



---

**Forschungszentrum Karlsruhe**  
Technik und Umwelt

---

**Wissenschaftliche Berichte**  
FZKA 5588

# **Thermal Properties of Water**

**K. Thurnay**

Institut für Neutronenphysik und Reaktortechnik  
Projekt Nukleare Sicherheitsforschung

Juni 1995

---



**Forschungszentrum Karlsruhe**  
**Technik und Umwelt**  
**Wissenschaftliche Berichte**  
**FZKA 5588**

# **Thermal Properties of Water**

**K. Thurnay**  
**Institut für Neutronenphysik und Reaktortechnik**  
**Projekt Nukleare Sicherheitsforschung**

**Forschungszentrum Karlsruhe GmbH, Karlsruhe**  
**1995**

Als Manuskript gedruckt  
Für diesen Bericht behalten wir uns alle Rechte vor

Forschungszentrum Karlsruhe GmbH  
Postfach 3640, 76021 Karlsruhe

ISSN 0947-8620

## **Abstract**

The report describes AQUA, a code developed in the Forschungszentrum Karlsruhe to calculate thermal properties of the water in steam explosions. AQUA bases on the H.G.K. water code, yet supplies - besides of the pressure and heat capacity functions - also the thermal conductivity, viscosity and the surface tension. AQUA calculates in a new way the thermal properties in the two phase region, which is more realistic as the one used in the H.G.K. code. AQUA is equipped with new, fast running routines to convert temperature-density dependent states into temperature-pressure ones.

AQUA has a version to be used on line and versions adapted for batch calculations. A complete description of the code is included.

## **Thermische Eigenschaften des Wassers.**

### **Zusammenfassung.**

Der Bericht befaßt sich mit dem Code AQUA. AQUA wurde im Forschungszentrum Karlsruhe entwickelt um bei der Untersuchung von Dampfexplosionen die thermischen Eigenschaften des Wassers zu liefern. AQUA ist eine Fortentwicklung des H.G.K.-Wassercodes, aber er berechnet - neben Druck- und Wärmeeigenschaften - auch die Wärmeleitfähigkeit, die Viskosität und die Oberflächenspannung. Im Zweiphasengebiet beschreibt AQUA die thermischen Eigenschaften mit einer neuen Methode, die realistischer ist, als das in der H.G.K.-Code dargebotene Verfahren. AQUA verfügt über schnell laufende Hilfsprogramme, die die Umwandlung der Temperatur-Dichte Darstellungen in die Druck-Dichte Darstellungen ermöglichen.

AQUA kann man sowohl im Dialogbetrieb benutzen als auch beim rechnen im Stapelverfahren.

Eine ausführliche Beschreibung des Programmsystems rundet den Bericht ab.

## Glossary

Name	Code	Description
<b>A</b>	A	specific free energy [ $J/kg$ ]
<b>A<sup>0</sup></b>		$A/(R_{gas} \cdot T)$
<b><math>\alpha</math></b>	ALF	-0.1, critical exponent for the heat capacity
<b><math>\beta</math></b>	BET	0.325, critical exponent for the density
<b>C<sub>v</sub></b>	CV	heat capacity at constant volume [ $J/(kg \cdot K)$ ]
	CVL	heat capacity at constant volume of the liquid [ $R_{gas}$ ]
	CVV	heat capacity at constant volume of the vapor [ $R_{gas}$ ]
<b>C<sub>v</sub><sup>0</sup></b>	CV0	$C_v/R_{gas}$
<b>C<sub>p</sub></b>	CP	heat capacity at constant pressure [ $J/(kg \cdot K)$ ]
<b>C<sub>p</sub><sup>0</sup></b>	CP0	$C_p/R_{gas}$
	CPC	$C_p/C_v$
<b><math>\gamma</math></b>	GAM	1.24, critical exponent for $\partial P/\partial \rho$
	DL	density of the liquid [ $g/cm^3$ ]
	DV	density of the vapor [ $g/cm^3$ ]
	DX	density in $g/cm^3$
<b><math>\eta</math></b>	ETA	dynamic viscosity [ $Pa \cdot sec$ ]
<b><math>\eta_c</math></b>	ETC	critical viscosity, $3.95 \cdot 10^{-5} Pa \cdot sec$
	ETL	dynamic viscosity of the liquid [ $Pa \cdot sec$ ]
	ETV	dynamic viscosity of the vapor [ $Pa \cdot sec$ ]
<b>G</b>	G	specific Gibbs function [ $J/kg$ ]
<b>G<sup>x</sup></b>	GL	G in the saturated states
<b>G<sup>0</sup></b>	G0	$G/(R_{gas} \cdot T)$
	GASCON	the gas-law constant of the water in $J/g \cdot K$
<b>H</b>	H	specific enthalpy [ $J/kg$ ]
<b>H<sup>0</sup></b>	H0	$H/(R_{gas} \cdot T)$
<b>H<sub>p</sub></b>	H1P	$\partial H/\partial P$ , pressure derivative of H
<b>H<sub><math>\rho</math></sub></b>		$\partial H/\partial \rho$ , density derivative of H

Name	Code	Description
$\chi_T$	CHIT	reduced isothermal compressibility ( s. Eq. [3.9] )
<b>IAPS</b>		The International Association for the Properties of Steam
$\lambda$	XLMB	thermal conductivity [ $W/(m \cdot K)$ ]
	XLL	thermal conductivity of the liquid [ $W/(m \cdot K)$ ]
	XLV	thermal conductivity of the vapor [ $W/(m \cdot K)$ ]
	MW	Molal mass ( 18.0152 $g/mol$ )
<b>P</b>	P	pressure [ Pa ]
$P_c$	PC	critical pressure, 22.0549 MPa
$P^x$	PS	vapor pressure
$P_{SL}$	PSL	pressure on the liquid spinodal
$P_{SV}$	PSV	pressure on the vapor spinodal
$P_\rho$	P1R	$\partial P / \partial \rho$ , density derivative of the pressure
	PL1R	density derivative of the pressure in the saturated liquid
	PV1R	density derivative of the pressure in the saturated vapor
$P_\rho^0$	P1R0	$P_\rho / (R_{gas} \cdot T)$
$\partial P / \partial \rho _s$	P1RS	density derivative of the pressure at constant entropy
$P_T$	P1T	$\partial P / \partial T$ , temperature derivative of the pressure
	PL1T	temperature derivative of the pressure in the saturated liquid
	PV1T	temperature derivative of the pressure in the saturated vapor
$P_T^0$	P1T0	$P_T / (R_{gas} \cdot \rho)$
$P^\#$	PRN	Prandtl number, $C_p \cdot (\eta / \lambda)$
	Q	pressure in MPa-s
$R_{gas}$	RGAS	gas-law constant of the water, 461.522 $J/(kg \cdot K)$
$\rho$	RH	density [ $kg/m^3$ ]
$\rho_c$	RHC	critical density, 322. $kg/m^3$
$\rho_{SL}$	RHSL	density of the liquid spinodal
$\rho_{SV}$	RHSV	density of the vapor spinodal
$\rho_T^0$	R1T0	reduced temperature-derivative of the density, $T / \rho \cdot \partial \rho / \partial T$

Name	Code	Description
$r_L$	RL1T0	reduced temperature-derivative of the density of the saturated liquid, $T/\rho \cdot d\rho_L/dT$
$r_v$	RV1T0	reduced temperature-derivative of the density of the saturated vapor, $T/\rho \cdot d\rho_v/dT$
<b>S</b>	S	specific entropy [ $J/(kg \cdot K)$ ]
$S_c$	SC	critical entropy, $9.43 \cdot R_{gas} = 435.244 J/(kg \cdot K)$
$\sigma$	SRFTS	surface tension [ $N/M$ ]
	SL	specific entropy of the liquid [ $J/(kg \cdot K)$ ]
	SREF	$7.6180802 \cdot R_{gas}$
$S^0$	S0	$S/R_{gas}$
<b>T</b>	T	temperature [ K ]
$T_c$	TC	critical temperature, 647.126 K
	TPS1T	$T \cdot dP^x/dT$ [ Pa ]
	TTPS2T	$T \cdot T \cdot d^2P^x/dT^2$ [ Pa ]
$dT/dP^x$	TS1P	inverse of the derivative of the vapor pressure [ $K/Pa$ ]
$\tau$	Y	scaled temperature, $1 - T/T_c$
<b>U</b>	U	specific internal energy [ $J/kg$ ]
	UREF	$-4328.455039 \cdot R_{gas} \cdot T$
$U^0$	U0	$U/(R_{gas} \cdot T)$
$U_\rho$	U1R	$\partial U/\partial \rho$ , density derivative of U
<b>V</b>	V	specific volume, $1/\rho$
$v_s$	VS	velocity of sound, $v_s^2 = \partial P/\partial \rho _s$
	VQ	vapor quality, $\frac{m_v}{m_L + m_v}$
<b>w</b>	W	$\sqrt{\tau}$
<b>Z</b>	Z	factor of reality, $(P \cdot V)/(R_{gas} \cdot T)$



## Contents

<b>1.0 Introduction</b>	<b>1</b>
<b>2.0 A short description of the H.G.K.-code</b>	<b>3</b>
2.1 The Helmholtz function	3
2.2 The base function	3
2.3 The ideal gas function	4
2.4 The residual function	5
2.5 The critical data	7
2.6 The validity domain	8
<b>3.0 Transport properties of the water.</b>	<b>9</b>
Reduced variables	9
3.1 Viscosity	9
3.2 Thermal conductivity	10
3.3 Surface tension	13
<b>4.0 Thermal properties in the two phase region</b>	<b>15</b>
4.1 Thermal states of the subcritical water	15
4.2 Water properties in the region of instability	16
<b>5.0 Describing the water properties in the saturated states</b>	<b>27</b>
5.1 Describing the saturation line of the water	27
The saturation pressure	27
The saturation line	29
5.2 Describing the spinodals	30
Calculating the spinodals of the water.	30
The spinodal densities	31
The spinodal pressures	33
5.3 The entropy on the saturation line	34
5.4 The transport properties on the saturation line	36
The viscosity	36
The thermal conductivity	37
5.5 Other saturated properties	39
The density derivative of the pressure	39
The temperature derivative of the pressure	41
The heat capacities	43
The free enthalpy, G	45
<b>6.0 Calculating the densities from the pressures</b>	<b>47</b>
6.1 Calculating small density changes	47
6.2 Calculating large density changes	49
<b>7.0 Using AQUA</b>	<b>55</b>
7.1 The codesystem AQUA	55
SERVUS : auxilliary datasets and routines	55
7.2 WATER	56
7.3 String-routines	58
KAISOT	58
KAISOK	61
7.4 IVA-routines	62
WASSER	62
DAMPF	62
RDRDP	63
SAET01	63

VD1	64
<b>8.0 References</b>	<b>65</b>
<b>Appendix A. Some thermal relations</b>	<b>67</b>
A.1 The derivatives of the free energy	67
A.2 The enthalpy and its derivatives	67
A.3 The sonic velocity	69
A.4 The Joule Thomson coefficient	70
<b>Appendix B. Derivatives of some saturated properties</b>	<b>71</b>
B.1 Derivatives of the vapor pressure	71
B.2 Derivatives of the saturated densities	72
<b>Appendix C. Adjusting the saturated states</b>	<b>75</b>
<b>Appendix D. Module trees</b>	<b>77</b>
D.1 Interactive routines	77
WATER	77
KAISOT	78
KAISOK	78
D.2 IVA3-ROUTINES	79
WASSER	79
DAMPF	79
RDRDP	80
SAET01	80
D.3 SUB-ROUTINES	81
HYDOR	81
SUBCRI	81
TWOPHA	81
GPCORR	82
DINSU	82
DINLI	82
DINVA	82
<b>Appendix E. Common blocks</b>	<b>83</b>
<b>Appendix F. Property routines</b>	<b>85</b>
F.1 INR105.AQUA.FORT(BLOCK)	85
F.2 INR105.AQUA.FORT(ONLYT)	86
F.3 INR105.AQUA.FORT(BASE)	87
F.4 INR105.AQUA.FORT(RESID)	88
F.5 INR105.AQUA.FORT(SIGMA)	90
F.6 INR105.AQUA.FORT(TRANSP)	90
F.7 INR105.AQUA.FORT(SSP)	92
F.8 INR105.AQUA.FORT(TS1)	93
F.9 INR105.AQUA.FORT(SSRHO)	93
F.10 INR105.AQUA.FORT(SPIRHO)	95
F.11 INR105.AQUA.FORT(SPIP)	96
F.12 INR105.AQUA.FORT(SSPER)	96
F.13 INR105.AQUA.FORT(SSPET)	97
F.14 INR105.AQUA.FORT(SSCEV)	98
F.15 INR105.AQUA.FORT(SSGIS)	99
F.16 INR105.AQUA.FORT(SSETA)	100
F.17 INR105.AQUA.FORT(SSLAM)	101
F.18 INR105.AQUA.FORT(DILE)	101
F.19 INR105.AQUA.FORT(DIVE)	102
F.20 INR105.AQUA.FORT(PLMAX)	102

F.21	INR105.AQUA.FORT(P1200)	102
F.22	INR105.AQUA.FORT(P600)	103
F.23	INR105.AQUA.FORT(P400)	103
F.24	INR105.AQUA.FORT(P322)	103
F.25	INR105.AQUA.FORT(P250)	104
<b>Appendix G. Secondary routines</b>		<b>105</b>
G.1	INR105.AQUA.FORT(DGFIND)	105
G.2	INR105.AQUA.FORT(DLFIND)	107
G.3	INR105.AQUA.FORT(DVFIND)	109
G.4	INR105.AQUA.FORT(DINSU)	110
G.5	INR105.AQUA.FORT(DINLI)	111
G.6	INR105.AQUA.FORT(DINVA)	112
G.7	INR105.AQUA.FORT(QUADO)	113
G.8	INR105.AQUA.FORT(MONIKA)	114
G.9	INR105.AQUA.FORT(SORBET)	115
G.10	INR105.AQUA.FORT(TCORR)	116
G.11	INR105.AQUA.FORT(GPCORR)	116
G.12	INR105.AQUA.FORT(SUBCRI)	117
G.13	INR105.AQUA.FORT(TWOPHA)	118
G.14	INR105.AQUA.FORT(REDUK)	118
G.15	INR105.AQUA.FORT(REDUKE)	119
G.16	INR105.AQUA.FORT(HYDOR)	121
<b>Appendix H. User-Interfaces</b>		<b>123</b>
H.1	INR105.AQUA.FORT(WATER)	123
H.2	INR105.AQUA.FORT(KAISOT)	126
H.3	INR105.AQUA.FORT(KAISOK)	127
<b>Appendix I. H.G.K.-users</b>		<b>131</b>
I.1	INR105.AQUA.FORT(HGKSPN)	131
I.2	INR105.AQUA.FORT(HGKSAT)	134
I.3	INR105.AQUA.FORT(TRNSAT)	136
<b>Appendix J. The IVA Routines</b>		<b>139</b>
J.1	INR105.AQUA.FORT(SAET01)	139
J.2	INR105.AQUA.FORT(WASSER)	140
J.3	INR105.AQUA.FORT(DAMPF)	142
J.4	INR105.AQUA.FORT(RDRDP)	143
J.5	INR105.AQUA.FORT(VD1)	144
<b>Appendix K. Service routines</b>		<b>147</b>
K.1	INR105.AQUA.FORT(SIGNAT)	147
K.2	INR105.AQUA.FORT(MUSTER)	149
K.3	INR105.AQUA.FORT(NAPSZM)	149
<b>Appendix L. Return codes</b>		<b>151</b>
<b>Appendix M. TSO-procedures, calculating thermal properties of the water</b>		<b>153</b>
<b>Appendix N. Thermal property labels</b>		<b>155</b>

## Figures

1.	Thermal conductivity of the steam in the critical region as a function of density at constant pressures . . . . .	10
2.	Thermal conductivity of the water on supercritical isotherms . . . . .	12
3.	Pressure surface of the water. H.G.K.-code . . . . .	14
4.	Sketch of P,V diagram with saturation and spinodal lines. $V = 1 / \rho$ . . . . .	15
5.	Pressure of the water in the two phase region. $T = 500$ K . . . . .	16
6.	Water pressures in the superheated liquid state . . . . .	17
7.	Free enthalpy of the water in the two phase region. $T = 500$ K . . . . .	18
8.	Entropy of the water in the two phase region. $T = 500$ K . . . . .	18
9.	Free energy of the water in the two phase region. $T = 500$ K . . . . .	19
10.	Sonic velocity of the water in the two phase region. $T = 500$ K . . . . .	20
11.	Heat capacity - $C_v$ - of the water in the two phase region. $T = 500$ K . . . . .	21
12.	Thermal conductivity of the water in the two phase region. $T = 500$ K . . . . .	22
13.	Viscosity of the water in the two phase region. $T = 500$ K . . . . .	23
14.	Pressure surface of the water. AQUA . . . . .	24
15.	Heat capacity surface of the water. AQUA. View from the cold side. . . . .	25
16.	Heat capacity surface of the water. AQUA. View from the liquid side. . . . .	26
17.	Vapor pressure formula of Wagner and Saul . . . . .	28
18.	Spinodal densities in the vicinity of the critical point . . . . .	31
19.	Saturation and spinodal densities of the water . . . . .	32
20.	Pressures on the spinodals of the water . . . . .	33
21.	Entropy on the saturation line of the water . . . . .	34
22.	Entropy of the saturated states in the vicinity of $T_c$ . . . . .	35
23.	Viscosity of the water in the saturated states . . . . .	36
24.	Thermal conductivity of the water as function of the temperature . . . . .	38
25.	Density derivatives of the pressure in the saturated states . . . . .	39
26.	Density derivatives of the pressure in the vicinity of $T_c$ . . . . .	40
27.	Temperature derivatives of the pressure in the saturated states . . . . .	41
28.	Temperature derivatives of the pressure in the vicinity of $T_c$ . . . . .	42
29.	Heat capacities of the water in the saturated states. . . . .	43
30.	Heat capacities of the water in the vicinity of $T_c$ . . . . .	44
31.	Free enthalpy - $G$ - of the water in the saturated states . . . . .	45
32.	Water pressures on the 647 K isotherm . . . . .	48
33.	Water. Density-pressure chart . . . . .	50
34.	The procedure userid.SERVUS.CNTL(IEBGENER). . . . .	55
35.	The procedure userid.AQUA.CLIST(KAISOT) . . . . .	58
36.	The procedure userid.AQUA.CLIST(KAISOK) . . . . .	61
37.	Module tree for the procedure WATER . . . . .	77
38.	Module tree for the procedure K AISOT . . . . .	78
39.	Module tree for the procedure K AISOK . . . . .	78
40.	Module tree for the procedure WASSER . . . . .	79
41.	Module tree for the procedure DAMPF . . . . .	79
42.	Module tree for the procedure RDRDP . . . . .	80
43.	Module tree for the procedure SAETO1 . . . . .	80
44.	Module tree for the procedure HYDOR . . . . .	81
45.	Module tree for the procedure SUBCRI . . . . .	81
46.	Module tree for the procedure TWOPHA . . . . .	81
47.	Module tree for the procedure GPCORR . . . . .	82
48.	Module tree for the function DINSU . . . . .	82
49.	Module tree for the function DINLI . . . . .	82
50.	Module tree for the function DINVA . . . . .	82

## Tables

1.	The coefficients of the base function	4
2.	The coefficients of the ideal gas function	4
3.	The exponents $i(1) - i(40)$ of the residual function	5
4.	The exponents $k(1) - k(40)$ of the residual function	5
5.	The coefficients of the near-critical description in the residual function	5
6.	The coefficients $g(1) - g(40)$ of the residual function	6
7.	The coefficients $B(j,m)$ of the viscosity	10
8.	The coefficients $b(i,n)$ of the thermal conductivity	11
9.	The coefficients $A(1) - A(8)$ of the HGK-vapor pressure equation	27
10.	The coefficients $A(1) - A(12)$ in the Eq. of the saturated liquid	29
11.	The coefficients $Z(1) - Z(12)$ in the Eq. of the saturated vapor	29
12.	Polynomial coefficients $A(1) - A(9)$ of the density of the liquid spinodal	31
13.	Polynomial coefficients $C(1) - C(7)$ of the density of the vapor spinodal	31
14.	Polynomial coefficients $A(1) - A(9)$ of the pressure of the liquid spinodal	33
15.	Polynomial coefficients $C(1) - C(8)$ of the pressure of the vapor spinodal	34
16.	Polynomial coefficients $A(1) - A(9)$ of the entropy of the saturated liquid	35
17.	Polynomial coefficients $A(1) - A(11)$ of the viscosity in the saturated liquid	37
18.	Polynomial coefficients $C(1) - C(9)$ of the viscosity in the saturated vapor	37
19.	Coefficients $A(1) - A(8)$ of the thermal conductivity in the saturated liquid	37
20.	Coefficients $C(1) - C(9)$ of the thermal conductivity in the saturated vapor	38
21.	Polynomial coefficients $A(1) - A(12)$ of $\partial P/\partial \rho$ in the saturated liquid	39
22.	Polynomial coefficients $C(1) - C(9)$ of $\partial P/\partial \rho$ in the saturated vapor	40
23.	Polynomial coefficients $A(1) - A(12)$ of $\partial P/\partial T$ in the saturated liquid	41
24.	Polynomial coefficients $C(1) - C(10)$ of $\partial P/\partial T$ in the saturated vapor	42
25.	Polynomial coefficients $A(1) - A(12)$ of $C_v$ in the saturated liquid	43
26.	Polynomial coefficients $C(1) - C(10)$ of $C_v$ in the saturated vapor	44
27.	Polynomial coefficients $A(1) - A(9)$ of the free enthalpy of the saturated states	45
28.	Coefficients of $P_{Lmax}(T)$	47
29.	Polynomial coefficients $U(1) - U(9)$ of the correcting term for $P'_{SL}$	49
30.	Polynomial coefficients $W(1) - W(6)$ of the correcting term for $P'_{SV}$	49
31.	Polynomial coefficients $A(1) - A(8)$ of $P_{250}(T)$	51
32.	Polynomial coefficients $A(1) - A(10)$ of $P_{322}(T)$	51
33.	Polynomial coefficients $A(1) - A(10)$ of $P_{400}(T)$	51
34.	Polynomial coefficients $A(1) - A(10)$ of $P_{500}(T)$	52
35.	Polynomial coefficients $A(1) - A(10)$ of $P_{800}(T)$	52
36.	Table of return codes, IECCH	151
37.	TSO-procedures in the dataset INR105.AQUA.CLIST	154
38.	Thermal property labels "KLASSE" used in AQUA.DATA	156

## 1.0 Introduction

Water is the most commonly used working substance in chemistry and engineering. It is also the cooling medium of the pressurized and boiling water nuclear reactors. Therefore a solid knowledge of the thermophysical properties of the water in the range of possible temperatures, pressures and densities is essential to calculate hazardous events - which may emerge once in a while in a normally operating reactor.

In the Institut für Neutronenphysik und Reaktortechnik we are presently developing a thermohydrodynamical code, IVA3, /21/ to be able to calculate the consequences of some hypothesized reactor failures. To supply IVA3 with the thermophysical properties of the water we adopted the H.G.K.-equation of state, developed in the years 1981-84 by Haar, Gallagher and Kell, /10/ and recommended by the IAPS ( s. /12/). The H.G.K.-equation of state calculates the various thermal properties via a continuous Helmholtz free energy function, which is defined over the whole range of the fluid states of water and is fitted to the available experimental data.

The corresponding computer code described in /10/ has some drawbacks and doesn't fit well to our IVA3-code. Item 1 : the code doesn't describe the properties viscosity, thermal conductivity and surface tension. Item 2 : the description of the properties in the mixed state is unphysical. Item 3 : IVA3 needs the thermophysical properties as functions of the temperature and pressure, whereas the H.G.K.-code supplies them as functions of the temperature and density. Although there is a modul in the H.G.K.-code, DFIND, which turns temperature-pressure states of water into corresponding temperature-density ones, yet DFIND consumes too much time during the iteration - needed for the conversion of these states.

In the following the code AQUA will be presented.

AQUA is an extension of the H.G.K.-code, modified to suit our special conditions. In the largest part of the temperature-density domain AQUA returns the same thermophysical property values as the H.G.K.-code, but in the unstable part of the two phase region the H.G.K.-properties are substituted with physically meaningful two phase state values.

AQUA calculates at a given (  $T$  ,  $\rho$  ) state besides of the pressure also the following thermal properties of the water :

- S, the specific entropy,
- the energy functions
  - U, the specific energy ,
  - H, the specific enthalpy ,
  - A, the specific free energy ,
  - G, the specific free enthalpy ,
- the pressure derivatives  $\partial P/\partial \rho$  and  $\partial P/\partial T$ ,
- the heat capacities  $C_p$  and  $C_v$ ,
- the derivatives  $\partial H/\partial P$  and  $\partial \rho/\partial T$
- the viscosity  $\eta$  and the thermal conductivity  $\lambda$ ,
- the Joule Thomson coefficient,  $\partial T/\partial P | H$ ,
- the surface tension, the velocity of sound and the Prandtl number.

AQUA calculates the properties also as temperature-pressure dependent functions. It incorporates additional functions and procedures to speed up the (  $T$  ,  $P$  )  $\rightarrow$  (  $T$  ,  $\rho$  ) conversion.

AQUA has a set of routines ( SAET01, WASSER, DAMPF, RDRDP, VD1 ) which serve to adjust AQUA to the conditions of the code IVA3.

AQUA needs and returns all the properties in SI-units.

AQUA calculates reasonable property-values if the temperature exceeds 260.00 K and if the density remains below 1200 kg/m<sup>3</sup>. Trying to convert pressures to densities above 5000 K may lead to dissatisfaction.



## 2.0 A short description of the H.G.K.-code

### 2.1 The Helmholtz function

Haar, Gallagher and Kell developed and tested - in close cooperation with the International Association for the Properties of Steam ( IAPS ) - an analytic description for the thermal and caloric properties of the water, /10/ to replace the standard tables used formerly. The resulting properties are the Provisional IAPS Formulation 1982 for the Thermodynamic Properties of Ordinary Water Substance for Scientific and General Use.

Haar, Gallagher and Kell constructed their algorithm by developing a mathematical description for the Helmholtz function ( alias free energy )  $A(\rho, T)$ . Using Eq. [A.4 - A.5] one can develop from A all the needed thermal and caloric properties ( s. Appendix A, "Some thermal relations" ).

The description of the Helmholtz function consist from a simple base function to describe the liquid and supercritical states, of an additional residual function for a proper fitting in the mixed states and of a solely temperature-dependent term for the ideal gas states.

$$A(\rho, T) = A_{base}(\rho, T) + A_{residual}(\rho, T) + A_{idealgas}(T) \quad . \quad [2.1]$$

### 2.2 The base function

is a simple analytic equation

$$A_{base}(\rho, T) = R_{gas} \cdot T \left[ -\ln(1-y) - \frac{\beta-1}{1-y} + \frac{\alpha+\beta+1}{2 \cdot (1-y)^2} + 4y \left( \frac{\bar{B}}{b} - \gamma \right) - \frac{\alpha-\beta+3}{2} + \ln \left( \frac{\rho \cdot R_{gas} \cdot T}{P_0} \right) \right] \quad [2.2]$$

In the above eq.

$$y = \frac{b(T) \cdot \rho}{4} \quad [2.3]$$

is a dimensionless density,

$$b = b_1 - b_2 \cdot \ln \frac{T_0}{T} + \sum_{j=3}^7 b_j \cdot \left( \frac{T_0}{T} \right)^{j-2} \quad [2.4]$$

is the 2nd virial and

$$\bar{B} = B_1 + \sum_{j=3}^7 B_j \cdot \left( \frac{T_0}{T} \right)^{j-2} \quad [2.5]$$

is the excluded volume. The pressure and temperature coefficients are

$$P_0 = 1.01325 \text{ bar} \quad \text{resp.} \quad T_0 = 647.073 \text{ K} \quad [2.6]$$



and the greek letters mean

$$\alpha = 11. \quad , \quad \beta = 133/3 \quad \text{and} \quad \gamma = 3.5 \quad .$$

The coefficients of the Eq.s [2.4] , [2.5] are the following ones :

$b_j (cm^3 / g)$	$j$	$B_j (cm^3 / g)$
0.7478629	1	1.1278334
-0.3540782	2	0.0
0.0	3	-0.5944001
0.0	4	-5.010996
0.007159876	5	0.0
0.0	6	0.63684256
-0.003528426	7	0.0

**Note:** The descriptions given in /10/ for the 2nd virial and for the excluded volume are not in accordance with those, used in the routine BB. I describe here the properties in accordance with the routine BB.

### 2.3 The ideal gas function

has the following form :

$$A_{idealgas}(T) = -R_{gas} \cdot T \left[ 1 + \left( \frac{C_1}{T_R} + C_2 \right) \ln(T_R) + \sum_{i=3}^{18} C_i \cdot T_R^{i-6} \right] \quad [2.8]$$

with  $T_R \equiv T / 100 K$  .

The C-s are given in the table below.

$j$	$C_j$	$j$	$C_j$
1	19.730271018	10	$4.1238460633 \cdot 10^{-3}$
2	20.9662681977	11	$-2.7929052852 \cdot 10^{-4}$
3	-0.483429455355	12	$1.4481695261 \cdot 10^{-5}$
4	6.05743189245	13	$-5.6473658748 \cdot 10^{-7}$
5	22.56023885	14	$1.6200446 \cdot 10^{-8}$
6	-9.87532442	15	$-3.303822796 \cdot 10^{-10}$
7	-4.3135538513	16	$4.51916067368 \cdot 10^{-12}$
8	$4.58155781 \cdot 10^{-1}$	17	$-3.70734122708 \cdot 10^{-14}$
9	$-4.7754901883 \cdot 10^{-2}$	18	$1.37546068238 \cdot 10^{-16}$

## 2.4 The residual function

has two parts :

$$A_{resl}(\rho, T) = \sum_{i=1}^{36} \frac{g_i}{k(i)} \left( \frac{647.073}{T} \right)^{k(i)} [1 - e^{-\rho}]^{k(i)} + \sum_{i=37}^{40} g_i \delta_i^{k(i)} \cdot \exp(-\alpha_i \delta_i^{k(i)} - \beta_i \tau_i^2) \quad [2.9]$$

The first 36 terms are needed to fit the description to the experimental data in the low temperature mixed states, the remaining 3 terms correct the function in the vicinity of the critical point.

$$\delta_i = \frac{\rho - \rho_i}{\rho_i} \quad \text{and} \quad \tau_i = \frac{T - T_i}{T_i} \quad [2.10]$$

are reduced densities and temperatures. The coefficients used by the residual function are given in the following tables :

0	0	0	0	1	1	1	1	2	2
2	2	3	3	3	3	4	4	4	4
5	5	5	5	6	6	6	6	8	8
8	8	2	2	0	4	2	2	2	4

2	3	5	7	2	3	5	7	2	3
5	7	2	3	5	7	2	3	5	7
2	3	5	7	2	3	5	7	2	3
5	7	1	4	4	4	0	2	0	0

i	37	38	39	40
$\alpha_i$	34.	40.	30.	1050.
$\beta_i$	20000	20000	40000	25.
$\rho_i$	0.319	0.319	0.319	1.550
$T_i$	640.	640.	641.6	270.

$-.53062968529023 \cdot 10^3$	$.22744901424408 \cdot 10^4$	$.78779333020687 \cdot 10^3$	$-.69830527374994 \cdot 10^2$	$.17863832875422 \cdot 10^5$
$-.39514731563338 \cdot 10^5$	$.33803884280753 \cdot 10^5$	$-.13855050202703 \cdot 10^5$	$-.25637436613260 \cdot 10^6$	$.48212575981415 \cdot 10^6$
$-.34183016969660 \cdot 10^5$	$.12223156417448 \cdot 10^6$	$.11797433655832 \cdot 10^7$	$-.21734810110373 \cdot 10^7$	$.10829952168620 \cdot 10^7$
$-.25441998064049 \cdot 10^6$	$-.31377774947767 \cdot 10^7$	$.52911910757704 \cdot 10^7$	$-.13802577177877 \cdot 10^7$	$-.25109914369001 \cdot 10^6$
$.46561826115608 \cdot 10^7$	$-.72752773275387 \cdot 10^7$	$.41774246148294 \cdot 10^6$	$.14016358244614 \cdot 10^7$	$-.31555231392127 \cdot 10^7$
$.47929666384584 \cdot 10^7$	$.40912664781209 \cdot 10^6$	$-.13626369388386 \cdot 10^7$	$.69625220862664 \cdot 10^6$	$-.10834900096447 \cdot 10^7$
$-.22722827401688 \cdot 10^6$	$.38365486000660 \cdot 10^6$	$.68833257944332 \cdot 10^4$	$.21757245522644 \cdot 10^5$	$-.26627944829770 \cdot 10^4$
$-.70730418082074 \cdot 10^5$	-.225	-1.68	0.055	-93.0

The following figures display the contributions of the basic, ideal and residual functions to the total pressure respective to the total heat capacity at the boiling point and at the temperatures 500 K and 800 K.

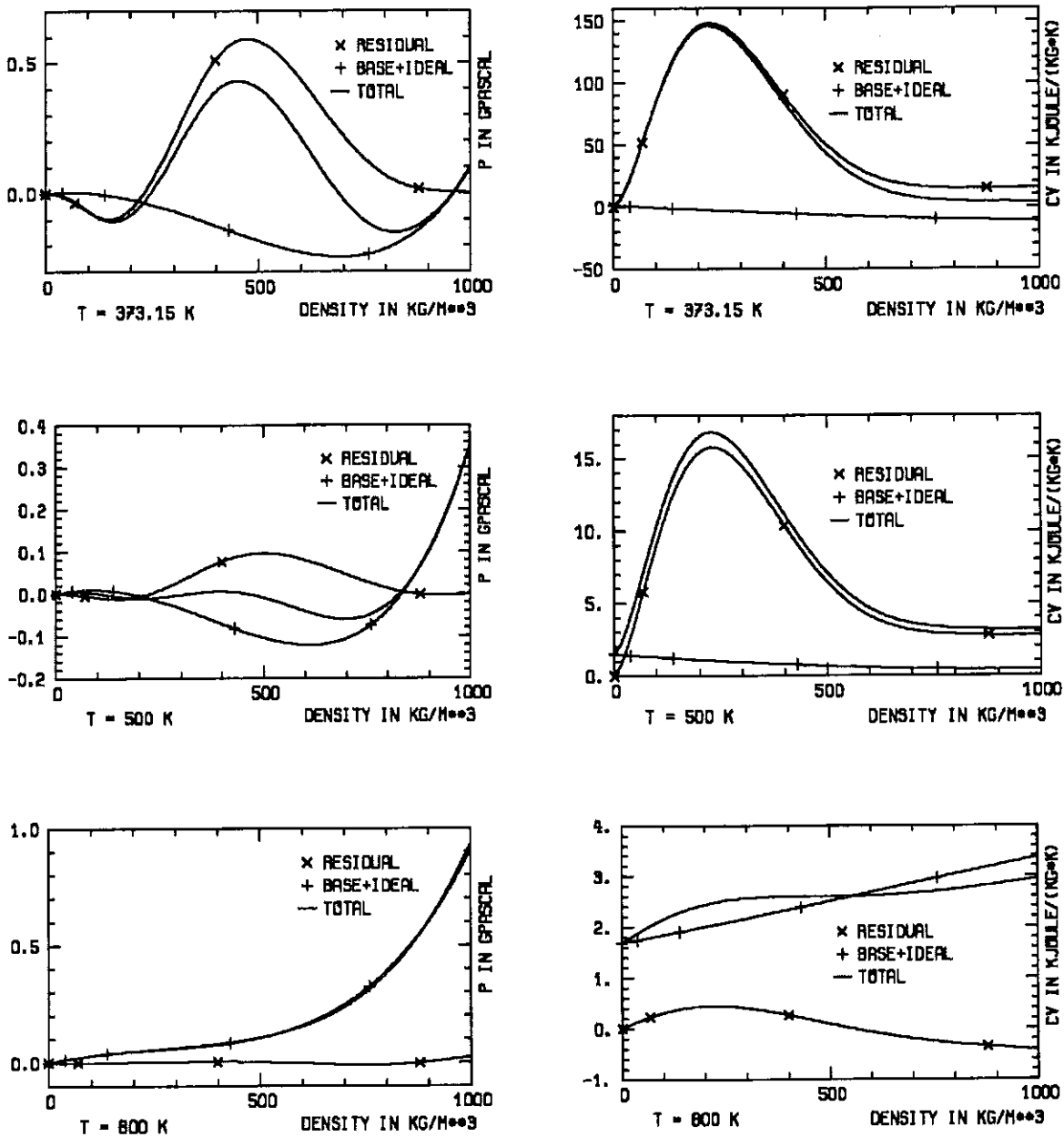


Figure 3 on page 14 displays the water pressure calculated by the H.G.K.-code as a surface of the temperature and the density.

### 2.5 The critical data

The H.G.K.-code uses the following values as critical data of the water :

$$T_c = 647.126 \text{ K} \quad , \quad P_c = 22.0549 \text{ MPa} \quad , \quad \rho_c = 322 \text{ kg/m}^3 \quad . \quad [2.11]$$

## 2.6 The validity domain

Haar and Gallagher, /13/ give - in accordance with the IAPS - the the following validity limits for the H.G.K.-code :

$$\begin{aligned} 273 \text{ K} &\leq T \leq 1273 \text{ K} \\ P &\leq 1500 \text{ MPa} \end{aligned} \quad [2.12]$$

**Note:** The density should be lower than  $1200 \text{ kg/m}^3$  especially at low temperatures to avoid getting negative values for the heat capacities and the pressure derivatives  $\partial P / \partial T$  !

### 3.0 Transport properties of the water.

For the description of the viscosity, of the thermal conductivity and of the surface tension AQUA uses the IAPS-82 equations, recommended on the VIII. International Conference on the Properties of Steam /9/ ( see also /10/, /11/ and /16/ ).

#### Reduced variables

In describing the transport properties of the water it is convenient to introduce the following reduced variables :

$$\begin{aligned}\tilde{r} &\equiv \frac{\rho}{\rho_k} \quad , \quad \rho_k = 317.763 \text{ kg/m}^3 \\ \tilde{t} &\equiv \frac{T}{T_k} \quad , \quad T_k = 647.27 \text{ K} \\ \tilde{p} &\equiv \frac{P}{P_k} \quad , \quad P_k = 22.115 \text{ MPa}\end{aligned}\quad [3.1]$$

along with the "ideal temperature" :

$$T_{id} \equiv \frac{P_k}{R_{gas} \cdot \rho_k} = 150.7965 \text{ K} \quad . \quad [3.2]$$

#### 3.1 Viscosity

With the reduced variables the viscosity can be described as

$$\eta(\rho, T) \equiv \eta_0(T) \cdot \exp \left[ \tilde{r} \cdot \sum_{j=0}^5 \sum_{m=0}^4 B_{jm} \cdot \left( \frac{1}{\tilde{t}} - 1 \right)^j \cdot (\tilde{r} - 1)^m \right] \quad [3.3]$$

with the temperature-dependent amplitude :

$$\eta_0(T) \equiv \frac{10^{-6} \cdot \sqrt{\tilde{t}}}{\sum_{l=0}^3 \frac{A_l}{\tilde{t}^l}} \quad . \quad [3.4]$$

The coefficients of the amplitude are

$$A_0 = 0.0181583 \quad , \quad A_1 = 0.0177624 \quad , \quad A_2 = 0.0105287 \quad , \quad A_3 = -0.0036744 \quad .$$

For the B-s in Eq. [3.3] see the table supplied.

Table 7. The coefficients B (j,m) of the viscosity					
j	m = 0	1	2	3	4
0	0.501938	0.235622	-0.274637	0.145831	-0.0270448
1	0.162888	0.789393	-0.743539	0.263129	-0.0273093
2	-0.130353	0.673665	-0.959456	0.347247	-0.0267758
3	0.907919	1.207552	-0.687343	0.213486	-0.0822904
4	-0.551119	0.0670665	-0.497089	0.100754	0.0602253
5	0.146543	-0.084337	0.195286	-0.032932	-0.0202595

### 3.2 Thermal conductivity

The thermal conductivity of all fluids reaches in the vicinity of the critical temperature anomalously high values ( s. e. g. /4/ ). The following figure of Sengers - presented in /16/ - shows the thermal conductivity in the water on isobars, nearing the critical one from the high-pressure side. The data points - compared here with the corresponding IAPS calculations - are measurements of Sirota and al., /7/.

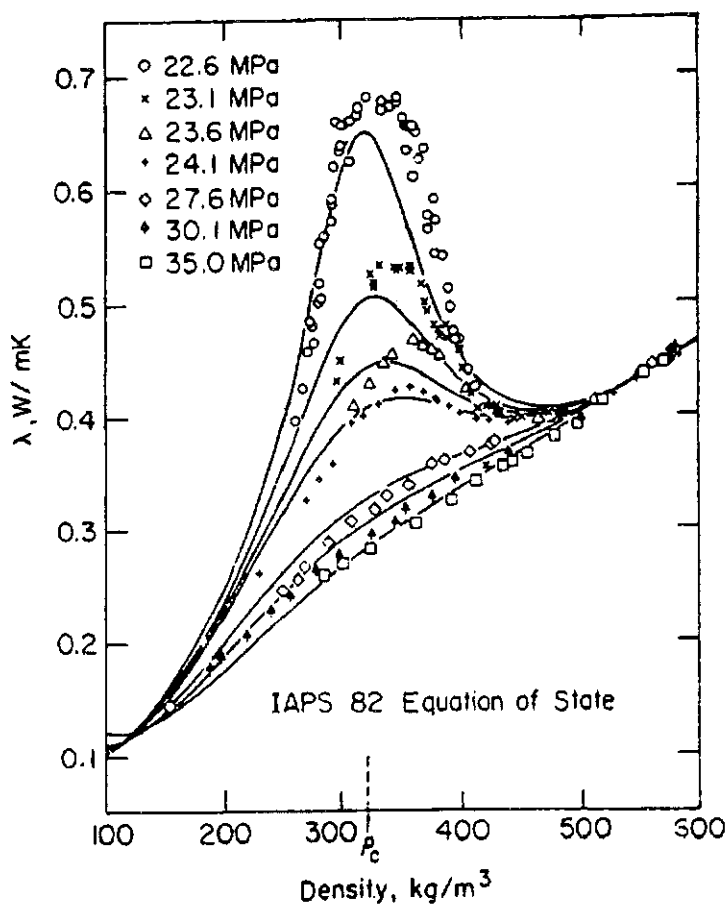


Figure 1. Thermal conductivity of the steam in the critical region as a function of density at constant pressures

Consequently Basu and Sengers proposed a two-termed description for the thermal conductivity. The first - regular - term of this description has the same form as the viscosity, the second term describes the near critical inflation of the thermal conductivity :

$$\lambda(\rho, T) \equiv \lambda_R(\rho, T) + \Delta\lambda(\rho, T) \quad . \quad [3.5]$$

The regular term is

$$\lambda_R(\rho, T) \equiv \lambda_0(T) \cdot \exp \left[ \tilde{r} \cdot \sum_{i=0}^4 \sum_{n=0}^5 b_{in} \cdot \left( \frac{1}{\tilde{t}} - 1 \right)^i \cdot (\tilde{r} - 1)^n \right] \quad [3.6]$$

with the amplitude

$$\lambda_0(T) \equiv \frac{\sqrt{\tilde{t}}}{\sum_{k=0}^3 \frac{a_k}{\tilde{t}^k}} \quad . \quad [3.7]$$

The coefficients of the amplitude are

$$a_0 = 2.02223 \quad , \quad a_1 = 14.11166 \quad , \quad a_2 = 5.25597 \quad , \quad a_3 = -2.01870 \quad .$$

The b-s of the Eq. [3.6] are tabulated as follows :

Table 8. The coefficients b (i,n) of the thermal conductivity					
n	i = 0	1	2	3	4
0	1.329304600	1.7018363	5.2246158	8.7127675	-1.8525999
1	-0.404524370	-2.2156845	-10.124111	-9.5000611	0.9340469
2	0.244094900	1.6511057	4.9874687	4.3786606	0.0
3	0.018660751	-0.76736002	-0.27297694	-0.91783782	0.0
4	-0.129610680	0.37283344	-0.43083393	0.0	0.0
5	0.044809953	-0.11203160	0.13333849	0.0	0.0

The nonregular term of the thermal conductivity is

$$\Delta\lambda(\rho, T) \equiv \frac{C}{\eta(\rho, T)} \cdot \left( \frac{\tilde{t}}{\tilde{r}} \cdot \frac{\partial \tilde{p}}{\partial \tilde{t}} \right)^2 \cdot (\chi_T)^\omega \cdot \sqrt{\tilde{r}} \cdot \exp \left[ -A \cdot (\tilde{t} - 1)^2 - B \cdot (\tilde{r} - 1)^4 \right] \quad . \quad [3.8]$$

Sengers call the property

$$\chi_T \equiv \tilde{r} \cdot \frac{\partial \tilde{r}}{\partial \tilde{p}} \quad [3.9]$$

"reduced isothermal compressibility". Actually,  $\chi_T$  is the factor of the nonregular term, which inflates the thermal conductivity, since



$$\frac{\partial \tilde{r}}{\partial \tilde{p}} \propto \frac{1}{\partial P / \partial \rho}$$

turns infinite at the critical point.

With the relation

$$\frac{\partial \tilde{p}}{\partial \tilde{r}} = \frac{P_k}{P_k} \cdot \frac{\partial P}{\partial \rho}$$

$\chi_T$  can be converted to

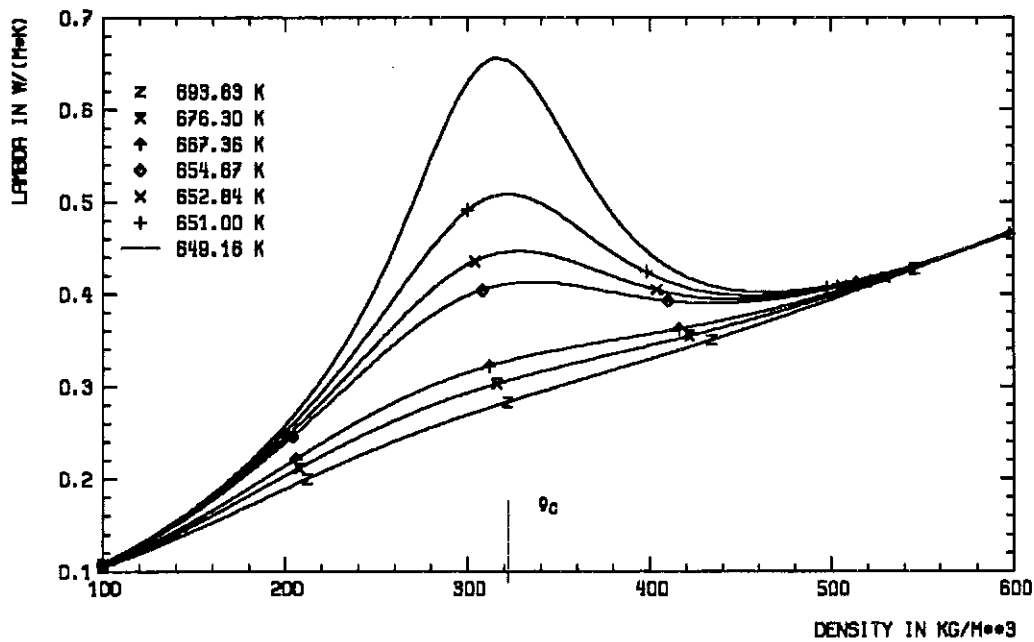
$$\chi_T = \frac{P_k}{\rho_k^2} \cdot \frac{\rho}{\partial P / \partial \rho} = \frac{\tilde{r}}{(T/T_{id}) \cdot P_\rho^0} \quad , \quad [3.10]$$

and the pressure derivative in Eq. [3.8] to

$$\frac{\tilde{t}}{\tilde{r}} \cdot \frac{\partial \tilde{p}}{\partial \tilde{t}} = \frac{T}{\tilde{r} \cdot P_k} \frac{\partial P}{\partial T} = \frac{T}{T_{id}} \cdot P_T^0 \quad . \quad [3.11]$$

The coefficients of the Eq. [3.8] are

$$A = 18.66 \quad , \quad B = 1.0 \quad , \quad C = 3.7711 \cdot 10^{-8} \text{ Pa} \cdot \text{sec} \cdot \text{W} / (\text{K} \cdot \text{m}) \quad , \quad \omega = 0.4678 \quad .$$



**Figure 2. Thermal conductivity of the water on supercritical isotherms**

Figure 2 shows the thermal conductivity on some supercritical isotherms. The isothermal pressures at the critical density are :

T / K	693.63	676.36	667.36	654.67	652.84	651.00	649.16
P322 / MPa	35.0	30.1	27.6	24.1	23.6	23.1	22.6

### 3.3 Surface tension

According to the IAPS Working Group III, 1975, /9/ the surface tension of the water can be described as

$$\sigma(T) = B \cdot \left( \frac{T_q - T}{T_q} \right)^\mu \cdot \left( 1 + b \frac{T_q - T}{T_c} \right) \quad [3.12]$$

with  $T_q = 647.15$  K,  $B = 0.2358$  N/M,  $b = -0.625$  and  $\mu = 1.256$  .

**Note:** AQUA uses as well in Eq. [3.1] as in Eq. [3.12] as critical temperature the now usually accepted value  $T_c = 647.126$  K. An attempt, to replace in Eq. [3.1]  $\rho_k = 317.763$  kg/m<sup>3</sup> with the correct value  $\rho_c = 322$ . kg/m<sup>3</sup> resulted in a drastic decrease of the thermal conductivity of the water at high densities and had been given up.

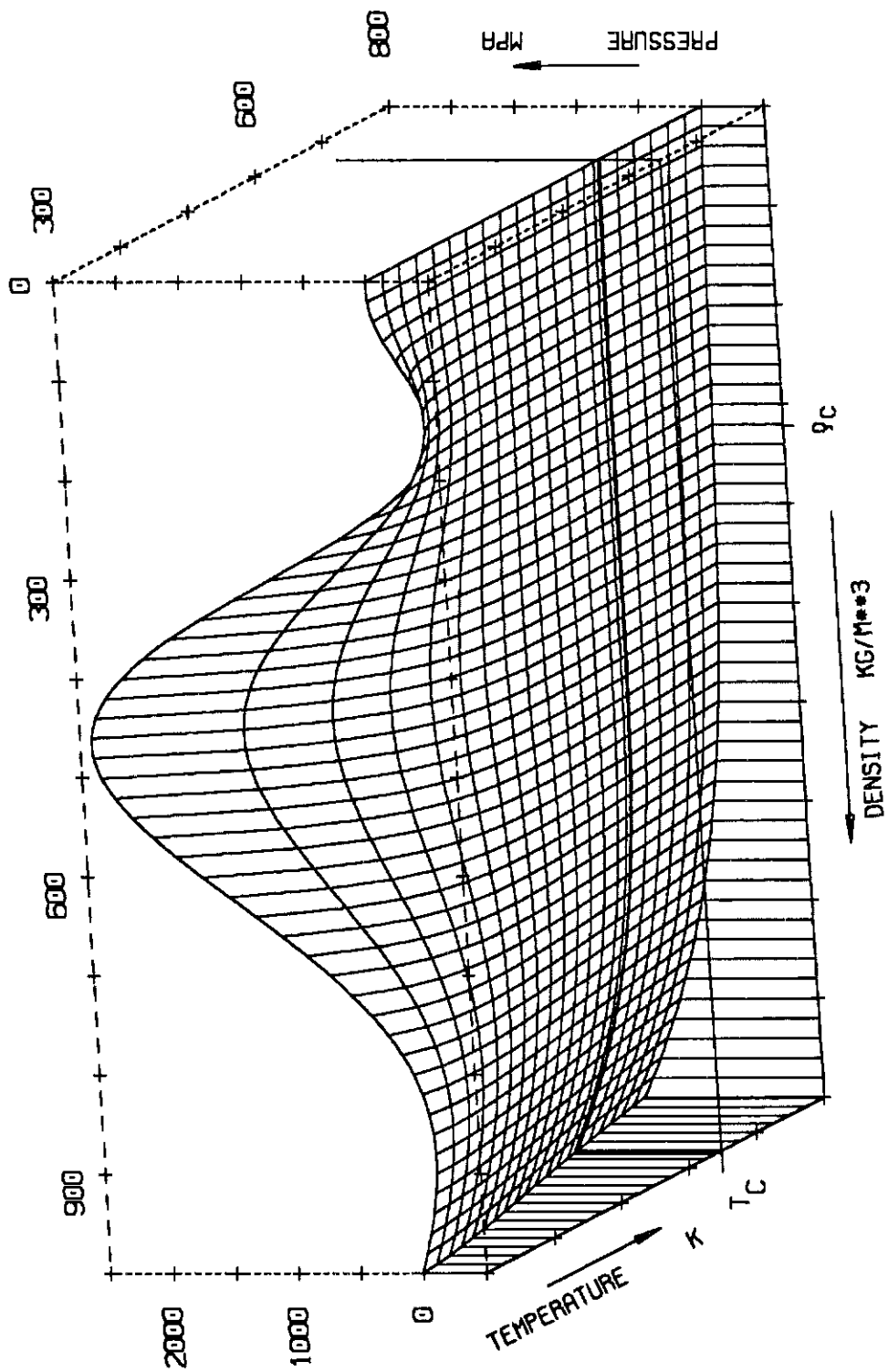


Figure 3. Pressure surface of the water. H.G.K.-code

## 4.0 Thermal properties in the two phase region

### 4.1 Thermal states of the subcritical water

At temperatures above the melting and below the critical Point water can be present not only as liquid or vapor but also as a mixture of both of them. This part of the subregion, the two phase region, or the "saturation dome" is bounded by the "saturation line", which is a collection of the saturated liquid and the saturated vapor states. Outside the saturation dome the water is either a compressed liquid or an expanded vapor.

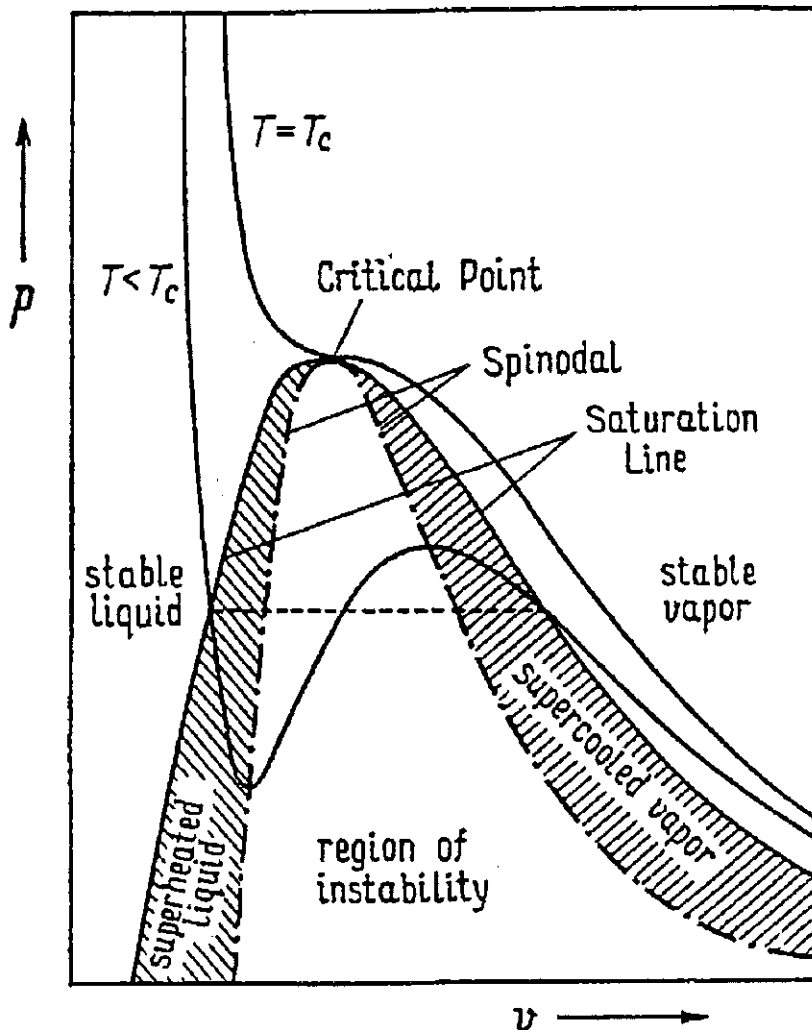


Figure 4. Sketch of P,V diagram with saturation and spinodal lines.  $V = 1 / \rho$

Normally an isothermal expansion of the water beyond the saturated density ( $\rho = \rho_L$ ) results in vaporizing the water. A part of the water - as large, as needed to restore the vapor pressure,  $P^*$  (dashed line on the sketch) - turns into vapor, to suffice the equilibrium conditions :

$$P^*(T) = P(\rho_L, T) = P(\rho_V, T) \quad , \quad [4.1]$$

$$G(\rho_L, T) = G(\rho_V, T) \quad . \quad [4.2]$$

Under particular circumstances ( rapid expansion, very clean water ) one can expand the saturated water to a distinct degree, without turning it into vapor and reach a state with a pressure below the  $P^*$  ( for a detailed description s e. g. /15/ ). This expanded water is known as superheated ( liquid ) water. The superheated water is metastable : it tends to turn into a mixture of saturated liquid and vapor with the equilibrium pressure  $P^*$ . The theoretical limit of the expansion of the liquid water is given by the highest density value on the isotherm, for which it holds :

$$\frac{\partial P}{\partial \rho}(T, \rho) = 0 \quad [4.3]$$

and is known as the ( liquid ) spinodal density,  $\rho_{SL}$ .

Similarly normal compression of the saturated vapor results in turning a part of it into liquid water. By carefully chosen conditions the vapor also can be compressed in some extent above the saturated vapor density without developing water droplets in it and the vapor can reach an overheated state with a pressure higher as  $P^*$ . Again, the limiting density is the smallest density, which fulfills Eq. [4.3] and is called the ( vapor ) spinodal density,  $\rho_{SV}$ .

On the remaining part of the isotherm - lying between the spinodal densities ( region of instability on Figure 4 ) - the water can occur only as a mixture of saturated liquid and vapor.

#### 4.2 Water properties in the region of instability

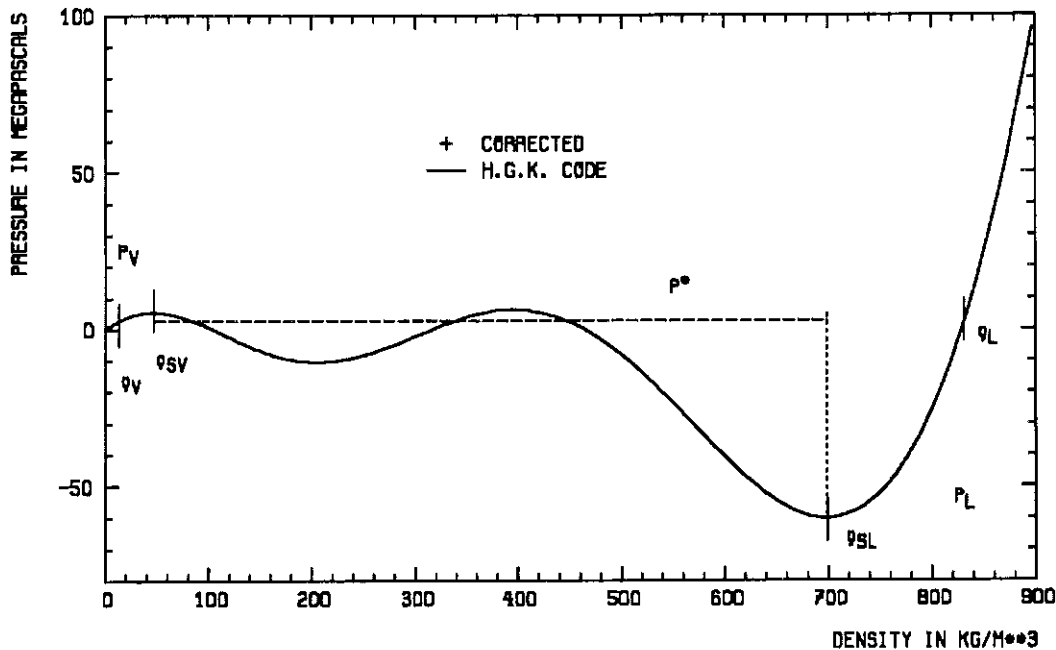


Figure 5. Pressure of the water in the two phase region. T = 500 K

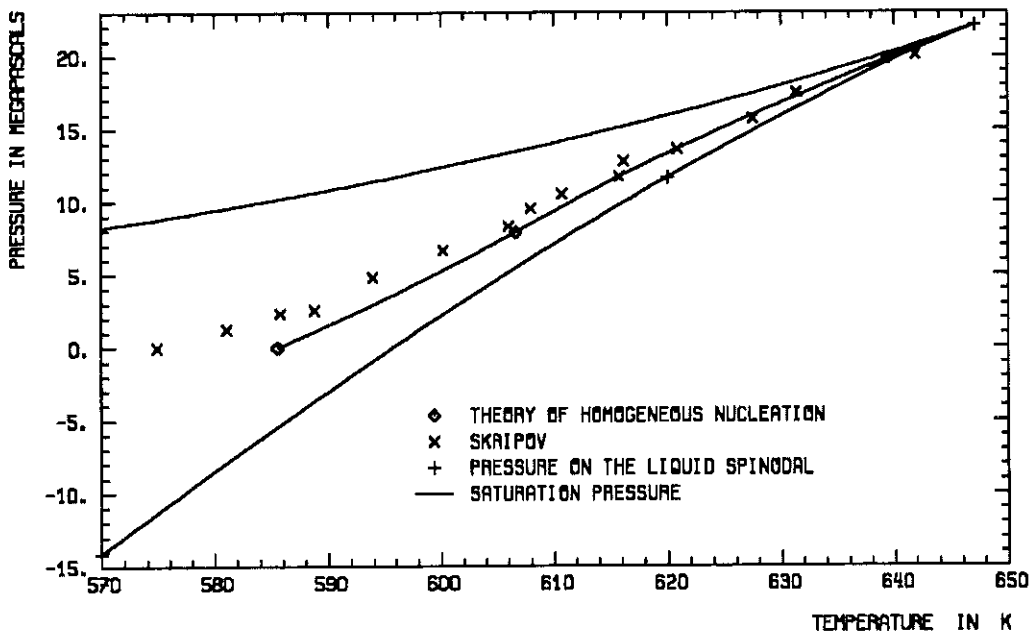
Figure 5 shows the water pressure along a typical isotherm as calculated by the codes H.G.K. resp. AQUA.

The code AQUA follows the pressure description of the H.G.K.-code from zero density till the density of the vapor-spinodal and from the liquid spinodal onward. But in the instable region, between the spinodal points AQUA describes the water as a two phase mixture :

$$\begin{aligned}
 P(\rho, T) &= P^x(T) \\
 \frac{\partial P}{\partial T}(\rho, T) &= \frac{dP^x(T)}{dT} \\
 \frac{\partial P}{\partial \rho}(\rho, T) &= 0
 \end{aligned}
 \tag{4.4}$$

( s. dashed line on Figure 5 ). A 3D-diagram of the pressure of the water - as AQUA calculates it - is depicted on Figure 14 on page 24.

**Note:** AQUA still idealizes the progress of the water pressure along the isotherm. In reality no liquid can be stretched to reach  $\rho_{sl}$  and no vapor can be compressed to have the density  $\rho_{sv}$ . The limiting values of the superheated pressure in the water can be seen on Figure 6. The measured points are from /3/.



**Figure 6. Water pressures in the superheated liquid state**

The Figure 7 on page 18 shows the free enthalpy of the water along the same isotherm. Here again outside of the region of instability AQUA and H.G.K. calculate identical values, between the spinodal points AQUA sets  $G$  equal to the common free enthalpy value :

$$G(\rho, T) = G_L(T) = G_V(T) .
 \tag{4.5}$$

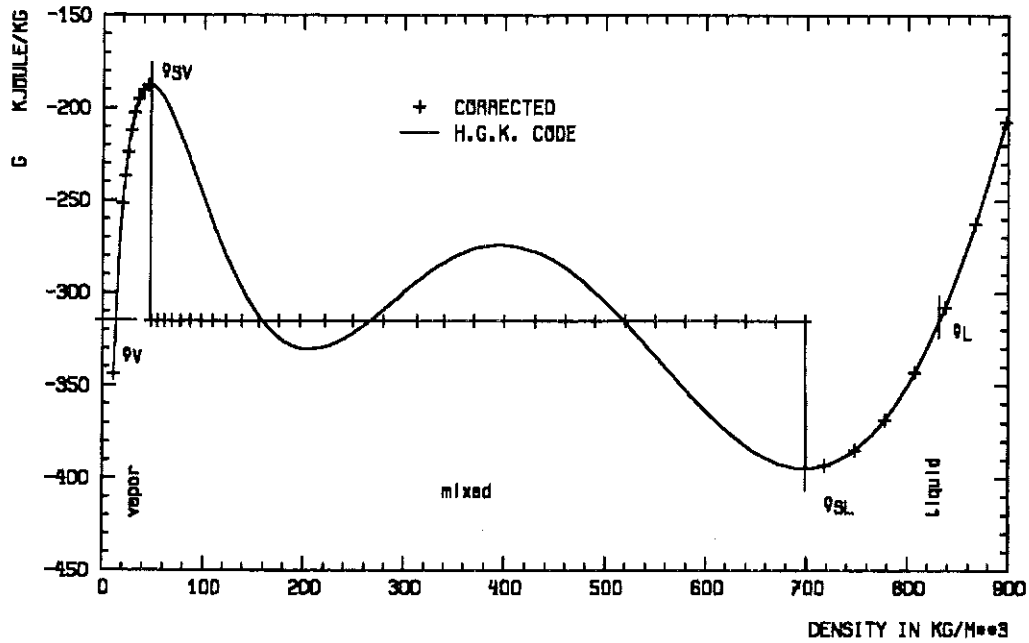


Figure 7. Free enthalpy of the water in the two phase region. T = 500 K

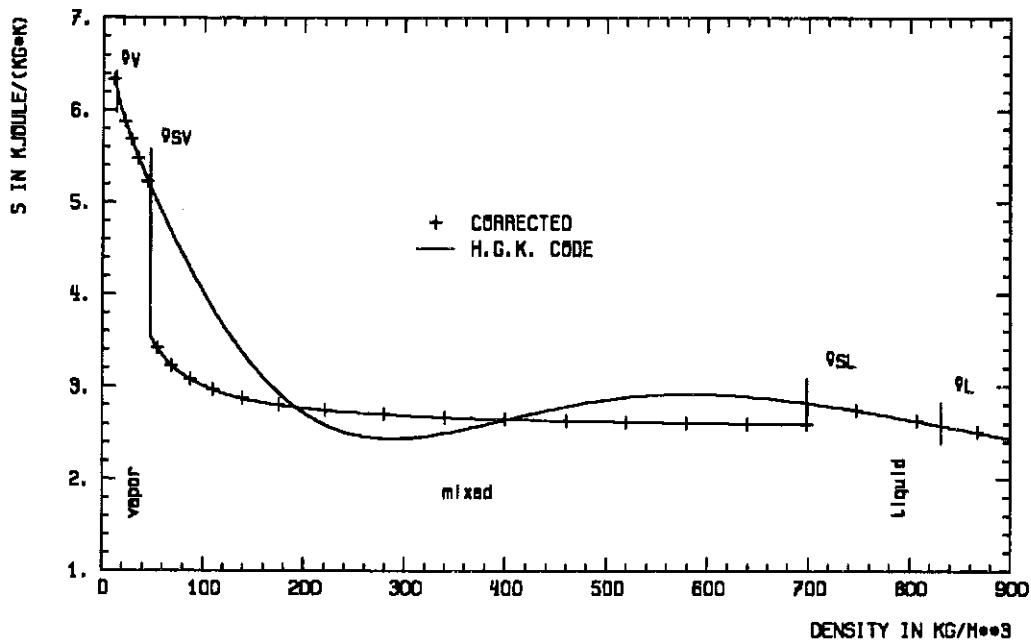


Figure 8. Entropy of the water in the two phase region. T = 500 K

As regards to the entropy, one can calculate this property in the mixed state using the density derivative [A.22]

$$\frac{\partial S}{\partial \rho} = -\frac{1}{\rho^2} \frac{\partial P}{\partial T} = -\frac{1}{\rho^2} \frac{dP^x}{dT} \quad [4.6]$$

which gives in the region of instability the following density dependence :

$$S(\rho, T) = S_L(T) + [V - V_L(T)] \cdot \frac{dP^x(T)}{dT} \quad . \quad [4.7]$$

The free energy in the region of instability ( s. Figure 9 ) AQUA calculates from the pressure and free enthalpy via

$$A = G - P/\rho \quad . \quad [4.8]$$

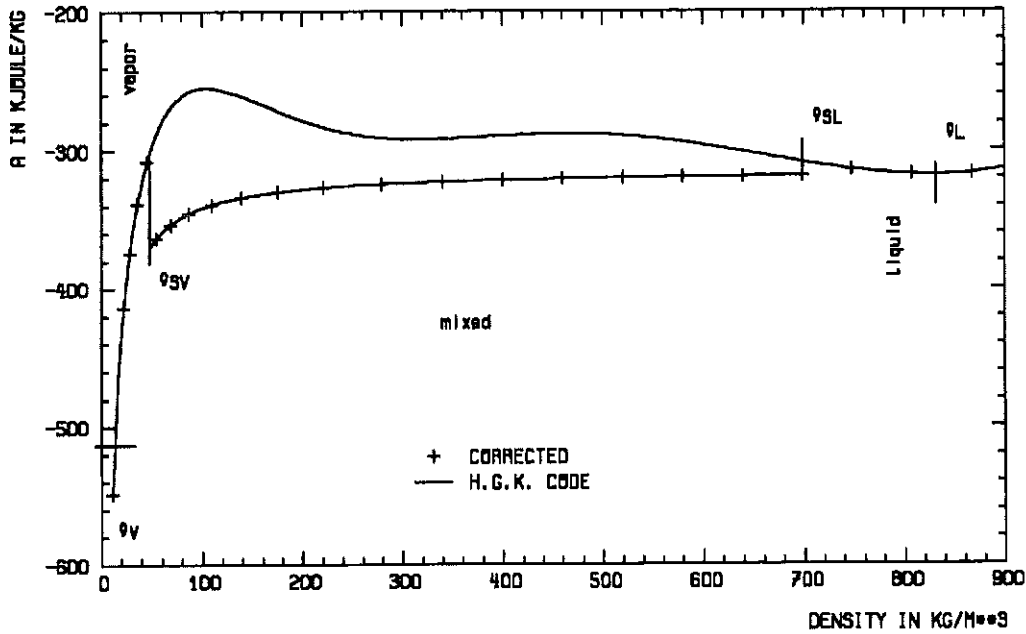


Figure 9. Free energy of the water in the two phase region. T = 500 K

For the enthalpy and the energy AQUA uses the equations

$$H = G + S \cdot T \quad , \quad [4.9]$$

respectively

$$U = H - P/\rho \quad . \quad [4.10]$$

The density derivative of the energy is given by

$$\frac{\partial U}{\partial \rho} = \frac{1}{\rho^2} \left( P - T \cdot \frac{\partial P}{\partial T} \right) = \frac{1}{\rho^2} \left( P^x - T \cdot \frac{dP^x}{dT} \right) \quad . \quad [4.11]$$

In the two phase region the derivatives  $\partial H/\partial P$  and  $C_p$  are not defined altogether.

For the sonic velocity in the region of instability the first version of the Eq. [A.23] can be used :

$$\frac{\partial P}{\partial \rho} \Big|_s = \frac{\partial P}{\partial \rho} + \frac{T}{C_V} \left( \frac{\partial P/\partial T}{\rho} \right)^2 = \frac{T}{C_V} \left( \frac{1}{\rho} \frac{dP^x}{dT} \right)^2 \quad [4.12]$$



with

$$v_s = \sqrt{\left. \frac{\partial P}{\partial \rho} \right|_s} = \frac{1}{\rho} \frac{dP^x}{dT} \sqrt{\frac{T}{C_V}} \quad [4.13]$$

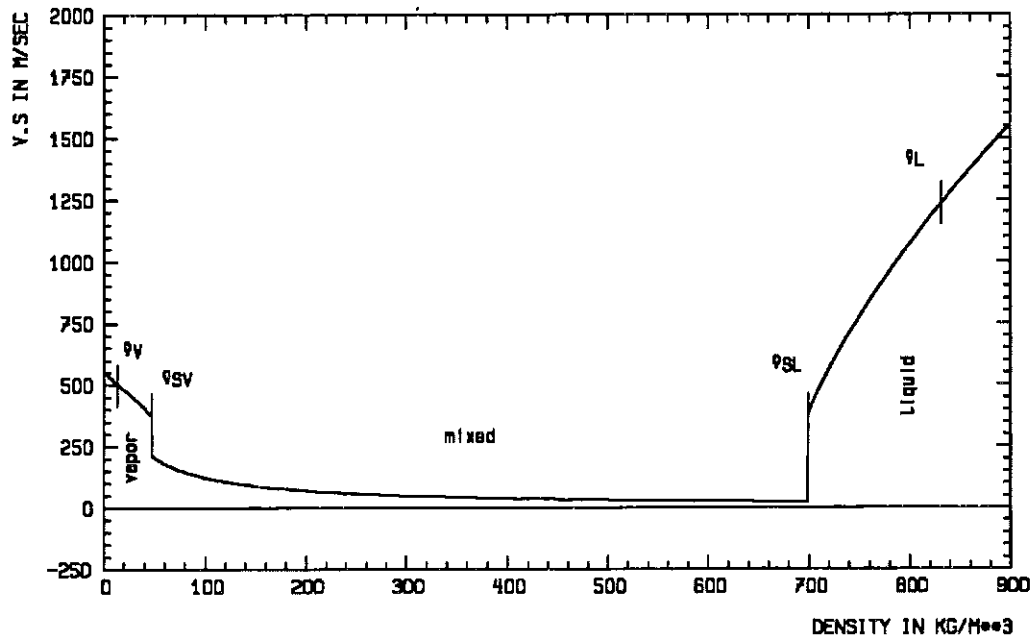


Figure 10. Sonic velocity of the water in the two phase region.  $T = 500 \text{ K}$

Not so easy is to calculate the constant volume heat capacity in the region of instability. The density derivative in the two phase region

$$\frac{\partial C_V}{\partial \rho} = \frac{\partial(T \cdot \partial S / \partial T)}{\partial \rho} = -\frac{T}{\rho^2} \frac{\partial^2 P}{\partial T^2} = -\frac{T}{\rho^2} \frac{d^2 P^x}{dT^2} \quad [4.14]$$

supplies here the difference term :

$$\Delta C_V(\rho, \rho_L) = T \frac{d^2 P^x}{dT^2} (V - V_L) \quad [4.15]$$

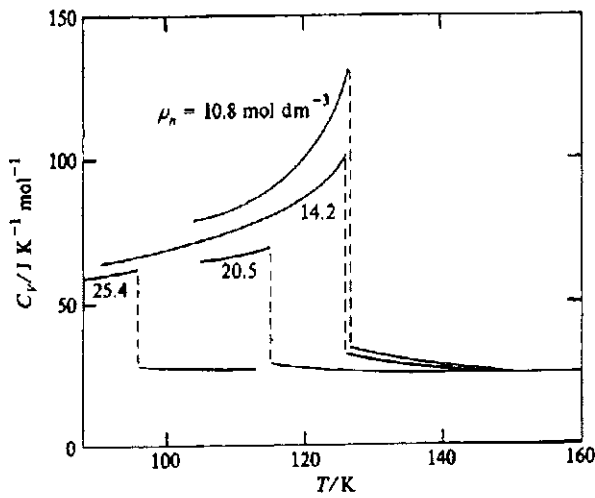
Besides of the difference term, one has also to account for the  $C_V$  - jumps, at the places, where the isotherm crosses the saturation line resp. where the liquid or the vapor turns into a vapor-liquid mixture ( s. 11/, 12/, 15/, 18/ ). Since  $\partial P / \partial T$  shows here a discontinuity, so does also  $C_V$ . At the saturated liquid this discontinuity amounts to

$$C_V(\rho_L - 0) - C_V(\rho_L) = \frac{r_L}{\rho_L} \left[ \frac{dP^x}{dT} - \frac{\partial P}{\partial T}(\rho_L) \right] \quad [4.16]$$

$r_L$  in the above eq. is the logarithmic derivative of the saturation line :

$$r_L \equiv \frac{T}{\rho_L} \cdot \frac{d\rho_L}{dT} \quad [4.17]$$

The following figure displays such  $C_V$ -jumps, measured in nitrogen along different isochores.



One- and two phase  $C_V$  results for nitrogen at several amount-of-substance densities.

after L. A. Weber, /6/.

The Eqs. [4.15], [4.16] together give the following heat capacity function in the region of instability :

$$C_V(\rho) = C_V(\rho_L) + \frac{r_L}{\rho_L} \left[ \frac{dP^x}{dT} - \frac{\partial P}{\partial T}(\rho_L) \right] + \left( \frac{1}{\rho} - \frac{1}{\rho_L} \right) T \frac{d^2 P^x}{dT^2} \quad . \quad [4.18]$$

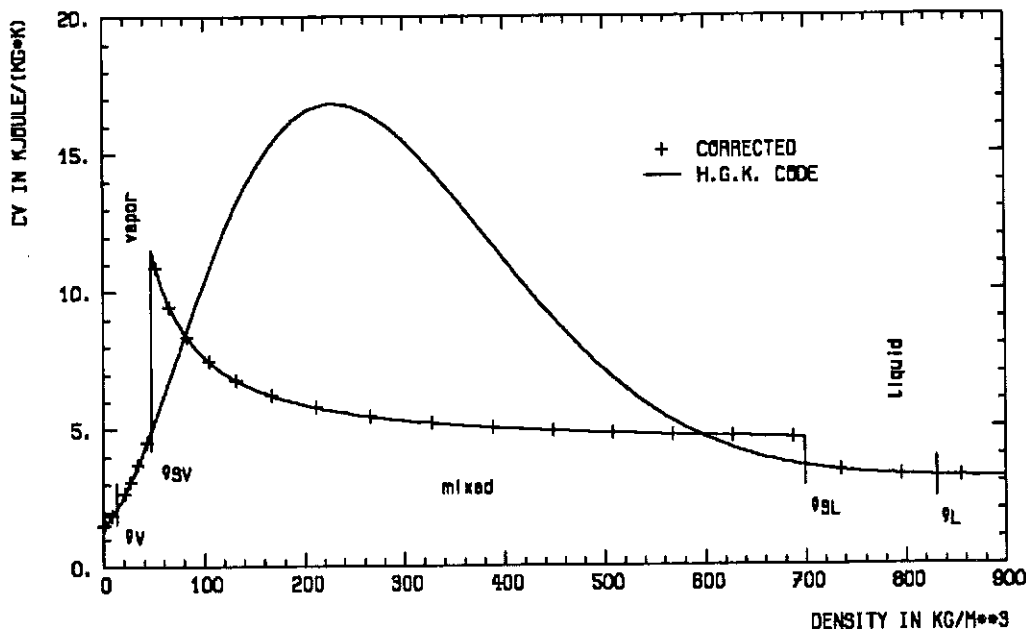


Figure 11. Heat capacity -  $C_V$  - of the water in the two phase region.  $T = 500 \text{ K}$

For a  $(T, \rho)$  - surface of the heat capacity of the water - calculated by AQUA - see Figure 15 on page 25 resp. Figure 16 on page 26.

In calculating the thermal conductivity in the two phase region AQUA uses in the different subregions different algorithms. In the superheated liquid and subcooled vapor the function - described in 3.0, "Transport properties of the water." , Eq. [3.5] ff. - is used as well as outside the saturation dome.

In the region of instability AQUA calculates the thermal conductivity using the following approximation : the liquid part of the water in this subregion is distributed homogeneously - as a multitude of tiny droplets - in the vapor part of the water. This mixture is furthermore static in its behaviour, so the convective part of the heat transfer can be neglected. The approximation above allows to calculate in the region of instability the thermal conductivity from the thermal conductivities of the saturated states -  $\lambda_L$  ,  $\lambda_V$  - as follows :

$$\frac{1}{\lambda} = \frac{1-x}{\lambda_L} + \frac{x}{\lambda_V + x^2 \cdot (\lambda_L - \lambda_V)} \quad [4.19]$$

( s. /5/, Eq. 71 ). x in the eq. above is

$$x = \sqrt[3]{\frac{\rho - \rho_V}{\rho_L - \rho_V}} \quad [4.20]$$

Figure 12 shows the thermal conductivity of the water along the 500 K - isotherm. The spikes in the vicinity of the spinodal points are a consequence of the description [3.8], using  $\partial P / \partial \rho$  - which vanishes at the spinodals - in the denominator.

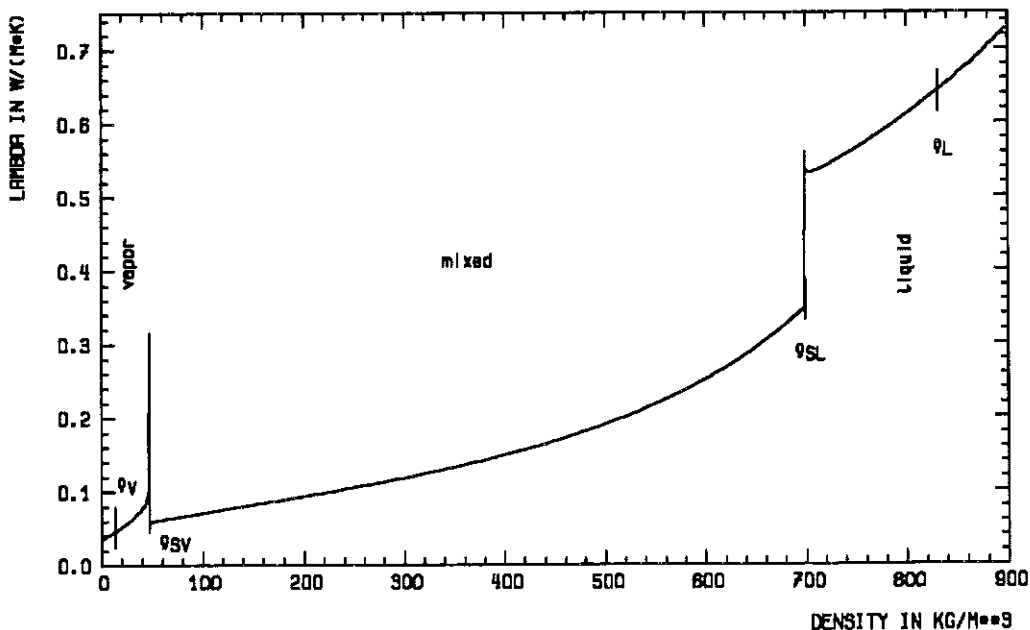


Figure 12. Thermal conductivity of the water in the two phase region. T = 500 K

As well as the thermal conductivity, the viscosity is also a transport process, the transport of the momentum. So AQUA calculates the viscosity by a procedure, very similar to the one, used for the calculation of the thermal conductivity.

At the saturation densities the  $\eta$  function Eq. [3.3]- [3.4] - will be continued until the spinodals are reached. In the region of instability AQUA uses the same fluid-model, as in the case of the thermal conductivity to calculate the viscosity from the saturated viscosities  $\eta_L$  ,  $\eta_V$ . The resulting equation is :

$$\frac{1}{\eta} = \frac{1-x}{\eta_L} + \frac{x}{\eta_V + x^2 \cdot (\eta_L - \eta_V)} \quad [4.21]$$

Figure 13 shows the water-viscosity as calculated by AQUA along the 500 K - isotherm.

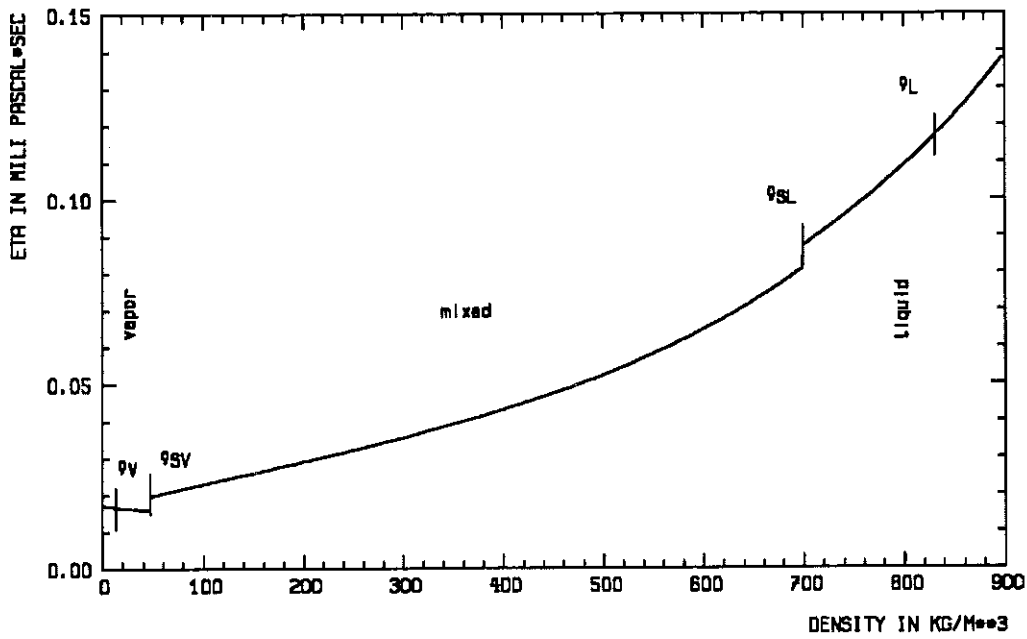


Figure 13. Viscosity of the water in the two phase region. T = 500 K

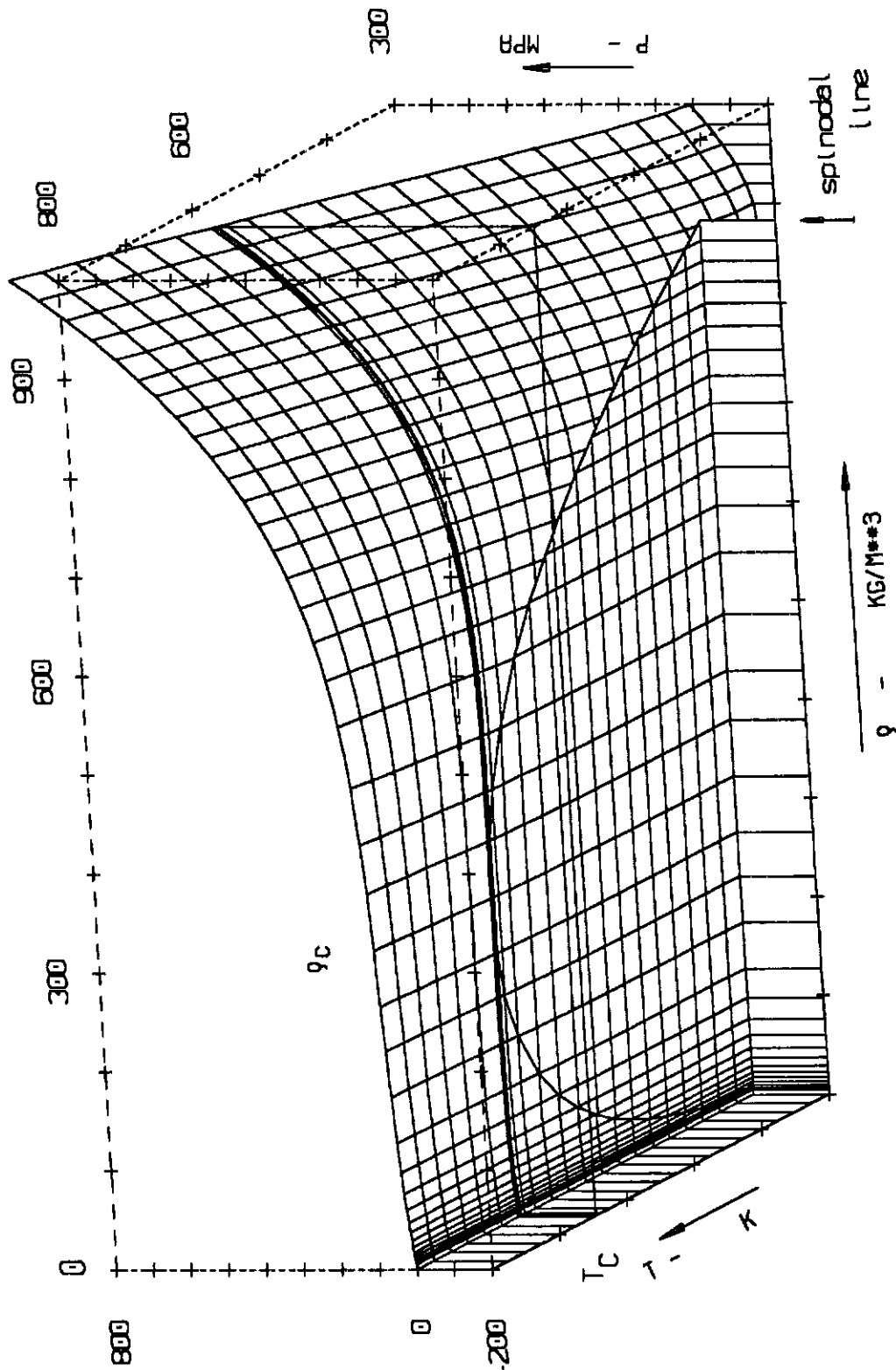


Figure 14. Pressure surface of the water. AQUA

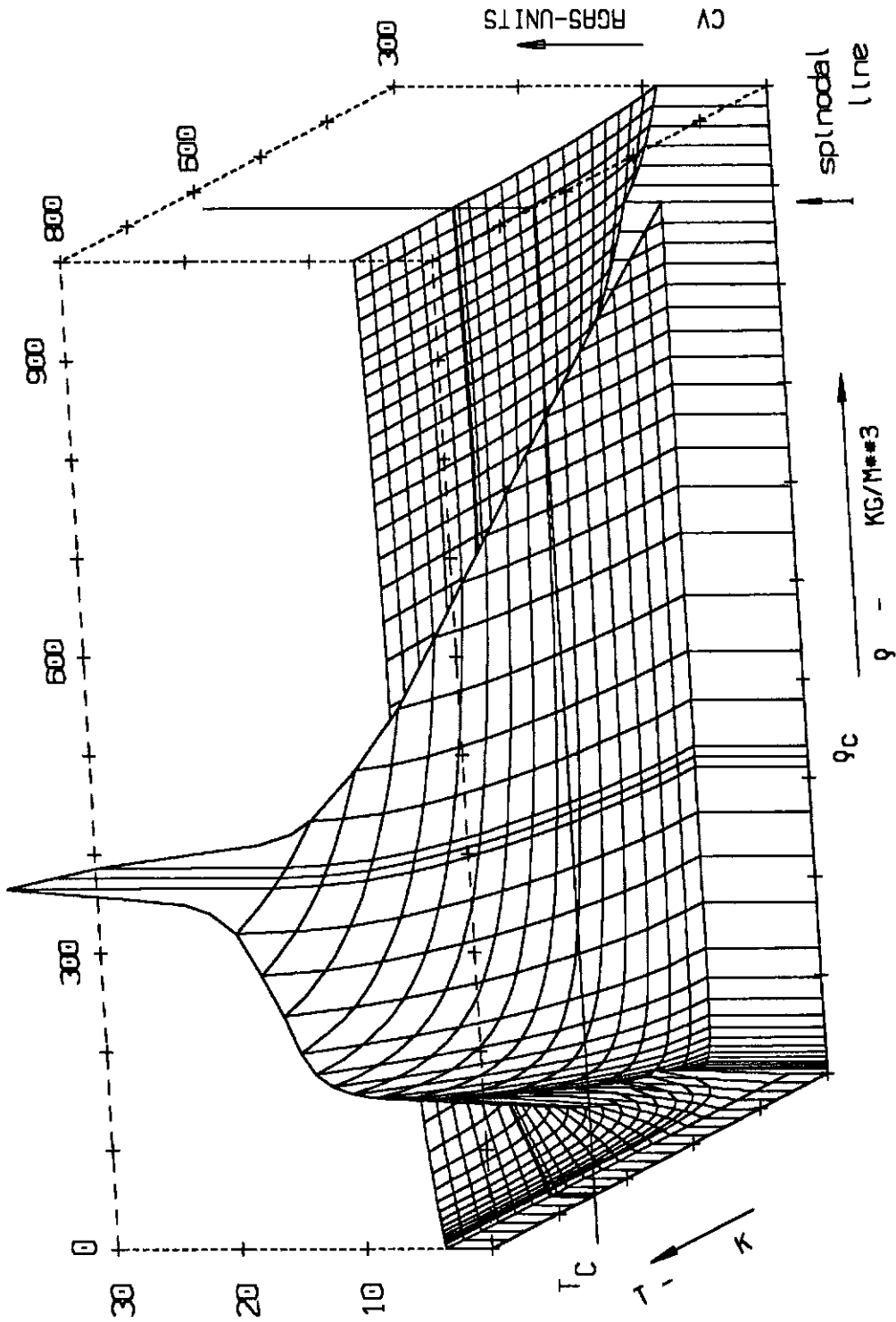


Figure 15. Heat capacity surface of the water. AQUA. View from the cold side.

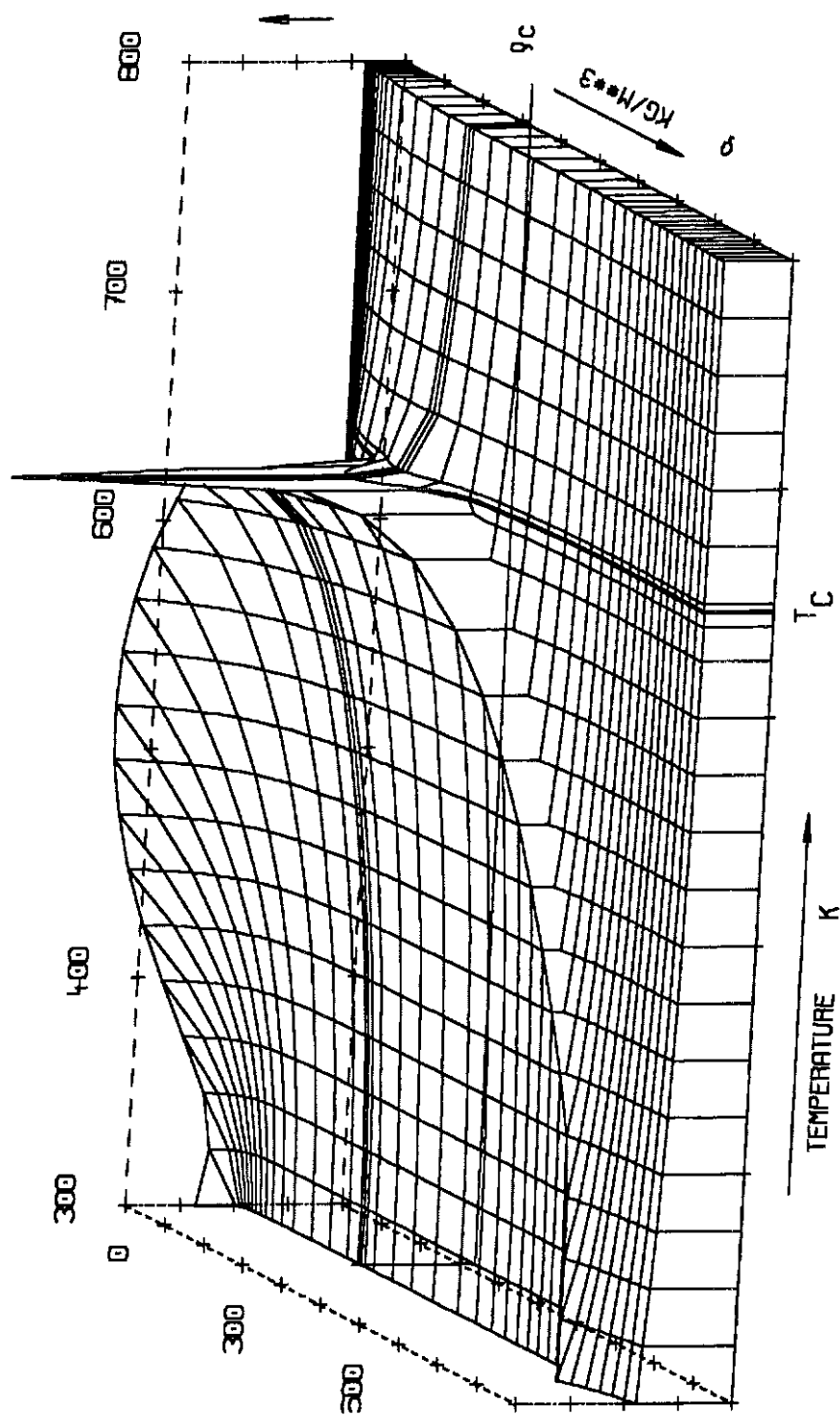


Figure 16. Heat capacity surface of the water. AQUA. View from the liquid side.

## 5.0 Describing the water properties in the saturated states

In order to be able to calculate the water properties in the region of instability AQUA needs a series of thermal properties of the water in the saturated states ( s. e.g. Eq. [4.18] ). To speed up the calculations AQUA doesn't calculate these properties (  $P^x$ ,  $\rho$ ,  $\partial P/\partial \rho$ ,  $\partial P/\partial T$ ,  $C_v$ ,  $\lambda$ , etc ) directly, but use pre-defined temperature-functions of the corresponding properties. In constructing these temperature-functions great care was taken

to have negligible deviations to the H.G.K values and

to show the correct behaviour in the vicinity of the critical point.

The remaining part of this chapter describes the temperature-descriptions of these properties of the water in the saturated states.

### 5.1 Describing the saturation line of the water

#### The saturation pressure

Haar, Gallagher and Kell describe the vapor pressure of the water ( at temperatures, exceeding 314 K ) as a function of the scaled temperature

$$\tau \equiv 1 - \frac{T}{T_c} \quad [5.1]$$

rsp. of the square root of it

$$w \equiv \sqrt{\tau} \quad [5.2]$$

as follows :

$$\ln P^x = \ln P_c + \frac{\tau}{1 - \tau} \sum_{j=1}^8 A_j \cdot w^{j-1} . \quad [5.3]$$

In the above Eq. H.G.K. use critical data

$$T_c = 647.25 \text{ K} , \quad P_c = 22.093 \text{ MPa} \quad [5.4]$$

which differ markedly from their regularly used values Eq. [2.11] .

The polynomial coefficients,  $A_j$  are :

1	-7.8889166	2	2.5514255	3	-6.716169
4	33.239495	5	-105.38479	6	174.35319
7	-148.39348	8	48.631602		

AQUA uses, instead of the Eq. [5.3] a description, proposed by Wagner and Saul at the conference in Moscow, /14/. The W&S-equation needs only 6 coefficients and yet repre-



sent the measured vapor pressure data within their experimental uncertainty ( see Figure 17 ).

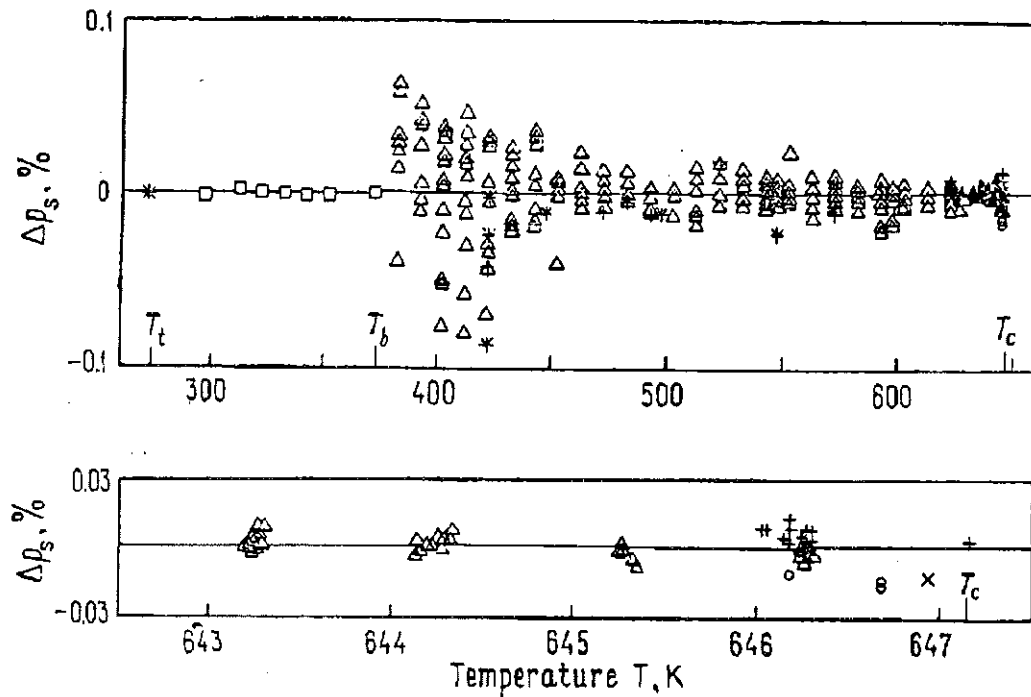


Fig. 1. Relative deviations  $\Delta p_s = (p_{s,exp} - p_{s,calc}) / p_{s,exp}$  of the experimental vapour pressures from values calculated with Eq. (5).  $\Delta$  Osborne et al. [14],  $+$  Rivkin et al. [16],  $\square$  Stimson [13],  $*$  Guildner et al. [12],  $\circ$  Hanafusa et al. [17],  $\times$  Kawai et al. [18],  $\Psi$  Kell et al. [19]

#### Figure 17. Vapor pressure formula of Wagner and Saul

The W&S-equation has the same form as the H.G.K. one, Eq. [5.3] ,

$$\ln P^x = \ln P_c + \frac{\tau}{1 - \tau} \sum_{j=1}^{13} N_j \cdot w^{j-1} \quad [5.5]$$

only the critical data and the coefficients differ. They use

$$T_c = 647.14 \text{ K} \quad , \quad P_c = 22.064 \text{ MPa} \quad [5.6]$$

as critical data and

$$\begin{aligned} N_1 &= -7.85823 & , & & N_2 &= 1.83991 & , & & N_5 &= -11.7811 & , \\ N_6 &= 22.6705 & , & & N_7 &= -15.9393 & , & & N_{13} &= 1.77516 & , \end{aligned} \quad [5.7]$$

as polynomial-coefficients. In calculating the vapor pressure AQUA uses here also the critical data of Haar, Gallagher and Kell ( Eq. [2.11] ) instead of the data [5.6] .

As well as in the H.G.K.-code,  $P^x$  of Eq. [5.5] is for AQUA only an approximated pressure. The true vapor pressure is calculated from this first value by correcting the saturated densities till the Eq.s [4.1 - 4.2] are satisfied.

## The saturation line

The saturated densities are needed not only as limiting borders, but also as properties of their own value. The calculation of various thermal properties in the region of instability requires mostly the saturated densities and in some cases their temperature derivatives too ( s. e.g. Eq. [4.18] ). Consequently one needs not only smooth functions for the saturated densities, but functions with smooth temperature derivatives.

The H.G.K.-code calculates - below 646.3 K -  $P^x$ ,  $\rho_L$  and  $\rho_V$  by searching for states  $(T, \rho_L)$  resp.  $(T, \rho_V)$  with equal free enthalpy values ( Eq.s [4.1 - 4.2] ). The corresponding equal pressures define then  $P^x(T)$ . Above 646.3 K Haar, Gallagher, and Kell use "scaled" equations for the saturated densities :

$$\begin{aligned} \frac{\rho_L(T)}{\rho_c} &= 1 + 2.04077 \cdot \tau^\beta \\ \frac{\rho_V(T)}{\rho_c} &= 1 - 2.04077 \cdot \tau^\beta \end{aligned} \quad [5.8]$$

with the critical exponent  $\beta = 0.325$ , without correcting the densities for common free enthalpy.

AQUA describes - as well as the vapor pressure - the saturation line also with the  $T_c$ -related variables  $\tau$  and  $w$ , Eq.s [5.1] , [5.2] .

**Below 620 K** the density of the saturated liquid is described with a polynomial of  $w$ ,

$$\rho_L(T) = \rho_c \cdot \sum_{j=1}^{12} A_j \cdot w^{j-1} \quad [5.9]$$

1	6.072746418	2	-149.9456089	3	2121.007381
4	-17164.83301	5	90176.84684	6	-323045.6094
7	805443.5472	8	-1398515.305	9	1658696.365
10	-1281195.416	11	580741.4278	12	-117181.9097

In the case of the saturated vapor instead of the density, the factor of reality is described by a polynomial of  $w$  :

$$\rho_V(T) = \frac{P^x(T)}{R_{gas} \cdot T \cdot Z_V(T)} \quad , \quad Z_V(T) = \sum_{j=1}^{12} Z_j \cdot w^{j-1} \quad [5.10]$$

1	-1.125999398	2	42.15284538	3	-538.0831003
4	4150.173161	5	-20809.47751	6	71295.07069
7	-170444.6088	8	284498.9747	9	-325135.7165
10	242459.309	11	-106255.203	12	20747.6267

As well for the liquid, as for the vapor the polynomial-coefficients were fitted to properties, calculated by the H.G.K-code. The vapor pressure in Eq. [5.10] was calculated by the W&S formula.

**Above 620 K** AQUA uses practically the same scaled description, as the H.G.K code ( Eq. [5.8] ) only fitting tails had been added :

$$\begin{aligned} \rho_L(T) &= \rho_c \cdot [1 + 2.03 \cdot \tau^\beta + \tau (U_1 + \tau (U_2 + \tau^2 U_4))] \quad , \\ \rho_V(T) &= \rho_c \cdot [1 - 2.03 \cdot \tau^\beta + \tau (W_1 + \tau (W_2 + \tau^2 W_4))] \quad . \end{aligned} \quad [5.11]$$

The coefficients of the fitting tails

$$\begin{aligned} U_1 &= 3.075668853 \quad , \quad U_2 = -19.58060683 \quad , \quad U_4 = 1383.086689 \\ W_1 &= 1.451962241 \quad , \quad W_2 = -5.041325559 \quad , \quad W_4 = 611.1810749 \end{aligned}$$

had ben calculated by demanding a double-smooth connection

$$f_-(T_s) = f_+(T_s) \quad , \quad f'_-(T_s) = f'_+(T_s) \quad , \quad f''_-(T_s) = f''_+(T_s) \quad [5.12]$$

for them to the low-T formulas at  $T_s = 620$  K. ( see Appendix B, "Derivatives of some saturated properties" on page 71 ).

The saturated densities, calculated by the Eq.s [5.9 - 5.11] are again only approximated ones. AQUA corrects them - below 646.0 K - to fulfil the Eq.s [4.1 - 4.2] ( s. Appendix C, "Adjusting the saturated states" on page 75 ).

## 5.2 Describing the spinodals

### Calculating the spinodals of the water.

The spinodals of the water are essential for the code AQUA : the spinodal densities are needed as domain borders for the calculations in the saturation dome, the spinodal pressures are needed as pressure bounds in converting the pressures to densities.

In order to calculate the densities of the spinodals one has to find the nearest roots of the Eq. [4.3] on the isotherms in the vicinity of the saturated densities. In the code-package AQUA there is a program HGKSPN wich calculates - using the routines QUADO, MONIKA and SORBET - the densities and pressures of the spinodals of the water.

HGKSPN calculates  $\partial P / \partial \rho$  at different densities on an isotherm using the H.G.K.-code.

QUADO selects 3 adjacent densities nearest to the root of Eq. [4.3] and fits a 2. grade polynomial -  $Q_2$  - to them.

MONIKA is needed only to ensure, that the densities are properly ordered. QUADO solves the resulting quadratic equation

$$Q_2(\rho) = 0$$

and selects the appropriate root as the first approximation for the spinodal density,  $\rho_{SL1}$ .

SORBET discards now one of the first 3 density values - the density, with the worst fitting - and replaces it with  $\rho_{SL1}$ . These 3 densities are then returned to QUADO and a new root for  $\partial P / \partial \rho$ ,  $\rho_{SL2}$  will be calculated. The procedure endures, till Eq. [4.3] is fulfilled with sufficient accuracy.

## The spinodal densities

Figure 18. shows the near-critical part of the calculated spinodal densities. As one can see, above the temperature of 635 K the spinodal lines get more and more irregular in shape. Above  $T = 646.3$  K the H.G.K. code prohibits altogether the calculation of the spinodals.

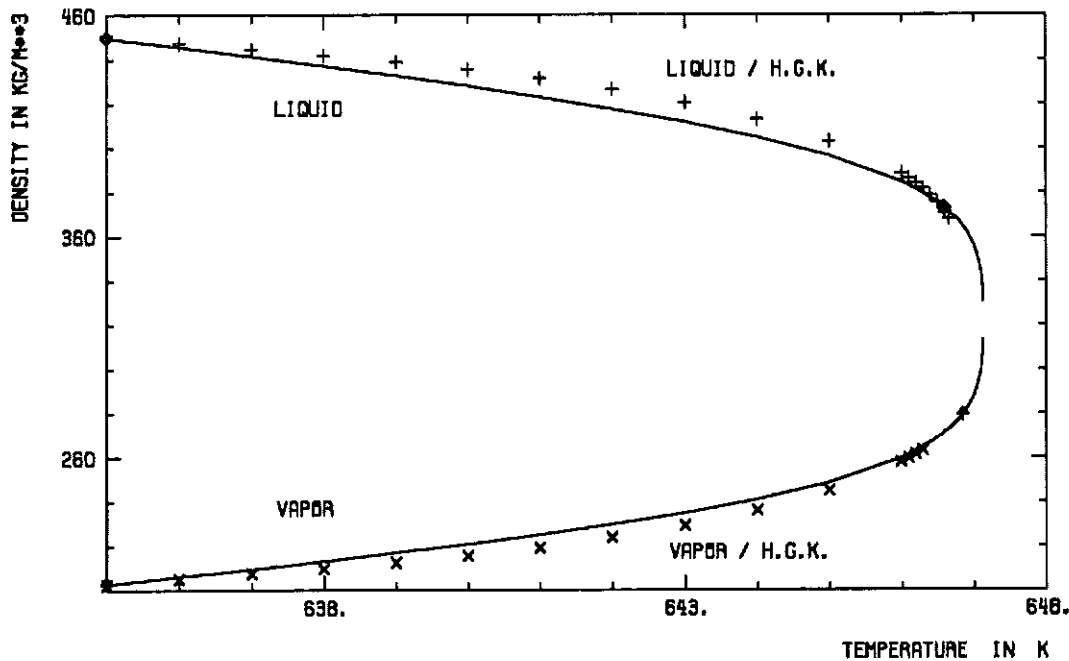


Figure 18. Spinodal densities in the vicinity of the critical point

AQUA describes the spinodal densities - in the same way as the saturated densities - as functions of  $w$  resp.  $\tau$  ( s. Eq. [5.1 - 5.2] ) .

**Below 620 K** the density as well of the liquid, as of the vapor spinodal is approximated with polynomials

$$\rho_{sl}(T) = \rho_c \cdot \sum_{j=1}^9 A_j \cdot w^{j-1} \quad , \quad \rho_{sv}(T) = \rho_c \cdot \sum_{j=1}^7 C_j \cdot w^{j-1} \quad . \quad [5.13]$$

Table 12. Polynomial coefficients A(1) - A(9) of the density of the liquid spinodal					
1	1.911888766	2	-16.54157899	3	179.7358194
4	-940.1374463	5	2957.964417	6	-5752.429056
7	6772.987553	8	-4428.001759	9	1234.441322

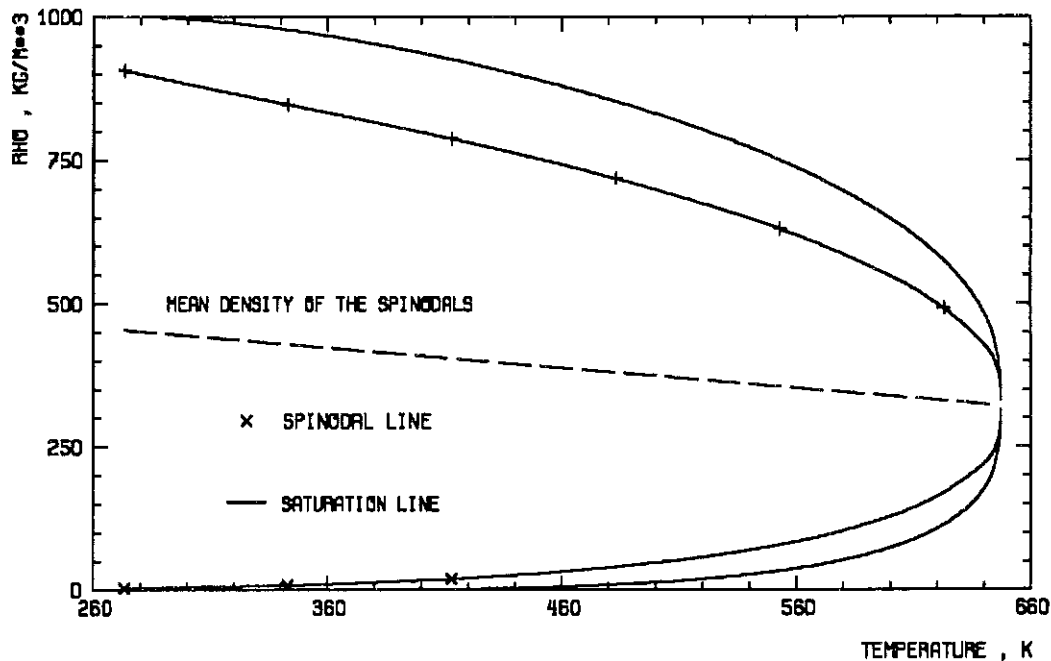
Table 13. Polynomial coefficients C(1) - C(7) of the density of the vapor spinodal					
1	0.9201508683	2	-2.159701944	3	-0.2932760492
4	6.324940268	5	-10.49597891	6	8.78261245
7	-3.119252324				

The coefficients in Eq. [5.13] were fitted to the liquid resp. vapor densities calculated by HGKSPN.

**Above 620 K** the spinodal densities of the H.G.K.-isotherms are not reliable to be used as a fitting source. Instead of using them, a [5.11] - type near-critical extension

$$\begin{aligned} \rho_{SL}(T) &= \rho_c \cdot [1 + 1.40 \cdot \tau^\beta + \tau (U_1 + \tau (U_2 + \tau^2 U_4))] \\ \rho_{SV}(T) &= \rho_c \cdot [1 - 1.40 \cdot \tau^\beta + \tau (W_1 + \tau (W_2 + \tau^2 W_4))] \end{aligned} \quad [5.14]$$

was constructed.



**Figure 19. Saturation and spinodal densities of the water**

$\rho_c$  and  $\beta$  in Eq. [5.14] ensure, that for all near-critical isotherms the density-relation

$$\rho_V < \rho_{SV} < \rho_c < \rho_{SL} < \rho_L$$

holds. The critical amplitude - 1.4 in Eq. [5.14] - was found by comparing the results of Eq. [5.14] with the H.G.K.-data at temperatures above 640 K. The coefficients of the fitting tails

$$U_1 = 0.4871717941 \quad , \quad U_2 = 24.26899415 \quad , \quad U_4 = -2461.494028$$

$$W_1 = 1.132819731 \quad , \quad W_2 = -30.70131059 \quad , \quad W_4 = 3324.944936$$

are calculated again - as in the case of the saturated densities - by demanding double smooth connections at 620 K.

Figure 19 displays the saturated and the spinodal densities as functions of the temperature. It is interesting to note that the mean spinodal

$$\rho_{SM} = (\rho_{SL} + \rho_{SV})/2$$

is a linear function of the temperature.

### The spinodal pressures

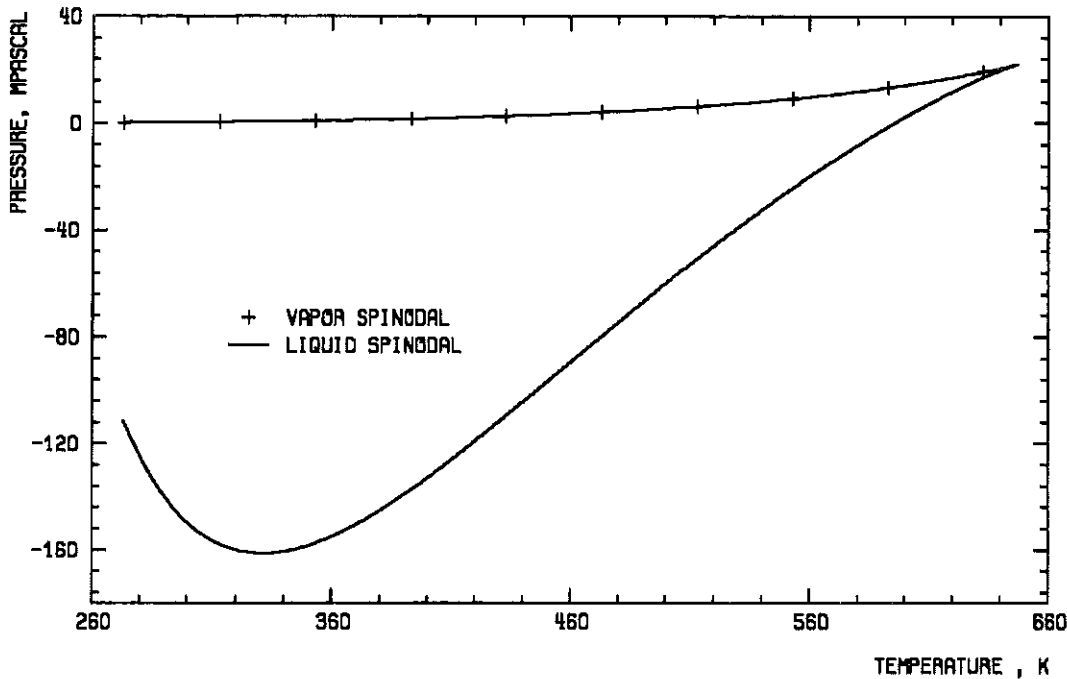


Figure 20. Pressures on the spinodals of the water

In the vicinity of the critical point the spinodal pressures - like the density - are not accessible via the H.G.K.-code. Moreover, at every subcritical T the pressure relation

$$P_{SL}(T) < P^x(T) < P_{SV}(T)$$

must hold. To cope with these conditions AQUA approximates below 620 K the spinodal pressures - calculated by HGKSPN - with  $\tau$ -polynomials and describes - near  $T_c$  - their deviations from  $P^x$ .

**Below 620 K** the pressures are described as :

$$P_{SL}(T) = \sum_{j=1}^9 A_j \cdot \tau^{j-1} \quad , \quad P_{SV}(T) = \sum_{j=1}^8 C_j \cdot \tau^{j-1} \quad . \quad [5.16]$$

1	-0.001387361801	2	0.1510341901	3	1.980521347
4	2.921317875	5	-38.3240633	6	162.2329521
7	-384.0027285	8	479.3961544	9	-264.1248035

Table 15. Polynomial coefficients C(1) - C(8) of the pressure of the vapor spinodal					
1	-0.0007497146775	2	0.162845372	3	-0.3190428782
4	-1.582269493	5	6.787420254	6	-10.41333815
7	7.599390008	8	-2.285884902		

Above 620 K the descriptions are :

$$P_{SL}(T) = P^x(T) - 4.98 \cdot 10^8 \cdot \sum_{j=1}^3 U_j \cdot \tau^j \quad , \quad [5.17]$$

$$P_{SV}(T) = P^x(T) + 2.00 \cdot 10^8 \cdot \sum_{j=1}^3 W_j \cdot \tau^j \quad .$$

The polynomials in Eq. [5.17] vanish at the critical point as  $\tau$ . Their coefficients

$$U_1 = 0.05235363429 \quad , \quad U_2 = 4.419437236 \quad , \quad U_3 = -18.86727538$$

$$W_1 = 0.1103469546 \quad , \quad W_2 = 0.913150777 \quad , \quad W_3 = -10.87001272$$

had been gained by using double smooth connections at 620 K.

### 5.3 The entropy on the saturation line

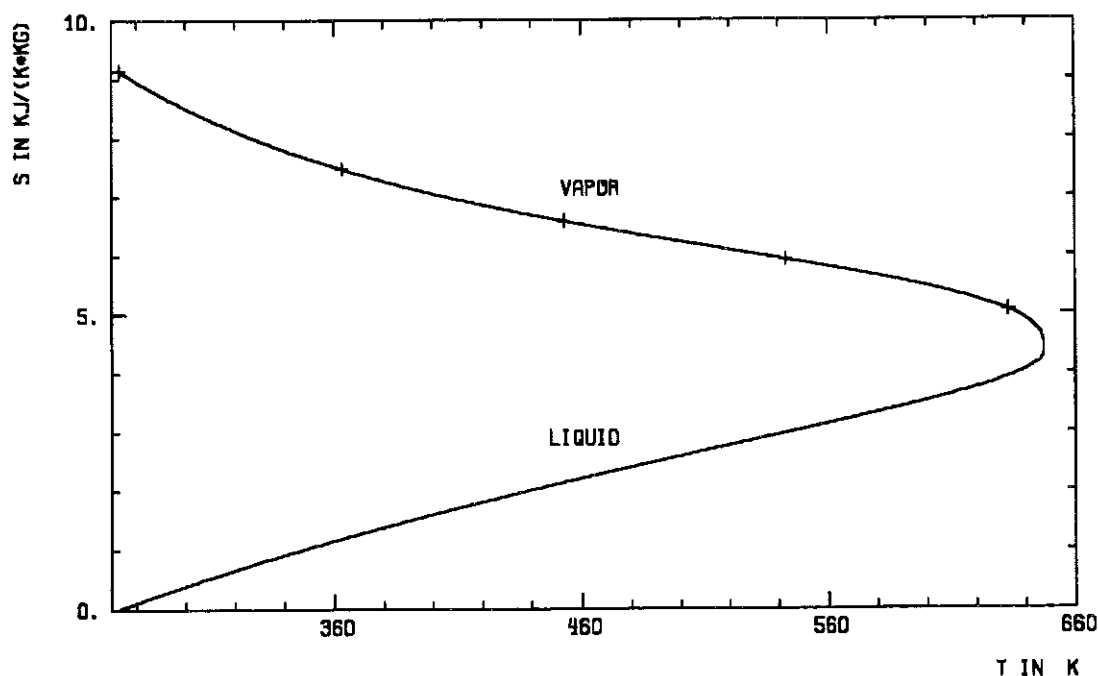


Figure 21. Entropy on the saturation line of the water

The entropy in the saturated states show a course similar to the course of the saturated densities. Consequently AQUA employs - for the description of the entropy of the saturated liquid - a formula, similar to the one used for the description of the saturated densities.

**Below 620 K** the entropy of the saturated liquid - calculated by the H.G.K. code - is described with a w - polynomial ( s. Eq. [5.1 - 5.2] )

$$S_L(T) = S_c \cdot \sum_{j=1}^9 A_j \cdot w^{j-1} \quad . \quad [5.18]$$

1	0.9528621116	2	0.2725494325	3	-8.457375782
4	44.57631748	5	-150.7441559	6	311.7914429
7	-391.8528748	8	273.5529633	9	-82.2848053

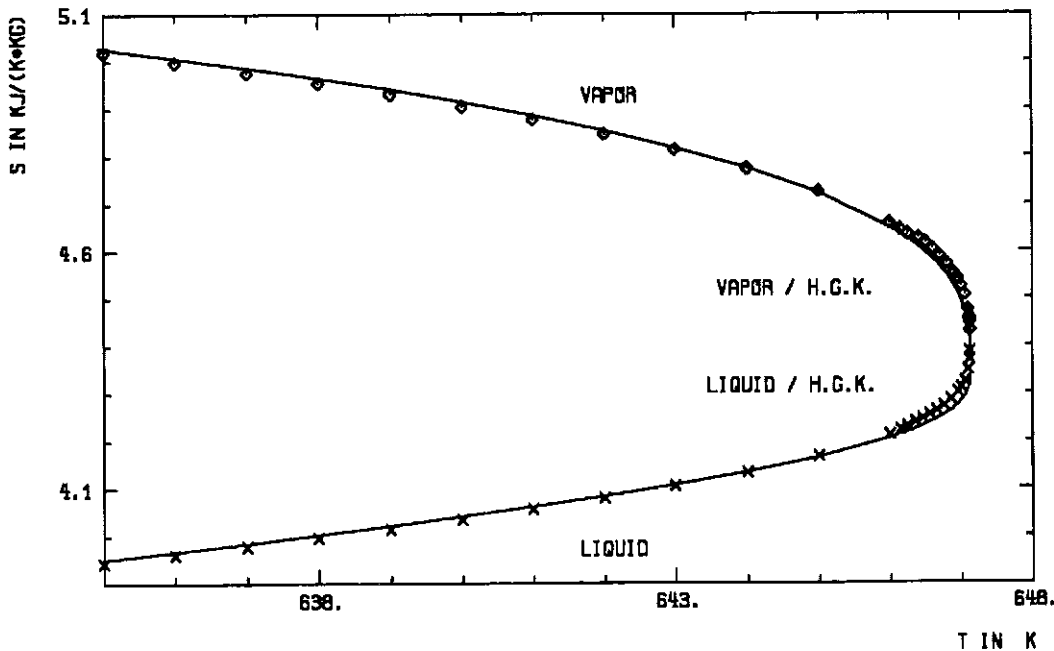


Figure 22. Entropy of the saturated states in the vicinity of  $T_c$

**Above 620 K** the entropy of the liquid is described with a "vapor-like" scaled formula :

$$S_L(T) = S_c \cdot [1 - 0.25 \cdot \tau^\beta + \tau (U_1 + \tau (U_2 + \tau^2 U_4))] \quad . \quad [5.19]$$

The critical entropy and the scaled amplitude

$$S_c = 9.43 \cdot R_{gas} \quad \text{and} \quad B_s = 0.25 \quad [5.20]$$

had been determined by fitting as well Eq. [5.19], as the corresponding formula for the entropy of the vapor to the H.G.K.-values above 635 K. The coefficients



$$U_1 = -1.23383958 \quad , \quad U_2 = -1.656615363 \quad , \quad U_4 = 911.5375812$$

result from using a double smooth extension to higher temperatures at the switching point.

The entropy on the saturated vapor needs no extra description, it can be calculated from  $S_L$  by the eq. of Clausius & Clapeyron

$$S_V = S_L + \left( \frac{1}{\rho_V} - \frac{1}{\rho_L} \right) \cdot \frac{dP^x}{dT} \quad .$$

#### 5.4 The transport properties on the saturation line

The following descriptions for the viscosity resp. thermal conductivity in the saturated states were constructed by calculating the properties on the saturation line with the algorithms, described in 3.0, "Transport properties of the water." on page 9 and approximating these values with adequate functions.

##### The viscosity

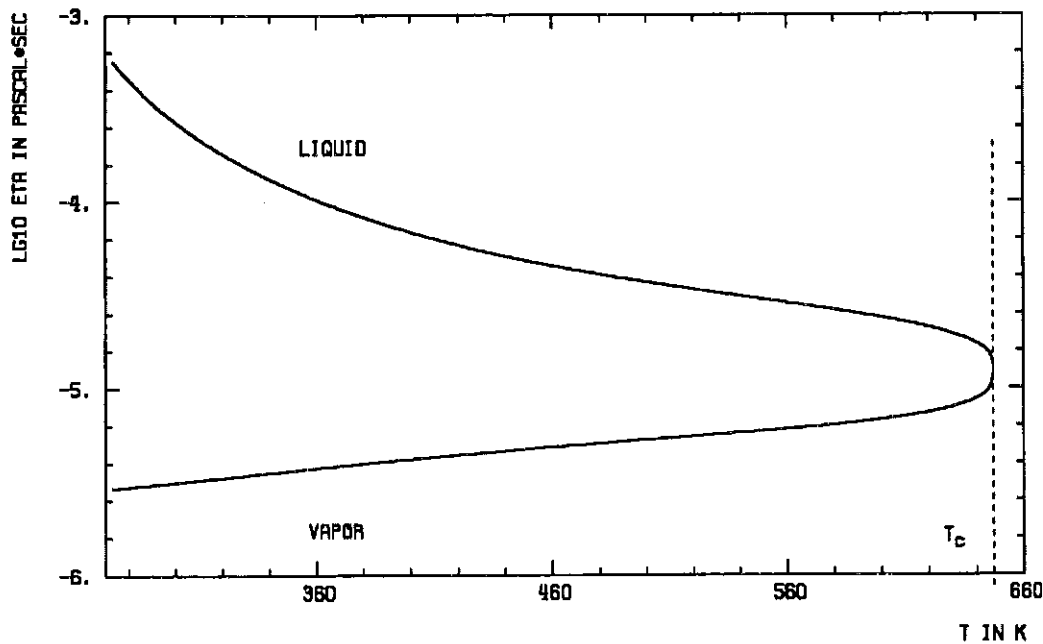


Figure 23. Viscosity of the water in the saturated states

In describing the viscosities in the cold liquid resp. vapor different functions of the critical variable  $w$  ( Eq. [5.2] ) were used :

$$\eta_L(T) = \frac{\eta_c}{\sum_{j=1}^{11} A_j \cdot w^{j-1}} \quad , \quad \eta_V(T) = \eta_c \cdot \sum_{j=1}^9 C_j \cdot w^{j-1} \quad . \quad [5.21]$$

1	0.945822019	2	-3.50567042	3	23.61506749
4	-176.2217806	5	960.5599634	6	-3533.866457
7	8609.716158	8	-13637.7187	9	13435.90472
10	-7453.116048	11	1776.272412		

1	0.9733280423	2	-4.061585251	3	20.72025593
4	-80.06616536	5	214.64856	6	-384.6194719
7	433.0818384	8	-278.1018786	9	78.25010044

In the vicinity of  $T_c$  density-like formulas were adapted :

$$\begin{aligned} \eta_L(T) &= \eta_c \cdot [1 + 1.50 \cdot \tau^\beta + \tau (U_1 + \tau (U_2 + \tau^2 U_4))] \quad , \\ \eta_V(T) &= \eta_c \cdot [1 - 1.50 \cdot \tau^\beta + \tau (W_1 + \tau (W_2 + \tau^2 W_4))] \quad . \end{aligned} \quad [5.22]$$

The polynomial coefficients

$$\begin{aligned} U_1 &= 4.669845392 \quad , \quad U_2 = -19.58930391 \quad , \quad U_4 = 1768.652432 \\ W_1 &= 4.262680597 \quad , \quad W_2 = -34.89896734 \quad , \quad W_4 = 2452.105565 \end{aligned}$$

were calculated using the restrictions of double-smooth connections at 620 K to the "cold" descriptions. By comparing the functions [5.22] with the calculated viscosities above 635 K the best fitting gives the following critical viscosity :

$$\eta_c = 3.95 \cdot 10^{-5} \quad \text{Pa} \cdot \text{sec} \quad . \quad [5.23]$$

### The thermal conductivity

AQUA describes the thermal conductivities in the whole T-region as function of  $\tau$ . Below 620 K the following polynomials were used :

$$\lambda_L(T) = \sum_{j=1}^8 A_j \cdot \tau^{j-1} \quad , \quad \lambda_V(T) = \sum_{j=1}^9 C_j \cdot \tau^{j-1} \quad . \quad [5.24]$$

1	0.397252283	2	1.361430056	3	0.2886579476
4	-1.137153088	5	-75.20016424	6	333.88123
7	-546.0632435	8	311.575529		

1	0.2012600289	2	-2.546881317	3	19.24073027
4	-79.8925815	5	178.6322576	6	-195.5137779
7	62.06051011	8	38.86696488	9	-15.11624233

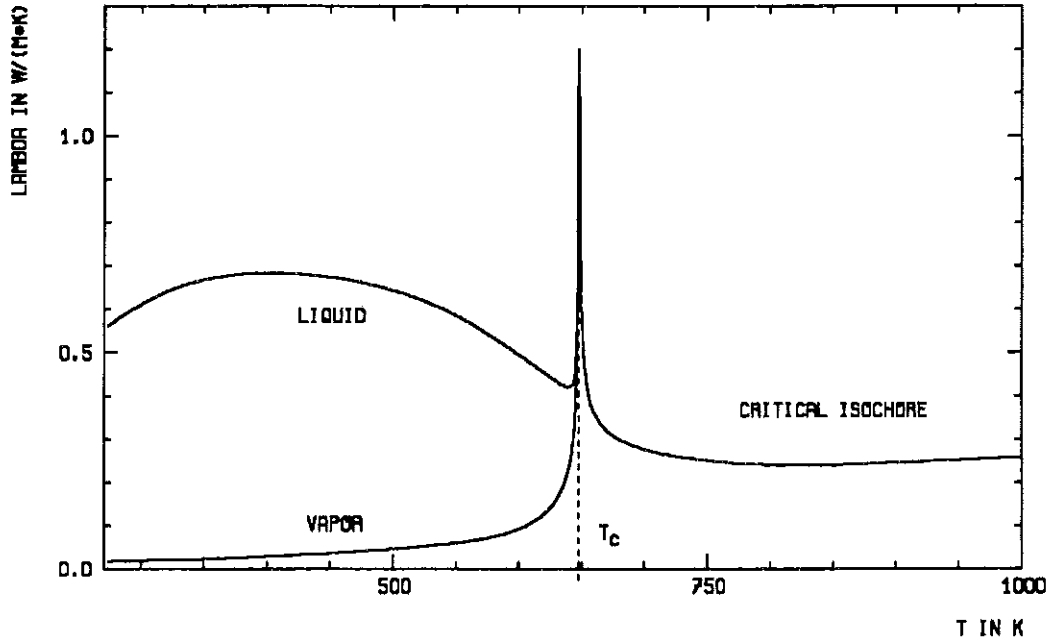


Figure 24. Thermal conductivity of the water as function of the temperature

At the critical point the thermal conductivity approaches infinity as

$$\lambda \propto \tau^{-0.5} \quad [5.25]$$

Consequently above 620 K the following near-critical descriptions had been applied :

$$\lambda_L(T) = \tau^{-0.5} \cdot \sum_{j=1}^5 U_j \cdot \tau^{j-1} \quad , \quad \lambda_V(T) = \tau^{-0.5} \cdot \sum_{j=1}^5 W_j \cdot \tau^{j-1} \quad . \quad [5.26]$$

The first coefficients of the appended polynomials are equal :

$$U_1 = W_1 = 0.02 .$$

For the remaining coefficients the condition of the double smooth continuation at 620 K gives :

$$\begin{aligned} U_2 &= 2.484618723 & , & & U_3 &= -31.16080409 & , \\ U_4 &= 421.0825394 & , & & U_5 &= -2390.608760 & , \\ W_2 &= 0.7221916099 & , & & W_3 &= -29.169893490 & , \\ W_4 &= 495.59365880 & , & & W_5 &= -3169.2527940 & . \end{aligned}$$

## 5.5 Other saturated properties

The following properties are all described in AQUA as functions of the scaled temperature  $\tau$ , only the form of the functions differ in the cold ( $T < 620 \text{ K}$ ) and in the near-critical ( $T > 620 \text{ K}$ ) part of saturation line. Below 620 K  $\tau$ -polynomials were fitted to the property-values calculated with the H.G.K. code. Above 620 K scaled descriptions ( s. e.g. Eq. [5.30] ) are used. The values of the critical parameters in these formula are gained by fitting the descriptions to the H.G.K.-functions in the region

$$635 \text{ K} < T < T_c \quad . \quad [5.27]$$

The remaining coefficients of the high-T descriptions are defined by demanding a double-smooth connection ( Eq. [5.12] ) for them to the low-T formulas at  $T_s = 620 \text{ K}$ .

In all the cases the descriptions had been developed for the reduced properties -  $P_r^0$ ,  $P_r^0$ ,  $C_v^0$  and  $G^0$  - since AQUA calculates internally with these functions.

### The density derivative of the pressure

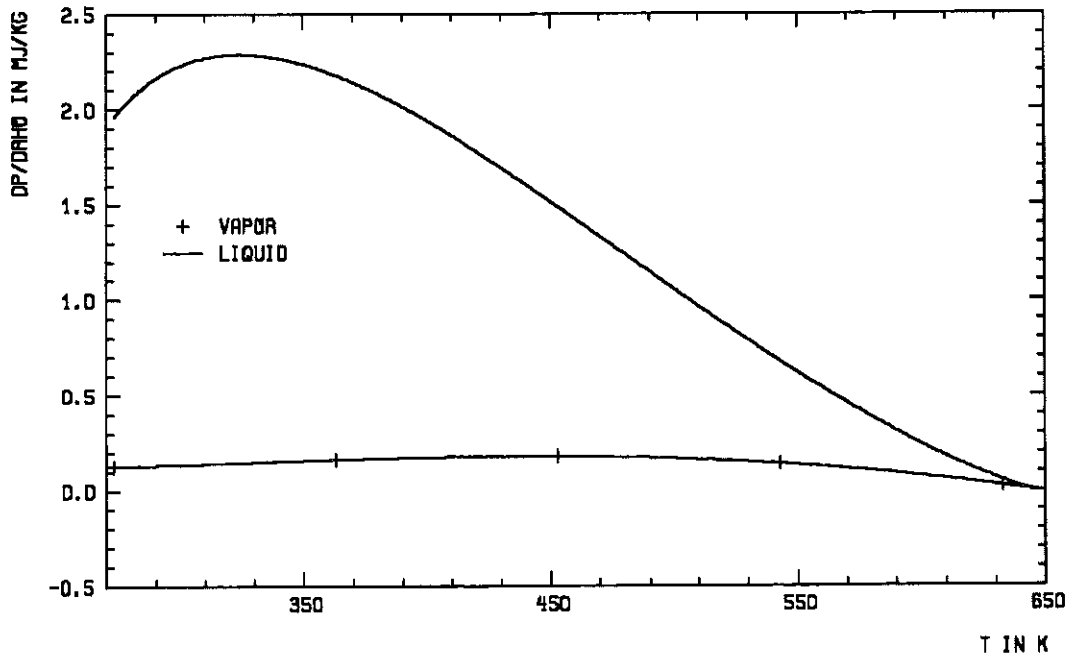


Figure 25. Density derivatives of the pressure in the saturated states

1	0.09168326854	2	0.1410358804	3	293.745773
4	-3694.718543	5	35315.9487	6	-224505.1913
7	968102.1275	8	-2837321.019	9	5560661.663
10	-6973719.089	11	5058540.246	12	-1615244.963

1	0.001435429119	2	3.738553917	3	3.552096947
4	-55.88507759	5	217.5743793	6	-535.2445068
7	812.4224243	8	-670.9285889	9	229.0827484

Table 21 and Table 22 contain the coefficients of the low temperature description of the density-derivatives for the saturated liquid resp. for the saturated vapor

$$\frac{\partial P}{\partial \rho_L}(T) = R_{gas} \cdot T \sum_{j=1}^{12} A_j \cdot \tau^{j-1} \quad , \quad \frac{\partial P}{\partial \rho_V}(T) = R_{gas} \cdot T \sum_{j=1}^9 C_j \cdot \tau^{j-1} \quad . \quad [5.28]$$

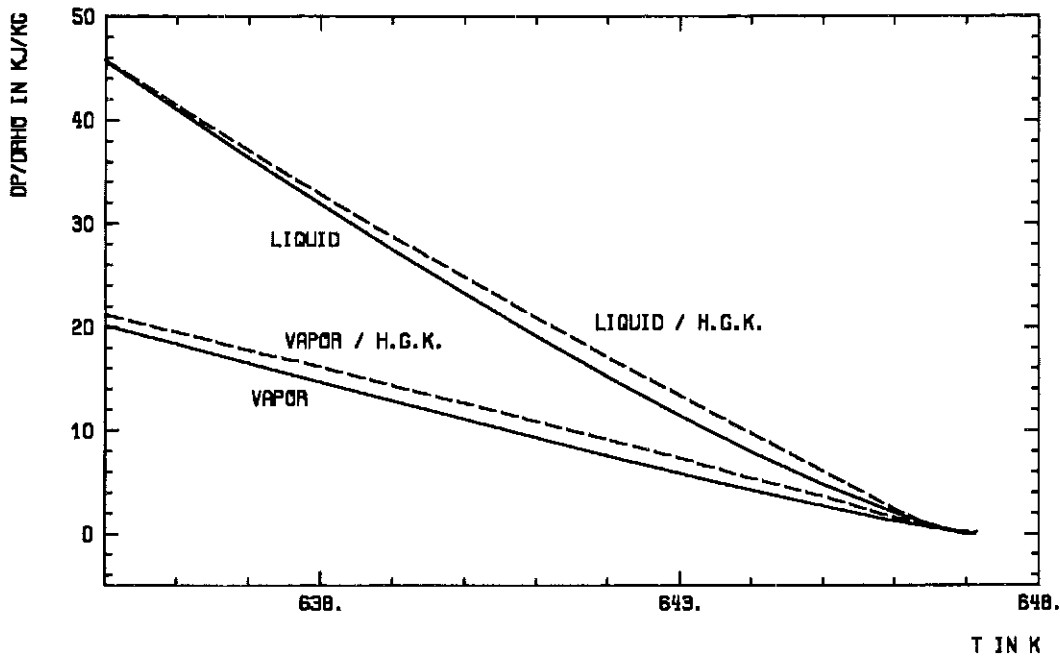


Figure 26. Density derivatives of the pressure in the vicinity of  $T_c$ .

At the critical point the density-derivatives vanish as

$$\frac{\partial P}{\partial \rho_L} \propto \rho_L \cdot \tau^\gamma \quad \text{resp.} \quad \frac{\partial P}{\partial \rho_V} \propto \rho_V \cdot \tau^\gamma \quad [5.29]$$

with the critical exponent  $\gamma = 1.24$  ( s. e.g. /17/ ) so the following near-critical descriptions had been applied :

$$\frac{\partial P}{\partial \rho_L}(T) = R_{gas} \cdot T \cdot \frac{\rho_L}{\rho_c} \cdot \tau^\gamma \sum_{j=1}^4 U_j \cdot \tau^{j-1} \quad , \quad [5.30]$$

$$\frac{\partial P}{\partial \rho_V}(T) = R_{gas} \cdot T \cdot \frac{\rho_V}{\rho_c} \cdot \tau^\gamma \sum_{j=1}^4 W_j \cdot \tau^{j-1} \quad .$$

The first coefficients of the polynomials in Eq. [5.30] are the same :

$$U_1 = W_1 = 15$$

and here are the remaining ones :

$$\begin{aligned} U_2 &= -79.73057836 & , & & U_3 &= -430.1476885 & , & & U_4 &= 14139.7632 & , \\ W_2 &= 369.6750971 & , & & W_3 &= -5110.089071 & , & & W_4 &= 46121.0908 & . \end{aligned}$$

### The temperature derivative of the pressure

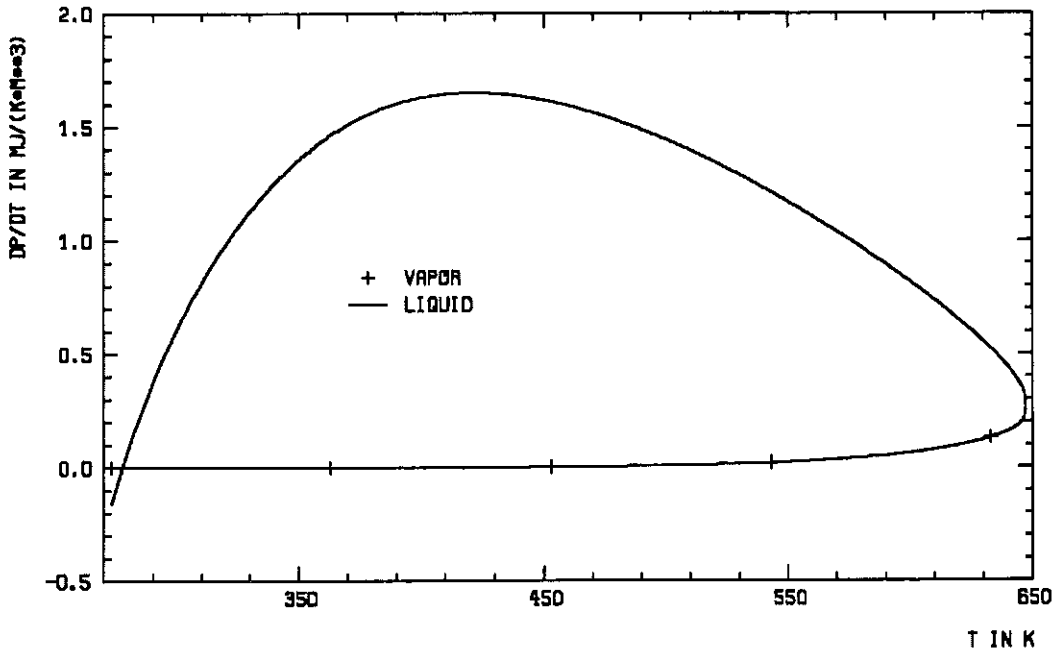


Figure 27. Temperature derivatives of the pressure in the saturated states

The temperature derivatives show quite a regular behaviour at  $T_c$ , so one can use polynomials as well on the cold saturation line as on the near critical one. Below 620 K the polynomial fittings resulted in

$$\frac{\partial P}{\partial T_L}(T) = R_{gas} \cdot \rho_L \sum_{j=1}^{12} A_j \cdot \tau^{j-1} \quad , \quad \frac{\partial P}{\partial T_V}(T) = R_{gas} \cdot \rho_V \sum_{j=1}^{10} C_j \cdot \tau^{j-1} \quad . \quad [5.31]$$

1	1.818884544	2	16.40737736	3	-65.85455759
4	52.04371914	5	4031.238729	6	-47257.0313
7	277423.4268	8	-988332.4003	9	2218919.758
10	-3077677.674	11	2414555.699	12	-821000.2244

1	2.029700729	2	-0.1255470649	3	-61.54407176
4	544.8751175	5	-2848.188899	6	9708.977783
7	-21262.51929	8	28710.96533	9	-21755.27246
10	7082.224609				

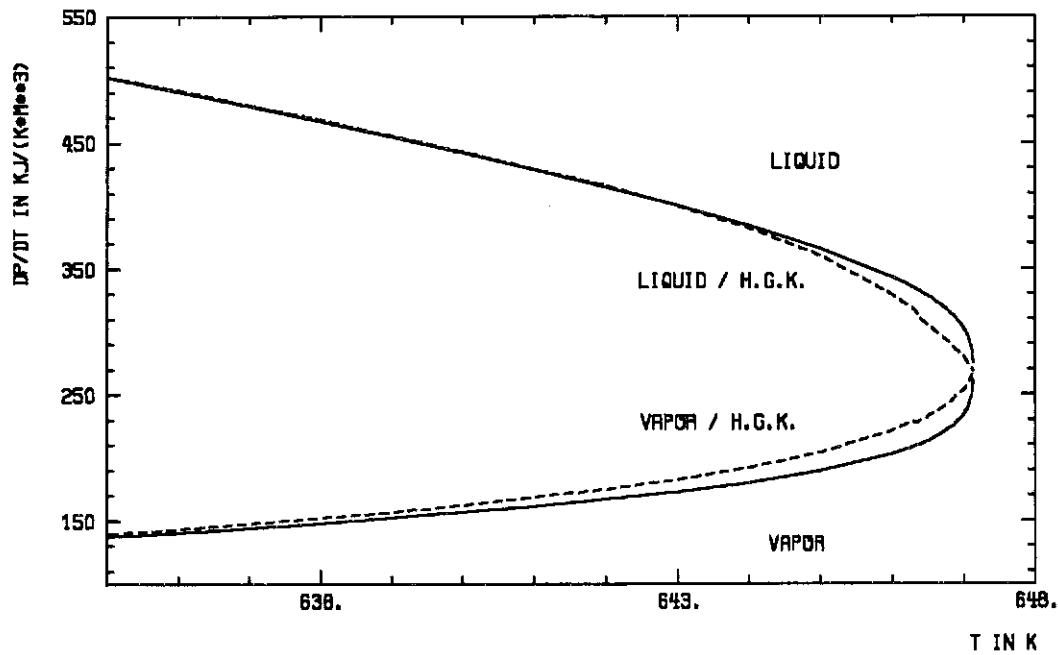


Figure 28. Temperature derivatives of the pressure in the vicinity of  $T_c$

Above 620 K the double smooth extension gives

$$\frac{\partial P}{\partial T_L}(T) = R_{gas} \cdot \rho_L \sum_{j=1}^4 U_j \cdot \tau^{j-1} \quad , \quad [5.32]$$

$$\frac{\partial P}{\partial T_V}(T) = R_{gas} \cdot \rho_V \sum_{j=1}^4 W_j \cdot \tau^{j-1} \quad .$$

The first coefficients of the polynomials in Eq. [5.32] must be the same

$$U_1 = W_1 = 1.80215732$$

to fulfil

$$\frac{\partial P}{\partial T_V}(T_c) = \frac{dP^x(T_c)}{dT} = \frac{\partial P}{\partial T_L}(T_c) \quad . \quad [5.34]$$

The remaining ones are :

$$\begin{aligned}
 U_2 &= 17.63135647 & , & & U_3 &= -97.95806903 & , & & U_4 &= 451.8053522 & , \\
 W_2 &= 16.00172011 & , & & W_3 &= -439.3063303 & , & & W_4 &= 3359.729201 & .
 \end{aligned}$$

### The heat capacities

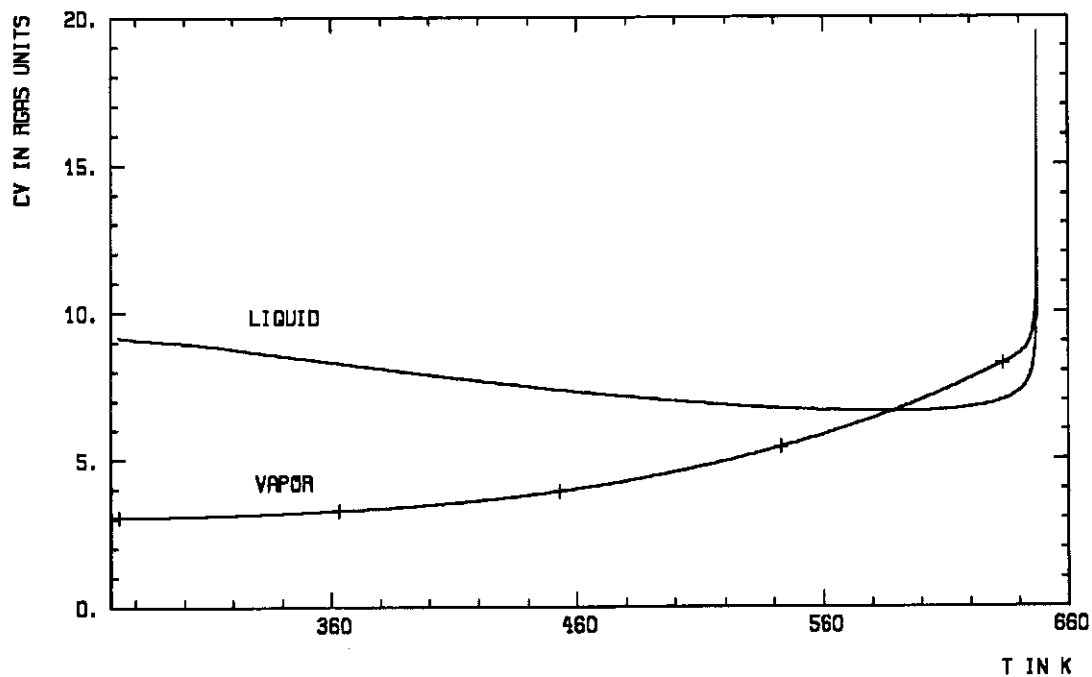


Figure 29. Heat capacities of the water in the saturated states.

AQUA describes the heat capacities in  $R_{gas}$  units. Below 620 K polynomials were used :

$$C_{VL}(T) = R_{gas} \cdot \sum_{j=1}^{12} A_j \cdot \tau^{j-1} \quad , \quad C_{VV}(T) = R_{gas} \cdot \sum_{j=1}^{10} C_j \cdot \tau^{j-1} \quad . \quad [5.35]$$

1	7.4305055	2	-24.93618016	3	195.5654567
4	1986.485797	5	-53305.43411	6	505697.1723
7	-2724774.677	8	9167737.673	9	-19622033.78
10	25984725.33	11	-19419431.35	12	6263206.554



1	8.956404735	2	-33.97230774	3	126.5786602
4	-513.7462523	5	1521.138693	6	-2182.266721
7	-73.57828067	8	4483.361889	9	-5407.324042
10	2099.821164				

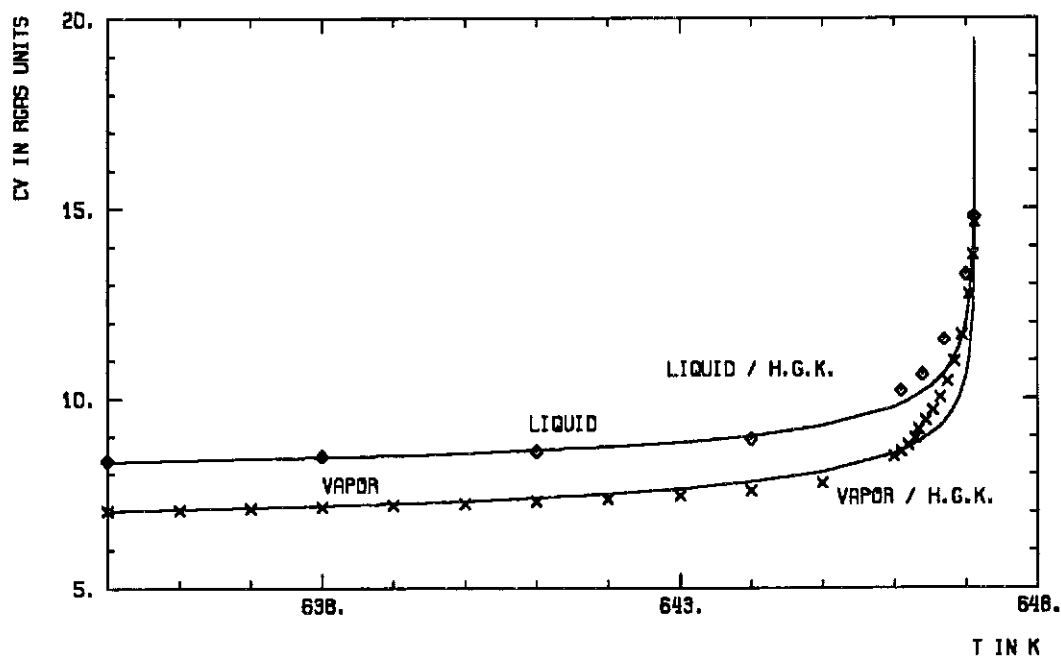


Figure 30. Heat capacities of the water in the vicinity of  $T_c$ .

At the critical point the heat capacity approaches infinity as

$$C_V \propto \tau^\alpha \quad [5.37]$$

with the critical exponent  $\alpha = -0.1$  (s. e.g. /17/ ). Consequently above 620 K the following near-critical descriptions had been applied :

$$C_{VL}(T) = R_{gas} \cdot \tau^\alpha \cdot \sum_{j=1}^4 U_j \cdot \tau^{j-1} \quad , \quad C_{VV}(T) = R_{gas} \cdot \tau^\alpha \cdot \sum_{j=1}^4 W_j \cdot \tau^{j-1} \quad . \quad [5.38]$$

The double smooth continuation at 620 K results in the polynomial-coefficients :

$$U_1 = 4.5 \quad , \quad U_2 = 15.21452007 \quad , \quad U_3 = -167.2951053 \quad , \quad U_4 = 1042.333618 \quad , \\ W_1 = 5.1 \quad , \quad W_2 = 42.35451412 \quad , \quad W_3 = -1002.675527 \quad , \quad W_4 = 6911.203183 \quad .$$

## The free enthalpy, G

The Gibbs function shows not even a trace of irregularity at  $T_c$ . A single polynomial suffices therefore, to describe it in the whole temperature region :

$$G_L(T) = G_V(T) = R_{gas} \cdot T \sum_{j=1}^9 A_j \cdot \tau^{j-1} \quad [5.39]$$

1	-2.569516966	2	5.187876405	3	0.6408677408
4	1.374106295	5	-18.01831251	6	68.53337198
7	-176.2024879	8	226.296896	9	-130.4997179

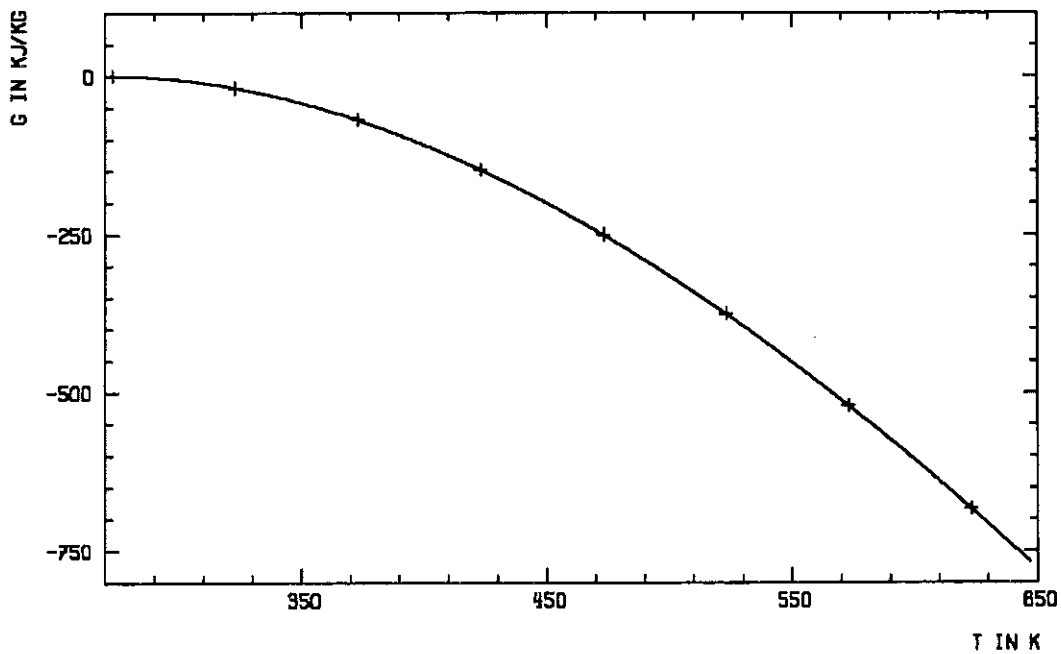


Figure 31. Free enthalpy - G - of the water in the saturated states



## 6.0 Calculating the densities from the pressures

IVA3 supplies the equation of state with the temperature and the pressure as independent variables, the H.G.K.-routines calculate the thermophysical properties as temperature-density functions. So obviously a procedure is needed to calculate the variable-pair ( T,  $\rho$  ) from the variable-pair ( T, P ). AQUA uses three routines to do this task : DLFIND seeks  $\rho$  ( T, P ) in the liquid, DVFIND in the vapor, and DGFIND in the supercritical water.

### 6.1 Calculating small density changes

All of these routines

```
DGFIND(T,P, $\rho$ ,IECCH)
DLFIND(T,P, $\rho$ ,IECCH)
DVFIND(T,P, $\rho$ ,IECCH)
```

use the same searching-algorithm : beginning with an initial density,  $\rho_0$  - supplied by the user - a first pressure,  $P_0$  and  $\partial P/\partial \rho$  are calculated, then a corrected density is derived - from the difference,  $P - P_0$  - by Newton's method.

Table 28. Coefficients of $P_{Lmax}(T)$ .			
$P_{Lmax} = A_0 + A_1 T^1 + A_2 T^2 + \dots$			
T >	0 K	373.15 K	647.126 K
$A_0$	$128.0317635 \cdot 10^9$	$1.121372093 \cdot 10^9$	$-731.6836907 \cdot 10^9$
$A_1$	$-1.503427271 \cdot 10^9$	$-6.278371391 \cdot 10^8$	$6.597512847 \cdot 10^8$
$A_2$	$6.608885188 \cdot 10^8$	$26.44577782 \cdot 10^3$	$-7.08407094 \cdot 10^3$
$A_3$	$-12.83080180 \cdot 10^3$	-33.36474460	5.764846806
$A_4$	9.305682808	$14.81761478 \cdot 10^{-3}$	$-3.188721412 \cdot 10^{-3}$
$A_5$			$1.193485372 \cdot 10^{-6}$
$A_6$			$-2.946300976 \cdot 10^{-10}$
$A_7$			$4.575561580 \cdot 10^{-14}$
$A_8$			$-4.040003535 \cdot 10^{-18}$
$A_9$			$1.543953233 \cdot 10^{-22}$

Before beginning with the iteration, the routines check, if there is a possibility at all to find a density, corresponding to the supplied pressure, P in the given area. To this aim P is compared with a pressure-window,

$$P_{min} \leq P \leq P_{max}$$

and if P don't fit into it, the routines end with an error message ( for the error handling s. Appendix L, "Return codes" on page 151 ). The borders of the pressure-window are

$$P_{min} = 0 \quad , \quad P_{max} = P_{Lmax}(T) \quad [6.1]$$

in the supercritical water,

$$P_{\min} = 0 \quad , \quad P_{\max} = P_{SV}(T) \quad [6.2]$$

in the vapor, and

$$P_{\min} = P_{SL}(T) \quad , \quad P_{\max} = P_{Lmax}(T) \quad [6.3]$$

in the liquid respectively.

$P_{Lmax}$  in the above eq.s is the pressure at the liquid edge of the validity-domain

$$P_{Lmax}(T) = P(T, 1200 \text{ kg/m}^3) \quad . \quad [6.4]$$

AQUA calculates  $P_{Lmax}$  with the temperature polynomials given in Table 28 ( routine PLMAX ). The description is valid in the temperature range

$$273.15 \text{ K} \leq T \leq 5000 \text{ K} \quad .$$

The remaining two temperature-functions in Eq. [6.1 - 6.3],  $P_{SV}(T)$  resp.  $P_{SL}(T)$  are the spinodal pressures in the vapor resp. in the liquid, Eq. [5.16 - 5.17]. At least as long the temperatures remain below 635. K. Above this point the shape of the H.G.K.-isotherms begins to differ markedly from the shape, one would expect in the neighbourhood of the critical point. As well the pressure, as the density - particularly of the liquid spinodal - differ from the descriptions given in 5.0, "Describing the water properties in the saturated states" on page 27 ( compare also Figure 18 on page 31 ). Figure 32 shows a massively enlarged part of the 647 K isotherm of the water. On this isotherm the liquid spinodal lies in the vapor region already.

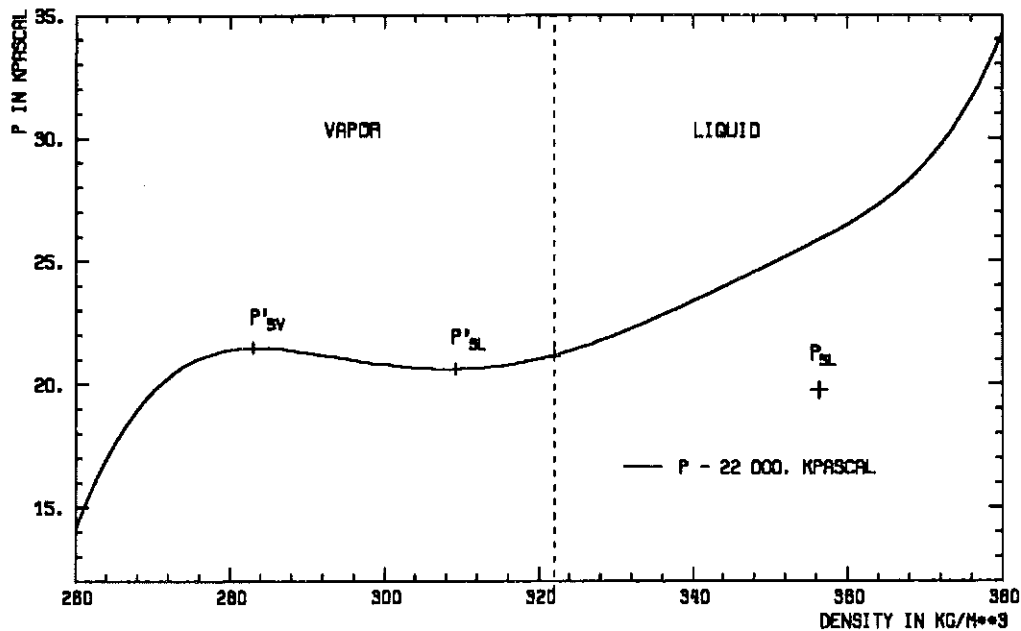


Figure 32. Water pressures on the 647 K isotherm

AQUA uses additional  $\tau$ -polynomials to describe the real pressure minima and maxima  $P'_{SL}$ ,  $P'_{SV}$  in the region above 635 K :

$$P'_{SL}(T) = P_{SL}(T) + \sum_{j=1}^9 U_j \cdot \tau^j \quad , \quad [6.5]$$

$$P'_{SV}(T) = P_{SV}(T) - \sum_{j=1}^6 W_j \cdot \tau^j \quad .$$

with the coefficients

1	40.49363544	2	$-20.73773694 \cdot 10^3$	3	$7.607260219 \cdot 10^8$
4	$-1.796520658 \cdot 10^9$	5	$258.9287847 \cdot 10^9$	6	$-22.71838331 \cdot 10^{12}$
7	$1.182066587 \cdot 10^{15}$	8	$-33.45912597 \cdot 10^{15}$	9	$396.4473916 \cdot 10^{15}$

resp.

1	17.17830396	2	$-3.434969255 \cdot 10^3$	3	$448.7956073 \cdot 10^3$
4	$-38.08457011 \cdot 10^6$	5	$1.646319316 \cdot 10^9$	6	$-27.19071374 \cdot 10^9$

( routines DILE resp. DIVE ).

In a thin region before the critical temperature,

$$647.0 \text{ K} < T < T_c$$

it can happen, that the liquid pressures of an isotherm remain always above the vapor pressures

$$P'_{SV} < P'_{SL} \quad .$$

This would lead to the unacceptable state, that at a distinct ( near critical ) temperature there would be a pressure, finding no corresponding density either in the vapor, or in the liquid. To avoid this unphysical state DLFIND checks at near critical temperatures the spinodal pressures and uses the smaller of them as  $P_{min}$  in the window [6.3].

## 6.2 Calculating large density changes

If the supplied density,  $\rho_0$  differs massively from the final density,  $\rho ( T, P )$ , then in some cases the 19 iteration-steps, allowed in the FIND-routines will not suffice to calculate the final density. The user can circumvent this difficulty by supplying in this case a zero initial density,

$$\rho_0 = 0$$

to the FIND-routines. Finding a zero initial density the routines search - after checking the pressure, yet before beginning with the iteration - for an adequate approximated density, which will then serve as an initial density for the iteration with Newton's method. To find this density, the routines use an algorithm, similar to the one, used to calculate the spinodal densities ( page 30 ).

DVFIND selects three densities with the corresponding pressures

$$\begin{aligned}
 \rho_1 &= 0 & , & & P_1 &= 0 \\
 \rho_2 &= \rho_V & , & & P_2 &= P^x \\
 \rho_3 &= \rho_{SV} & , & & P_3 &= P_{SV}
 \end{aligned}
 \tag{6.6}$$

and calls the routine DINVA to find a density, corresponding the supplied pressure, P with the help of these points. DINVA supplies the points to the routine QUADO, which constructs from them a polynomial of second grade and solves it for  $\rho_0 ( P )$ . DINVA then calculates the pressure, P (  $\rho_0$  ) and checks, if the  $\rho_0$  is good enough to begin with. If P (  $\rho_0$  ) differs to much from P, then DINVA uses SORBET to discard the "worst" of the four points

$$\rho_1 \quad , \quad \rho_2 \quad , \quad \rho_3 \quad , \quad \rho_0$$

and seeks with the remainig three a better  $\rho_0$ .  
DLFIND uses

$$\begin{aligned}
 \rho_1 &= \rho_{SL} & , & & P_1 &= P_{SL} \\
 \rho_2 &= \rho_L & , & & P_2 &= P^x \\
 \rho_3 &= 1200\text{kg/m}^3 & , & & P_3 &= P_{Lmax}
 \end{aligned}
 \tag{6.7}$$

as pivotal points and calls the routine DINLI to find the density,  $\rho_0$ . DINLI works the same way as DINVA.

DGFIND has a more complicated task to fulfill, as either DVFIND or DLFIND. Since at supercritical temperatures there are no ready-made density-pressure relations to use - as the saturation line or the spinodals - some density-pressure relations must be constructed.

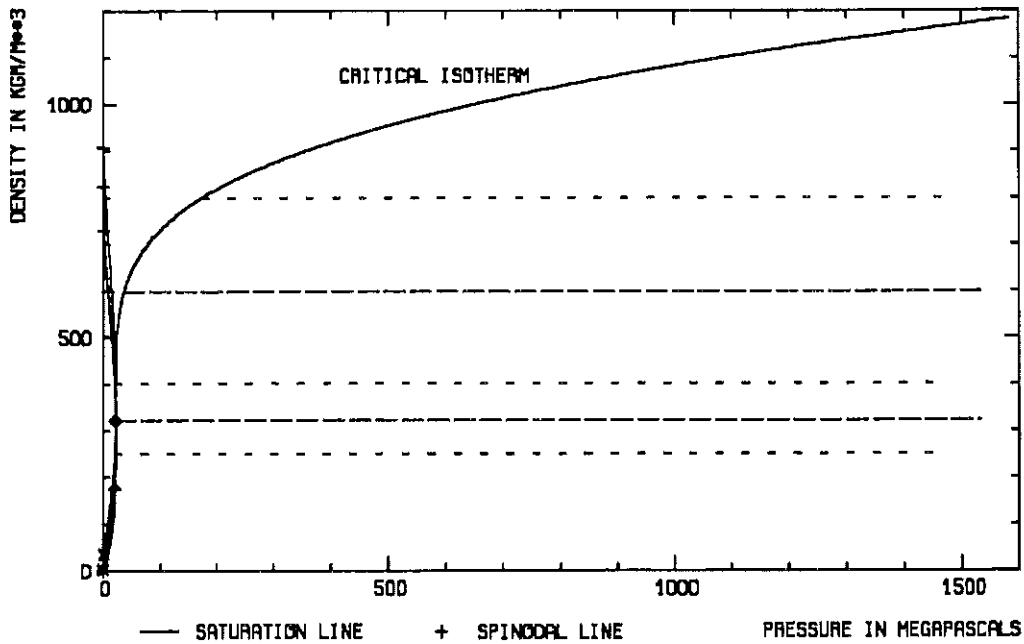


Figure 33. Water. Density-pressure chart

To this end DGFIND divides the supercritical region in three sub-areas:

$$\text{"LD"} : \quad 0 \leq \rho \leq \rho_c$$

the area of low-densities,

$$\text{"MD"} : \quad \rho_c \leq \rho \leq 600 \text{ kg/m}^3$$

the area of moderate-densities and

$$\text{"HD"} : \quad 600 \text{ kg/m}^3 \leq \rho \leq 1200 \text{ kg/m}^3$$

the area of high-densities ( s. Figure 33 ). AQUA supplies DGFIND with the following isochoral pressures :

$$P_{250}(T) = P(T, 250 \text{ kg/m}^3) \quad [6.8]$$

Table 31. Polynomial coefficients A(1) - A(8) of $P_{250}(T)$ . $P_{250} = A_0 + A_1 \cdot T + A_2 \cdot T^2 + \dots$					
1	$-161.6042550 \cdot 10^6$	2	$370.3861090 \cdot 10^3$	3	$-183.5936079$
4	$93.96635527 \cdot 10^{-3}$	5	$-30.07808886 \cdot 10^{-6}$	6	$5.819644183 \cdot 10^{-9}$
7	$-619.7247350 \cdot 10^{-15}$	8	$27.81950727 \cdot 10^{-18}$		

$$P_{322}(T) = P(T, \rho_c) \quad [6.9]$$

Table 32. Polynomial coefficients A(1) - A(10) of $P_{322}(T)$ . $P_{322} = A_0 + A_1 \cdot T + A_2 \cdot T^2 + \dots$					
1	$-115.3411242 \cdot 10^6$	2	$-14.61390831 \cdot 10^3$	3	$713.8479434$
4	$-843.0259976 \cdot 10^{-3}$	5	$558.6511529 \cdot 10^{-6}$	6	$-228.9982666 \cdot 10^{-9}$
7	$59.25205508 \cdot 10^{-12}$	8	$-9.423230731 \cdot 10^{-15}$	9	$841.0373279 \cdot 10^{-21}$
10	$-32.24390358 \cdot 10^{-24}$				

$$P_{400}(T) = P(T, 400 \text{ kg/m}^3) \quad [6.10]$$

Table 33. Polynomial coefficients A(1) - A(10) of $P_{400}(T)$ . $P_{400} = A_0 + A_1 \cdot T + A_2 \cdot T^2 + \dots$					
1	$-48.22391500 \cdot 10^6$	2	$-534.0793594 \cdot 10^3$	3	$1.893391113 \cdot 10^3$
4	$-2.052518308$	5	$1.304520061 \cdot 10^{-3}$	6	$-521.2344512 \cdot 10^{-8}$
7	$132.5272444 \cdot 10^{-12}$	8	$-20.80959374 \cdot 10^{-15}$	9	$1.839373352 \cdot 10^{-18}$
10	$-69.98460931 \cdot 10^{-24}$				



$$P_{600}(T) = P(T, 600 \text{ kg/m}^3) \quad [6.11]$$

1	$-193.5182531 \cdot 10^6$	2	$-662.0824295 \cdot 10^3$	3	$3.041300733 \cdot 10^3$
4	$-3.331588396$	5	$2.091973967 \cdot 10^{-3}$	6	$-822.3881849 \cdot 10^{-9}$
7	$205.8600153 \cdot 10^{-12}$	8	$-31.88132588 \cdot 10^{-15}$	9	$2.784677333 \cdot 10^{-18}$
10	$-104.8772468 \cdot 10^{-24}$				

$$\text{and } P_{800}(T) = P(T, 800 \text{ kg/m}^3) \quad [6.12]$$

1	$-576.9907617 \cdot 10^6$	2	$229.0441320 \cdot 10^3$	3	$3.275463692 \cdot 10^3$
4	$-4.331244582$	5	$3.008662103 \cdot 10^{-3}$	6	$-1.272178994 \cdot 10^{-6}$
7	$337.9386420 \cdot 10^{-12}$	8	$-55.07102702 \cdot 10^{-15}$	9	$5.030253110 \cdot 10^{-18}$
10	$-197.1414663 \cdot 10^{-24}$				

( routines P250, P322, P400, P600 and P800 ). The above polynomials are valid in the temperature range

$$647.126 \text{ K} \leq T \leq 5000 \text{ K} \quad .$$

If now, in the beginnig DGFIND finds a zero for the initial density, then it will compare the supplied pressure with three pressure-windows :

$$0 \leq P \leq P_{322} \quad , \quad [6.13]$$

$$P_{322} \leq P \leq P_{600} \quad \text{and} \quad [6.14]$$

$$P_{600} \leq P \leq P_{Lmax} \quad [6.15]$$

to decide, in which of the three sub-areas to expect the density. If the pressure P lies in window [6.13] ( sub-area "LD" ), then DGFIND selects the following pivotal points

$$\begin{aligned} \rho_1 &= 0 \quad , \quad P_1 = 0 \\ \rho_2 &= 250 \text{ kg/m}^3 \quad , \quad P_2 = P_{250} \\ \rho_3 &= \rho_c \quad , \quad P_3 = P_{322} \end{aligned} \quad [6.16]$$

( s. Figure 33 ) and supplies them to the routine DINSU to get the initial density  $\rho_0$ . If P suffices the inequality [6.14] ( sub-area "MD" ), then the supplied corner-points will be

$$\begin{aligned} \rho_1 &= \rho_c \quad , \quad P_1 = P_{322} \\ \rho_2 &= 400 \text{ kg/m}^3 \quad , \quad P_2 = P_{400} \\ \rho_3 &= 600 \text{ kg/m}^3 \quad , \quad P_3 = P_{400} \end{aligned} \quad [6.17]$$

and if P lies inside the window [6.15] ( sub-area "HD" ), then DGFIND uses the points

$$\begin{aligned} \rho_1 &= 600 \text{ kg/m}^3 & , & & P_1 &= P_{600} \\ \rho_2 &= 800 \text{ kg/m}^3 & , & & P_2 &= P_{800} \\ \rho_3 &= 1200 \text{ kg/m}^3 & , & & P_3 &= P_{Lmax} \end{aligned} \quad [6.18]$$

to get an initial density,  $\rho_0$ .

DINSU calculates  $\rho_0$  the same way as DINVA or DINLI.



## 7.1 The codesystem AQUA

can be used in different ways :

1. the user can call the interactive procedure WATER, to get the water properties at a given  $(T, \rho)$  resp.  $(T, P)$  point displayed on the screen immediately,
2. it is also possible to acquire water properties for a whole string of points, either along a number of isotherms or along a set of isochores in a TSO-session by calling the procedure KAISOT or KAISOK,
3. finally, there are also the routines WASSER, DAMPF, RDRDP, SAET01 and VD1 which supply the water properties to the code IVA3 and which could be used by other codes too.

### SERVUS : auxiliary datasets and routines

The procedures KAISOT and KAISOK calculate the water properties along a string of density resp. temperature points. To be able to do that, they need

- a SERVUS-dataset to supply a sample isotherm resp. a sample isochore for these routines and to receive the calculated properties,
- some routines from the code SERVUS to read, inspect, handle, sort and store the functions appearing in the procedures.

The code SERVUS ( s. /19/) resides in the dataset **INR105.SERVUS.LOAD**.

The sample functions are either in the dataset **INR105.AQUA.DATA** or in **INR105.AQUAT.DATA**. AQUA holds sample isotherms with some 300 density points in the range  $\rho < 1200 \text{ kg/m}^3$ . In AQUAT there are temperature functions with 400 temperature points between the melting and the critical points.

The user can acquire the above datasets by copying and adapting the procedure **INR105.SERVUS.CNTL(IEBGENER)** to **userid.SERVUS.CNTL(IEBGENER)** (s. Figure 34 ).

```
//userida1 JOB (0000,999,abcde),dombrowsky,MSGLEVEL=(1,1),
// NOTIFY=userid,MSGCLASS=H
//IEBG EXEC PGM=IEBGENER
//SYSPRINT DD SYSOUT=*
//SYSUT1 DD UNIT=INR,DSNAME=INR105.AQUA.DATA,DISP=SHR
//SYSUT2 DD UNIT=iii,DSNAME=userid.AQUA.DATA,DISP=(,CATLG),
// SPACE=(TRK,(50,10),RLSE),DCB=(RECFM=VBS,LRECL=32760,BLKSIZE=19069)
//SYSIN DD DUMMY
//
```

Figure 34. The procedure **userid.SERVUS.CNTL(IEBGENER)**.

The "string"-routines KAISOT and KAISOK need, besides of the SERVUS routines also the AQUA-routines SIGNAT, NAPSZM and MUSTER.

- MUSTER helps in selecting the sample-function in the SERVUS dataset,
- SIGNAT sets the characteristic numbers and words - needed in registering the function in the dataset - for the properties calculated and
- NUMBER returns the number of day in the year.

## 7.2 WATER

WATER calculates the water properties either as pressure-temperature or as density-temperature functions. In the stable and metastable states WATER returns property-values, corresponding to the ones, calculated by the H.G.K.-code. In the region of instability WATER calculates mixed state properties, as described in 4.0, "Thermal properties in the two phase region."

WATER needs the "FIND"-routines for the  $(T, P) \Rightarrow (T, \rho)$  conversion and HYDOR to calculate the water properties in the  $(\rho, T)$  - dependence ( s. Appendix D, "Module trees" on page 77 ).

WATER resides in the dataset 'INR105.KATHER.LOAD' and can be activated by typing simply "WATER".

The following frozen screens exemplify a typical conversation with WATER :

```

----- TSO COMMAND PROCESSOR -----
ISPF COMMAND ===>

ENTER TSO COMMAND, CLIST, OR REXX EXEC BELOW:

===> WATER

```

```

+++++
+
+   W   W   A   TTTTTT EEEEEEE RRRRRR      STATUS : 94/12/09  +
+   W   W   AAA  TTTTTT EEEEEEE RRRRRR      +
+   W   W   AA AA   T   E   R   R      +
+   W W W AA  AA   T   E EEE  R RRRR      +
+   W WWW W A   A   T   E EEE  R RRRR      +
+   WWW WWW A AAA A   T   E   R  RR      +
+   WW  WW A AAA A   T   EEEEEEE R  RR      +
+   W   W A   A   T   EEEEEEE R  RR      +
+
+
+           WATER CALCULATES THERMAL PROPERTIES OF THE WATER  +
+           IN THE LIQUID, VAPOR AND GASEOUS STATES           +
+           EITHER AS FUNCTIONS OF TEMPERATURE AND DENSITY    +
+           OR AS FUNCTIONS OF TEMPERATURE AND PRESSURE.      +
+
+           THE PROPERTIES CALCULATED ARE THE FOLLOWING ONES:  +
+
+   T = TEMPERATURE           P = PRESSURE           RHO = DENSITY  +
+   CV / CP = ISOBARIC / ISOCHORIC HEAT CAPACITIES          S = ENTROPY      +
+   G = GIBBS FUNCTION       U = ENERGY             H = ENTHALPY    +
+   VEL.S = SONIC VELOCITY   SIGMA = SURFACE TENSION      ETA = VISCOSITY  +
+   LAMBDA = THERMAL CONDUCTIVITY                               PR.NR = PRANDTL NUMBER  +
+   VAP.QU = M.VAP/M.TOT                                         +
+
+
+           FOR MORE INFORMATION => K. THURNAY, T: 2477      +
+++++

```

ENTER TEMPERATURE ( IN K ) / 373.150 / =: ( STOP WITH T = 0 )  
00011 ?  
**373.15**

DO YOU WANT THE PROPERTIES AS  
1 = TEMPERATURE-DENSITY FUNCTIONS  
2 = TEMPERATURE-PRESSURE FUNCTIONS ( STOP WITH 0 )  
00014 ?  
**2**

ENTER PRESSURE ( IN MPASCAL ) / 0.1013250 / =: ( STOP WITH P = 0 )  
00019 ?  
**0.1013250**

LIQUID ( 1 ) OR VAPOR ( 2 ) STATE ? ( 0 = STOP ) =:  
00028 ?  
**1**

T = 373.150 K	RHO = 958.3926 KG/M**3	P = 0.1013458 MPASCAL
DP/DRHO = 2125.38 KJ/KG	DP/DT = 1.52786 MPASCAL/K	DRHO/DT = -0.718864 KG/(K*M**3)
CV = 3.77087 KJ/(KG*K)	CP = 4.21707 KJ/(KG*K)	S = 1.30688 KJ/(KG*K)
U = 418.958 KJ/KG	H = 419.064 KJ/KG	G = -68.5997 KJ/KG
DU/DRHO = -0.620588 KJ*M**3/KG**2	DH/DP = 0.751373E-03 M**3/KG	VEL.S = 1541.71 M/SEC
ETA = 0.282103E-03 KG/M/SEC	LAMBDA = 0.679215 W/M/K	SIGMA = 0.589152E-01 KG/SEC**2
PR.NR = 1.75150	VAP.QU = 0.352389E-07	PS = 0.1013127 MPASCAL
RH.V = 0.5975324 KG/M**3	RH.SV = 12.67164 KG/M**3	P.SV = 1.21103 MPASCAL
RH.L = 958.4467 KG/M**3	RH.SL = 823.0227 KG/M**3	P.SL = -149.037 MPASCAL

```

ENTER PRESSURE ( IN MPASCAL ) / 0.1013250 / =: ( STOP WITH P = 0 )
00019 ?
0

```

```

DO YOU WANT THE PROPERTIES AS
  1 = TEMPERATURE-DENSITY FUNCTIONS
  2 = TEMPERATURE-PRESSURE FUNCTIONS ( STOP WITH 0 )
00014 ?
0

```

```

ENTER TEMPERATURE ( IN K ) / 373.150 / =: ( STOP WITH T = 0 )
00011 ?
0

```

### 7.3 String-routines

The routines KAISOT and KAISOK are placed in the dataset 'INR105.AQUA.LOAD'

#### KAISOT

KAISOT calculates the water properties as density-temperature functions along isotherms.

KAISOT needs ( s. Appendix D, "Module trees" on page 77 ) :

- HYDOR for the calculation of the (  $\rho, T$  )-properties,
- the AQUA-routines SIGNAT, NAPSZM, MUSTER
- the SERVUS-routines SERDIO, SDEXG8, FENST8, SDING8
- the dataset **userid.AQUA.DATA** and
- the procedure **userid.AQUA.CLIST(KAISOT)** ( s. Figure 35 ).

```

ALLOC DA('userid.AQUA.DATA') F(FT15F001) SHR REU
ALLOC DA('userid.AQUA.DATA') F(FT16F001) SHR REU
CALL 'INR105.AQUA.LOAD(KAISOT)'
FREE DA('userid.AQUA.DATA')
FREE DA('userid.AQUA.DATA')

```

Figure 35. The procedure **userid.AQUA.CLIST(KAISOT)**

A sample conversation with KAISOT is something, like this :

READY  
 ex aqua(kaisot)

```

+++++
+
+ PROGRAM KAISOT STATUS : 94/11/30 +
+
+ CALCULATES THERMAL PROPERTIES OF THE WATER ALONG AN ISOTHERM +
+ AS FUNCTIONS OF THE DENSITY +
+ USING A SAMPLE (RH,FRH)-FUNCTION AS A DENSITY-PATTERN +
+
+++++
  
```

```

+++++
+
+ 12.12.94 LIST OF THE GRAPHS OF THE DATASET AQUA .DATA +
+
+ STORAGE-NORM IS "GRAPHIC8" +
+
+ DATE OF THE ALLOCATION: 23. 5.91 LAST SCRATCH: 27. 9.94 +
+
+ THE DATASET CONTAINS 275 GRAPHS TIME: 16H 57' 11" +
+
+++++
  
```

```

+++++
+
+ 12.12.94 THE GRAPHS WILL BE COPIED INTO THE DATASET AQUA .DATA +
+
+ STORAGE-NORM IS "GRAPHIC8" +
+
+ DATE OF THE ALLOCATION: 23. 5.91 LAST SCRATCH: 27. 9.94 +
+
+ THE DATASET CONTAINS 275 GRAPHS TIME: 16H 57' 11" +
+
+++++
  
```

ENTER TEMPERATURE OF THE ISOTHERM =: ( 0 = END )  
 00016 ?  
 373.15



TABLE OF THE PROPERTY-NUMBERS : ( END = 0 )

	1	2	3	4	5	6	7	8
0	P*	P*	DP/DT	DT/DP	T*			
10	RH	RH	RH.SP		P	P.SP	V	Z
20	P1R	P1T	R1P	R1T		CV	CP	
30	V.S	SIG		ETA	LAMBDA	PRN		
40	A	G	U	H	S	U1R	H1P	

ENTER NUMBER OF THE PROPERTY / 0 / =:

00012 ?

26

ENTER SERIAL NUMBER OF THE (T,FT) MASTER FUNCTION =: ( 0 = END )

00008 ?

22

# 22 GRAPH(1015/ 5000) X : +DENSITY +KG/M\*\*3 + F : + P.T+PASCAL +  
X1= 0.5550E-03,X(293)= 1190. , 0.0000E+00 < X < 1200. , U=0.10E+03  
F1= 128.1 ,F(293)= 0.1283E+10 , -0.6100E+08 < F < 0.1300E+10 , U=0.10E+09  
FUNCTION OK =: ( 0 = YES | 9 = NO )

00012 ?

0

ENTER BEGIN, END & INCREMENTUM ( 0 = END ) / 1 ,1000 , 1 / =: , =: , =:

00016 ?

1 300 1

FUNCTION TO BE SAVED ? ( 0 = NO ) =:

00024 ?

3

==> 173 GRAPH( 26/ 3731) X : +DENSITY +KG/M\*\*3 + F : + CV.ITH+J/KG/K +  
X1= 0.5550E-03,X(293)= 1190. , 0.0000E+00 < X < 1200. , U=0.10E+03  
F1= 1428. ,F(293)= 3412. , 1400. < F < 7000. , U=0.10E+04

TABLE OF THE PROPERTY-NUMBERS : ( END = 0 )

	1	2	3	4	5	6	7	8
0	P*	P*~	DP/DT	DT/DP	T*			
10	RH	RH~	RH.SP		P	P.SP	V	Z
20	P1R	P1T	R1P	R1T		CV	CP	
30	V.S	SIG		ETA	LAMBDA	PRN		
40	A	G	U	H	S	U1R	H1P	

ENTER NUMBER OF THE PROPERTY / 0 / =:

00012 ?

0

ENTER TEMPERATURE OF THE ISOTHERM =: ( 0 = END )

00016 ?

0

## KAISOK

KAISOK calculates the water properties as density-temperature functions along isochores.

KAISOK uses, similarly to KAISOT :

- HYDOR for the calculation of the (  $\rho, T$  )-properties,
- the AQUA-routines SIGNAT, NAPSZM, MUSTER
- the SERVUS-routines SERDIO, SDEXG8, FENST8, SDING8
- the dataset *userid.AQUAT.DATA* and
- the procedure *userid.AQUA.CLIST(KAISOK)* ( s. Figure 35 ).

```

ALLOC DA('userid.AQUAT.DATA') F(FT15F001) SHR REU
ALLOC DA('userid.AQUAT.DATA') F(FT16F001) SHR REU
CALL 'INR105.AQUA.LOAD(KAISOK)'
FREE DA('userid.AQUAT.DATA')
FREE DA('userid.AQUAT.DATA')

```

Figure 36. The procedure *userid.AQUA.CLIST(KAISOK)*

A dialogue with KAISOK runs in the same way, as one with KAISOT.

```

READY
ex aqua(kaisok)

```

etc.

## 7.4 IVA-routines

A set of the following routines - completed with all the other AQUA-routines needed by them - are deposited in the dataset **INR105.IVHZO.FORT**.

### WASSER

WASSER calculates thermophysical properties of the liquid water as functions of the temperature and pressure for the codesystem IVA3. WASSER calculates only in the stable and metastable states of the liquid, so the supplied P must be higher as the spinodal-pressure of the liquid. In calling WASSER

```
CALL WASSER(T,P
            ,RW,HW,SW,CPW,VSW,ETW,XLMBW,PRNW,HW1P,RW1T,RW1P,SRFTS
            ,LCATAW)
```

1. T, the temperature and
2. P, the pressure  
should be supplied. WASSER then returns
3. RW, the density,
4. HW, the specific enthalpy,
5. SW, the specific entropy,
6. CPW, the heat capacity at constant pressure,
7. VSW, the velocity of sound,
8. ETW, the viscosity,
9. XLMBW, the thermal conductivity,
10. PRND, the Prandtl number of the water ,
11. HW1P, the pressure derivative of the enthalpy,  $\partial H/\partial P$ ,
12. RW1T, the temperature derivative of the density,  $\partial \rho/\partial T$ ,
13. RW1P, the pressure derivative of the density,  $\partial \rho/\partial P$  and
14. SRFTS, the surface tension.

All of the above properties are in SI-units.

LCATAW, a logical variable will be set .true. if WASSER fails to find the density  $\rho(P,T)$ .

### DAMPF

DAMPF calculates thermophysical properties of the vaporized water as functions of the temperature and pressure for the codesystem IVA3. DAMPF calculates only in the stable and metastable vapor states, so the supplied P must be lower as the spinodal-pressure of the vapor. DAMPF

```
CALL DAMPF(T,P
            ,RGAS,WGM,RD,HD,SD,CPD,VSD,ETD,XLMBD,PRND,HD1P,RD1T,RD1P
            ,LCATAD)
```

needs initial values for

1. T, the temperature and
  2. P, the pressure
- and it returns the following properties :
3. RGAS, the gas-law constant of the water,
  4. WGM, the molal mass,
  5. RD, the density,
  6. HD, the specific enthalpy,
  7. SD, the specific entropy,
  8. CPD, the heat capacity at constant pressure,
  9. VSD, the velocity of sound,
  10. ETD, the viscosity,
  11. XLMBD, the thermal conductivity,
  12. PRND, the Prandtl number of the steam
  13. HD1P, the pressure derivative of the enthalpy,  $\partial H/\partial P$ ,
  14. RD1T, the temperature derivative of the density,  $\partial \rho/\partial T$  and
  15. RD1P, the pressure derivative of the density,  $\partial \rho/\partial P$ .

All of the above properties - as in the case of WASSER - are in SI-units.  
 LCATAD will be again set .true. if DAMPF fails to find the density RD(P,T).

### RDRDP

RDRDP is an abridged version of DAMPF, only to find the density corresponding to the pressure P and the temperature T in the vapor state for the code IVA3.

```
CALL RDRDP(T,P, RD,RD1P, LCATAD)
```

For a given T and P RDRDP returns RD, the density and RD1P, the pressure derivative of the density (  $\partial \rho/\partial P$  ) of the steam.

### SAET01

```
CALL SAET01(T,P
            ,VW,VD,HW,HD,HWD,SW,SD,TS1P,CPW,CPD,ETW,ETD,XLMBW,XLMBD,SRFTS
            ,KIND,LCATAS)
```

SAET01 calculates saturation properties of the water for IVA3 as function of the pressure P ( if KIND is 1 ) or the temperature T ( if KIND equals 0 ).

SAET01 returns - besides of T or P - the following thermophysical properties :

- VW and VD are the specific volume of the liquid resp. of the vapor,
- HW and HD are the corresponding specific enthalpies,
- SW and SD are the specific entropies in the liquid and vapor states,
- CPW and CPD are the corresponding heat capacities at constant pressure,
- ETW and ETD are the viscosities of the liquid resp. of the vapor,
- XLMBW and XLMBD are the corresponding thermal conductivities,
- SRFTS is the surface tension ( of the liquid ),
- HWD = HD - HW and

- $TS1P = 1 / \partial P / \partial T$ .

All the properties are again in SI-units.

LCATAS will be set .true. if SAET01 fails to find saturated properties ( i. e. if TS is supercritical ).

**Note:** There exists also an extended version of the above routine, SAETEX, created as a testing facility for the routine SAET01. Besides of the properties listed above, SAETEX calculates also the pressure derivatives  $dP^*/dT$ ,  $\partial P / \partial T$  and  $\partial P / \partial \rho$ , the constant volume heat capacities  $C_v$  and the sonic velocities.

## VD1

VD1 calculates approximated specific volumes for the saturated vapor, VD in  $kg/m^3$ -s at the temperature T for the code IVA3.

```
CALL VD1(T, VD, DUMMY)
```

For a complete list of available water property procedures see Appendix M, "TSO-procedures, calculating thermal properties of the water" on page 153.

## 8.0 References

- /1/ V. V. Sychev, Character of the Change of Thermodynamic Quantities on Boundary Curves  
High Temperature, Vol. 5, No. 6, p. 1022, 1967.
- /2/ Kh I. Amirkhanov, B. G. Alibekov, B. A. Mursalov, and G. V. Stepanov, Calculation of the Derivatives of Thermal and Caloric Quantities on a Saturation Line  
High Temperature, Vol. 10, No. 3, p. 475, 1972.
- /3/ V. P. Skripov, Metastable Liquids, English translation  
John Wiley and sons, New York, 1974.
- /4/ J. V. Sengers, Thermal Conductivity of Steam in the Critical Region, P. 822 in  
Proceedings of the Seventh Symposium on Thermophysical Properties, ASME, 1977.
- /5/ K.Thurnay, Thermophysical Properties of Sodium in the Liquid and Gaseous States,  
KFK 2863, Februar 1981.
- /6/ L. A. Weber, Measurements of the heat capacities  $C_V$  of dense gaseous and liquid nitrogen and nitrogen trifluoride.  
J. Chem. Thermodynamics, Vol. 13, p. 389-403, 1981.
- /7/ A. M. Sirota, V. I. Latunin, N. E. Nikolaeva, An Experimental Investigation of the Thermal Conductivity Maxima of Water in the Critical Region  
Thermal Engineering Vol. 28, No. 4, p. 246, 1981.
- /8/ K.Thurnay, Evaluation of Thermophysical Properties of Sodium as Surfaces of the Temperature and the Density,  
Nucl. Sci. Eng., 82, p. 181, 1982.
- /9/ C. A. Meyer, R. B. McClintock, G. J. Silvestri, R. C. Spencer, Jr., ASME steam tables : thermodynamic and transport properties of steam comprising tables and charts for steam and water, 5. ed.  
American Soc. of Mechanical Engineers, New York, 1983.
- /10/ L. Haar, J. S. Gallagher, G. S. Kell, NBS/NRC steam tables : thermodynamic and transport properties and computer programs for vapor and liquid states of water in SI units  
Hemisphere Publ. Co., Washington, 1984.
- /11/ J. V. Sengers and B. Kamgar Parsi, Representative Equations for the Viscosity of Water Substance  
J. Phys. Chem. Ref. Data, Vol. 13, No. 1, p. 185, 1984.
- /12/ H. J. White, Jr., Activities of the International Association for the Properties of Steam Between 1979 and 1984, pp. 31 - 34 in  
V. V. Sytchev ( ed. ) The properties of steam  
Proc. of the 10th internat. conf., Moscow, USSR, 3 - 7 Sept. 1984.  
Moscow : Mir Publishers, London : Plenum Press, 1984.
- /13/ L. Haar and J. S. Gallagher, Thermodynamic Values Near the Critical Point of Water, pp. 167 - 179 in  
V. V. Sytchev ( ed. ) The properties of steam  
Proc. of the 10th internat. conf., Moscow, USSR, 3 - 7 Sept. 1984.  
Moscow : Mir Publishers, London : Plenum Press, 1984.
- /14/ W. Wagner and A. Saul, Correlation Equations for the Vapor Pressure and for the Orthobaric Densities of Water Substance, pp. 199 - 209 in  
V. V. Sytchev ( ed. ) The properties of steam  
Proc. of the 10th internat. conf., Moscow, USSR, 3 - 7 Sept. 1984.  
Moscow : Mir Publishers, London : Plenum Press, 1984.
- /15/ V. P. Skripov, Phase-Change Behaviour of Water in Nonequilibrium Processes, pp. 227 - 251 in  
V. V. Sytchev ( ed. ) The properties of steam  
Proc. of the 10th internat. conf., Moscow, USSR, 3 - 7 Sept. 1984.  
Moscow : Mir Publishers, London : Plenum Press, 1984.

- /16/ J. V. Sengers, Transport Properties of Fluids Near Critical Points, pp. 363 - 389 in  
V. V. Sytchev ( ed. ) The properties of steam  
Proc. of the 10th internat. conf., Moscow, USSR, 3 - 7 Sept. 1984.  
Moscow : Mir Publishers, London : Plenum Press, 1984.
- /17/ V. V. Sychev, V. A. Rabinovich, and Yu. E. Sheludyak, Using Differential Thermodynamic Equations at the Critical Point of Pure Materials  
High Temperature, Vol. **22**, No. 6, p. 829, 1985.
- /18/ P. W. Atkins, Physical chemistry, 3. ed.  
Oxford University Press, 1986.
- /19/ K. Thurnay, SERVUS. Ein System von Dateien und Prozeduren zur Speicherung und Präsentation von Rechenergebnissen, die Funktionen ( einer Variablen ) sind.  
Version V, April 1989.  
Kfk 4587, Juli 1989.
- /20/ K. Thurnay, SODIUM. A Code for Calculating Thermophysical Properties of the Sodium in the Liquid and Gaseous States.  
KfK 4609, September 1989.
- /21/ N. I. Kolev, IVA3 : Computer Code For Modeling of Transient Three Dimensional Three Phase Flow in Complicated Geometry  
Kfk 4950, Dezember 1991.
- /22/ V. P. Skripov, Metastable States  
J. Non-Equilib. Thermodyn., Vol. **17**, No. 3, p. 193 - 206, 1992.
- /23/ C. A. Meyer, R. B. McClintock, G. J. Silvestri, R. C. Spencer, Jr., ASME Steam Tables : Thermodynamic and Transport Properties of Steam Comprising Tables and Charts For Steam and Water, 6. ed.  
New York, NY : American Soc. of Mechanical Engineers, 1993.

## Appendix A. Some thermal relations

### A.1 The derivatives of the free energy

In developing the thermal properties of a substance as functions of the **temperature and density** the appropriate energy property to begin with is the free energy :

$$A \equiv U - S \cdot T \quad [A.1]$$

also known as Helmholtz-function or total work.

Since it holds ( s. e. g. /18/ )

$$dU = T \cdot dS - P \cdot dV \quad [A.2]$$

one has for A the following ( $\rho, T$ )-differential :

$$dA(\rho, T) = dU - S dT - T dS = -S dT + \frac{P}{\rho^2} d\rho \quad . \quad [A.3]$$

The differential [A.3] gives following relations for entropy and pressure :

$$S(\rho, T) = - \frac{\partial A}{\partial T} \quad , \quad [A.4]$$

$$P(\rho, T) = \frac{\partial A}{\partial \rho} \rho^2 \quad . \quad [A.5]$$

### A.2 The enthalpy and its derivatives

For the enthalpy one can also apply a ( $d\rho, dT$ ) - expression :

$$dH(\rho, T) = \frac{\partial H}{\partial T} dT + \frac{\partial H}{\partial \rho} d\rho \quad . \quad [A.6]$$

The T-derivative in the Eq. [A.6] can be calculated from the equation

$$H \equiv U + \frac{P}{\rho} \quad [A.7]$$

as

$$\left. \frac{\partial H}{\partial T} \right|_{\rho} = C_V + \frac{1}{\rho} \frac{\partial P}{\partial T} \quad [A.8]$$

and the  $\rho$ -derivative from the equation

$$H \equiv A + \frac{P}{\rho} + S \cdot T \quad [A.9]$$

transformed with the relations [A.4] and [A.5] properly. The result is



$$\left. \frac{\partial H}{\partial \rho} \right|_T = \frac{1}{\rho} \left( \frac{\partial P}{\partial \rho} - \frac{T}{\rho} \frac{\partial P}{\partial T} \right) . \quad [\text{A.10}]$$

Setting the pressure-differential

$$dP(\rho, T) = \frac{\partial P}{\partial T} dT + \frac{\partial P}{\partial \rho} d\rho$$

to zero gives the important relation

$$\frac{\partial P / \partial T}{\partial P / \partial \rho} = - \left. \frac{\partial \rho}{\partial T} \right|_P . \quad [\text{A.11}]$$

Using this relation and the reduced temperature-derivative of the density

$$\rho_T^0 \equiv \left. \frac{T}{\rho} \frac{\partial \rho}{\partial T} \right|_P \quad [\text{A.12}]$$

allows to describe the density derivative [A.10] in a more simple form

$$\left. \frac{\partial H}{\partial \rho} \right|_T = \frac{1}{\rho} \frac{\partial P}{\partial \rho} (1 + \rho_T^0) . \quad [\text{A.13}]$$

To develop the derivatives of the **(P,T)-description** the differential [A.6] is to be transformed accordingly. Using a (P,T)-differential for the density results in

$$\begin{aligned} dH &= \frac{\partial H}{\partial T} dT + \frac{\partial H}{\partial \rho} \left( \frac{\partial \rho}{\partial T} dT + \frac{\partial \rho}{\partial P} dP \right) = \\ &= \left( \left. \frac{\partial H}{\partial T} \right|_\rho + \left. \frac{\partial H}{\partial \rho} \right|_T \frac{\partial \rho}{\partial T} \right) dT + \left. \frac{\partial H}{\partial \rho} \right|_T \frac{\partial \rho}{\partial P} dP . \end{aligned} \quad [\text{A.14}]$$

Eq. [A.14] combined with Eqs. [A.8], [A.11] and [A.13] results in the following descriptions for the (P,T)-derivatives of the enthalpy :

$$\begin{aligned} \left. \frac{\partial H}{\partial T} \right|_P &\equiv C_P(P, T) = \left. \frac{\partial H}{\partial T} \right|_\rho + \left. \frac{\partial H}{\partial \rho} \right|_T \left. \frac{\partial \rho}{\partial T} \right|_P = \\ &= C_V(\rho, T) + \frac{T}{\partial P / \partial \rho} \left( \frac{1}{\rho} \frac{\partial P}{\partial T} \right)^2 , \end{aligned} \quad [\text{A.15}]$$

respectively

$$\left. \frac{\partial H}{\partial P} \right|_T = \left. \frac{\partial H}{\partial \rho} \right|_T \left. \frac{\partial \rho}{\partial P} \right|_T = \frac{1}{\rho} (1 + \rho_T^0) . \quad [\text{A.16}]$$

This last derivative is also known as the 'isothermal factor'

$$\left. \frac{\partial H}{\partial P} \right|_T = V - T \cdot \left. \frac{\partial V}{\partial T} \right|_P . \quad [\text{A.17}]$$

### A.3 The sonic velocity

is defined as

$$v_s^2 = \left. \frac{\partial P}{\partial \rho} \right|_s \quad . \quad [\text{A.18}]$$

To develop the derivative in the equation above one needs the following differential

$$dS(\rho, P) = \frac{\partial S}{\partial T} \left( \frac{\partial T}{\partial \rho} d\rho + \frac{\partial T}{\partial P} dP \right) + \frac{\partial S}{\partial \rho} d\rho \quad . \quad [\text{A.19}]$$

Eq. [A.19] can be transformed to

$$dS(\rho, P) = \frac{\partial S / \partial T}{\partial P / \partial T} dP + \left( \frac{\partial S}{\partial \rho} + \frac{\partial S / \partial T}{\partial \rho / \partial T} \right) d\rho \quad [\text{A.20}]$$

Setting  $dS = 0$  in the above equation and using Eq. [A.11] gives

$$\left. \frac{\partial P}{\partial \rho} \right|_s = - \frac{\left( \frac{\partial S}{\partial \rho} + \frac{\partial S / \partial T}{\partial \rho / \partial T} \right)}{\frac{\partial S / \partial T}{\partial P / \partial T}} = \frac{\partial P}{\partial \rho} - \frac{\partial P}{\partial T} \frac{\partial S / \partial \rho}{\partial S / \partial T} \quad . \quad [\text{A.21}]$$

Differentiating the equations Eq. [A.1] respectively Eq. [A.4] result in the entropy-derivatives

$$\frac{\partial S}{\partial T} = \frac{C_V}{T} \quad , \quad \frac{\partial S}{\partial \rho} = - \frac{1}{\rho^2} \frac{\partial P}{\partial T} \quad . \quad [\text{A.22}]$$

Setting the above derivatives into the Eq. [A.21] and comparing the result with the Eq. [A.15] ends in

$$\left. \frac{\partial P}{\partial \rho} \right|_s = \frac{\partial P}{\partial \rho} + \frac{T}{C_V} \left( \frac{\partial P / \partial T}{\rho} \right)^2 = \frac{\partial P}{\partial \rho} \frac{C_P}{C_V} \quad . \quad [\text{A.23}]$$

Related to the sonic velocity are the adiabatic resp. the isothermal compressibilities :

$$\kappa_S = \frac{1}{\rho \cdot \left. \partial P / \partial \rho \right|_s} = \frac{1}{\rho \cdot v_s^2} \quad , \quad [\text{A.24}]$$

$$\kappa_T = \frac{1}{\rho \cdot \partial P / \partial \rho} = \frac{C_P}{C_V} \cdot \kappa_S \quad . \quad [\text{A.25}]$$

#### A.4 The Joule Thomson coefficient

This property is defined as

$$\text{Joule Thomson coefficient} \equiv \left. \frac{\partial T}{\partial P} \right|_H \quad [\text{A.26}]$$

Using the restraint

$$dH = \frac{\partial H}{\partial P} dP + \frac{\partial H}{\partial T} dT = 0$$

and the result of Eq. [A.16] one has

$$\left. \frac{\partial T}{\partial P} \right|_H = - \frac{\frac{\partial H}{\partial P}}{\frac{\partial H}{\partial T}} = - \frac{\frac{\partial H}{\partial P}}{C_P} = - \frac{1 + \rho_T^0}{\rho \cdot C_P} \quad [\text{A.27}]$$

## Appendix B. Derivatives of some saturated properties

### B.1 Derivatives of the vapor pressure

Since the derivative of  $w$  in the Eq. [5.2] can be expressed as

$$\frac{dw}{dT} = \frac{-0.5}{T_c} \left(1 - \frac{1}{T_c}\right)^{-0.5} = \frac{-0.5}{w \cdot T_c} \quad [B.1]$$

it holds for the first T-derivative of the vapor pressure

$$\begin{aligned} Q_1 &\equiv \frac{T}{P^x} \frac{dP^x}{dT} = T \frac{d \ln P^x}{dT} = \\ &= -\frac{T_c}{T} \cdot \sum_{j=1}^{13} A_j \cdot w^{j-1} + \left(\frac{T_c}{T} - 1\right) \sum_{j=1}^8 T \cdot \frac{dw}{dT} \cdot (j-1) \cdot A_j \cdot w^{j-2} = \\ &= -\sum_{j=1}^8 A_j \cdot w^{j-1} \left[ \frac{T_c}{T} + \frac{j-1}{2} \right] \end{aligned} \quad [B.2]$$

The derivative of  $Q_1$  is

$$\begin{aligned} Q_2 &\equiv T \frac{dQ_1}{dT} = \\ &= \sum_{j=1}^8 A_j \cdot w^{j-1} \left[ \frac{T_c}{T} + \frac{j-1}{2} \left(1 + \frac{j+1}{2} \frac{T}{T_c - T}\right) \right] \end{aligned} \quad [B.3]$$

On the other hand one can develop a pressure relation for  $Q_2$  :

$$\begin{aligned} Q_2 &\equiv T \frac{d}{dT} \left[ \frac{T}{P^x} \frac{dP^x}{dT} \right] = \\ &= \frac{T}{P^x} \frac{dP^x}{dT} + \frac{T^2}{P^x} \frac{d^2P^x}{dT^2} - \left[ \frac{T}{P^x} \frac{dP^x}{dT} \right]^2 = \\ &= \frac{T^2}{P^x} \frac{d^2P^x}{dT^2} + Q_1(1 - Q_1) \end{aligned} \quad [B.4]$$

Transforming Eq. [B.2] and [B.4] result in the following equations for the temperature derivatives of the vapor pressure :

$$\frac{dP^x}{dT} = \frac{P^x}{T} Q_1 \quad \text{and} \quad [B.5]$$

$$\frac{d^2P^x}{dT^2} = \frac{P^x}{T^2} [Q_2 - Q_1(1 - Q_1)] \quad [B.6]$$

To ease the descriptions of the Q-s the coefficients

$$B_j \equiv \frac{j-1}{2} A_j \quad \text{and} \quad [B.7]$$

$$C_j \equiv \frac{j+1}{2} B_j \quad [B.8]$$

will be introduced, resulting in the Q-equations

$$Q_1 = -\frac{T_c}{T} \sum_{j=1}^8 A_j \cdot w^{j-1} - \sum_{j=1}^8 B_j \cdot w^{j-1} \quad [B.9]$$

and

$$Q_2 = \frac{T_c}{T} \sum_{j=1}^8 A_j \cdot w^{j-1} + \sum_{j=1}^8 B_j \cdot w^{j-1} + \frac{T}{T_c - T} \sum_{j=1}^8 C_j \cdot w^{j-1} \quad [B.10]$$

## B.2 Derivatives of the saturated densities

For

**the cold water** one can develop from Eq. [5.9] with Eq. [B.1] the derivative

$$\frac{d\rho_L}{dT} = -\frac{0.5 \cdot \rho_c}{T_c} \sum_{j=2}^{12} (j-1) \cdot A_j \cdot w^{j-3} \quad [B.11]$$

Correspondingly the second derivative is here

$$\frac{d^2\rho_L}{dT^2} = \left[ \frac{0.5}{T_c} \right]^2 \cdot \rho_c \sum_{j=2}^{12} (j-1) \cdot (j-3) \cdot A_j \cdot w^{j-5} \quad [B.12]$$

To develop the density derivatives for

**the cold steam** is a bit more complicated, since here is not the density, but the factor of reality given by a polynomial ( s. Eq. [5.10] ).

From the gas law

$$P^x = \rho_V \cdot R_{gas} \cdot T \cdot Z_V$$

one has the following relations between the first and second derivatives :

$$\frac{T}{P^x} \frac{dP^x}{dT} = 1 + \frac{T}{\rho_V} \frac{d\rho_V}{dT} + \frac{T}{Z_V} \frac{dZ_V}{dT} \quad [B.13]$$

and

$$\begin{aligned} & \frac{T^2}{P^x} \frac{d^2 P^x}{dT^2} - 2 \cdot \frac{T}{P^x} \frac{dP^x}{dT} + 2 = \\ & = \frac{T^2}{\rho_V} \frac{d^2 \rho_V}{dT^2} + 2 \cdot \frac{T}{\rho_V} \frac{d\rho_V}{dT} \cdot \frac{T}{Z_V} \frac{dZ_V}{dT} + \frac{T^2}{Z_V} \frac{d^2 Z_V}{dT^2} \end{aligned} \quad [B.14]$$

Since the Z-derivatives are

$$\frac{dZ_V}{dT} = -\frac{0.5}{T_c} \sum_{j=2}^{12} (j-1) \cdot Z_j \cdot w^{j-3} \quad \text{and} \quad [B.15]$$

$$\frac{d^2 Z_V}{dT^2} = \left[ \frac{0.5}{T_c} \right]^2 \sum_{j=2}^{12} (j-1) \cdot (j-3) \cdot Z_j \cdot w^{j-5} \quad [B.16]$$

so the density derivatives in the cold steam can be described as

$$\frac{d\rho_V}{dT} = \frac{\rho_V}{T} \left[ Q_1 - 1 + \frac{0.5 \cdot T}{Z_V \cdot T_c} \sum_{j=2}^{12} (j-1) \cdot Z_j \cdot w^{j-3} \right] \quad [B.17]$$

$$\begin{aligned} \frac{d^2 \rho_V}{dT^2} &= \frac{\rho_V}{T^2} [Q_1^2 - 3Q_1 + Q_2 + 2] + \\ &+ \frac{1}{Z_V \cdot T_c} \left[ \frac{d\rho_V}{dT} \sum_{j=2}^{12} (j-1) \cdot Z_j \cdot w^{j-3} + \right. \\ &\quad \left. - 0.25 \frac{\rho_V}{T_c} \sum_{j=2}^{12} (j-1) \cdot (j-3) \cdot Z_j \cdot w^{j-5} \right] \end{aligned} \quad [B.18]$$

**In the vicinity of the critical point** AQUA describes the densities as

$$\begin{aligned} \rho(T) &= \rho_c \cdot \omega(\tau) \\ \omega(\tau) &\equiv 1 + B\tau^\beta + \tau [U_1 + \tau (U_2 + \tau^2 U_4)] \end{aligned} \quad [B.19]$$

with  $B = 2.03$  for the liquid and  $B = -2.03$  for the vapor. The  $\tau$ -derivative is

$$\frac{d\tau}{dT} = -\frac{\rho_c}{T_c} \quad [B.20]$$

so the first density derivatives will be

$$\frac{d\rho}{dT} = -\frac{\rho_c}{T_c} \frac{d\omega}{d\tau} = -\frac{\rho_c \cdot g}{T_c - T} \quad [B.21]$$

the function  $g$  being

$$g \equiv \tau \frac{d\omega}{d\tau} = B\tau^\beta \beta + \tau [U_1 + \tau(2U_2 + \tau^2 4U_4)] \quad . \quad [B.22]$$

The second density-derivatives one can develop as

$$\frac{d^2\rho}{dT^2} = \frac{\rho_c}{T_c^2} \frac{d^2\omega}{d\tau^2} = \frac{\rho_c}{(\tau \cdot T_c)^2} \left[ \tau \frac{dg}{d\tau} - g \right] = \frac{\rho_c \cdot h}{(T_c - T)^2} \quad [B.23]$$

with

$$h \equiv \tau \frac{dg}{d\tau} - g = B\tau^\beta \beta(\beta - 1) + \tau^2(2U_2 + \tau^2 12U_4) \quad . \quad [B.24]$$

Requesting smooth transitions at the switching point for the density and its first and second derivatives leads to the following system of equations for the coefficients U :

$$\begin{aligned} U_4 &= \frac{2(o - g) + h - (1 - \beta)(2 - \beta)b}{6\tau^4} \\ U_2 &= - \frac{4(o - g) + h - (1 - \beta)(4 - \beta)b}{2\tau^2} \\ U_1 &= \frac{8o - 5g + h - (2 - \beta)(4 - \beta)b}{3\tau} \end{aligned} \quad [B.25]$$

with the abbreviations

$$b = B\tau^\beta \quad \text{and} \quad o = \frac{\rho - \rho_c}{\rho_c} \quad . \quad [B.26]$$

## Appendix C. Adjusting the saturated states

To be in thermal equilibrium the pressures and the Gibbs-energies of the saturated liquid and the saturated vapor must be in accordance with each other :

$$G_L = G_V \quad \text{and} \quad P_L = P_V . \quad [\text{C.1}]$$

For the Gibbs-function

$$G \equiv A + \frac{P}{\rho} \quad [\text{C.2}]$$

along an isotherm it holds :

$$dG|_T = d\rho \cdot \frac{\partial G}{\partial \rho} \quad [\text{C.3}]$$

with

$$\frac{\partial G}{\partial \rho} = \frac{\partial A}{\partial \rho} - \frac{P}{\rho^2} + \frac{1}{\rho} \frac{\partial P}{\partial \rho} = \frac{1}{\rho} \frac{\partial P}{\partial \rho} . \quad [\text{C.4}]$$

Suppose the liquid and the vapor state differ from each other by  $\Delta P = P_L - P_V$  and  $\Delta G = G_L - G_V$ . To get rid of these differences one needs small changes in the liquid density,  $x$  and in the vapor density,  $y$ .

The density changes,  $x$  and  $y$  can be calculated as follows : The pressure difference can be transformed to

$$\Delta P = P_L - P^x + P^x - P_V = -\Delta P_L + \Delta P_V = -x \frac{\partial P}{\partial \rho_L} + y \frac{\partial P}{\partial \rho_V} \quad [\text{C.5}]$$

with  $\Delta P_L = P^x - P_L$  and  $\Delta P_V = P^x - P_V$ . Correspondingly one has for the G-s

$$\Delta G = G_L - G^x + G^x - G_V = -\Delta G_L + \Delta G_V = -\frac{x}{\rho_L} \frac{\partial P}{\partial \rho_L} + \frac{y}{\rho_V} \frac{\partial P}{\partial \rho_V} . \quad [\text{C.6}]$$

The Eqs. [C.5]- [C.6] can be solved for  $x$  and  $y$  :

$$\begin{aligned} x &= \frac{\rho_L}{\rho_L - \rho_V} \cdot \frac{\Delta G \cdot \rho_V - \Delta P}{\partial P / \partial \rho_L} , \\ y &= \frac{\rho_V}{\rho_L - \rho_V} \cdot \frac{\Delta G \cdot \rho_L - \Delta P}{\partial P / \partial \rho_V} . \end{aligned} \quad [\text{C.7}]$$

Using the properties in their reduced forms leads to :

$$\begin{aligned} x &= \frac{\rho_L}{\rho_L - \rho_V} \cdot \frac{\rho_V (\Delta G^0 + Z_V) - \rho_L \cdot Z_L}{PR_L^0} , \\ y &= \frac{\rho_V}{\rho_L - \rho_V} \cdot \frac{\rho_L (\Delta G^0 - Z_L) + \rho_V \cdot Z_V}{PR_V^0} . \end{aligned} \quad [\text{C.8}]$$





**D.1 Interactive routines**

**WATER**

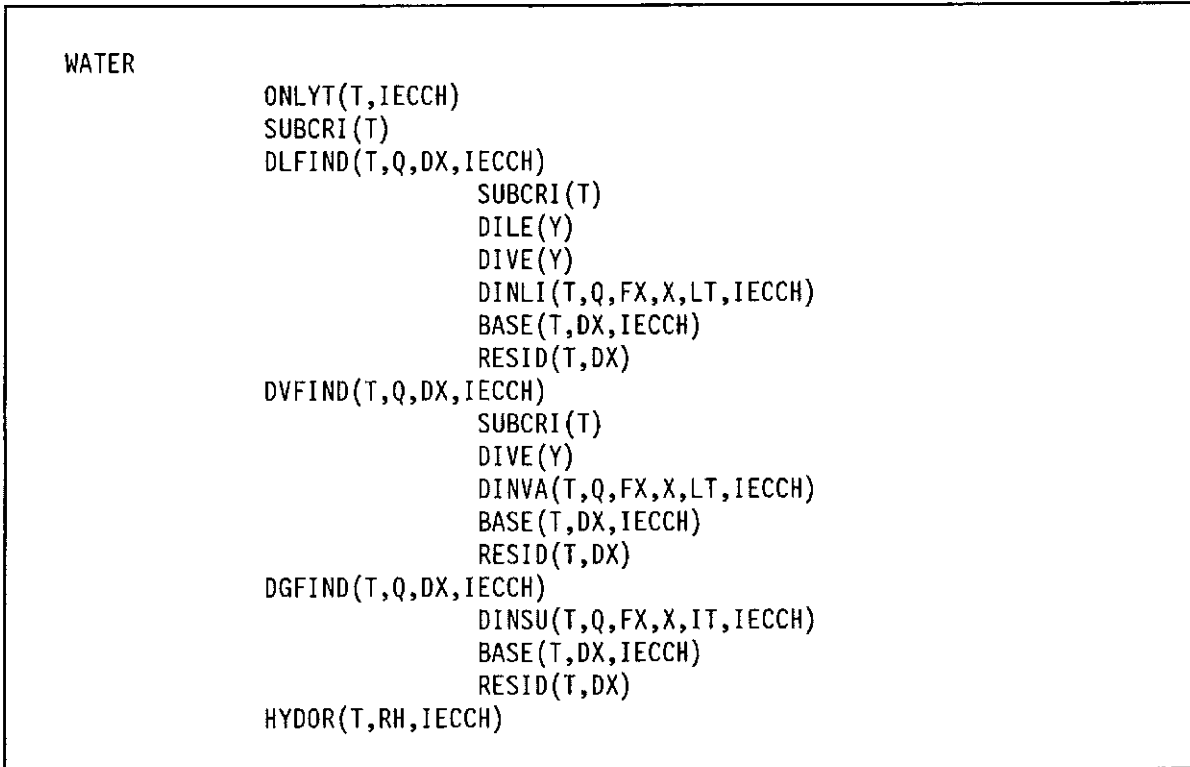


Figure 37. Module tree for the procedure WATER

## KAISOT

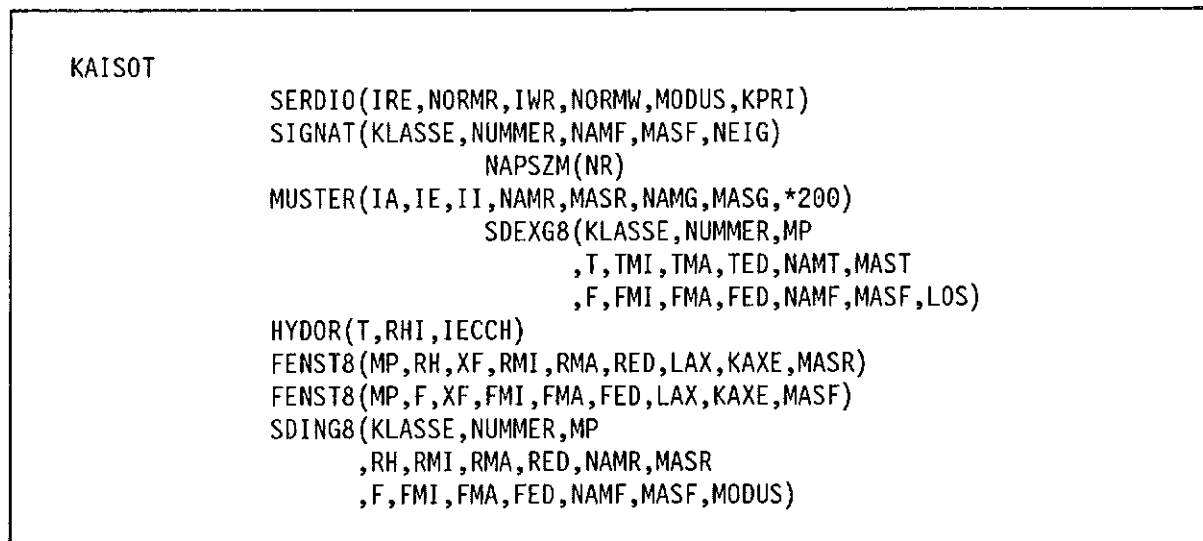


Figure 38. Module tree for the procedure KAISOT

## KAISOK

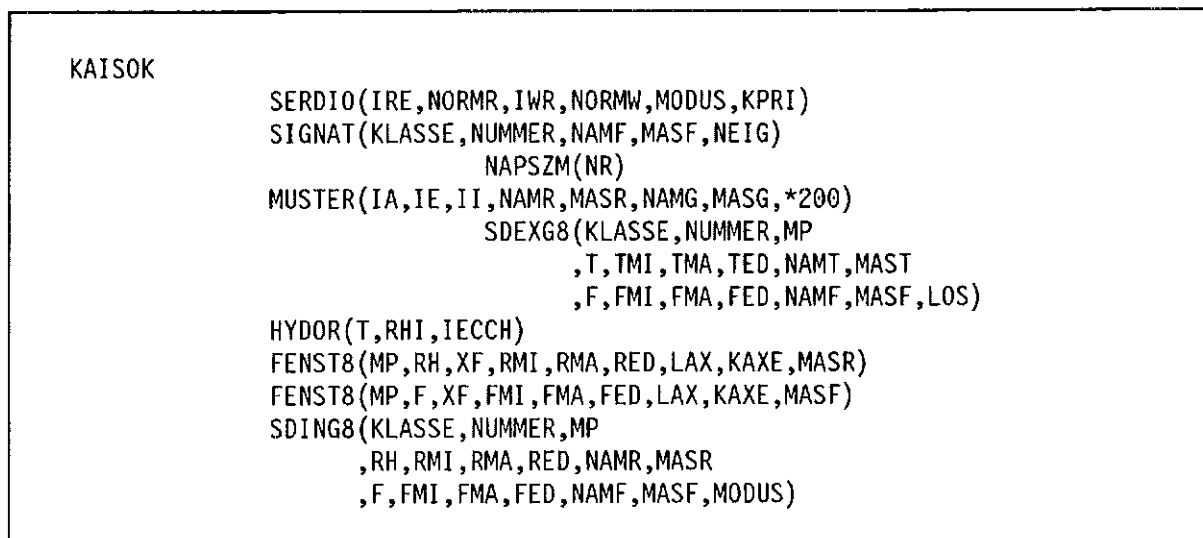


Figure 39. Module tree for the procedure KAISOK

## D.2 IVA3-ROUTINES

### WASSER

```
WASSER(T,P,RW,HW,SW,CPW,VSW
,ETW,XLMBW,PRNW,HW1P,RW1T,RW1P,SRFTS, LCATAW)
  ONLYT(T,IECCH)
  DGFIND(T,Q,DX,IECCH)
    DINSU(T,Q,FX,X,IT,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
  DLFIND(T,Q,DX,IECCH)
    SUBCRI(T)
    DILE(Y)
    DIVE(Y)
    DINLI(T,Q,FX,X,LT,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
  REDUK(T,DX,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
    TRANSP(T,RH,P1R0,P1T0,ETA,XLMB)
  SRFTS=SIGMA(T)
```

Figure 40. Module tree for the procedure WASSER

### DAMPF

```
DAMPF(T,P,RGAS,WGM,RD,HD,SD,CPD,VSD
,ETD,XLMBD,PRND,HD1P,RD1T,RD1P, LCATAD)
  ONLYT(T,IECCH)
  DGFIND(T,Q,DX,IECCH)
    DINSU(T,Q,FX,X,IT,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
  DVFIND(T,Q,DX,IECCH)
    SUBCRI(T)
    DIVE(Y)
    DINVA(T,Q,FX,X,LT,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
  REDUK(T,DX,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
    TRANSP(T,RH,P1R0,P1T0,ETA,XLMB)
```

Figure 41. Module tree for the procedure DAMPF

## RDRDP

```
RDRDP(T,P,RD,RD1P,LCATAD)
  ONLYT(T,IECCH)
  DGFIND(T,Q,DX,IECCH)
    DINSU(T,Q,FX,X,IT,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
  DVFIND(T,Q,DX,IECCH)
    SUBCRI(T)
    DINVA(T,Q,FX,X,LT,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
  REDUK(T,DX,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
    TRANSP(T,RH,P1R0,P1T0,ETA,XLMB)
```

Figure 42. Module tree for the procedure RDRDP

## SAET01

```
SAET01(T,P,VW,VD,HW,HD,HWD,SW,SD,TS1P,CPW,CPD
,ETW,ETD,XLMBW,XLMBD,SRFTS,LCATAS)
  TCORR(T,Q,DLK,DVK,IECCH)
    TS1(P,T,TS1P,IECCH)
    ONLYT(T,IECCH)
    SUBCRI(T)
    GPCORR(T,Q,DLK,DVK)
  ONLYT(T,IECCH)
  SUBCRI(T)
  GPCORR(T,Q,DLK,DVK)
  TWOPHA(T)
  REDUK(T,DX,IECCH)
    BASE(T,DX,IECCH)
    RESID(T,DX)
    TRANSP(T,RH,P1R0,P1T0,ETA,XLMB)
  SRFTS=SIGMA(T)
```

Figure 43. Module tree for the procedure SAET01

### D.3 SUB-ROUTINES

#### HYDOR

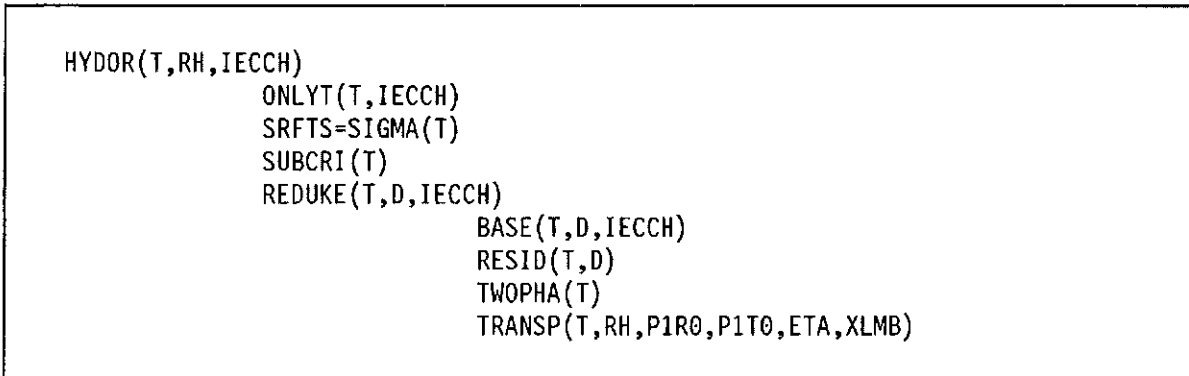


Figure 44. Module tree for the procedure HYDOR

#### SUBCRI

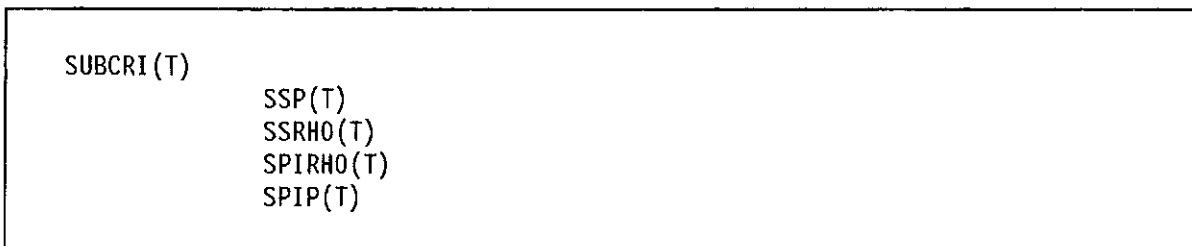


Figure 45. Module tree for the procedure SUBCRI

#### TWOPHA

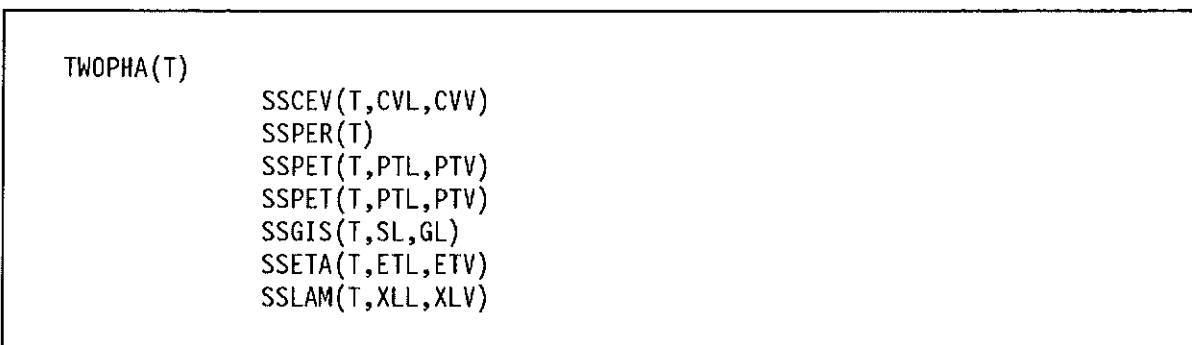


Figure 46. Module tree for the procedure TWOPHA

## GPCORR

```
GPCORR(T,Q,DLK,DVK)
  BASE(T,DLK,IECCH)
  RESID(T,DLK)
  BASE(T,DVK,IECCH)
  RESID(T,DVK)
```

Figure 47. Module tree for the procedure GPCORR

## DINSU

```
DINSU(T,Q,FX,X,IT,IECCH)
  QUADO(FX,X,Q,DX,JECCH)
    MONIKA(FX,X,JECCH)
  SORBET(KI,FX,X,Q,DX)
  BASE(T,DX,IECCH)
  RESID(T,DX)
```

Figure 48. Module tree for the function DINSU

## DINLI

```
DINLI(T,Q,FX,X,IT,IECCH)
  QUADO(FX,X,Q,DX,JECCH)
    MONIKA(FX,X,JECCH)
  SORBET(KI,FX,X,Q,DX)
  BASE(T,DX,IECCH)
  RESID(T,DX)
```

Figure 49. Module tree for the function DINLI

## DINVA

```
DINVA(T,Q,FX,X,IT,IECCH)
  QUADO(FX,X,Q,DX,JECCH)
    MONIKA(FX,X,JECCH)
  SORBET(KI,FX,X,Q,DX)
  BASE(T,DX,IECCH)
  RESID(T,DX)
```

Figure 50. Module tree for the function DINVA

## Appendix E. Common blocks

A substantial part of the data transfer between the moduls of the AQUA code takes place via common blocks. Present appendix serves to describe the most important ones of these common blocks in an abridged form.

**/SATIF/** PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV

carries thermal properties in the saturated and spinodal states, namely

PS, TPS1T, TTPS2T

the vapor pressure and its first and second derivatives,  
SL, GL

the entropy and the Gibbs-function of the saturated liquid,

DL, DV, RL1T0, RV1T0

the saturated densities with their reduced derivatives,

PL1R0, PV1R0, PL1T0, PV1T0

the density- resp. temperature-derivatives of the pressure in the saturated states,  
CVL, CVV, ETL, ETV, XLL, XLV

the constant volume heat capacities, the viscosities and the thermal conductivities  
in the saturated states,

and the densities and pressures on the liquid and vapor spinodals

RHSL, RHSV, PSL, PSV .

SATIF is supplied with data - under the direction of SUBCRI and TWOPHA - by the moduls SSP, SSRHO, SPIRHO, SSPER and SPIP. SATIF supplies data to the moduls DLFIND, DVFIND, REDUKE, TS1, TCORR, SAET01 and VD1.

**/REDUF/** G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLMB,R1T0,CPC,H0,VQ

holds thermal properties of the water, mostly in the reduced form, namely

S0, H0, G0

the entropy, the enthalpy and the Gibbs-function,

Z

the factor of reality,

CV0, CP0, CPC

the constant volume and constant pressure heat capacities and their ratio,

P1T0, P1R0, R1T0

the density- resp. temperature-derivative of the pressure

and the temperature-derivative of the density,

ETA, XLMB

the viscosity and the thermal conductivity

and the vapor quality

VQ .

REDUF is filled either by REDUK or by REDUKE, and used by the moduls HYDOR, WATER, WASSER, DAMPF and RDRDP.

**/HYDUF/** S,G,H,H1P,U1R,CV,CP,P,P1T,P1R,R1T,R1P,ETA,XLMB,SRFTS,P1RS,VS,PRN,V,VQ

is needed to supply the thermal properties of the water - calculated by HYDOR to the prorams WATER, KAISOK AND KAISOT.



The following common blocks

**/BASIF/** GB,SB,ZB,PB1T0,PB1R0,CVB  
and

**/RESEF/** GR,SR,ZR,PR1T0,PR1R0,CVR  
carry contributions - in the reduced form - to the thermal properties as calculated by the modul BASE resp. RESID. The properties are :

S, G  
the entropy and the Gibbs-function,  
Z  
the factor of reality,  
P1T0, P1R0  
the density- resp. temperature-derivative of the pressure, and  
CV  
the constant volume heat capacity.

BASIF and RESEF are needed by the moduls DGFIND, DLFIND, DVFIND, REDUK, REDUKE and GPCORR.

**/TEMPO/** RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI  
holds properties calculated in ONLYT, which depend only from the temperature as e.g.

SI, GI, CVI  
the entropy, the Gibbs-function and the constant volume heat capacity  
of the ideal gas.

RT is  $RGAS \cdot T$  in J / g. The remainig properties are needed only in BASE.

## Appendix F. Property routines

The following routines serve directly to describe some thermal property of the water. The essential ones are basically the same, as the corresponding routines in the H.G.K.-code.

### F.1 INR105.AQUA.FORT(BLOCK)

```

BLOCK DATA                                BLOCKD
IMPLICIT REAL*8(A-H,O-Z)                  BLOCKD
COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW  BLOCKD
COMMON /ADDCON/ ATZ(4),ADZ(4),AAT(4),AAD(4)                BLOCKD
COMMON /NCONST/ G(40),II(40),JJ(40),NC                    BLOCKD
C THIS BLOCKDATA SUBROUTINE SUPPLIES PARAMETERS AND COEFFICIENTS  BLOCKD
C USED IN THE REST OF THE ROUTINES.                               93/05/04  BLOCKD
C   D IN G/CM**3, GASCON IN J/(G*K), T IN K                    BLOCKD
C   G1, G2, AND GF ARE THE ALPHA, BETA AND GAMMA   OF EQ. A-2,   BLOCKD
C   G,II,JJ ARE THE G(I),K(I) AND L(I)           OF EQ. A-5.   BLOCKD
C   VALIDITY-LIMITS : TLOW=260. K, THIGH=2500. K, PHIGH=3000. MPA  BLOCKD
C   DMAX=1.2 G/CM**3                                           BLOCKD
DATA ATZ/2*64.D1,641.6D0,27.D1/,ADZ/3*.319D0,1.55D0/,AAT/2*2.D4  BLOCKD
=,4.D4,25.D0/,AAD/34.D0,4.D1,3.D1,1.05D3/                       BLOCKD
DATA WM/18.0152D0/,GASCON/.461522D0/,TZ/647.073D0/,NC/36/      BLOCKD
=,TLOW/260.D0/                                                  BLOCKD
DATA UREF,SREF,TC,DC/-4328.455039D0,7.6180802D0,647.126D0,0.322D0/  BLOCKD
DATA G1,G2,GF/11.D0,44.333333333333D0,3.5D0/                    BLOCKD
DATA G/- .53062968529023D3, .22744901424408D4, .78779333020687D3  BLOCKD
1,-.69830527374994D2, .17863832875422D5,-.39514731563338D5     BLOCKD
2, .33803884280753D5,-.13855050202703D5,-.25637436613260D6     BLOCKD
3, .48212575981415D6,-.34183016969660D6, .12223156417448D6     BLOCKD
4, .11797433655832D7,-.21734810110373D7, .10829952168620D7     BLOCKD
5,-.25441998064049D6,-.31377774947767D7, .52911910757704D7     BLOCKD
6,-.13802577177877D7,-.25109914369001D6, .46561826115608D7     BLOCKD
7,-.72752773275387D7, .41774246148294D6, .14016358244614D7     BLOCKD
8,-.31555231392127D7, .47929666384584D7, .40912664781209D6     BLOCKD
9,-.13626369388386D7, .69625220862664D6,-.10834900096447D7     BLOCKD
A,-.22722827401688D6, .38365486000660D6, .68833257944332D4     BLOCKD
B, .21757245522644D5,-.26627944829770D4,-.70730418082074D5     BLOCKD
C,-.225D0,-1.68D0,.055D0,-93.0D0/                                BLOCKD
DATA II/4*0,4*1,4*2,4*3,4*4,4*5,4*6,4*8,2*2,0,4,3*2,4/        BLOCKD
DATA JJ/2,3,5,7,2,3,5,7,2,3,5,7,2,3,5,7,2,3,5,7,2,3,5,7,2,3,5,7  BLOCKD
=,2,3,5,7,1,3*4,0,2,0,0/                                       BLOCKD
END                                                                    BLOCKD

```

## F.2 INR105.AQUA.FORT(ONLYT)

```

SUBROUTINE ONLYT(T,IECCH)                                ONLYT
C THIS SUBROUTINE CALCULATES THERMAL PROPERTIES          ONLYT
C DEPENDING ONLY FROM THE TEMPERATURE                    93/08/30 ONLYT
IMPLICIT REAL*8(A-H,O-Z)                                  ONLYT
COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW ONLYT
COMMON /TEMPO/  CGT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI  ONLYT
DIMENSION BP(7),BQ(7),C(18)                               ONLYT
DATA BP/.7478629D0,-.3540782D0,2*0.D0,.7159876D-2,0.D0    ONLYT
=,-.3528426D-2/, BQ/1.1278334D0,0.D0,-.5944001D0         ONLYT
=,-5.010996D0,0.D0,.63684256D0,0.D0/                    ONLYT
DATA C/.19730271018D2,.209662681977D2,-.483429455355D0   ONLYT
=,.605743189245D1,22.56023885D0,-9.87532442D0,-.43135538513D1 ONLYT
=,.458155781D0,-.47754901883D-1,.41238460633D-2,-.27929052852D-3 ONLYT
=,.14481695261D-4,-.56473658748D-6,.16200446D-7,-.3303822796D-9 ONLYT
=,.451916067368D-11,-.370734122708D-13,.137546068238D-15/ ONLYT
DATA UN/1.D0/                                             ONLYT
C-----
IF(T .GE. TLOW) GO TO 10                                  ONLYT
WRITE(6,'(16H TEMPERATURE T =,F10.4,13H K TOO LOW ! /)' ) T ONLYT
IECCH=-4                                                  ONLYT
RETURN                                                    ONLYT
C-----
C THIS PART CALCULATES THE B'S OF EQS. 3,4 USING COEFFICIENTS ONLYT
C FROM BLOCKDATA , CALCULATING ALSO THE FIRST AND SECOND DERIVS ONLYT
C W.R. TO TEMP. THE B'S CALCULATED HERE ARE IN CM3/G.    ONLYT
10 CGT=GASCON*T                                          ONLYT
T1=UN/T                                                  ONLYT
TV=TZ*T1                                                ONLYT
C-----
B1=BP(7)*TV                                             ONLYT
B2=BQ(7)*TV                                             ONLYT
B1T=B1*5                                                ONLYT
B2T=B2*5                                                ONLYT
B1TT=B1T*6                                              ONLYT
B2TT=B2T*6                                              ONLYT
DO 20 I=1,4                                             ONLYT
JI=7-I                                                  ONLYT
AP=BP(JI)                                              ONLYT
AQ=BQ(JI)                                              ONLYT
B1=(B1+AP)*TV                                          ONLYT
B2=(B2+AQ)*TV                                          ONLYT
AP=AP*(JI-2)                                          ONLYT
AQ=AQ*(JI-2)                                          ONLYT
B1T=(B1T+AP)*TV                                       ONLYT
B2T=(B2T+AQ)*TV                                       ONLYT
AP=AP*(JI-1)                                          ONLYT
AQ=AQ*(JI-1)                                          ONLYT
B1TT=(B1TT+AP)*TV                                     ONLYT
20 B2TT=(B2TT+AQ)*TV                                   ONLYT
B1=B1-BP(2)*DLOG(TV)+BP(1)                             ONLYT
B2=B2+BQ(1)                                            ONLYT
B1T=(BP(2)-B1T)*T1                                    ONLYT

```

```

B2T=-B2T*T1
B1TT=(-BP(2)+B1TT)*T1*T1
B2TT=B2TT*T1*T1
C-----
C THIS PART CALCULATES THE THERMODYNAMIC PROPERTIES FOR
C WATER IN THE IDEAL GAS STATE FROM FUNCTION OF H.W. WOOLLEY
50 TT=T*1.D-2
TT1=UN/TT
TL=DLOG(TT)
GI=-(C(2)+C(1)*TT1)*TL
HI= C(2)+C(1)*(UN-TL)*TT1
CPI=C(2)-C(1)*TT1
DO 60 J=3,18
CIT=C(J)*TT**(J-6)
GI=GI-CIT
HI=HI+(J-6)*CIT
60 CPI=CPI+(J-5)*(J-6)*CIT
SI=HI-GI
CVI=CPI-UN
C-----
RETURN
END

```

### F.3 INR105.AQUA.FORT(BASE)

```

SUBROUTINE BASE(T,D,IECCH)
C BASE CALCULATES REDUCED THERMAL PROPERTIES OF THE WATER
C VIA THE BASE-FUNCTION FOR A GIVEN TEMPERATURE, T AND DENSITY, D.
C THE PROPERTIES ARE: ZB=P/(D*RT), PB1T0=(DP/DT)/(D*R),
C PB1R0=(DP/DRH)/(T*R), GB=G/RT, SB=S/R, CVB=CV/R.
C T IS IN K, D IS IN G/CM**3.
C IMPLICIT REAL*8 (A-H,O-Z)
C G1,G2 AND GF ARE THE ALPHA, BETA AND GAMMA OF EQ 2, WHICH ARE
C SUPPLIED BY THE BLOCKDATA ROUTINE. B1 AND B2 ARE THE "EXCLUDED
C VOLUME" AND "2ND VIRIAL" (EQS 3 AND 4) SUPPLIED BY THE SUBROUTINE
C BB(T), WHICH ALSO SUPPLIES THE 1ST AND 2ND DERIVATIVES WITH
C RESPECT TO T (B1T,B2T,B1TT,B2TT).
COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW
COMMON /TEMPO/ CGT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI
COMMON /BASIF/ GB,SB,ZB,PB1T0,PB1R0,CVB
PARAMETER(DMAX=1.2D0)
DATA UN/1.D0/
IF(D .GT. DMAX) THEN
WRITE(6,9100) D
IECCH = 4
RETURN
ENDIF
9100 FORMAT(12H DENSITY D =,G15.6,20H G/CM**3 TOO HIGH ! /)
10 YK=2.5D-1*B1*D
W=UN/(UN-YK)
W2=W*W
Z0=(UN+(G1+G2*YK)*YK)*W2*W
BPB=B2/B1

```

Z1=4.D0*(BPB-GF)	BASE
ZB = Z0+Z1*YK	BASE
DZ0=( 3.D0*Z0 + (G1+2.D0*G2*YK)*W2 ) *W	BASE
DBZ = DZ0+Z1	BASE
AB = 17.4560887731D0 + Z1*YK + W*(UN-G2+28.16666667D0*W)	BASE
=+ DLOG(D*CGT*W)	BASE
GB = AB + ZB	BASE
UBB1=B1T/B1	BASE
UB = -T*(UBB1*(ZB-UN-D*B2) + D*B2T)	BASE
SB = UB - AB	BASE
CVB = 2.D0*UB + T*T*( (Z0-UN)*(UBB1*UBB1-B1TT/B1)	BASE
- D*(B2TT-GF*B1TT) - UBB1*UBB1*YK*DZO )	BASE
PB1T0 = ZB + T*D*(B2T+(2.5D-1*DBZ-BPB)*B1T)	BASE
PB1R0 = ZB+DBZ*YK	BASE
RETURN	BASE
END	BASE

#### F.4 INR105.AQUA.FORT(RESID)

SUBROUTINE RESID(T,D)	RESID
C RESID CALCULATES REDUCED THERMAL PROPERTIES OF THE WATER VIA	RESID
C THE RESIDUAL-FUNCTION FOR A GIVEN TEMPERATURE, T AND DENSITY, D.	RESID
C THE PROPERTIES ARE: ZR=P/(D*RT), PR1T0=(DP/DT)/(D*R),	RESID
C PR1R0=(DP/DRH)/(T*R), GR=G/RT, SR=S/R, CVR=CV/R.	RESID
C T IS IN K, D IS IN G/CM**3.	94/11/30 RESID
C TERMS 37 THRU 39 ARE THE ADDITIONAL TERMS AFFECTING ONLY THE	RESID
C IMMEDIATE VICINITY OF THE CRITICAL POINT, AND TERM 40 IS THE	RESID
C ADDITIONAL TERM IMPROVING THE LOW T, HIGH P REGION.	RESID
IMPLICIT REAL*8(A-H,O-Z)	RESID
DIMENSION QR(11),QT(10),QZR(9),QZT(9)	RESID
COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW	RESID
COMMON /ADDCON/ ATZ(4),ADZ(4),AAT(4),AAD(4)	RESID
COMMON /NCONST/ G(40),II(40),JJ(40),NC	RESID
COMMON /TEMPO/ CGT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI	RESID
COMMON /RESEF/ GR,SR,ZR,PR1T0,PR1R0,CVR	RESID
EQUIVALENCE (QR(3),QZR(1)),(QT(2),QZT(1))	RESID
DATA ZERO,UN/0.D0,1.D0/	RESID
GASI=UN/GASCON	RESID
QR(1)=ZERO	RESID
PRR=ZERO	RESID
PR=ZERO	RESID
AR=ZERO	RESID
DADT=ZERO	RESID
CVR=ZERO	RESID
PTR=ZERO	RESID
IF(D .LT. 1.D-15) GO TO 90	RESID
W = D	RESID
TWPD=2.D0/D	RESID
IF(W .GE. 1.D-3) THEN	RESID
E=DEXP(-W)	RESID
Q20=UN-E	RESID
ELSE	RESID
Q20=W*(UN-5.D-1*W*(UN-3.3333333D-1*W))	RESID

E=UN-Q20	RESID
ENDIF	RESID
VER0=E/Q20	RESID
VER1=VER0*D*D	RESID
Q10=D*D*E	RESID
Q10RT=Q10*CGT	RESID
QR(2)=Q10	RESID
QT(1)=T/TZ	RESID
DO 11 I=2,10	RESID
IF(QR(I) .LE. 1.E-35) QR(I)=ZERO	RESID
QR(I+1)=QR(I)*Q20	RESID
11 QT(I)=QT(I-1)*TV	RESID
DO 21 I=1,NC	RESID
K=II(I)+1	RESID
L=JJ(I)	RESID
GPP=G(I)*QZT(L)	RESID
QP=GPP*QZR(K-1)	RESID
PR = PR + QP	RESID
PRR = PRR + QP*( TWPD - (UN-(K-1)*VER0) )	RESID
BUAM=K*LOG(Q20)	RESID
IF(BUAM .LE. -35.) AUA=ZERO	RESID
IF(BUAM .GT. -35.) AUA=EXP(BUAM)	RESID
AR=AR+GPP*QZR(K)/(K*Q10RT)	RESID
DFDT=AUA*(1-L)*QZT(L+1)/(TZ*K)	RESID
GPDT=G(I)*DFDT	RESID
DADT=DADT+GPDT	RESID
DPT=GPDT*K*VER1	RESID
PTR=PTR+DPT	RESID
D2F=L*GPDT*GASI	RESID
21 CVR=CVR+D2F	RESID
QP=ZERO	RESID
Q2A=ZERO	RESID
DO 31 J=37,40	RESID
IF(G(J) .EQ. ZERO) GO TO 31	RESID
K=II(J)	RESID
KM=JJ(J)	RESID
DDZ = ADZ(J-36)	RESID
DDZ1 = UN/DDZ	RESID
DEL = D*DDZ1 - UN	RESID
IF(DABS(DEL) .LT. 1.D-10) DEL=1.D-10	RESID
EX1 = -AAD(J-36)*DEL**K	RESID
IF(EX1 .GT. -170.D0) DEX=DEXP(EX1)*DEL**KM	RESID
IF(EX1 .LE. -170.D0) DEX=ZERO	RESID
ATT = AAT(J-36)	RESID
TX1=UN/ATZ(J-36)	RESID
TA=T*TX1	RESID
TAU=TA-UN	RESID
EX2 = -ATT*TAU*TAU	RESID
IF(EX1+EX2 .GT. -170.D0) THEN	RESID
TEX=DEXP(EX2)	RESID
Q10 = DEX*TEX	RESID
IF(DABS(Q10) .LT. 1.D-14) Q10=ZERO	RESID
ELSE	RESID
Q10 = ZERO	RESID
GO TO 31	RESID

```

ENDIF RESID
QM1 = KM/DEL RESID
QM2 = K*AAD(J-36)*DEL**(K-2) RESID
QM = QM1 - QM2*DEL RESID
FCT1=D*D*Q10*DDZ1 RESID
FCT=QM*FCT1 RESID
Q5T = (FCT1*DDZ1)*(QM*(TWPD*DDZ+QM) - (QM1/DEL+(K-1)*QM2)) RESID
PRR = PRR + Q5T*G(J) RESID
QP = QP + G(J)*FCT RESID
GATT = 2.D0*G(J)*ATT*TX1 RESID
GATTA = GATT*TAU RESID
IF(DABS(GATTA) .LT. 1.D-14) GATTA=ZERO RESID
DADT = DADT - Q10*GATTA RESID
PTR = PTR - GATTA*FCT RESID
Q2A = Q2A + Q10*GATT*(2.D0*EX2+UN)*TA RESID
AR = AR + Q10*G(J)/CGT RESID
31 CONTINUE RESID
CVR = CVR + Q2A*GASI RESID
PR = PR + QP RESID
ZR = PR/(CGT*D) RESID
PR1T0 = PTR/(GASCON*D) RESID
PR1R0 = PRR/CGT RESID
90 GR = AR + ZR RESID
SR = -DADT*GASI RESID
RETURN RESID
END RESID

```

### F.5 INR105.AQUA.FORT(SIGMA)

```

FUNCTION SIGMA(T) SIGMA
C CALCULATES THE SURFACE TENSION OF THE WATER IN N/M 94/12/07 SIGMA
IMPLICIT REAL*8(A-H,O-Z) SIGMA
PARAMETER(TC=647.126D0,BG=0.2358D0,BK=-0.625D0,AMU=1.256D0) SIGMA
Y=(TC-T)/TC SIGMA
SIGMA=BG*(1+BK*Y)*Y**AMU SIGMA
RETURN SIGMA
END SIGMA

```

### F.6 INR105.AQUA.FORT(TRANSP)

```

SUBROUTINE TRANSP(T,RH,P1R0,P1T0,ETA,XLMB) TRANSP
C TRANSP CALCULATES THE TRANSPORT PROPERTIES TRANSP
C VISCOSITY ( ETA ) AND THERMAL CONDUCTIVITY ( XLMB ) TRANSP
C AS FUNCTIONS OF THE TEMPERATURE ( T ) AND THE DENSITY ( RH ). TRANSP
C ALL THE PROPERTIES ARE IN IN S.I. UNITS . TRANSP
C 'TRANSP' NEEDS ALSO THE REDUCED PRESSURE DERIVATIVES TRANSP
C P1R0 = DP/(RGAS*T*DRHO) AND P1T0 = DP/(RGAS*RHO*DT) TRANSP
C FOR CALCULATING THE NONREGULAR PART OF THE TRANSP
C THERMAL CONDUCTIVITY ( DELLAM ) 94/12/01 TRANSP
IMPLICIT REAL*8(A-H,O-Z) TRANSP
DIMENSION E0(4),X0(4),BE(6,5),BX(5,6) TRANSP
PARAMETER(TC=647.126D0,RGAS=4.61522D+2) TRANSP

```

	DATA RZO,TZO,PZO/317.763D+0,647.126D+0,22.115D+6/	TRANSP
	=,CC,OMEGA,AC,BC/3.7711D-8,0.4678D+0,18.66D+0,1.0D+0/	TRANSP
	DATA E0/1.81583D-2,1.77624D-2,1.05287D-2,-0.36744D-2/	TRANSP
	DATA X0/2.02223D+0,1.411166D+1,5.25597D+0,-2.01870D+0/	TRANSP
	DATA BE/5.01938D-1,1.62888D-1,-1.30356D-1,9.07919D-1,-5.51119D-1	TRANSP
	=,1.46543D-1,2.35622D-1,7.89393D-1,6.73665D-1,1.207552D-0	TRANSP
	=,6.70665D-2,-8.43370D-2,-2.74637D-1,-7.43539D-1,-9.59456D-1	TRANSP
	=,-6.87343D-1,-4.97089D-1,1.95286D-1,1.45831D-1,2.63129D-1	TRANSP
	=,3.47247D-1,2.13486D-1,1.00754D-1,-3.29320D-2,-2.70448D-2	TRANSP
	=,-2.53093D-2,-2.67758D-2,-8.22904D-2,6.02253D-2,-2.02595D-2/	TRANSP
	DATA BX/1.32930460D-0,1.7018363D-0,5.2246158D-0,8.7127675D-0	TRANSP
	=,-1.852599D-0,-4.0452437D-1,-2.2156845D-0,-1.0124111D+1	TRANSP
	=,-9.5000611D-0,9.340469D-1,2.440949D-1,1.6511057D-0,4.9874687D-0	TRANSP
	=,4.3786606D-0,0.D-0,1.8660751D-2,-7.6736002D-1,-2.7297694D-1	TRANSP
	=,-9.1783782D-1,0.D-0,-1.2961068D-1,3.7283344D-1,-4.3083393D-1	TRANSP
	=,0.D-0,0.D-0,4.4809953D-2,-1.120316D-1,1.3333849D-1,0.D-0,0.D-0/	TRANSP
C	THE IDEAL TEMPERATURE	TRANSP
	TID=PZO/(RZO*RGAS)	TRANSP
	TR=T/TID	TRANSP
C	REDUCED VARIABLES	TRANSP
	TT=T/TZO	TRANSP
	T1=1.D0/TT	TRANSP
	T2=SQRT(TT)	TRANSP
	T3=T1-1.D-0	TRANSP
	T4=TT-1.D-0	TRANSP
	RR=RH/RZO	TRANSP
	R2=SQRT(RR)	TRANSP
	R3=RR-1.D-0	TRANSP
C	VISCOSITY	TRANSP
	EN=E0(1)+T1*(E0(2)+T1*(E0(3)+T1*E0(4)))	TRANSP
	ETA0=T2*1.D-6/EN	TRANSP
	ESU=0.D0	TRANSP
	DO 110 J=6,1,-1	TRANSP
	BEJ = BE(J,1)+R3*(BE(J,2)+R3*(BE(J,3)+R3*(BE(J,4)+ R3*BE(J,5))))	TRANSP
110	ESU=T3*ESU+BEJ	TRANSP
	ETA=ETA0*EXP(RR*ESU)	TRANSP
C	THERMAL CONDUCTIVITY	TRANSP
	XN=X0(1)+T1*(X0(2)+T1*(X0(3)+T1*X0(4)))	TRANSP
	XLA0=T2/XN	TRANSP
	EXU=0.D0	TRANSP
	DO 210 I=5,1,-1	TRANSP
	BXI = BX(I,1)+R3*(BX(I,2)+R3*(BX(I,3)	TRANSP
	=+R3*(BX(I,4)+R3*(BX(I,5)+R3*BX(I,6))))	TRANSP
210	EXU=T3*EXU+BXI	TRANSP
	QXQ=RR*EXU	TRANSP
	XLMB=XLA0*EXP(RR*EXU)	TRANSP
C	NONREGULAR PART OF THE THERMAL CONDUCTIVITY	TRANSP
C	ISOTHERMAL COMPRESSIBILITY	TRANSP
230	IF(P1R0 .NE. 0.D0) CHIT = RR/(TR*P1R0)	TRANSP
	IF(P1R0 .EQ. 0.D0) CHIT = 1.D40	TRANSP
	FIO = ABS(CHIT)	TRANSP
	IF(FIO .EQ. 0.D0) FIO=1.D-6	TRANSP
	CHITE = FIO**OMEGA	TRANSP
	ZW=R3*R3	TRANSP
C	DAMPNG FUNCTION OF SENGERS, J. V. , 10TH INT. CONF. ON STEAM	TRANSP



```

CHECKA=R2*EXP(-AC*T4*T4-BC*ZW*ZW)             TRANSP
ZW=TR*P1T0                                     TRANSP
DELLAM =(CC/ETA)*(ZW*ZW)*CHITE*CHECKA          TRANSP
XLMB=XLMB+DELLAM                               TRANSP
999 RETURN                                     TRANSP
END                                              TRANSP

```

## F.7 INR105.AQUA.FORT(SSP)

```

SUBROUTINE SSP(T)                               SSP
C  SSP(T) CALCULATES APPROXIMATIONS             SSP
C  TO THE VAPOR PRESSURE      P(T)              ( PS      ) SSP
C  TO ITS FIRST              T*DPS(T)/DT         ( TPS1T  ) SSP
C  AND SECOND DERIVATIVE     T*T*D2PS(T)/DT**2  ( TTPS2T ) SSP
C  AS FUNCTIONS OF THE INPUT TEMPERATURE T IN PA-S.   SSP
C  AFTER WAGNER ET SAUL, 1984, IAPS MOSCOW CONFERENCE. 94/11/30 SSP
C  IMPLICIT REAL*8(A-H,O-Z)                     SSP
COMMON /SATIF/  PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL  SSP
              =,RHVS,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV  SSP
PARAMETER(TC=647.126D0,PC=22.0549D6)             SSP
DATA UN/1.D0/                                     SSP
IF(T .GT. TC)            RETURN                   SSP
TR=T/TC                                               SSP
TI=UN/TR                                               SSP
W=SQRT(UN-TR)                                         SSP
W3=W*W*W                                             SSP
W7=W*W3*W3                                           SSP
POLA = -7.85823D+0 + W*( 1.83991D+0 - W3*( 11.7811D+0          SSP
= - W*( 22.67050D+0 - W*( 15.9393D+0 - W7*1.77516D+0 )))     SSP
POLB = W*( 9.19955D-1 - W3*( 23.5622D+0                       SSP
= - W*( 56.67625D+0 - W*( 47.8179D+0 - W7*11.53854D+0 )))   SSP
POLC = W*( 1.3799325D-0 - W3*( 70.6866D+0                    SSP
= - W*( 198.366875D+0 - W*( 191.2716D+0 - W7*86.53905D+0 ))) SSP
SSPLN=(TI-UN)*POLA                                     SSP
PS=EXP(SSPLN)*PC                                       SSP
Q1=-(TI*POLA+POLB)                                     SSP
TPS1T=PS*Q1                                           SSP
IF(T .EQ. TC)            RETURN                       SSP
Q2=(TI*POLA+POLB+POLC/(TI-UN))                         SSP
TTPS2T=PS*(Q2-Q1*(UN-Q1))                             SSP
RETURN                                                SSP
END                                                    SSP

```

## F.8 INR105.AQUA.FORT(TS1)

```

SUBROUTINE TS1(P,TS,TS1P,IECCH) TS1
C   TS1 RETURNS - AT A GIVEN PRESSURE P IN PA - APPROXIMATED VALUES TS1
C   FOR THE SATURATION TEMPERATURE, TS AND TS1
C   FOR THE DERIVATIVE TS1P = DTS/DP 94/12/01 TS1
  IMPLICIT REAL*8(A-H,O-Z) TS1
  COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL TS1
  =,RHSLV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV TS1
  PARAMETER(PMIN=6.1D+2,PCN=21.75966D6,PC=22.0549D6) TS1
C   PARAMETER(TMIN=273.15D0,TCN=646.00,TC=647.126D0) TS1
  DATA FINIS/1.D-5/ TS1
  IF(P .GE. PMIN .AND. P .LE. PC) GO TO 10 TS1
  WRITE(6,9100) P TS1
  IF(P .LT. PMIN) IECCH=-2 TS1
  IF(P .GT. PC) IECCH=+2 TS1
  RETURN TS1
C   ANALYTIC DESCRIPTION TS1
10 IF(P .GT. PCN) THEN TS1
    TS=562.764692D0+3.82515211D-6*P TS1
  ELSE TS1
    PL=DL0G(P)-11.512925465D0 TS1
    TS=372.83D0+ TS1
    = PL*(27.7589D0+PL*(2.3819D0+PL*(0.24834D0+PL*0.193855D-1))) TS1
  ENDIF TS1
C   CORRECTURES TS1
  DO 21 K=1,7 TS1
  CALL SSP(TS) TS1
  IF(P .GE. PCN) GO TO 90 TS1
  DP=P-PS TS1
  DEV=ABS(DP/P) TS1
  IF(DEV .LT. FINIS) GO TO 90 TS1
21 TS = TS*(1.00+DP/TPS1T) TS1
  IECCH=3 TS1
  WRITE(6,9300) TS1
  RETURN TS1
  90 TS1P=TS/TPS1T TS1
9100 FORMAT(27H PRESSURE RANGE EXCEED! P =,G15.6,7H PASCAL) TS1
9300 FORMAT(51H ITERATION FOR THE SATURATIION TEMPERATURE FAILED ! ) TS1
  RETURN TS1
  END TS1

```

## F.9 INR105.AQUA.FORT(SSRHO)

```

SUBROUTINE SSRHO(T) SSRHO
C   SSRHO RETURNS APPROXIMATED VALUES OF THE DENSITIES OF SSRHO
C   THE SATURATED LIQUID ( DL ) AND THE SATURATED VAPOR ( DV ) SSRHO
C   - IN G/CM**3 ! - AS FUNCTIONS OF THE TEMPERATURE ( T ). SSRHO
C   SSRHO CALCULATES ALSO THE REDUCED DENSITY DERIVATIVES SSRHO
C   RL1T0 = T/RHL * DRHL/DT AND RV1T0 = T/RHV * DRHV/DT. 94/12/01 SSRHO
  IMPLICIT REAL*8(A-H,O-Z) SSRHO
  DIMENSION PUL(12),DPUL(12),PUV(12),DPUV(12) SSRHO

```

```

COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL      SSRHO
=,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV  SSRHO
PARAMETER(TC=647.126D0,DC=0.322D0,RGAS=4.61522D+2)              SSRHO
PARAMETER(BR=2.03D0,BET=0.325D0)                                SSRHO
DATA PUL/ 6.072746418D+0,-149.9456089D+0,2.121007381D+3          SSRHO
=,-17.16483301D+3,90.17684684D+3,-323.0456094D+3,805.4435472D+3  SSRHO
=,-1.398515305D+6,1.658696365D+6,-1.281195416D+6,580.7414278D+3  SSRHO
=,-117.1819097D+3/                                              SSRHO
DATA DPUL/ 0.0D+0,-74.97280445D+0, 2.121007381D+3              SSRHO
=,-25.747249515D+3, 180.35369368D+3,-807.6140235D+3           SSRHO
=, 2.4163306416D+6,-4.8948035675D+6, 6.63478546D+6           SSRHO
=,-5.765379372D+6, 2.903707139D+6, -644.50050335D+3/         SSRHO
DATA PUV/-1.125999398D+0,42.15284538D+0,-538.0831003D+0       SSRHO
=,4.150173161D+3,-20.80947751D+3,71.29507069D+3,-170.4446088D+3  SSRHO
=,284.4989747D+3,-325.1357165D+3,242.4593090D+3,-106.2552030D+3  SSRHO
=,20.74762670D+3/                                              SSRHO
DATA DPUV/0.0D+0,21.07642269D+0,-538.0831003D+0,6.225259741D+3  SSRHO
=,-41.61895501D+3,178.2376767D+3,-511.3338264D+3,995.7464115D+3  SSRHO
=,-1.300542866D+6,1.091066891D+6,-531.2760148D+3,114.1119469D+3/  SSRHO
IF(T .GE. TC) THEN                                             SSRHO
  DL = DC                                                       SSRHO
  DV = DC                                                       SSRHO
  RETURN                                                         SSRHO
ENDIF                                                            SSRHO
Y = (TC-T)/TC                                                  SSRHO
CHI = T/(T-TC)                                                 SSRHO
IF(T .LE. 620.D0) THEN                                        SSRHO
C  LOW TEMPERATURES                                           SSRHO
  W = SQRT(Y)                                                  SSRHO
  OML = PUL(12)                                               SSRHO
  TDOML = DPUL(12)                                           SSRHO
  ZV = PUV(12)                                               SSRHO
  TDZV = DPUV(12)                                           SSRHO
  DO 31 K = 11,1,-1                                          SSRHO
  TDOML = TDOML*W+DPUL(K)                                    SSRHO
  TDZV = TDZV*W+DPUV(K)                                    SSRHO
  OML = OML*W+PUL(K)                                       SSRHO
31  ZV = ZV*W+PUV(K)                                       SSRHO
  DLOML = TDOML/OML                                         SSRHO
  DV = PS*1.D-3/(RGAS*T*ZV)                                 SSRHO
  DLZV = CHI*TDZV/ZV                                       SSRHO
  RV1T0 = TPS1T/PS -1.D0 -DLZV                              SSRHO
ELSE                                                            SSRHO
C  NEAR CRITICAL TEMPERATURES                                SSRHO
  BEG = BR*Y**BET                                           SSRHO
  EB = BEG*BET                                               SSRHO
  AL = Y*(3.075668853D-0-Y*(19.58060683D-0-Y*Y*1.383086689D+3))  SSRHO
  OML = 1.D0 + BEG + AL                                       SSRHO
  DAL = AL -Y*Y*(19.58060683D-0-Y*Y*4.149260067D+3)        SSRHO
  DLOML = (EB + DAL)/OML                                     SSRHO
  AV = Y*(1.451962241D-0-Y*(5.041325559D-0-Y*Y*611.1810749D-0))  SSRHO
  OMV = 1.D0 -BEG + AV                                       SSRHO
  DAV = AV -Y*Y*(5.041325559D-0 -Y*Y*1.833543225D+3)        SSRHO
  DV = OMV*DC                                               SSRHO
  RV1T0 = CHI*(-EB + DAV)/OMV                               SSRHO

```

```

ENDIF                                SSRHO
DL = OML*DC                          SSRHO
RL1T0 = DLOML*CHI                    SSRHO
RETURN                                SSRHO
END                                    SSRHO

```

### F.10 INR105.AQUA.FORT(SPIRHO)

```

SUBROUTINE SPIRHO(T)                 SPIRHO
C   SPIRHO RETURNS - FOR A GIVEN TEMPEREATURE T - APPROXIMATED VALUES SPIRHO
C   OF THE DENSITY OF THE LIQUID SPINODAL ( RHSL ) AND                SPIRHO
C   OF THE VAPOR SPINODAL ( RHSV ) IN KG/M**3.                       94/12/01 SPIRHO
IMPLICIT REAL*8(A-H,O-Z)            SPIRHO
DIMENSION PUL(9),PUV(9)             SPIRHO
COMMON /SATIF/ PS,TPS1T,TPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL           SPIRHO
=,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV   SPIRHO
PARAMETER(TC=647.126D0,RHC=322.D0,RGAS=4.61522D+2)                 SPIRHO
PARAMETER(BSP=1.40D0,BET=0.325D0)                                   SPIRHO
DATA PUL/1.911888766D+0,-16.54157899D+0,179.7358194D+0             SPIRHO
=,-940.1374463D+0,2.957964417D+3,-5.752429056D+3,6.772987553D+3   SPIRHO
=,-4.428001759D+3,1.234441322D+3/                                  SPIRHO
DATA PUV/0.9201508683D-0,-2.159701944D-0,-0.2932760492D-0       SPIRHO
=,6.324940268D-0,-10.49597891D-0,8.78261245D-0,-3.119252324D-0   SPIRHO
=,0.D0,0.D0/                                                       SPIRHO
IF(T .GE. TC) THEN                                                 SPIRHO
    RHSL = RHC                                                       SPIRHO
    RHSV = RHC                                                       SPIRHO
    RETURN                                                           SPIRHO
ENDIF                                                                SPIRHO
Y = (TC-T)/TC                                                       SPIRHO
IF(T .LE. 620.D0) THEN                                             SPIRHO
C   LOW TEMPERATURES                                               SPIRHO
    W = SQRT(Y)                                                       SPIRHO
    OML = PUL(9)                                                       SPIRHO
    OMV = PUV(9)                                                       SPIRHO
    DO 31 I=8,1,-1                                                    SPIRHO
    OML = OML*W+PUL(I)                                                SPIRHO
31  OMV = OMV*W+PUV(I)                                              SPIRHO
ELSE                                                                    SPIRHO
C   NEAR CRITICAL TEMPERATURES                                     SPIRHO
    AL = Y*(487.1717941D-3+Y*(24.26899415D-0-Y*Y*2.461494028D+3)) SPIRHO
    OML = 1.D0 + BSP*Y**BET + AL                                       SPIRHO
    AV = Y*(1.132819731D-0-Y*(30.70131059D-0-Y*Y*3.324944936D+3)) SPIRHO
    OMV = 1.D0 - BSP*Y**BET + AV                                       SPIRHO
ENDIF                                                                SPIRHO
RHSL = OML*RHC                                                       SPIRHO
RHSV = OMV*RHC                                                       SPIRHO
RETURN                                                                SPIRHO
END                                                                    SPIRHO

```

## F.11 INR105.AQUA.FORT(SPIP)

```

SUBROUTINE SPIP(T)
C SPIP RETURNS - FOR A GIVEN TEMPERATURE T - APPROXIMATED VALUES
C OF THE PRESSURE OF THE LIQUID SPINODAL ( PSL ) AND
C OF THE VAPOR SPINODAL ( PSV ) IN PASCALS.
C
C 94/12/01
IMPLICIT REAL*8(A-H,O-Z)
DIMENSION PUL(9),PUV(9)
COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL
=,RHSLV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV
PARAMETER(TC=647.126D0,RHC=322.D0,RGAS=4.61522D+2)
PARAMETER(PSVCL=498.D6,PSVCV=200.D6,PC=22.0549D6)
DATA PUL/-1.387361801D-3,151.0341901D-3,1.980521347D-0
=, 2.921317875D-0,-38.3240633D-0,162.2329521D-0
=, -384.0027285D-0,479.3961544D-0,-264.1248035D-0/
DATA PUV/-749.7146775D-6,162.845372D-3,-319.0428782D-3
=, -1.582269493D-0, 6.787420254D-0,-10.41333815D-0
=, 7.599390008D-0,-2.285884902D-0,0.D0/
IF(T .GE. TC) THEN
PSL=PC
RETURN
ENDIF
Y=(TC-T)/TC
IF(T .LE. 620.D0) THEN
C LOW TEMPERATURES
DPSL = PUL(9)
DPSV = PUV(9)
DO 31 I=8,1,-1
DPSL = DPSL*Y+PUL(I)
31 DPSV = DPSV*Y+PUV(I)
ELSE
C NEAR CRITICAL TEMPERATURES
DPSL = Y*(52.35363429D-3+Y*(4.419437236D-0-Y*18.86727538D-0))
DPSV = Y*(110.3469546D-3+Y*(913.150777D-3 -Y*10.87001272D-0))
ENDIF
PSL=PS-DPSL*PSVCL
PSV=PS+DPSV*PSVCV
RETURN
END

```

## F.12 INR105.AQUA.FORT(SSPER)

```

SUBROUTINE SSPER(T)
C SSPER RETURNS - FOR A GIVEN TEMPERATURE T -
C APPROXIMATED DENSITY DERIVATIVES OF THE PRESSURE
C IN THE SATURATED STATES - IN REDUCED FORM .
C
C 94/12/01
C PL1R0 = D(P,L)/DRH * 1/(RGAS*T)
C PV1R0 = D(P,V)/DRH * 1/(RGAS*T)
IMPLICIT REAL*8(A-H,O-Z)
DIMENSION PUL(12),PUV(12)
COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL
=,RHSLV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV

```

	PARAMETER(TC=647.126D0,DC=322.D-3,PROC=15.D0,GAM=1.24D0)	SSPER
	DATA PUL/91.68326854D-3,141.0358804D-3,293.745773D-0	SSPER
	=,-3.694718543D+3,35.3159487D+3,-224.5051913D+3,968.1021275D+3	SSPER
	=,-2.837321019D+6,5.560661663D+6,-6.973719089D+6,5.058540246D+6	SSPER
	=,-1.615244963D+6/	SSPER
	DATA PUV/1.435429119D-3,3.738553917D-0,3.552096947D-0	SSPER
	=,-55.88507759D-0,217.5743793D-0,-535.2445068D-0,812.4224243D-0	SSPER
	=,-670.9285889D-0,229.0827484D-0,0.D0,0.D0,0.D0/	SSPER
	IF(T .GE. TC) THEN	SSPER
	SSPRHL = 0.D0	SSPER
	RETURN	SSPER
	ENDIF	SSPER
	Y=(TC-T)/TC	SSPER
	IF(T .LE. 620.D0) THEN	SSPER
C	LOW TEMPERATURES	SSPER
	PL1R0 = PUL(12)	SSPER
	PV1R0 = PUV(12)	SSPER
	DO 31 I=11,1,-1	SSPER
	PL1R0 = PL1R0*Y+PUL(I)	SSPER
31	PV1R0 = PV1R0*Y+PUV(I)	SSPER
	ELSE	SSPER
C	NEAR CRITICAL TEMPERATURES	SSPER
	YGA = Y**GAM	SSPER
	FIL = PROC	SSPER
	= -Y*(79.73057836D-0+Y*(430.1476885D-0-Y*14139.7632D+0))	SSPER
	FIV = PROC	SSPER
	= +Y*(369.6750971D-0-Y*(5110.089071D-0-Y*46121.0908D-0))	SSPER
	PL1R0 = (DL/DC)*FIL*YGA	SSPER
	PV1R0 = (DV/DC)*FIV*YGA	SSPER
	ENDIF	SSPER
	RETURN	SSPER
	END	SSPER

### F.13 INR105.AQUA.FORT(SSPET)

	SUBROUTINE SSPET(T,PL1T0,PV1T0)	SSPET
C	SSPET RETURNS - FOR A GIVEN TEMPERATURE T -	SSPET
C	APPROXIMATED TEMPERATURE DERIVATIVES OF THE PRESSURE	SSPET
C	IN THE SATURATED STATES - IN REDUCED FORM .	94/12/01 SSPET
C	PL1T0 = D(P,L)/DT * 1/(RGAS*RH,L)	SSPET
C	PV1T0 = D(P,V)/DT * 1/(RGAS*RH,V)	SSPET
	IMPLICIT REAL*8(A-H,0-Z)	SSPET
	DIMENSION PUL(12),PUV(12)	SSPET
	PARAMETER(TC=647.126D0,PT0C=1.80215732D-0)	SSPET
	DATA PUL/1.818884544D-0,16.40737736D-0,-65.85455759D-0	SSPET
	=,52.04371914D-0,4.031238729D+3,-47.2570313D+3,277.4234268D+3	SSPET
	=,-988.3324003D+3,2.218919758D+6,-3.077677674D+6,2.414555699D+6	SSPET
	=,-821.0002244D+3/	SSPET
	DATA PUV/2.029700729D-0,-125.5470649D-3,-61.54407176D-0	SSPET
	=,544.8751175D-0,-2.848188899D+3,9.708977783D+3,-21.26251929D+3	SSPET
	=,28.71096533D+3,-21.75527246D+3,7.082224609D+3,0.D0,0.D0/	SSPET
	IF(T .GE. TC) THEN	SSPET
	PL1T0 = PT0C	SSPET

```

        PV1T0 = PT0C
        RETURN
    ENDIF
    Y=(TC-T)/TC
    IF(T .LE. 620.D0)      THEN
C     LOW TEMPERATURES
        PL1T0 = PUL(12)
        PV1T0 = PUV(12)
        DO 31 I=11,1,-1
        PL1T0 = PL1T0*Y+PUL(I)
    31  PV1T0 = PV1T0*Y+PUV(I)
    ELSE
C     NEAR CRITICAL TEMPERATURES
        PL1T0 = PT0C
        = + Y*(17.63135647D-0-Y*(97.95806903D-0-Y*451.8053522D-0))
        PV1T0 = PT0C
        = + Y*(16.00172011D-0-Y*(439.3063303D-0-Y*3359.729201D-0))
    ENDIF
    RETURN
    END

```

#### F.14 INR105.AQUA.FORT(SSCEV)

```

    SUBROUTINE SSCEV(T,CVL,CVV)
C     SSCEV RETURNS - FOR A GIVEN TEMPERATURE T - APPROXIMATED VALUES
C     OF THE CONSTANT VOLUME HEAT CAPACITY
C     IN THE SATURATED LIQUID ( CVL ) AND
C     IN THE SATURATED VAPOR ( CVV ) IN RGAS UNITS.          94/12/01
    IMPLICIT REAL*8(A-H,O-Z)
    DIMENSION PUL(12),PUV(12)
    PARAMETER(TC=647.126D0,ALF=-1.0D-1)
    DATA PUL/7.4305055D-0,-24.93618016D-0,195.5654567D-0
    =,1.986485797D+3,-53.30543411D+3,505.6971723D+3,-2.724774677D+6
    =,9.167737673D+6,-19.62203378D+6,25.98472533D+6,-19.41943135D+6
    =,6.263206554D+6/
    DATA PUV/8.956404735D-0,-33.97230774D-0,126.5786602D-0
    =,-513.7462523D-0,1.521138693D+3,-2.182266721D+3,-73.57828067D-0
    =,4.483361889D+3,-5.407324042D+3,2.099821164D+3,0.D0,0.D0/
    IF(T .GE. TC)      THEN
        CVL = 0.D0
        CVV = 0.D0
        RETURN
    ENDIF
    Y = (TC-T)/TC
    IF(T .LE. 620.D0)      THEN
C     LOW TEMPERATURES
        CVL = PUL(12)
        CVV = PUV(12)
        DO 31 I=11,1,-1
        CVL = CVL*Y+PUL(I)
    31  CVV = CVV*Y+PUV(I)
    ELSE
C     NEAR CRITICAL PART OF THE SATURATION LINE

```

```

      BEG = Y**ALF                               SSCEV
      PSIL = 4.5D-0 +                             SSCEV
=      Y*(15.21452007D-0-Y*(167.2951053D-0-Y*1.042333618D+3)) SSCEV
      PSIV = 5.1D-0 +                             SSCEV
=      Y*(42.35451412D-0-Y*(1.002675527D+3-Y*6.911203183D+3)) SSCEV
      CVL = PSIL*BEG                               SSCEV
      CVV = PSIV*BEG                               SSCEV
ENDIF                                             SSCEV
RETURN                                           SSCEV
END                                              SSCEV

```

### F.15 INR105.AQUA.FORT(SSGIS)

```

      SUBROUTINE SSGIS(T,SL,GL)                   SSGIS
C      SSGIS RETURNS APPROXIMATED VALUES FOR THE ENTROPY AND FOR SSGIS
C      THE GIBBS-FUNCTION OF THE SATURATED LIQUID SSGIS
C      AS FUNCTIONS OF THE INPUT TEMPERATURE T. SSGIS
C      SL = S.LIQUID/RGAS , GL = G.LIQUID/(RGAS*T) 94/06/10 SSGIS
      IMPLICIT REAL*8(A-H,O-Z)                   SSGIS
      DIMENSION PSL(9),PGL(9)                   SSGIS
      PARAMETER(TC=647.126D0,SC=9.430D0,BR=0.25D0,BET=0.325D0) SSGIS
      DATA PSL/952.8621116D-3,272.5494325D-3,-8.457375782D-0 SSGIS
=      ,44.57631748D-0,-150.7441559D-0,311.7914429D-0,-391.8528748D-0 SSGIS
=      ,273.5529633D-0,-82.2848053D-0/ SSGIS
      DATA PGL/-2.569516966D-0,5.187876405D-0,640.8677408D-3 SSGIS
=      ,1.374106295D-0,-18.01831251D-0,68.53337198D-0,-176.2024879D-0 SSGIS
=      ,226.296896D-0,-130.4997179D-0/ SSGIS
      IF(T .GE. TC) THEN SSGIS
          SL = 0.D0 SSGIS
          GL = 0.D0 SSGIS
          RETURN SSGIS
      ENDIF SSGIS
      Y = (TC-T)/TC SSGIS
C      ENTROPY SSGIS
      IF(T .LE. 620.D0) THEN SSGIS
C      LOW TEMPERATURES SSGIS
          W = SQRT(Y) SSGIS
          OMV = PSL(9) SSGIS
          DO 31 K = 8,1,-1 SSGIS
31      OMV = OMV*W+PSL(K) SSGIS
      ELSE SSGIS
C      NEAR CRITICAL TEMPERATURES SSGIS
          BEG = BR*Y**BET SSGIS
          AV = -Y*(1.23383958D-0+Y*(1.656615363D-0-Y*Y*911.5375812D-0)) SSGIS
          OMV = 1.D0 - BEG + AV SSGIS
      ENDIF SSGIS
      SL = OMV*SC SSGIS
C      GIBBS-FUNCTION SSGIS
      GL = PGL(9) SSGIS
          DO 71 K = 8,1,-1 SSGIS
71      GL = GL*Y+PGL(K) SSGIS
      RETURN SSGIS
      END SSGIS

```



## F.16 INR105.AQUA.FORT(SSETA)

```

SUBROUTINE SSETA(T,ETL,ETV)                                SSETA
C  SSETA RETURNS - FOR A GIVEN TEMPERATURE T - APPROXIMATED VALUES SSETA
C  OF THE DYNAMIC VISCOSITY                               SSETA
C  IN THE SATURATED LIQUID ( ETL ) AND                  SSETA
C  IN THE SATURATED VAPOR ( ETV ) IN PA*SEC.           94/12/01 SSETA
IMPLICIT REAL*8(A-H,O-Z)                                  SSETA
DIMENSION PUL(11),PUV(11)                                SSETA
PARAMETER(TC=647.126D0,ETC=3.95D-5,BET=0.325D0,BE=1.50D0) SSETA
DATA PUL/945.822019D-3,-3.50567042D-0,23.61506749D-0      SSETA
=,-176.2217806D-0,960.5599634D-0,-3.533866457D+3,8.609716158D+3 SSETA
=,-13.6377187D+3,13.43590472D+3,-7.453116048D+3,1.776272412D+3/ SSETA
DATA PUV/973.3280423D-3,-4.061585251D-0,20.72025593D-0   SSETA
=,-80.06616536D-0,214.64856D-0,-384.6194719D-0,433.0818384D-0 SSETA
=,-278.1018786D-0,78.25010044D-0,0.D0,0.D0/             SSETA
DATA UN/1.D0/                                             SSETA
IF(T .GE. TC)      THEN                                  SSETA
    ETL=ETC                                               SSETA
    ETV=ETC                                               SSETA
    RETURN                                                SSETA
ENDIF                                                      SSETA
Y = (TC-T)/TC                                           SSETA
CHI = T/(T-TC)                                          SSETA
IF(T .LE. 620.D0)  THEN                                  SSETA
C  LOW TEMPERATURES                                     SSETA
    W = SQRT(Y)                                          SSETA
    FL = PUL(11)                                         SSETA
    FV = PUV(11)                                         SSETA
    DO 31 K = 10,1,-1                                    SSETA
    FL = FL*W+PUL(K)                                     SSETA
31  FV = FV*W+PUV(K)                                    SSETA
C  VISCOSITY OF THE LIQUID ( IN PA*SEC )               SSETA
    ETL=ETC/FL                                           SSETA
C  VISCOSITY OF THE VAPOR ( IN PA*SEC )               SSETA
    ETV=ETC*FV                                           SSETA
C  NEAR CRITICAL TEMPERATURES                         SSETA
ELSE                                                       SSETA
C  NEAR CRITICAL TEMPERATURES                         SSETA
    BEG = BE*Y**BET                                       SSETA
    AL = Y*(4.669845392D-0-Y*(19.58930391D-0-Y*Y*1.768652432D+3)) SSETA
    ETL=ETC*( UN + BEG + AL )                             SSETA
    AV = Y*(4.262680597D-0-Y*(34.89896734D-0-Y*Y*2.452105565D+3)) SSETA
    ETV=ETC*( UN - BEG + AV )                             SSETA
ENDIF                                                      SSETA
RETURN                                                    SSETA
END                                                       SSETA

```

## F.17 INR105.AQUA.FORT(SSLAM)

```

SUBROUTINE SSLAM(T,XLL,XLV)                                SSLAM
C  SSLAM RETURNS - FOR A GIVEN TEMPERATURE T - APPROXIMATED VALUES SSLAM
C  OF THE THERMAL CONDUCTIVITY                            SSLAM
C  IN THE SATURATED LIQUID ( XLL ) AND                   SSLAM
C  IN THE SATURATED VAPOR ( XLV ) IN W/(M*K).           94/12/01 SSLAM
IMPLICIT REAL*8(A-H,O-Z)                                  SSLAM
DIMENSION PUL(9),PUV(9)                                   SSLAM
PARAMETER(TC=647.126D0,XLC=1.D+70,ALA=-0.50D0)           SSLAM
DATA PUL/397.252283D-3,1.361430056D-0,288.6579476D-3     SSLAM
=,-1.137153088D-0,-75.20016424D-0,333.88123D-0,-546.0632435D-0 SSLAM
=,311.575529D-0,0.D0/                                     SSLAM
DATA PUV/201.2600289D-3,-2.546881317D-0,19.24073027D-0   SSLAM
=,-79.8925815D-0,178.6322576D-0,-195.5137779D-0,62.06051011D-0 SSLAM
=,38.86696488D-0,-15.11624233D-0/                         SSLAM
IF(T .GE. TC)      THEN                                    SSLAM
  XLL=XLC          SSLAM
  XLV=XLC          SSLAM
  RETURN           SSLAM
ENDIF              SSLAM
Y=(TC-T)/TC       SSLAM
IF(T .LE. 620.D0) THEN                                    SSLAM
C  LOW TEMPERATURES                                       SSLAM
  XLL = PUL(9)     SSLAM
  XLV = PUV(9)     SSLAM
  DO 31 I=8,1,-1  SSLAM
    XLL = XLL*Y+PUL(I) SSLAM
31    XLV = XLV*Y+PUV(I) SSLAM
  ELSE             SSLAM
C  NEAR CRITICAL TEMPERATURES                             SSLAM
  BEG=Y**ALA       SSLAM
  AL = 20.D-3 +Y*( 2.484618723D-0 -Y*( 31.16080409D-0     SSLAM
=  -Y*( 421.0825394D-0 -Y*2390.60876D-0 )))              SSLAM
  XLL=BEG*AL       SSLAM
  AV = 20.D-3 +Y*( 722.1916099D-3 -Y*( 29.16989349D-0     SSLAM
=  -Y*( 495.5936588D-0 -Y*3169.252794D-0 )))              SSLAM
  XLV=BEG*AV       SSLAM
ENDIF              SSLAM
RETURN             SSLAM
END                SSLAM

```

## F.18 INR105.AQUA.FORT(DILE)

```

FUNCTION DILE(Y)                                          DILE
C  RETURNS AN ADDITIVE TERM FOR PSL(T) TO DESCRIBE THE PRESSURE DILE
C  OF THE LIQUID SPINODAL ON A HGK-ISOTHERM.           94/07/12 DILE
C  DILE IS A FUNCTION OF THE SCALED TEMPERATURE Y=1-T/TC DILE
DILE = (40.49363544D+0 -Y*(20.73773694D+3 -Y*(7.607260219D+6 DILE
= -Y*(1.796520658D+9 -Y*(258.928784714D+9 -Y*(22.71838331D+12 DILE
= -Y*(1.182066587D+15 -Y*(33.45912597D+15 -Y*396.4473916D+15 DILE
= )))))))**Y                                           DILE

```

```

RETURN                                     DILE
END                                         DILE

```

**F.19 INR105.AQUA.FORT(DIVE)**

```

FUNCTION DIVE(Y)                           DIVE
C RETURNS AN ADDITIVE TERM FOR PSV(T) TO DESCRIBE THE PRESSURE DIVE
C OF THE VAPOR SPINODAL ON A HGK-ISOTHERM. 94/07/12 DIVE
C DIVE IS A FUNCTION OF THE SCALED TEMPERATURE Y=1-T/TC DIVE
DIVE = Y*(17.17830396D+0 -Y*(3.434969255D+3 DIVE
= -Y*(448.795607300D+3 -Y*(3.808457011D+7 -Y*(1.646319316D+9 DIVE
= -Y*2.719071374D+10)))))) DIVE
RETURN DIVE
END DIVE

```

**F.20 INR105.AQUA.FORT(PLMAX)**

```

FUNCTION PLMAX(T)                           PLMAX
C PLMAX RETURNS THE PRESSURE IN PA-S ON THE WHOLE ISOCHORE PLMAX
C RHO = 1200 KG/M**3 PLMAX
C AS FUNCTION OF THE INPUT TEMPERATURE T 95/01/18 PLMAX
IMPLICIT REAL*8(A-H,O-Z) PLMAX
IF(T .LE. 373.15D0) THEN PLMAX
PLMAX = 128.0317635D+9 -T*( 1.503427271D+9 PLMAX
= -T*( 6.608885188D+6 -T*( 12.83080180D+3 -T*9.305682808D-0 )) PLMAX
ELSE IF(T .GT. 373.15D0 .AND. T .LE. 647.126D0) THEN PLMAX
PLMAX = 1.121372093D+9 -T*( 6.278371391D+6 PLMAX
= -T*( 26.44577782D+3 -T*( 33.36474460D-0 -T*14.81761478D-3 )) PLMAX
ELSE IF(T .GT. 647.126D0) THEN PLMAX
PLMAX = -731.6836907D+6 +T*( 6.597512847D+6 PLMAX
= -T*( 7.08407094D+3 -T*( 5.764846806D-0 -T*( 3.188721412D-3 PLMAX
= -T*( 1.193485372D-6 -T*( 2.946300976D-10 -T*( 4.575561580D-14 PLMAX
= -T*( 4.040003535D-18 -T*1.543953233D-22 )))))))) PLMAX
ENDIF PLMAX
RETURN PLMAX
END PLMAX

```

**F.21 INR105.AQUA.FORT(P1200)**

```

FUNCTION P800(T)                             P800
C P800 RETURNS THE PRESSURE IN PA-S ON THE SUPERCRITICAL PART P800
C OF THE ISOCHORE RHO = 800 KG/M**3 P800
C AS FUNCTION OF THE INPUT TEMPERATURE T 95/01/18 P800
P800 = -576.9907617D+6 +T*( 229.0441320D+3 P800
= +T*( 3.275463692D+3 -T*( 4.331244582D-0 -T*( 3.008662103D-3 P800
= -T*( 1.272178994D-6 -T*( 337.938642D-12 -T*( 55.07102702D-15 P800
= -T*( 5.03025311D-18 -T*197.1414663D-24 )))))))) P800
RETURN P800
END P800

```

## F.22 INR105.AQUA.FORT(P600)

```
FUNCTION P600(T) P600
C P600 RETURNS THE PRESSURE IN PA-S ON THE SUPERCRITICAL PART P600
C OF THE ISOCHORE RHO = 600 KG/M**3 P600
C AS FUNCTION OF THE INPUT TEMPERATURE T 95/01/18 P600
P600 = -193.5182531D+6 -T*( 662.0824295D+3 P600
= -T*( 3.041300733D+3 -T*( 3.331588396D-0 -T*( 2.091973967D-3 P600
= -T*( 822.3881849D-9 -T*( 205.8600153D-12 -T*( 31.88132588D-15 P600
= -T*( 2.784677333D-18 -T*104.8772468D-24 ))))))) P600
RETURN P600
END P600
```

## F.23 INR105.AQUA.FORT(P400)

```
FUNCTION P400(T) P400
C P400 RETURNS THE PRESSURE IN PA-S ON THE SUPERCRITICAL PART P400
C OF THE ISOCHORE RHO = 400 KG/M**3 P400
C AS FUNCTION OF THE INPUT TEMPERATURE T 95/01/18 P400
P400 = -48.22391500D+6 -T*( 534.0793594D+3 P400
= -T*( 1.893391113D+3 -T*( 2.052518308D-0 -T*( 1.304520061D-3 P400
= -T*( 521.2344512D-9 -T*( 132.5272444D-12 -T*( 20.80959374D-15 P400
= -T*( 1.839373352D-18 -T* 69.98460931D-24 ))))))) P400
RETURN P400
END P400
```

## F.24 INR105.AQUA.FORT(P322)

```
FUNCTION P322(T) P322
C P322 RETURNS THE PRESSURE IN PA-S ON THE SUPERCRITICAL PART P322
C OF THE CRITICAL ISOCHORE P322
C AS FUNCTION OF THE INPUT TEMPERATURE T 95/01/18 P322
P322 = -115.3411242D+6 -T*( 14.61390831D+3 P322
= -T*( 713.8479434D-0 -T*( 843.0259976D-3 -T*( 558.6511529D-6 P322
= -T*( 228.9982666D-9 -T*( 59.25205508D-12 -T*( 9.423230731D-15 P322
= -T*( 841.0373279D-21 -T*32.24390358D-24 ))))))) P322
RETURN P322
END P322
```

## F.25 INR105.AQUA.FORT(P250)

```
FUNCTION P250(T) P250
C P250 RETURNS THE PRESSURE IN PA-S ON THE SUPERCRITICAL PART P250
C OF THE ISOCHORE RHO = 250 KG/M**3 P250
C AS FUNCTION OF THE INPUT TEMPERATURE T 95/01/18 P250
P250 = -161.604255D+6 +T*( 370.3861090D+3 -T*( 183.5936079D+0 P250
= -T*( 93.96635527D-3 -T*( 30.07808886D-6 -T*( 5.819644183D-9 P250
= -T*( 619.724735D-15 -T*27.81950727D-18 )))) P250
RETURN P250
END P250
```

## Appendix G. Secondary routines

The following routines either organize the calculation of the thermal properties of the water, or help to find a density to a given pressure.

### G.1 INR105.AQUA.FORT(DGFIND)

```

SUBROUTINE DGFIND(T,Q,DX,IECCH)                                DGFIND
C  DGFIND SEEKS A DENSITY DX (G/CM**3)                        DGFIND
C  CORRESPONDING TO THE PRESSURE Q(MPA) AND                  DGFIND
C  TO THE - SUPERCRITICAL - TEMPERATURE T (K).              94/12/06 DGFIND
IMPLICIT REAL*8(A-H,O-Z)                                       DGFIND
COMMON /TEMPO/  RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI       DGFIND
COMMON /BASIF/  GB,SB,ZB,PB1T0,PB1R0,CVB                       DGFIND
COMMON /RESEF/  GR,SR,ZR,PR1T0,PR1R0,CVR                       DGFIND
REAL*8 FX(3),X(3)                                             DGFIND
PARAMETER(DC=0.322D0,DMAX=1.2D0,FINIS=5.D-5)                  DGFIND
DATA PF/1.D-6/,ZERO/0.D0/                                     DGFIND
C  RANGE CONTROL                                              DGFIND
QSMAX=PLMAX(T)*PF                                             DGFIND
IF(Q .GT. QSMAX)          IECCH=+2                            DGFIND
IF(IECCH .NE. 0)         GOTO 900                             DGFIND
QSMIN=P600(T)*PF                                               DGFIND
IF(Q .LT. QSMIN)         GO TO 200                             DGFIND
C  REGION OF HIGH DENSITIES                                  DGFIND
100 X(3)=DMAX                                                  DGFIND
X(1)=0.6D0                                                      DGFIND
X(2)=0.80D0                                                      DGFIND
FX(2)=P800(T)*PF                                               DGFIND
SAM=2.D-1                                                        DGFIND
      GOTO 600                                                  DGFIND
C  REGION OF MODERATE DENSITIES                              DGFIND
200 X(3)=0.6D0                                                  DGFIND
X(1)=DC                                                         DGFIND
QSMAX=QSMIN                                                      DGFIND
QSMIN=P322(T)*PF                                                DGFIND
IF(Q .LT. QSMIN)         GO TO 300                             DGFIND
X(2)=0.40D0                                                      DGFIND
FX(2)=P400(T)*PF                                               DGFIND
SAM=1.D-1                                                        DGFIND
      GOTO 600                                                  DGFIND
C  REGION OF LOW DENSITIES                                  DGFIND
300 X(3)=DC                                                      DGFIND
X(1)=ZERO                                                        DGFIND
QSMAX=QSMIN                                                      DGFIND
QSMIN=ZERO                                                       DGFIND
IF(Q .LT. QSMIN)          IECCH=-2                             DGFIND
IF(IECCH .NE. 0)         GOTO 900                             DGFIND
X(2)=0.25D0                                                      DGFIND
FX(2)=P250(T)*PF                                               DGFIND
SAM=5.D-2                                                        DGFIND
600 FX(1)=QSMIN                                                 DGFIND

```

```

      FX(3)=QSMAX
C     WRITE(6,4400) QSMIN,QSMAX,SAM
C     FIRST DENSITY APPROXIMATIONS
      LI=1
C     WRITE(6,6660)
      IF(DX .LE. ZERO)      DX=DINSU(T,Q,FX,X,LI,IECCH)
      IF(IECCH .NE. 0)      RETURN
C     REFINED DENSITY APPROXIMATIONS IN THE MANNER OF MR. I. NEWTON
C     WRITE(6,' ( )')
C     ITERATING
      LF=LI+19
      DO 71 L=LI,LF
      IF(DX .GT. DMAX)      THEN
          DX=DMAX
          IECCH=2
      ENDIF
      IF(DX .LE. ZERO)      DX=Q/RT
      CALL BASE(T,DX,IECCH)
      CALL RESID(T,DX)
      ZP = ZB + ZR
      QAX = ZP*DX*RT
      DELQ=Q-QAX
      EPS=ABS(DELQ/Q)
      IF(EPS .LT. FINIS)    THEN
          IECCH=0
          RETURN
      ENDIF
      IF(IECCH .NE. 0)      GOTO 900
      P1R0 = PB1R0+PR1R0
      DQD = P1R0*RT
      DELR = DELQ/DQD
C     WRITE(6,6666) L,DX,QAX,DELQ,FINIS,DELR
      IF(DELR .GT. ZERO)    DELR=MIN(DELR,SAM)
      IF(DELR .LT. ZERO)    DELR=MAX(DELR,-SAM)
      DX = DX + DELR
71    CONTINUE
      IECCH=3
      WRITE(6,9000)
      RETURN
900   IF(IECCH .GT. 0)      WRITE(6,9100) Q
      IF(IECCH .LT. 0)      WRITE(6,9300) Q
      RETURN
C4400 FORMAT(8H QSMIN =,F14.9,2X,7HQSMAX =,F14.9,2X,5HSAM =,2X,F7.3)
C6660 FORMAT(3H L,5X,2HDX,12X,3HQAX,13X,4HDELQ,11X,3HQGR,7X,4HDELR)
C6666 FORMAT(I3,F13.9,G18.9,G15.6,F9.5,F13.9)
9000  FORMAT(41H ITERATION FOR THE DENSITY VALUE FAILED ! )
9100  FORMAT(13H PRESSURE P =,G15.7,16H MPA TOO HIGH ! )
9300  FORMAT(13H PRESSURE P =,G15.7,16H MPA TOO LOW ! )
      END

```

## G.2 INR105.AQUA.FORT(DLFIND)

```

SUBROUTINE DLFIND(T,Q,DX,IECCH)                                DLFIND
C  DLFIND SEEKS A LIQUID DENSITY DX (G/CM**3)                DLFIND
C  CORRESPONDING TO THE PRESSURE Q(MPA) AND                  DLFIND
C  TO THE - SUBCRITICAL - TEMPERATURE T (K).                94/12/08 DLFIND
IMPLICIT REAL*8(A-H,O-Z)                                       DLFIND
COMMON /TEMPO/  RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI        DLFIND
COMMON /BASIF/  GB,SB,ZB,PB1T0,PB1R0,CVB                       DLFIND
COMMON /RESEF/  GR,SR,ZR,PR1T0,PR1R0,CVR                      DLFIND
COMMON /SATIF/  PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL    DLFIND
=,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV DLFIND
REAL*8 FX(3),X(3)                                             DLFIND
PARAMETER(TC=647.126D0,TCN=646.0D0,DC=0.322D0)                DLFIND
PARAMETER(DMAX=1.2D0,FINIS=1.D-4)                             DLFIND
DATA ZERO/0.D0/,PF/1.D-6/                                     DLFIND
CALL SUBCRI(T)                                                DLFIND
C  RANGE CONTROL                                             DLFIND
QSV = PSV*PF                                                  DLFIND
QSL = PSL*PF                                                  DLFIND
IF(T .GE. 635.D0)      THEN                                    DLFIND
  Y=1-T/TC                                                    DLFIND
  QSL = QSL + DILE(Y)                                         DLFIND
  QSV = QSV - DIVE(Y)                                         DLFIND
  IF(QSL .GE. QSV)    QSL=QSV-FINIS                          DLFIND
ENDIF                                                         DLFIND
QMAX = PLMAX(T)*PF                                           DLFIND
IF(Q .GT. QSV*1.3D0)    SAM=2.D-1                             DLFIND
IF(Q .LE. QSV*1.3D0)    SAM=1.D-1                             DLFIND
DMI=RHSL*1.D-3                                                DLFIND
DS0=(DMI-DC)*5.D-2                                           DLFIND
C  VICINITY OF THE CRITICAL POINT ?                            DLFIND
IF(T .GE. TCN .AND. Q .LE. QSV*1.005D0)    DMI=DC*0.965D0  DLFIND
C  WRITE(6,4400) QSL,QSV,DMI,SAM                              DLFIND
IF(Q .GT. QMAX)        IECCH=+2                                DLFIND
IF(Q .LT. QSL)         IECCH=-2                                DLFIND
IF(IECCH .NE. 0)      GOTO 900                                DLFIND
C  FIRST DENSITY APPROXIMATIONS                               DLFIND
C  WRITE(6,6660)                                             DLFIND
LI=1                                                           DLFIND
IF(DX .LE. ZERO)      THEN                                    DLFIND
  X(3)=DMAX                                                    DLFIND
  FX(3)=QMAX                                                    DLFIND
  X(2)=DL                                                       DLFIND
  FX(2)=PS*PF                                                  DLFIND
  X(1)=DMI                                                       DLFIND
  FX(1)=QSL                                                    DLFIND
  DX=DINLI(T,Q,FX,X,LI,IECCH)                                  DLFIND
  IF(IECCH .NE. 0)    RETURN                                    DLFIND
ENDIF                                                         DLFIND
C  REFINED DENSITY APPROXIMATIONS IN THE MANNER OF MR. I. NEWTON DLFIND
C  WRITE(6,'( )')                                           DLFIND
LF=LI+19                                                       DLFIND
DO 71 L=LI,LF                                                  DLFIND

```



IF(DX .GT. DMAX)	THEN	DLFIND
DX=DMAX		DLFIND
IECCH=2		DLFIND
ENDIF		DLFIND
IF(DX .LT. DMI)	DX=DMI	DLFIND
CALL BASE(T,DX,IECCH)		DLFIND
CALL RESID(T,DX)		DLFIND
ZP = ZB + ZR		DLFIND
QAX = ZP*DX*RT		DLFIND
DELQ=Q-QAX		DLFIND
EPS=ABS(DELQ)		DLFIND
IF(EPS .LT. FINIS)	THEN	DLFIND
IECCH=0		DLFIND
RETURN		DLFIND
ENDIF		DLFIND
IF(IECCH .GT. 0)	GO TO 900	DLFIND
P1R0 = PB1R0+PR1R0		DLFIND
DQD = P1R0*RT		DLFIND
IF(DQD .LE. ZERO)	THEN	DLFIND
DX=DX+DS0		DLFIND
GO TO 71		DLFIND
ENDIF		DLFIND
DELR = DELQ/DQD		DLFIND
C WRITE(6,6666) L,DX,QAX,DELQ,FINIS,DELR		DLFIND
IF(ABS(DELR) .LT. 1.D-6)	THEN	DLFIND
IECCH=0		DLFIND
RETURN		DLFIND
ENDIF		DLFIND
IF(DELR .GT. ZERO)	DELR=MIN(DELR,SAM)	DLFIND
IF(DELR .LT. ZERO)	DELR=MAX(DELR,-SAM)	DLFIND
DX = DX + DELR		DLFIND
71 CONTINUE		DLFIND
IECCH=3		DLFIND
800 WRITE(6,9000)		DLFIND
RETURN		DLFIND
900 IF(IECCH .GT. 0)	WRITE(6,9100) Q	DLFIND
IF(IECCH .LT. 0)	WRITE(6,9300) Q	DLFIND
RETURN		DLFIND
C4400 FORMAT(7H QSPL =,F14.9,3X,6HQSPV =,F14.9,3X,5HDMI =,3X,F10.6,3X		DLFIND
C =,5HSAM =,F7.3 )		DLFIND
C6660 FORMAT(3H L,5X,2HDX,12X,3HQAX,13X,4HDELQ,11X,3HQGR,7X,4HDELR)		DLFIND
C6666 FORMAT(I3,F13.9,G18.9,G15.6,F9.5,F13.9)		DLFIND
9000 FORMAT(41H ITERATION FOR THE DENSITY VALUE FAILED !)		DLFIND
9100 FORMAT(13H PRESSURE P =,G15.7,16H MPA TOO HIGH ! )		DLFIND
9300 FORMAT(13H PRESSURE P =,G15.7,16H MPA TOO LOW ! )		DLFIND
END		DLFIND

### G.3 INR105.AQUA.FORT(DVFIND)

```

SUBROUTINE DVFIND(T,Q,DX,IECCH)                                DVFIND
C  DVFIND SEEKS A VAPOR DENSITY DX (G/CM**3)                  DVFIND
C  CORRESPONDING TO THE PRESSURE Q(MPA) AND                   DVFIND
C  TO THE - SUBCRITICAL - TEMPERATURE T (K).                 94/12/08 DVFIND
IMPLICIT REAL*8(A-H,O-Z)                                       DVFIND
COMMON /TEMPO/  RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI        DVFIND
COMMON /BASIF/  GB,SB,ZB,PB1T0,PB1R0,CVB                       DVFIND
COMMON /RESEF/  GR,SR,ZR,PR1T0,PR1R0,CVR                       DVFIND
COMMON /SATIF/  PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL    DVFIND
=,RHVS,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV DVFIND
REAL*8 FX(3),X(3)                                              DVFIND
PARAMETER(TC=647.126D0,TCN=646.0D0,DC=0.322D0,FINIS=5.D-5)    DVFIND
DATA ZERO/0.D0/,PF/1.D-6/                                     DVFIND
CALL SUBCRI(T)                                                DVFIND
C  RANGE CONTROL                                             DVFIND
QSL = PSL*PF                                                  DVFIND
QSV = PSV*PF                                                  DVFIND
IF(Q .LT. QSL*0.95D0)  SAM=1.D-1                             DVFIND
IF(Q .GE. QSL*0.95D0)  SAM=5.D-2                             DVFIND
DMX=RHSV*1.D-3                                               DVFIND
C  VICINITY OF THE CRITICAL POINT ?                          DVFIND
IF(T .GE. TCN .AND. Q .GE. QSL*0.9999D0)  SAM=5.D-2         DVFIND
IF(T .GE. 635.D0)    THEN                                     DVFIND
    Y=1-T/TC                                                  DVFIND
    QSV = QSV - DIVE(Y)                                       DVFIND
ENDIF                                                         DVFIND
C  WRITE(6,4400) QSL,QSV,DMX,SAM                             DVFIND
IF(Q .GT. QSV)        IECCH=+2                                 DVFIND
IF(Q .LT. ZERO)       IECCH=-2                                 DVFIND
IF(IECCH .NE. 0)     GO TO 900                                 DVFIND
C  FIRST DENSITY APPROXIMATIONS                               DVFIND
C  WRITE(6,6660)                                             DVFIND
LI=1                                                           DVFIND
IF(DX .LE. ZERO)     THEN                                     DVFIND
    X(3)=DMX                                                  DVFIND
    FX(3)=QSV                                                  DVFIND
    X(2)=DV                                                    DVFIND
    FX(2)=PS*1.D-6                                           DVFIND
    X(1)=ZERO                                                  DVFIND
    FX(1)=ZERO                                                  DVFIND
    DX=DINVA(T,Q,FX,X,LI,IECCH)                               DVFIND
    IF(IECCH .NE. 0)  RETURN                                  DVFIND
ENDIF                                                         DVFIND
C  REFINED DENSITY APPROXIMATIONS IN THE MANNER OF MR. I. NEWTON DVFIND
C  WRITE(6,'( )' )                                           DVFIND
LF=LI+19                                                       DVFIND
IF(DX .GT. DMX)       DX=DMX                                   DVFIND
DO 71 L=LI,LF                                                 DVFIND
IF(DX .LE. ZERO)     DX=Q/RT                                  DVFIND
CALL BASE(T,DX,IECCH)                                         DVFIND
CALL RESID(T,DX)                                              DVFIND
ZP = ZB + ZR                                                  DVFIND

```



```

REAL*8 FX(3),X(3)                                DINSU
PARAMETER(TC2=648.126D0,DUG=0.122D0,DOG=0.522D0,DMAX=1.2D0) DINSU
DATA ZERO/0.D0/                                  DINSU
FINIS=1.D-2                                       DINSU
C ITERATING                                       DINSU
C QUADRATIC INTERPOLATION TO THE POINTS 1 2 3    DINSU
50 CALL QUADO(FX,X,Q,DX,JECCH)                   DINSU
IF(DX .GT. DMAX) DX=DMAX                         DINSU
IF(DX .LE. ZERO) DX=1.D-6                       DINSU
C SELECTING THE NEW PILLARS                       DINSU
IF(IT .EQ. 1) THEN                               DINSU
IF(T .LT. TC2 .AND.                             DINSU
= (DX .GT. DUG .AND. DX .LT. DOG)) FINIS=1.D-3 DINSU
ELSE                                             DINSU
IF(IT .GT. 3 .OR. EPS .LT. FINIS) GO TO 900    DINSU
ENDIF                                           DINSU
KI=0                                             DINSU
CALL SORBET(KI,FX,X,Q,DX)                       DINSU
IF(KI .EQ. 2 .OR. JECCH .NE. 0) GO TO 900      DINSU
C PRESSURE AT THE NEW DENSITY VALUE              DINSU
52 DELR=DX-DXA                                   DINSU
IF(ABS(DELR) .LT. 1.D-4) GO TO 900             DINSU
CALL BASE(T,DX,IECCH)                           DINSU
CALL RESID(T,DX)                                DINSU
ZP = ZB + ZR                                    DINSU
FX(KI)= ZP*DX*RT                                DINSU
DELQ=FX(KI)-Q                                   DINSU
EPS=ABS(DELQ)                                   DINSU
DXA=DX                                           DINSU
C WRITE(6,6666) IT,DX,FX(KI),DELQ,FINIS,DELR   DINSU
IT=IT+1                                          DINSU
GO TO 50                                         DINSU
900 DINSU=DX                                     DINSU
6666 FORMAT(I3,F13.9,G18.9,G15.6,F9.5,F13.9)    DINSU
999 RETURN                                       DINSU
END                                              DINSU

```

## G.5 INR105.AQUA.FORT(DINLI)

```

FUNCTION DINLI(T,Q,FX,X,IT,IECCH)                DINLI
C DINLI IS AN INITIAL DENSITY FOR THE ROUTINE DLFIND. DINLI
C DINLI IS THE ROOT OF A 2. GRADE EQUATION Q.IS(DX)-Q = 0 DINLI
C PRESSURES ARE IN MPA-S, DENSITIES IN G/CM**3-S. 93/09/03 DINLI
IMPLICIT REAL*8(A-H,O-Z)                        DINLI
COMMON /TEMPO/ RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI DINLI
COMMON /BASIF/ GB,SB,ZB,PT0B,PROB,CVB          DINLI
COMMON /RESEF/ GR,SR,ZR,PT0R,PR0R,CVR          DINLI
REAL*8 FX(3),X(3)                              DINLI
PARAMETER(TC1=646.126D0,DOG=0.522D0)          DINLI
FINIS=1.D-2                                     DINLI
DMX=X(3)                                        DINLI
C ITERATING                                       DINLI
C QUADRATIC INTERPOLATION TO THE POINTS 1 2 3    DINLI

```

```

50 CALL QUADO(FX,X,Q,DX,JECCH)          DINLI
   IF(DX .GT. DMX)      DX=DMX          DINLI
C   SELECTING THE NEW PILLARS            DINLI
   IF(IT .EQ. 1)      THEN              DINLI
     IF(T .GT. TC1)   THEN              DINLI
       KI=1            DINLI
     IF(DX .LT. DOG)          FINIS=1.D-3  DINLI
     ELSE                    DINLI
       KI=0            DINLI
     ENDIF                  DINLI
   ENDIF                    DINLI
   CALL SORBET(KI,FX,X,Q,DX)            DINLI
   IF(KI .EQ. 2 .OR. JECCH .NE. 0)      GO TO 900  DINLI
   IF(IT .LE. 1)      GO TO 52          DINLI
   IF(IT .GT. 3 .OR. EPS .LT. FINIS)    GO TO 900  DINLI
C   PRESSURE AT THE NEW DENSITY VALUE    DINLI
52 DELR=DX-DXA                      DINLI
   IF(ABS(DELQ) .LT. 1.D-4)            GO TO 900  DINLI
   CALL BASE(T,DX,IECCH)                DINLI
   CALL RESID(T,DX)                      DINLI
   ZP = ZB + ZR                          DINLI
   FX(KI)= ZP*DX*RT                      DINLI
   DELQ=FX(KI)-Q                          DINLI
   EPS=ABS(DELQ)                          DINLI
   DXA=DX                                  DINLI
C   WRITE(6,6666) IT,DX,FX(KI),DELQ,FINIS,DELR  DINLI
   IT=IT+1                                DINLI
   GO TO 50                                DINLI
900 DINLI=DX                              DINLI
6666 FORMAT(I3,F13.9,G18.9,G15.6,F9.5,F13.9)  DINLI
   RETURN                                  DINLI
   END                                      DINLI

```

## G.6 INR105.AQUA.FORT(DINVA)

```

FUNCTION DINVA(T,Q,FX,X,IT,IECCH)      DINVA
C   DINVA IS AN INITIAL DENSITY FOR THE ROUTINE DVFIND.  DINVA
C   DINVA IS THE ROOT OF A 2. GRADE EQUATION Q.IS(DX)-Q = 0.  DINVA
C   PRESSURES ARE IN MPA-S, DENSITIES IN G/CM**3-S.          93/09/17  DINVA
   IMPLICIT REAL*8(A-H,O-Z)            DINVA
   COMMON /TEMPO/  RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI  DINVA
   COMMON /BASIF/  GB,SB,ZB,PT0B,PR0B,CVB                    DINVA
   COMMON /RESEF/  GR,SR,ZR,PT0R,PR0R,CVR                    DINVA
   REAL*8 FX(3),X(3)                                          DINVA
   PARAMETER(TC1=646.126D0,DUG=0.122D0)                       DINVA
   FINIS=1.D-2                                                DINVA
C   ITERATING                                                  DINVA
C   QUADRATIC INTERPOLATION TO THE POINTS 1 2 3              DINVA
50 CALL QUADO(FX,X,Q,DX,JECCH)          DINVA
   IF(DX .LT. 0.D0)      DX=1.D-6          DINVA
C   SELECTING THE NEW PILLARS            DINVA
   IF(IT .EQ. 1)      THEN              DINVA
     IF(T .GT. TC1)   THEN              DINVA

```

	KI=3		DINVA
	IF(DX .GT. DUG)	FINIS=1.D-3	DINVA
	ELSE		DINVA
	KI=0		DINVA
	ENDIF		DINVA
	ENDIF		DINVA
	CALL SORBET(KI,FX,X,Q,DX)		DINVA
	IF(KI .EQ. 2 .OR. JECCH .NE. 0)	GO TO 900	DINVA
	IF(IT .LE. 1)	GO TO 52	DINVA
	IF(IT .GT. 3 .OR. EPS .LT. FINIS)	GO TO 900	DINVA
C	PRESSURE AT THE NEW DENSITY VALUE		DINVA
	52 DELR=DX-DXA		DINVA
	IF(ABS(DELR) .LT. 1.D-4)	GO TO 900	DINVA
	CALL BASE(T,DX,IECCH)		DINVA
	CALL RESID(T,DX)		DINVA
	ZP = ZB + ZR		DINVA
	FX(KI)= ZP*DX*RT		DINVA
	DELQ=FX(KI)-Q		DINVA
	EPS=ABS(DELQ/Q)		DINVA
	DXA=DX		DINVA
C	WRITE(6,6666) IT,DX,FX(KI),DELQ,FINIS,DELR		DINVA
	IT=IT+1		DINVA
	GO TO 50		DINVA
	900 DINVA=DX		DINVA
	6666 FORMAT(I3,F13.9,G18.9,G15.6,F9.5,F13.9)		DINVA
	999 RETURN		DINVA
	END		DINVA

## G.7 INR105.AQUA.FORT(QUADO)

	SUBROUTINE QUADO(FX,X,FXS,XS,JECCH)		QUADO
C	SEARCHES AN XS WITH THE PROPERTY $FX(XS) = FXS$	93/09/07	QUADO
C	BY FITTING A QUADRATIC POLYNOMIAL TO THE POINTS $FX(1) - FX(3)$		QUADO
	IMPLICIT REAL*8(A-H,O-Z)		QUADO
	REAL*8 FX(3),X(3),FY(3)		QUADO
	DATA ZERO,UN/0.00,1.00/		QUADO
	CALL MONIKA(FX,X,JECCH)		QUADO
	IF(JECCH .NE. 0) RETURN		QUADO
	A=ZERO		QUADO
	B=ZERO		QUADO
	C=ZERO		QUADO
	IECCH=0		QUADO
C	THE COEFFICIENTS OF THE POLYNOMIAL		QUADO
	DO 21 K=1,3		QUADO
	21 FY(K)=FX(K)-FXS		QUADO
	DO 31 I=1,3		QUADO
	J=I+1		QUADO
	IF(J .GT. 3) J=J-3		QUADO
	XD=X(J)-X(I)		QUADO
	K=J+1		QUADO
	IF(K .GT. 3) K=K-3		QUADO
	YXD= FY(K)*XD		QUADO
	YX2=-YXD*(X(J)+X(I))		QUADO

	YX3= YXD*X(J)*X(I)		QUADO
	A=A+YXD		QUADO
	B=B+YX2		QUADO
	C=C+YX3		QUADO
31	CONTINUE		QUADO
C	THE ROOTS OF THE POLYNOMIAL		QUADO
	XG=-B/(A+A)		QUADO
	US=UN-C/(A*XG*XG)		QUADO
	IF(US .LE. ZERO) THEN		QUADO
	XS=XG		QUADO
	RETURN		QUADO
	ENDIF		QUADO
	W=SQRT(US)		QUADO
	XA=XG*(UN+W)		QUADO
	XB=XG*(UN-W)		QUADO
C	THE APPROPRIATE ROOT		QUADO
80	XM=(X(1)+X(2)+X(3))/3.D0		QUADO
	XS=XA		QUADO
	IF(ABS(XB-XM) .LT. ABS(XA-XM))	XS=XB	QUADO
	RETURN		QUADO
	END		QUADO

## G.8 INR105.AQUA.FORT(MONIKA)

	SUBROUTINE MONIKA(FX,X,JECCH)		MONIKA
C	ARRANGES THE POINTS FX,X TO A SET WITH		MONIKA
C	MONOTONOUS ABSCISSAS	93/09/07	MONIKA
	IMPLICIT REAL*8(A-H,0-Z)		MONIKA
	DIMENSION X(3),FX(3),W(3),FW(3)		MONIKA
	IF((X(1) .EQ. X(2)) .OR. (X(2) .EQ. X(3)) .OR.		MONIKA
	= (X(1) .EQ. X(3))) THEN		MONIKA
	JECCH = 1		MONIKA
	RETURN		MONIKA
	ENDIF		MONIKA
	U=MIN(X(1),X(2),X(3))		MONIKA
	IF(U .EQ. X(1)) I=1		MONIKA
	IF(U .EQ. X(2)) I=2		MONIKA
	IF(U .EQ. X(3)) I=3		MONIKA
	J=I+1		MONIKA
	IF(J .GT. 3) J=J-3		MONIKA
	K=J+1		MONIKA
	IF(K .GT. 3) K=K-3		MONIKA
	V=MIN(X(J),X(K))		MONIKA
	IF(V .EQ. X(J)) GO TO 90		MONIKA
	JF=J		MONIKA
	J=K		MONIKA
	K=JF		MONIKA
	GO TO 92		MONIKA
90	IF(I .EQ. 1) RETURN		MONIKA
92	W(1)=X(I)		MONIKA
	FW(1)=FX(I)		MONIKA
	W(2)=X(J)		MONIKA
	FW(2)=FX(J)		MONIKA

W(3)=X(K)	MONIKA
FW(3)=FX(K)	MONIKA
DO 91 N=1,3	MONIKA
X(N)=W(N)	MONIKA
91 FX(N)=FW(N)	MONIKA
END	MONIKA

## G.9 INR105.AQUA.FORT(SORBET)

	SUBROUTINE SORBET(KI,FX,X,FXS,XS)	SORBET
C	REPLACES ONE OF THE BOUNDARY POINTS X(1) OR X(3) WITH XS.	SORBET
C	NORMALLY THE POINT WITH THE LARGER (FXS-FX) WILL BE REPLACED.	SORBET
C	IF 2 OF THE 3 POINTS ARE VERY NEAR TO EACH OTHER	SORBET
C	( X(K)/X(L) < 1.05 ), THEN AT FIRST ONE OF THESE POINTS WILL BE	SORBET
C	REPLACED. 93/07/09	SORBET
	IMPLICIT REAL*8(A-H,O-Z)	SORBET
	DIMENSION X(3),FX(3)	SORBET
C	-----	SORBET
	IF(KI .NE. 0) GO TO 10	SORBET
	KI=1	SORBET
	D3=ABS(FX(3)-FXS)	SORBET
	D1=ABS(FX(1)-FXS)	SORBET
	IF(D3 .GT. D1) KI=3	SORBET
C	SORTING	SORBET
C	FIRST POINT DISCARDED	SORBET
10	IF(KI .EQ. 3) GO TO 50	SORBET
	IF(XS .GT. X(2)) GO TO 20	SORBET
	X(KI) = XS	SORBET
	FX(KI) = FXS	SORBET
	GO TO 90	SORBET
20	X(1) = X(2)	SORBET
	FX(1) = FX(2)	SORBET
	IF(XS .GT. X(3)) GO TO 30	SORBET
	KI=2	SORBET
	X(KI) = XS	SORBET
	FX(KI) = FXS	SORBET
	GO TO 90	SORBET
30	X(2) = X(3)	SORBET
	FX(2) = FX(3)	SORBET
	KI=3	SORBET
	X(KI) = XS	SORBET
	FX(KI) = FXS	SORBET
	GO TO 90	SORBET
C	LAST POINT DISCARDED	SORBET
50	IF(XS .LT. X(2)) GO TO 60	SORBET
	X(KI) = XS	SORBET
	FX(KI) = FXS	SORBET
	GO TO 90	SORBET
60	X(3) = X(2)	SORBET
	FX(3) = FX(2)	SORBET
	IF(XS .LT. X(1)) GO TO 70	SORBET
	KI=2	SORBET
	X(KI) = XS	SORBET



	FX(KI) = FXS	SORBET
	GO TO 90	SORBET
70	X(2) = X(1)	SORBET
	FX(2) = FX(1)	SORBET
	KI=1	SORBET
	X(KI) = XS	SORBET
	FX(KI) = FXS	SORBET
90	RETURN	SORBET
	END	

### G.10 INR105.AQUA.FORT(TCORR)

	SUBROUTINE TCORR(TS,Q,DLK,DVK,IECCH)	TCORR
C	RETURNS FOR A GIVEN Q ( IN MPA ) THE CORRESPONDING SATURATION	TCORR
C	TEMPERATURE TS AND THE SATURATED DENSITIES	TCORR
C	DLK, DVK ( IN G/CM**3 ).	94/11/30 TCORR
	IMPLICIT REAL*8 (A-H,O-Z)	TCORR
	COMMON /SATIF / PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL	TCORR
	=,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV	TCORR
	COMMON /TCOFF / DQ,DTS,QC	TCORR
	PARAMETER(TCN=646.0D0,QCN=21.83815D0,FINIS=1.D-5)	TCORR
C	FIRST APPROXIMATE VALUES OF TS AND DTS/DQ	TCORR
	P=Q*1.D6	TCORR
	CALL TS1(P,TS,TS1P,IECCH)	TCORR
	TS1Q=TS1P*1.D6	TCORR
	IF(IECCH .NE. 0) RETURN	TCORR
C	ITERATING FOR CORRECTED TEMPERATURES	TCORR
	DO 22 I=1,5	TCORR
	CALL ONLYT(TS,IECCH)	TCORR
	IF(IECCH .NE. 0) RETURN	TCORR
	CALL SUBCRI(TS)	TCORR
	DLK=DL	TCORR
	DVK=DV	TCORR
	QC=Q	TCORR
	IF(Q .GT. QCN .OR. TS .GE. TCN) RETURN	TCORR
	CALL GPCORR(TS,QC,DLK,DVK,IECCH)	TCORR
	IF(IECCH .NE. 0) RETURN	TCORR
	DQ=Q-QC	TCORR
	DTS=DQ*TS1Q	TCORR
	IF(ABS(DTS) .LT. FINIS) RETURN	TCORR
22	TS = TS+DTS	TCORR
	RETURN	TCORR
	END	TCORR

### G.11 INR105.AQUA.FORT(GPCORR)

	SUBROUTINE GPCORR(T,Q,DLK,DVK,IECCH)	GPCORR
C	RETURNS FOR A GIVEN T THE SATURATED PRESSURE Q ( IN MPA )	GPCORR
C	AND THE SATURATED DENSITIES DLK, DVK ( IN G/CM**3 )	GPCORR
C	CORRECTED FOR G,L = G,V AND Q,L = Q,V	94/11/30 GPCORR
	IMPLICIT REAL*8(A-H,O-Z)	GPCORR
	COMMON /TEMPO/ RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI	GPCORR

```

COMMON /BASIF/  GB,SB,ZB,PB1T0,PB1R0,CVB          GPCORR
COMMON /RESEF/  GR,SR,ZR,PR1T0,PR1R0,CVR          GPCORR
COMMON /GPCOF/  ZL,ZV,GL,GV                       GPCORR
PARAMETER(TCN=646.0D0,FINIS=1.D-7)                GPCORR
DO 55 K=1,10                                       GPCORR
C SATURATED LIQUID STATE                           GPCORR
  CALL BASE(T,DLK,IECCH)                           GPCORR
  CALL RESID(T,DLK)                                 GPCORR
  ZL = ZB + ZR                                      GPCORR
  GL = GB + GR                                      GPCORR
  PR0L = PB1R0+PR1R0                               GPCORR
C SATURATED VAPOR STATE                           GPCORR
  CALL BASE(T,DVK,IECCH)                           GPCORR
  CALL RESID(T,DVK)                                 GPCORR
  ZV = ZB + ZR                                      GPCORR
  GV = GB + GR                                      GPCORR
  PR0V = PB1R0+PR1R0                               GPCORR
  Q=RT*(ZV*DVK+ZL*DLK)*5.D-1                       GPCORR
  IF(T .GE. TCN) RETURN                             GPCORR
C DEVIATIONS AND CORRECTIONS                       GPCORR
  DD=DLK-DVK                                        GPCORR
  DELG=GL-GV                                        GPCORR
  CL=DVK*(DELG+ZV)-DLK*ZL                          GPCORR
  CV=DLK*(DELG-ZL)+DVK*ZV                          GPCORR
  CXL=CL*DLK/(DD*PR0L)                             GPCORR
  CXV=CV*DVK/(DD*PR0V)                             GPCORR
  IF(ABS(CXL) .LT. FINIS) RETURN                   GPCORR
  DLK=DLK+CXL                                       GPCORR
55 DVK=DVK+CXV                                     GPCORR
  RETURN                                           GPCORR
END                                                 GPCORR

```

## G.12 INR105.AQUA.FORT(SUBCRI)

```

SUBROUTINE SUBCRI(T)                                SUBCRI
C SUBCRI CALCULATES DENSITIES, PRESSURES AND DERIVATIVES SUBCRI
C OF THE PRESSURES IN THE SATURATED STATES          SUBCRI
C   DL   DV                                         G/CM**3          SUBCRI
C   PS   TPS1T   TTPS2T                             PASCAL            SUBCRI
C AS WELL AS THE DENSITIES AND PRESSURES OF THE SPINODALS SUBCRI
C   RHSL RHSV                                       KG/M**3           SUBCRI
C   PSL  PSV                                         PASCAL            SUBCRI
C AS FUNCTIONS OF THE TEMPERATURE T.                94/12/01          SUBCRI
  IMPLICIT REAL*8(A-H,O-Z)                          SUBCRI
  COMMON /SATIF/  PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL SUBCRI
  =,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV SUBCRI
C-----APPROXIMATION TO THE VAPOR PRESSURE ----- SUBCRI
  CALL SSP(T)                                        SUBCRI
C-----APPROXIMATIONS TO THE SATURATED DENSITIES----- SUBCRI
  CALL SSRHO(T)                                     SUBCRI
C-----APPROXIMATIONS TO THE----- SUBCRI
C-----DENSITIES AND PRESSURES ON THE LIQUID AND VAPOR SPINODAL----- SUBCRI
  CALL SPIRHO(T)                                    SUBCRI

```

CALL SPIP(T)	SUBCRI
C-----	SUBCRI
RETURN	SUBCRI
END	SUBCRI

### G.13 INR105.AQUA.FORT(TWOPHA)

SUBROUTINE TWOPHA(T)	TWOPHA
C THIS SUBROUTINE CALCULATES DIFFERENT SATURATION PROPERTIES	TWOPHA
C IN S.I. UNITS	94/11/30 TWOPHA
IMPLICIT REAL*8(A-H,O-Z)	TWOPHA
COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL	TWOPHA
=,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV	TWOPHA
PARAMETER(TC=647.126D0)	TWOPHA
C-----	TWOPHA
IF(T .LE. TC) GO TO 10	TWOPHA
WRITE(6,1000) T	TWOPHA
T=-1.D0	TWOPHA
RETURN	TWOPHA
C-----HEAT CAPACITY OF THE SATURATED STATES CVL CVV -----	TWOPHA
10 CALL SSCEV(T,CVL,CVV)	TWOPHA
C-----DENSITY DERIVATIVE OF THE SATURATED STATES PLR10 PVR10 -----	TWOPHA
CALL SSPER(T)	TWOPHA
C-----TEMPERATURE DERIVATIVE OF THE SATURATED STATES PL1T0 PV1T0 -----	TWOPHA
CALL SSPET(T,PL1T0,PV1T0)	TWOPHA
C-----ENTROPY AND GIBBS FUNCION OF THE SATURATED LIQUID SL GL -----	TWOPHA
CALL SSGIS(T,SL,GL)	TWOPHA
C-----VISCOSITY OF THE SATURATED STATES ETL ETV -----	TWOPHA
CALL SSETA(T,ETL,ETV)	TWOPHA
C-----THERMAL CONDUCTIVITY OF THE SATURATED STATES XLL XLV -----	TWOPHA
CALL SSLAM(T,XLL,XLV)	TWOPHA
C-----	TWOPHA
1000 FORMAT(16H TEMPERATURE T =,F10.4,22H K IS SUPERCRITICAL ! /)	TWOPHA
RETURN	TWOPHA
END	TWOPHA

### G.14 INR105.AQUA.FORT(REDUK)

SUBROUTINE REDUK(T,D,IECCH)	REDUK
C CALCULATES THERMAL PROPERTIES OF THE WATER	REDUK
C IN REDUCED ( DIMENSIONLESS ) UNITS	REDUK
C IN THE LIQUID, VAPOR AND GAS STATES	REDUK
C D : DENSITY IN G/CM**3 , RH : IN KG/M**3	94/12/08 REDUK
C R1T0 : REDUCED DENSITY DERIVATIVE = (T/RH)*(DRH/DT)	REDUK
C P1T0, P1R0 : REDUCED DERIVATIVES OF THE PRESSURE	REDUK
C H0 , S0 : REDUCED ENTHALPY RESP. ENTROPY , CPC : CP/CV	REDUK
C ETA , XLMB : VISCOSITY RESP. THERMAL CONDUCTIVITY	REDUK
IMPLICIT REAL*8(A-H,O-Z)	REDUK
COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW	REDUK
COMMON /TEMPO/ RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,SI,CVI	REDUK
COMMON /BASIF/ GB,SB,ZB,PB1T0,PB1R0,CVB	REDUK
COMMON /RESEF/ GR,SR,ZR,PR1T0,PR1R0,CVR	REDUK

```

COMMON /REDUF/  G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLMB,R1T0,CPC,H0 REDUK
PARAMETER(RGAS=4.61522D+2) REDUK
DATA FNI/1.D+70/ REDUK
C----- REDUK
CALL BASE(T,D,IECCH) REDUK
IF(IECCH .NE. 0) RETURN REDUK
CALL RESID(T,D) REDUK
200 Z = ZB + ZR REDUK
P1T0 = PR1T0 + PB1T0 REDUK
P1R0 = PR1R0 + PB1R0 REDUK
S0 = SB + SR + SI - SREF REDUK
G0 = GB + GR + GI - 1.D0 - UREF/T + SREF REDUK
CV0 = CVB + CVR + CVI REDUK
IF(P1R0 .EQ. 0.D0) THEN REDUK
R1T0 = -FNI REDUK
CP0 = FNI REDUK
CPC = FNI REDUK
ELSE REDUK
R1T0 = -P1T0/P1R0 REDUK
CP0 = CV0-P1T0*R1T0 REDUK
CPC = CP0/CV0 REDUK
ENDIF REDUK
900 H0 = S0 + G0 REDUK
RH=D*1.03 REDUK
CALL TRANSP(T,RH,P1R0,P1T0,ETA,XLMB) REDUK
C----- REDUK
RETURN REDUK
END REDUK

```

### G.15 INR105.AQUA.FORT(REDUKE)

```

SUBROUTINE REDUKE(T,D,IECCH) REDUKE
C CALCULATES THERMAL PROPERTIES OF THE WATER REDUKE
C IN REDUCED ( DIMENSIONLESS ) UNITS REDUKE
C MIXED STATE INCLUDED REDUKE
C D : DENSITY IN G/CM**3 , RH : IN KG/M**3 94/12/08 REDUKE
C R1T0 : REDUCED DENSITY DERIVATIVE = (T/RHO)*(DRO/DT) REDUKE
C P1T0, P1R0 : REDUCED DERIVATIVES OF THE PRESSURE REDUKE
C H0 , S0 : REDUCED ENTHALPY RESP. ENTROPY, CPC : CP/CV REDUKE
C ETA , XLMB : VISCOSITY RESP. THERMAL CONDUCTIVITY REDUKE
C VQ : VAPOR QUALITY REDUKE
IMPLICIT REAL*8(A-H,O-Z) REDUKE
COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW REDUKE
COMMON /TEMPO/ RT,B1,B2,B1T,B2T,B1TT,B2TT,TV,GI,S1,CVI REDUKE
COMMON /BASIF/ GB,SB,ZB,PB1T0,PB1R0,CVB REDUKE
COMMON /RESEF/ GR,SR,ZR,PR1T0,PR1R0,CVR REDUKE
COMMON /REDUF/ G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLMB,R1T0,CPC,H0,VQ REDUKE
COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL REDUKE
=,RHSL,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV REDUKE
PARAMETER(RGAS=4.61522D+2) REDUKE
DATA ZERO/0.D0/,FNI/1.D+70/,UN/1.D0/ REDUKE
C----- REDUKE
CALL BASE(T,D,IECCH) REDUKE

```

IF(IECCH .NE. 0)	RETURN	REDUKE
CALL RESID(T,D)		REDUKE
RH=D*1.D3		REDUKE
IF(T .GE. TC)	GO TO 200	REDUKE
RDY = (D-DV)/(DL-DV)		REDUKE
IF(D .GT. DL)	THEN	REDUKE
VQ = ZERO		REDUKE
ELSEIF(D .LT. DV)	THEN	REDUKE
VQ = UN		REDUKE
ELSE		REDUKE
VQ = (UN-RDY)*DV/D		REDUKE
ENDIF		REDUKE
IF(RH .LT. RHSV .OR. RH .GT. RHSL)	GO TO 200	REDUKE
C-----TWO PHASED MIXTURE-----		REDUKE
100 CALL TWOPHA(T)		REDUKE
RGAT=RGAS*T		REDUKE
ROT=RGAT*D*1.D3		REDUKE
ROTL=RGAT*DL*1.D+3		REDUKE
DIFF= UN - D/DL		REDUKE
X = RDY		REDUKE
IF(X .GT. ZERO .AND. X .LT. UN)		REDUKE
= X = DEXP(DLOG(RDY)/3.D0)		REDUKE
P1R0 = ZERO		REDUKE
Z = PS/ROT		REDUKE
P1T0 = TPS1T/ROT		REDUKE
S0 = SL+DIFF*P1T0		REDUKE
G0 = GL		REDUKE
DCVS = RL1T0*RL1T0*PL1R0		REDUKE
AU = TTPS2T/ROT		REDUKE
CV0 = CVL + DCVS + DIFF*AU		REDUKE
R1T0 = -FNI		REDUKE
CP0 = FNI		REDUKE
CPC = FNI		REDUKE
ETAR=(UN-X)/ETV + X/(ETV + X*X*(ETL-ETV))		REDUKE
XLAR=(UN-X)/XLV + X/(XLV + X*X*(XLL-XLV))		REDUKE
ETA = UN/ETAR		REDUKE
XLMB = UN/XLAR		REDUKE
GO TO 900		REDUKE
C-----LIQID, VAPOR OR SUPERCRITICAL REGIONS-----		REDUKE
200 P1R0 = PR1R0 + PB1R0		REDUKE
Z = ZB + ZR		REDUKE
P1T0 = PR1T0 + PB1T0		REDUKE
S0 = SB + SR + SI - SREF		REDUKE
G0 = GB + GR + GI - UN - UREF/T + SREF		REDUKE
CV0 = CVB + CVR + CVI		REDUKE
IF(P1R0 .EQ. 0.D0)	THEN	REDUKE
R1T0 = -FNI		REDUKE
CP0 = FNI		REDUKE
CPC = FNI		REDUKE
ELSE		REDUKE
R1T0 = -P1T0/P1R0		REDUKE
CP0 = CV0-P1T0*R1T0		REDUKE
CPC = CP0/CV0		REDUKE
ENDIF		REDUKE

	CALL TRANSP(T,RH,P1R0,P1T0,ETA,XLMB)	REDUKE
C		REDUKE
	900 H0 = S0 + G0	REDUKE
C	-----	REDUKE
	RETURN	REDUKE
	END	REDUKE

### G.16 INR105.AQUA.FORT(HYDOR)

	SUBROUTINE HYDOR(T,RH,IECCH)	HYDOR
C	CALCULATES THERMAL PROPERTIES OF WASSER	HYDOR
C	S G H H1P U1R CV CP P P1T P1R R1T R1P	HYDOR
C	ETA XLMB SRFTS P1RS VS PRN VQ	HYDOR
C	AS FUNCTIONS OF THE TEMPERATURE, T AND THE DENSITY, RH	HYDOR
C	ALL PROPERTIES ARE IN S.I. UNITS	94/12/08 HYDOR
	IMPLICIT REAL*8(A-H,O-Z)	HYDOR
	COMMON /REDUF/ G0,S0,Z,P1T0,P1R0,CV0,CP0,ET,XLB,R1T0,CPC,H0,VQ1	HYDOR
	COMMON /HYDUF/ S,G,H,H1P,U1R,CV,CP,P,P1T,P1R,R1T,R1P	HYDOR
	=,ETA,XLMB,SRFTS,P1RS,VS,PRN,V,VQ	HYDOR
	PARAMETER(TC=647.126D0,RGAS=4.61522D+2)	HYDOR
	DATA TA/0.D0/,ZERO/0.D0/,UN/1.D0/	HYDOR
	IF(RH .LE. ZERO) THEN	HYDOR
	WRITE(6,9100) RH	HYDOR
	IECCH = 5	HYDOR
	RETURN	HYDOR
	ENDIF	HYDOR
100	D=RH*1D-3	HYDOR
	V=UN/RH	HYDOR
C	VAPOR, LIQUID OR GAS STATE	HYDOR
200	IF(T .EQ. TA) GO TO 210	HYDOR
	ROT=T*RGAS	HYDOR
	CALL ONLYT(T,IECCH)	HYDOR
	IF(IECCH .NE. 0) RETURN	HYDOR
	SRFTS=ZERO	HYDOR
	IF(T .GE. TC) GO TO 210	HYDOR
	SRFTS=SIGMA(T)	HYDOR
	CALL SUBCRI(T)	HYDOR
210	CALL REDUKE(T,D,IECCH)	HYDOR
	IF(IECCH .NE. 0) RETURN	HYDOR
	VQ=VQ1	HYDOR
	G=G0*ROT	HYDOR
	H=H0*ROT	HYDOR
	S=S0*RGAS	HYDOR
	CP=CP0*RGAS	HYDOR
	CV=CV0*RGAS	HYDOR
	P1R=P1R0*ROT	HYDOR
	P1T=P1T0*RGAS*RH	HYDOR
	IF(P1R0 .GT. ZERO) THEN	HYDOR
	P1RS=CPC*P1R	HYDOR
	R1P=UN/P1R	HYDOR
	ELSE	HYDOR
	P1RS=(P1T0*P1T0/CV0)*ROT	HYDOR
	R1P=-R1T0/ROT	HYDOR

ENDIF	HYDOR
U1R=(Z-P1T0)*V*ROT	HYDOR
H1P=(UN+R1T0)*V	HYDOR
R1T=(RH/T)*R1T0	HYDOR
VS=SQRT(P1RS)	HYDOR
P=Z*ROT*RH	HYDOR
ETA=ET	HYDOR
XLMB=XLB	HYDOR
PRN=CP*ETA/XLMB	HYDOR
999 TA=T	HYDOR
RETURN	HYDOR
9100 FORMAT(13H DENSITY RH =,G15.6,20H KG/M**3 TOO LOW ! /)	HYDOR
END	HYDOR

## Appendix H. User-Interfaces

The following pages list the programs, which are needed directly by the users to calculate the water properties.

### H.1 INR105.AQUA.FORT(WATER)

```

PROGRAM WATER
C CALCULATES THERMAL PROPERTIES OF WATER AS FUNCTIONS
C OF TWO VARIABLES : (T,RHO) OR (T,P) 94/12/09
IMPLICIT REAL*8(A-H,O-Z)
COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL
=,RHVS,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV
COMMON /HYDUF/ S,G,H,H1P,U1R,CV,CP,P,P1T,P1R,R1T,R1P
=,ETA,XLMB,SRFTS,P1RS,VS,PRN,V,XI
PARAMETER(TC=647.126D0,FR=1.D+3,FK=1.D-3,FM=1.D-6)
DATA ZERO/0.D0/,IMO/0/,QIS,T,RHI/0.101325D0,373.15D0,958.4467D0/
WRITE(6,1001)
WRITE(6,1002)
WRITE(6,1003)
100 WRITE(6,1050) T
READ(5,*) T
IF(T .LE. ZERO) GO TO 999
200 WRITE(6,1100)
READ(5,*) IMO
IF(IMO .LE. 0) GO TO 100
400 CALL PG3277
IF(IMO .EQ. 1) GO TO 600
500 WRITE(6,1200) QIS
READ(5,*) QIS
IF(QIS .EQ. ZERO) GO TO 200
DX=ZERO
IECCH=0
CALL ONLYT(T,IECCH)
IF(IECCH .NE. 0) GO TO 500
IF(T .LT. TC) THEN
CALL SUBCRI(T)
WRITE(6,1300)
READ(5,*) ISTA
IF(ISTA .EQ. 1) THEN
CALL DLFIND(T,QIS,DX,IECCH)
IF(IECCH .EQ. 0) GO TO 700
WRITE(6,9300)
ELSEIF(ISTA .EQ. 2) THEN
CALL DVFIND(T,QIS,DX,IECCH)
IF(IECCH .EQ. 0 .AND. DX .GT. ZERO) GO TO 700
WRITE(6,9400)
ENDIF
ELSE
CALL DGFIND(T,QIS,DX,IECCH)
IF(IECCH .EQ. 0 .AND. DX .GT. ZERO) GO TO 700
WRITE(6,9500)

```



```

ENDIF
GO TO 500
600 WRITE(6,1400) RHI
READ(5,*) RHI
IF(RHI .LE. ZERO) GO TO 200
IF(RHI .GT. ZERO) GO TO 800
700 RHI=DX*FR
800 IECCH=0
CALL HYDOR(T,RHI,IECCH)
IF(IECCH .NE. 0) GO TO 300
A=G-P/RHI
U=H-P/RHI
WRITE(6,1610) T,RHI,P*FM
IF(P1R .NE. ZERO) THEN
WRITE(6,1620) P1R*FK,P1T*FM,R1T
WRITE(6,1630) CV*FK,CP*FK,S*FK
WRITE(6,1640) U*FK,H*FK,G*FK
WRITE(6,1650) U1R*FK,H1P,VS
ELSE
WRITE(6,1622) P1R*FK,P1T*FM
WRITE(6,1632) CV*FK,S*FK
WRITE(6,1640) U*FK,H*FK,G*FK
WRITE(6,1652) U1R*FK,VS
ENDIF
IF(T .LE. TC) THEN
IF(XI .LT. 1.D0) WRITE(6,1660) ETA,XLMB,SRFTS
IF(XI .GE. 1.D0) WRITE(6,1662) ETA,XLMB
IF(P1R .NE. ZERO) WRITE(6,1670) PRN,XI,PS*FM
IF(P1R .EQ. ZERO) WRITE(6,1674) XI,PS*FM
ELSE
WRITE(6,1662) ETA,XLMB
WRITE(6,1672) PRN
ENDIF
IF(T .LE. TC) THEN
WRITE(6,1680) DV*FR,RHSV,PSV*FM
WRITE(6,1682) DL*FR,RHSL,PSL*FM
ENDIF
300 WRITE(6,2000)
GO TO 400
1001 FORMAT(/2H +,39(2H+)/2H +,77X,1H+/2H +,5X,1HW,5X,1HW,5X,1HA,5X
=,7(1HT),2X,7(1HE),2X,6(1HR),9X,17HSTATUS : 94/12/09,4X,1H+/2H +
=,5X,1HW,5X,1HW,4X,3(1HA),4X,7(1HT),2X,7(1HE),2X,7(1HR),29X,1H+
=/2H +,5X,1HW,5X,1HW,3X,5HAA AA,6X,1HT,5X,1HE,6X,2X,1HR,5X,1HR,29X
=,1H+/2H +,5X,7HW W W,2X,7HAA AA,5X,1HT,5X,5HE EEE,4X,1HR,1X
=,5(1HR),29X,1H+/2H +,5X,7HW WWW W,2X,1HA,5X,1HA,5X,1HT,5X,5HE EEE
=,4X,1HR,1X,4(1HR),30X,1H+/2H +,5X,7HWWW WWW,2X,7HA AAA A,5X,1HT
=,5X,1HE,8X,5HR RR,31X,1H+/2H +,5X,7HWW WW,2X,7HA AAA A,5X,1HT
=,5X,7(1HE),2X,6HR RR,30X,1H+/2H +,5X,1HW,5X,1HW,2X,1HA,5X
=,1HA,5X,1HT,5X,7(1HE),2X,1HR,4X,2HRR,29X,1H+ )
1002 FORMAT(2H +,77X,1H+/2H +,14X
=,48HWATER CALCULATES THERMAL PROPERTIES OF THE WATER,15X,1H+
=/2H +,19X,39HIN THE LIQUID, VAPOR AND GASEOUS STATES,19X,1H+
=/2H +,15X,46HEITHER AS FUNCTIONS OF TEMPERATURE AND DENSITY,16X
=,1H+/2H +,16X,44HOR AS FUNCTIONS OF TEMPERATURE AND PRESSURE.
=,17X,1H+)

```

```

1003 FORMAT(2H +,77X,1H+/2H +,14X WATER
      =,49HTHE PROPERTIES CALCULATED ARE THE FOLLOWING ONES: WATER
      =,14X,1H+/2H +,77X,1H+/2H +,5X,3HT =,12H TEMPERATURE,11X,3HP = WATER
      =,9H PRESSURE,15X,5HRHO =,8H DENSITY,6X,1H+/2H +,5X WATER
      =,46HCV / CP = ISOBARIC / ISOCHORIC HEAT CAPACITIES,7X WATER
      =,11HS = ENTROPY,8X,1H+/2H +,5X,18HG = GIBBS FUNCTION,8X WATER
      =,10HU = ENERGY,17X,12HH = ENTHALPY,7X,1H+/2H +,5X WATER
      =,22HVEL.S = SONIC VELOCITY,4X,23HSIGMA = SURFACE TENSION,4X WATER
      =,15HETA = VISCOSITY,4X,1H+/2H +,5X WATER
      =,29HLAMBDA = THERMAL CONDUCTIVITY,12X,22HPR.NR = PRANDTL NUMBER WATER
      =,9X,1H+/2H +,5X,20HVAP.QU = M.VAP/M.TOT,52X,1H+ WATER
      =/2H +,77X,1H+/2H +,28X,21HFOR MORE INFORMATION ,5H => WATER
      =,19HK. THURNAY, T: 2477,4X,1H+/2H +,39(2H++) ///) WATER
1050 FORMAT(/29H ENTER TEMPERATURE ( IN K ) /,F10.3 WATER
      =,31H / =: ( STOP WTIH T = 0 ) ) WATER
1100 FORMAT(/30H DO YOU WANT THE PROPERTIES AS WATER
      =/ 5X,35H 1 = TEMPERATURE-DENSITY FUNCTIONS WATER
      =/ 5X,35H 2 = TEMPERATURE-PRESSURE FUNCTIONS WATER
      =,15X,15H( STOP WITH 0 ) ) WATER
1200 FORMAT(/33H ENTER PRESSURE ( IN MPASCAL ) /,G15.7 WATER
      =,28H / =: ( STOP WTIH P = 0 ) ) WATER
1300 FORMAT(/36H LIQUID ( 1 ) OR VAPOR ( 2 ) STATE ? WATER
      =,17H ( 0 = STOP ) =: ) WATER
1400 FORMAT(/31H ENTER DENSITY ( IN KG/M**3 ) /,G15.7 WATER
      =,28H / =: ( STOP WTIH RHO = 0 ) ) WATER
1610 FORMAT(/4H T =,F10.3,13X,5HRHO =,G14.7,8X,3HP =,G15.7 WATER
      =/2X,2H K,25X,7HKG/M**3,20X,7HMPASCAL) WATER
1620 FORMAT(10H DP/DRHO =,G13.6,4X,7HDP/DT =,G13.6,7X,9HDRHO/DT = WATER
      =,G14.6/2X,6H KJ/KG,21X,9HMPASCAL/K,18X,10HKG/(K*M**3 ) WATER
1622 FORMAT(10H DP/DRHO =,G13.6,4X,7HDP/DT =,G13.6,7X,9HDRHO/DT = WATER
      =,6H -INF./2X,6H KJ/KG,21X,9HMPASCAL/K,17X,9HKG/M**3/K ) WATER
1630 FORMAT(5H CV =,G13.6,9X,4HCP =,G13.6,10X,3HS =,G14.6 WATER
      =/3X,9HKJ/(KG*K),17X,9HKJ/(KG*K),18X,9HKJ/(KG*K) ) WATER
1632 FORMAT(5H CV =,G13.6,9X,4HCP =,5H INF.,18X,3HS =,G14.6 WATER
      =/3X,9HKJ/(KG*K),17X,9HKJ/(KG*K),18X,9HKJ/(KG*K) ) WATER
1640 FORMAT(4H U =,G14.6,9X,3HH =,G14.6,10X,3HG =,G14.6 WATER
      =/2X,6H KJ/KG,21X,5HKJ/KG,22X,5HKJ/KG ) WATER
1650 FORMAT(10H DU/DRHO =,G14.6,3X,7HDH/DP =,G14.6,6X WATER
      =,7HVEL.S =,G13.6/2X,14H KJ*M**3/KG**2,13X,7HM**3/KG,20X,5HM/SEC) WATER
1652 FORMAT(10H DU/DRHO =,G14.6,3X,7HDH/DP =,6H -INF.,14X WATER
      =,7HVEL.S =,G13.6/2X,14H KJ*M**3/KG**2,13X,7HM**3/KG,20X,5HM/SEC) WATER
1660 FORMAT(6H ETA =,G13.6,8X,8HLAMBDA =,G13.6,6X WATER
      =,7HSIGMA =,G13.6/2X,9H KG/M/SEC,18X,5HW/M/K,22X,9HKG/SEC**2) WATER
1662 FORMAT(6H ETA =,G13.6,8X,8HLAMBDA =,G13.6 WATER
      =/2X,9H KG/M/SEC,18X,5HW/M/K) WATER
1670 FORMAT(8H PR.NR =,G13.6,6X,8HVAP.QU =,G13.6,6X,4HPS =,G14.7 WATER
      =/56X,7HMPASCAL ) WATER
1672 FORMAT(8H PR.NR =,G13.6/) WATER
1674 FORMAT(8H PR.NR =,6H -INF.,13X,8HVAP.QU =,G13.6,6X,4HPS =,G14.7 WATER
      =/56X,7HMPASCAL ) WATER
1680 FORMAT(7H RH.V =,G14.7,6X,7HRH.SV =,G14.7,6X,6HP.SV = WATER
      =,G13.6/2X,8H KG/M**3,19X,7HKG/M**3,20X,7HMPASCAL ) WATER
1682 FORMAT(7H RH.L =,G14.7,6X,7HRH.SL =,G14.7,6X,6HP.SL = WATER
      =,G13.6/2X,8H KG/M**3,19X,7HKG/M**3,20X,7HMPASCAL ) WATER
2000 FORMAT(//////////) WATER

```

9300	FORMAT(/35H NO LIQUID STATE AT THIS PRESSURE ! )	WATER
9400	FORMAT(/35H NO VAPOR STATE AT THIS PRESSURE ! )	WATER
9500	FORMAT(/42H NO SUPERCRITICAL STATE AT THIS PRESSURE ! )	WATER
999	STOP	WATER
	END	WATER

## H.2 INR105.AQUA.FORT(KAISOT)

	PROGRAM KAISOT		KAISOT
C	CALCULATES THERMAL PROPERTIES OF WATER		KAISOT
C	ALONG DIFFERENT ISOTHERMS	94/11/30	KAISOT
	IMPLICIT REAL*8(A-H,O-Z)		KAISOT
	REAL*8 FRH(1000),RH(1000),F(1000),XF(1000)		KAISOT
	INTEGER NEIG(50)		KAISOT
	CHARACTER*8 NAMR,MASR,NAMF,MASF,NORMR,NORMW,MODUS		KAISOT
	=,NAMG,MASG		KAISOT
	COMMON /MUFUN/ RH,RMI,RMA,RED,FRH,FMI,FMA,FED		KAISOT
	COMMON /HYDUF/ S,G,H,H1P,U1R,CV,CP,P,P1T,P1R,R1T,R1P		KAISOT
	=,ETA,XLMB,SRFTS,PIRS,VS,PRN,V,XI		KAISOT
	DATA IRE/15/,IWR/16/,MODUS/'**OLD**'/,KPRI/2/		KAISOT
	DATA IA,IE,II/1,1000,1/		KAISOT
	DATA NEIG/11,15,17,21,22,23,24,26,27,31,32,34,35,36,41,42,43		KAISOT
	=,44,45,46,47,29*0/		KAISOT
	WRITE(6,1000)		KAISOT
	CALL SERDIO(IRE,NORMR,IWR,NORMW,MODUS,KPRI)		KAISOT
	IF(NORMR .NE. 'GRAPHIC8') GO TO 999		KAISOT
	IF(NORMW .NE. 'GRAPHIC8') GO TO 999		KAISOT
120	WRITE(6,1100)		KAISOT
	READ(5,*) T		KAISOT
	IF(T .LE. 0) GO TO 999		KAISOT
200	CALL SIGNAT(KLASSE,NUMMER,NAMF,MASF,NEIG)		KAISOT
	IF(KLASSE .LE. 0) GO TO 120		KAISOT
	CALL MUSTER(IA,IE,II,NAMR,MASR,NAMG,MASG,*200)		KAISOT
	CALL PG3277		KAISOT
	NAMF(5:8)='.ITH'		KAISOT
	WRITE(6,1200)		KAISOT
	READ(5,*) ISCRI		KAISOT
	MP=0		KAISOT
	DO 111 I=IA,IE,II		KAISOT
	RHI=RH(I)		KAISOT
	CALL HYDOR(T,RHI,IECCH)		KAISOT
	IF(IECCH .NE. 0) GO TO 111		KAISOT
	MP=MP+1		KAISOT
	IF(ISCRI .EQ. 0 .AND. MOD(MP,25) .EQ. 1)		KAISOT
	= WRITE(6,1500) NAMF,NAMG		KAISOT
	IF(KLASSE .EQ. 11) F(I)=RH(I)		KAISOT
	IF(KLASSE .EQ. 15) F(I)=P		KAISOT
	IF(KLASSE .EQ. 17) F(I)=V		KAISOT
	IF(KLASSE .EQ. 21) F(I)=P1R		KAISOT
	IF(KLASSE .EQ. 22) F(I)=P1T		KAISOT
	IF(KLASSE .EQ. 23) F(I)=R1P		KAISOT
	IF(KLASSE .EQ. 24) F(I)=R1T		KAISOT
	IF(KLASSE .EQ. 26) F(I)=CV		KAISOT

```

IF(KLASSE .EQ. 27)      F(I)=CP          KAISOT
IF(KLASSE .EQ. 31)      F(I)=VS          KAISOT
IF(KLASSE .EQ. 32)      F(I)=SRFTS       KAISOT
IF(KLASSE .EQ. 34)      F(I)=ETA         KAISOT
IF(KLASSE .EQ. 35)      F(I)=XLMB       KAISOT
IF(KLASSE .EQ. 36)      F(I)=PRN         KAISOT
IF(KLASSE .EQ. 41)      F(I)=G-P*V      KAISOT
IF(KLASSE .EQ. 42)      F(I)=G          KAISOT
IF(KLASSE .EQ. 43)      F(I)=H-P*V      KAISOT
IF(KLASSE .EQ. 44)      F(I)=H          KAISOT
IF(KLASSE .EQ. 45)      F(I)=S          KAISOT
IF(KLASSE .EQ. 46)      F(I)=U1R        KAISOT
IF(KLASSE .EQ. 47)      F(I)=H1P        KAISOT
IF(ISCRI .EQ. 0) WRITE(6,1600) I,RH(I),F(I),FRH(I) KAISOT
IF(ISCRI .EQ. 0 .AND. MOD(MP,25) .EQ. 0) WRITE(6,1400) KAISOT
111 CONTINUE          KAISOT
IF(ISCRI .EQ. 0)      GO TO 199          KAISOT
LAX=-1               KAISOT
RMI=RMA              KAISOT
CALL FENST8(MP,RH,XF,RMI,RMA,RED,LAX,KAXE,MASR) KAISOT
FMI=FMA              KAISOT
CALL FENST8(MP,F,XF,FMI,FMA,FED,LAX,KAXE,MASF) KAISOT
NUMMER=T*10.D0       KAISOT
CALL SDING8(KLASSE,NUMMER,MP          KAISOT
=,RH,RMI,RMA,RED,NAMR,MASR,F,FMI,FMA,FED,NAMF,MASF,MODUS) KAISOT
199 WRITE(6,2000)    KAISOT
GO TO 200            KAISOT
1000 FORMAT(2H +,39(2H+)/2H +,77X,1H+ KAISOT
= /2H +,6X,14HPROGRAM KAISOT,34X,17HSTATUS : 94/11/30,6X,1H+ KAISOT
= /2H +,77X,1H+/2H +,6X,30HCALCULATES THERMAL PROPERTIES KAISOT
=,30HOF THE WATER ALONG AN ISOTHERM,11X,1H+ KAISOT
= /2H +,24X,27HAS FUNCTIONS OF THE DENSITY,26X,1H+ KAISOT
= /2H +,12X,35HUSING A SAMPLE (RH,FRH)-FUNCTION AS KAISOT
=,18H A DENSITY-PATTERN,12X,1H+/2H +,77X,1H+/2H +,39(2H++) ) KAISOT
1100 FORMAT(49H ENTER TEMPERATURE OF THE ISOTHERM =: ( 0 = END ) ) KAISOT
1200 FORMAT(37H FUNCTION TO BE SAVED ? ( 0 = NO ) =: ) KAISOT
1400 FORMAT(///)     KAISOT
1500 FORMAT(/1H0,3X,1HI,9X,2HRH,8X,5HCALC.,A8,6X,5HMAST.,A8) KAISOT
1600 FORMAT(I5,2G16.6,G19.6) KAISOT
2000 FORMAT(////////////////////////////////////) KAISOT
999 STOP            KAISOT
END                 KAISOT

```

### H.3 INR105.AQUA.FORT(KAISOK)

```

PROGRAM KAISOK          KAISOK
C  CALCULATES THERMAL PROPERTIES OF WATER KAISOK
C  ALONG DIFFERENT ISOCHORES          94/11/30 KAISOK
IMPLICIT REAL*8(A-H,O-Z) KAISOK
REAL*8 FX(1000),X(1000),F(1000),XF(1000) KAISOK
INTEGER NEIG(50)       KAISOK
CHARACTER*8 NAMX,MASX,NAMF,MASF,NORMR,NORMW,MODUS KAISOK
=,NAMG                 KAISOK

```

COMMON /MUFUN/	X,XMI,XMA,XED,FX,FMI,FMA,FED	KAISOK
COMMON /HYDUF/	S,G,H,H1P,U1R,CV,CP,P,P1T,P1R,R1T,R1P	KAISOT
=,ETA,XLMB,SRFTS,P1RS,VS,PRN,V,XI		KAISOT
DATA IRE/15/,IWR/16/,MODUS/'**OLD**'/,KPRI/2/		KAISOK
DATA IA,IE,II/1,1000,1/		KAISOK
DATA NEIG/11,15,17,21,22,23,24,26,27,31,32,34,35,36,41,42,43		KAISOT
=,44,45,46,47,29*0/		KAISOT
WRITE(6,1000)		KAISOK
CALL SERDIO(IRE,NORMR,IWR,NORMW,MODUS,KPRI)		KAISOK
IF(NORMR .NE. 'GRAPHIC8')	GO TO 999	KAISOK
IF(NORMW .NE. 'GRAPHIC8')	GO TO 999	KAISOK
120 WRITE(6,1100)		KAISOK
READ(5,*) RHI		KAISOK
IF(RHI .LE. 0)	GO TO 999	KAISOK
200 CALL SIGNAT(KLASSE,NUMMY,NAMF,MASF,NEIG)		KAISOK
IF(KLASSE .LE. 0)	GO TO 120	KAISOK
CALL MUSTER(IA,IE,II,NAMX,MASX,NAMG,MASF,*200)		KAISOK
CALL PG3277		KAISOK
NAMF(5:8)=' .ICH'		KAISOK
WRITE(6,1200)		KAISOK
READ(5,*) ISCRI		KAISOK
MP=0		KAISOK
DO 111 I=IA,IE,II		KAISOK
T=X(I)		KAISOK
CALL HYDOR(T,RHI,IECCH)		KAISOK
IF(IECCH .NE. 0)	GO TO 111	KAISOK
MP=MP+1		KAISOK
IF(ISCRI .EQ. 0 .AND. MOD(MP,25) .EQ. 1)		KAISOK
= WRITE(6,1500) NAMF,NAMG		KAISOK
IF(KLASSE .EQ. 15)	F(I)=P	KAISOK
IF(KLASSE .EQ. 17)	F(I)=V	KAISOK
IF(KLASSE .EQ. 21)	F(I)=P1R	KAISOK
IF(KLASSE .EQ. 22)	F(I)=P1T	KAISOK
IF(KLASSE .EQ. 23)	F(I)=R1P	KAISOK
IF(KLASSE .EQ. 24)	F(I)=R1T	KAISOK
IF(KLASSE .EQ. 26)	F(I)=CV	KAISOK
IF(KLASSE .EQ. 27)	F(I)=CP	KAISOK
IF(KLASSE .EQ. 31)	F(I)=VS	KAISOK
IF(KLASSE .EQ. 32)	F(I)=SRFTS	KAISOK
IF(KLASSE .EQ. 34)	F(I)=ETA	KAISOK
IF(KLASSE .EQ. 35)	F(I)=XLMB	KAISOK
IF(KLASSE .EQ. 36)	F(I)=PRN	KAISOK
IF(KLASSE .EQ. 41)	F(I)=G-P*V	KAISOT
IF(KLASSE .EQ. 42)	F(I)=G	KAISOK
IF(KLASSE .EQ. 43)	F(I)=H-P*V	KAISOT
IF(KLASSE .EQ. 44)	F(I)=H	KAISOK
IF(KLASSE .EQ. 45)	F(I)=S	KAISOK
IF(KLASSE .EQ. 46)	F(I)=U1R	KAISOT
IF(KLASSE .EQ. 47)	F(I)=H1P	KAISOK
IF(ISCRI .EQ. 0) WRITE(6,1600) I,T,F(I),FX(I)		KAISOK
IF(ISCRI .EQ. 0 .AND. MOD(MP,25) .EQ. 0) WRITE(6,1400)		KAISOK
111 CONTINUE		KAISOK
IF(ISCRI .EQ. 0)	GO TO 199	KAISOK
LAX=-1		KAISOK
XMI=XMA		KAISOK

CALL FENST8(MP,X,XF,XMI,XMA,XED,LAX,KAXE,MASR)	KAISOK
FMI=FMA	KAISOK
CALL FENST8(MP,F,XF,FMI,FMA,FED,LAX,KAXE,MASF)	KAISOK
NUMMER=RHI	KAISOK
CALL SDING8(KLASSE,NUMMER,MP	KAISOK
=,X,XMI,XMA,XED,NAMX,MASX,F,FMI,FMA,FED,NAMF,MASF,MODUS)	KAISOK
199 WRITE(6,2000)	KAISOK
GO TO 200	KAISOK
1000 FORMAT(2H +,39(2H++)/2H +,77X,1H+	KAISOK
=/2H +,6X,14HPROGRAM KAISOK,34X,17HSTATUS : 94/11/30,6X,1H+	KAISOK
=/2H +,77X,1H+/2H +,6X,30HCALCULATES THERMAL PROPERTIES	KAISOK
=,30HOF THE WATER ALONG AN ISOCHORE,11X,1H+	KAISOK
=/2H +,23X,31HAS FUNCTIONS OF THE TEMPERATURE,23X,1H+	KAISOK
=/2H +,12X,33HUSING A SAMPLE (T,FT)-FUNCTION AS	KAISOK
=,22H A TEMPERATURE-PATTERN,10X,1H+/2H +,77X,1H+/2H +,39(2H++ )	KAISOK
1100 FORMAT(1H1/45H ENTER DENSITY OF THE ISOCHORE =: ( 0 = END ) )	KAISOK
1200 FORMAT(37H FUNCTION TO BE SAVED ? ( 0 = NO ) =: )	KAISOK
1400 FORMAT(///)	KAISOK
1500 FORMAT(/1H0,3X,1HI,5X,1HT,8X,5HCALC.,A8,6X,5HMAST.,A8)	KAISOK
1600 FORMAT(I5,F10.3,G16.6,G19.6)	KAISOK
2000 FORMAT(////////////////////)	KAISOK
999 STOP	KAISOK
END	KAISOK

**Note:** The routine PG3277 in the above programs serves only to clear the screen for a new batch of output and can be omitted, or replaced with an alternative routine.



## Appendix I. H.G.K.-users

The following programs serve to calculate different water properties using the H.G.K.-routines and the IAPS-82 equations ( 3.0, "Transport properties of the water." on page 9 ).

### I.1 INR105.AQUA.FORT(HGKSPN)

```

PROGRAM HGKSPN
C   CALCULATES THE SPINODALS OF THE WATER
C   USING THE HGK-CODE
C                                     94/12/13
IMPLICIT REAL*8(A-H,O-Z)
REAL*8 P(1000),F(1000),R(1000),W(1000)
REAL*8 Y(3),FY(3)
CHARACTER*8 NORMR,NORMW,MODUS
=,NAMW,MASW,NAMR,MASR,NAMP,MASP
COMMON /MUFUN/ W,WMI,WMA,WED ,F,FMI,FMA,FED
COMMON /REDUF/  G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLMB,R1T0,CPC,H0
COMMON /SATIF/  PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL
=,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,EV,XLV
PARAMETER(TC=647.126D0,RHC=0.322D3,RGAS=4.61522D+2,TCN=646.3D0)
DATA IRE/15/,IWR/16/,MODUS/'**OLD**'/,KPRI/2/
DATA IA,IE,II/1,1000,1/,IP/0/
DATA ITE/10/,ZERO/0.D0/,PRM0/1.D-10/,RHMAX/1.2D3/
DATA NAMW/'  T  '/
DATA MASW,MASR,MASP/'  K  ',' KG/M**3',' PASCAL'/
C   -----
WRITE(6,1000)
CALL SERDIO(IRE,NORMR,IWR,NORMW,MODUS,KPRI)
IF(NORMR.NE.'GRAPHIC8'.OR.NORMW.NE.'GRAPHIC8') GO TO 999
100 WRITE(6,1100) IP
READ(5,*) IP
IPP=MIN(IPP,2)
IF(IP.LE.0) GO TO 999
CALL MUSTER(IA,IE,II,NAMT,MAST,NAMP,MASP,*100)
CALL PG3277
WRITE(6,1200)
READ(5,*) ISCRI
C   DENSITY ITERATION FOR THE "DP/DRH = 0" POINT
MP=0
DO 211 J=IA,IE,II
T=W(J)
C   IF(T.GT.TCN) GO TO 490
CALL ONLYT(T,IECCH)
CALL SUBCRI(T)
RED=2.D-2*RHC
IF(IP.EQ.1) THEN
NAMR='RHSL.HGK'
NAMP='PSPL.HGK'
RHL=DL*1.D3
RHI=MIN(1.25D0*RHL,RHMAX)
RED=ABS(RED)

```



ELSEIF(IP .EQ. 2)	THEN	HGKSPN
NAMR=' RHSV.HGK'		HGKSPN
NAMP=' PSPV.HGK'		HGKSPN
RHV=DV*1.D3		HGKSPN
RHI=7.5D-1*RHV		HGKSPN
RED=-ABS(RED)		HGKSPN
ENDIF		HGKSPN
MP=MP+1		HGKSPN
IF(ISCRI .EQ. 0 .AND. MOD(MP,25) .EQ. 1)	WRITE(6,1600)	HGKSPN
RH=RHI		HGKSPN
IT=0		HGKSPN
DO 311 L=1,1000		HGKSPN
ROF=RH-RHC		HGKSPN
IF(IP .EQ. 1 .AND. ROF .LT. ZERO)	THEN	HGKSPN
WRITE(6,9000) T		HGKSPN
GO TO 490		HGKSPN
ELSEIF(IP .EQ. 2 .AND. ROF .GT. ZERO)	THEN	HGKSPN
WRITE(6,9001) T		HGKSPN
GO TO 490		HGKSPN
ENDIF		HGKSPN
DX=RH*1.D-3		HGKSPN
CALL REDUK(T,DX,IECCH)		HGKSPN
IF(P1R0 .EQ. ZERO)	THEN	HGKSPN
R(J)=RH		HGKSPN
GO TO 450		HGKSPN
ENDIF		HGKSPN
IF(L .LE. 2)	GO TO 320	HGKSPN
IF(IP .EQ. 1)	THEN	HGKSPN
Y(3)=Y(2)		HGKSPN
FY(3)=FY(2)		HGKSPN
ELSE		HGKSPN
Y(1)=Y(2)		HGKSPN
FY(1)=FY(2)		HGKSPN
ENDIF		HGKSPN
320 IF(L .LE. 1)	GO TO 330	HGKSPN
IF(IP .EQ. 1)	THEN	HGKSPN
Y(2)=Y(1)		HGKSPN
FY(2)=FY(1)		HGKSPN
ELSE		HGKSPN
Y(2)=Y(3)		HGKSPN
FY(2)=FY(3)		HGKSPN
ENDIF		HGKSPN
330 IF(IP .EQ. 1)	THEN	HGKSPN
Y(1)=DX		HGKSPN
FY(1)=P1R0		HGKSPN
ELSE		HGKSPN
Y(3)=DX		HGKSPN
FY(3)=P1R0		HGKSPN
ENDIF		HGKSPN
IF(P1R0 .GT. ZERO)	GO TO 311	HGKSPN
IF(P1R0 .EQ. ZERO)	THEN	HGKSPN
R(J)=RH		HGKSPN
GO TO 450		HGKSPN
ENDIF		HGKSPN
IF(L .GT. 2)	GO TO 410	HGKSPN

```

311 RH=RH-RED
410 IT=IT+1
420 CALL QUADO(FY,Y,ZERO,DX,ID)
      IF(ID .NE. 0) THEN
          WRITE(6,1500) ID,(Y(K),K=1,3),(FY(K),K=1,3)
      ENDIF
      KI=0
      CALL SORBET(KI,FY,Y,ZERO,DX)
      CALL REDUK(T,DX,IECCH)
      R(J)=DX*1.D3
      IF(ABS(P1R0) .LT. PRM0) GO TO 450
      FY(KI)=P1R0
440 IF(IT .LT. ITE) GO TO 410
450 P(J)=R(J)*T*RGAS*Z
      IF(ISCRI .EQ. 0) WRITE(6,1700) T,L,IT,R(J),P(J),P1R0
      IF(ISCRI .EQ. 0 .AND. MOD(MP,25) .EQ. 0) WRITE(6,1400)
211 CONTINUE
490 IF(ISCRI .EQ. 0) GO TO 99
C SAVING THE FUNCTIONS
500 LAX=-1
      WMI=WMA
      CALL FENST8(MP,W,F,WMI,WMA,WED,LAX,KAXE,MASW)
      RMI=RMA
      CALL FENST8(MP,R,F,RMI,RMA,RED,LAX,KAXE,MASR)
      PMI=PMA
      CALL FENST8(MP,P,F,PMI,PMA,PED,LAX,KAXE,MASP)
      MMM=0
      NUMMER=NAPSZM(MMM)
C DENSITY ON THE SPINODALE
      KLASSE=1013 + (IP-1)*40
      CALL SDING8(KLASSE,NUMMER,MP
=,W,WMI,WMA,WED,NAMW,MASW,R,RMI,RMA,RED,NAMR,MASR,MODUS)
C PRESSURE ON THE SPINODALE
      KLASSE=1016 + (IP-1)*40
      CALL SDING8(KLASSE,NUMMER,MP
=,W,WMI,WMA,WED,NAMW,MASW,P,PMI,PMA,PED,NAMP,MASP,MODUS)
99 WRITE(6,2000)
      GO TO 100
C -----
1000 FORMAT(2H +,39(2H++)/ 2H +,77X,1H+
= /2H +,6X,15HPROGRAM HGKSPN ,33X,17HSTATUS : 94/12/09,6X,1H+
= /2H +,77X,1H+/2H +,17X
=,42HCALCULATES THE LIQUID AND VAPOR SPINODALS ,18X,1H+
= / 2H +,77X,1H+/2H +,39(2H++) )
1100 FORMAT(/11H PROPERTY /,I3,5H / =:
= /20H " 0 " STOP :
= /20H " 1 " LIQID :
= /20H " 2 " VAPOR : )
1200 FORMAT(37H FUNCTION TO BE SAVED ? ( 0 = NO ) =: )
1400 FORMAT(////)
1500 FORMAT(' KI =',I3,2X,'Y =',3G17.8/10X,'FY =',3G17.8)
1600 FORMAT(4X,'T',6X,'L',2X,'IT',7X,'RHSP',12X,'PSP',13X,'P1R0')
1700 FORMAT(F8.3,2I4,3G16.6)
2000 FORMAT(//////////)
9000 FORMAT(22H NO P.MIN FOUND AT T =,F10.3,4H !!! )

```

```

9001 FORMAT(22H NO P.MAX FOUND AT T =,F10.3,4H !!! )           HGKSPN
C -----                                                     HGKSPN
999 STOP                                                         HGKSPN
END                                                               HGKSPN

```

## 1.2 INR105.AQUA.FORT(HGKSAT)

```

PROGRAM HGKSAT                                                 HGKSAT
C CALCULATES SATURATED PROPERTIES OF THE WATER                 HGKSAT
C USING THE HGK-CODE                                           94/07/29 HGKSAT
IMPLICIT REAL*8(A-H,O-Z)                                       HGKSAT
INTEGER NEIG(50)                                               HGKSAT
REAL*8 X(1000),TS(1000),FL(1000),FV(1000),XF(1000),FX(1000)  HGKSAT
CHARACTER*8 NAMT,MAST,NAMF,MASF,NORMR,NORMW,MODUS,NAMP,MASP,NAMV HGKSAT
COMMON /MUFUN/ X,XMI,XMA,XED,FX,FMI,FMA,FED                   HGKSAT
COMMON /ACONST/ WM,GASCON,TZ,AA,ZB,DZB,YB,UREF,SREF           HGKSAT
COMMON /FCTS/ AD,GD,SD,UD,HD,CVD,CPD,DPDT,DVDT,CJTT,CJTH,Z,DPDD HGKSAT
DATA IRE/15/,IWR/16/,MODUS/'**OLD**'/,KPRI/2/,IA,IE,II/1,1000,1/ HGKSAT
=,TAUS/1.D3/                                                  HGKSAT
DATA NEIG/1,11,15,21,22,26,27,42,44,45,40*0/                 HGKSAT
WRITE(6,1000)                                                  HGKSAT
CALL SERDIO(IRE,NORMR,IWR,NORMW,MODUS,KPRI)                   HGKSAT
IF(NORMR .NE. 'GRAPHIC8') GO TO 999                           HGKSAT
IF(NORMW .NE. 'GRAPHIC8') GO TO 999                           HGKSAT
100 WRITE(6,1100)                                              HGKSAT
READ(5,*) NEP                                                  HGKSAT
IF(NEP .LE. 0) GO TO 999                                       HGKSAT
200 CALL SIGNAT(KLASSE,NUMMER,NAMF,MASF,NEIG)                 HGKSAT
IF(KLASSE .LE. 0) GO TO 100                                       HGKSAT
CALL MUSTER(IA,IE,II,NAMT,MAST,NAMP,MASP,*200)               HGKSAT
220 CALL PG3277                                               HGKSAT
WRITE(6,1200)                                                  HGKSAT
READ(5,*) ISCRI                                               HGKSAT
NAMF(5:8)='-HGK'                                             HGKSAT
NAMV=NAMF                                                       HGKSAT
MP=0                                                            HGKSAT
J=0                                                            HGKSAT
DL=1.D0                                                        HGKSAT
DV=5.D-3                                                       HGKSAT
RGAS=GASCON*TAUS                                             HGKSAT
DO 111 I=IA,IE,II                                           HGKSAT
J=J+1                                                         HGKSAT
IF(ISCRI .EQ. 0 .AND. MOD(J,25) .EQ. 1)                       HGKSAT
= WRITE(6,1500) NAMF,NAMV                                       HGKSAT
T=X(I)                                                         HGKSAT
TS(J)=T                                                       HGKSAT
CALL PCORR(T,P,DL,DV)                                         HGKSAT
IF(KLASSE .EQ. 1) THEN                                         HGKSAT
FL(J)=P*1.D+6                                                 HGKSAT
FV(J)=PS(T)*1.D6                                             HGKSAT
GO TO 110                                                       HGKSAT
ENDIF                                                           HGKSAT
IF(KLASSE .EQ. 11) THEN                                       HGKSAT

```

```

        FL(J)=DL*TAUS
        FV(J)=DV*TAUS
        GO TO 110
ENDIF
ROT = RGAS*T
ZDUM = BASE(DL,T)
CALL QQ(T,DL)
CALL THERM(DL,T)
IF(KLASSE .EQ. 15)   FL(J)=Z*ROT*DL*TAUS
IF(KLASSE .EQ. 21)   FL(J)=DPDD*TAUS
IF(KLASSE .EQ. 22)   FL(J)=DPDT*1.D6
IF(KLASSE .EQ. 26)   FL(J)=CVD*RGAS
IF(KLASSE .EQ. 27)   FL(J)=CPD*RGAS
IF(KLASSE .EQ. 42)   FL(J)=GD*ROT
IF(KLASSE .EQ. 44)   FL(J)=HD*ROT
IF(KLASSE .EQ. 45)   FL(J)=SD*RGAS
ZDUM = BASE(DV,T)
CALL QQ(T,DV)
CALL THERM(DV,T)
IF(KLASSE .EQ. 15)   FV(J)=Z*ROT*DV*TAUS
IF(KLASSE .EQ. 21)   FV(J)=DPDD*TAUS
IF(KLASSE .EQ. 22)   FV(J)=DPDT*1.D6
IF(KLASSE .EQ. 26)   FV(J)=CVD*RGAS
IF(KLASSE .EQ. 27)   FV(J)=CPD*RGAS
IF(KLASSE .EQ. 42)   FV(J)=GD*ROT
IF(KLASSE .EQ. 44)   FV(J)=HD*ROT
IF(KLASSE .EQ. 45)   FV(J)=SD*RGAS
110 IF(ISCRI .EQ. 0)  WRITE(6,1700) J,T,FX(I),FL(J),FV(J)
    IF(ISCRI .EQ. 0 .AND. MOD(J,25) .EQ. 0) WRITE(6,1400)
111 CONTINUE
    MP=J
    IF(ISCRI .EQ. 0)  GO TO 199
    LAX=-1
    XMI=XMA
    CALL FENST8(MP,X,XF,XMI,XMA,XED,LAX,KAXE,MASX)
    KLASSE=KLASSE+1000
    FMI=FMA
    CALL FENST8(MP,FL,XF,FMI,FMA,FED,LAX,KAXE,MASF)
    CALL SDING8(KLASSE,NUMMER,MP
    =,X,XMI,XMA,XED,NAMT,MAST,FL,FMI,FMA,FED,NAMF,MASF,MODUS)
    IF(KLASSE .EQ. 1001) THEN
        KLASSV=1002
    ELSE
        KLASSV=KLASSE+40
    ENDIF
    FMI=FMA
    CALL FENST8(MP,FV,XF,FMI,FMA,FED,LAX,KAXE,MASF)
    CALL SDING8(KLASSV,NUMMER,MP
    =,X,XMI,XMA,XED,NAMT,MAST,FV,FMI,FMA,FED,NAMV,MASF,MODUS)
199 WRITE(6,2000)
    GO TO 200
1000 FORMAT(2H +,39(2H+)/2H +,77X,1H+/2H +,6X,14HPROGRAM HGKSAT
    =,34X,17HSTATUS : 94/07/29,6X,1H+/2H +,77X,1H+/2H +,14X
    =,48HCALCULATES THE SATURATED PROPERTIES OF THE WATER,15X,1H+
    =/2H + ,13X,51HAS FUNCTIONS OF THE TEMPERATURE WITH THE H.G.K-CODE HGKSAT

```

```

      =,13X,1H+/2H +,7X,30HUSING A SAMPLE (T,P)-FUNCTION          HGKSAT
      =,33HAS A TEMPERATURE/PRESSURE-PATTERN,7X,1H+              HGKSAT
      =/2H +,77X,1H+/2H +,39(2H++) )                             HGKSAT
1100 FORMAT(31H NEW PROPERTY ? ( 0 = STOP ) =: )                 HGKSAT
1200 FORMAT(37H FUNCTION TO BE SAVED ? ( 0 = NO ) =: )          HGKSAT
1400 FORMAT(/////)                                              HGKSAT
1500 FORMAT(1H0,2X,1HI,8X,2HT ,8X,2HP ,11X,A8,8X,A8)           HGKSAT
1700 FORMAT(I4,F12.3,3G16.8)                                     HGKSAT
2000 FORMAT(////////////////////////////////////)                HGKSAT
  999 STOP                                                       HGKSAT
      END                                                         HGKSAT

```

### I.3 INR105.AQUA.FORT(TRNSAT)

```

      PROGRAM TRNSAT                                             TRNSAT
C     CALCULATES TRANSPORT PROPERTIES OF THE WATER              TRNSAT
C     ON THE SATURATTION LINE                                  94/12/07 TRNSAT
      IMPLICIT REAL*8(A-H,O-Z)                                   TRNSAT
      REAL*8 X(1000),F(1000),FL(1000),FV(1000)                 TRNSAT
      CHARACTER*8 NAMT,MAST,NAMF,MASF,NORMR,NORMW,MODUS         TRNSAT
      COMMON /MUFUN/ X,TMI,TMA,TED ,F,FMI,FMA,FED              TRNSAT
      COMMON /REDUF/ G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLMB,R1T0,CPC,H0 TRNSAT
      COMMON /SATIF / PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL TRNSAT
      =,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV TRNSAT
      PARAMETER(TC=647.126D0,RGAS=4.61522D+2,DC=0.322)         TRNSAT
      DATA IRE/15/,IWR/16/,MODUS/'**OLD**'/,KPRI/2/          TRNSAT
      DATA NR/0/,IA,IE,II/1,1000,1/                           TRNSAT
      WRITE(6,1000)                                             TRNSAT
      CALL SERDIO(IRE,NORMR,IWR,NORMW,MODUS,KPRI)               TRNSAT
      IF(NORMR .NE. 'GRAPHIC8')                                GO TO 999              TRNSAT
      IF(NORMW .NE. 'GRAPHIC8')                                GO TO 999              TRNSAT
100  WRITE(6,1100) IPP                                          TRNSAT
      READ(5,*) IPP                                             TRNSAT
      IF(IPP .LE. 0)                                           GO TO 999              TRNSAT
      CALL MUSTER(IA,IE,II,NAMT,MAST,NAMF,MASF,*100)          TRNSAT
      CALL PG3277                                               TRNSAT
      IF(IPP .LE. 12)     THEN                                  TRNSAT
          KLASSE=34                                             TRNSAT
          NAMF='  ETA.L'                                         TRNSAT
      ELSEIF(IPP .GE. 13)   THEN                                  TRNSAT
          KLASSE=35                                             TRNSAT
          NAMF='LAMBDA.L'                                       TRNSAT
      ENDIF                                                    TRNSAT
      WRITE(6,1200)                                             TRNSAT
      READ(5,*) ISCRI                                           TRNSAT
      MP=0                                                       TRNSAT
      DO 111 I=IA,IE,II                                         TRNSAT
      MP=MP+1                                                    TRNSAT
      IF(ISCRI .EQ. 0 .AND. MOD(MP,25) .EQ. 1)                 TRNSAT
      = WRITE(6,1500)                                           TRNSAT
      OR=F(I)                                                   TRNSAT
      T=X(I)                                                    TRNSAT
C     SATURATION PRESSURE AND DENSITIES                         TRNSAT

```

	CALL ONLYT(T,IECCH)	TRNSAT
	CALL SUBCRI(T)	TRNSAT
C	CORRECTED VALUES FOR PS,RHOL,RHOV	TRNSAT
	DLK=DL	TRNSAT
	DVK=DV	TRNSAT
	CALL GPCORR(T,Q,DLK,DVK,IECCH)	TRNSAT
C	HEAT CONDUCTIVITIES	TRNSAT
	IF(T .LE. 645.0D0) THEN	TRNSAT
C	SATURATED LIQUID	TRNSAT
	CALL REDUK(T,DLK,IECCH)	TRNSAT
	ETAL=ETA	TRNSAT
	XLAL=XLMB	TRNSAT
C	SATURATED VAPOR	TRNSAT
	CALL REDUK(T,DVK,IECCH)	TRNSAT
	ETAV=ETA	TRNSAT
	XLAV=XLMB	TRNSAT
	ELSE	TRNSAT
	CALL TWOPHA(T)	TRNSAT
	RHL=DL*1.D3	TRNSAT
	CALL TRANSP(T,RHL,PL1R0,PL1T0,ETAL,XLAL)	TRNSAT
	RHV=DV*1.D3	TRNSAT
	CALL TRANSP(T,RHV,PV1R0,PV1T0,ETAV,XLAV)	TRNSAT
	ENDIF	TRNSAT
	IF(IPP .LE. 12) THEN	TRNSAT
	FL(I)=ETAL	TRNSAT
	FV(I)=ETAV	TRNSAT
	ELSEIF(IPP .GE. 13) THEN	TRNSAT
	FL(I)=XLAL	TRNSAT
	FV(I)=XLAV	TRNSAT
	ENDIF	TRNSAT
	FF=FL(I)	TRNSAT
	IF(ISCRI .EQ. 0) THEN	TRNSAT
	RR=1.D2*(OR-FF)/OR	TRNSAT
	WRITE(6,1700) I,MP,T,OR,FF,RR	TRNSAT
	IF(MOD(MP,25) .EQ. 0) WRITE(6,1400)	TRNSAT
	ENDIF	TRNSAT
111	CONTINUE	TRNSAT
	IF(ISCRI .EQ. 0) GO TO 99	TRNSAT
	NUMMER=NAPSZM(NR)	TRNSAT
	KLASSE=1034	TRNSAT
	LAX=-1	TRNSAT
	TMI=TMA	TRNSAT
	CALL FENST8(MP,X,XF,TMI,TMA,TED,LAX,KAXE,MASX)	TRNSAT
	FMI=FMA	TRNSAT
	CALL FENST8(MP,FL,XF,FMI,FMA,FED,LAX,KAXE,MASF)	TRNSAT
	CALL SDING8(KLASSE,NUMMER,MP	TRNSAT
	=,X,TMI,TMA,TED,NAMT,MAST,FL,FMI,FMA,FED,NAMF,MASF,MODUS)	TRNSAT
	KLASSE=KLASSE+40	TRNSAT
	NAMF(8:8)='V'	TRNSAT
	FMI=FMA	TRNSAT
	CALL FENST8(MP,FV,XF,FMI,FMA,FED,LAX,KAXE,MASF)	TRNSAT
	CALL SDING8(KLASSE,NUMMER,MP	TRNSAT
	=,X,TMI,TMA,TED,NAMT,MAST,FV,FMI,FMA,FED,NAMF,MASF,MODUS)	TRNSAT
99	WRITE(6,2000)	TRNSAT
	GO TO 100	TRNSAT

1000	FORMAT(2H +,39(2H++)/ 2H +,77X,1H+/2H +,6X,14HPROGRAM TRNSAT	TRNSAT
	=,34X,17HSTATUS : 94/12/07,6X,1H+	TRNSAT
	=/2H +,77X,1H+/2H +,22X,32HCALCULATING A TRANSPORT PROPERTY	TRNSAT
	=,23X,1H+/ 2H +,77X,1H+/2H +,39(2H++ )	TRNSAT
1100	FORMAT(/11H PROPERTY /,I3,5H / =:	TRNSAT
	=/50H " 0 " STOP :	TRNSAT
	=/52H " 12 " VISCOSITY OF THE SATURATED STATES :	TRNSAT
	=/52H " 13 " THERMAL CONDUCTIVITY OF THE SAT. STATES : )	TRNSAT
1200	FORMAT(37H FUNCTION TO BE SAVED ? ( 0 = NO ) =: )	TRNSAT
1400	FORMAT(///)	TRNSAT
1500	FORMAT(/1H0,3X,1H1,3X,2HMP,6X,8HTS ( K ),5X,10HOR ( S.I ),2X	TRNSAT
	=,8HFU.APPR.,3X,13HR. DEV. ( % ) )	TRNSAT
1700	FORMAT(2I5,F12.4,2G15.6,F14.8)	TRNSAT
2000	FORMAT(////////////////////////////////////)	TRNSAT
999	STOP	TRNSAT
	END	TRNSAT

## Appendix J. The IVA Routines

The following routines are specially fitted for the codesystem IVA3, /21/. The main variables are the temperature and the pressure, the corresponding density must be gained by inverting the isotherm T. All the properties used are strictly in SI-units.

### J.1 INR105.AQUA.FORT(SAET01)

```

SUBROUTINE SAET01(T,P                                SAET01
=,VW,VD,HW,HD,HWD,SW,SD,TS1P,CPW,CPD,ETW,ETD,XLMBW,XLMBD,SRFTS SAET01
=,KIND,LCATAS)                                     SAET01
C   CALCULATES SATURATION PROPERTIES OF THE WATER SAET01
C   AS FUNCTIONS OF THE TEMPERATURE T ( IF KIND = 0 ) OR SAET01
C   AS FUNCTIONS OF THE PRESSURE P ( IF KIND = 1 ) SAET01
C   FOR THE CODESYSTEM IVA3 94/12/09 SAET01
C   Q = PRESSURE IN MEGAPASCALS, D = DENSITY IN G/CM**3 SAET01
IMPLICIT REAL*8(A-H,O-Z) SAET01
LOGICAL LCATAS SAET01
COMMON /SATIF/ PS,TPS1T,TTPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL SAET01
=,RHSV,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV SAET01
PARAMETER(TC=647.126D0,TCN=646.0D0,RGAS=4.61522D+2) SAET01
DATA UN/1.D0/,RAF/1.D3/ SAET01
LCATAS=.FALSE. SAET01
IECCH = 0 SAET01
Q=P*1.D-6 SAET01
IF(KIND .EQ. 0) GOTO 100 SAET01
C   SATURATION TEMPERATURE AND DENSITIES SAET01
CALL TCORR(T,Q,DLK,DVK,IECCH) SAET01
IF(IECCH .NE. 0) THEN SAET01
LCATAS=.TRUE. SAET01
WRITE(6,9001) SAET01
RETURN SAET01
ENDIF SAET01
GO TO 200 SAET01
100 IF(T .GT. TC) THEN SAET01
LCATAS=.TRUE. SAET01
WRITE(6,9004) T SAET01
RETURN SAET01
ENDIF SAET01
C   SATURATION PRESSURE AND DENSITIES SAET01
CALL ONLYT(T,IECCH) SAET01
IF(IECCH .NE. 0) THEN SAET01
LCATAS=.TRUE. SAET01
WRITE(6,9005) SAET01
RETURN SAET01
ENDIF SAET01
CALL SUBCRI(T) SAET01
C   CORRECTED VALUES FOR PS,RHOL,RHOV SAET01
DLK=DL SAET01
DVK=DV SAET01
IF(T .GE. TCN) GOTO 200 SAET01

```



	CALL GPCORR(T,Q,DLK,DVK,IECCH)	SAET01
C	OTHER PROPERTIES OF THE SATURATED STATES	SAET01
C	VAPOR PRESSURE DERIVATIVE T*DPS/DT	SAET01
200	ROT=T*RGAS	SAET01
	TS1P=T/TPS1T	SAET01
	CALL TWOPHA(T)	SAET01
C	LIQUID PROPERTIES	SAET01
	RHL=DLK*RAF	SAET01
	VW=UN/RHL	SAET01
	GW=GL*ROT	SAET01
	SW=SL*RGAS	SAET01
	HW=GW+SW*T	SAET01
	CPL=CVL+PL1T0*PL1T0/PL1R0	SAET01
	CPW=CPL*RGAS	SAET01
	ETW=ETL	SAET01
	XLMBW=XLL	SAET01
	SRFTS=SIGMA(T)	SAET01
C	VAPOR PROPERTIES	SAET01
	RHV=DVK*RAF	SAET01
	VD=UN/RHV	SAET01
	GD=GW	SAET01
	HWD=TPS1T*(VD-VW)	SAET01
	SD=SW+HWD/T	SAET01
	HD=HW+HWD	SAET01
	CPV=CVV+PV1T0*PV1T0/PV1R0	SAET01
	CPD=CPV*RGAS	SAET01
	ETD=ETV	SAET01
	XLMBD=XLV	SAET01
9001	FORMAT(40H SAET01-ERROR. PRESSURE RANGE EXCEEDED. )	SAET01
9004	FORMAT(' T =',F12.2,' IS A SUPERCRITICAL TEMPERATURE !' )	SAET01
9005	FORMAT(42H SAET01-ERROR. TEMPERATURE BELOW FREEZING. )	SAET01
9008	FORMAT(39H SAET01-ERROR. DENSITY RANGE EXCEEDED. )	SAET01
	RETURN	SAET01
	END	SAET01

## J.2 INR105.AQUA.FORT(WASSER)

	SUBROUTINE WASSER(T,P	WASSER
	=,RW,HW,SW,CPW,VSW,ETW,XLMBW,PRNW,HW1P,RW1T,RW1P,SRFTS, LCATAW)	WASSER
C	CALCULATES THERMOPHYSICAL PROPERTIES OF THE WATER	WASSER
C	AS FUNCTIONS OF THE TEMPERATURE AND PRESSURE	WASSER
C	FOR THE CODESYSTEM IVA3	94/12/07 WASSER
C	Q = PRESSURE IN MEGAPASCALS, D = DENSITY IN G/CM**3	WASSER
	IMPLICIT REAL*8(A-H,O-Z)	WASSER
	LOGICAL LCATAW	WASSER
	COMMON /REDUF/ G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLMB,R1T0,CPC,H0	WASSER
	PARAMETER(TC=647.126D0,RGAS=4.61522D+2)	WASSER
	DATA ZERO/0.D0/,UN/1.D0/	WASSER
	LCATAW=.FALSE.	WASSER
	Q=P*1.D-6	WASSER
	IECCH=0	WASSER
	CALL ONLYT(T,IECCH)	WASSER
	IF(IECCH .NE. 0) THEN	WASSER

```

        WRITE(6,9001)
        LCATAW=.TRUE.
        GO TO 99
ENDIF
C   CALCULATING THE DENSITY
    IF(T .GE. TC)      CALL DGFIND(T,Q,DX,IECCH)
    IF(T .LT. TC)      CALL DLFIND(T,Q,DX,IECCH)
    IF(IECCH .NE. 0)   THEN
        IF(IECCH .LT. 0) WRITE(6,9005)
        IF(IECCH .GT. 0) WRITE(6,9006)
        LCATAW=.TRUE.
        GO TO 99
ENDIF
    RW=DX*1.D3
    V=UN/RW
C   PROPERTIES AS TEMPERATURE-DENSITY-FUNCTIONS
    ROT=T*RGAS
    CALL REDUK(T,DX,IECCH)
C   WRITE(6,'(16H PREDUK IECCH =,I6)' ) IECCH
    IF(P1R0 .LE. ZERO) THEN
        WRITE(6,9003)
        LCATAW=.TRUE.
        GO TO 99
ENDIF
    HW=H0*ROT
    SW=S0*RGAS
    CPW=CP0*RGAS
    P1R=P1R0*ROT
    ETW=ETA
    XLMBW=XLMB
C   VELOCITY OF SOUND
    P1RS=CPC*P1R
    VSW=SQRT(P1RS)
C   DH/DP , DRHO/DT , DRHO/DP
    HW1P=(UN+R1T0)*V
    RW1T=(RW/T)*R1T0
    RW1P=UN/P1R
C   PRANDTL # , SURFACE TENSION
    PRNW=CPW*ETW/XLMB
    SRFTS=0.D0
    IF(T .LT. TC)      SRFTS=SIGMA(T)
9001 FORMAT(34H WASSER-ERROR. TEMPERATURE TO LOW. )
9002 FORMAT(55H WASSER-ERROR. NO WATER DENSITY FOUND AT THIS PRESSURE.)
9003 FORMAT(45H WASSER-ERROR. DP/DRHO OF THE WATER VANISHES. )
9005 FORMAT(51H WASSER-ERROR. PRESSURE TO LOW FOR A WATER DENSITY. )
9006 FORMAT(52H WASSER-ERROR. PRESSURE TO HIGH FOR A WATER DENSITY. )
99  RETURN
END

```

### J.3 INR105.AQUA.FORT(DAMPF)

```

SUBROUTINE DAMPF(T,P,RGAS,WGM                                DAMPF
=,RD,HD,SD,CPD,VSD,ETD,XLMBD,PRND,HD1P,RD1T,RD1P, LCATAD) DAMPF
C CALCULATES THERMOPHYSIVCAL PROPERTIES OF THE STEAM      DAMPF
C AS FUNCTIONS OT THE TEMPERATURE AND PRESSURE ( IF RD = 0 ) DAMPF
C OR                                                       DAMPF
C AS FUNCTIONS OT THE TEMPERATURE AND DENSITY ( IF RD > 0 ) DAMPF
C FOR THE CODESYSTEM IVA3                                94/12/07 DAMPF
C Q = PRESSURE IN MEGAPASCALS, D = DENSITY IN G/CM**3      DAMPF
IMPLICIT REAL*8(A-H,O-Z)                                  DAMPF
LOGICAL LCATAD                                            DAMPF
COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW DAMPF
COMMON /REDUF / G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLMB,R1T0,CPC,H0 DAMPF
DATA ZERO/0.D0/,UN/1.D0/                                  DAMPF
LCATAD=.FALSE.                                           DAMPF
RGAS=GASCON*1.D3                                         DAMPF
WGM=WM                                                    DAMPF
Q=P*1.D-6                                                 DAMPF
IECCH=0                                                   DAMPF
CALL ONLYT(T,IECCH)                                       DAMPF
IF(IECCH .NE. 0) THEN                                     DAMPF
WRITE(6,9001)                                             DAMPF
LCATAD=.TRUE.                                            DAMPF
GO TO 90                                                 DAMPF
ENDIF                                                    DAMPF
DX=RD*1.D-3                                              DAMPF
IF(RD .GT. ZERO) GO TO 50                                DAMPF
C CALCULATING THE DENSITY                                  DAMPF
IF(T .GE. TC) CALL DGFIND(T,Q,DX,IECCH)                 DAMPF
IF(T .LT. TC) CALL DVFIND(T,Q,DX,IECCH)                 DAMPF
IF(IECCH .NE. 0) THEN                                     DAMPF
IF(IECCH .LT. 0) WRITE(6,9005)                            DAMPF
IF(IECCH .GT. 0) WRITE(6,9006)                            DAMPF
LCATAD=.TRUE.                                            DAMPF
GO TO 90                                                 DAMPF
ENDIF                                                    DAMPF
RD=DX*1.D3                                              DAMPF
50 V=UN/RD                                               DAMPF
C PROPERTIES AS TEMPERATURE-DENSITY-FUNCTIONS           DAMPF
ROT=T*RGAS                                               DAMPF
CALL REDUK(T,DX,IECCH)                                    DAMPF
IF(P1R0 .EQ. ZERO) THEN                                   DAMPF
WRITE(6,9003)                                             DAMPF
LCATAD=.TRUE.                                            DAMPF
GO TO 90                                                 DAMPF
ENDIF                                                    DAMPF
IF(P .EQ. ZERO) P=Z*ROT*RD                                DAMPF
HD=H0*ROT                                                DAMPF
SD=S0*RGAS                                               DAMPF
CPD=CP0*RGAS                                             DAMPF
P1R=P1R0*ROT                                             DAMPF
ETD=ETA                                                  DAMPF

```

```

        XLMBD=XLMB
C      VELOCITY OF SOUND
        P1RS=CPC*P1R
        VSD=SQRT(P1RS)
C      DH/DP , DRHO/DT , DRHO/DP
        HD1P=(UN+R1T0)*V
        RD1T=(RD/T)*R1T0
        RD1P=UN/P1R
C      PRANDTL #
        PRND=CPD*ETD/XLMB
9001  FORMAT(33H DAMPF-ERROR. TEMPERATURE TO LOW. )
9002  FORMAT(54H DAMPF-ERROR. NO STEAM DENSITY FOUND AT THIS PRESSURE. )
9003  FORMAT(44H DAMPF-ERROR. DP/DRHO OF THE STEAM VANISHES. )
9005  FORMAT(50H DAMPF-ERROR. PRESSURE TO LOW FOR A STEAM DENSITY. )
9006  FORMAT(51H DAMPF-ERROR. PRESSURE TO HIGH FOR A STEAM DENSITY. )
90  RETURN
      END

```

#### J.4 INR105.AQUA.FORT(RDRDP)

```

      SUBROUTINE RDRDP(T,P,RD,RD1P,LCATAD)
C      RETURNS FOR A GIVEN TEMPERATURE TX AND PRESSURE P
C      THE DENSITY, RD AND THE DENSITY DERIVATIVE, RD1P = DRD/DP
C      OF THE STEAM FOR THE CODESYSTEM IVA3.
C      Q = PRESSURE IN MEGAPASCALS, D = DENSITY IN G/CM**3
      IMPLICIT REAL*8(A-H,O-Z)
      LOGICAL LCATAD
      COMMON /AKONST/ WM,TC,DC,GASCON,G1,G2,GF,TZ,UREF,SREF,TLOW
      COMMON /REDUF/  G0,S0,Z,P1T0,P1R0,CV0,CP0,ETA,XLAMB,R1T0,CPC,H0
      DATA ZERO/0.D0/,UN/1.D0/
      LCATAD=.FALSE.
      RGAS=GASCON*1.D3
      Q=P*1.D-6
      IECCH=0
      CALL ONLYT(T,IECCH)
      IF(IECCH .NE. 0) THEN
        WRITE(6,9001)
        LCATAD=.TRUE.
        GO TO 90
      ENDIF
C      CALCULATING THE DENSITY
      IF(T .GE. TC) CALL DGFIND(T,Q,DX,IECCH)
      IF(T .LT. TC) CALL DVFIND(T,Q,DX,IECCH)
      IF(IECCH .NE. 0) THEN
        IF(IECCH .LT. 0) WRITE(6,9005)
        IF(IECCH .GT. 0) WRITE(6,9006)
        LCATAD=.TRUE.
        GO TO 90
      ENDIF
      RD=DX*1.D3
      IF(DX .LT. ZERO) THEN
        WRITE(6,9002)
        LCATAD=.TRUE.

```

```

          GO TO 90
        ENDIF
C      PROPERTIES AS TEMPERATURE-DENSITY-FUNCTIONS
50    CALL REDUK(T,DX,IECCH)
      IF(P1R0 .EQ. ZERO) THEN
        WRITE(6,9003)
        LCATAD=.TRUE.
        GO TO 90
      ENDIF
      P1R=P1R0*T*RGAS
C      DRHO/DP
      RD1P=UN/P1R
9001  FORMAT(33H RDRDP-ERROR. TEMPERATURE TO LOW. )
9002  FORMAT(54H RDRDP-ERROR. NO STEAM DENSITY FOUND AT THIS PRESSURE. )
9003  FORMAT(44H RDRDP-ERROR. DP/DRHO OF THE STEAM VANISHES. )
9005  FORMAT(50H RDRDP-ERROR. PRESSURE TO LOW FOR A STEAM DENSITY. )
9006  FORMAT(51H RDRDP-ERROR. PRESSURE TO HIGH FOR A STEAM DENSITY. )
90    RETURN
      END

```

### J.5 INR105.AQUA.FORT(VD1)

```

      SUBROUTINE VD1(T,VD,DUMMY)
C      VD1 CALCULATES - FOR A GIVEN TEMPOERATURE ( T ) IN K -
C      AN APPROXIMATED SPECIFIC VOLUME OF THE SATURATED STEAM ( VD )
C      IN M**3/KG. VD1 IS A SUBROUTINE OF THE IVA3-CODE.      94/12/01
      IMPLICIT REAL*8(A-H,O-Z)
      DIMENSION PU(12)
      COMMON /SATIF/ PS,TPS1T,TPS2T,DL,DV,RL1T0,RV1T0,RHSL,PSL
      =,RHSL,PSV,SL,GL,PL1R0,PL1T0,CVL,ETL,XLL,PV1R0,PV1T0,CVV,ETV,XLV
      PARAMETER(TC=647.126D0,RHC=0.322D3,RGAS=4.61522D+2,TSW=620.D0)
      PARAMETER(BR=2.03D0,BET=0.325D0)
      DATA PU/-1.125999398D+0,42.15284538D+0,-538.0831003D+0
      =,4.150173161D+3,-20.80947751D+3,71.29507069D+3,-170.4446088D+3
      =,284.4989747D+3,-325.1357165D+3,242.4593090D+3,-106.2552030D+3
      =,20.74762670D+3/, UN/1.D0/
      DUMMY = 0.D0
      IF(T .GE. TC) THEN
        DUMMY = -1.D0
        VD = UN/RHC
        RETURN
      ENDIF
      Y = (TC-T)/TC
      IF(T .LE. TSW) THEN
C      LOW TEMPERATURES
C      FACTOR OF REALITY OF THE VAPOR
        W = SQRT(Y)
        CALL SSP(T)
        ZV = PU(12)
        DO 31 I=11,1,-1
31      ZV = ZV*W+PU(I)
        VD = RGAS*T*ZV/PS
C      NEAR CRITICAL TEMPERATURES

```

```
ELSE VD1
  AV = Y*(1.451962241D-0-Y*(5.041325559D-0-Y*Y*611.1810749D-0)) VD1
  OMV = UN-BR*Y**BET+AV VD1
  VD = UN/(RHC*OMV) VD1
ENDIF VD1
RETURN VD1
END VD1
```



## Appendix K. Service routines

The following routines are needed only by the 'string'-routines KAISOT and KAISOK

### K.1 INR105.AQUA.FORT(SIGNAT)

SIGNAT creates the labels and distinguishing names for the properties calculated.

The first label of the property, KLASSE will be set according to the Table 38 on page 156, the second label, NUMMER will be equated with the date of the calculation.

```

SUBROUTINE SIGNAT(KLASSE,NUMMER,NAMF,MASF,NEIG)          SIGNAT
C  SIGNAT SETS THE CHARACTERISTICA OF A SERVUS-FUNCTION SIGNAT
C  THERMAL PROPERTIES                                     94/12/09 SIGNAT
IMPLICIT REAL*8(A-H,O-Z)                                SIGNAT
DIMENSION NEIG(50)                                       SIGNAT
CHARACTER*8 NAMF,MASF                                     SIGNAT
DATA NR/0/                                               SIGNAT
NUMMER=NAPSZM(NR)                                        SIGNAT
MASF='SI-UNIT'                                           SIGNAT
CALL PG3277                                              SIGNAT
WRITE(6,1000)                                           SIGNAT
KLASSE=MOD(KLASSE,100)                                   SIGNAT
100 WRITE(6,1100) KLASSE                                  SIGNAT
READ(5,*) KLASSE                                         SIGNAT
IF(KLASSE .EQ. 0)          GO TO 999                      SIGNAT
MP=0                                                       SIGNAT
DO 111 I=1,50                                           SIGNAT
IF(NEIG(I) .EQ. 0)      GO TO 190                        SIGNAT
IF(NEIG(I) .EQ. KLASSE) GO TO 200                      SIGNAT
111 MP=MP+1                                              SIGNAT
190 KAPRO = 0                                            SIGNAT
WRITE(6,1300) (NEIG(I),I=1,MP)                          SIGNAT
GO TO 100                                                SIGNAT
C.....                                                  SIGNAT
200 IF(KLASSE .EQ. 1 .OR. KLASSE .EQ. 2 .OR.           SIGNAT
= KLASSE .EQ. 15 .OR. KLASSE .EQ. 16) MASF='PASCAL '   SIGNAT
IF(KLASSE .EQ. 3 .OR. KLASSE .EQ. 22) MASF='PASCAL/K'  SIGNAT
IF(KLASSE .EQ. 4) MASF='K/PASCAL '                     SIGNAT
IF(KLASSE .EQ. 5) MASF='K '                            SIGNAT
IF(KLASSE .GE. 11 .AND. KLASSE .LE. 13) MASF='KG/M**3 ' SIGNAT
IF(KLASSE .EQ. 17) MASF='M**3/KG '                     SIGNAT
IF(KLASSE .EQ. 18 .OR. KLASSE .EQ. 36) MASF='- 1 - '   SIGNAT
IF(KLASSE .EQ. 26 .OR. KLASSE .EQ. 27 .OR.           SIGNAT
= KLASSE .EQ. 45) MASF='J/KG/K '                       SIGNAT
IF(KLASSE .EQ. 31) MASF='M/SEC '                       SIGNAT
IF(KLASSE .EQ. 32) MASF='N/M '                         SIGNAT
IF(KLASSE .EQ. 34) MASF='PA SEC '                      SIGNAT
IF(KLASSE .EQ. 35) MASF='W/(K*M) '                    SIGNAT
IF(KLASSE .GE. 41 .AND. KLASSE .LE. 44) MASF='J/KG '   SIGNAT
C.....                                                  SIGNAT
IF(KLASSE .EQ. 1) NAMF=' P* '                          SIGNAT
IF(KLASSE .EQ. 2) NAMF='P*AX '                        SIGNAT

```



```

IF(KLASSE .EQ. 3)      NAMF=' DP*/DT  '      SIGNAT
IF(KLASSE .EQ. 4)      NAMF=' DT*/DP  '      SIGNAT
IF(KLASSE .EQ. 5)      NAMF='  T*    '      SIGNAT
C.....
IF(KLASSE .EQ. 11)     NAMF='  RH????'      SIGNAT
IF(KLASSE .EQ. 12)     NAMF=' RHAX????'      SIGNAT
IF(KLASSE .EQ. 13)     NAMF='  RHS????'      SIGNAT
IF(KLASSE .EQ. 15)     NAMF='  P????'      SIGNAT
IF(KLASSE .EQ. 16)     NAMF='  PS????'      SIGNAT
IF(KLASSE .EQ. 17)     NAMF='  V????'      SIGNAT
IF(KLASSE .EQ. 18)     NAMF='  Z????'      SIGNAT
C.....
IF(KLASSE .EQ. 21)     NAMF='  PR????'      SIGNAT
IF(KLASSE .EQ. 22)     NAMF='  PT????'      SIGNAT
IF(KLASSE .EQ. 23)     NAMF='  RHP????'      SIGNAT
IF(KLASSE .EQ. 24)     NAMF='  RHT????'      SIGNAT
IF(KLASSE .EQ. 26)     NAMF='  CV????'      SIGNAT
IF(KLASSE .EQ. 27)     NAMF='  CP????'      SIGNAT
C.....
IF(KLASSE .EQ. 31)     NAMF='  VS????'      SIGNAT
IF(KLASSE .EQ. 32)     NAMF=' SIGMA  '      SIGNAT
IF(KLASSE .EQ. 34)     NAMF='  ETA????'      SIGNAT
IF(KLASSE .EQ. 35)     NAMF='  LAM????'      SIGNAT
IF(KLASSE .EQ. 36)     NAMF='  PRN????'      SIGNAT
C.....
IF(KLASSE .EQ. 41)     NAMF='  A????'      SIGNAT
IF(KLASSE .EQ. 42)     NAMF='  G????'      SIGNAT
IF(KLASSE .EQ. 43)     NAMF='  U????'      SIGNAT
IF(KLASSE .EQ. 44)     NAMF='  H????'      SIGNAT
IF(KLASSE .EQ. 45)     NAMF='  S????'      SIGNAT
IF(KLASSE .EQ. 46)     NAMF='  UR????'      SIGNAT
IF(KLASSE .EQ. 47)     NAMF='  HP????'      SIGNAT
1000 FORMAT(//44H TABLE OF THE PROPERTY-NUMBERS : ( END = 0 )      SIGNAT
=/' | 1 2 3 4 5 6 7 8 ' SIGNAT
=/' SIGNAT
=/' 0 | P* P*~ DP/DT DT/DP T* SIGNAT
=/' 10 | RH RH~ RH.SP P P.SP V Z SIGNAT
=/' 20 | P1R P1T R1P R1T CV CP SIGNAT
=/' 30 | V.S SIG ETA LAMBDA PRN SIGNAT
=/' 40 | A G U H S U1R H1P SIGNAT
1100 FORMAT(31H ENTER NUMBER OF THE PROPERTY /,I4,5H / =: ) SIGNAT
1300 FORMAT(50H ONLY THE FOLLOWING PROPERTIES CAN BE CALCULATED : SIGNAT
=/(5X,18(I3, :, 1H,)) SIGNAT
999 RETURN SIGNAT
END SIGNAT

```

## K.2 INR105.AQUA.FORT(MUSTER)

MUSTER helps to select a sample function from a SERVUS dataset.

	SUBROUTINE MUSTER(IA,IE,II,NAMT,MAST,NAMF,MASF,*)		MUSTER
C	SELECTS A SAMPLE FUNCTION F ( T )	94/07/29	MUSTER
	IMPLICIT REAL*8(A-H,O-Z)		MUSTER
	REAL*8 T(1000),F(1000)		MUSTER
	CHARACTER*8 NAMT,MAST,NAMF,MASF		MUSTER
	COMMON /MUFUN/ T,TMI,TMA,TED,F,FMI,FMA,FED		MUSTER
	DATA LOK,LOS/0,0/		MUSTER
100	WRITE(6,1000) LOS		MUSTER
	READ(5,*) LOS		MUSTER
	IF(LOS .LE. 0) RETURN 1		MUSTER
	CALL SDEXG8(KLASSE,NUMMER,MP		MUSTER
	=,T,TMI,TMA,TED,NAMT,MAST,F,FMI,FMA,FED,NAMF,MASF,LOS)		MUSTER
	WRITE(6,1100)		MUSTER
	READ(5,*) LOK		MUSTER
	IF(LOK .NE. 0) GO TO 100		MUSTER
	IE=MAX(IE,MP)		MUSTER
120	WRITE(6,1300) IA,IE,II		MUSTER
	READ(5,*) IA,IE,II		MUSTER
	IF(IA*IE*II .EQ. 0) RETURN 1		MUSTER
	IE=MIN(IE,MP)		MUSTER
	IA=MAX(IA,1)		MUSTER
C			MUSTER
1000	FORMAT(52H ENTER SERIAL NUMBER OF THE (X,FX)-MASTER-FUNCTION /		MUSTER
	=,I4,19H / =: ( 0 = END ) )		MUSTER
1100	FORMAT(36H FUNCTION OK =: ( 0 = YES   9 = NO ) )		MUSTER
1300	FORMAT(45H ENTER BEGIN, END & INCREMENTUM ( 0 = END ) /		MUSTER
	=,I3,2H ,,I4,2H ,,I3,16H / =: , =: , =: )		MUSTER
	RETURN		MUSTER
	END		MUSTER

## K.3 INR105.AQUA.FORT(NAPSZM)

NAPSZM turns the actual date into a number, the number of the day in the year.

	INTEGER FUNCTION NAPSZM(K)		NAPSZM
C	REURNS THE NUMBER OF THE DAY	92/01/13	NAPSZM
	CHARACTER JAN*64		NAPSZM
	CALL JOBINF(JAN)		NAPSZM
	NAPSZM=K-K		NAPSZM
	JUF=1		NAPSZM
	DO 11 J=32,30,-1		NAPSZM
	I=ICHAR(JAN(J:J))		NAPSZM
	IF(I .GT. 240 .AND. I .LT. 250) NAPSZM=NAPSZM+(I-240)*JUF		NAPSZM
11	JUF=JUF*10		NAPSZM
	RETURN		NAPSZM
	END		NAPSZM



## Appendix L. Return codes

IECCH	Reason
0	density found
-1,-2	pressure to low for a density
1, 2	pressure to high for a density
3	iteration failed
-4	temperature to low for any density
4	density too high for any temperature



**Appendix M. TSO-procedures, calculating thermal properties of the water**

Table 37. TSO-procedures in the dataset INR105.AQUA.CLIST			
Member	Routine	Procedure	Dataset needed
WATER	WATER	Water properties point by point / results = > dataset	-
KAISOT	KAISOT	Explores water properties along isochores	AQUAT.DATA
KAISOK	KAISOK	Explores water properties along isotherms	AQUA.DATA
HGK	NBSNRC	Water properties point by point / H.G.K. code	-
HGKIST	HGKIST	Explores water properties along isotherms / H.G.K. code	AQUA.DATA
HGKSAT	HGKSAT	Calculates properties in the saturated states / H.G.K. code	AQUAT.DATA
HGKSPN	HGKSPN	Calculates the spinodals / H.G.K. code	AQUAT.DATA
TRNSAT	TRNSAT	Calculates transport properties in the saturated states	AQUAT.DATA
TCORR	\$TCORR	Checks the routines TCORR, GPCORR	-
TDFIND	\$TDFND	Checks the routines DGFIND, DLFIND, DVFIND	-
TAXSL	\$TAXSL	Checks properties in the saturated states	AQUAT.DATA
TWASSER	\$TWSSR	Checks the routine WASSER	AQUA.DATA
TDAMPF	\$TDMPF	Checks the routine DAMPF	AQUA.DATA
TST01	\$TST01	Checks the routine SAET01	AQUAT.DATA
TSTEX	\$TSTEX	Checks the routine SAETEX	AQUAT.DATA

## Appendix N. Thermal property labels



Table 38. Thermal property labels "KLASSE" used in AQUA.DATA										
	1	2	3	4	5	6	7	8	9	10
0	$P^*$	$P^*_z$	$dP^*/dT$	$dT/dP^*$	$T^*$					
10	$\rho_L$	$\rho_{Lz}$	$\rho_{SL}$		$P_L$	$P_{SL}$	$v_L$	$Z_L$		
20	$\partial P/\partial \rho_L$	$\partial P/\partial T_L$	$\partial \rho/\partial P_L$	$\partial \rho/\partial T_L$		$C_{vL}$	$C_{pL}$			
30	$v_{sL}$	$\sigma$		$\eta_L$	$\lambda_L$	$PrN_L$				
40	A.L	G.L	U.L	H.L	S.L	$\partial U/\partial \rho_L$	$\partial H/\partial P_L$			
50	$\rho_v$	$\rho_{vz}$	$\rho_{sv}$		$P_v$	$P_{sv}$	$v_v$	$Z_v$		
60	$\partial P/\partial \rho_v$	$\partial P/\partial T_v$	$\partial \rho/\partial P_v$	$\partial \rho/\partial T_v$		$C_{vv}$	$C_{pv}$			
70	$v_{sv}$			$\eta_v$	$\lambda_v$	$PrN_v$				
80	A.V	G.V	U.V	H.V	S.V	$\partial U/\partial \rho_v$	$\partial H/\partial P_v$			