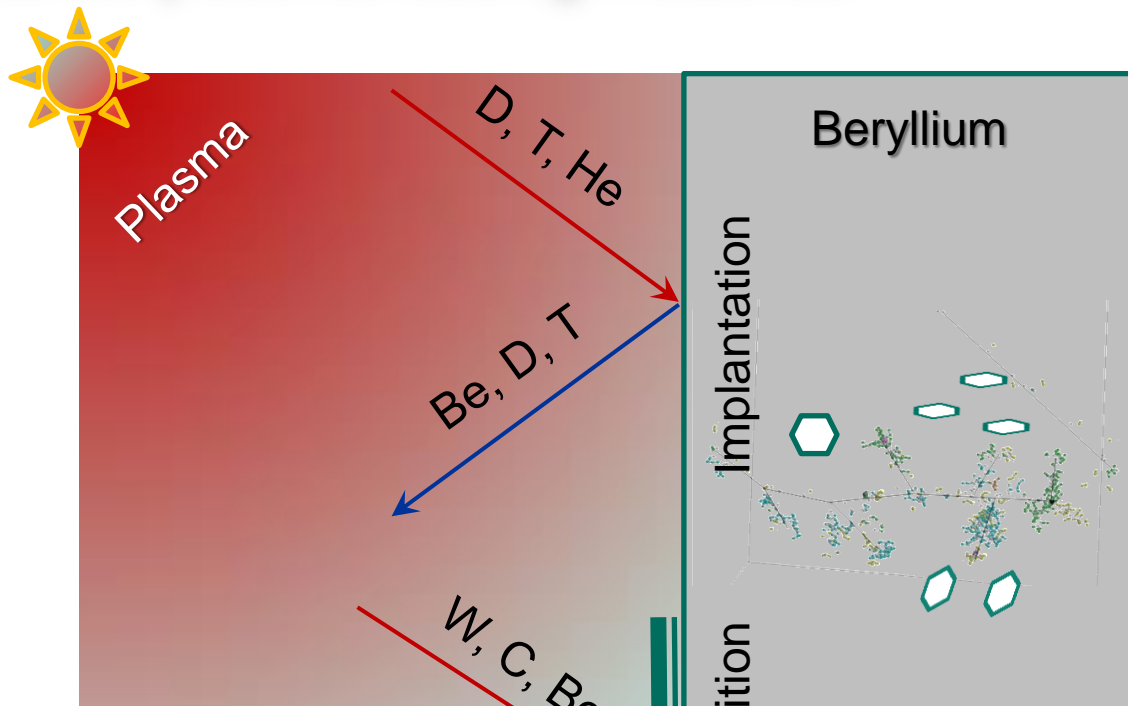
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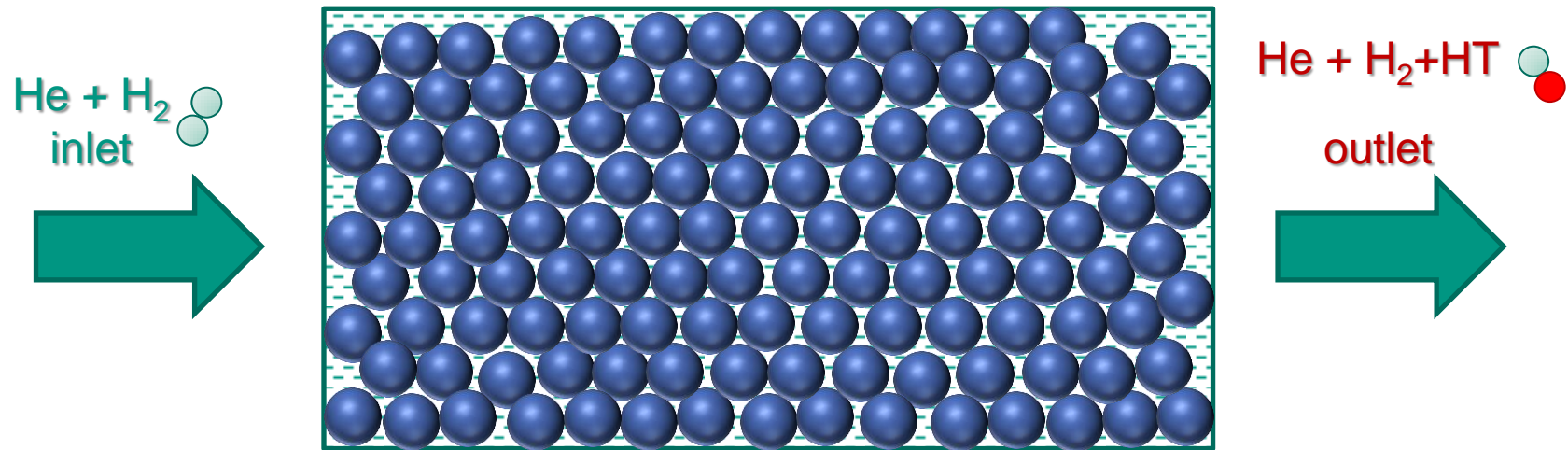


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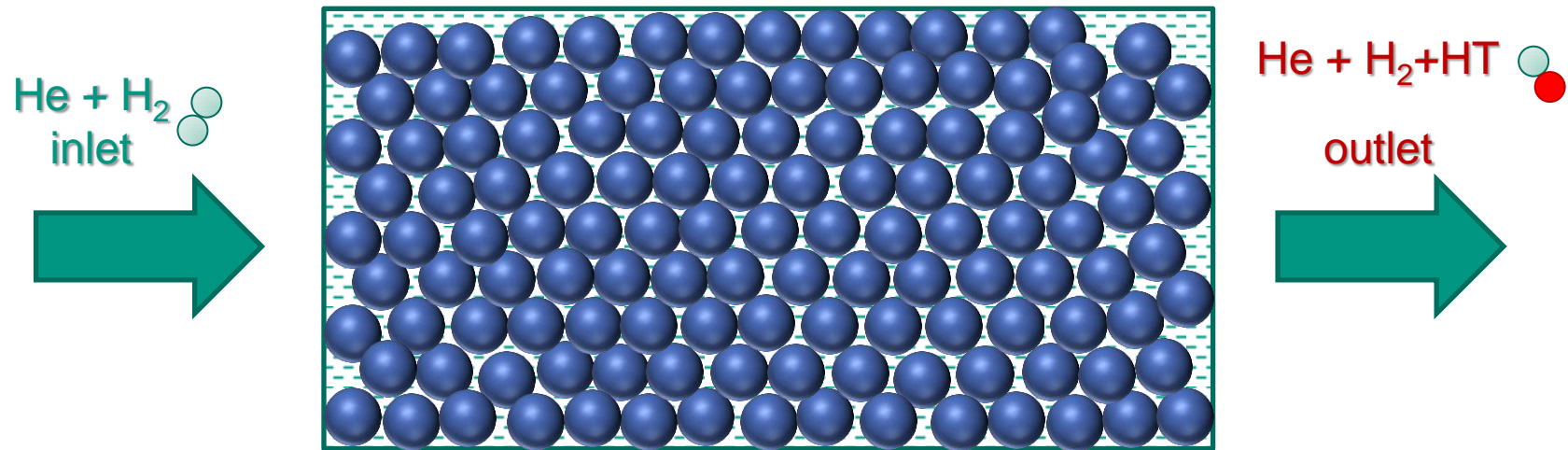


- Helium and hydrogen isotopes are produced in Be by nuclear transmutations as well as implanted from the hot plasma.
- He and H can be trapped within vacancies and vacancy clusters produced by neutron irradiation and facilitate formation of gas filled bubbles.

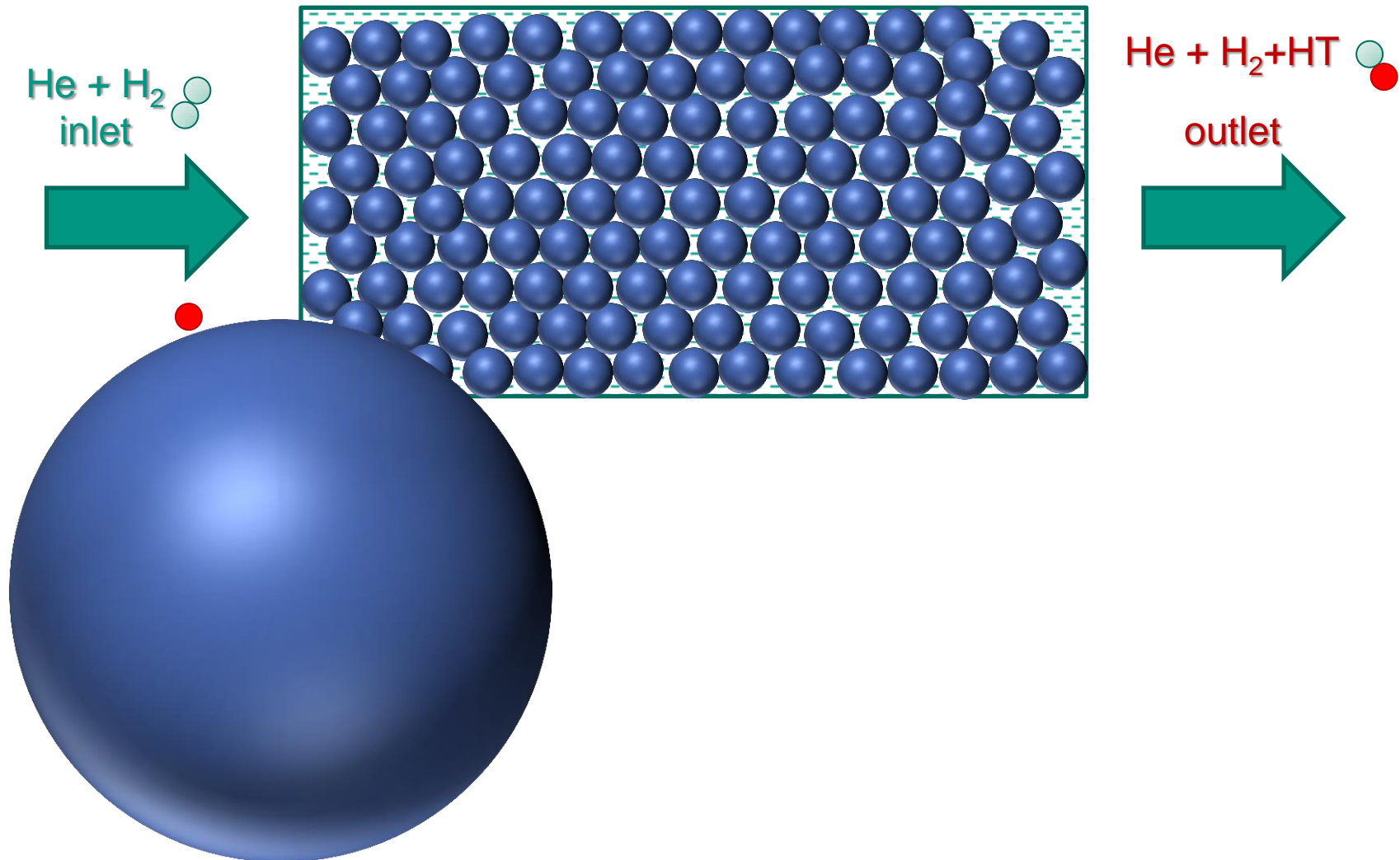
Be Pebble Bed of T-breeding Blanket



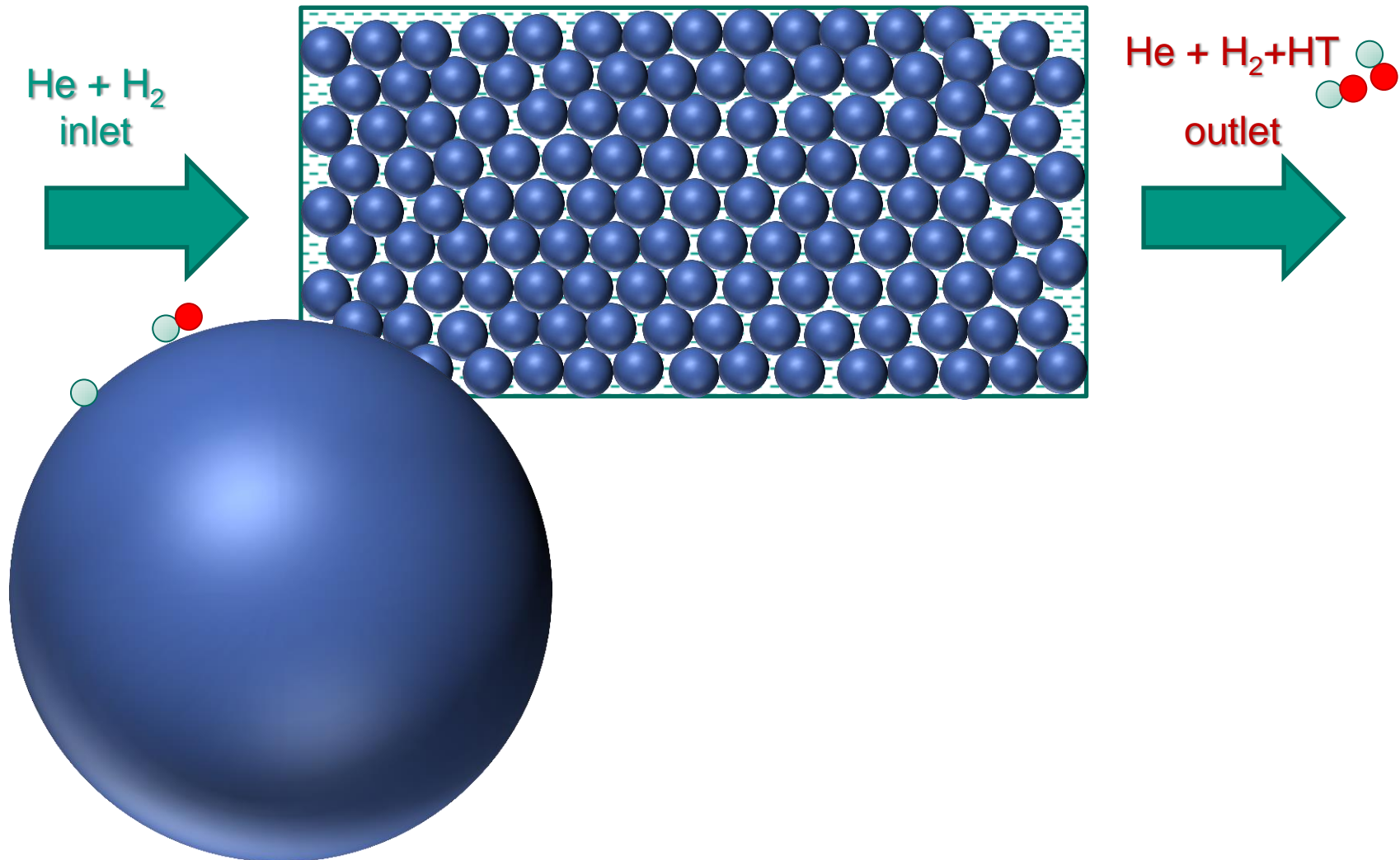
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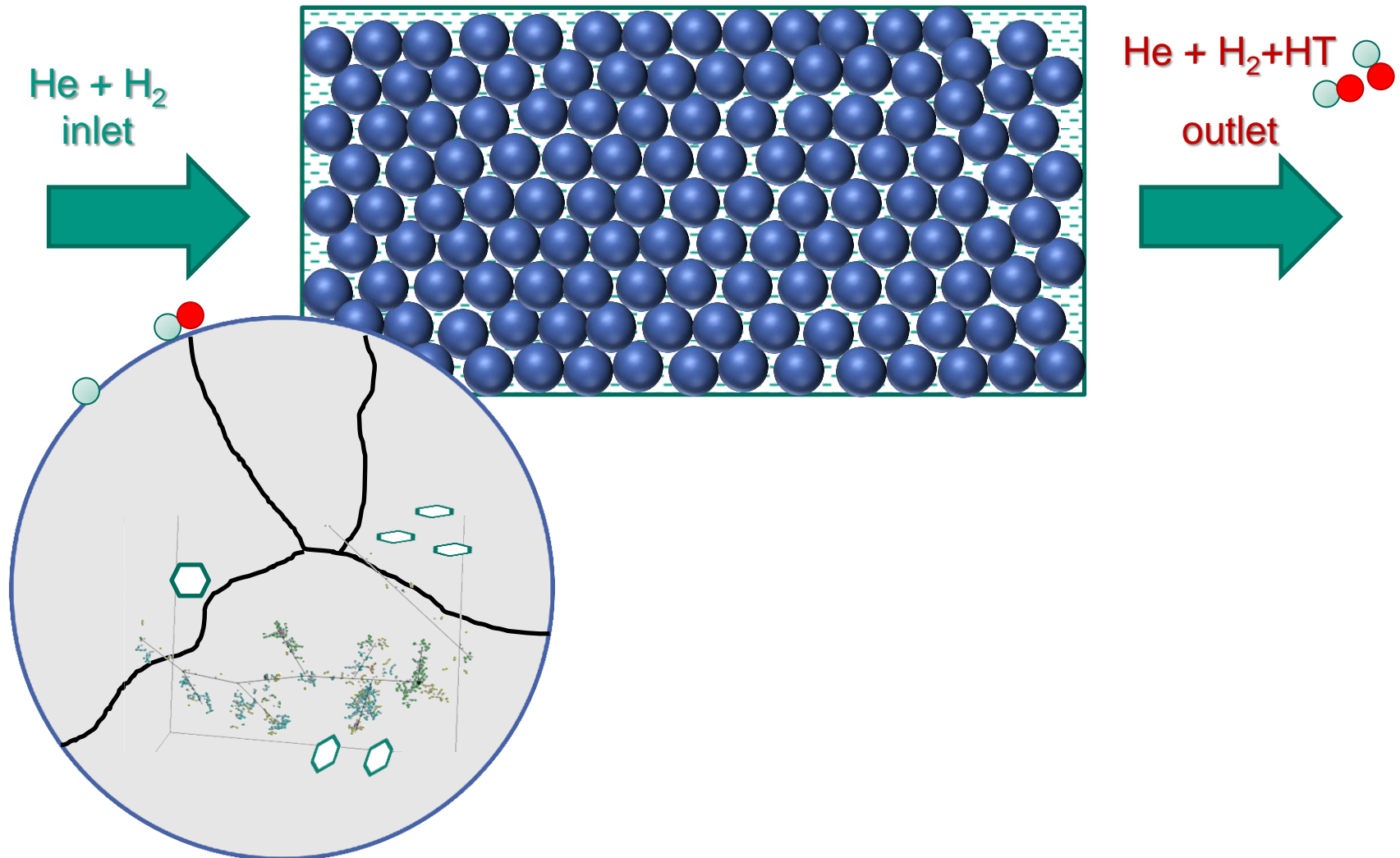
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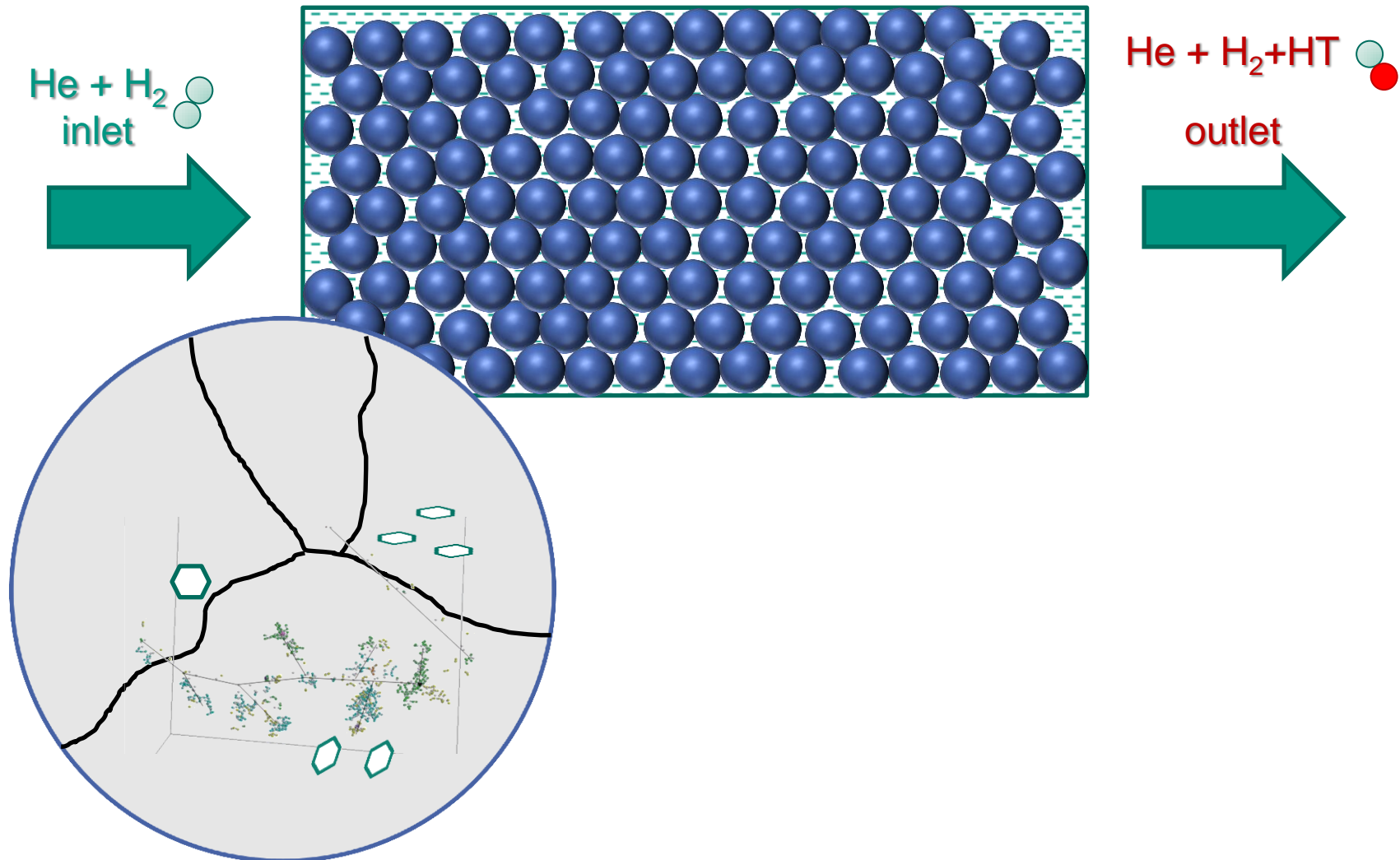
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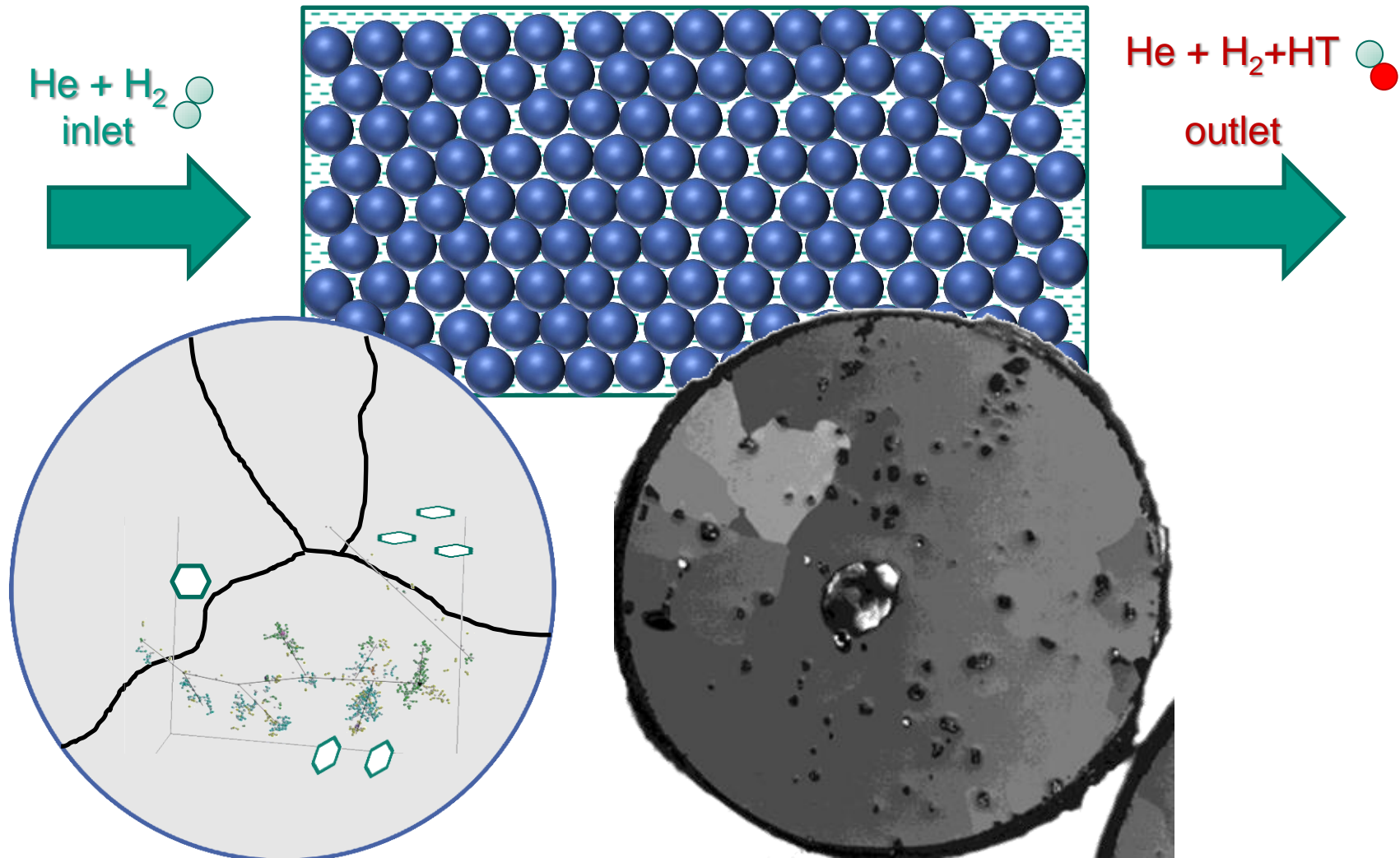
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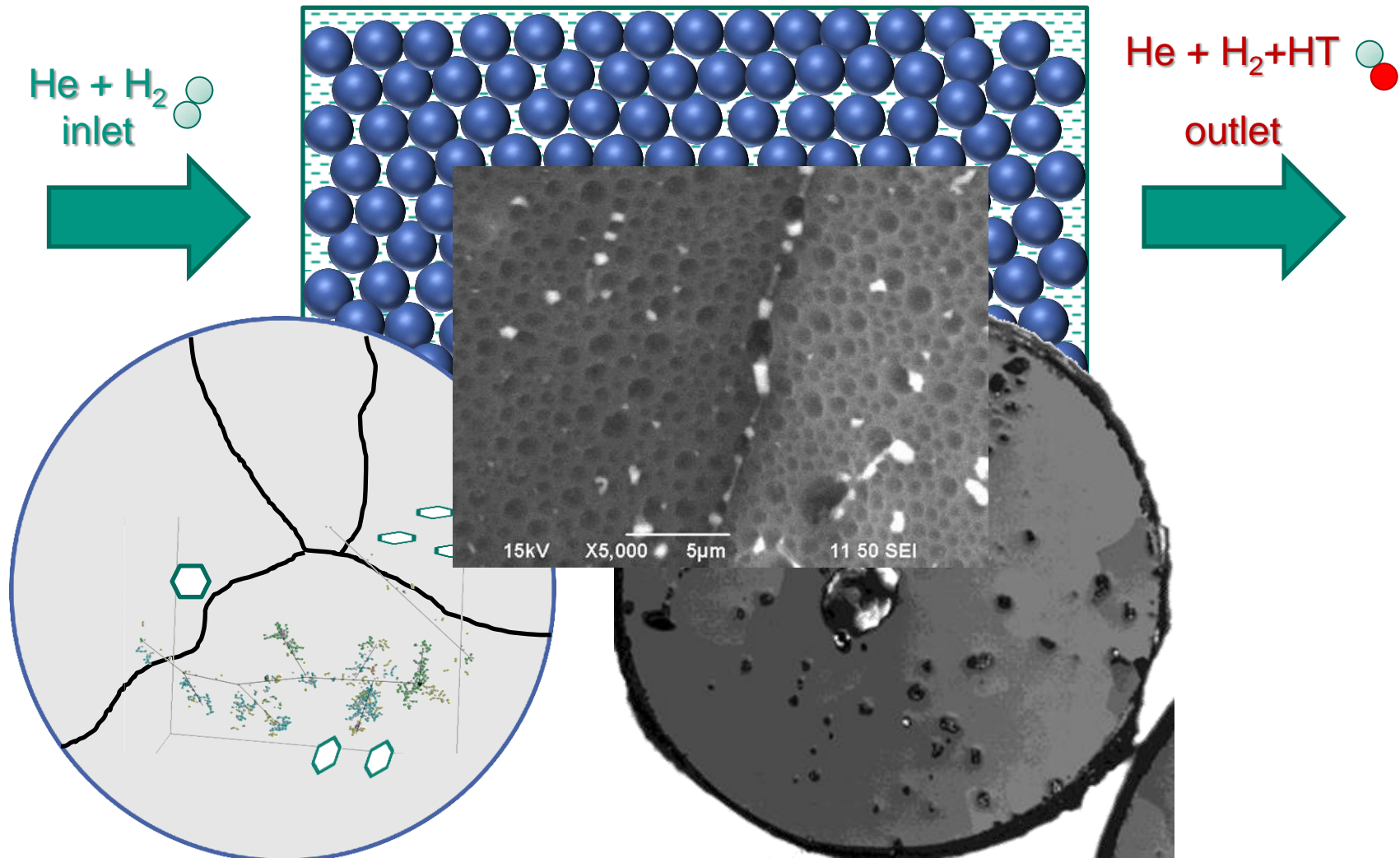
Be T-breeding Pebble Bed



Be T-breeding Pebble Bed



Be T-breeding Pebble Bed



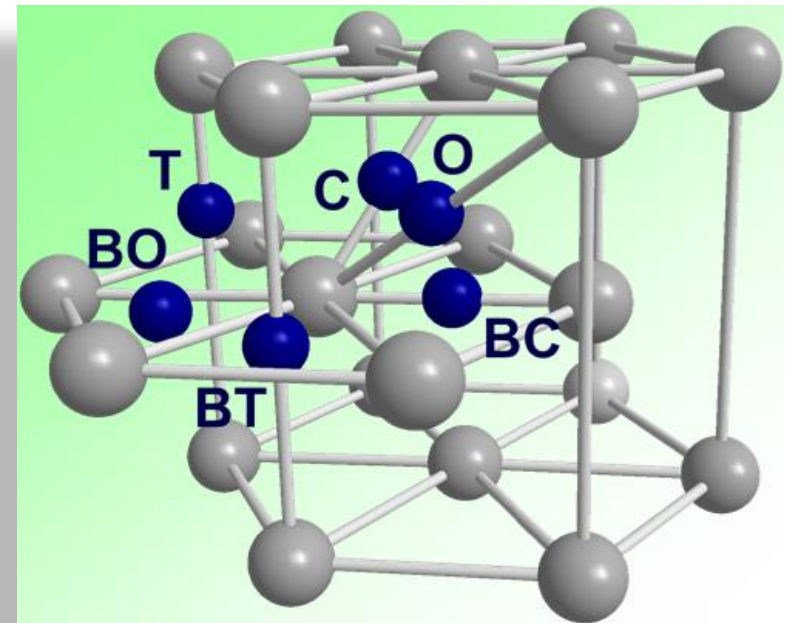
Simulation methods

Density Functional Theory (*ab initio*)

- VASP 4.6 / VASP 5.2
- Generalized Gradient Approximation (GGA)
- Pseudopotentials:
 - Plain Augmented Waves (PAW)
- Gamma centered Monkhorst-Pack k-point grid $\geq 13 \times 13 \times 13$
- Energy cutoff = 450 eV

Ab initio Molecular Dynamics (VASP)

- Time step 0.3 fs
- Run duration ~3000-4000 steps (~1 ps)
- Simulation cell size: $4 \times 4 \times 2 = 64$ atoms
- Temperature: 200-1000 K
- k-point grid: $7 \times 7 \times 7$
- Energy cutoff 250 eV

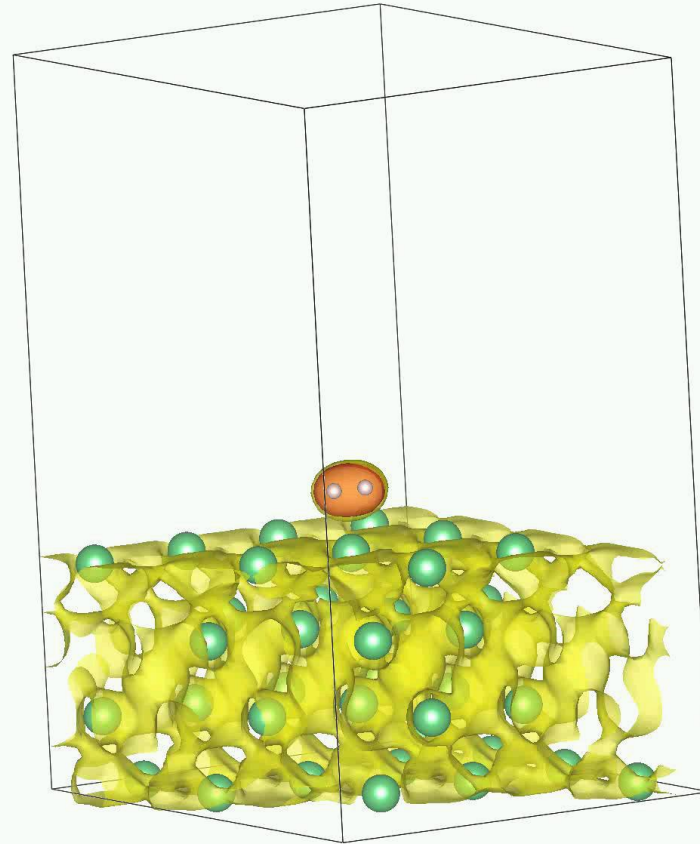


G.Kresse, J.Hafner, Phys. Rev. B (1993) 47, 558; *ibid.* (1994) 49, 14251;
 G.Kresse, J.Furthmüller, Comput. Mat. Sci. (1996) 6, 15;
 G.Kresse and J.Furthmüller, Phys. Rev. B (1996) 54, 11169

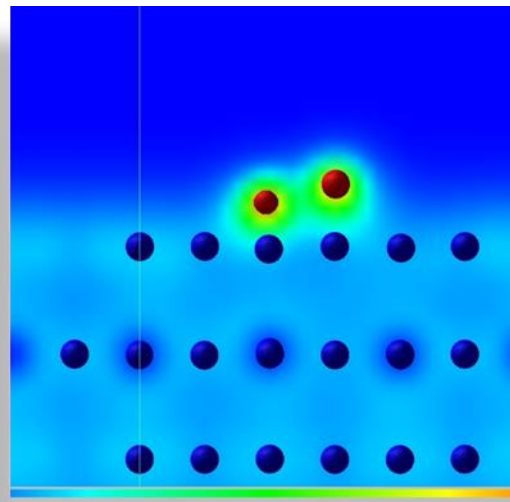
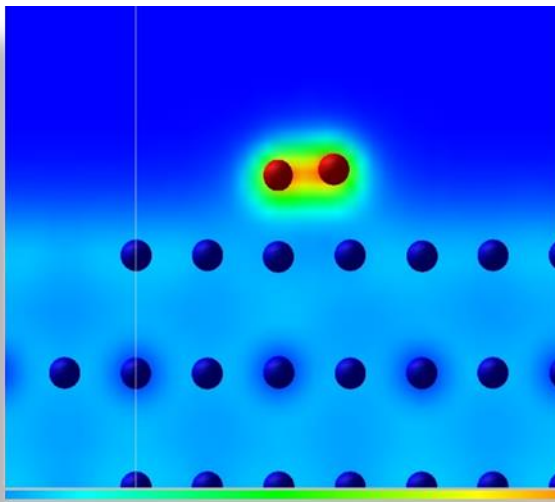
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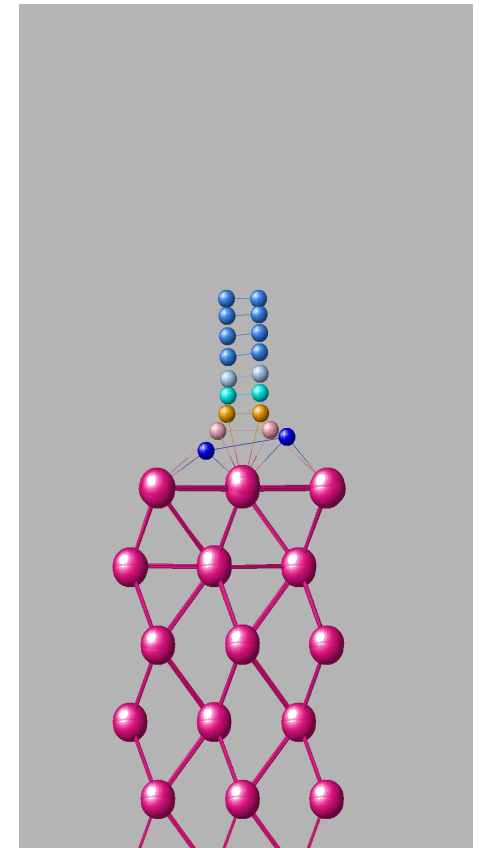
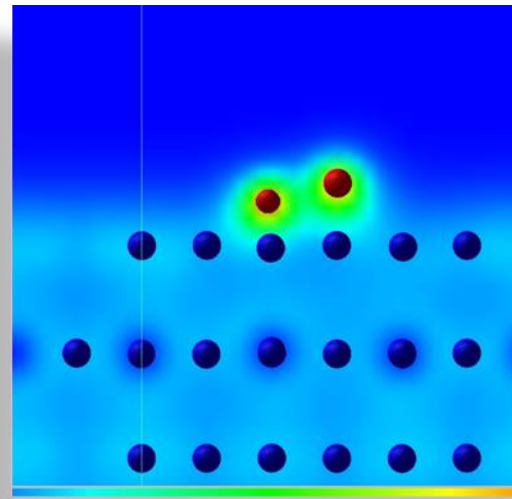
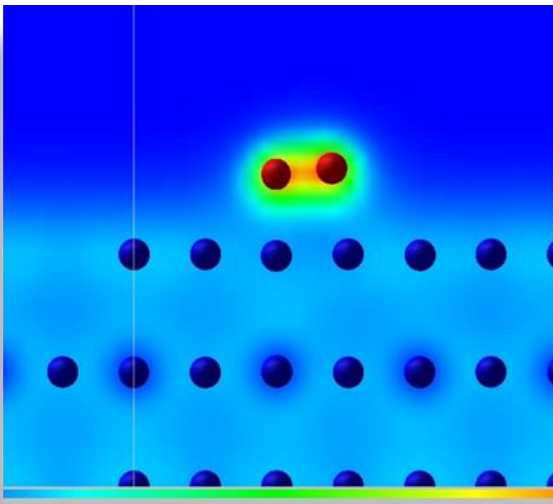
H₂ dissociative adsorption on (0001)Be



H₂ dissociative adsorption on (0001)Be

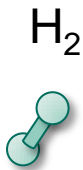


H₂ dissociative adsorption on (0001)Be

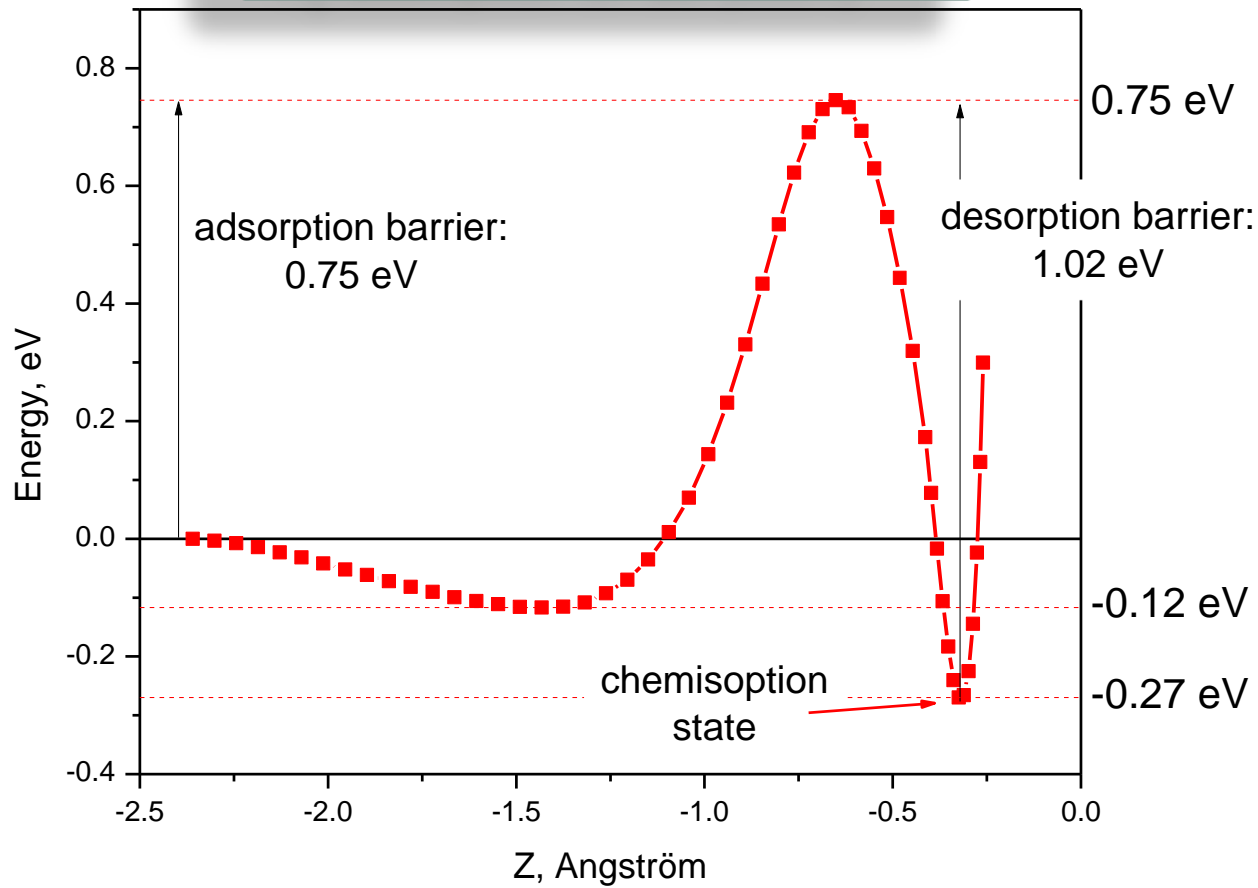


H₂ dissociative adsorption on (0001)Be

Vacuum

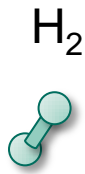


H₂ adsorption on Be (0001)

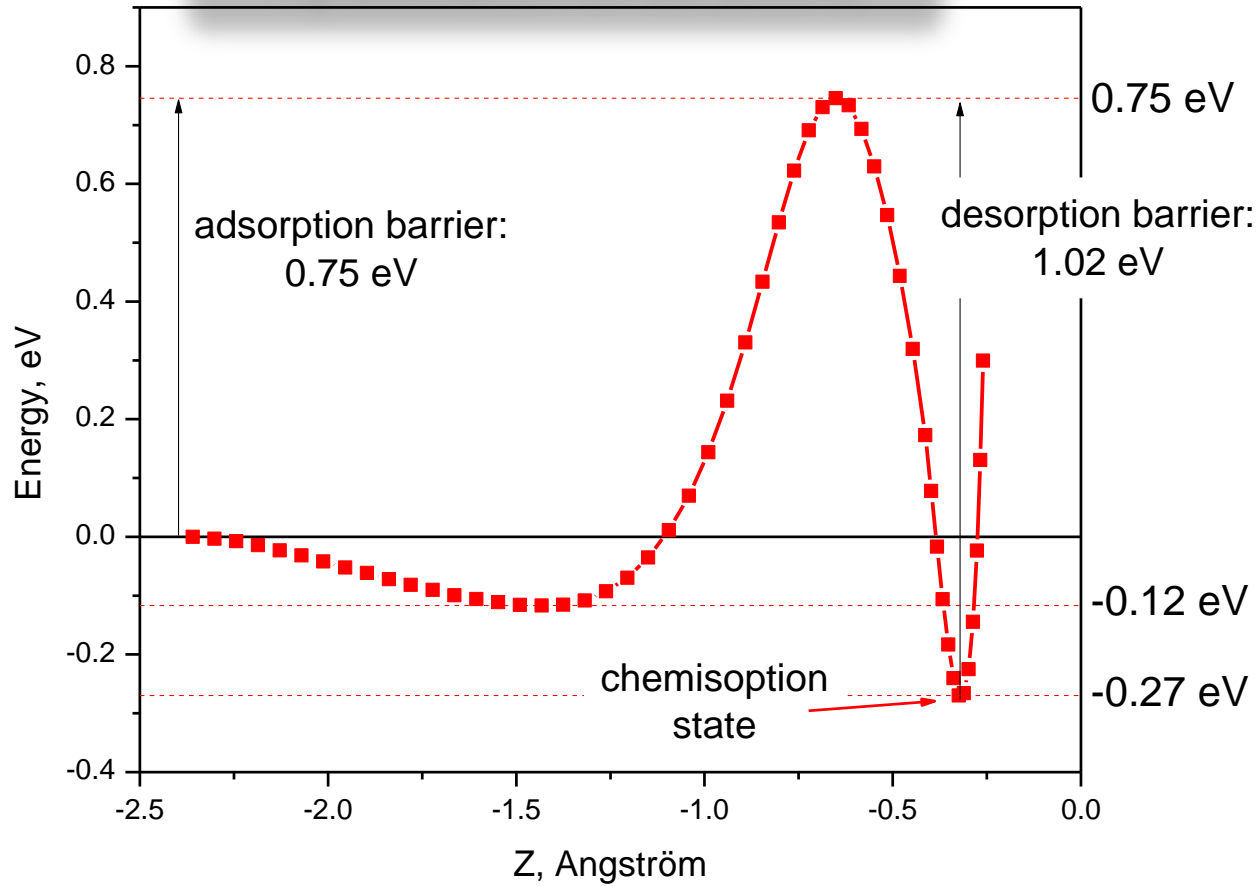


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Vacuum

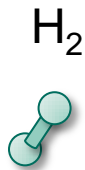


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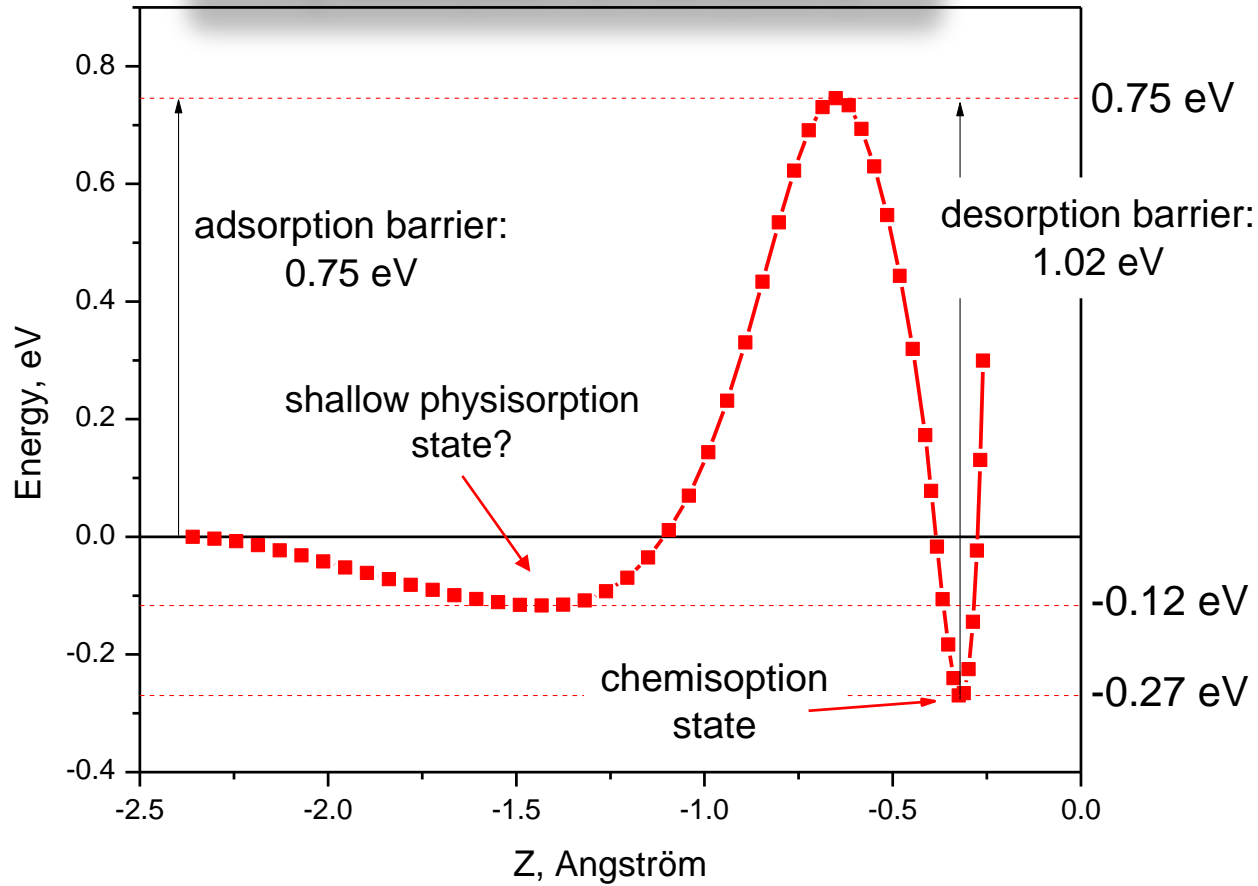


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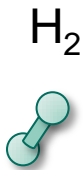


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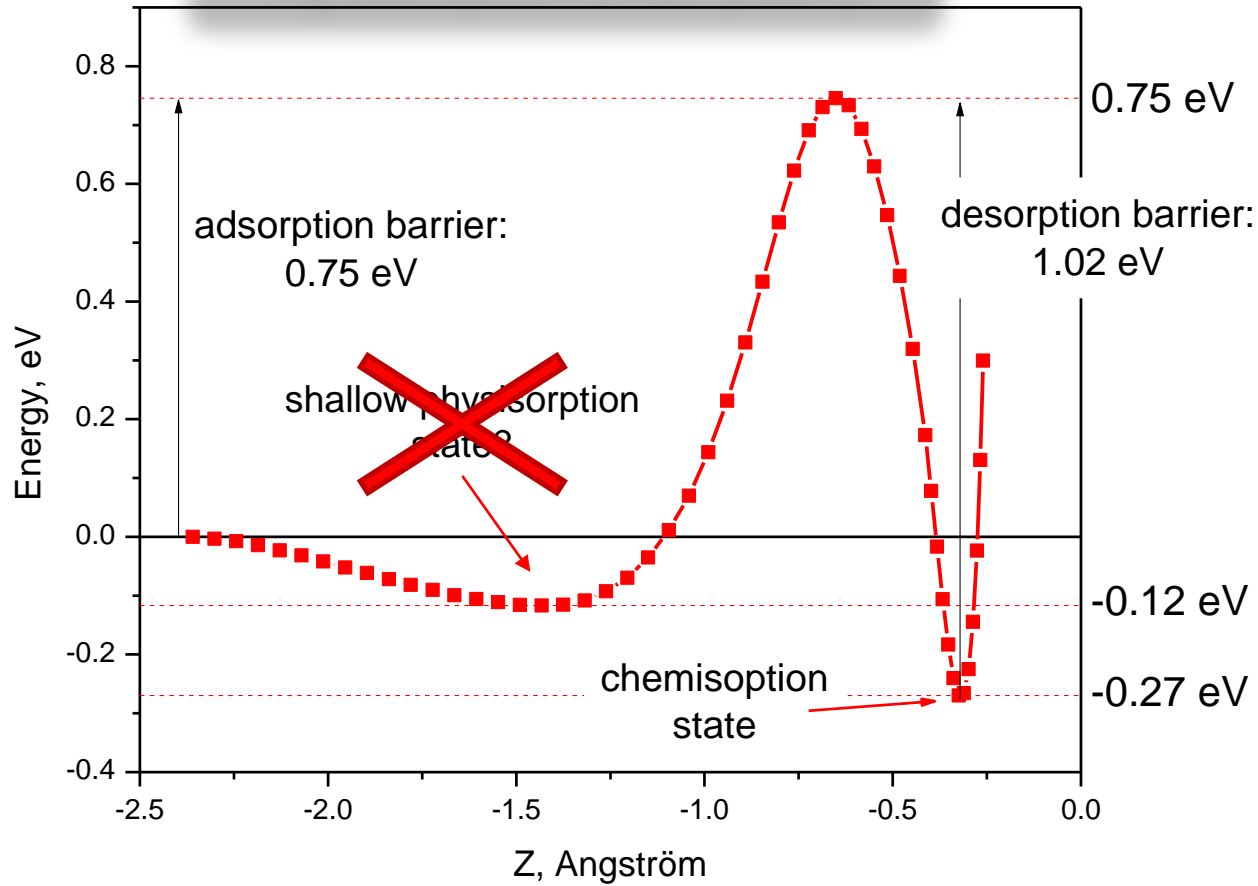


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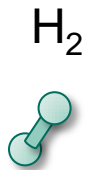


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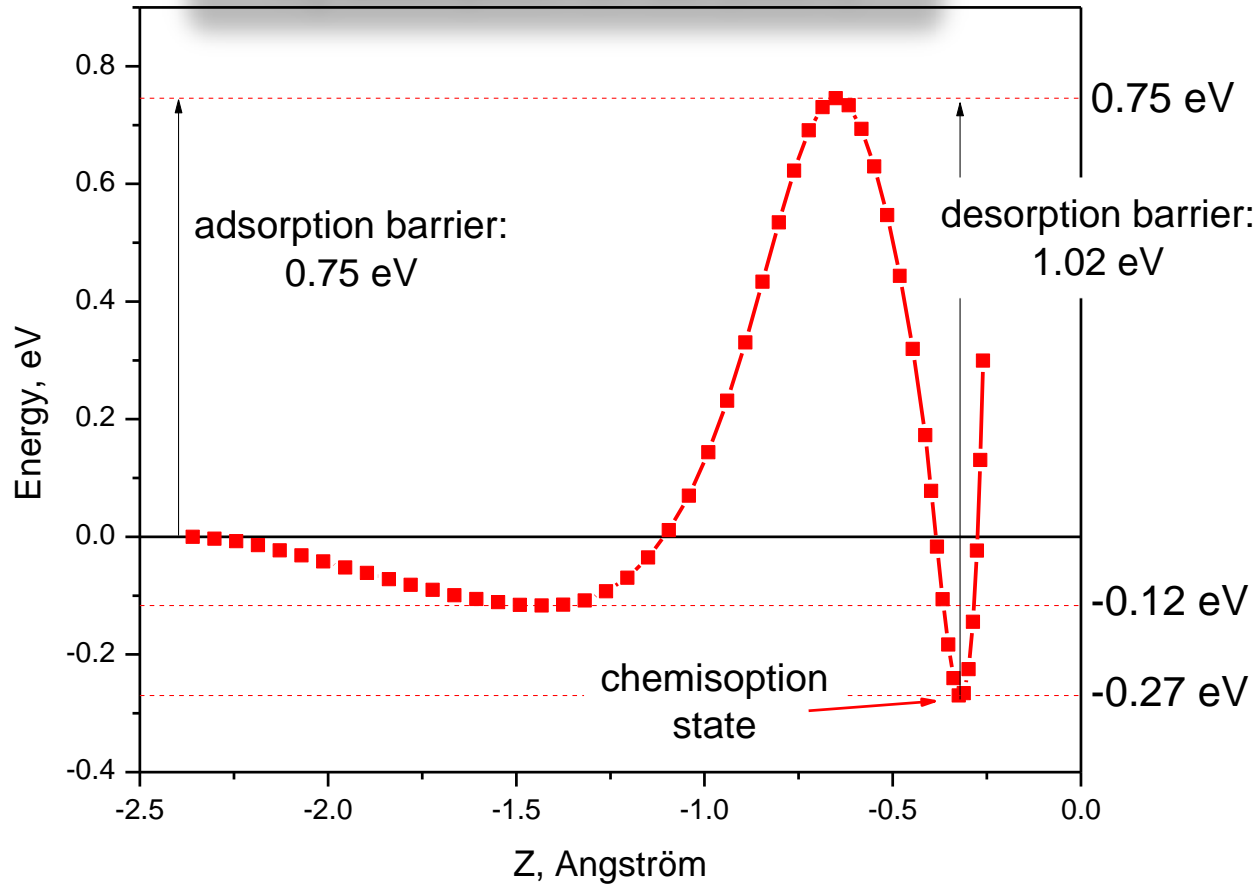


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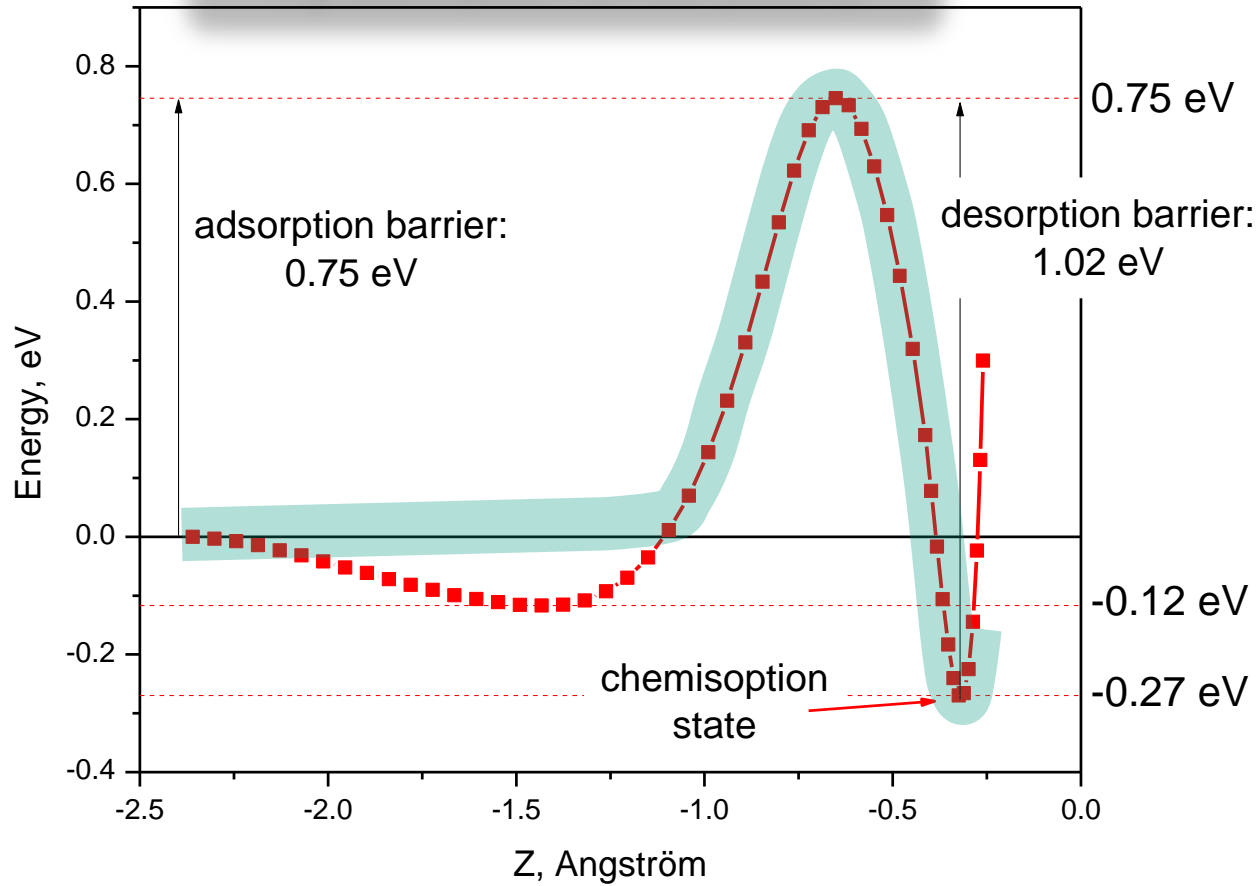


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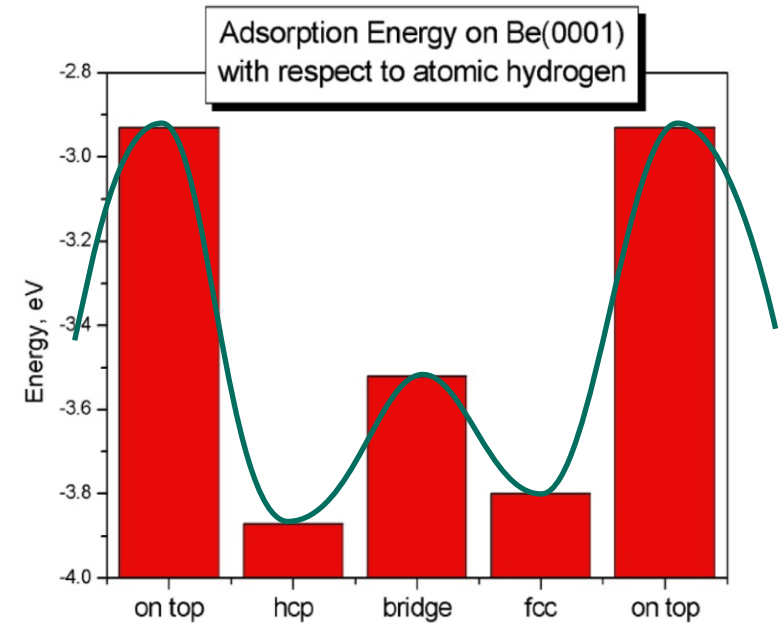
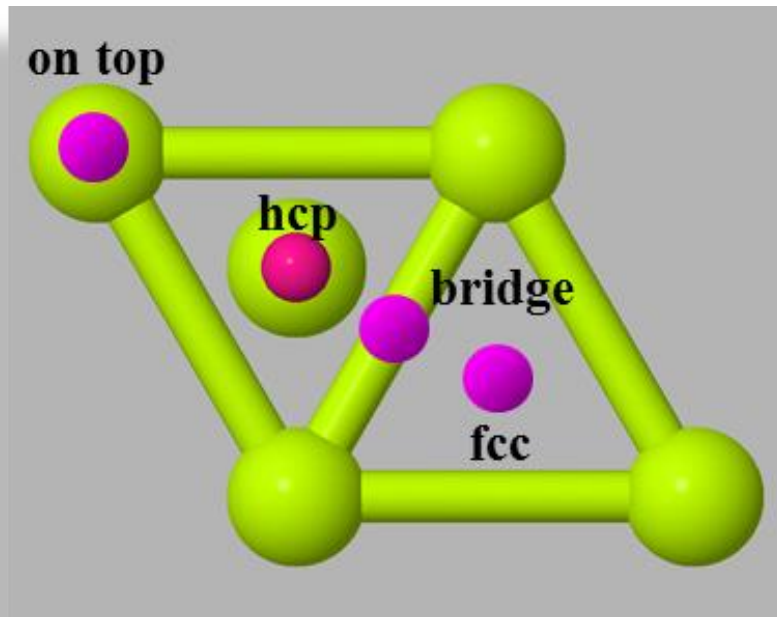
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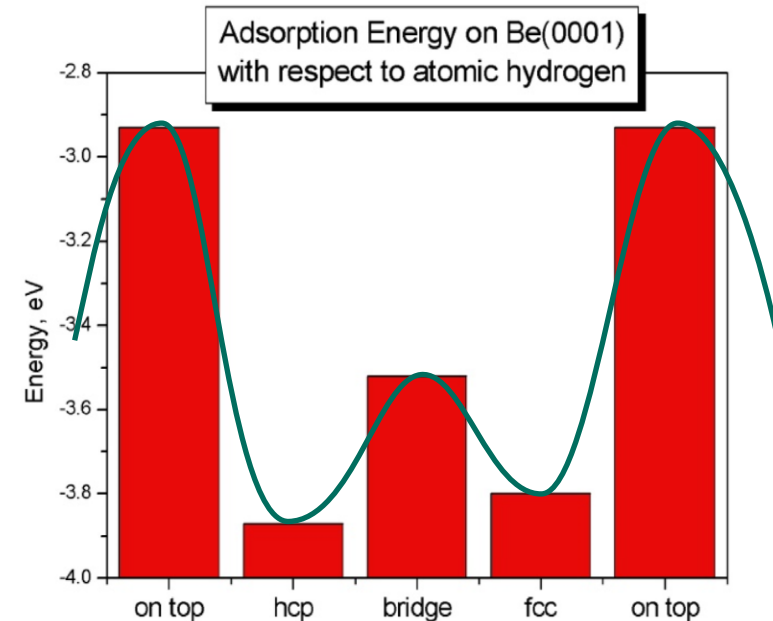
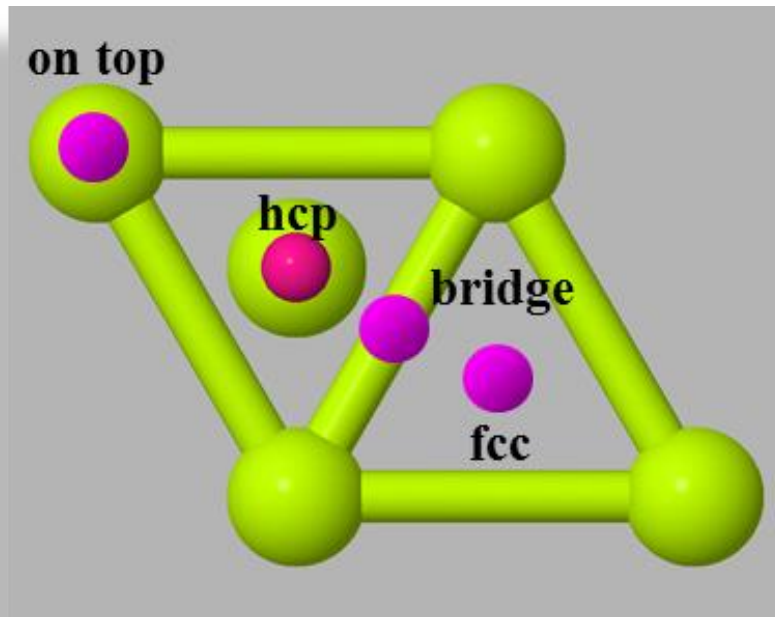
H₂ adsorption on Be (0001)



H on Be(0001) surface: Adsorption sites

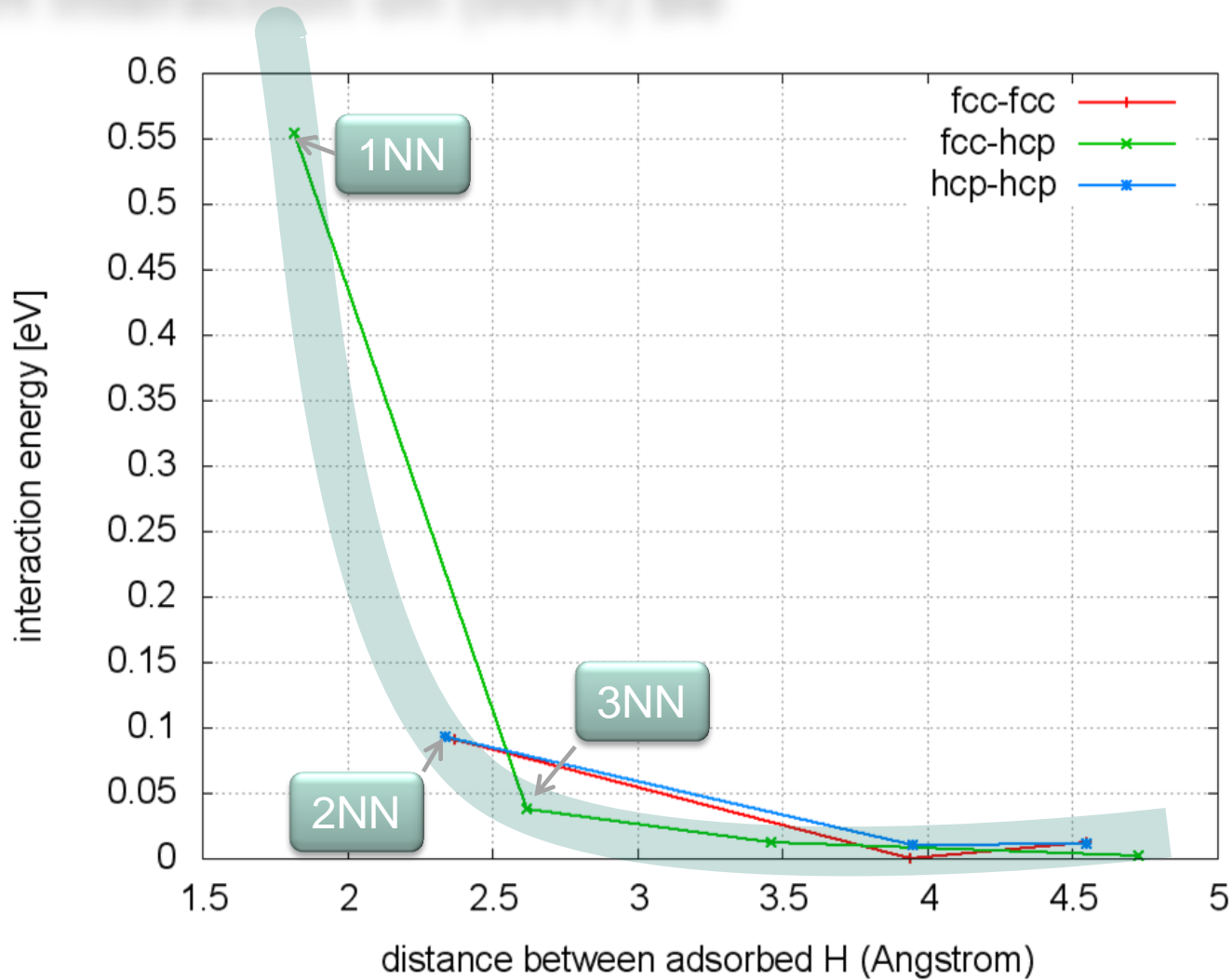


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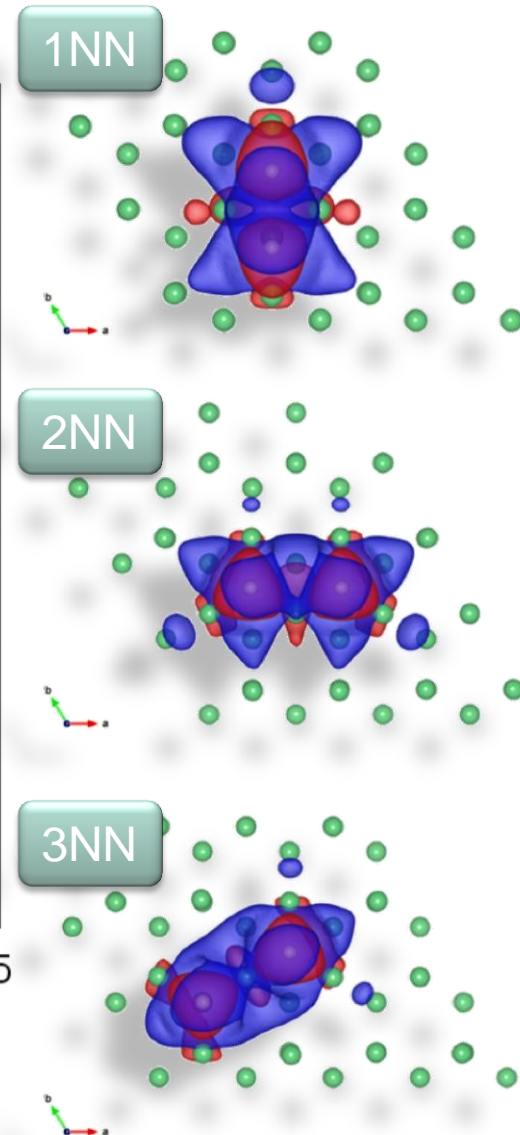
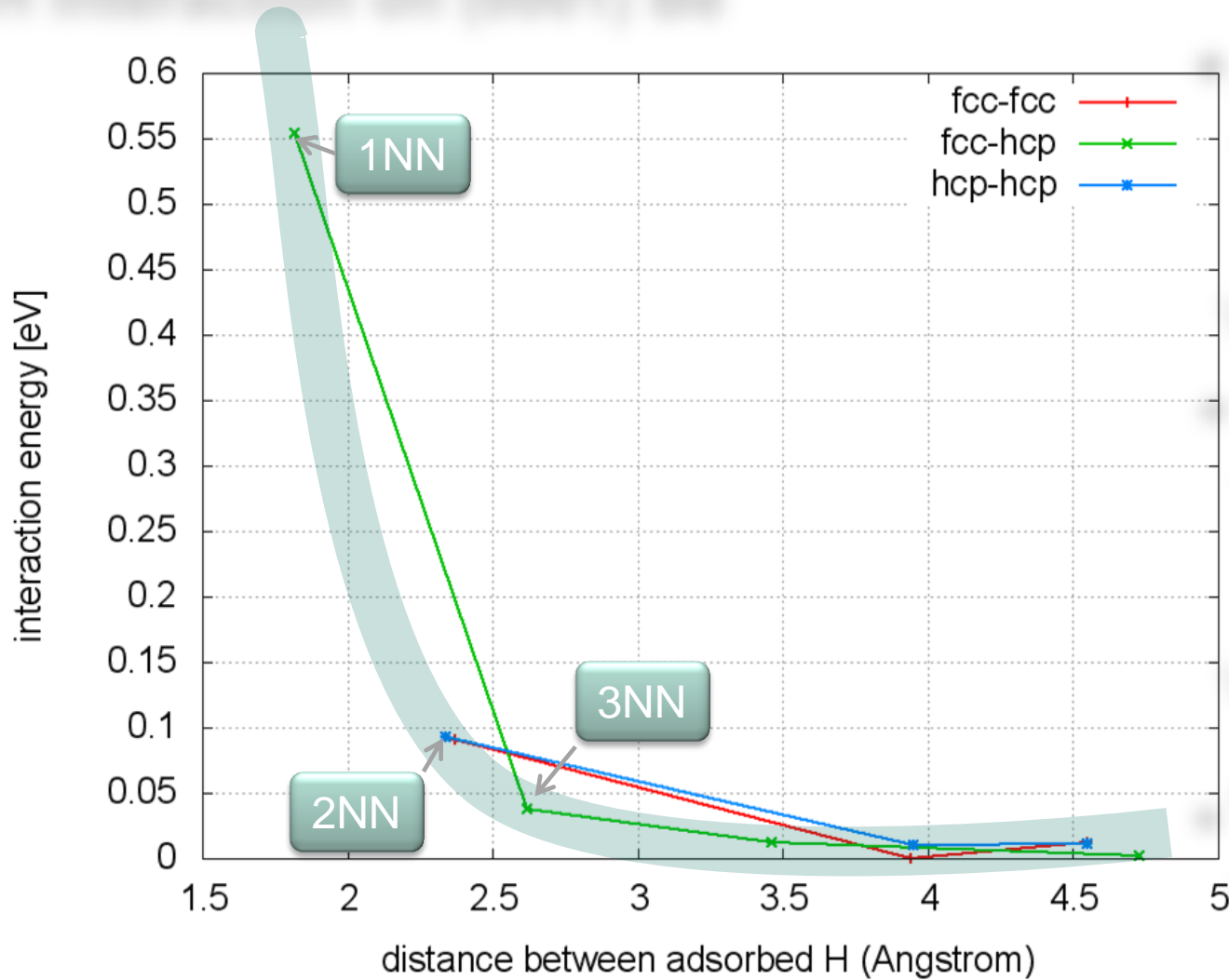


- Two stable adsorption sites for hydrogen (hcp and fcc) exist at (0001)Be surface
- Hydrogen coverage calculated as a fraction of occupied sites (1ML – all sites occupied)

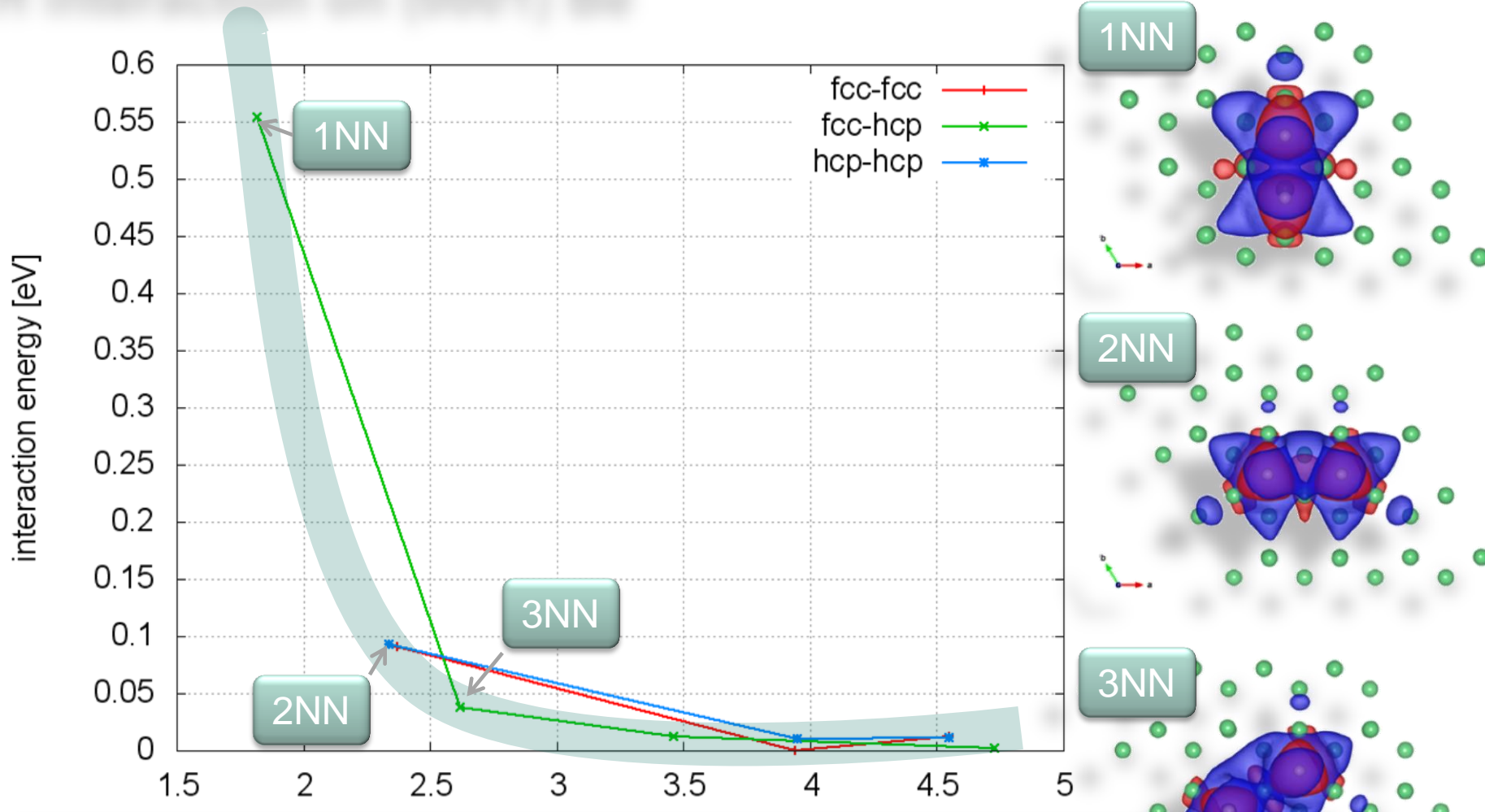
H interaction on (0001) Be



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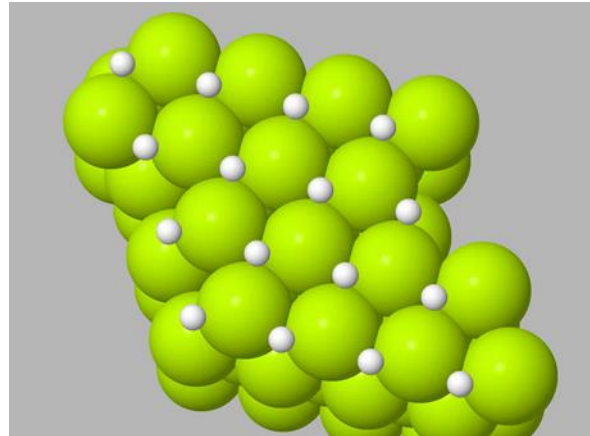


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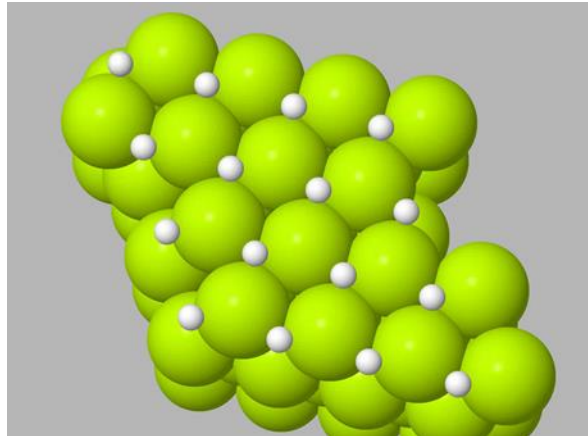


- Hydrogen atoms at the surface prefer to stay far from each other

H₂ adsorption on H pre-covered surface



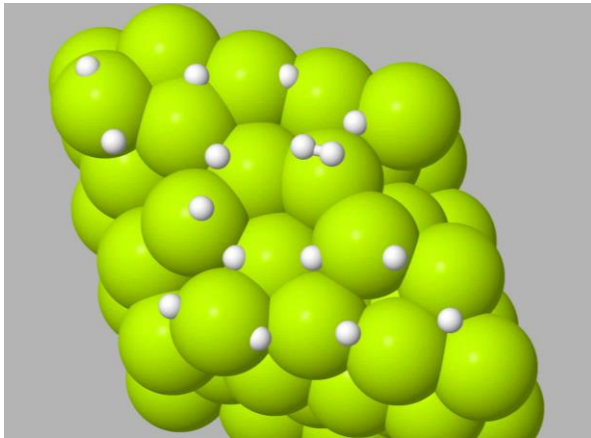
H₂ adsorption on H pre-covered surface



- There is no hydrogen adsorption on fully precovered (0.5 ML) Be (0001) surface
- One H-vacancy is also insufficient for adsorption

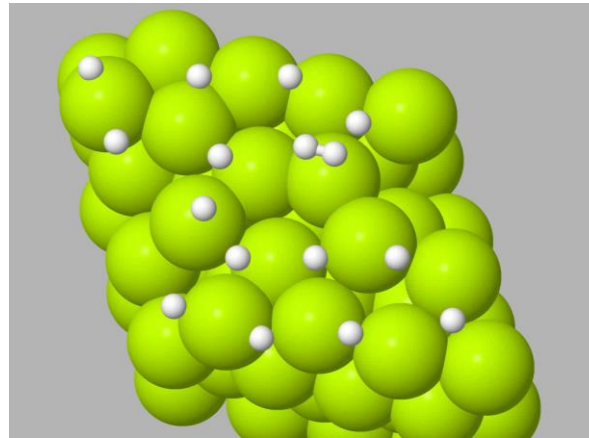
H₂ adsorption on H pre-covered surface

H coverage 0.5ML with two adjacent vacancies;
 T=200K



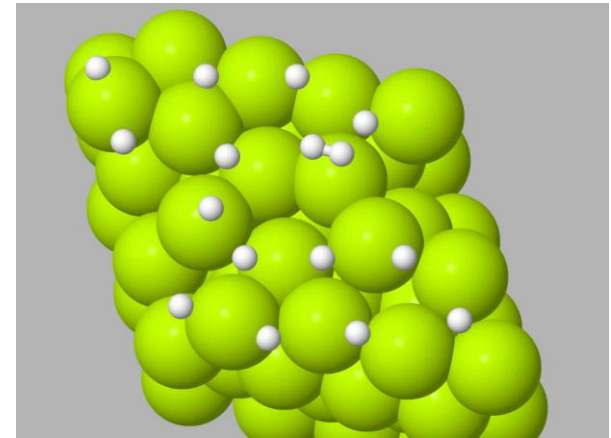
$$E_k(\text{H}_2) = 2.0 \text{ eV}$$

No adsorption



$$E_k(\text{H}_2) = 4.6 \text{ eV}$$

Adsorption!

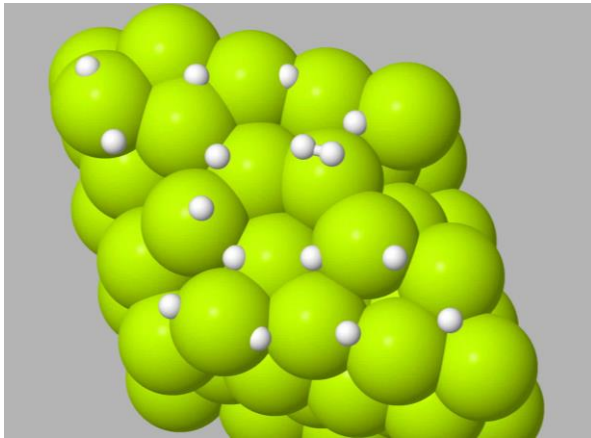


$$E_k(\text{H}_2) = 8.2 \text{ eV}$$

No adsorption

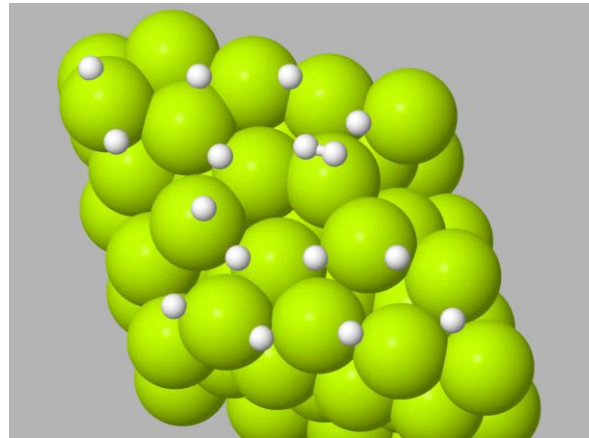
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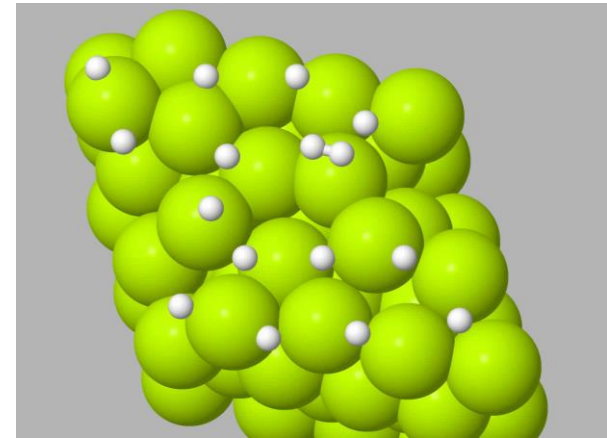
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No adsorption



$$E_k(\text{H}_2) = 4.6 \text{ eV}$$

Adsorption!



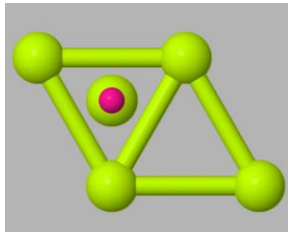
$$E_k(\text{H}_2) = 8.2 \text{ eV}$$

No adsorption

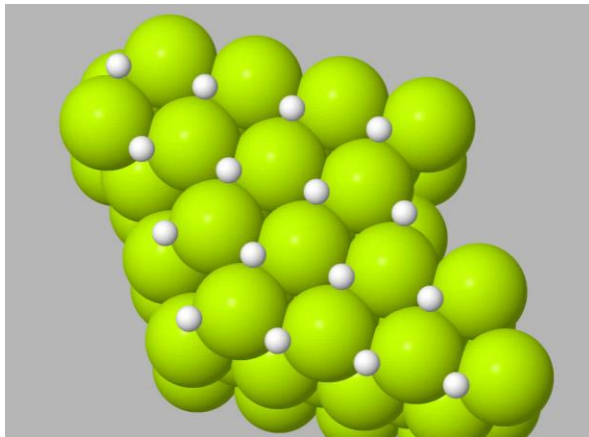
- Two hydrogen vacancies on H-covered surface are required for H₂ adsorption.
- The adsorption energy barrier on pre-covered surface is higher than for the clean surface.
- The energy of incident molecule should be in a rather narrow range!

Hydrogen at Be(0001) surface: Desorption

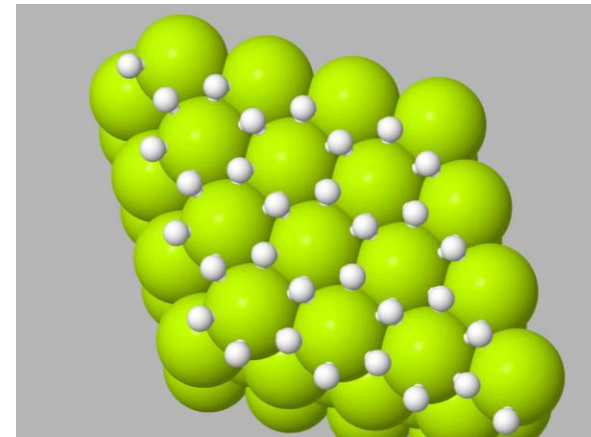
Surface coverage 0.5ML
 (half of sites occupied by H)
 hcp sites



T=900K



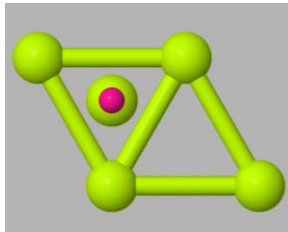
No desorption



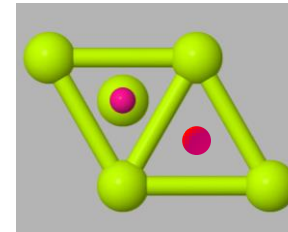
Desorption!

Hydrogen at Be(0001) surface: Desorption

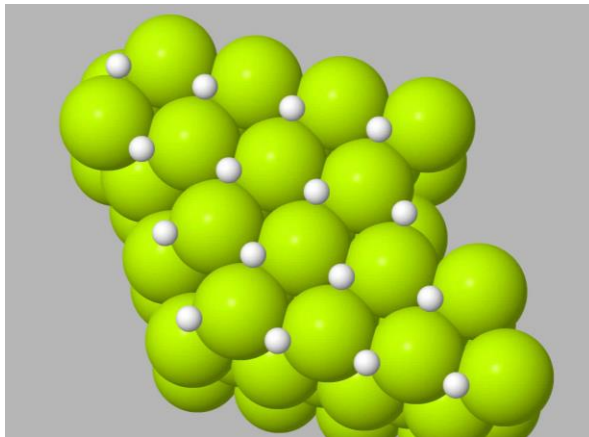
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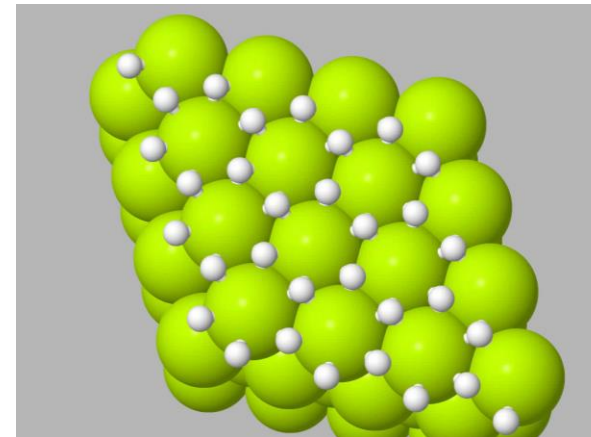
Surface coverage 1.0ML
(all sites occupied by H)
hcp-fcc sites



$T=900\text{K}$



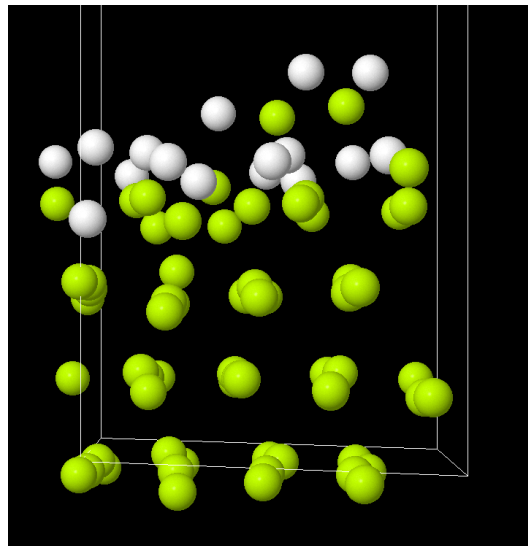
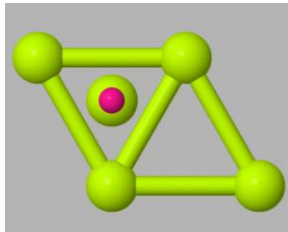
No desorption



Desorption!

Hydrogen at Be(0001) surface: Desorption

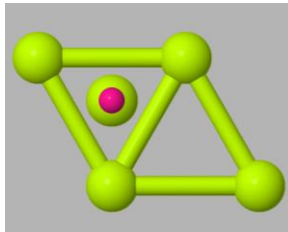
Surface coverage 0.5ML
(half of sites occupied by H)
hcp sites



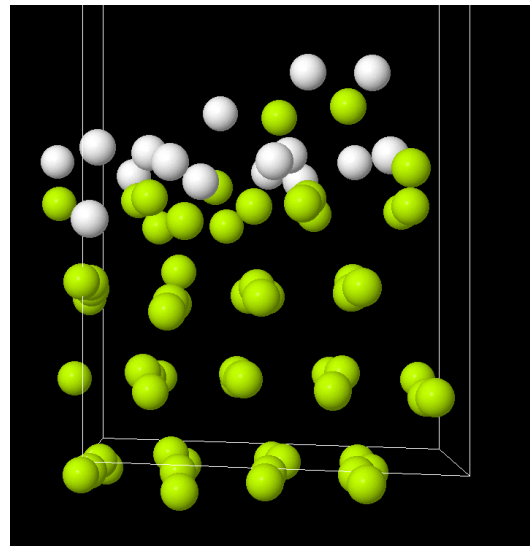
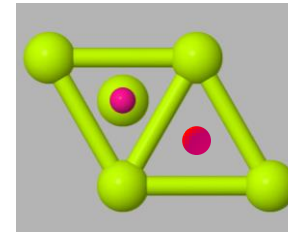
33 fs

Hydrogen at Be(0001) surface: Desorption

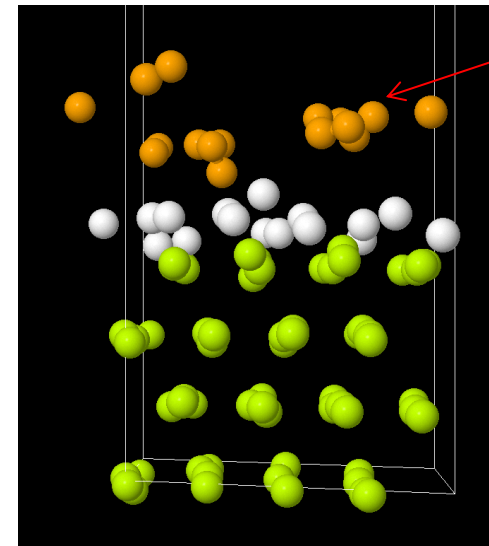
Surface coverage 0.5ML
 (half of sites occupied by H)
 hcp sites



Surface coverage 1.0ML
 (all sites occupied by H)
 hcp-fcc sites



33 fs

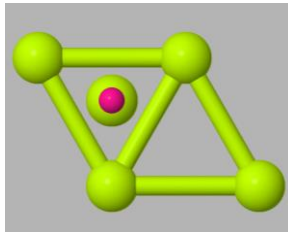


H₂

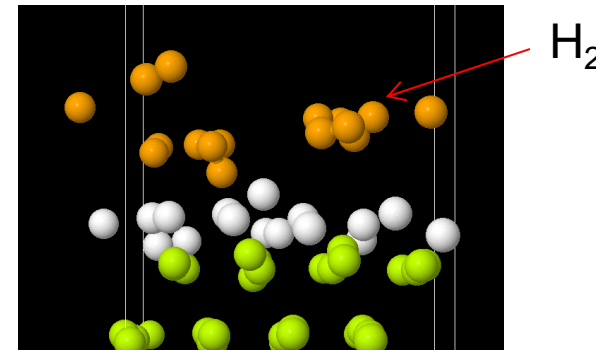
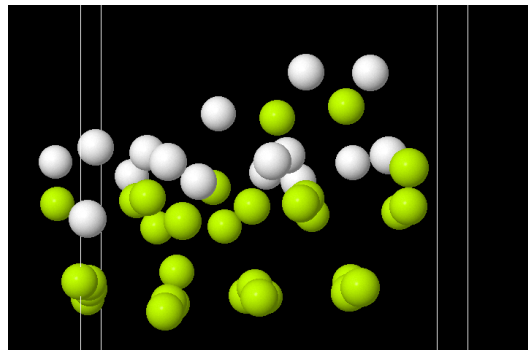
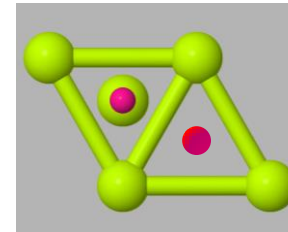
22 fs

Hydrogen at Be(0001) surface: Desorption

Surface coverage 0.5ML
 (half of sites occupied by H)
 hcp sites

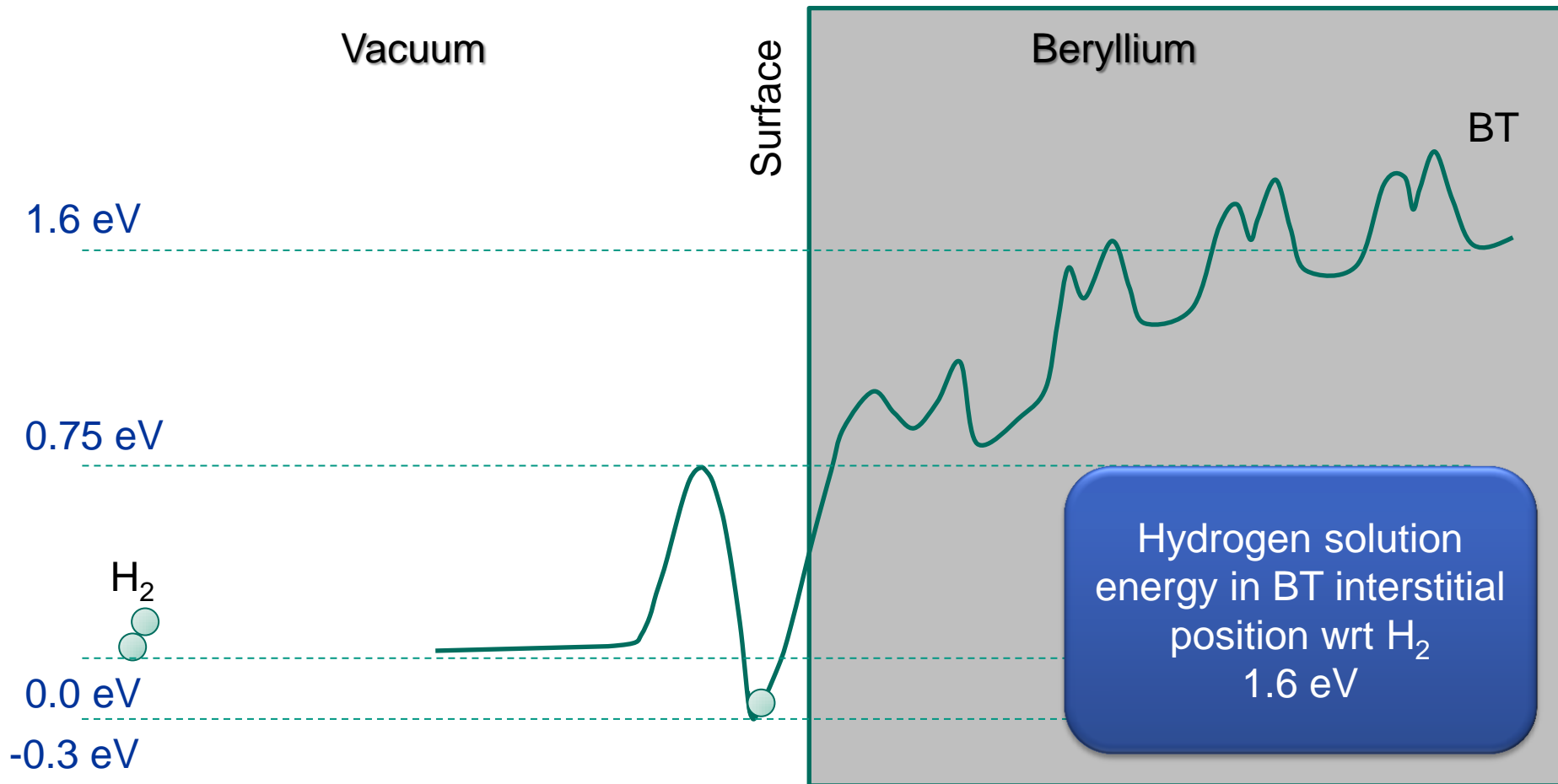


Surface coverage 1.0ML
 (all sites occupied by H)
 hcp-fcc sites

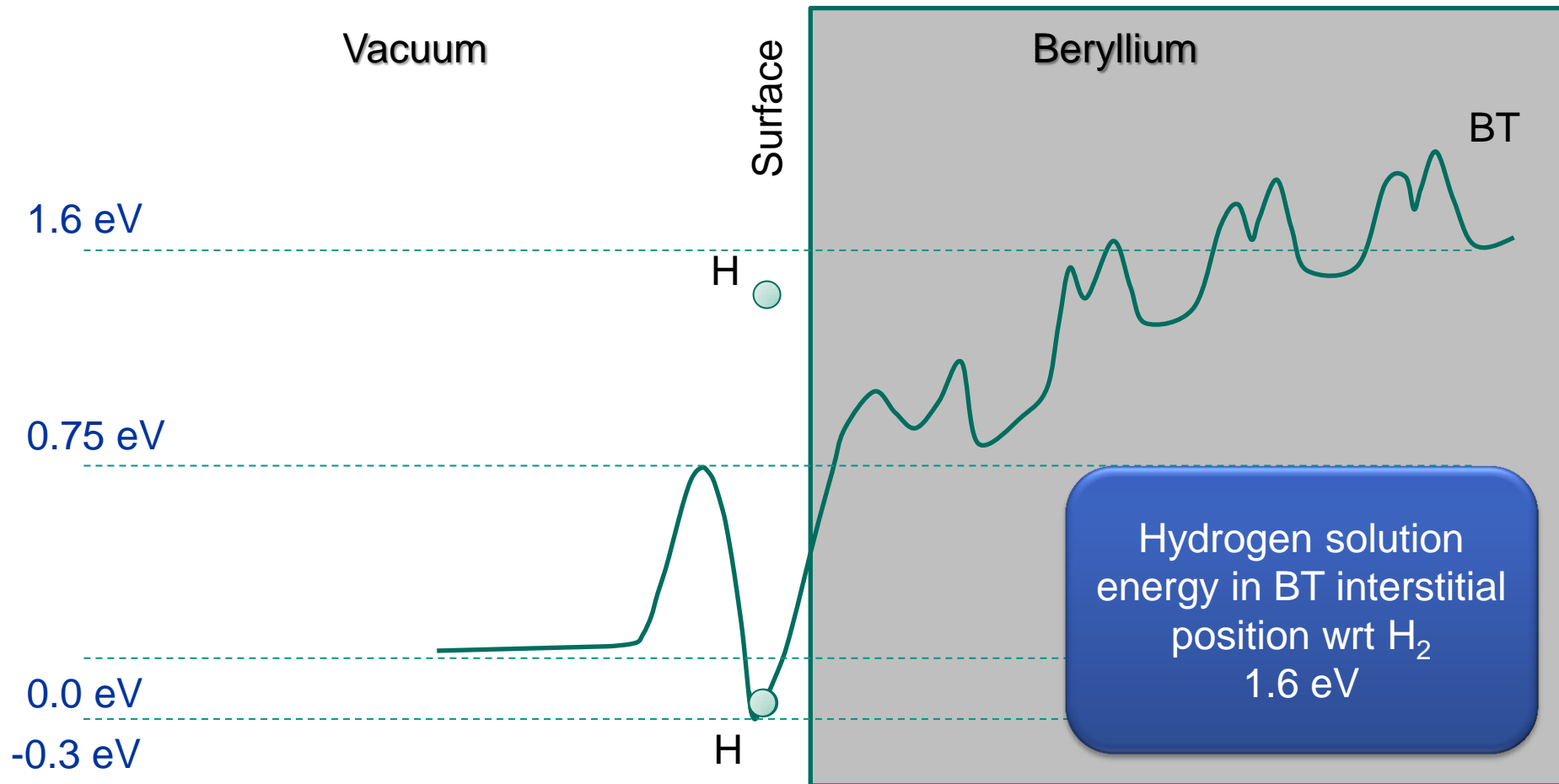


- At surface coverage of 0.5 ML severe surface reconstruction is observed, but no hydrogen desorption occurs.
 - At surface coverage of 1.0 ML immediate associative desorption occurs.
- ⇒ **Maximum critical H coverage of 0.5 ML**

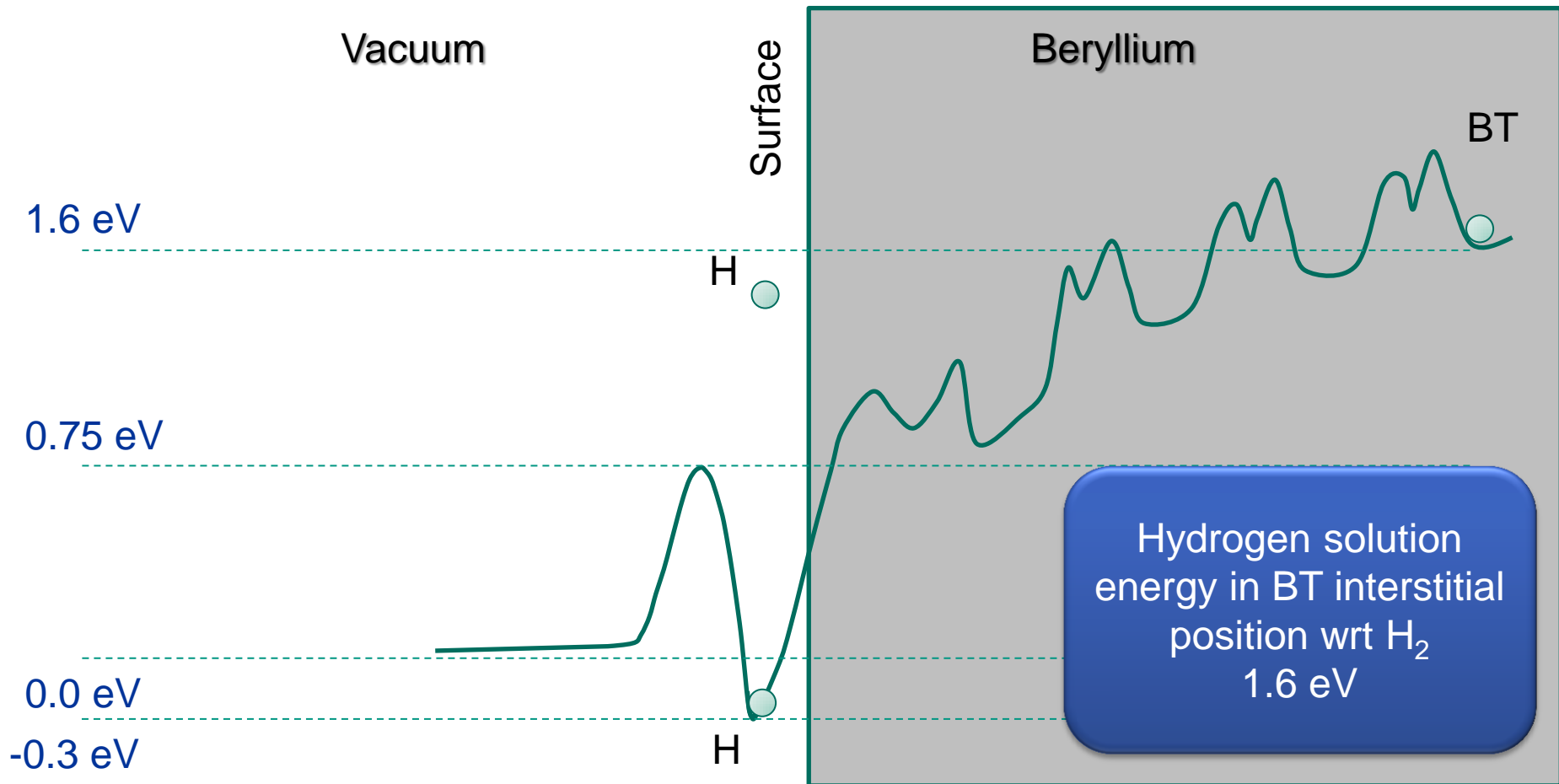
Energy landscape for H₂ absorption



Energy landscape for H₂ absorption



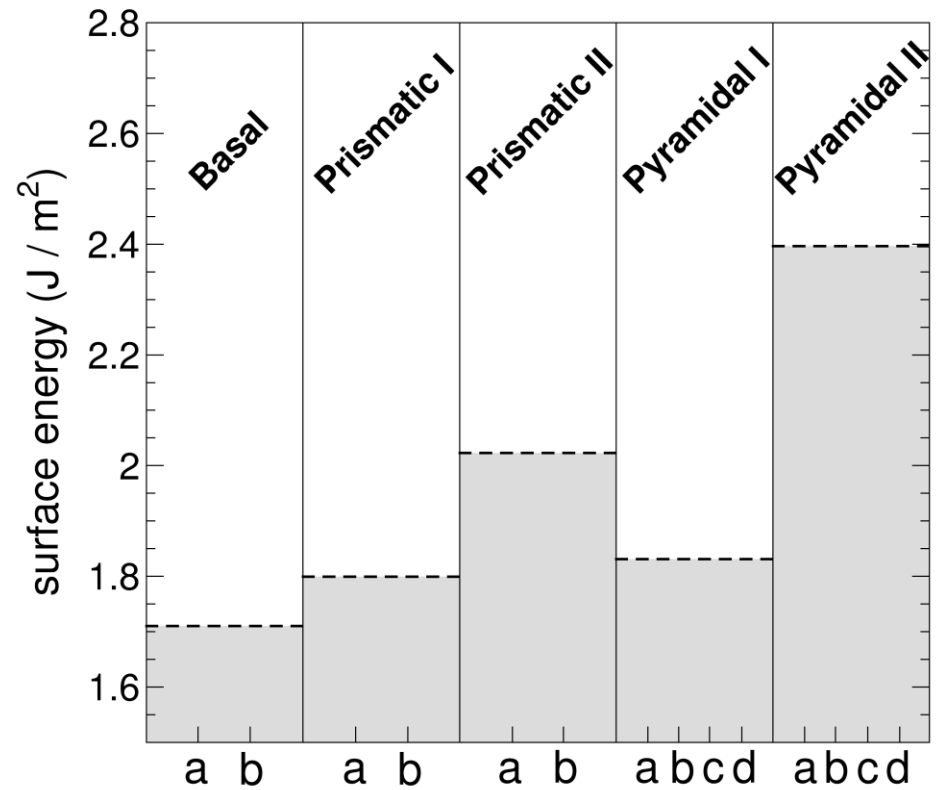
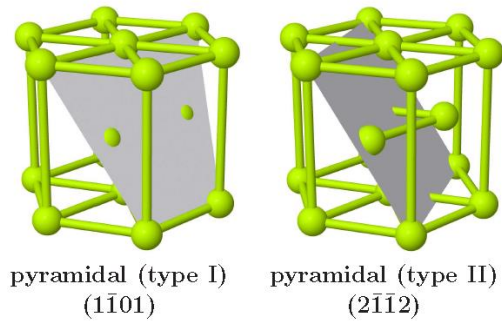
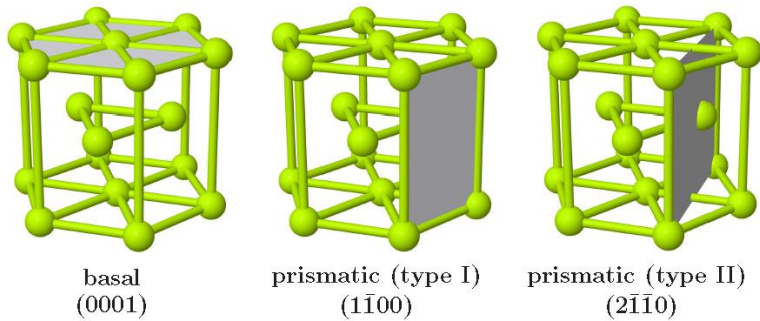
Energy landscape for H₂ absorption



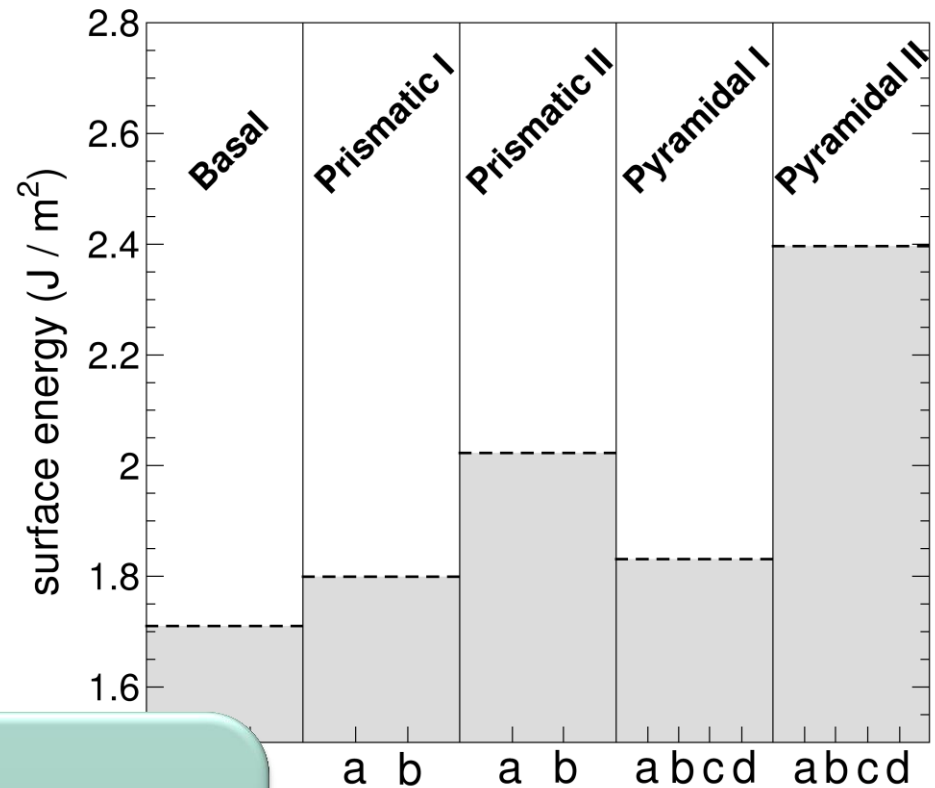
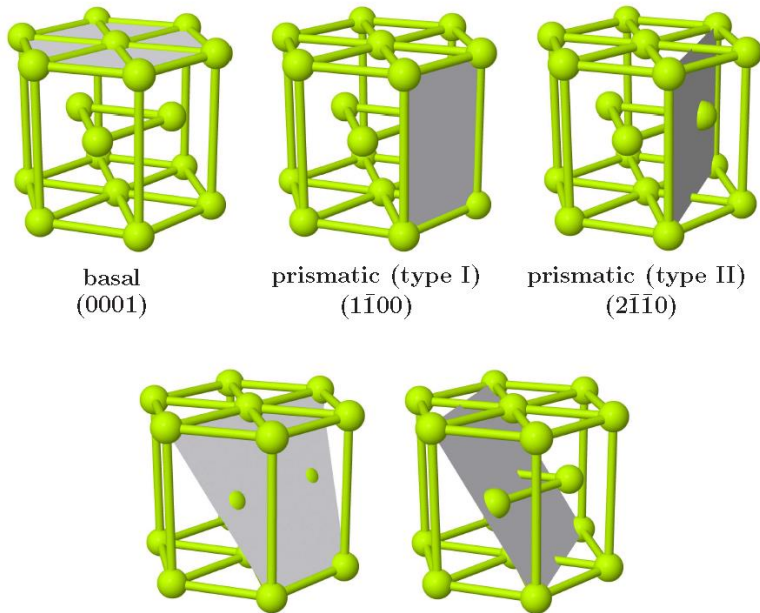
Outline

- **Beryllium in Fusion Reactor**
 - Be as plasma facing material
 - Be in T-breeding pebble bed
- **Simulation method**
- **Adsorption and desorption**
 - H₂ dissociative adsorption on (0001)Be
 - Hydrogen surface adsorption sites
 - H₂ adsorption on precovered surface
 - Hydrogen interaction on Be surface
 - Hydrogen associative desorption
 - Hydrogen absorption in bulk
- **Surface energy modification**
 - Clean surface energies
 - Surface energy with hydrogen
 - Faceting of H-covered bubble
- **Conclusions**

Be principal surfaces



Be principal surfaces

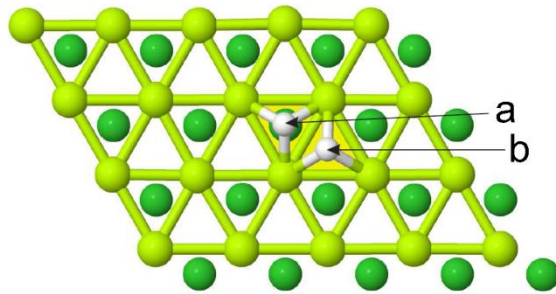


The lowest surface energy has

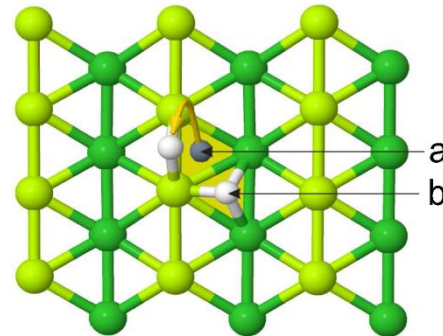
- **basal (0001)** plane,
- followed by **prismatic I ($1\bar{1}00$)** and
- **pyramidal II ($\bar{2}112$)** surfaces

The last two being very close in energy.

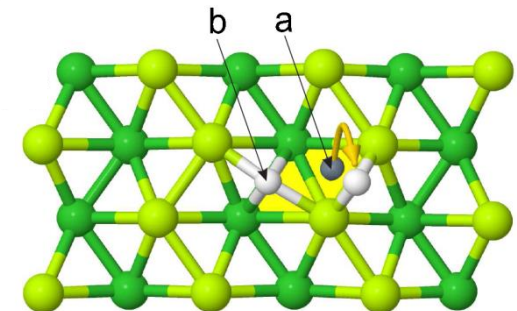
H adsorption sites on Be principal surfaces



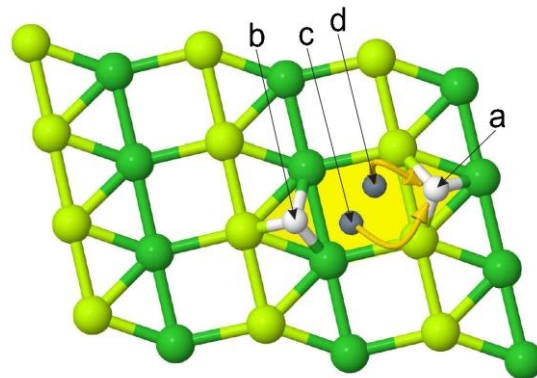
basal
(0001)



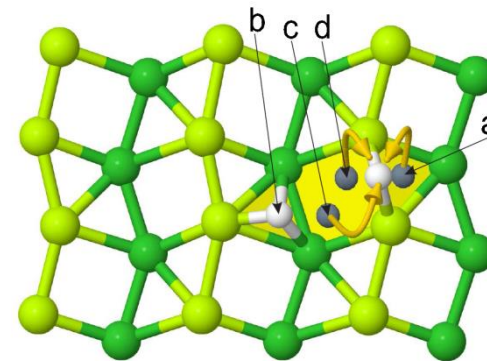
prismatic (type I)
($1\bar{1}00$)



prismatic (type II)
($2\bar{1}10$)

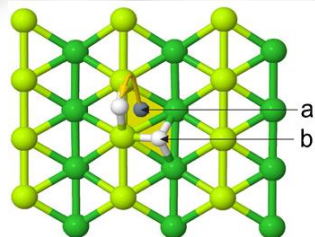


pyramidal (type I)
($1\bar{1}01$)

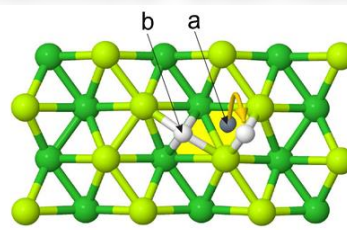


pyramidal (type II)
($2\bar{1}12$)

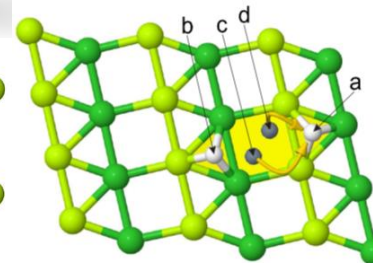
Adsorbed hydrogen modifies surface energy



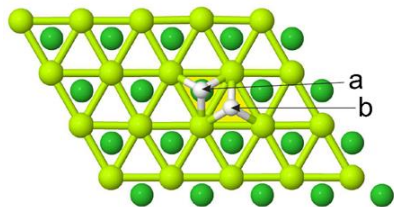
prismatic (type I)
($1\bar{1}00$)



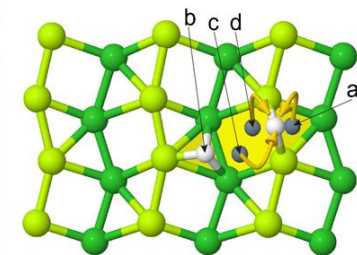
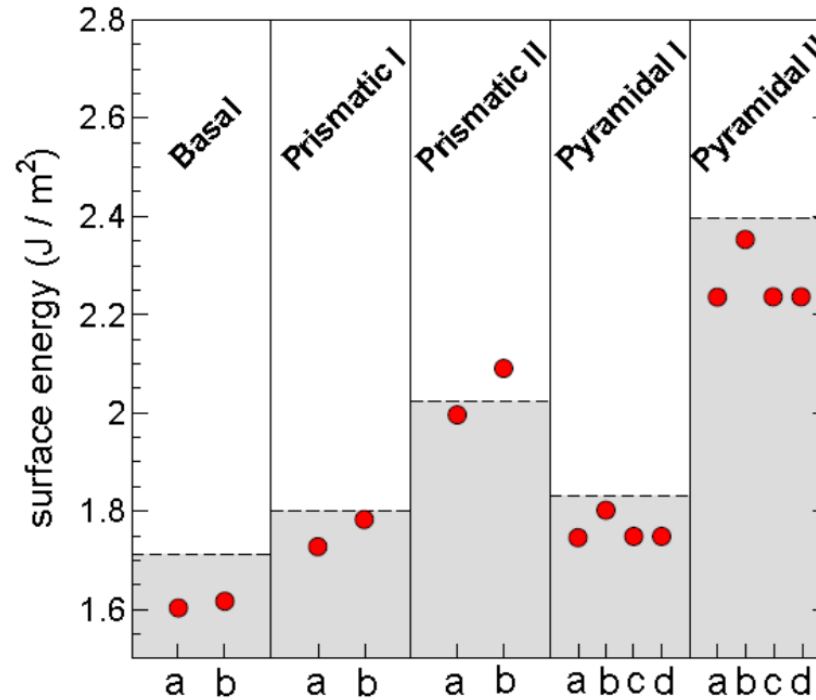
prismatic (type II)
($2\bar{1}\bar{1}0$)



pyramidal (type I)
($1\bar{1}01$)

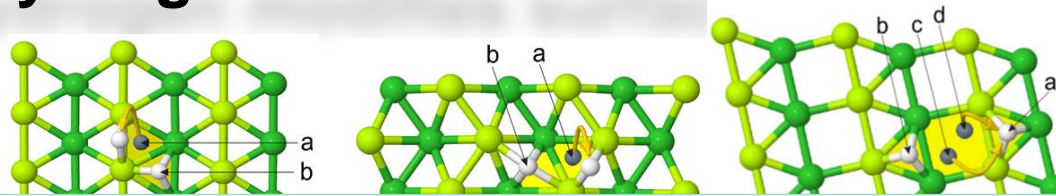


basal
(0001)

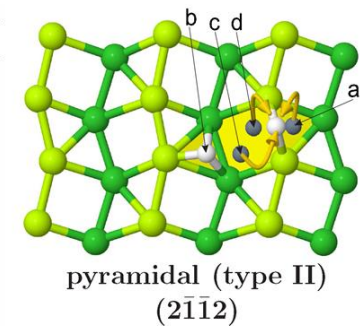
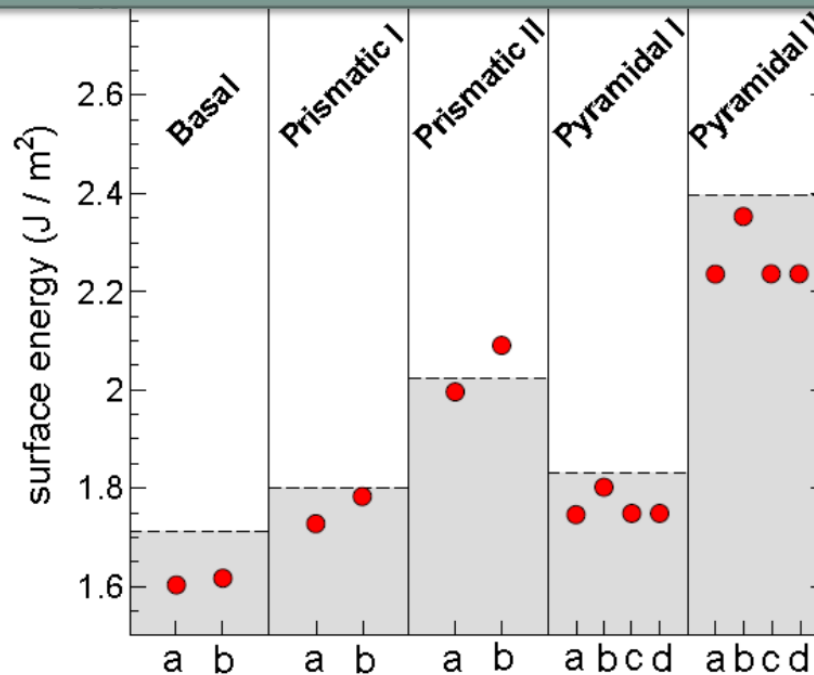
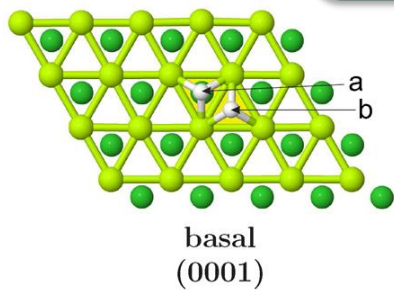


pyramidal (type II)
($2\bar{1}\bar{1}2$)

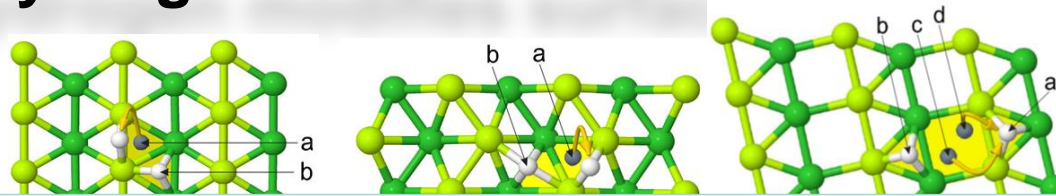
Adsorbed hydrogen modifies surface energy



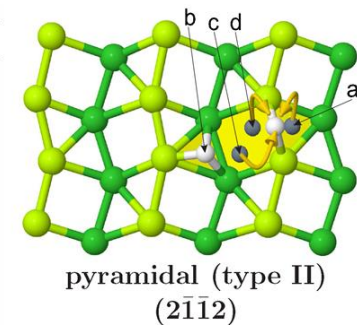
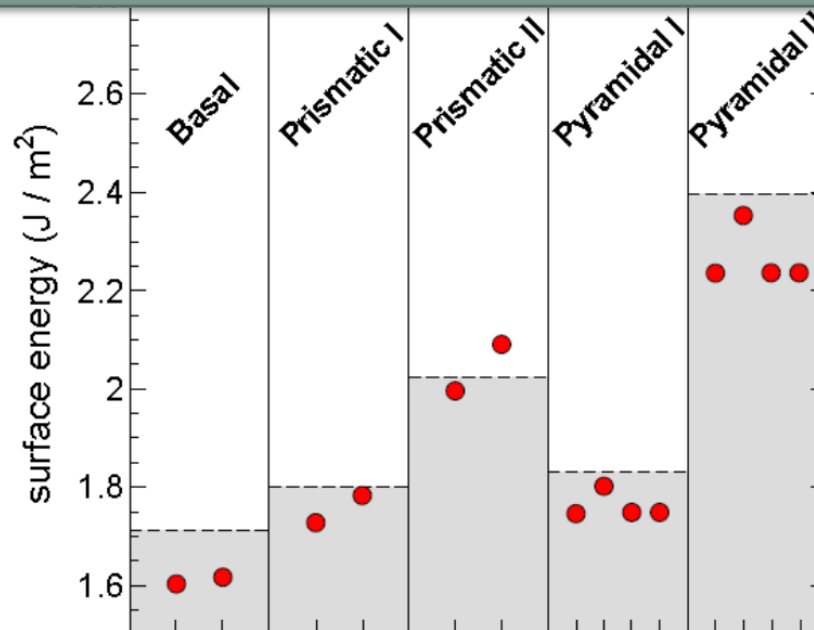
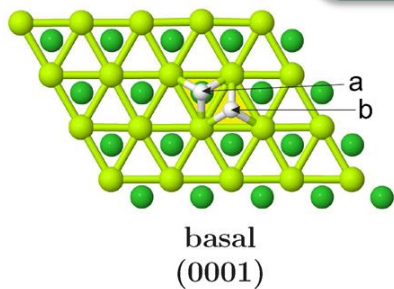
Single hydrogen atom adsorption results in **notable decrease of surface energy** in most of the cases.



Adsorbed hydrogen modifies surface energy



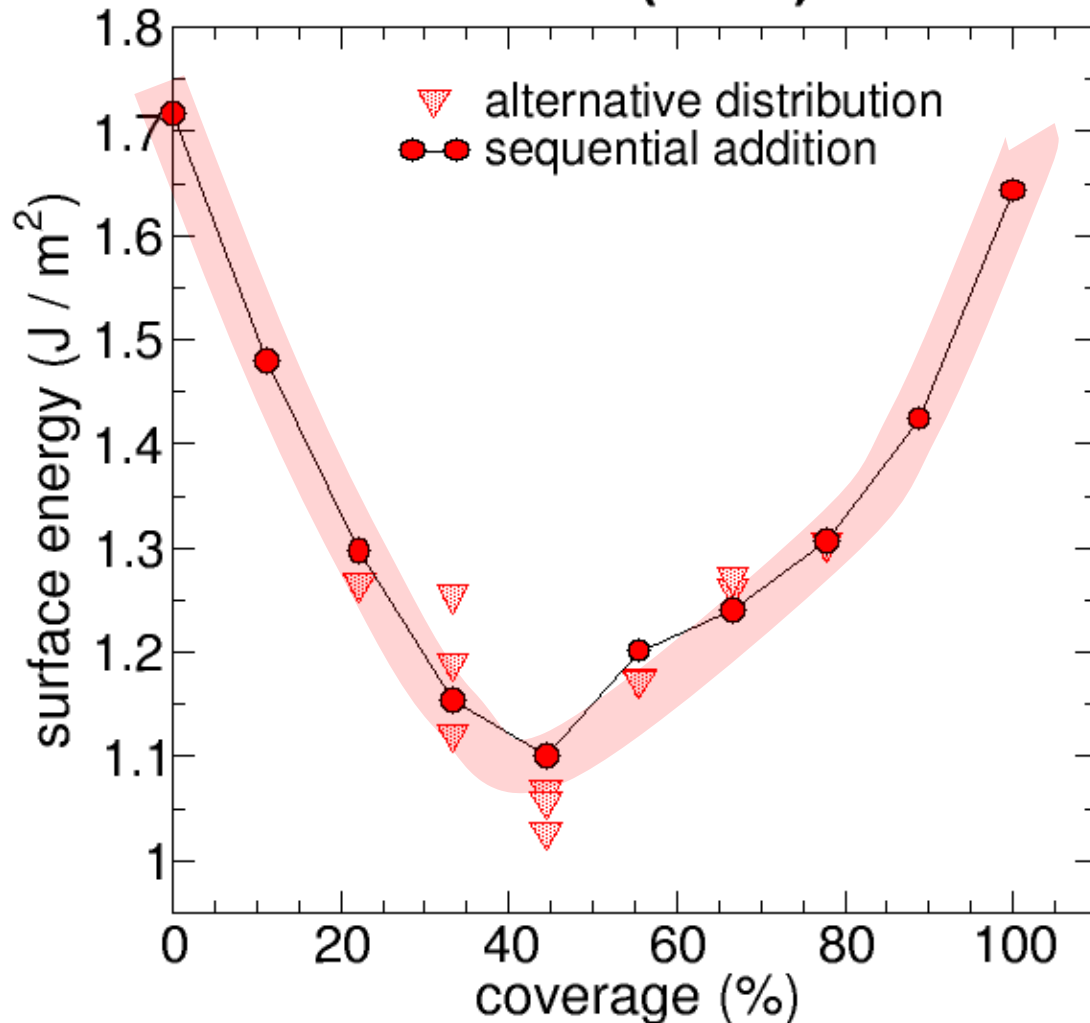
Single hydrogen atom adsorption results in **notable decrease of surface energy** in most of the cases.



How multiple hydrogen adsorption would affect the surface energy?

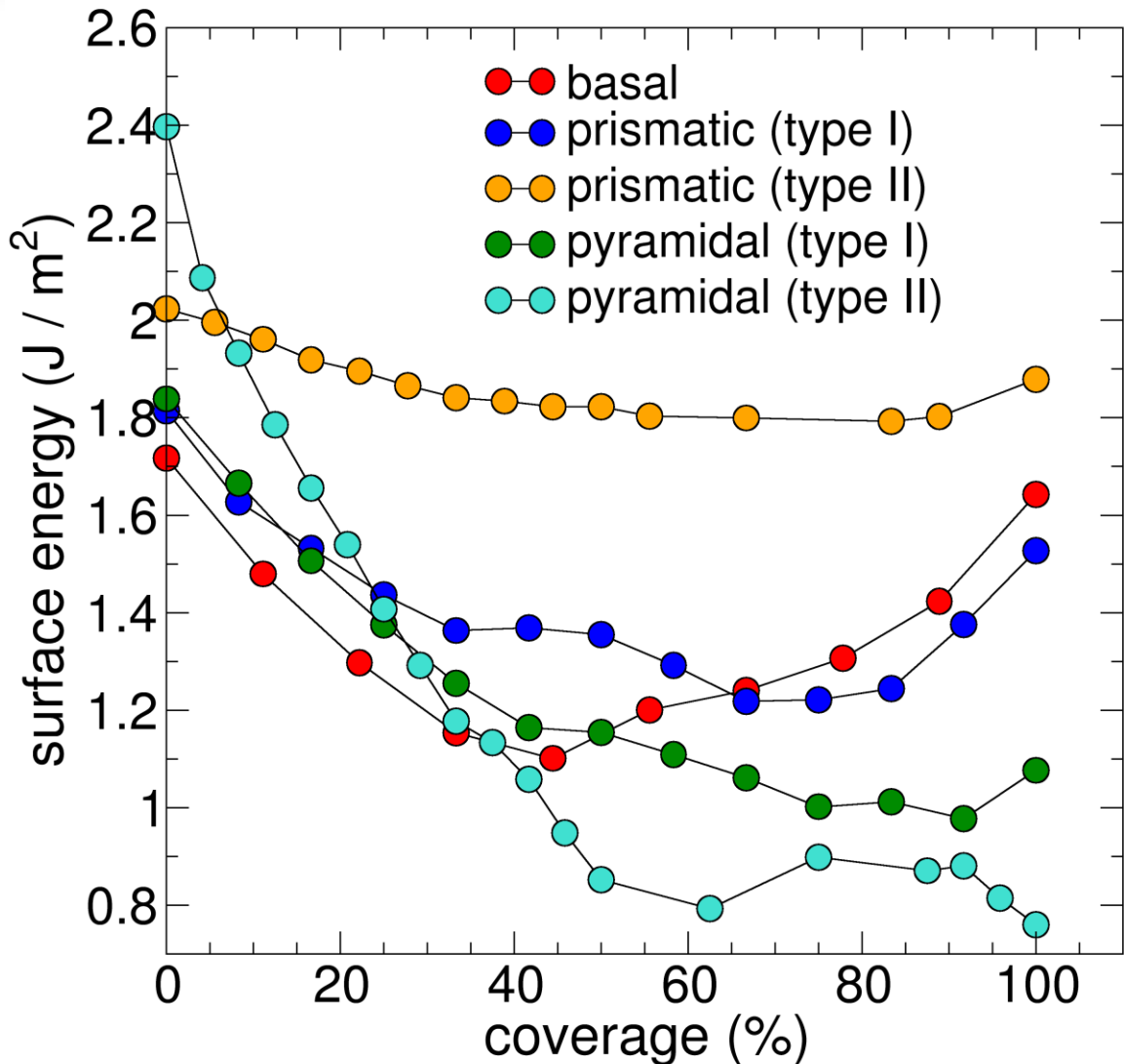
Adsorbed hydrogen modifies surface energy

basal (0001)



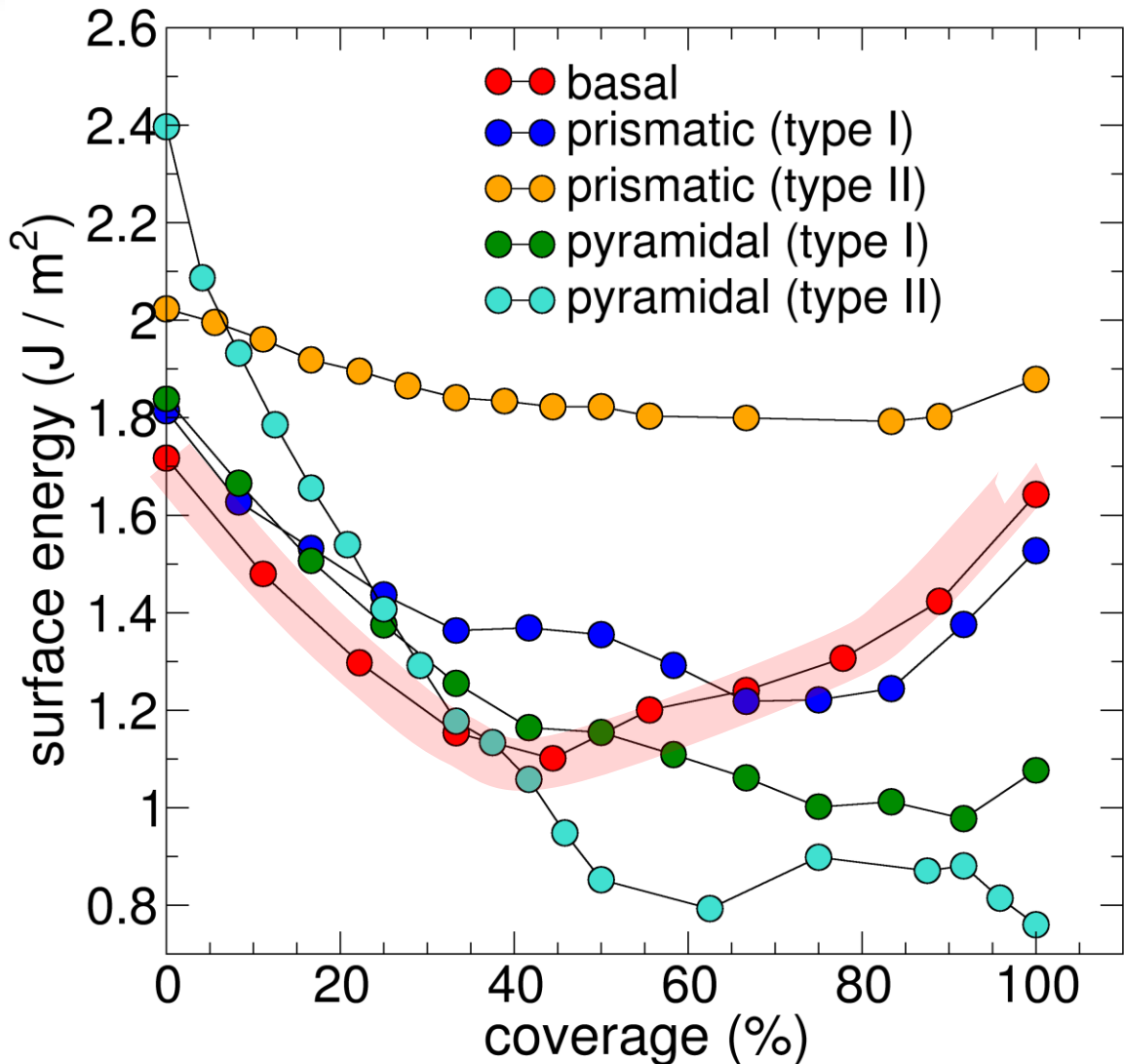
- **At low coverage:**
Decrease of E_{surf} due to single H adsorption
- **At high coverage:**
Increase of E_{surf} due to H-H repulsion

Adsorbed hydrogen modifies surface energy



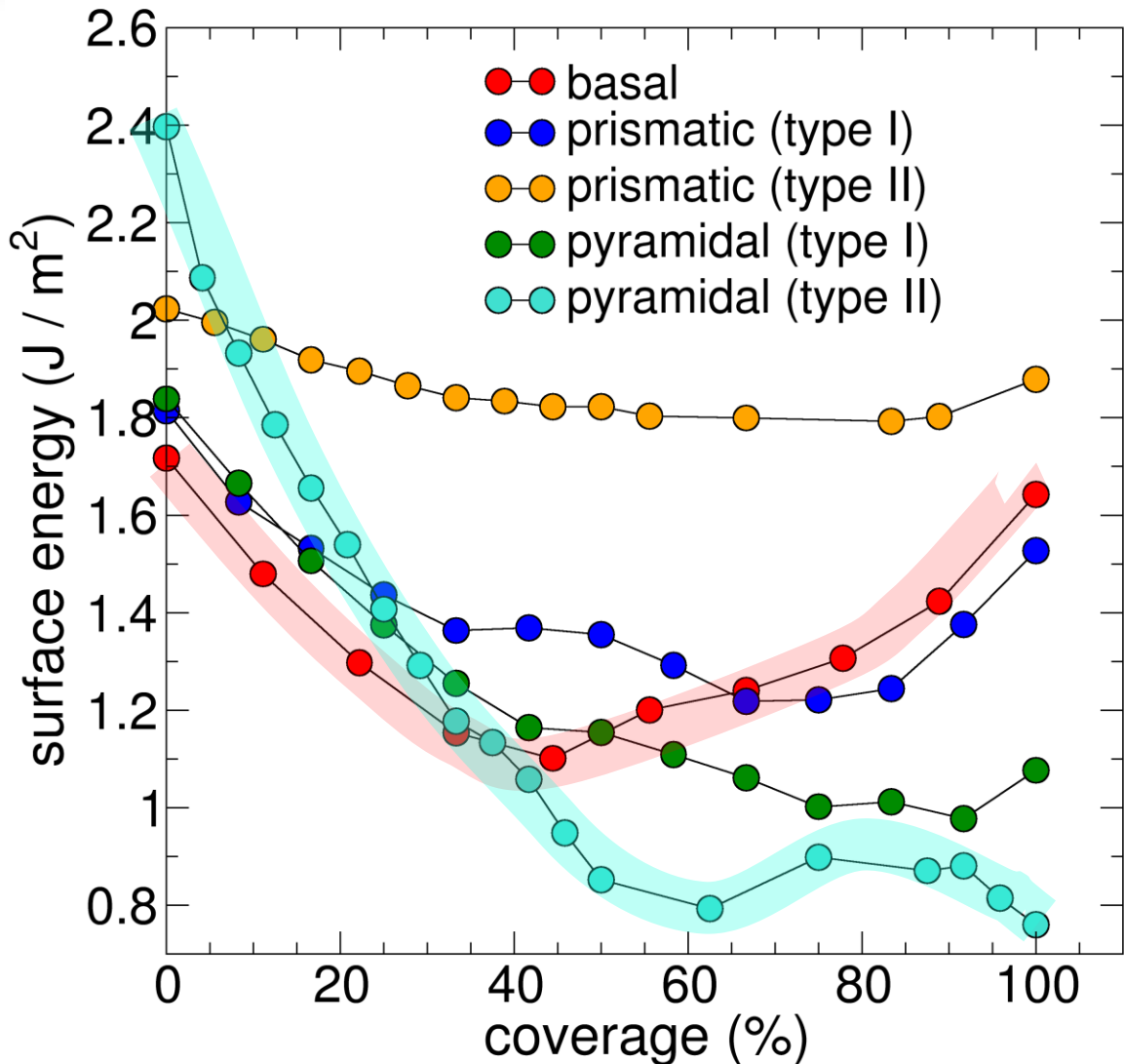
- **At low coverage:**
Decrease of E_{surf} due to single H adsorption
- **At high coverage:**
Increase of E_{surf} due to H-H repulsion

Adsorbed hydrogen modifies surface energy



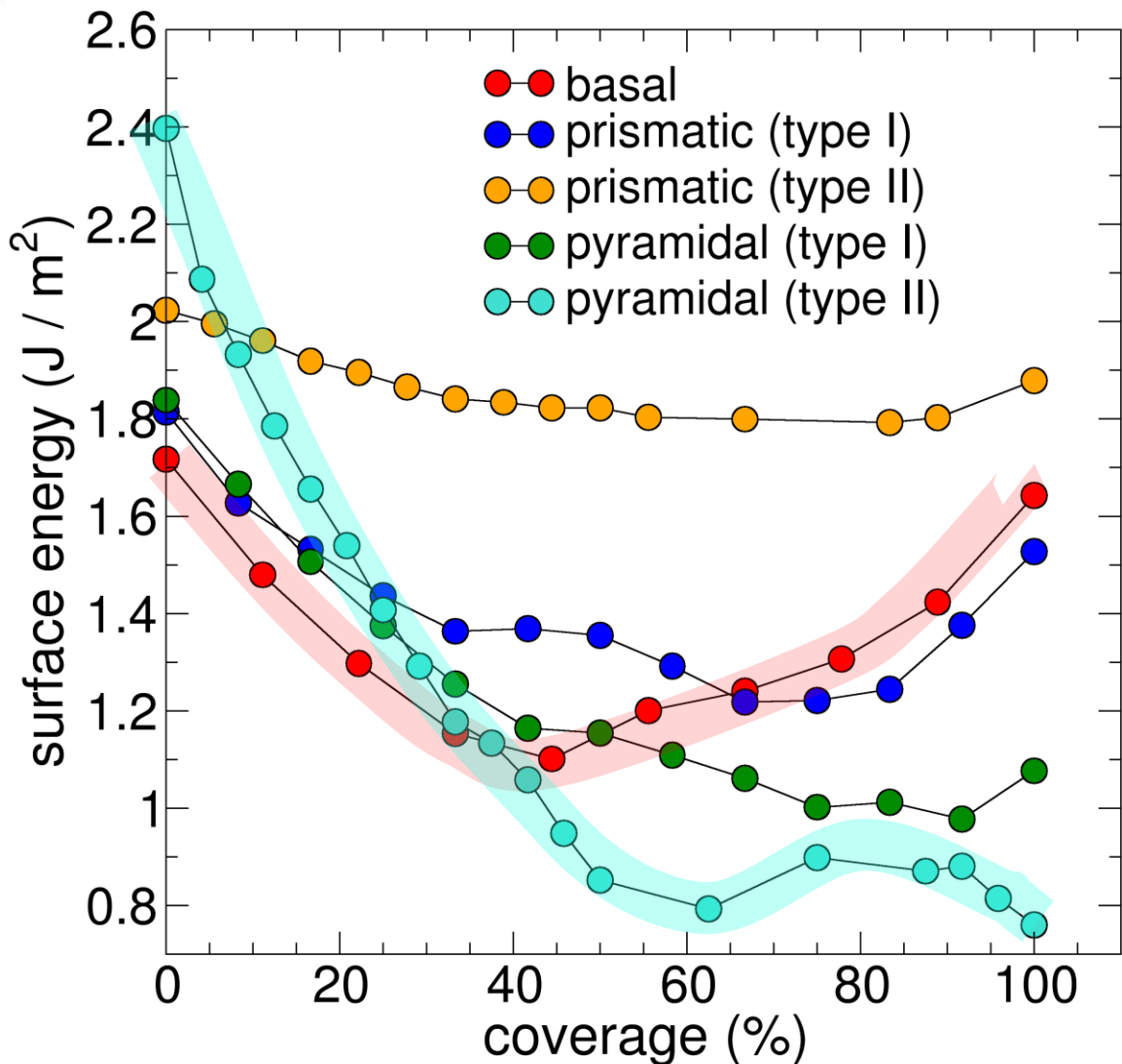
- **At low coverage:**
Decrease of E_{surf} due to single H adsorption
- **At high coverage:**
Increase of E_{surf} due to H-H repulsion

Adsorbed hydrogen modifies surface energy



- **At low coverage:**
Decrease of E_{surf} due to single H adsorption
- **At high coverage:**
Increase of E_{surf} due to H-H repulsion

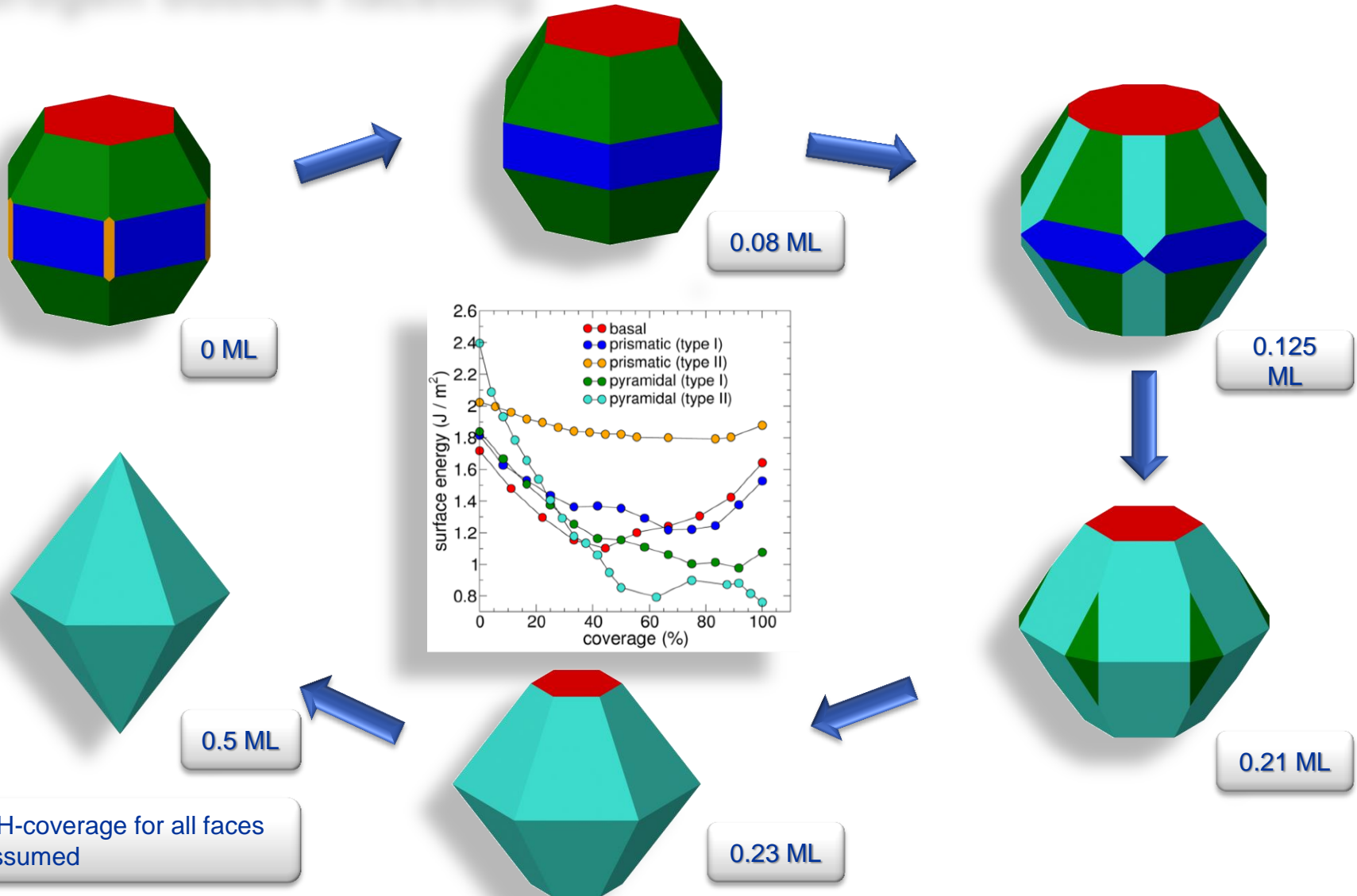
Adsorbed hydrogen modifies surface energy



- **At low coverage:**
Decrease of E_{surf} due to single H adsorption
- **At high coverage:**
Increase of E_{surf} due to H-H repulsion

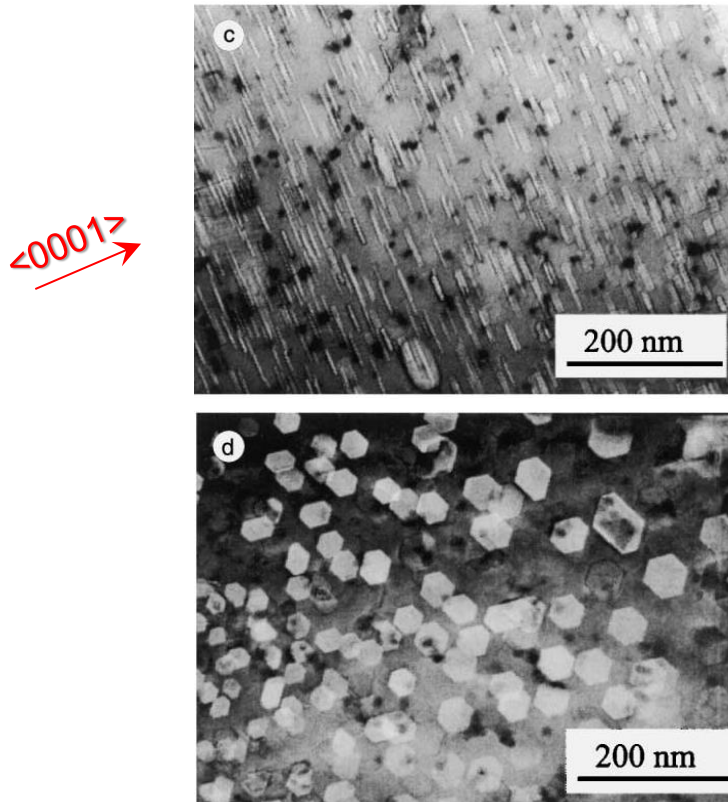
⇒ Complex modification of hydrogen-filled gas bubble faceting

Hydrogen bubble faceting



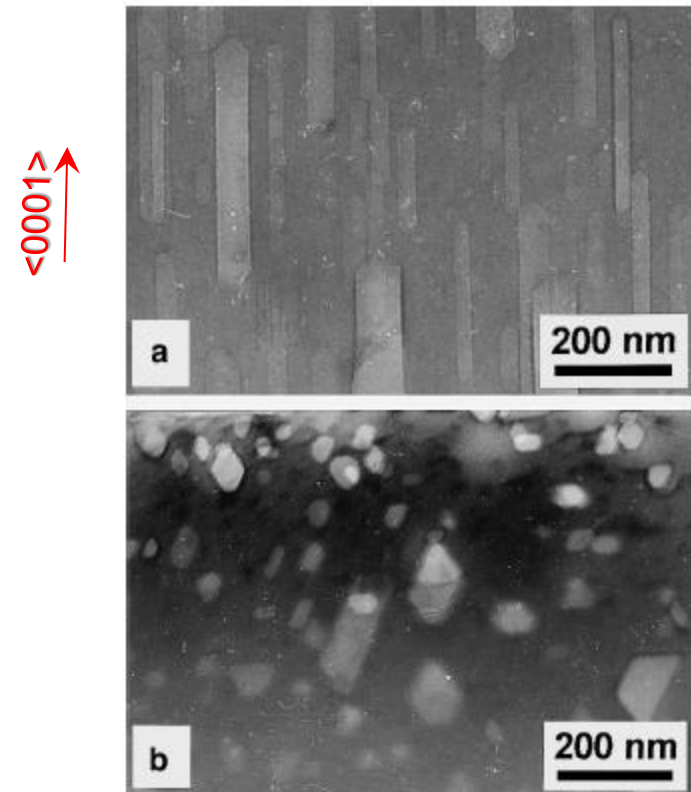
Shape of gas bubbles in Be

n-irradiated (He-bubbles)



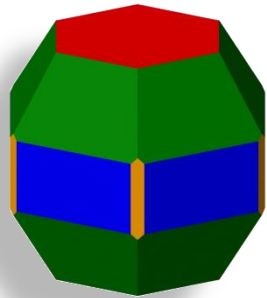
V. Chakin, Z. Ostrovsky, J. Nucl. Mater. **307–311** (2002) 657–663

H-implanted (H-bubbles)

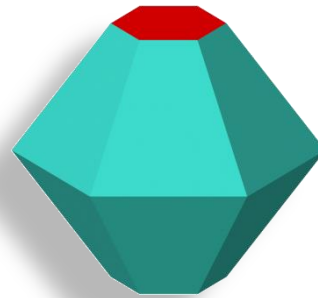


S.P. Vagin et al. J. Nucl. Mater. **258-263** (1998) 719-723

Equilibrium shape of bubbles (Wulff construction)



0 ML



0.23 ML



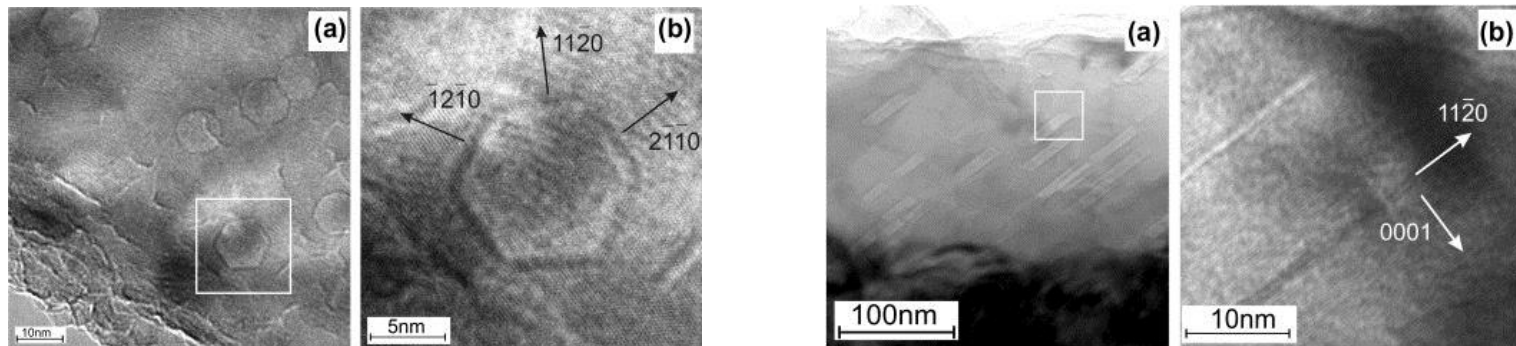
0.5 ML

- basal
- prismatic type I
- prismatic type II
- pyramidal type I
- pyramidal type II

Equilibrium shape of bubbles (Wulff construction)



“Comparison” with experiment



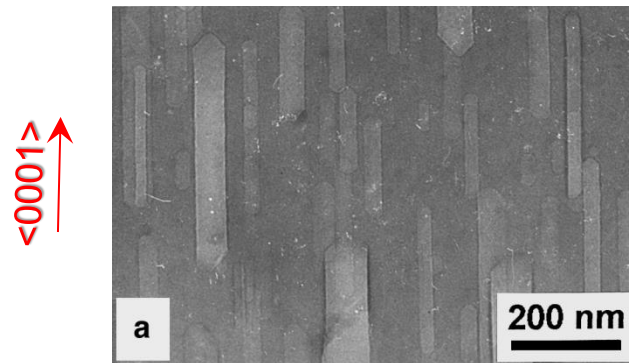
HRTEM investigations of a 8 nm bubble in a pebble irradiated at 686 K: the bubble with a regular hexagonal form and with an elongated shape

Taken from: M. Klimenkov et al. J. Nucl. Mat. 443 (2013) 409-413.

Equilibrium shape of bubbles (Wulff construction)



“Comparison” with experiment



Cavities in hydrogen-implanted beryllium after annealing for 15 min at 600°C.

Taken from: S.P. Vagin et al. J. Nucl. Mat. 258-263 (1998) 719.

Conclusions

- H₂ adsorption and desorption on clean and H precovered Be(0001) surface has been studied
- Hydrogen atom is adsorbed without barrier, while ~0.8 eV should be overcome during H₂ molecule adsorption
- Hydrogen adsorption is completely blocked by H-coverage of 0.5 ML
- At least two vacant sites are necessary for H₂ adsorption on H precovered surface
- There is a critical H surface coverage of 0.5ML, above which non-activated H₂ desorption occurs
- H repulsion on the surface results in severe surface reconstruction
- Adsorbed H significantly modifies surface energy of various Be surfaces, so that equilibrium shape of H-covered bubble is changed drastically

MoD-PMI 2015

International workshop on Models and Data for Plasma-Material Interaction in Fusion Devices

25-27 May 2015 - Aix-Marseille Université - Site St. Charles, 3 place Victor Hugo 13001 Marseille (France)



Thank you for your attention!