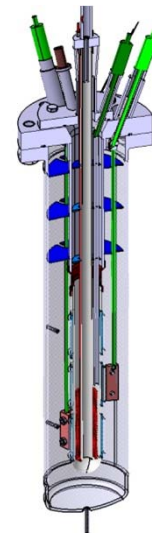
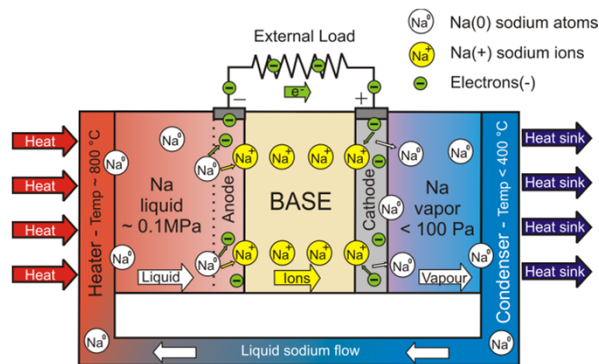


# Direct energy conversion of heat to electricity using AMTEC

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IYCE 2015, Pisa

Karlsruhe Institute of Technology (KIT) – Institute for Neutron Physics and Reactor Technology (INR)

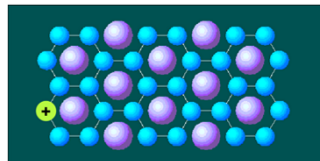
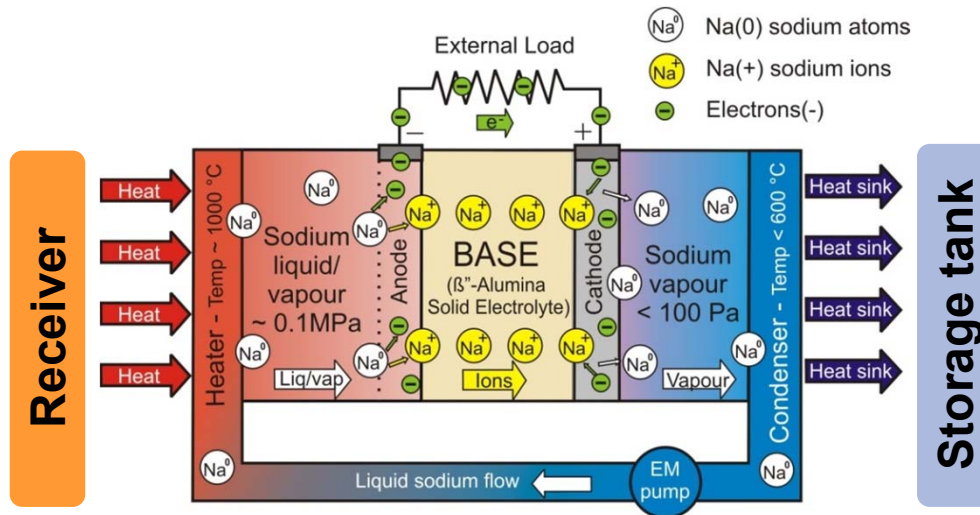


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2. Overview of AMTEC systems
3. AMTEC TEst FAcility (ATEFA) & Test cell
4. Scoping tests of PEEK in liquid sodium environment
5. Summary and outlook

# AMTEC – direct heat to electricity conversion

## Alkali Metal Thermal to Electric Converter



- Key component:  $\beta''$ -Alumina Solid Electrolyte (BASE)
- Key process: Na-ionization  
 $\Delta P$  across BASE  $\rightarrow \Delta(\text{sodium activity})$



### Main advantages

- High theoretical efficiency (40 %)
- Flexible regarding the heat source
- Suitable for modular design
- No moving components

### Issues

- Real efficiency of 20 %
  - Heat losses (10 – 15 %)
  - Electrical losses (12 – 15 %)
- Power degradation over time
  - Degradation of BASE
  - Electrode sintering (grain growth)

# Overview of AMTEC systems

Year	Author	AMTEC cell characteristics	Comments / Electrical properties
1969	Kummer & Weber	single tube, liquid anode	Developed Sodium Heat Engine (SHE)
1986	Hunt	single tube, liquid anode	$P_{out} = 8 \text{ W}$ ( $0.4 - 0.5 \text{ W/cm}^2$ ); $T \sim 800 \text{ }^\circ\text{C}$ ; $\eta = 19 \%$ ; $t_{op} = 1500 \text{ h}$
1989 - 1993	Heinzel et al. (KIT - INR)	single tube, liquid anode	$P_{out} = 112 \text{ W}$ ( $1.8 \text{ W/cm}^2$ ); $910 \text{ }^\circ\text{C}$ ; $t_{op} = \text{few hours}$
1992 - 1998	Air Force Philips Laboratory	multitube, vapor anode	Series connected AMTEC for advanced radioisotope power system to fulfill the mission of Pluto Express. $P_{out} = 4.3 \text{ W}$ ( $0.12 \text{ W/cm}^2$ ); $750 \text{ }^\circ\text{C}$ ; $\eta \sim 12 \%$ ; $t_{op} = 600 \text{ h}$
2006	Hunt & Rasmussen	Small scale, multitube, vapor anode	$P_{out} = 0.5 \text{ W}$ ( $0.7 - 0.8 \text{ W/cm}^2$ ); $740 \text{ }^\circ\text{C}$ ; $\eta \sim 13 \%$ ; $t_{op} = 5 \text{ years}$



Heinzel et al. 1992

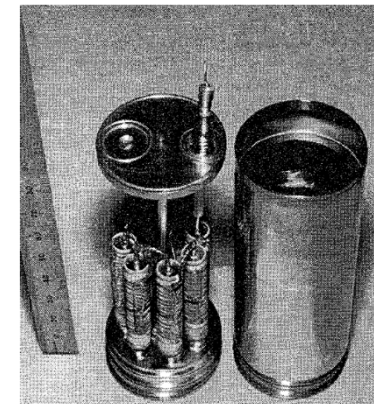


Figure 2 Series II, PX-2 AMTEC Cell

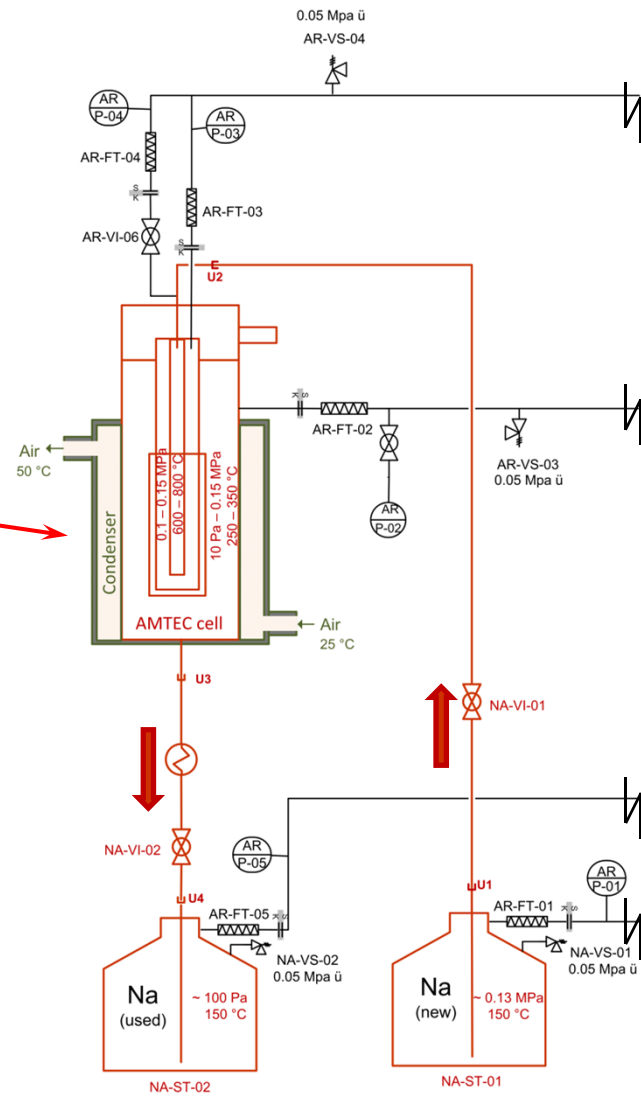
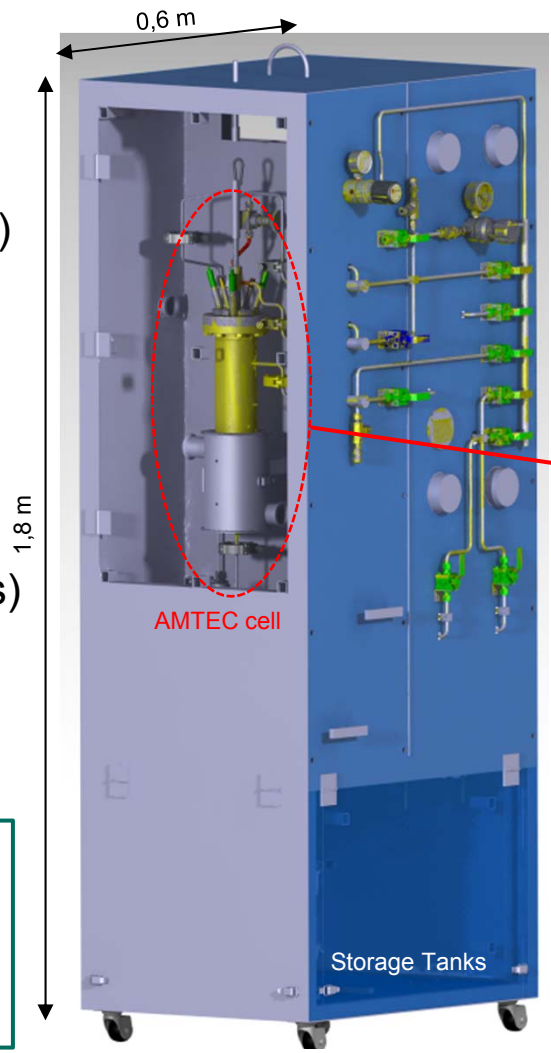
Sievers et al. 1997

# AMTEC Test Facility (ATEFA)

- Research on AMTEC cell
- Test facility consists of Na-side + Ar-side.
- **Na-side** includes:
  - Two storage tanks (~ 4,5 l)
    - $T_{Na} \sim 150 \text{ }^\circ\text{C}$
  - AMTEC test cell
    - $T_{Na} = 600 - 800 \text{ }^\circ\text{C}$
  - Air-cooled condenser
    - $T_{Na} \sim 300 \text{ }^\circ\text{C}$
  - Heating trace
  - $Vol_{Na} \sim 0.3 \text{ l}$  (without tanks)
- **Ar-side** controls Na-flow and pressure in cell and tank
  - Max. p = 1.5 bar

**Safe design!**

- ✓ In-housed in a metallic box
- ✓ Possibility to float with Ar
- ✓ Safety equipped lab.



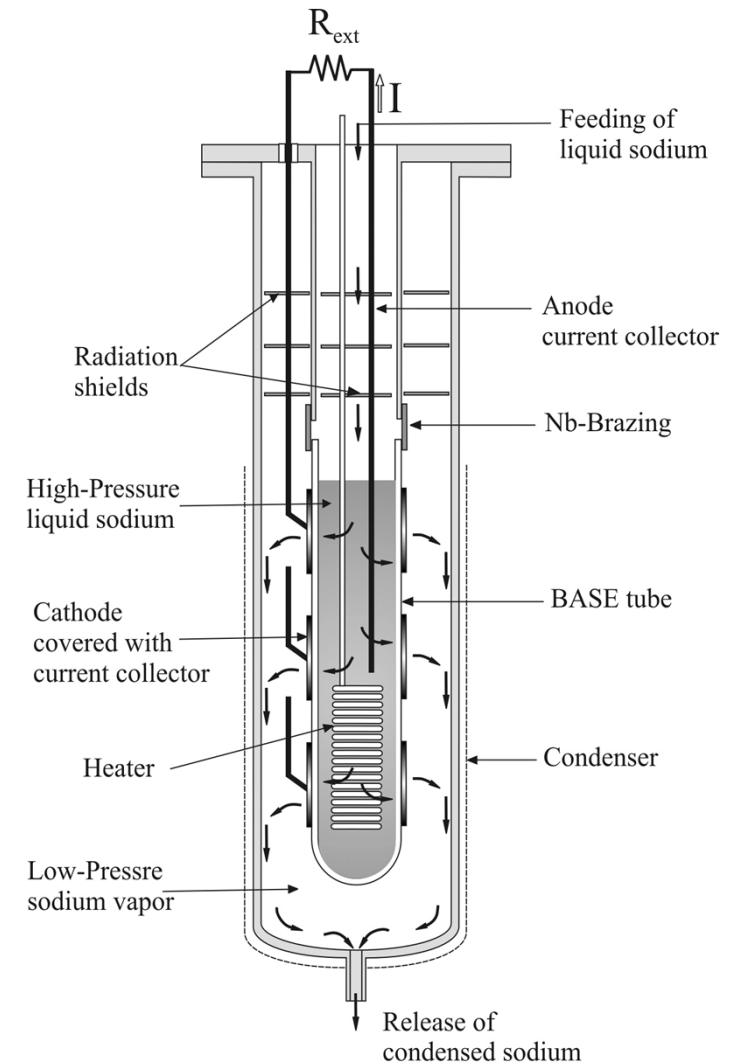
# AMTEC test cell

- Single BASE tube  
(D=3cm, L=22cm, wall thickness 1mm )
- Na-liquid anode
- Sputtered cathodes (~ 5 μm thickness)
- BASE-Nb-Steel brazing
- 21 thermocouples → ΔT measurement in axial, radial, circumferential direction
- Pressure sensors in Argon atmosphere
- Electronic load → IV curve measurements

## Tests\*

- BASE (thickness, chemical stability, homogeneity, ionic conductivity)
- Electrode composition (Mo, TiN, TiC)
- Electrode and current collector structure
- Ceramic-metal brazing

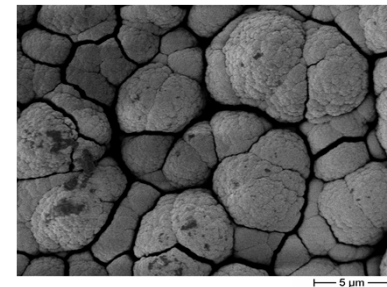
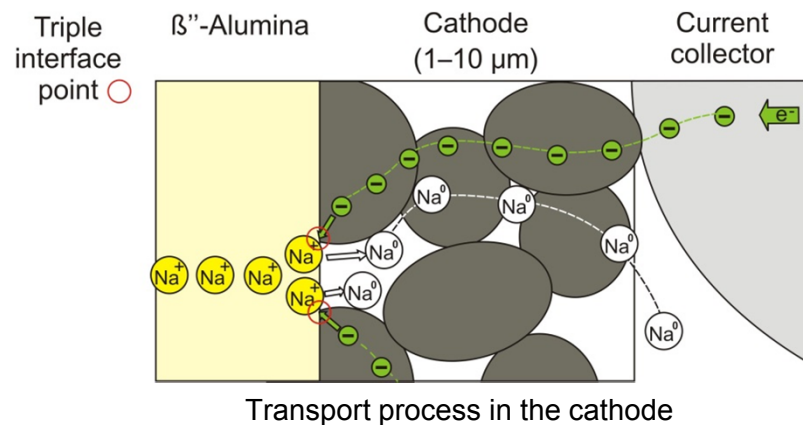
\*together with other institutes in KIT





# AMTEC test cell

- Electrode – current collector structure
  - Magnetron-Sputtered porous cathode (~ 5  $\mu\text{m}$  grain size)
  - Transport in cathode
    - Recombination of Na ions at triple interface points
    - Porosity vs. current density
    - Thickness vs. electrical resistance
  - Current collector
    - Metallic grid, metallic foam, knitted wire...



Molybdenum cathode (h~ 5  $\mu\text{m}$ )



- Sintering of electrode decreases cell performance with time  
→ refractory materials or ceramics (Mo, TiN, TiC)

# Scoping tests of PEEK in liquid sodium

- Polyether ether ketone (PEEK) is a polymer with excellent mechanical and chemical resistance properties at high temperatures.
- Liquid / vapor sodium is a challenge for sealing materials in valves and instrumentation
- Little information about chemical test of polymers – liquid sodium
- Seven PEEK samples of ~1cm<sup>2</sup> were introduced in a sodium bath at different temperatures inside a glove box with argon atmosphere
- Chemical and mechanical analysis of the samples foreseen

Temp.	Duration	Remark
100 °C	15 min.	No changes
120 °C	10 h	No changes
150 °C	10 h	Appearance of black spots
200 °C	1h	Sample completely darkened and apparent changes in mechanical properties

Overview of the performed PEEK chemical test



0 sec. – 120 °C



10h – 120 °C



0 sec. – 150 °C



10h – 150 °C



5 min. – 200 °C



10h – 200 °C

VICTREX-PEEK-450G™  
samples before / after  
chemical test in liquid sodium



# Summary and outlook

- AMTEC test cell and test facility have been designed and are under construction
- Tests in sputtering and brazing started
- Scoping test of PEEK in liquid sodium
  
- Experimental infrastructure (ATEFA facility, AMTEC prototype) to be set into operation by September 2015
- Experimental campaign focused on:
  - Open issues (BASE analysis, electrodes)
  - Performance evaluation
  - Technology improvement
- Further analysis of the PEEK samples



ATEFA facility

LIMTECH Alliance *and* HEMCP:  
*Helmholtz Energy Materials  
Characterization Platform*

# Thank you for your attention

Contact:

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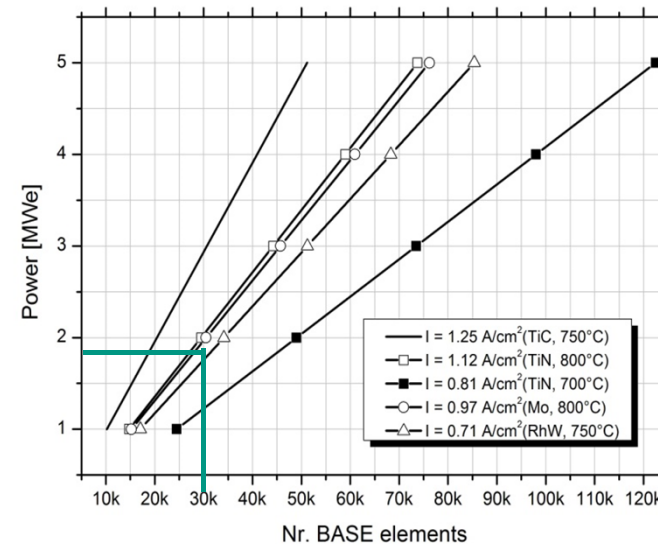
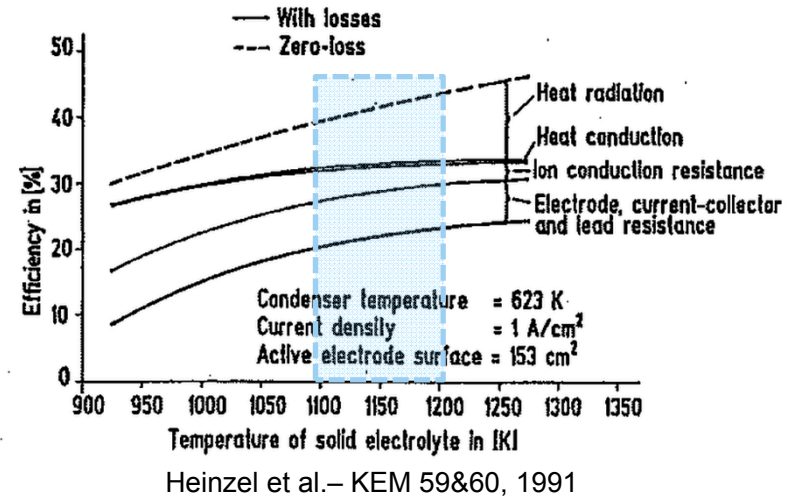
Funding:

LIMTECH (Liquid Metal Technology) Alliance → HGF Alliance

HEMCP (Helmholtz Energy Materials Characterization Platform)

# Challenges of AMTEC for CSP:

- Efficiency optimization
  - Heat losses (10 – 15 %)
  - Electrical losses (12 – 15 %)
- Constant performances for long time operation (~ 60000 h)
  - Degradation of BASE (formation of molten dendrites, internal crack formation, contamination through impurities, grain growth)
  - Electrode sintering (grain growth)
- Cyclic operation (fatigue)
- Costs: special materials for construction, safety design
- Present technology: ~2MWe with ~30k BASE elements at 750-800°C and 1 A/cm<sup>2</sup>



Onea et al. – accepted for publication MHD, 2015

# Liquid metals as HTF for CSP

Heat transfer fluid (HTF)	Therminol VP-1	Solar Salt (60% NaNO <sub>3</sub> /40% KNO <sub>3</sub> )	Liquid metals (General)	Liquid Sodium	Steam@ 10bar
Stability	400 °C	~ 585 °C	800-2600 °C	873 °C	-
Therm. Cond. 400 / 600 °C	0.0956 W/mK	≥ 0.58 W/mK	> 15 W/mK	60 W/mK	0.09 W/mK
Melting Point	12 °C	228 °C	< 0-300 °C	97.7 °C	-
Density 400 / 600 °C	696 kg/m <sup>3</sup>	1867 kg/m <sup>3</sup>	500 kg/m <sup>3</sup> - 10000 kg/m <sup>3</sup>	810 kg/m <sup>3</sup>	2.5 kg/m <sup>3</sup>
Spec. heat 400 / 600 °C	~ 2,5 kJ/kgK	1,5 kJ/kgK	0,15 - 4,16 kJ/kgK	1,25 kJ/kgK	2.2 kJ/kgK

## Advantages of Na:

- Thermal stability 870 °C
- High thermal conductivity
- Low melting point
- Low density

## Disadvantages of Na

- Reactive with air and water



Safety oriented design and maintenance

## ■ Comparison of HTFs \*:

Na identified as the best HTF for CSP-TS

- $\frac{(E_{grid})^{Na}}{(E_{grid})^{Sol\ Salt}} \sim 1.06$

- $\frac{(E_{grid})^{Na}}{(E_{grid})^{Steam}} \sim 1.11$

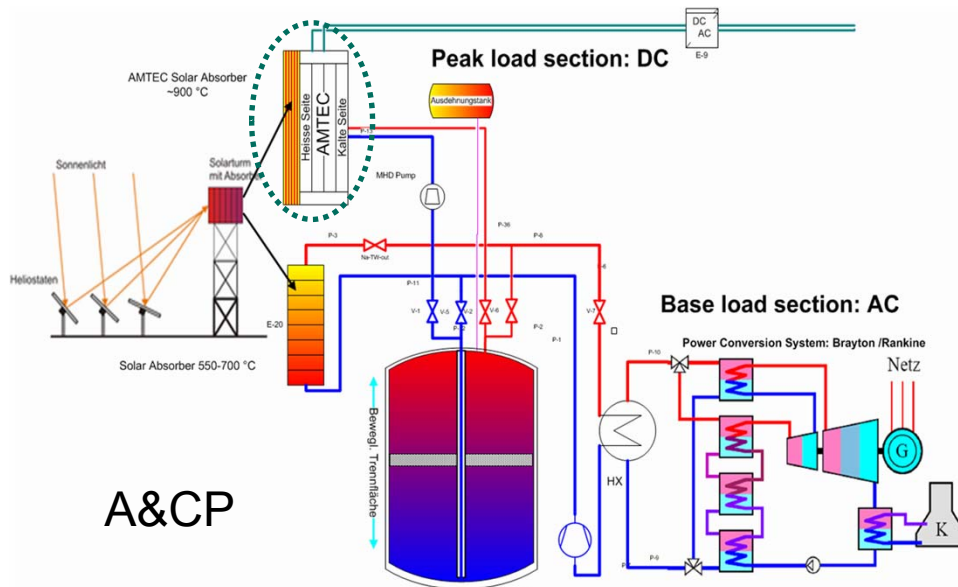


Sodium good candidate as a HTF in CSP systems

\* Liu et al. SE 101, 220-231, 2014

# Long time perspective

## Hybrid system CSP + Na + AMTEC

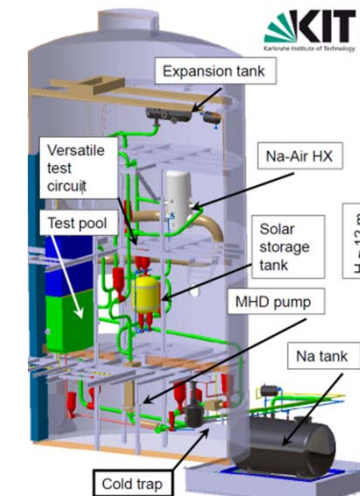


CSP - AMTEC hybrid cycle (\*Hering et al., E2C Maastricht, 2012)

- HTF + storage fluid: Na
- $T_{\text{receiver}} \sim 550 - 700 \text{ } ^\circ\text{C}$
- AMTEC as topping system
- Plant size:  $\sim 100 \text{ MWth}$
- Compensation of fluctuations in source
- AMTEC excess energy reused in TS
- Longer and more flexible operation
- Increase PCS lifetime
- Increase system efficiency\* ( $\eta \sim 25 \%$ )

### Research in projects LIMTECH, HEMCP

- AMTEC prototype
- Material characterization in Na  
→ Sodium Loop for Test Materials and Corrosion (SOLTEC)
- High temperature thermal storage device
- A&CP → integration of small prototype in Karlsruhe Sodium Laboratory (KASOLA)





# Different direct energy conversion systems

Energy Conversion System	$T_{\min} / T_{\max}$ [°C]	$\eta$ [%]	Comments
Thermoelectric converter (Seebeck effect)	100 / 900	< 20	Availability of good semiconductors that sustain high temperatures
Thermionic converter	$T_{\max} > 1500$	< 20	High operating temperatures
Solar cell	30 / 60	15 - 30	High costs, "environmentally" expensive, contain toxic elements
Fuel cell	200 / 1000	35 – 60	Fuel storage, fuel generation and lifetime research needed. High costs.
<b>AMTEC</b>	200 / 1000	~ 20 (40)	Lifetime and efficiency under research

- All ECS are under development for being competent in the industry
- Issues like lifetime, efficiency, output power and costs need still to be treated

## AMTEC - Advantages

- **High theoretical efficiency (40%)**
- **Flexible regarding the heat source**
- Suitable for modular design
- No moving parts
- Silent operation
- **Flexible connection (to other heat-to-electric systems)**



**Application for solar energy conversion**